

Supplementary Information for
**Topological insulating states in 2D transition metal dichalcogenides
induced by defects and strain**

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Supplementary Tables

Table S1. Structural parameters and band gap of sho-TMD.

Table S2. Structural parameters, total energy, and band gap of shⁿo-MoS₂ ($n = 1-6$).

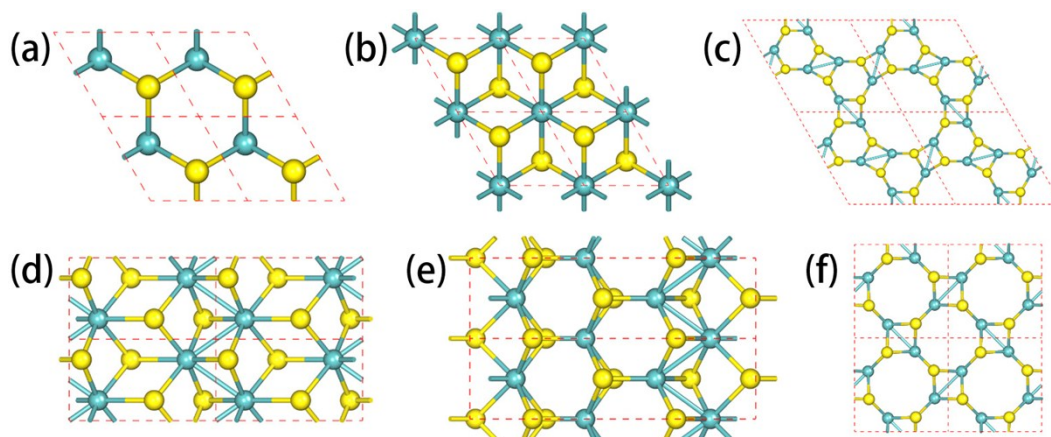


Figure S1. Optimized structures of (a) H-MoS₂, (b) 1T-MoS₂, (c) H'-MoS₂, (d) 1T'-MoS₂, (e) T''-MoS₂ and (f) so-MoS₂.

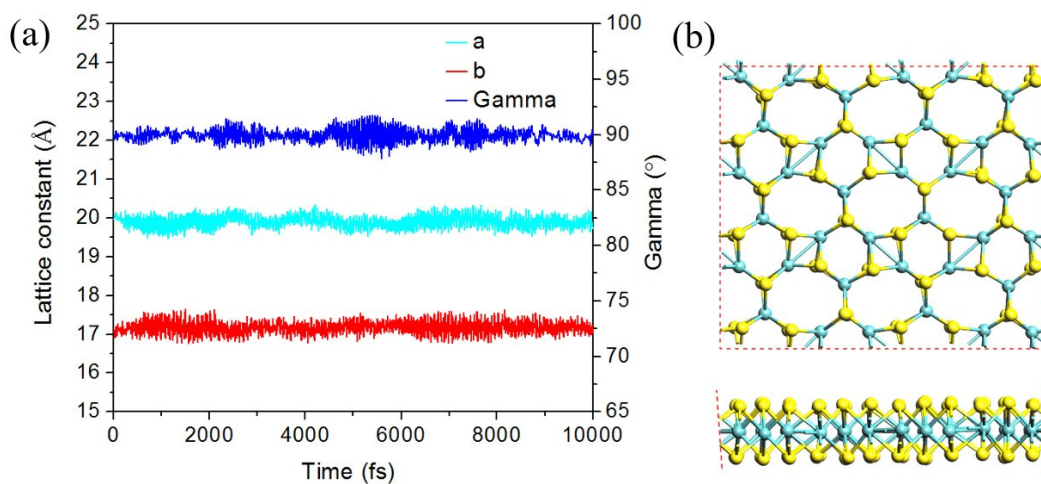


Figure S2. (a) Evolution of lattice parameters a , b , and γ of sho-MoS₂ during the NpT-AIMD simulation at temperature of 300 K. (b) Snapshots of the structure (2×3 supercell) at the end of the simulation.

