

Supplementary Information

Homogeneous Cationic Substitution for Two-Dimensional Layered Metal Oxide Nanosheet via Galvanic Exchange Reaction

Joohyun Lim, Jang Mee Lee, Boyeon Park, Xiaoyan Jin, and Seong-Ju Hwang*

Department of Chemistry and Nanoscience, College of Natural Sciences, Ewha Womans University, Seoul 03760, Korea

*E-mail: hwangsju@ewha.ac.kr (S.-J. H.)

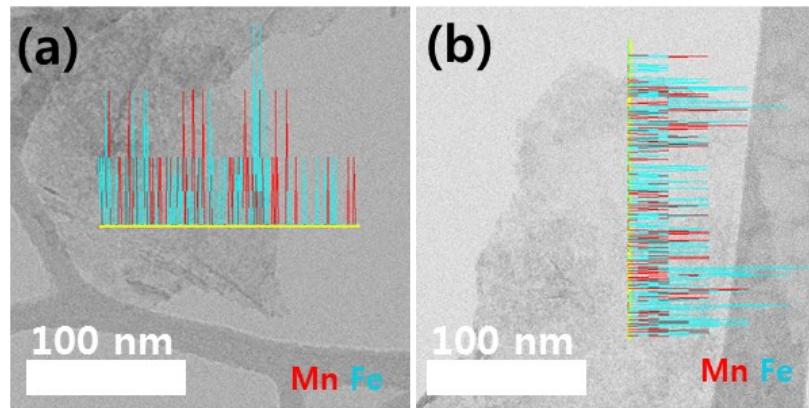


Fig. S1 Scanning tunneling microscopy–Energy dispersive spectrometry (STEM–EDS) line profiles of (a) FMO-GE1 and (b) FMO-GE3.

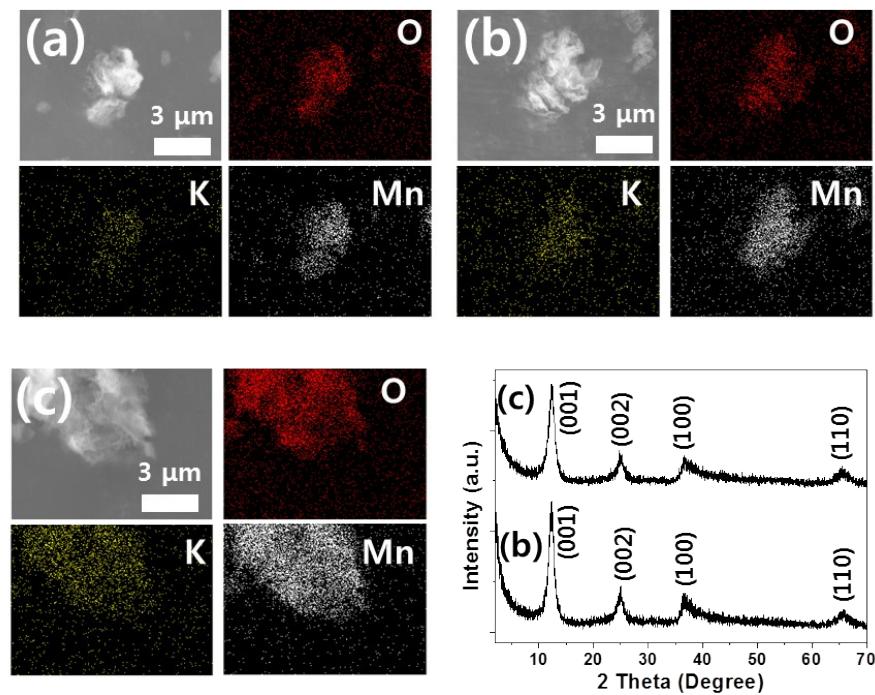


Fig. S2 EDS–elemental maps and powder X-ray diffraction (XRD) patterns of (a) KMO-1, (b) KMO-3, and (c) KMO-RS.

