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# A One-pot Synthesis of Tetrazolones from Acid Chlorides: Understanding Functional Group Compatibility, and Application to the Late-Stage Functionalization of Marketed Drugs

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#### LITERATURE STRUCTURES CONTAINING A TETRAZOLONE MOTIF

The following sub-structure search was conducted within SciFinder® on 01st October 2014 to ascertain the number of compounds attached to a mono-substituted tetrazolone motif.

This resulted in 9410 substances contained within 269 references (when duplicates were removed). The results of this search can be found using the following link

https://scifinder.cas.org/scifinder/view/link\_v1/answerset.html?l=guJIPNevZdEYhFK6sQswYGkco7Yz

XzXs2-pfH6Ki1yQ5q4Iv3\_UQHnjzA1XKEO8UJbeTI\_be1M2X8c-3iqIVTw

The 269 references consisted of 164 patents (or patent applications) and 105 journal articles.

Of the 164 patents (or patent applications), approximately 95 described the use of a tetrazolone within agrochemistry. Approximately 48 patents (or patent applications) described the use of a tetrazolone within medicinal chemistry. It should be noted that most of the 164 patent (or patent applications) synthesize a mono-substituted tetrazolone intermediate, like that shown in the substructure above, and then acylate, or alkylate, the mono-substituted tetrazolone to obtain 1,3-disubstituted tetrazolone final compounds.

Of the 105 journal articles, approximately 8 described applications of tetrazolones in agrochemistry, and approximately 14 described applications of tetrazolones in medicinal chemistry. Included in the agrochemistry articles were those describing research with the agrochemical, fentrazamide (YRC 2388).

S5

#### **IMPORTANT SAFETY NOTICE FOR ALL EXPERIMENTS**

Azidotrimethylsilane (also known as trimethylsilylazide) is highly toxic and is potentially explosive.<sup>1-4</sup> Azidotrimethylsilane can also be hydrolyzed to hydrazoic acid (HN<sub>3</sub>), a volatile, highly toxic and highly explosive substance.<sup>2-4</sup> Therefore, personnel should be well-versed in the safe-handling and safe disposal of azidotrimethylsilane before attempting to repeat any of the experiments described within this *Supplementary Information*.<sup>4</sup> It should be noted that all experiments were undertaken behind a blast shield, with personnel wearing appropriate protective clothing for working with azidotrimethylsilane, azide intermediates and tetrazolone products.

It should also be noted that the tetrazolone products arising from these reactions may also be potentially explosive, and possess unknown toxicities. Therefore, appropriate care and safety precautions should also be exercised when performing workups and purifications of the compounds described within this *Supplementary Information*.

- 1. For a Material Safety Data Sheet (MSDS) on azidotrmethylsilane, see: <a href="http://www.sigmaaldrich.com/MSDS/MSDS/DisplayMSDSPage.do?country=US&language=en&productNumber=155071&brand=ALDRICH&PageToGoToURL=http%3A%2F%2Fwww.sigmaaldrich.com">http://www.sigmaaldrich.com/MSDS/MSDS/DisplayMSDSPage.do?country=US&language=en&productNumber=155071&brand=ALDRICH&PageToGoToURL=http%3A%2F%2Fwww.sigmaaldrich.com</a> %2Fcatalog%2Fproduct%2Faldrich%2F155071%3Flang%3Den
- 2. K. J. KerBeek, Chem. Eng. News, 1998, 76, 6.
- 3. M. Jafarzadeh, Synlett, 2007, 2144-2145.
- 4. For a recent discussion regarding some important considerations when using, or scaling-up azide chemistry, see: F. González-Bobes, N. Kopp, L. Li, J. Deerberg, P. Sharma, S. Leung, M. Davies, J. Bush, J. Hamm, M. Hrytsak, *Org. Proc. Res. Develop.*, 2012, 16, 2051-2057 and references cited therein.

#### **GENERAL EXPERIMENTAL**

All reagents and solvents were purchased from commercial suppliers and used without further purification. Reactions were monitored by thin-layer chromatography or high-performance liquid chromatography. NMR was performed on a 300 MHz NMR spectrometer and all chemical shifts are reported relative to a tetramethylsilane internal standard, or by referencing on the deuterated solvent. Reverse-phase high-performance liquid chromatography was performed on standard equipment and was coupled to diode array and mass spectra detectors — the mass-spectra detector operating under the electrospray ionization (ESI) mode. The column-gradient system was as follows

Column: Phenomenex Gemini 4.6 x 100 mm, C18, 5µm, 110Å

Column temperature 30°C

Sample temperature 15°C

Solvent A -0.05% Formic acid in Water

Solvent B -0.05% Formic acid in Acetonitrile

Flow rate -1.5 mL/min

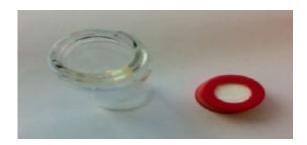
Gradient:

Time	A%	В%
0	95	5
10	0	100 (curve=6)
11.1	0	100
11.2	95	5
12.1	95	5

High resolution mass spectrometry were obtained on a LCT Premier XE mass spectrometer (time-of-flight) operating under the electrospray ionization mode. Mass analysis was performed in extended W-

mode using leucine enkephalin as reference lock mass (556.2771 Da for positive ion & 554.2615 Da for negative ion).

20 mL Vials with pressure-release caps were obtained from Chemglass (catalogue # CG-4912-05).



**Supporting Figure 1**. Photo of 20 mL vials with pressure-release tops.

Reactions could also be undertaken in parallel using the setup illustrated in Supporting Figure 2. The last position of the heating block is used to house a vial containing heat-resistant silicone oil. A thermocouple is then inserted in to the oil to control the temperature of the block.



Supporting Figure 2. Photo of reactions undertaken in parallel on heating block (behind a blast shield).

Note about the reactions: It is likely that the reactions described in this Supporting Information are complete within a short timeframe (2 hours) - e.g. see page S18. However, we typically undertake the majority of reactions at the end of a working day, and hence, leave them overnight for convenience.

## PREPARATION OF 1-(2-BROMO-4-FLUOROPHENYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 4

F

Br

TMS-N<sub>3</sub>, 100 °C

F

N

N

N

N

N

N

A

$$N = N$$

## Table 1, entry 1: See 'Important safety notice for all experiments' (page S6); Blast shield employed

Azidotrimethylsilane (5 mL, 38.0 mmol) was added in one portion to 2-bromo-4-fluorobenzoyl chloride **3** (1.5 g, 6.3 mmol). The mixture was placed under nitrogen and then heated to 100 °C with stirring [Note: 100 °C refers to temperature of heating block]. The mixture was left to stir at 100 °C overnight. After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (50 mL) and H<sub>2</sub>O (50 mL). The organic layer was then extracted with a saturated aqueous solution of NaHCO<sub>3</sub> (4 x 40 mL) [Note: extraction continued until TLC showed all tetrazolone product removed from the organic layer]. EtOAc (50 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned, and the aqueous layer extracted with EtOAc (1 x 50 mL). The combined organic layers were dried (MgSO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (1.34 g, 82%) as a solid.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz): δ 7.90 (dd, J = 8.4, 2.7 Hz, 1H), 7.77 (dd, J = 8.7, 5.7 Hz, 1H), 7.52-7.45 (m, 1H), -1.2 (br. s, 1H)

<sup>19</sup>F NMR (DMSO- $d_6$ , MHz): - 107.6 (dd, J = 13.8, 7.6 Hz)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  164.2 (d, J = 251 Hz), 150.7, 132.0 (d, J = 9.3 Hz), 128.7 (d, J = 3.8 Hz), 122.6 (d, J = 11.0 Hz), 121.0 (d, J = 26 Hz), 116.4 (d, J = 23 Hz)

 $m/z = 257.28 \text{ [M-H]}^+ \text{ for }^{79} \text{Br}$ 

HRMS (EI): [M-H]<sup>+</sup> calc'd for C<sub>7</sub>H<sub>4</sub>BrFN<sub>4</sub>O *m/z* 256.9474, found 256.9527

#### Table 1, entry 2:

The above reaction could be repeated using 4.5 equivalents of azidotrimethylsilane (3.75 mL) to give the product (1.24 g, 76%) as a solid.

#### Table 1, entry 3:

The above reaction could be repeated using 3.0 equivalents of azidotrimethylsilane (2.5 mL) to give the product (0.6 g , 37%) as a solid. Also isolated from the EtOAc layer after extraction with saturated NaHCO<sub>3</sub> was a symmetrical urea by-product, *bis*-(2-bromo-4-fluorphenyl)urea.

Data for bis-(2-bromo-4-fluorphenyl)urea

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz): δ 8.82 (br. s, 2H), 7.88 (dd, J = 9.0, 5.7 Hz, 2H), 7.57 (dd, J = 8.7, 3.0 Hz, 2H), 7.22 (ddd, J = 9.0, 8.1, 3.0, Hz, 2H)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  159.2 (d, J = 244 Hz), 152.7, 133.7 (d, J = 3.3 Hz), 125.1 (d, J = 8.3 Hz), 119.4 (d, J = 25 Hz), 115.0 (d, J = 22 Hz), 114.8 (d, J = 10 Hz)

 $m/z = 407.26 \text{ [M+H]}^{+} \text{ and } 405.37 \text{ [M-H]}^{+}$ 

HRMS (EI):  $[M+H]^+$  calc'd for  $C_{13}H_8Br_2F_2N_2O$  m/z 404.9050, found 404.9047; m/z 406.9030, found 406.9025; m/z 408.9011, found 408.9003

HRMS (EI):  $[M-H]^+$  calc'd for  $C_{13}H_8Br_2F_2N_2O$  m/z 402.8893, found 402.8912; m/z 404.8873, found 404.8815; m/z 406.8854, found 406.8802

bis-(2-bromo-4-fluorphenyl)urea

**Table 1, entry 4**: Note: For large-scale reactions, the mixtures are only placed under a nitrogen atmosphere after cessation of the gas evolution

The above reaction was repeated on a large-scale using azidotrimethylsilane (65 mL, 494 mmol) and 2-bromo-4-fluorobenzoyl chloride **3** (20.4 g, 85.9 mmol). The mixture was heated slowly and evolution of a gas (presumably, arising from a Curtius rearrangement) was noted from 50-60 °C (block temperature), which became a vigorous evolution when the block temperature was raised to *ca.* 65 °C. The mixture was removed from the heat at a block temperature of *ca.* 70 °C, and when gas evolution subsided, was re-subjected to heating stepwise from *ca.* 70 °C to 90 °C. The mixture was stirred at 90 °C overnight. Usual workup, partitioning between EtOAc (200 mL) and H<sub>2</sub>O (100 mL) and then extracting the organic layer with a saturated solution of NaHCO<sub>3</sub> (5 x 150 mL), followed by acidification and extraction with EtOAc, gave the product (19.9 g, 89%) as a solid.

A separate reaction using 20.0g of 2-bromo-4-fluorobenzoyl chloride **3** and 47 mL of azidotrimethylsilane (4.0 equivalents) gave the product (17.4 g, 80%) as a solid.

A separate reaction using 36g of 2-bromo-4-fluorobenzoyl chloride **3** and 120 mL of azidotrimethylsilane (6.0 equivalents) gave the product (37 g, 94%) as a solid. For this reaction, the mixture was heated slowly to *ca*. 55 °C, whereupon the evolution of gas became regular. Evolution of gas stopped after *ca*. 15 min at 55 °C. The mixture was then slowly heated to 90-95 °C (block temperature), placed under a nitrogen atmosphere and stirred overnight.

PREPARATION OF 1-(2-BROMO-4-FLUOROPHENYL)-1,4-DIHYDRO-5*H*-TETRAZOL-5-ONE <u>4</u> FROM COMMERCIALLY-AVAILABLE 2-BROMO-4-FLUORO-1-ISOCYANOATOBENZENE

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

Azidotrimethylsilane (10 mL, 76 mmol) was added in one portion to 2-bromo-4-fluoro-1-isocyanatobenzene (3.0 g, 13.9 mmol). The mixture was placed under nitrogen and then heated to 90 °C with stirring [Note: 90 °C refers to temperature of heating block]. The mixture was left to stir at 90 °C overnight. After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (50 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (100 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 50 mL). EtOAc (100 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 2N HCl with efficient stirring. The aqueous and organic layers were partitioned, and the organic layer was dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (3.44 g, 95%) as a solid.

A separate reaction was repeated using 20.0g of 2-bromo-4-fluoro-1-isocyanatobenzene (92.6 mmol) and 50 mL of azidotrimethylsilane (380 mmol; *ca.* 4.0 equivalents) to give the product (20.1 g, 84%) after workup.

