ELECTRONIC SUPPLEMENTARY INFORMATION

Phosphorothioate Anti-sense Oligonucleotides: The Kinetics and Mechanism of the Generation of the Sulfurising Agent from Phenylacetyl Disulfide (PADS)

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1. Typical HPLC plot for the degradation of PADS at 25°C



2. NMR spectral data for the degradation of PADS in 50% acetonitrile 3-picoline



Samples prepared by quenching PADS solutions (0.2M) in 50% Picoline/ACN at T=n with 2M HCI

3. Mass spec of phenylacetyl polysulfides





CH₂ and CHD Concentration (M) vs Time in D2O exchange experiment in ACN-d₃: D₂O (5M) + 3-Picoline (0.37M)

5. X-Ray Crystal structure of benzoyl disulfide

6. Graphical display of data provided in Table 1

7. Graphical display of data provided in Table 2

log kex(M-1s-1) for deuterium exchange catalysed by 3-picoline against PADS phenyl substituent Hammet σ

8. ¹H and ¹³C NMR Spectra for *Methyl 2'-iodo-2-phenylacetate*

9. ¹H and ¹³C NMR Spectra for *Methyl 2'-allyl-2-phenylacetate*

10. ¹H and ¹³C NMR Spectra for 2'-allyl-2-phenylacid

11. ¹H and ¹³C NMR Spectra for 2'2'-Diallyl 2,2-phenylacetyl disulfide (8)

12. ¹H and ¹³C NMR Spectra for 7,7a-dihydro-1H-cyclobuta[a]inden-2(2aH)-one (9)

13. ¹H and ¹³C NMR Spectra for 2,2,2',2'-Tetramethyl 2,2'phenylacetyl disulfide (7)

14. ¹H and ¹³C NMR Spectra for 2,2'-(4-cyanophenyl)acetyl disulfide

15. ¹H and ¹³C NMR Spectra for 2,2'-(4-methoxyphenyl)acetyl disulfide

16. ¹H and ¹³C NMR Spectra for *Bis-benzoyl disulfide (see also section 5 for X-Ray crystal structure)*