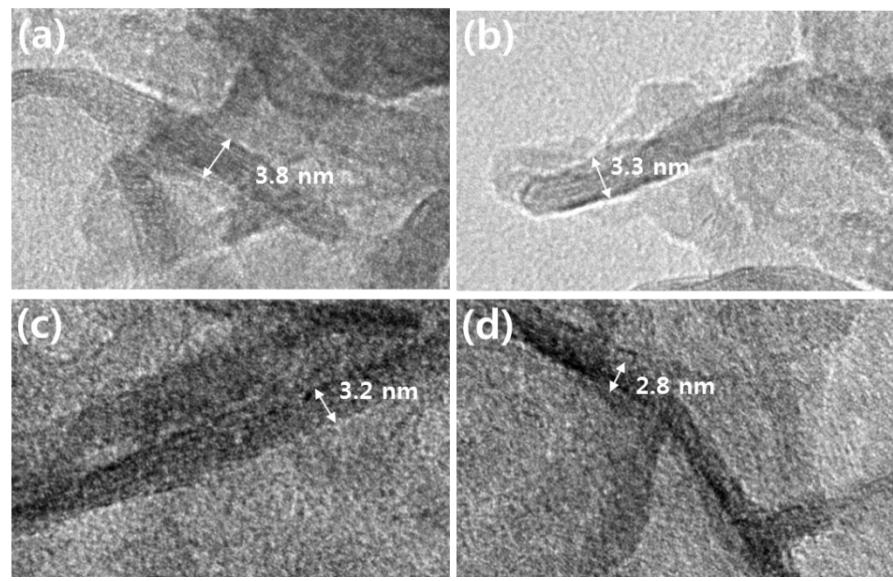


Fig. S3 Transmission electron microscopy (TEM) images of (a), (b) FMO-RS and (c), (d) FMO-GE2.

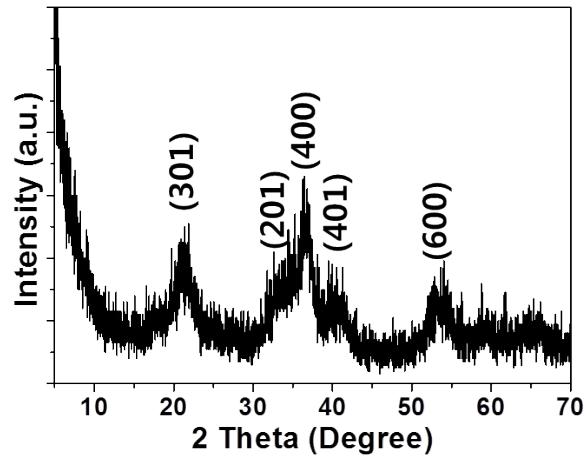


Fig. S4 Powder XRD pattern of FMO-GE material prepared with 0.5 mmol (in 1 mL water) Fe^{2+} solution.

Table. S1 Comparison of the capacity of manganese oxidized based materials reported in literatures and this work.

| <i>Material</i> | <i>Current density</i> (mA g ⁻¹) | <i>Capacity</i> (mAh g ⁻¹) | <i>Reference</i> |
|---|---|---|------------------|
| FMO-GE2@C | 200 | 1019 | [Our work] |
| MnO@C nanowire seeded by Si | 200 | 854 | 1 |
| MnO@reduced graphene oxide | 200 | ~620 | 2 |
| MnO/nanoporous Au | 200 | ~700 | 3 |
| MnO ₂ @CNT microsphere | 200 | ~1000 | 4 |
| MnO ₂ @N-doped graphene | 200 | ~900 | 5 |
| MnO ₂ -PEI-graphene | 200 | ~870 | 6 |
| Mn ₂ O ₃ /PEDOT:PSS | 200 | ~420 | 7 |
| Mn ₃ O ₄ @graphene membrane | 200 | ~650 | 8 |
| (011) exposed Mn ₃ O ₄ single crystal | 100 | ~600 | 9 |
| 3D Mn ₃ O ₄ | 200 | 1166 | 10 |

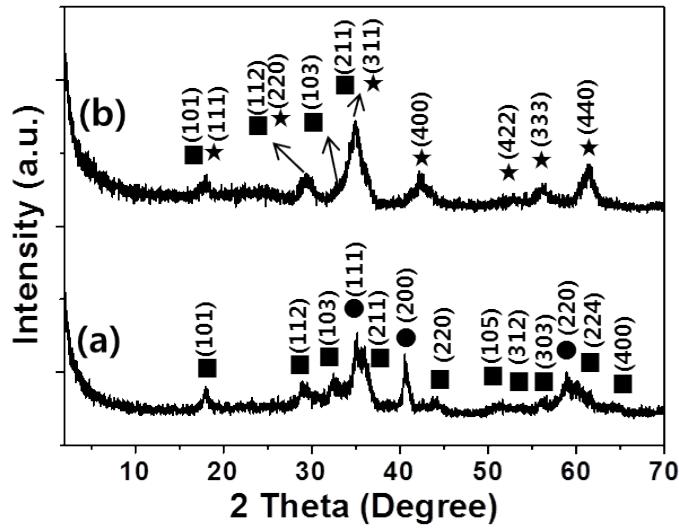


Fig. S5 Powder XRD patterns of the carbon-coated derivatives of (a) FMO-GE1 and (b) FMO-GE3. The Bragg reflections of Mn_3O_4 , MnO , and MnFe_2O_4 phases are denoted as squares, circles, and stars, respectively.

