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Two Dimensional Layered Co_{0.85}Se Nanosheets as a high-capacity anode

for Lithium-ion batteries

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Figure S1. EDS spectrum of layered $Co_{0.85}Se$ nanosheets.



Figure S2. (a) TEM image of layered $Co_{0.85}Se$ nanosheets and (b) HRTEM image of Co0.85Se nanosheets in the selected area in the blue box of image (a).



Figure S3. XRD patterns of the samples prepared at various mole ratios of precursorsCo(AC)₂/Na₂SeO₃: (a) 1:0.5, (b) 1:0.75 and (c) 1:1.5.



Figure S4. (a) Initial discharge/charge curves of $Co_{0.85}$ Se nanosheets at the current density of 100 mAg⁻¹; XRD patterns of $Co_{0.85}$ Se nanosheet electrode at various voltage of (b) 1.1 V, (c) 0.01 V, and (d) 3.0 V; and HRTEM image of $Co_{0.85}$ Se nanosheet at (e) 1.1 V and (f) 3.0 V.



Figure S5. Nitrogen adsorption-desorption isotherms of $Co_{0.85}Se$ nanosheets and $Co_{0.85}Se$ microspheres.



Figure S6. Equivalent circuit: R_e is the electrolyte resistance; C_f and R_f are the capacitance and resistance of the surface film formed on the electrodes, respectively; C_{dl} and R_{ct} are the double-layer capacitance and charge-transfer resistance, respectively; Z_w is the Warburg impedance related to the diffusion of lithium ions into the bulk electrodes.



Figure S7. SEM images of (a, b) $Co_{0.85}Se$ nanosheets electrode and (c, d) $Co_{0.85}Se$ microspheres electrodes.