Note: Experiments with related starting materials have indicated that the isocyanate to tetrazolone formation is complete within ca. 2 hours. Therefore, it is likely that the reaction of acid chlorides to tetrazolones is similarly complete within a few hours, since it is observed that the Curtius rearrangement occurs quickly (evolution of gas complete within 15-30 min). However, we typically

convenience.	

undertake the majority of reactions at the end of a working day, and hence, leave them overnight for

#### PREPARATION OF 1-PHENYL-1,4-DIHYDRO-5*H*-TETRAZOL-5-ONE 6A

## See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and benzoyl chloride (422 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (10 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (10 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 10 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (20 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 10 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (369 mg, 76%) as a solid. A sample was recrystallized from EtOAc.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz): δ 7.85-7.81 (m, 2H), 7.57-7.50 (m, 2H), 7.41 (dt, J = 7.5, 1.5 Hz, 1H), -1.3 (br. s, 1H)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  150.3, 134.2, 129.5, 127.6, 119.5

 $m/z = 163.18 \text{ [M+H]}^+ \text{ and } 161.24 \text{ [M-H]}^+$ 

HRMS (EI):  $[M-H]^+$  calc'd for  $C_7H_6N_4O$  m/z 161.0463, found 161.0532

# PREPARATION OF 1-(NAPHTHALEN-2-YL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 6B

## See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 2-naphthoyl chloride (572 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (10 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (10 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 10 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (20 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 10 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (419 mg, 66%) as a solid. A sample was recrystallized from EtOAc.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz): δ 8.42 (d, J = 2.1 Hz, 1H), 8.09 (d, J = 8.7 Hz, 1H), 8.02-7.93 (m, 3H), 7.60-7.52 (m, 2H), -1.1 (br. s, 1H)

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 75MHz): δ 150.4, 132.7, 131.8, 131.6, 129.5, 128.1, 127.8, 127.2, 126.6, 118.1, 117.0

 $m/z = 216.36 \text{ [M+H]}^{+} \text{ and } 215.45 \text{ [M-H]}^{+}$ 

HRMS (EI):  $[M+H]^+$  calc'd for  $C_{11}H_8N_4O$  m/z 213.0776, found 213.0764

## PREPARATION OF 1-(4-(TERT-BUTYL)PHENYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 6C

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 4-(*tert*-butyl)benzoyl chloride (590 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (20 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (20 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (10 x 15 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (40 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 40 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (401 mg, 61%) as a solid.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz):  $\delta$  7.74 (dt, J = 8.7, 2.4 Hz, 2H), 7.57 (dt, J = 8.7, 2.4 Hz, 2H), 1.29 (s, 9H), -1.3 (br. s, 1H)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  150.3, 139.0, 131.7, 126.2, 119.5, 34.4, 31.0

 $m/z = 219.31[M+H]^{+}$  and 217.38  $[M-H]^{+}$ 

HRMS (EI):  $[M-H]^+$  calc'd for  $C_{11}H_{14}N_4O$  m/z 217.1089, found 213.1139

## PREPARATION OF 1-(3,5-BIS(TRIFLUOROMETHYL)PHENYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 6D

$$F_{3}C$$

$$\begin{array}{c}
CF_{3} \\
\hline
TMS-N_{3}, 100 \, ^{\circ}C
\end{array}$$

$$F_{3}C$$

$$\begin{array}{c}
CF_{3} \\
\hline
N \\
N=N
\end{array}$$

$$\begin{array}{c}
O \\
N=N
\end{array}$$

# See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 3,5-bis(trifluoromethyl)benzoyl chloride (830 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (10 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (10 mL). This base-acid extraction did not work for this compound (only 20 mg obtained after acidification / extraction), and the tetrazolone remained in the organic (EtOAc) layer. Therefore, the organic layer from the initial partition above was concentrated under vacuum and purified by column chromatography on silica gel using hexane / EtOAc as eluent (ISCO Combiflash System) to give the product (503 mg, 56%) as a solid.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz):  $\delta$  8.48 (s, 2H), 8.14 (s, 1H), -1.2 (br. s, 1H)

<sup>19</sup>F NMR (DMSO- $d_6$ , 282MHz): δ -61.8 (s)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz): δ 150.4, 136.0, 131.7 (q, J = 33 Hz), 124.6 (q, J = 271 Hz), 120.6 (m), 118.9 (m)

 $m/z = 297.36 \text{ [M-H]}^+$ 

HRMS (EI):  $[M-H]^+$  calc'd for  $C_9H_4F_6N_4O$  m/z 297.0211, found 297.0157

## PREPARATION OF 1-(3-NITROPHENYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 6E

$$O_2N$$

$$O_2N$$

$$O_2N$$

$$O_2N$$

$$O_2N$$

$$O_2N$$

$$O_2N$$

$$O_2N$$

$$O_3N$$

$$O_4N$$

$$O_2N$$

$$O_4N$$

$$O_2N$$

$$O_4N$$

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 3-nitrobenzoyl chloride (554 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C for 2 hr (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (10 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (10 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 10 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (20 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 10 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (560 mg, 90%) as a solid. A sample was recrystallized from EtOAc.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz): δ 8.69 (t, J = 4.2 Hz, 1H), 8.28-8.19 (m, 2H), 7.82 (t, J = 8.3 Hz, 1H), -1.0 (br. s, 1H)

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 75MHz): δ 150.1, 148.1, 135.1, 131.2, 124.8, 121.8, 113.2

 $m/z = 208.25 \text{ [M+H]}^+ \text{ and } 206.30 \text{ [M-H]}^+$ 

HRMS (EI):  $[M-H]^+$  calc'd for  $C_7H_5N_5O_3$  m/z 206.0314, found 206.0398

## PREPARATION OF 1-(4-FLUORO-5-NITROPHENYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 6F

## See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (7.9 mL, 60 mmol) and 4-fluoro-5-nitrobenzoyl chloride (2.04 g, 10 mmol) in a round bottom flask with reflux condenser was heated slowly from room temperature to 90 °C under an atmosphere of nitrogen (block temperature). The mixture was then stirred at 90 °C overnight. After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (50 mL) and H<sub>2</sub>O (50 mL). The organic layer was extracted with a saturated aqueous solution of NaHCO<sub>3</sub> (3 x 50 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (100 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the organic layer was dried (MgSO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (1.96 g) as a solid. <sup>1</sup>H NMR analysis indicated that the product was *ca*. 80-90% purity with approximately 10-20% of 4-fluoro-3-nitrobenzoic acid as a contaminant. Adjusting the yield for purity gives *ca*. 70% of desired product.

A sample of the above material was recrystallized from EtOAc to provide pure material.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz): δ 8.63 (dd, J = 9.6, 2.7 Hz, 1H), 8.25 (ddd, J = 9.3, 3.9, 2.7 Hz, 1H), 7.80 (dd, J = 11.1, 9.0 Hz, 1H), -1.0 (br. s, 1H)

<sup>19</sup>F NMR (DMSO- $d_6$ , 272MHz): δ -119.7 (m)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz): δ 154.9 (d, J = 261 Hz), 150.2, 136.9 (d, J = 8 Hz), 130.6 (d, J = 3 Hz), 127.0 (d, J = 9 Hz), 120.1 (d, J = 23 Hz), 116.6 (d, J = 3 Hz)

 $m/z = 226.24 \text{ [M+H]}^{+} \text{ and } 224.32 \text{ [M-H]}^{+}$ 

HRMS (EI):  $[M-H]^+$  calc'd for  $C_7H_4FN_5O_3$   $\emph{m/z}$  224.0220, found 224.0227

## PREPARATION OF 1-(2-FLUORO-5-NITROPHENYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 6G

$$O_2N \xrightarrow{\mathsf{F}} O \xrightarrow{\mathsf{TMS-N_3}, \ 90 \ ^{\circ}\mathsf{C}} O_2N \xrightarrow{\mathsf{N}} O_{\mathsf{N}} \overset{\mathsf{F}}{\underset{\mathsf{N}=\mathsf{N}}{\bigvee}} O_{\mathsf{N}} \overset{\mathsf{C}}{\underset{\mathsf{N}=\mathsf{N}}{\bigvee}} O_{\mathsf{N}} O_{\mathsf{N}} \overset{\mathsf{C}}{\underset{\mathsf{N}=\mathsf{N}}{\bigvee}} O_{\mathsf{N}} O_{\mathsf{N}} \overset{\mathsf{C}}{\underset{\mathsf{N}=\mathsf{N}}{\bigvee}} O_{\mathsf{N}} O_{\mathsf{$$

## See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (19.7 mL, 150 mmol) and 2-fluoro-5-nitrobenzoyl chloride (5.1 g, 25 mmol) in a round bottom flask with reflux condenser was heated slowly from room temperature to 90 °C (block temperature) (note: evolution of nitrogen is observed from 70 °C). The mixture was then stirred at 90 °C for 5-6 hr (TLC indicated complete reaction). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (100 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (150 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 50 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (200 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (1 x 100 mL). The combined organic layers were dried (MgSO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (4.8 g, 86%) as a solid. A sample was recrystallized from EtOAc.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz):  $\delta$  8.66-8.63 (m, 1H), 8.49-8.43 (m, 1H), 7.82 (t, J = 7.5 Hz, 1H), -1.0 (br. s, 1H)

<sup>19</sup>F NMR (DMSO- $d_6$ , 272MHz): δ -110.4 (m)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  160.9 (d, J = 262 Hz), 150.5, 144.0 (d, J = 3.1 Hz), 127.2 (d, J = 9.9 Hz), 123.4 (d, J = 2.2 Hz), 121.6 (d, J = 13.7 Hz), 118.7 (d, J = 22.1 Hz)

 $m/z = 226.24 \text{ [M+H]}^{+} \text{ and } 224.30 \text{ [M-H]}^{+}$ 

## PREPARATION OF 1-(3-CHLOROPHENYL)-1,4-DIHYDRO-5*H*-TETRAZOL-5-ONE 6H

## See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 3-chlorobenzoyl chloride (525 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (10 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (10 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 10 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (20 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 10 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (355 mg, 60%) as a solid. A sample was recrystallized from EtOAc.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz): δ 7.92 (t, J = 2.1 Hz, 1H), 7.82 (ddd, J = 8.2, 2.1, 1.2 Hz, 1H), 7.59 (t, J = 8.3 Hz, 1H), 7.46 (ddd, J = 8.1, 2.1, 0.9 Hz, 1H), -1.1 (br. s, 1H)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  150.1, 135.4, 133.7, 131.3, 127.3, 118.7, 117.6

 $m/z = 197.30 \text{ [M+H]}^+ \text{ and } 195.38 \text{ [M-H]}^+ \text{ for } ^{35}\text{Cl}$ 

HRMS (EI): [M+H]<sup>+</sup> calc'd for C<sub>7</sub>H<sub>5</sub>ClN<sub>4</sub>O *m/z* 195.0074, found 195.0099

## PREPARATION OF 1-(2-BROMOPHENYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 61

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 2-bromobenzoyl chloride (590 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C for 24hr (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (20 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (20 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 20 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (30 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 30 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (546 mg, 76%) as a solid. A sample was recrystallized from EtOAc.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz):  $\delta$  7.89 (dd, J = 8.1, 1.5 Hz, 1H), 7.68 (dd, J = 7.8, 2.1 Hz, 1H),

7.61-7.50 (m, 2H), -1.3 (br. s, 1H)

 $^{13}$ C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  150.7, 133.5, 132.4, 131.9, 130.3, 129.0, 121.3

 $m/z = 239.24 \text{ [M-H]}^+$ 

HRMS (EI):  $[M-H]^+$  calc'd for  $C_7H_5BrN_4O$  m/z 238.9568, found 238.9577

## PREPARATION OF 1-(2-IODOPHENYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 6J

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 2-iodobenzoyl chloride (799 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (10 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (10 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 10 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (20 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 10 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (567 mg, 66%) as a solid. A sample was recrystallized from EtOAc.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz):  $\delta$  8.07 (m, 1H), 7.61-7.57 (m, 2H), 7.38-7.29 (m, 1H), -1.3 (br. s, 1H)

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 75MHz): δ 150.5, 139.6, 135.5, 132.3, 129.6, 129.6, 98.2

 $m/z = 289.31 \text{ [M+H]}^{+} \text{ and } 287.42 \text{ [M-H]}^{+}$ 

HRMS (EI): [M+H]<sup>+</sup> calc'd for C<sub>7</sub>H<sub>5</sub>IN<sub>4</sub>O *m/z* 289.9586, found 289.9588

#### PREPARATION OF 1-((2,4-DICHLOROPHENOXY)METHYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 6K

Oxalyl chloride (2.0 M in  $CH_2Cl_2$ ; 3.0 mL, 6.0 mmol) was added dropwise over 2-3 min to a stirred suspension of 2,4-dichlorophenoxyacetic acid, also known as 2,4-D (663 mg, 3.0 mmol) and DMF (1-2drops) in  $CH_2Cl_2$  (6 mL) at 0 °C under nitrogen. After complete addition, the mixture was allowed to warm to room temperature and stirred over a weekend (note: a solution develops after warming to rt; reaction left over a weekend as did not want to undertake the next step on a Friday evening). The mixture was concentrated under vacuum to leave the acid chloride 5k, which was used directly in the tetrazolone-forming step below (yield assumed quantitative = 718 mg).

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 2,4-dichlorophenoxyacetyl chloride (718 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a round bottom flask with reflux condenser under an atmosphere of nitrogen. The mixture was then stirred at 100 °C for 3 hours. After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (30 mL) and a saturated solution of NaHCO<sub>3</sub> (30 mL). The organic layer was extracted with a saturated solution of NaHCO<sub>3</sub> (1 x 30 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (30 mL) was added to the filtrate and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 20 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the crude product (600 mg) of *ca*. 90% purity. The crude product was purified by column chromatography on silica gel (ISCO Combiflash)

using CH<sub>2</sub>Cl<sub>2</sub> / MeOH (1:0 to 9:1) as eluent to give pure product (336 mg, 43%) as a solid. Less pure product (*ca.* 130 mg) was also obtained from the column, but was not purified further.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz):  $\delta$  7.59 (d, J = 2.4 Hz, 1H), 7.49 (d, J = 8.7 Hz, 1H), 7.43 (dd, J = 8.7, 2.4 Hz, 1H), 5.94 (s, 2H), -1.4 (br. s, 1H)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  151.3, 150.6, 129.7, 128.3, 127.2, 124.2, 118.6, 71.2

 $m/z = 261.30 \text{ [M+H]}^+ \text{ and } 259.37 \text{ [M-H]}^+ \text{ for } ^{35}\text{Cl}$ 

HRMS (EI): [M-H]<sup>+</sup> calc'd for C<sub>8</sub>H<sub>6</sub>Cl<sub>2</sub>N<sub>4</sub>O<sub>2</sub> *m/z* 258.9789, found 258.9795

## PREPARATION OF 1-(4-ACETYLPHENYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 6L

Oxalyl chloride (2.0 M in CH<sub>2</sub>Cl<sub>2</sub>; 3.0 mL, 6.0 mmol) was added dropwise over 2-3 min to a stirred suspension of 4-acetylbenzoic acid (493 mg, 3.0 mmol) and DMF (1-2drops) in CH<sub>2</sub>Cl<sub>2</sub> (6 mL) at 0 °C under nitrogen. After complete addition, the mixture was allowed to warm to room temperature and stirred overnight. The mixture was concentrated under vacuum to leave the acid chloride 51, which was used directly in the tetrazolone-forming step below (yield assumed quantitative = 548 mg).