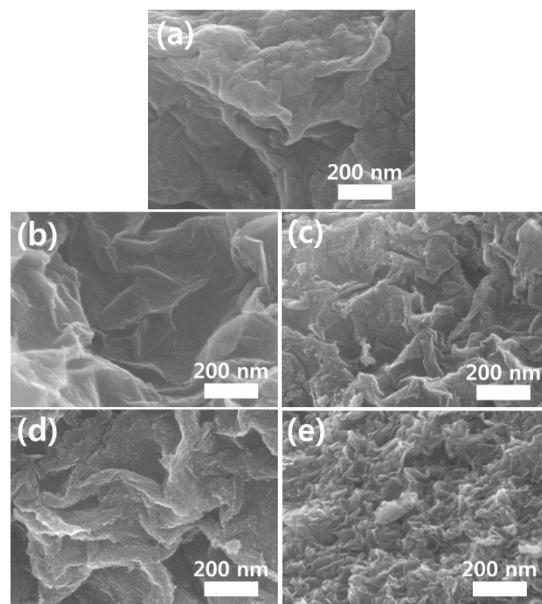


Fig. S6 Field emission-scanning electron microscopy (FE-SEM) images of (a) MnO_2 nanosheet (NS), (b) SMO-GE1, (c) SMO-GE2, (d) SMO-GE3, and (e) SMO-RS.

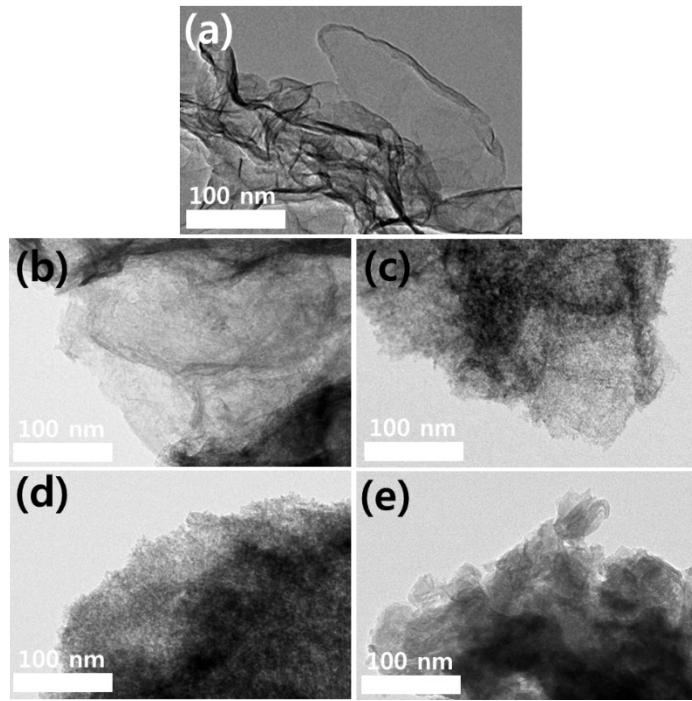


Fig. S7 TEM images of (a) MnO₂ NS, (b) SMO-GE1, (c) SMO-GE2, (d) SMO-GE3, and (e) SMO-RS.

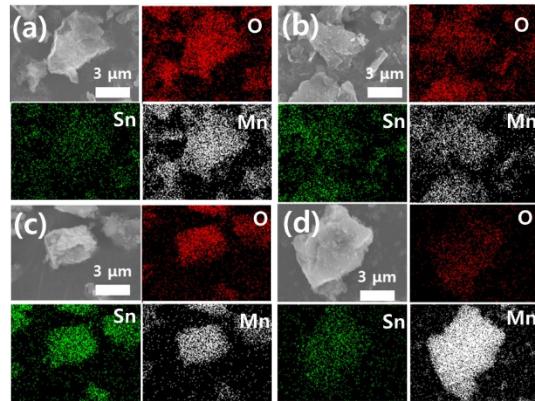


Fig. S8 SEM images and SEM-EDS maps of (a) SMO-GE1, (b) SMO-GE2, (c) SMO-GE3, and (d) SMO-RS.

