**Support Information** 

## Highly fluorescent Zn-doped carbon dots as Fenton reaction-based

## bio-sensor: An integrative experimental-theoretical consideration

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Figure S1. The quantum yield results by (a) ratio of zinc chloride to sodium citrate; (b) temperature of the hydrothermal reaction (from 140 to  $210^{\circ}$ C);(c) time of hydrothermal reaction.

The quantum yield of PLQY was determined in a standard method.[1] And the absolute PLQY value of CDs was calculated by the following formula:

$$\phi_{X} = \phi_{ST} \left( \frac{Grad_{X}}{Grad_{ST}} \right) \left( \frac{\eta_{X}^{2}}{\eta_{ST}^{2}} \right)$$

Where the subscripts ST and X denote standard and test, respectively.  $\Phi$  is the fluorescence quantum yield. Grad is the gradient from the plot of integrated fluorescence intensity vs absorbance. And  $\eta$  is the refractive index of the solvent.



Figure S4. Photoluminescence spectrum of the Zn-CDs



Figure S5. UV absorption spectrum of Zn-CDs



Figure S6. QY according to different Zn %



Figure S7 High resolution TEM picture of Zn-CDs



Figure S8 TEM-EDX analysis of Zn-CDs



Figure S9 EDX mapping analysis of Zn-CDs

| ratio   | Zn wt% |
|---------|--------|
| 1:0.25  | 6.5%   |
| 1 : 0.5 | 7.9%   |
| 1:0.75  | 11.6%  |
| 1:1     | 13.2%  |

Table S1 Zn wt% content tested by ICP

It is found that the PL quantum yield reach the maximum when the molar ratio for sodium citrate and zinc chloride is 1:0.5. Compared with other reported CDs doped with heteroatom,[1-3] by testing the Zn content with ICP (Table S1), the as-prepared Zn-doped carbon dots has a high doping ratio (from 6% to 9%)

Supplementary references

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