## See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 4-acetylbenzoyl chloride (548 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (100 mL) and a 1:1 mixture of saturated NaHCO<sub>3</sub> (75 mL) and H<sub>2</sub>O (75 mL). The organic layer was extracted with a 1:1 mixture of saturated NaHCO<sub>3</sub> (25 mL) and H<sub>2</sub>O (25 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. The aqueous layer was filtered to remove minor insoluble items and then EtOAc (100 mL) was added to the filtrate. The pH was adjusted to < 3 using 6N HCl with efficient stirring (care: if an azidohydrin is formed from the reaction, then treatment with acid may release HCN). The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 75 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the crude product (445 mg) of *ca*.

90-95% purity. The crude product was purified by column chromatography on silica gel (ISCO Combiflash) using CH<sub>2</sub>Cl<sub>2</sub> / MeOH (1:0 to 9:1) as eluent to give the product (360 mg, 59%) as a solid.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz):  $\delta$  8.11 (d, J = 7.2 Hz, 2H), 8.02 (d, J = 9.0 Hz, 2H), 2.58 (s, 3H), -1.0 (br. s, 1H)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  196.8, 150.1, 137.8, 135.1, 129.7, 118.3, 26.7

 $m/z = 203.30 \text{ [M-H]}^+$ 

HRMS (EI): [M+H]<sup>+</sup> calc'd for C<sub>9</sub>H<sub>8</sub>N<sub>4</sub>O<sub>2</sub> m/z 205.0726, found 205.0721

## PREPARATION OF 1-(3-CYANOPHENYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 6M

## See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 3-cyanobenzoyl chloride (497 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (10 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (10 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 10 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (20 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 10 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (505 mg, 90%) as a solid. A sample was recrystallized from EtOAc

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz):  $\delta$  8.26 (m, 1H), 8.16 (ddd, J = 8.7, 2.4, 1.2 Hz, 1H), 7.87 (dt, J = 7.8, 2.7 Hz, 1H), 7.48 (dt, J = 8.3, 0.6 Hz, 1H), -1.1 (br. s, 1H)

 $^{13}$ C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  150.1, 134.9, 131.1, 131.0, 123.7, 122.1, 117.9, 112.4

 $m/z = 188.30 \text{ [M+H]}^{+} \text{ and } 186.42 \text{ [M-H]}^{+}$ 

HRMS (EI):  $[M-H]^+$  calc'd for  $C_8H_5N_5O$  m/z 186.0486, found 186.0427

## PREPARATION OF 1-(3-(PENTAFLUOROSULFANYL)PHENYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 6N

Oxalyl chloride (2.0 M in  $CH_2Cl_2$ ; 0.75 mL, 1.5 mmol) was added dropwise over 2-3 min to a stirred suspension of 3-(pentafluorosulfanyl)benzoic acid (248 mg, 1.0 mmol) and DMF (2-3drops) in  $CH_2Cl_2$  (3 mL) at 0 °C under nitrogen. After complete addition, the mixture was stirred at 0 °C for 1 hr then allowed to warm to room temperature and stirred for 3 hr. The mixture was concentrated under vacuum to leave the acid chloride 5n, which was used directly in the tetrazolone-forming step below (yield assumed quantitative = 267 mg).

See 'Important safety notice for all experiments' (page S6); Blast shield employed

Note: 18 equivalent of TMS-N<sub>3</sub> were used in order to ensure complete coverage of the material in a vial with pressure-release cap

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 3-(pentafluorosulfanyl)benzoyl chloride (267 mg, 1.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (20 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (10 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (8 x 10 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (30 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 30 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (187 mg, 65%) as a solid.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz): δ 8.40 (t, J = 2.1 Hz, 1H), 8.14 (d, J = 8.1 Hz, 1H), 7.92-7.96 (m, 1H), 7.80 (t, J = 8.6 Hz, 1H), -1.1 (br. s, 1H)

<sup>19</sup>F NMR (DMSO- $d_6$ , 272MHz):  $\delta$  -114.4 (quin., J = 145 Hz), -136.2 (d, J = 145 Hz)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  153.0 (quin., J = 17.3 Hz), 150.2, 134.7, 130.9, 124.5 (quin., J = 4.7 Hz), 123.0, 116.1 (quin., J = 5.0 Hz)

 $m/z = 287.45 \text{ [M-H]}^+$ 

HRMS (EI):  $[M-H]^+$  calc'd for  $C_7H_5F_3N_4OS$  m/z 287.0026, found 287.0059

## PREPARATION OF 1-(4-(METHYLTHIO)PHENYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 60

Oxalyl chloride (2.0 M in  $CH_2Cl_2$ ; 3.0 mL, 6.0 mmol) was added dropwise over 2-3 min to a stirred suspension of 4-(methylthio)benzoic acid (505 mg, 3.0 mmol) and DMF (1-2drops) in  $CH_2Cl_2$  (6 mL) at 0 °C under nitrogen. After complete addition, the mixture was allowed to warm to room temperature and stirred overnight. The mixture was concentrated under vacuum to leave the acid chloride **50**, which was used directly in the tetrazolone-forming step below (yield assumed quantitative = 560 mg).

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 4-(methylthio)benzoyl chloride (560 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (100 mL) and a 1:1 mixture of saturated NaHCO<sub>3</sub> (50 mL) and H<sub>2</sub>O (50 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (2 x 30 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (75 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was carefully adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 50 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford a crude residue containing product (*ca.* 93% purity). The residue was purified by column chromatography on silica gel (ISCO Combiflash) using CH<sub>2</sub>Cl<sub>2</sub> / MeOH (1:0 to 9:1) as eluent to give the product (473 mg, 73%) as a solid.

 $^{1}$ H NMR (DMSO- $d_{6}$ , 300MHz):  $\delta$  7.78-7.73 (m, 2H), 7.42-7.38 (m, 2H), 2.49 (s, 3H), -1.3 (br. s, 1H)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  150.2, 137.9, 131.1, 126.6, 120.1, 14.7

 $m/z = 209.27 \text{ [M+H]}^+ \text{ and } 207.41 \text{ [M-H]}^+$ 

HRMS (EI):  $[M+H]^+$  calc'd for  $C_8H_8N_4OS$  m/z 209.0497, found 209.0482

#### PREPARATION OF 1-(3,4,5-TRIMETHOXYPHENYL)-1,4-DIHYDRO-5*H*-TETRAZOL-5-ONE 6P

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 3,4,5-trimethoxybenzoyl chloride (692 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (10 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (10 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 10 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (20 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was carefully adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 10 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford a crude residue containing product and carboxylic acid impurity. The residue was purified by column chromatography on silica gel (ISCO Combiflash) using hexanes / EtOAc (1:0 to 0:1) as eluent to give the product (155 mg, 20%) as a solid.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz):  $\delta$  7.14 (s, 2H), 3.79 (s, 6H), 3.67 (s, 3H)

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 75MHz): δ 153.3, 150.3, 136.6, 130.1, 97.6, 60.2, 56.1

 $m/z = 253.34 \text{ [M+H]}^{+} \text{ and } 251.39 \text{ [M-H]}^{+}$ 

HRMS (EI):  $[M+H]^+$  calc'd for  $C_{10}H_{12}N_4O_4$  m/z 253.0937, found 253.0933

## PREPARATION OF 2-(5-OXO-4,5-DIHYDRO-1H-TETRAZOL-1YL)PHENYL ACETATE 6Q

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 2-chlorocarbonyl)phenyl acetate (566 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (10 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (10 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 10 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (20 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was carefully adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 10 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford a crude residue containing product and other impurities. The residue was purified by column chromatography on silica gel (ISCO Combiflash) using hexanes / EtOAc (1:0 to 0:1) as eluent to give the product (367 mg, 56%) as a solid.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz): δ 7.66-7.54 (m, 2H), 7.50-7.38 (m, 2H), 2.16 (s, 3H), -1.3 (br. s, 1H) <sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz): δ 168.2, 150.5, 144.4, 130.5, 126.9, 126.7, 125.4, 124.6, 20.6  $m/z = 219.45 \text{ [M-H]}^+$ 

HRMS (EI):  $[M-H]^+$  calc'd for  $C_9H_8N_4O_3$  m/z 219.0518, found 219.0507

# PREPARATION OF 1-((1-(4-CHLOROBENZOYL)-5-METHOXY-1H-INDOL-3-YL)METHYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 6R

Oxalyl chloride (2.0 M in  $CH_2Cl_2$ ; 2.3 mL, 4.6 mmol) was added dropwise over 2-3 min to a stirred suspension of 2-(1-(4-chlorobenzoyl)-5-methoxy-1H-indol-3-yl)acetic acid, also known as Indomethacin (1.07 g, 3.0 mmol) and DMF (1-2drops) in  $CH_2Cl_2$  (9 mL) at 0 °C in a vial with pressure-release top. After complete addition, the mixture was allowed to warm to room temperature and stirred at room temperature for ca. 60 min. The mixture was concentrated under vacuum to leave the acid chloride  $\mathbf{5r}$ , which was used directly in the tetrazolone-forming step below after drying on a high vacuum for 30 min (yield assumed quantitative = 1.13 g).

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

Note: 4.8 mL (12 equiv.) of azidotrimethylsilane used in order to give sufficient volume for acid chloride to dissolve

A stirred mixture of azidotrimethylsilane (4.8 mL, 18 mmol) and 2-(1-(4-chlorobenzoyl)-5-methoxy-1*H*-indol-3-yl)acetyl chloride (1.13 g, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The temperature was lowered to 90 °C and the mixture was stirred at 90 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and then the mixture was dry-loaded on to silica gel by evaporation from EtOAc. The mixture was purified by column chromatography on silica gel (ISCO Combiflash) using hexanes / EtOAc (1:0 to 0:1) as eluent to the product (0.98 g, 82%) as a solid.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz): δ 7.69-7.60 (m, 4H), 7.21 (d, J = 2.4 Hz, 1H), 6.89 (d, J = 9.0 Hz, 1H), 6.73 (dd, J = 9.0, 2.4 Hz, 1H), 5.17 (s, 2H), 3.73 (s, 3H), 2.39 (s, 3H), -1.6 (br. s, 1H)

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 75MHz): δ 168.0, 155.6, 151.6, 137.9, 137.2, 133.7, 131.3, 130.3, 129.3, 129.1, 114.7, 113.4, 111.6, 101.6, 55.3, 37.1, 13.0

 $m/z = 398.41 \text{ [M+H]}^+ \text{ and } 396.53 \text{ [M-H]}^+$ 

HRMS (EI):  $[M+H]^+$  calc'd for  $C_{19}H_{16}CIN_5O_3$  m/z 398.1020, found 398.1034

HRMS (EI):  $[M-H]^+$  calc'd for  $C_{19}H_{16}ClN_5O_3$  m/z 396.0863, found 396.0823

# Preparation of 4-(5-0x0-4,5-dihydro-1*h*-tetrazol-1-yl)-dipropylbenzenesulfonamide 6s

Oxalyl chloride (2.0 M in  $CH_2Cl_2$ ; 3.0 mL, 6.0 mmol) was added dropwise over 2-3 min to a stirred suspension of 4-(N,N-dipropylsulfamoyl)benzoic acid (856 mg, 3.0 mmol) and DMF (1-2drops) in  $CH_2Cl_2$  (6 mL) at 0 °C under nitrogen. After complete addition, the mixture was allowed to warm to room temperature and stirred overnight. The mixture was concentrated under vacuum to leave the acid chloride **5s**, which was used directly in the tetrazolone-forming step below (yield assumed quantitative = 911 mg).

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 4-(*N*,*N*-dipropylsulfamoyl)benzoyl chloride (911 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (30 mL) and an aqueous saturated NaHCO<sub>3</sub> (30 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (8 x 30 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (60 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was carefully adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 30 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (665 mg, 68%) as a solid.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz): δ 8.13-8.08 (m, 2H), 7.98-7.93 (m, 2H), 3.03 (t, J = 7.7 Hz, 2H), 1.47 (app. sextet, J = 7.4 Hz, 2H), 0.79 (t, J = 7.4 Hz, 3H), -1.1 (br. s, 1H)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  150.1, 137.8, 137.3, 128.4, 119.0, 49.6, 21.6, 11.0

 $m/z = 326.35 \text{ [M+H]}^{+} \text{ and } 324.43 \text{ [M-H]}^{+}$ 

HRMS (EI):  $[M+H]^+$  calc'd for  $C_{13}H_{19}N_5O_3S$  m/z 326.1287, found 326.1257

HRMS (EI):  $[M-H]^+$  calc'd for  $C_{13}H_{19}N_5O_3S$  m/z 324.1130, found 324.1116

PREPARATION OF 1-(4'-((1,7-dimethyl-2'-propyl-1H,3'H-[2,5'-dibenzo[D]imidazol]-3'-yl)methyl-[1,1'-biphenyl-2-yl)-1,4-dihydro-5H-tetrazol-5-one  $\underline{6}\underline{T}$ 

Oxalyl chloride (2.0 M in CH<sub>2</sub>Cl<sub>2</sub>; 0.75 mL, 1.5 mmol) was added dropwise over 2-3 min to a stirred suspension of 4'-((1,7-dimethyl-2'-propyl-1*H*,3'*H*-[2,5'-dibenzo[*d*]imidazol]-3'-yl)methyl-[1,1'-biphenyl)-2-carboxylic acid, also known as Telmisartan (514 mg, 1.0 mmol) and DMF (3 drops) in CH<sub>2</sub>Cl<sub>2</sub> (6 mL) at 0 °C under nitrogen. After complete addition, the mixture was stirred at 0 °C for 10 min, then allowed to warm to room temperature and stirred for 30 min (a yellow then orange solution develops). The mixture was concentrated under vacuum, then fresh CH<sub>2</sub>Cl<sub>2</sub> (2 mL) was added to the residue and the mixture concentrated under vacuum again to leave the acid chloride **5t**, which was used directly in the tetrazolone-forming step below, after drying on a high vacuum for 15 min (yield assumed quantitative = 533 mg).

