# Designed Formation through Metal Organic Frameworks Route of ZnO/ZnCo<sub>2</sub>O<sub>4</sub> Hollow Core-shell Nanocages with Enhanced Gas Sensing Properties

Fengdong Qu, Huifang Jiang and Minghui Yang<sup>\*</sup>

<sup>†</sup>Dalian Institute of Chemical Physics, Chinese Academy of Sciences, Dalian 116023, P. R. China.

E-mail: myang@dicp.ac.cn

### **Experimental Section**

#### Synthesis of ZnCo<sub>2</sub>O<sub>4</sub> shells

The ZnCo<sub>2</sub>O<sub>4</sub> shells were synthesized according to a previous report with a modification. Typically, 0.04g of ZIF-8/Co–Zn hydroxides precursor was first dispersed in a mixed solvent consisting of 20 mL ethanol and refluxed at 90 °C for 1 h, during which the ZIF-8 cores can be completely removed. Then, the as-obtained Co–Zn hydroxides shells were annealed in air at 450 °C for 3 h with a ramp rate of 1 °C min<sup>-1</sup> to generate the  $ZnCo_2O_4$  shells.

# Supplementary Figures



Fig. S1 The schematic illustration of (a) gas sensing analysis system and (b) gas mixing line equipment.





Fig. S3 XRD pattern, SEM images and TEM images of ZIF-8.



The following equations may describe the reaction process of Co(OH)<sub>2</sub> and Zn(OH)<sub>2</sub>:<sup>1</sup> Co(NO<sub>3</sub>)<sub>2</sub>· $^{6}H_{2}O \rightarrow Co^{2^{+}} + 2NO_{3}^{-} + 6H_{2}O$  (1)  $H_{2}O \rightarrow OH^{-} + H^{+}$  (2)  $Co^{2^{+}} + 2OH^{-} \rightarrow Co(OH)_{2} \downarrow$  (3)  $Zn^{2^{+}} + 2OH^{-} \rightarrow Zn(OH)_{2} \downarrow$  (4)



Fig. S5 SEM images and TEM images of ZnO. XRD pattern,



Fig. S6 XRD pattern, SEM images and TEM images of  $ZnCo_2O_4$  shells.



Fig. S7 The dynamic 3-cycles response measurements to 100 ppm xylene for  $ZnO/ZnCo_2O_4$  HCSNCs at 320 °C.



Fig. S8 The  $S_{wet}/S_{dry}$  of ZnO/ZnCo<sub>2</sub>O<sub>4</sub> HCSNCs exposed to 100 ppm xylene in dry and humid conditions (r.h. 20%, 50%, and 80%).

## **Reference:**

[1] Jiang Z, Li Z, Qin Z, et al. LDH nanocages synthesized with MOF templates and their high performance as supercapacitors. Nanoscale, 2013, 5(23): 11770-11775.