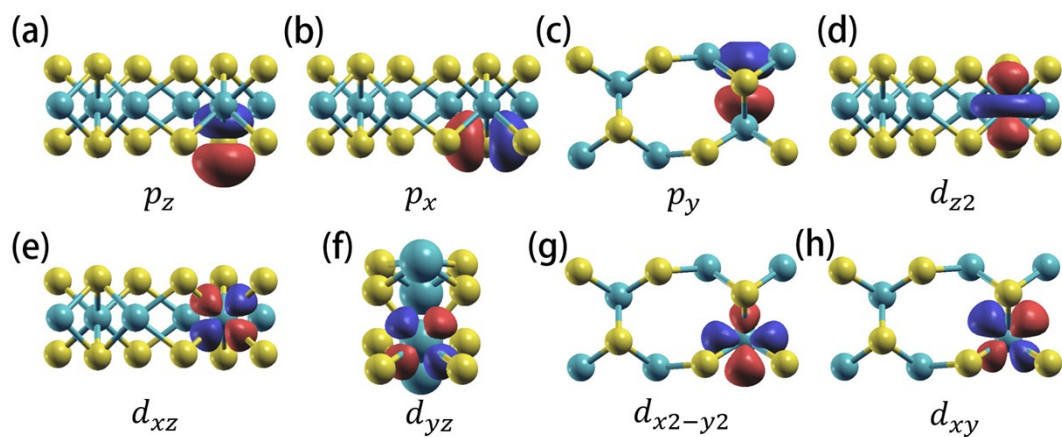


Figure S3. Maximally Localized Wannier functions (MLWFs) of sho-MoS₂, including three S-3*p* orbitals and five Mo-4*d* orbitals.

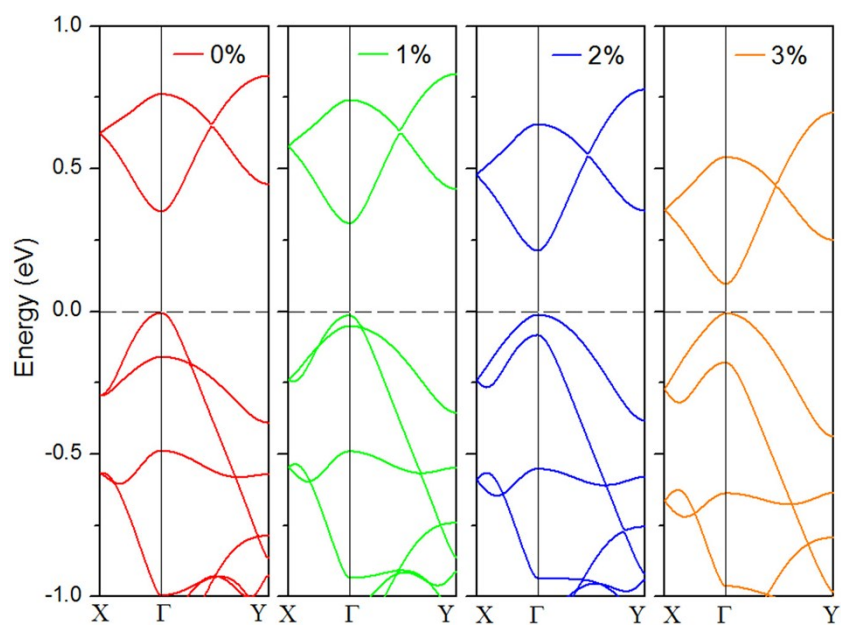


Figure S4. Electronic band structure of sho-MoS₂ under different magnitudes of equibiaxial tensile strain at the PBE level.