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

Note: 4.8 mL (36 equiv.) of azidotrimethylsilane used in order to give sufficient volume for the reaction

A stirred mixture of azidotrimethylsilane (4.8 mL, 36 mmol) and 4'-((1,7-dimethyl-2'-propyl-1*H*,3'*H*-[2,5'-dibenzo[*d*]imidazol]-3'-yl)methyl-[1,1'-biphenyl)-2-carbonyl chloride (533 mg, 1.0 mmol) was heated from room temperature to 100 °C (block temperature) in a round bottom flask with reflux condenser under an atmosphere of nitrogen. The mixture was then stirred at 100 °C for 2 hours. After cooling, the mixture was concentrated under vacuum and MeOH was added to the residue. The mixture was dry-loaded on to silica gel and then purified by column chromatography on silica gel (ISCO Combiflash) using CH<sub>2</sub>Cl<sub>2</sub> / MeOH (1:0 to 92:8) as eluent to give the pure product (87 mg) and mixed fractions. The mixed fractions were re-purified by column chromatography on silica gel (ISCO

Combiflash) using  $CH_2Cl_2$  / MeOH (1:0 to 92:8) as eluent to give the product (97 mg) as a solid. Total yield of product = 184 mg (33%). Also obtained, was a faster-eluting unidentified by-product (181 mg).

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz):  $\delta$  7.73 (s, 1H), 7.64-7.47 (m, 7H), 7.29-7.11 (m, 6H), 5.58 (s, 2H), 3.80 (s, 3H), 2.86 (t, J = 7.5 Hz, 2H), 2.61 (s, 3H), 1.75 (sextet, J = 7.5 Hz, 2H), 0.94 (t, J = 7.5 Hz, 3H), -1.5 (br. s, 1H)

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 75MHz): δ 156.2, 154.0, 151.3, 142.6, 142.4, 139.2, 136.7, 136.7, 136.6, 134.7, 130.9, 130.7, 130.3, 128.8, 128.6, 128.6, 128.3, 126.8, 123.3, 123.2, 122.1, 121.8, 118.6, 110.4, 109.1, 45.9, 31.7, 28.7, 20.7, 16.4, 13.8

 $m/z = 555.66 \text{ [M+H]}^{+} \text{ and } 553.75 \text{ [M-H]}^{+}$ 

HRMS (EI):  $[M+H]^+$  calc'd for  $C_{33}H_{30}N_8O$  m/z 555.2621, found 555.2585

HRMS (EI):  $[M-H]^+$  calc'd for  $C_{33}H_{30}N_8O$  m/z 553.2465, found 553.2411

A separate HRMS (EI) obtained:  $[M+H]^+$  calc'd for  $C_{33}H_{30}N_8O$  m/z 555.2621, found 555.2633

PREPARATION OF 1-(4-(1-(3,5,5,8,8-PENTAMETHYL-5,6,7,8-TETRAHYDRONAPHTHALEN-2-YL)VINYL)PHENYL)-1,4-DIHYDRO-5*H*-TETRAZOL-5-ONE 6U

$$\begin{array}{c} \text{TMS-N}_3 \\ \text{100 °C} \\ \text{OH} \end{array}$$

Oxalyl chloride (2.0 M in  $CH_2Cl_2$ ; 0.38 mL, 0.75 mmol) was added dropwise over 1 min to a stirred suspension of 4-(1-(3,5,5,8,8-pentamethyl-5,6,7,8-tetrahydronaphthalen-2-yl)vinyl)benzoic acid, also known as Bexarotene (175 mg, 0.5 mmol) and DMF (1-2drops) in  $CH_2Cl_2$  (5 mL) at 0 °C under nitrogen. After complete addition, the mixture was allowed to warm to room temperature and for 2 hr. The mixture was concentrated under vacuum to leave the acid chloride  $\mathbf{5u}$ , which was used directly in the tetrazolone-forming step below (yield assumed quantitative = 184 mg).

See 'Important safety notice for all experiments' (page S6); Blast shield employed

Note: 1.0 mL (15 equiv.) of azidotrimethylsilane used in order to give sufficient volume for the reaction

A stirred mixture of azidotrimethylsilane (1.0 mL, 7.5 mmol) and 4-(1-(3,5,5,8,8-pentamethyl-5,6,7,8-tetrahydronaphthalen-2-yl)vinyl)benzoyl chloride (184 mg, 0.5 mmol) was heated from room temperature to 100 °C (block temperature) in a 10 mL round-bottom flask under an atmosphere of nitrogen. The mixture was then stirred at 100 °C for 2 hr, by which time LC/MS and TLC analysis indicates completion of the reaction. After cooling, CH<sub>2</sub>Cl<sub>2</sub> and MeOH was added to the mixture, which was then dry-loaded on to silica gel and purified by column chromatography on silica gel (ISCO Combiflash) using hexanes / EtOAc (1:0 to 0:1) as eluent to give the product (173 mg, 89%) as a solid.

<sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>, 300MHz): δ 7.82-7.79 (m, 2H), 7.39-7.36 (m, 2H), 7.13 (s, 1H), 7.06 (s, 1H), 5.84 (s, 1H), 5.18 (s, 1H), 1.91 (s, 3H), 1.63 (br. s, 4H), 1.24 (s, 6H), 1.21 (s, 6H), -1.0 (br. s, 1H)

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 75MHz): δ 150.2, 147.9, 143.7, 141.8, 138.3, 137.9, 133.4, 132.1, 127.9, 127.3, 127.1, 119.5, 115.9, 34.7, 34.6, 33.6, 33.5, 31.7, 31.6, 19.5

 $m/z = 389.63 \text{ [M+H]}^+ \text{ and } 387.68 \text{ [M-H]}^+$ 

HRMS (EI):  $[M+H]^+$  calc'd for  $C_{24}H_{28}N_4O$   $\emph{m/z}$  389.2341, found 389.2344

HRMS (EI):  $[M-H]^+$  calc'd for  $C_{24}H_{28}N_4O$  m/z 387.2185, found 387.2211

#### PREPARATION OF 1-(PYRIDIN-3-YL)-1,4-DIHYDRO-5*H*-TETRAZOL-5-ONE 6V

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and nicotinoyl chloride hydrochloride (425 mg, 2.4 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (10 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (10 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 10 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (20 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. Once acidified, the aqueous layer was re-adjusted to pH 6-7 using saturated NaHCO<sub>3</sub>. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 10 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (330 mg, 65%) as a solid. A sample was recrystallized from EtOAc.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz):  $\delta$  9.04 (d, J = 2.1 Hz, 1H), 8.60 (d, J = 4.5 Hz, 1H), 8.22 (ddd, J = 8.4, 2.7, 1.5 Hz, 1H), 7.58 (dd, J = 8.4, 4.8, Hz, 1H), -1.1 (br. s, 1H)

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 75MHz): δ 150.4, 148.5, 140.7, 131.1, 127.1, 12.2

 $m/z = 162.20 [M-H]^{+}$ 

HRMS (EI):  $[M+H]^+$  calc'd for  $C_6H_5N_5O$  m/z 164.0572, found 164.0542

#### PREPARATION OF 1-(3-CHLOROTHIOPHEN-2-YL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE 6W

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 3-chlorothiophene-2-carbonyl chloride (543 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (10 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (10 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 10 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (20 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 10 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford a mixture of the desired product and 3-chlorothiophene-2-carboxylic acid. This mixture was purified by column chromatography on silica gel (ISCO Combiflash) using hexanes / EtOAc (1:0 to 0:1) as eluent to give the product (185 mg, 30%) as a solid [also obtained from the column was 3-chlorothiophene-2-carboxylic acid (20mg)].

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz):  $\delta$  7.88 (d, J = 6.0 Hz, 1H), 7.25 (d, J = 6.0 Hz, 1H), -1.1 (br. s, 1H)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  150.5, 128.0, 126.6, 125.8, 123.6

 $m/z = 203.26 \text{ [M+H]}^{+} \text{ and } 201.37 \text{ [M-H]}^{+}$ 

# PREPARATION OF 1-(3-(2-CHLOROPHENYL)5-METHYLISOXAZOL-4-YL)-1,4-DIHYDRO-5*H*-TETRAZOL-5-ONE 6X

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and 3-(2-chlorophenyl)-5-methylisoxazole-4-carbonyl chloride (768 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (10 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (10 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 10 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (20 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 10 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (713 mg, 86%) as a solid. A sample was recrystallized from EtOAc.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz):  $\delta$  7.58-7.44 (m, 4H), 2.55 (s, 3H), -1.2 (br. s, 1H)

<sup>13</sup>C NMR (DMSO- $d_6$ , 75MHz):  $\delta$  167.4, 157.9, 150.8, 132.7, 132.1, 130.3, 128.2, 126.2, 115.5, 111.6  $m/z = 278.36 \, [\text{M+H}]^+ \, \text{and} \, 276.51 \, [\text{M-H}]^+$ 

HRMS (EI):  $[M+H]^+$  calc'd for  $C_{11}H_8ClN_5O_2$  m/z 278.0445, found 278.0450

#### PREPARATION OF METHYL (E)-3-(5-0XO-4,5-DIHYDRO-1H-TETRAZOL-1-YL)ACRYLATE 6Y

Oxalyl chloride (2.0 M in  $CH_2Cl_2$ ; 15.2 mL, 30.3 mmol) was added dropwise over 2-3 min to a stirred suspension of monomethyl fumarate (2.63 g, 20.2 mmol) in  $CH_2Cl_2$  (80 mL) at 0 °C under nitrogen. After complete addition, the mixture was stirred at 0 °C for 5min then allowed to warm to room temperature and stirred for 3 hr (a yellow solution develops). A small aliquot was removed and quenched with MeOH – TLC indicated no acid remaining. The mixture was concentrated under vacuum and  $CH_2Cl_2$  (50 mL) was added to the residue and the mixture concentrated under vacuum once more to leave the acid chloride  $5\mathbf{w}$ , which was used directly in the tetrazolone-forming step below (yield assumed quantitative = 3.0 g).

#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

Azidotrimethylsilane (16.1 mL, 121.2 mmol) was added in one portion to the acid chloride from the above procedure (3.0 g, 20.2 mmol) at room temperature (gas evolution was noted). The mixture was place under nitrogen and heated from room temperature to 100 °C (block temperature), then stirred at 100 °C for 90 min. After cooling to room temperature, the excess solvent was removed under vacuum to leave a crude residue. EtOAC (150 mL) and saturated NaHCO<sub>3</sub> (150 mL) were added to the residue. A solid was noticed, so the mixture was filtered. The filter cake was dissolved in H<sub>2</sub>O (350 mL) and then combined with the saturated NaHCO<sub>3</sub> layer of the filtrate. EtOAc (150 mL) was added to the combined aqueous system described above, and the mixture was acidified to *ca.* pH 3 with 1N HCl. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 100 mL). The combined organic extracts were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to leave a crude residue (2.6 g; *ca.* 90% purity of desired product). The residue was purified by column

chromatography on silica gel using  $CH_2Cl_2$  / MeOH (1:0 to 9:1) as eluent to give the product (2.14 g, 62% over 2 steps) as a solid.

<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz): δ 7.73 (d, J = 14.4 Hz, 1H), 6.49 (d, J = 14.4 Hz, 1H), 3.71 (s, 3H), -1.0 (br. s, 1H)

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 75MHz): δ 165.7, 149.8, 132.3, 106.3, 51.9

 $m/z = 169.34 \text{ [M-H]}^+$ 

HRMS (EI):  $[M+H]^+$  calc'd for  $C_5H_6N_4O_3$  m/z 171.0518, found 171.0522

#### PREPARATION OF 1-ETHYL-1,4-DIHYDRO-5*H*-TETRAZOL-5-ONE 6Z

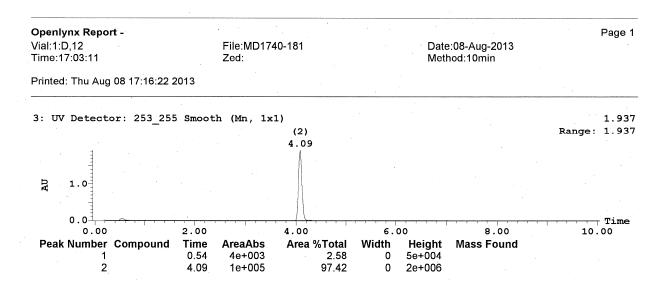
#### See 'Important safety notice for all experiments' (page S6); Blast shield employed

A stirred mixture of azidotrimethylsilane (2.4 mL, 18 mmol) and propionyl chloride (278 mg, 3.0 mmol) was heated from room temperature to 100 °C (block temperature) in a sealed vial with pressure-release cap. The mixture was then stirred at 100 °C overnight (Note: pressure develops during heating). After cooling, the mixture was concentrated under vacuum and the residue partitioned between EtOAc (10 mL) and a saturated aqueous solution of NaHCO<sub>3</sub> (10 mL). The organic layer was extracted with a further quantity of saturated aqueous NaHCO<sub>3</sub> (1 x 10 mL) [note: organic layer assessed by TLC to ascertain if tetrazolone product completely removed. If tetrazolone still present in organic layer, then further extractions with saturated NaHCO<sub>3</sub> are used]. EtOAc (20 mL) was added to the combined NaHCO<sub>3</sub> layers, and the pH was adjusted to < 3 using 6N HCl with efficient stirring. The aqueous and organic layers were partitioned and the aqueous layer was extracted with EtOAc (2 x 10 mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered and the solvent removed under vacuum to afford the product (49 mg, 14%) as a solid.