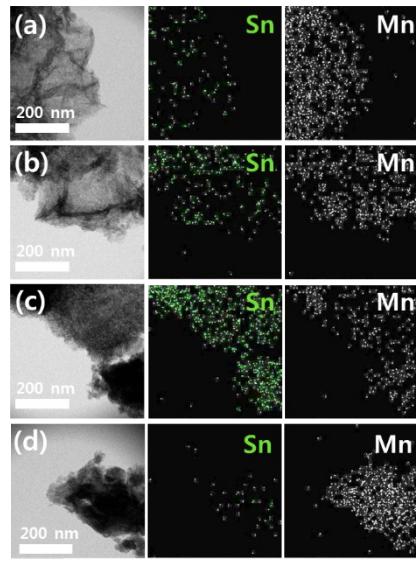


Fig. S9 TEM images and TEM–EDS maps of (a) SMO-GE1, (b) SMO-GE2, (c) SMO-GE3, and (d) SMO-RS.

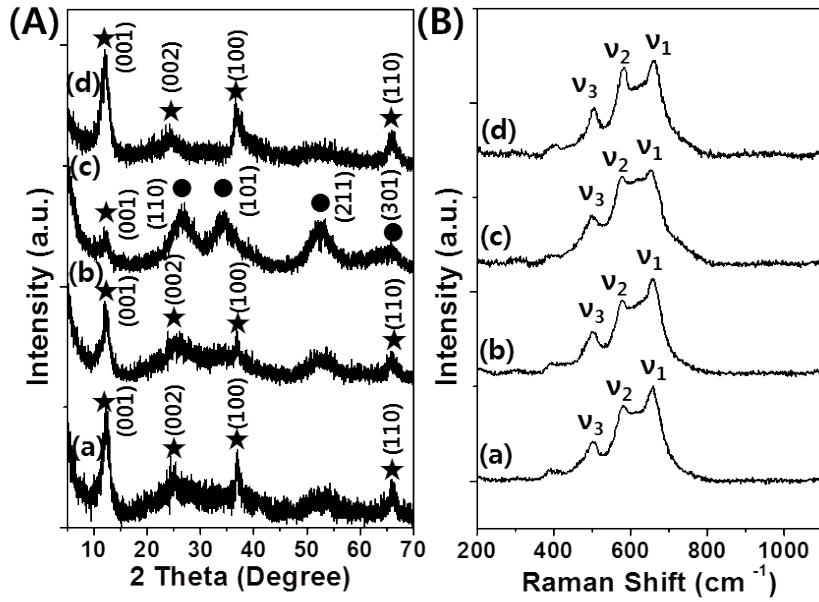


Fig. S10 (A) Powder XRD patterns and (B) micro-Raman spectra of (a) SMO-GE1, (b) SMO-GE2, (c) SMO-GE3, and (d) SMO-RS. The Bragg reflections of MnO_2 NS and SnO_2 phases are denoted as stars and circles, respectively.

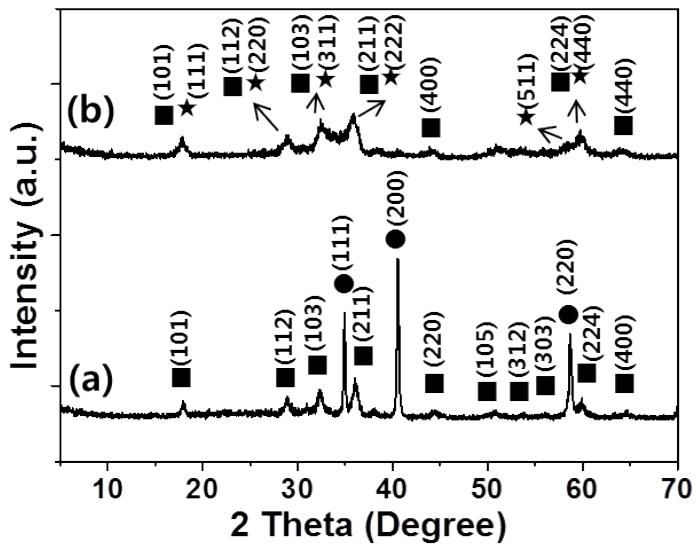


Fig. S11 Powder XRD patterns of the carbon-coated derivatives of (a) MnO_2 NS and (b) SMO-GE1. The Bragg reflections of Mn_3O_4 , MnO , and SnMn_2O_4 phases are denoted as squares, circles, and stars, respectively.