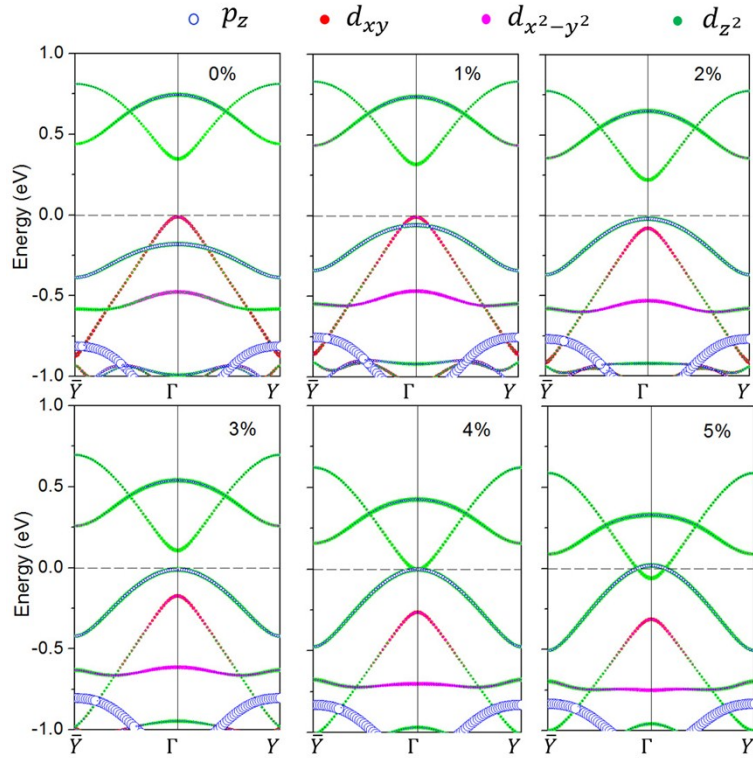


Figure S5. Orbital projected band structure of sho-MoS₂ under different magnitudes of equi-biaxial tensile strains.

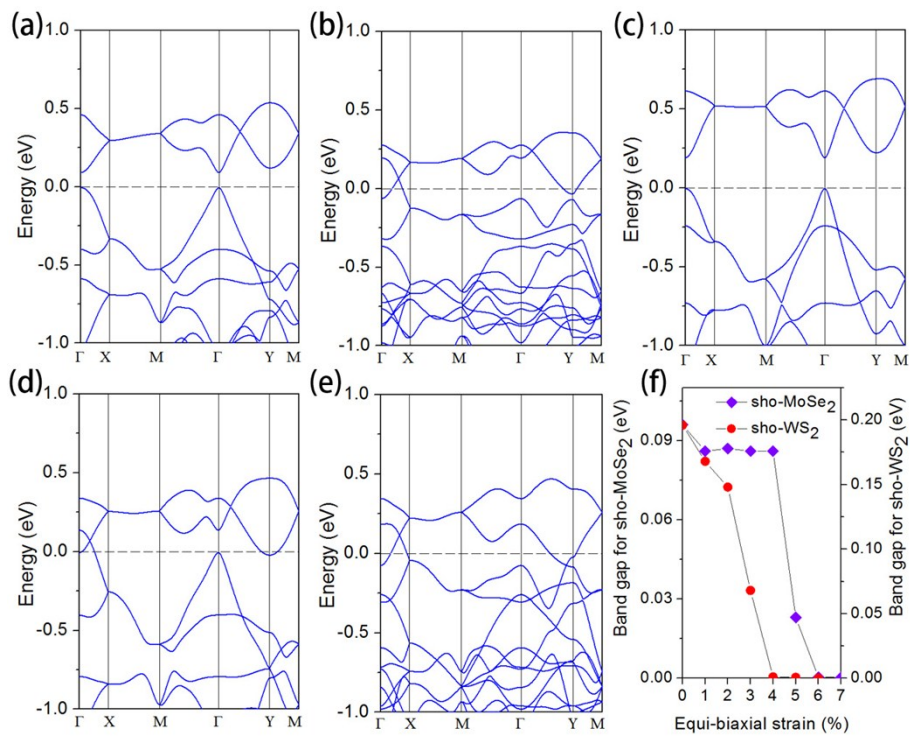


Figure S6. Electronic band structure of (a) sho-MoSe₂, (b) sho-MoTe₂, (c) sho-WS₂, (d) sho-WSe₂, and (e) sho-WTe₂ calculated using the PBE functional. (f) Energy band

gap as a function of equi-biaxial tensile strain for sho-MoSe₂ and sho-WS₂, respectively.