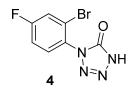
<sup>1</sup>H NMR (DMSO- $d_6$ , 300MHz):  $\delta$  3.86 (q, J = 7.3 Hz, 2H), 1.26 (t, J = 7.3 Hz, 3H), -1.7 (br. s, 1H)

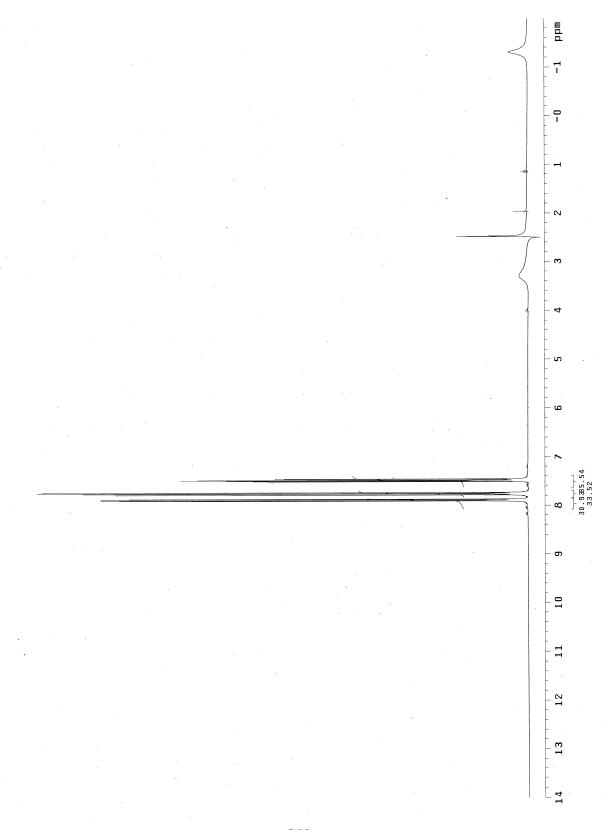
<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 75MHz): δ 151.6, 38.7, 13.7

## LC data for 1-(2-bromo4-fluorophenyl)-1,4-dihydro-5*H*-tetrazol-5-one 4

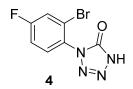


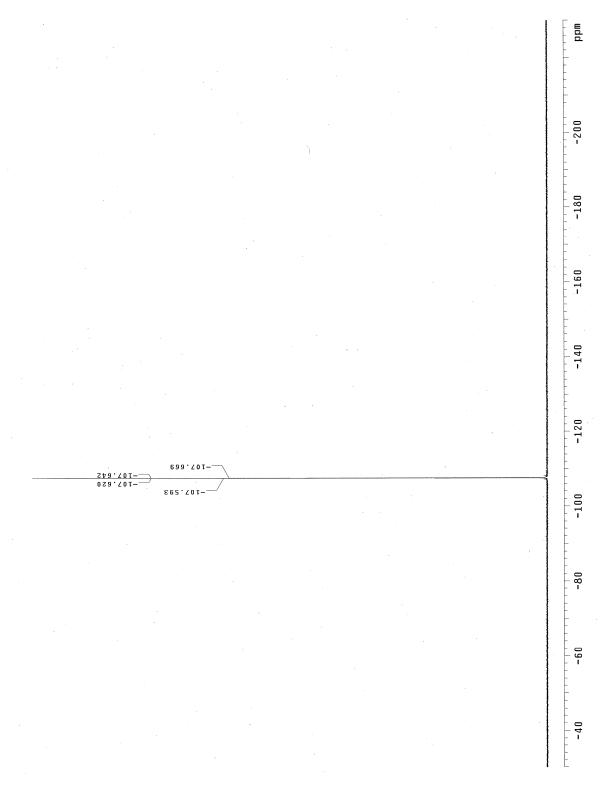
# $^{1}\mathrm{H}$ NMR for 1-(2-bromo4-fluorophenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{4}$



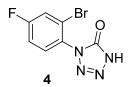


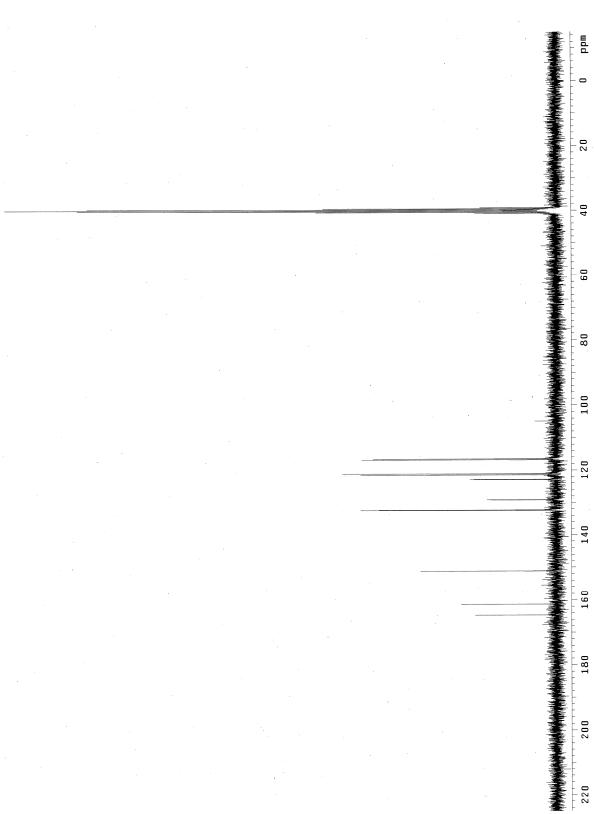
## $^{19}\mathrm{F}$ NMR for 1-(2-bromo4-fluorophenyl)-1,4-dihydro-5*H*-tetrazol-5-one $\underline{4}$







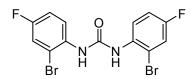


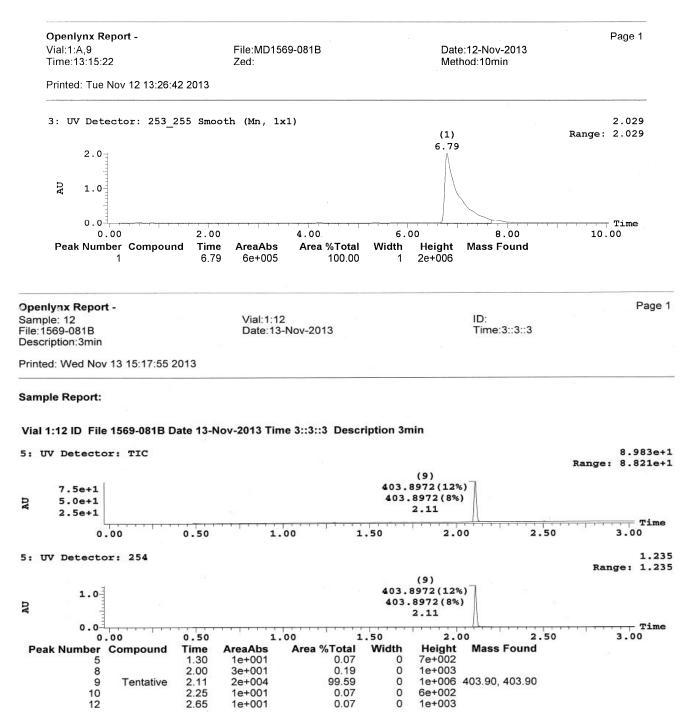


#### LC DATA FOR SYMMETRICAL UREA BIS-(2-BROMO-4-FLUORPHENYL)UREA

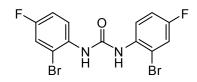
#### BY-PRODUCT ISOLATED FROM OPTIMIZATION REACTION IN TABLE 1, ENTRY 3

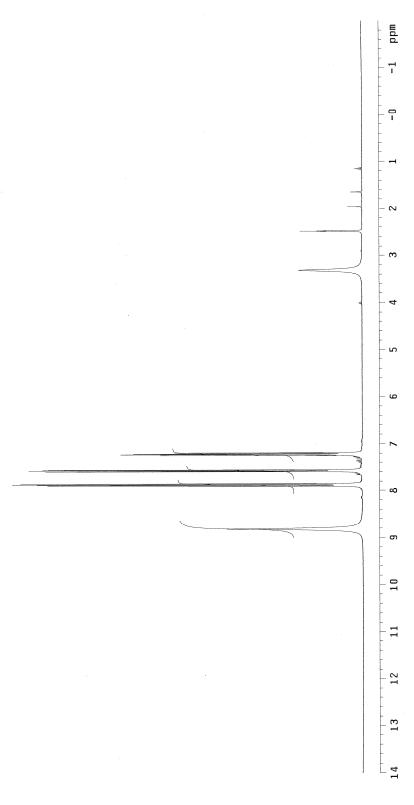
[NOTE: TWO LC ARE SHOWN DUE TO SIGNIFICANT TAILING WITH FIRST METHOD]



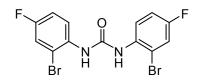


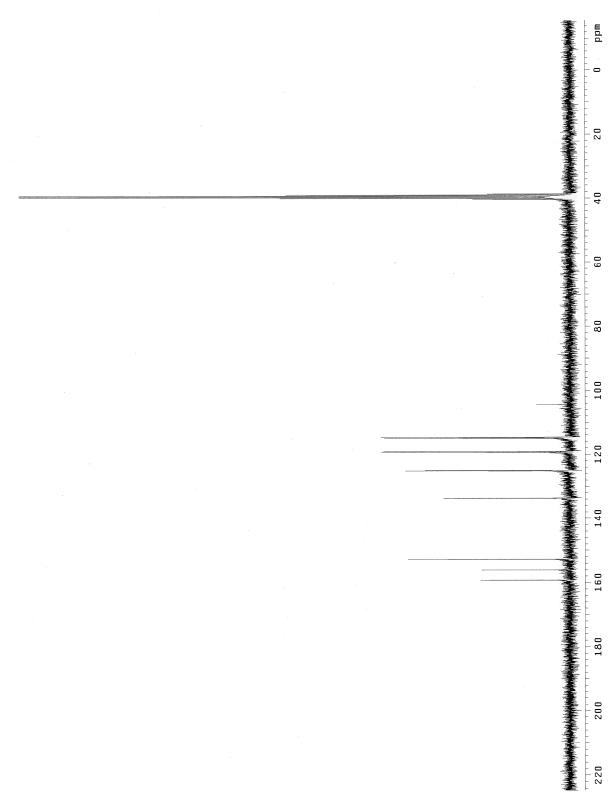
 $^1\mathrm{H}$  NMR for symmetrical urea *bis*-(2-bromo-4-fluorphenyl)urea (by-product isolated from optimization reaction in table 1, entry 3)



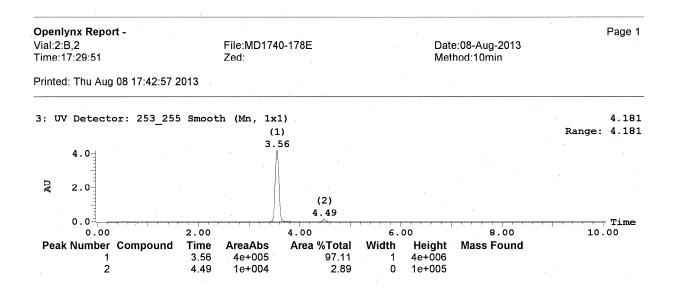


 $^{13}\mathrm{C}$  NMR for symmetrical urea  $\emph{bis}\text{-}(2\text{-bromo-4-fluorphenyl})$  urea (by-product isolated from optimization reaction in table 1, entry 3)