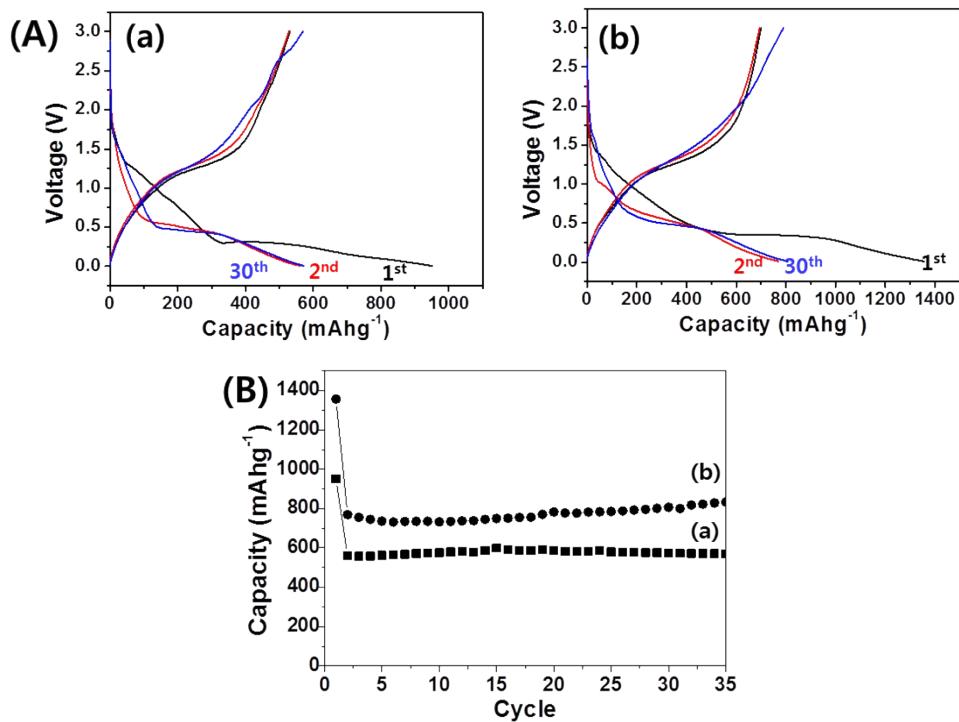


Fig. S12 (A) Galvanostatic charge–discharge potential profiles and (B) capacity retention plots of carbon-coated derivatives of (a) MnO_2 NS and (b) SMO-GE1.

References

- 1 H. Wei, J. Ma, B. Li, L. An, J. Kong, P. Yu and D. Xia, *NPG Asia Mater.*, 2016, **8**, e255.
- 2 P. Xia, H. B. Lin, W. Q. Tu, X. Q. Chen, X. Cai, X. W. Zheng, M. Q. Xu and W. S. Li, *Electrochim. Acta*, 2016, **198**, 66-76.
- 3 X. Guo, J. Han, L. Zhang, P. Liu, A. Hirata, L. Chen, T. Fujita and M. Chen, *Nanoscale*, 2015, **7**, 15111-15116.
- 4 W. Mao, G. Ai, Y. Dai, Y. Fu, Y. Ma, S. Shi, R. Soe, X. Zhang, D. Qu, Z. Tang and V. S. Battaglia, *J. Power Sources*, 2016, **310**, 54-60.
- 5 C. Jiang, C. Yuan, P. Li, H.-g. Wang, Y. Li and Q. Duan, *J. Mater. Chem. A*, 2016, **4**, 7251-7256.
- 6 C. Chae, K. W. Kim, Y. J. Yun, D. Lee, J. Moon, Y. Choi, S. S. Lee, S. Choi and S. Jeong, *ACS Appl. Mater. Interfaces*, 2016, **8**, 11499-11506.
- 7 I.-H. Ko, S.-J. Kim, J. Lim, S.-H. Yu, J. Ahn, J.-K. Lee and Y.-E. Sung, *Electrochim. Acta*, 2016, **187**, 340-347.
- 8 J.-G. Wang, D. Jin, R. Zhou, X. Li, X.-r. Liu, C. Shen, K. Xie, B. Li, F. Kang and B. Wei, *ACS Nano*, 2016, **10**, 6227-6234.
- 9 S.-Z. Huang, J. Jin, Y. Cai, Y. Li, H.-Y. Tan, H.-E. Wang, G. Van Tendeloo and B.-L. Su, *Nanoscale*, 2014, **6**, 6819-6827.
- 10 X.-Y. Fan, Y. Cui, P. Liu, L. Gou, L. Xu and D.-L. Li, *Phys. Chem. Chem. Phys.*, 2016, **18**, 22224-22234.