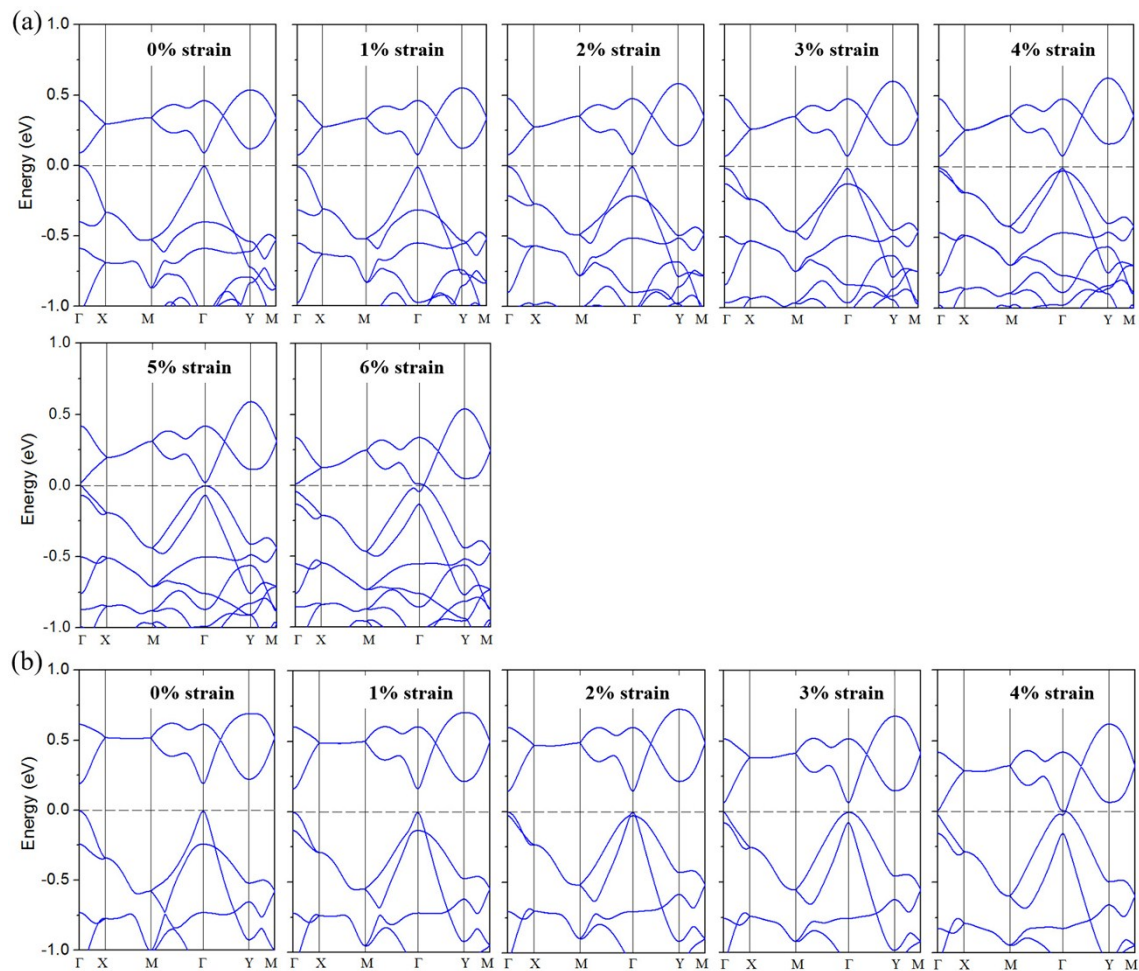


Figure S7. Electronic band structure of (a) sho-MoSe₂ and (b) sho-WS₂ under different magnitudes of equi-biaxial tensile strain (at the PBE level).

Table S1. Optimized lattice parameters and electronic band gaps E_g at the PBE level for sho- MX_2 structures, where $M = \text{Mo}$ and W , and $X = \text{S}$, Se and Te .

sho- MX_2	a (Å)	b (Å)	E_g (eV)
MoS ₂	9.98	5.70	0.35
MoSe ₂	10.41	5.92	0.10
MoTe ₂	11.12	6.18	metallic
WS ₂	10.02	5.73	0.20
WSe ₂	10.42	5.92	metallic
WTe ₂	11.08	6.18	metallic