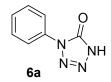


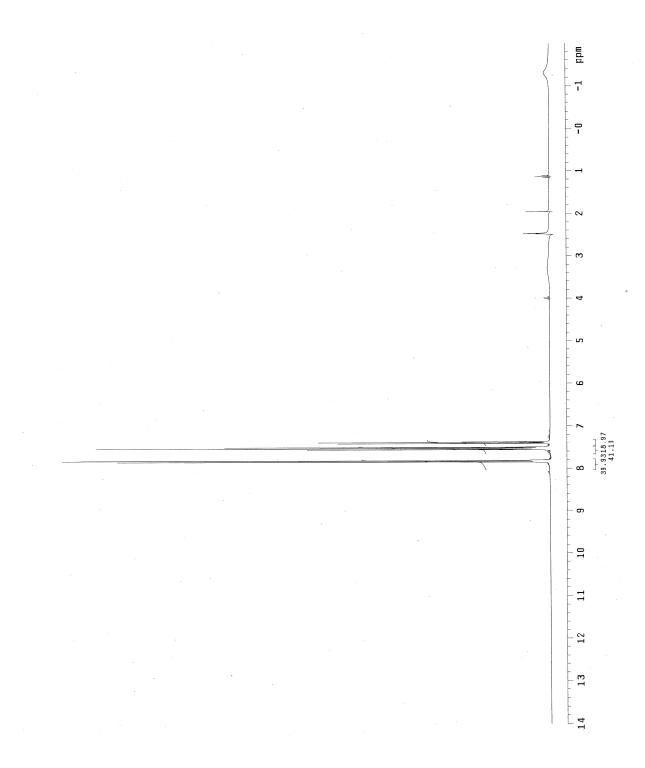


## LC data for 1-phenyl-1,4-dihydro-5*H*-tetrazol-5-one <u>6a</u>

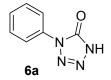


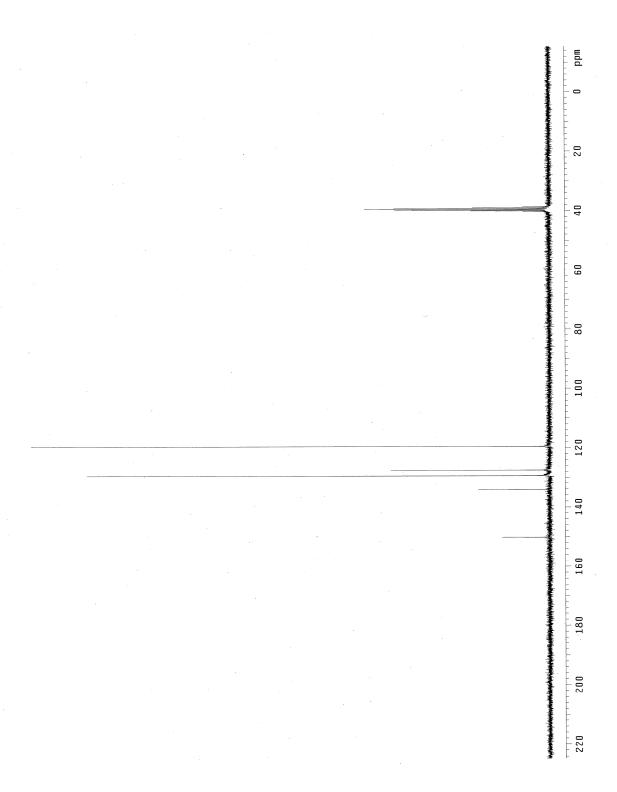
## $^{1}$ H NMR for 1-phenyl-1,4-dihydro-5H-tetrazol-5-one $\underline{6a}$



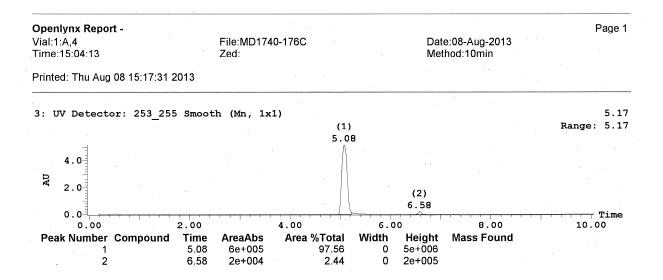


## $^{13}$ C NMR for 1-phenyl-1,4-dihydro-5H-tetrazol-5-one $\underline{6}$ A

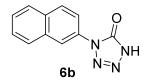


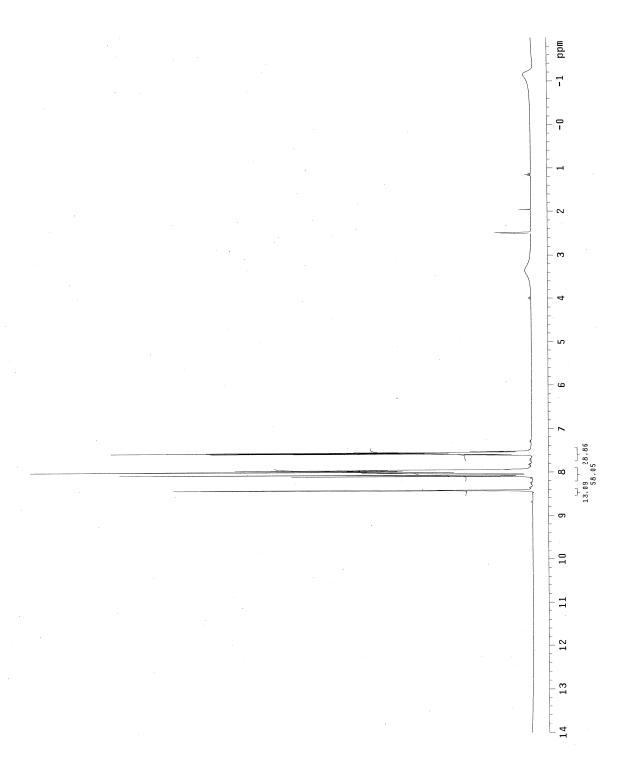


## LC data for 1-(naphthalen-2-yl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6</u>B

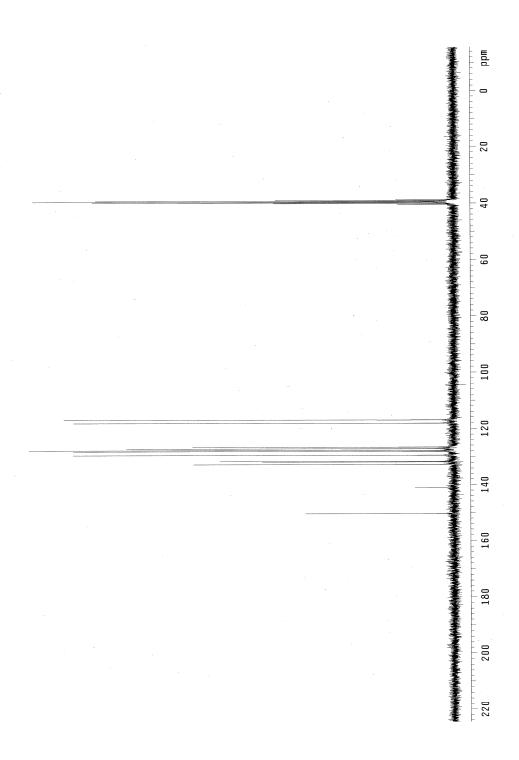


## $^{1}$ H NMR for 1-(naphthalen-2-yl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}$ B

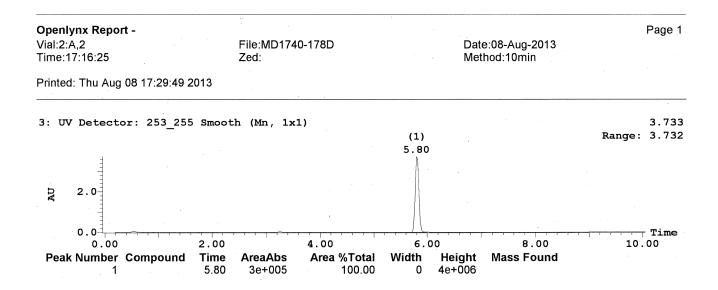




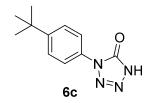
## $^{1}$ H NMR for 1-(naphthalen-2-yl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}$ B

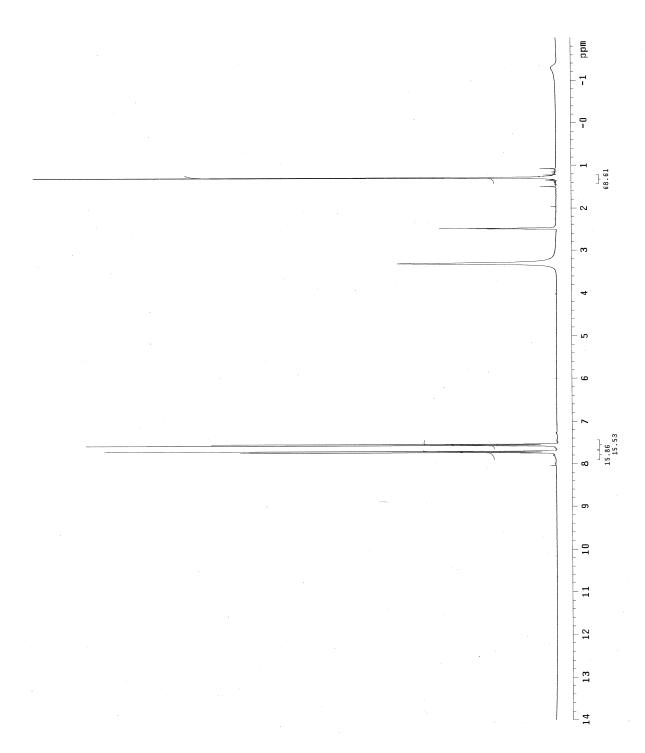


## LC data for 1-(4-(*tert*-butyl)phenyl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6</u>C

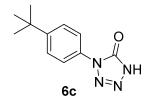


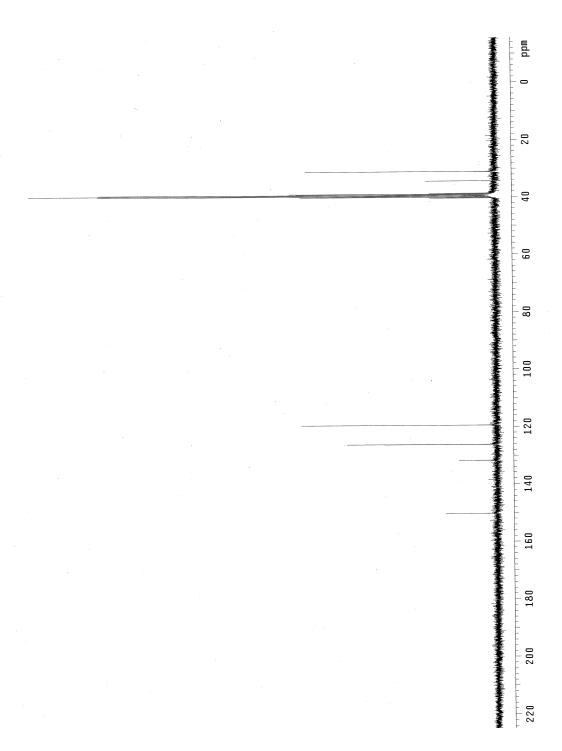
## $^{1}$ H NMR for 1-(4-(tert-butyl)phenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}$ C





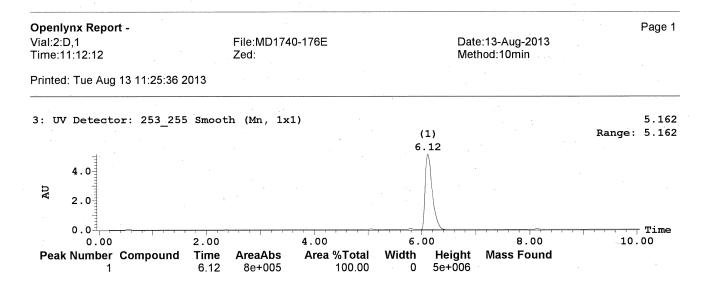
# $^{13}\mathrm{C}$ NMR for 1-(4-(*tert*-butyl)phenyl)-1,4-dihydro-5*H*-tetrazol-5-one $\underline{6\mathrm{C}}$



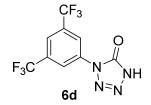


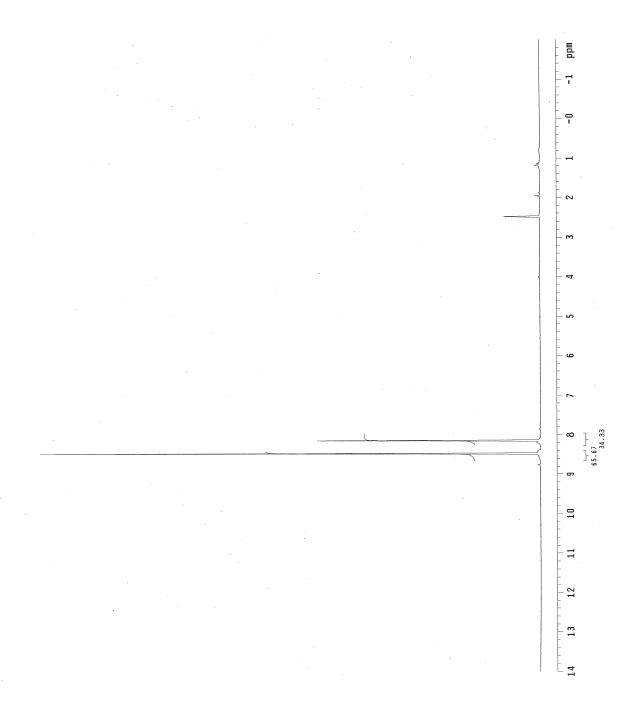
## LC data for 1-(3,5-bis(trifluoromethyl)phenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6D}$

$$F_3C$$
 $O$ 
 $N=N$ 
 $N=N$ 

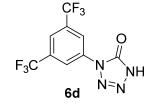


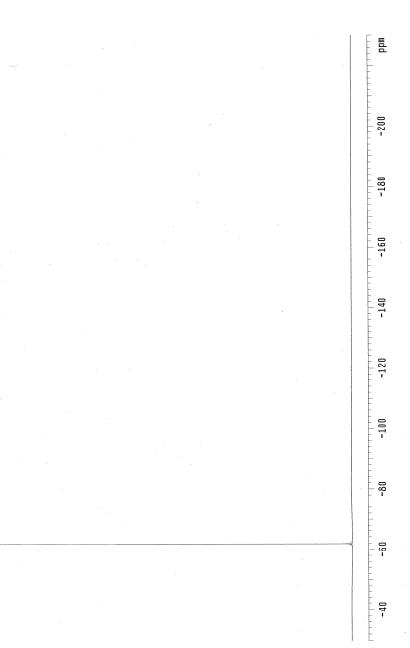
## $^{1}$ H NMR for 1-(3,5-bis(trifluoromethyl)phenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}$ D