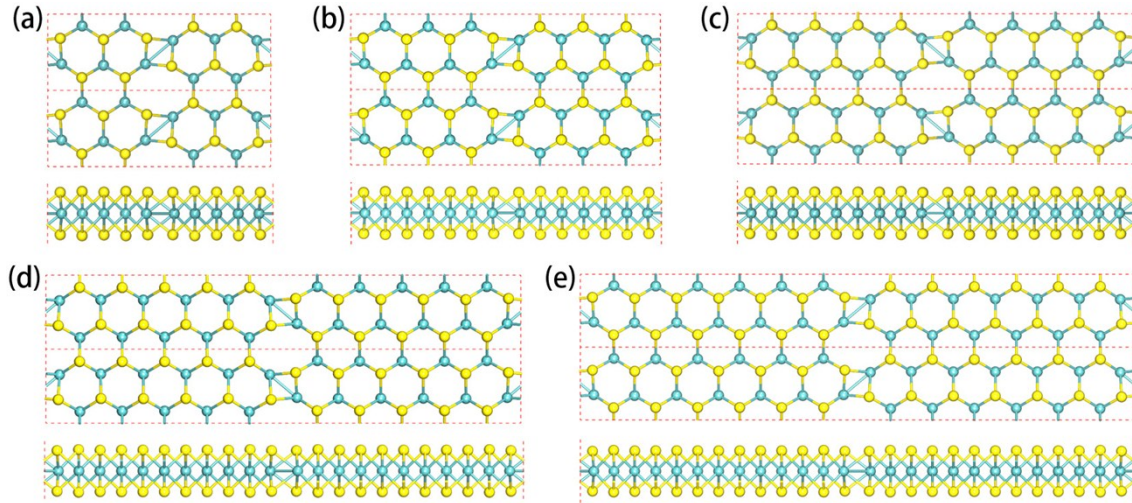


Figure S8. Optimized structures of (a) sh²o-MoS₂, (b) sh³o-MoS₂, (c) sh⁴o-MoS₂, (d) sh⁵o-MoS₂, and (e) sh⁶o-MoS₂ from the top and side views.

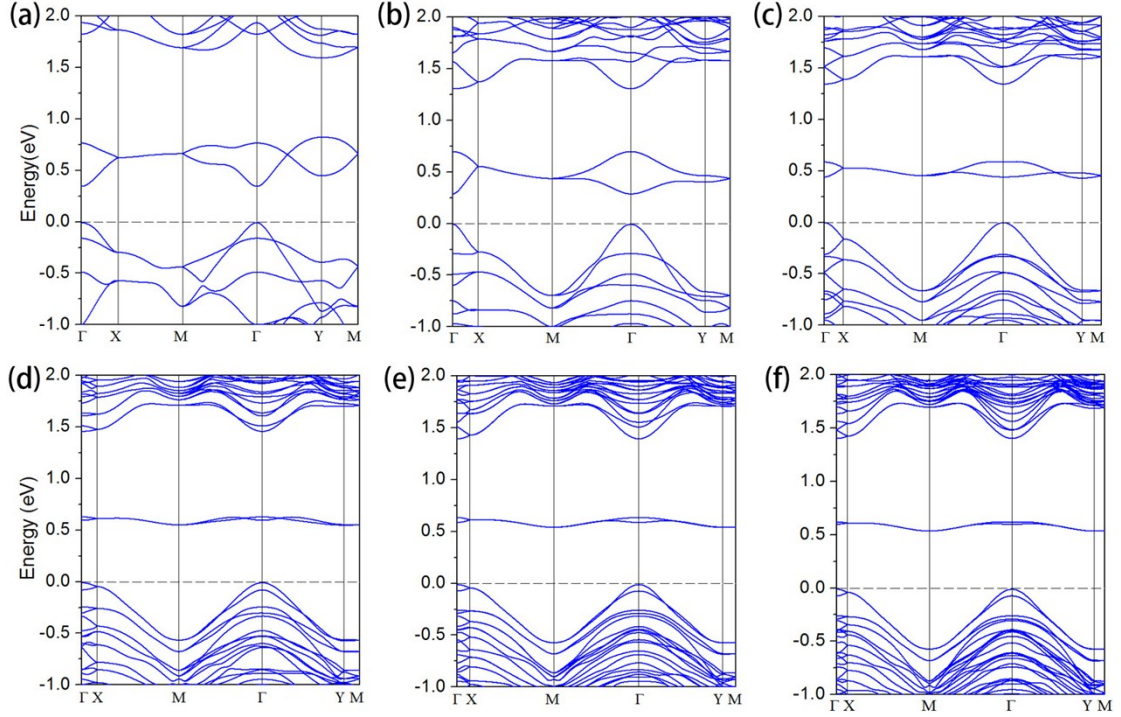


Figure S9. Electronic band structure of (a) sho-MoS₂, (b) sh²o-MoS₂, (c) sh³o-MoS₂, (d) sh⁴o-MoS₂, (e) sh⁵o-MoS₂, and (f) sh⁶o-MoS₂ at the PBE level.