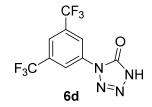


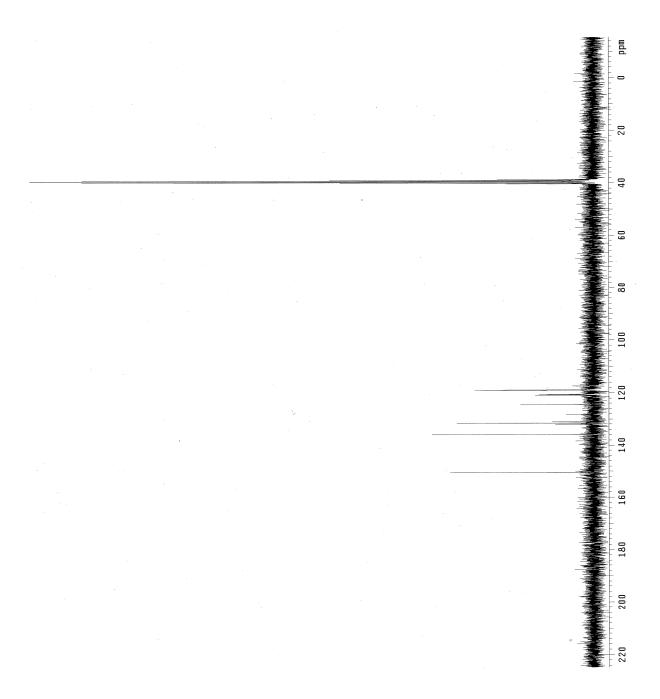
# $^{19}\mathrm{F}$ NMR for 1-(3,5-bis(trifluoromethyl)phenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}\underline{0}$



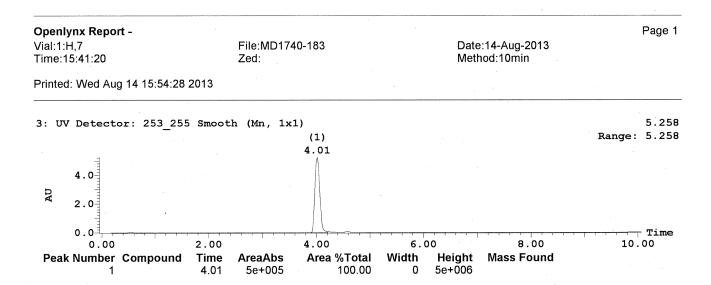


# $^{13}\mathrm{C}$ NMR for 1-(3,5-bis(trifluoromethyl)phenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6D}$

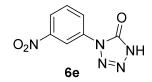


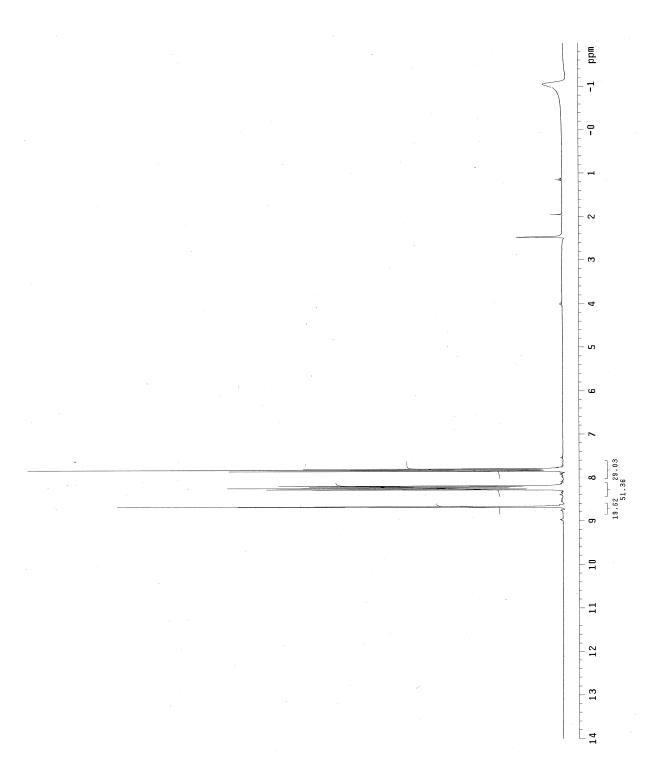


## LC data for 1-(3-nitrophenyl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6e</u>

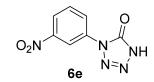


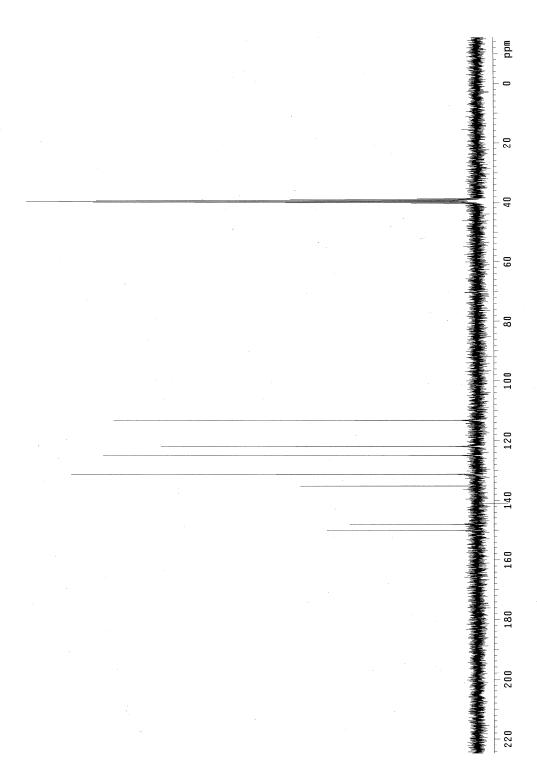
## $^{1}$ H NMR for 1-(3-nitrophenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}$ E





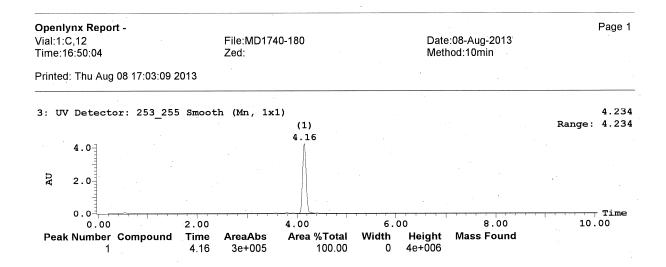
## $^{13}$ C NMR for 1-(3-nitrophenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}$ E



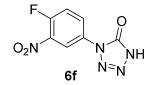


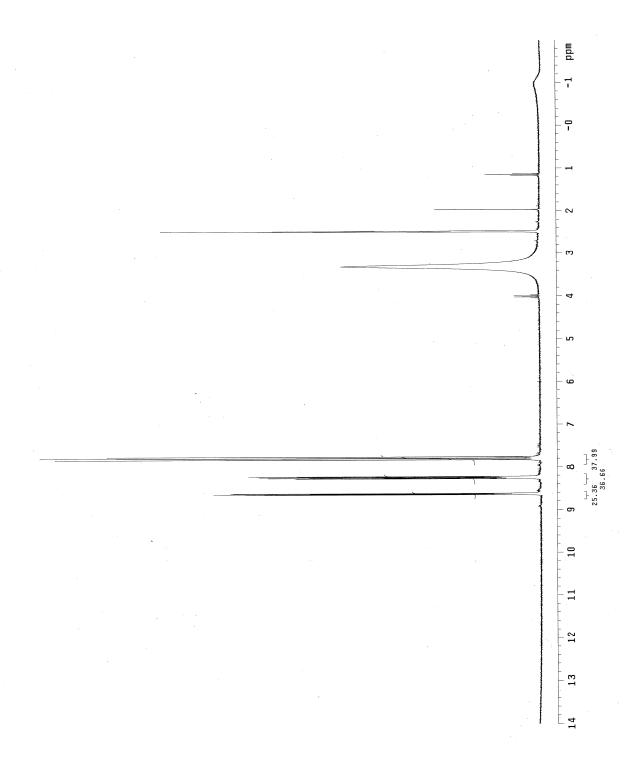
#### LC data for 1-(4-fluoro-5-nitrophenylphenyl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6</u>f

$$O_2N$$
 $O_2N$ 
 $O_2N$ 

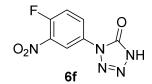


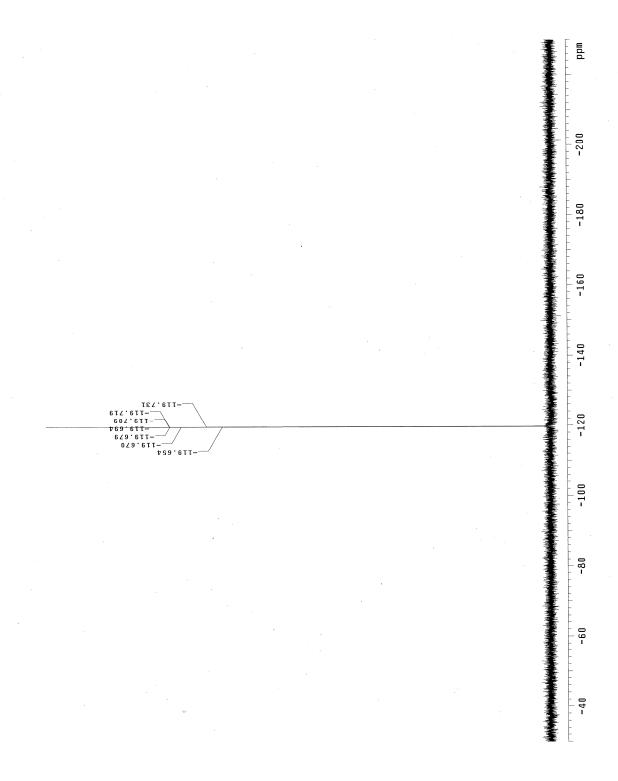
## $^{1}$ H NMR for 1-(4-fluoro-5-nitrophenylphenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}$ F



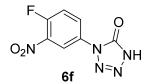


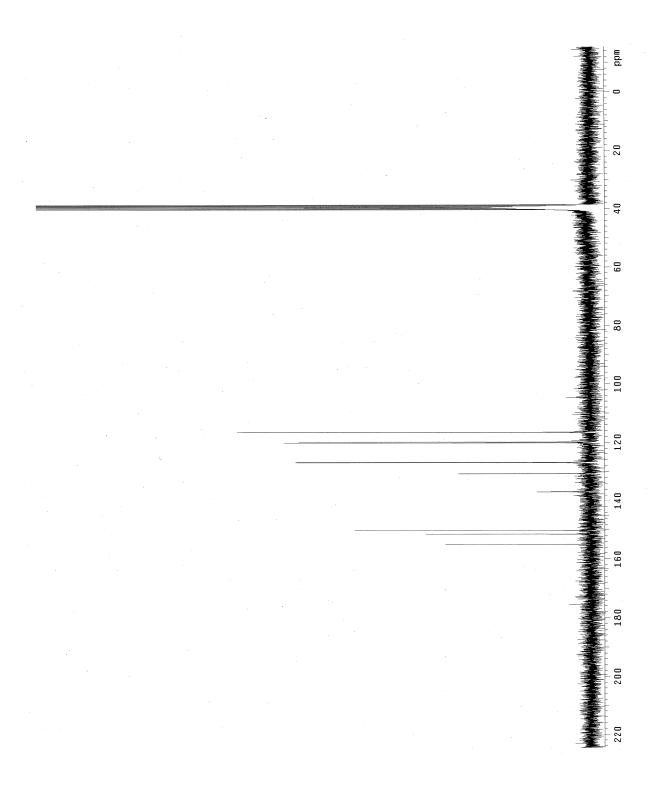
## $^{19}\mathrm{F}$ NMR for 1-(4-fluoro-5-nitrophenylphenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6\mathrm{f}}$