From the results in Table S2, one can see that the band gaps of shⁿo-MoS₂ structures increase as the value of n increases except for the case of $n = 1$. This can be attributed to the structure uniqueness of sho-MoS₂. It is found that the charge is mainly distributed in the octagon rings by calculating band-decomposed charge density corresponding to the two lowest unoccupied bands of shⁿo-MoS₂. For shⁿo-MoS₂ with n value larger than 1, where the neighboring octagons are separated by hexagons along the crystal axis a , so that suppressing the interaction of localized charge in two neighboring octagons. However, for sho-structure, there is no hexagon unit between the neighboring octagonal rings along the crystal axis a , thus the charge localized in the neighboring octagons can directly interact with each other, resulting in the abnormal value of band gap.

Table S2. Optimized lattice parameters, relative energy (E) calculated with respect to H-MoS₂, and band gap (E_g , calculated by using the PBE functional) for sh^{*n*}o-MoS₂ ($n=1-6$).

n in sh ^{<i>n</i>} o-MoS ₂	a (Å)	b (Å)	E (eV/f.u.)	E_g (eV)
1	9.98	5.70	0.60	0.35
2	16.40	5.60	0.37	0.29
3	22.80	5.56	0.27	0.43
4	29.18	5.55	0.21	0.56
5	35.54	5.54	0.17	0.55
6	41.91	5.54	0.15	0.55

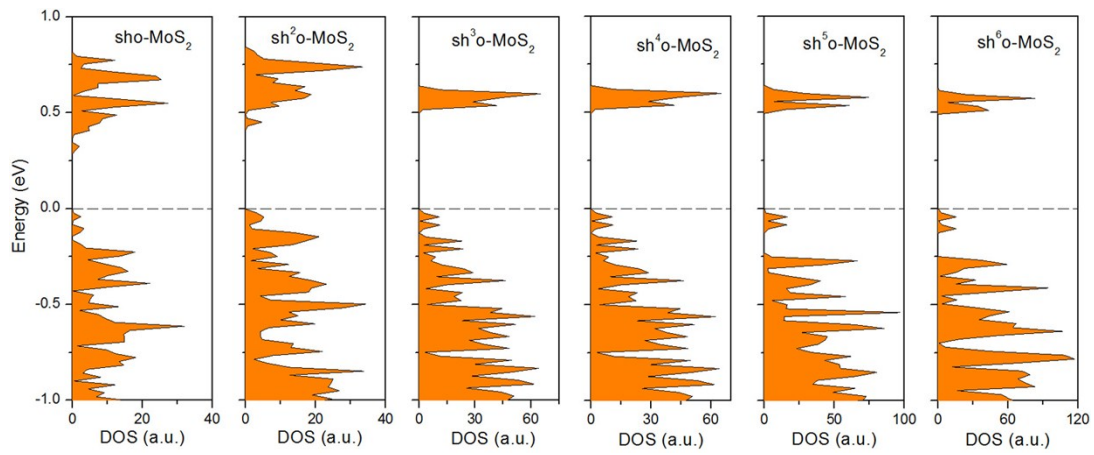


Figure S10. Electronic density of states (DOS) of sh⁰o-, sh²o-, sh³o-, sh⁴o-, sh⁵o-, and sh⁶o-MoS₂. Here a.u. represents arbitrary unit.

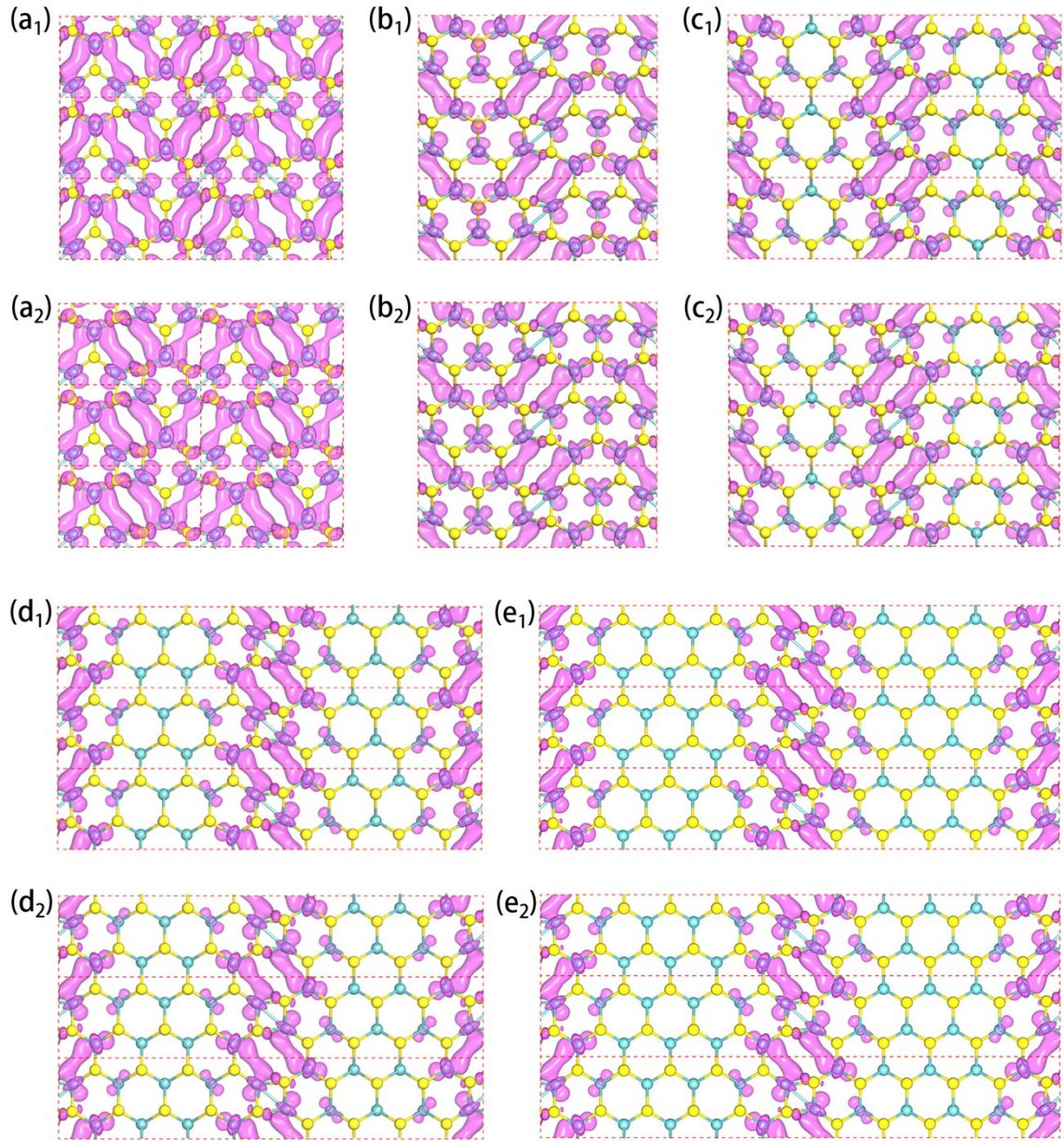


Figure S11. Band-decomposed charge density isosurfaces (isovalue: $0.01 e/\text{\AA}^3$) corresponding to the lowest and the second lowest unoccupied energy bands of sh^n - MoS_2 : (a₁) and (a₂) for sh^0 - MoS_2 ; (b₁) and (b₂) for sh^2 - MoS_2 ; (c₁) and (c₂) for sh^3 - MoS_2 ; (d₁) and (d₂) for sh^4 - MoS_2 ; (e₁) and (e₂) for sh^5 - MoS_2 . The top and bottom panels are the charge of the lowest and second lowest unoccupied bands, respectively.