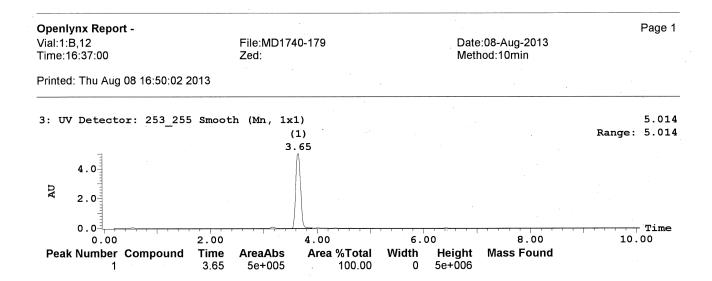


## $^{13}\mathrm{C}$ NMR for 1-(4-fluoro-5-nitrophenylphenyl)-1,4-dihydro-5*H*-tetrazol-5-one $\underline{6\mathrm{f}}$



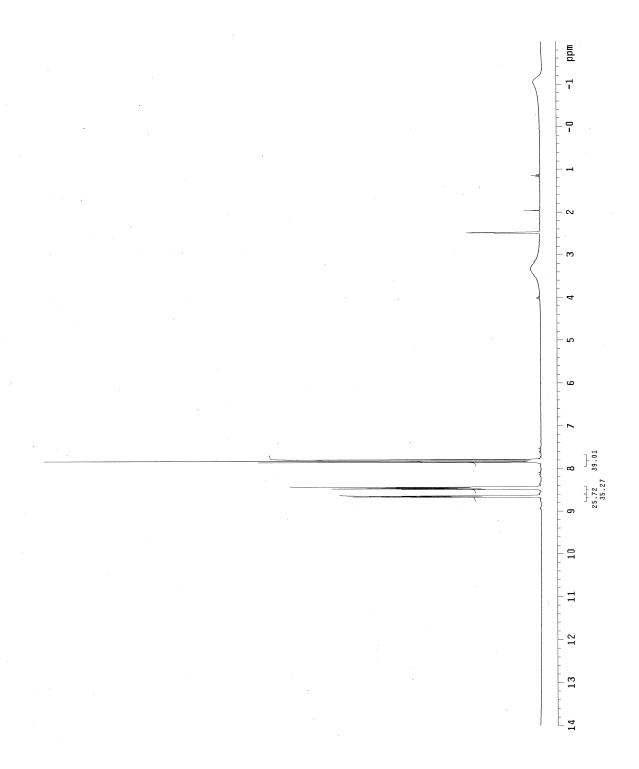


#### LC data for 1-(2-fluoro-5-nitrophenylphenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6G}$

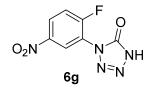


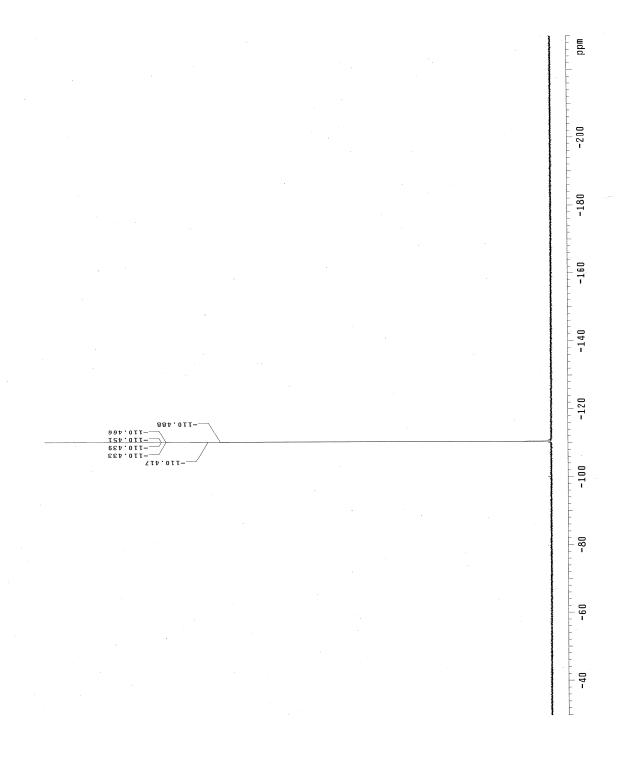
## $^1\mathrm{H}$ NMR for 1-(2-fluoro-5-nitrophenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}\mathrm{G}$





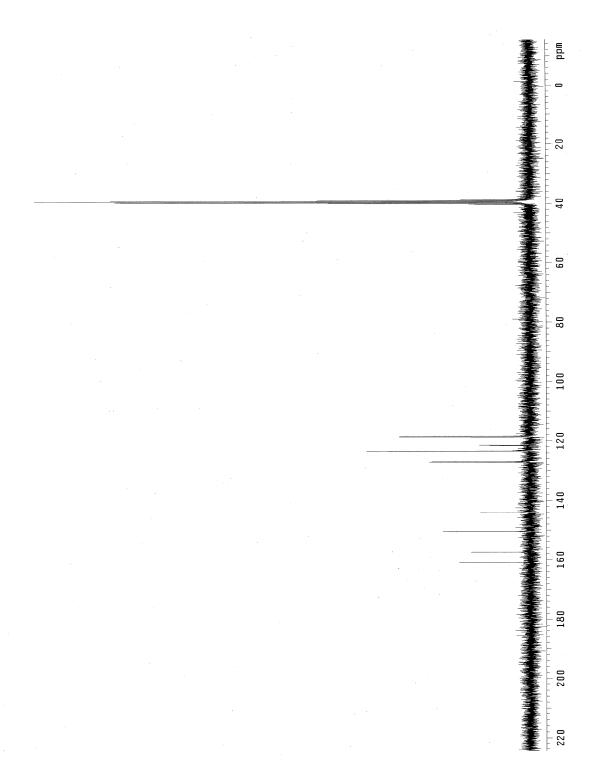
## $^{19}\mathrm{F}$ NMR for 1-(2-fluoro-5-nitrophenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6G}$



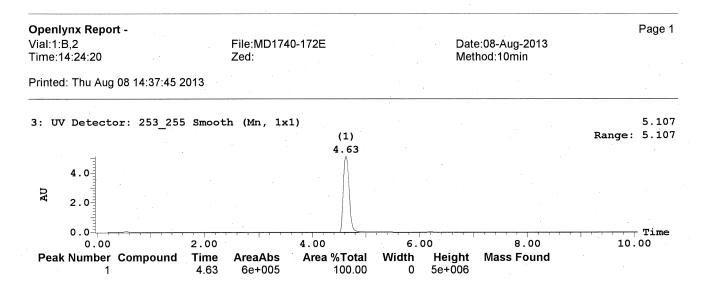


## $^{13}\mathrm{C}$ NMR for 1-(2-fluoro-5-nitrophenyl)-1,4-dihydro-5*H*-tetrazol-5-one $\underline{6G}$

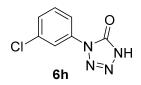


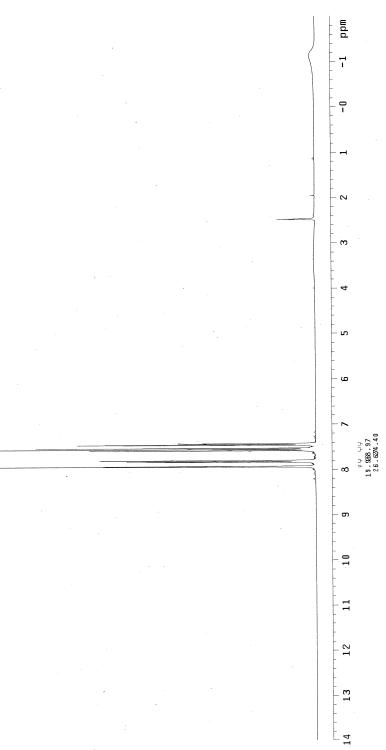


#### LC data for 1-(3-chlorophenyl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6h</u>

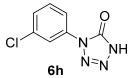


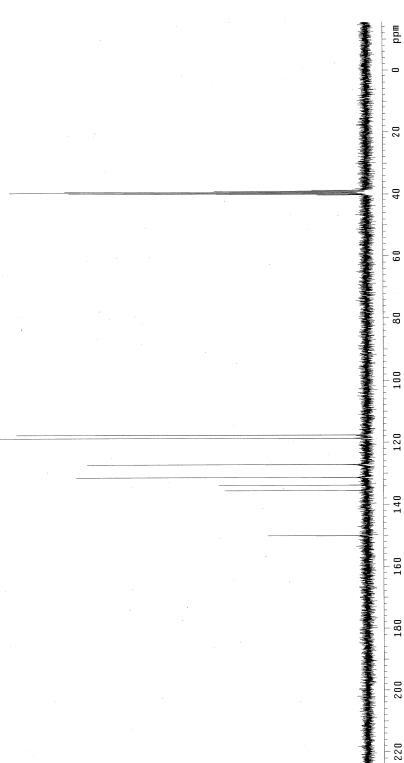
## $^{1}$ H NMR for 1-(3-chlorophenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}$ H



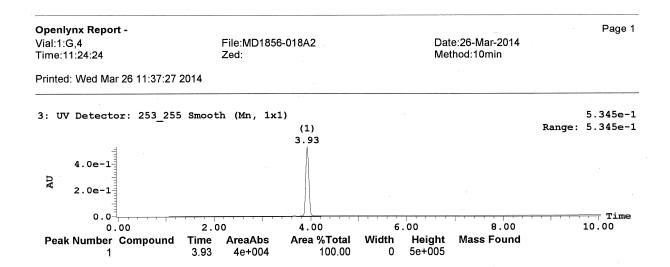


## $^{13}$ C NMR for 1-(3-chlorophenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}$ H

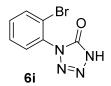


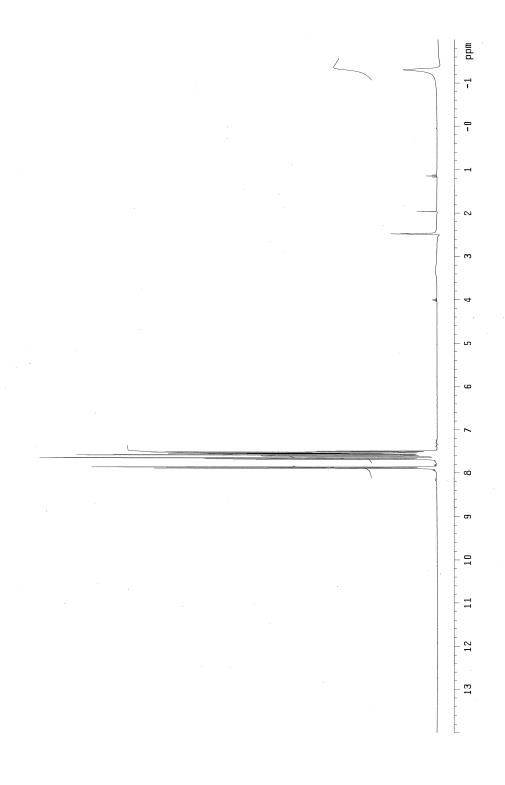


#### LC data for 1-(2-bromophenyl)-1,4-dihydro-5*H*-tetrazol-5-one <u>61</u>

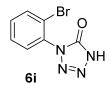


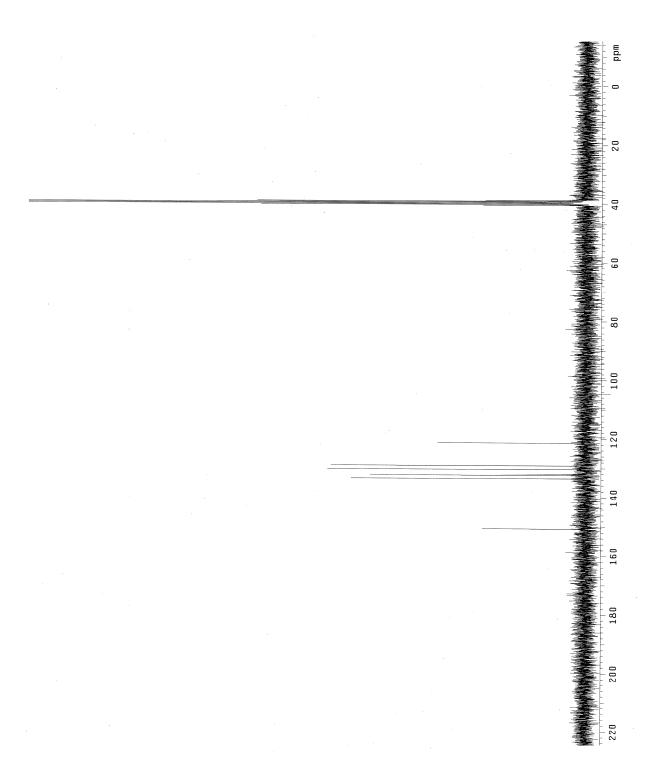
## $^{1}$ H NMR for 1-(2-bromophenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{61}$



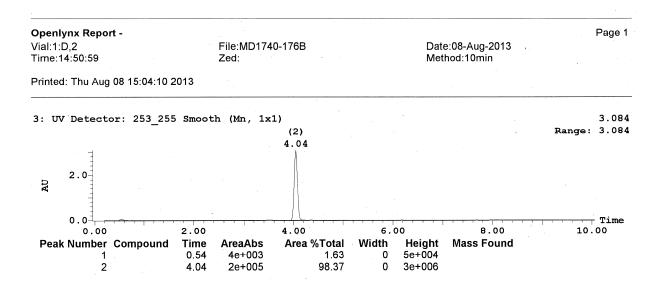


# $^{13}\mathrm{C}$ NMR for 1-(2-bromophenyl)-1,4-dihydro-5*H*-tetrazol-5-one $\underline{61}$

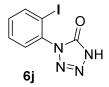


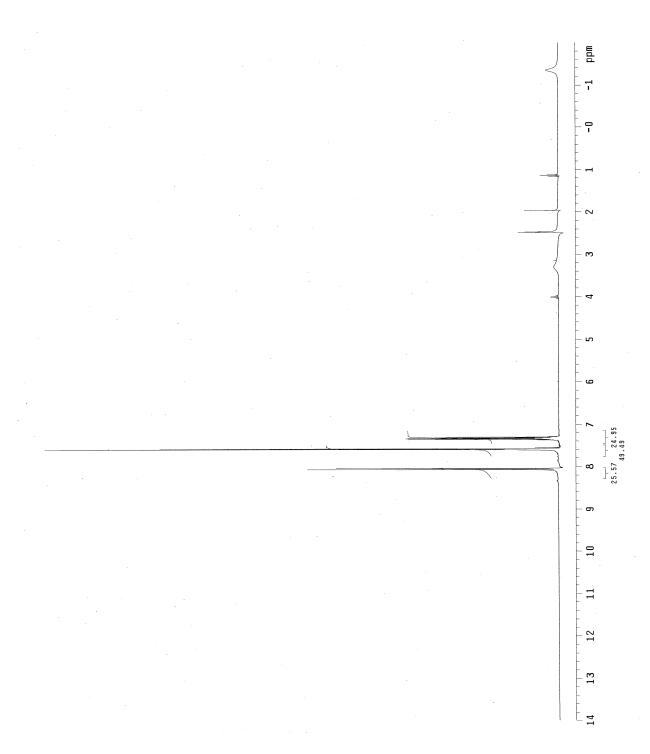


#### LC data for 1-(2-iodophenyl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6j</u>

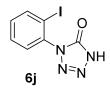


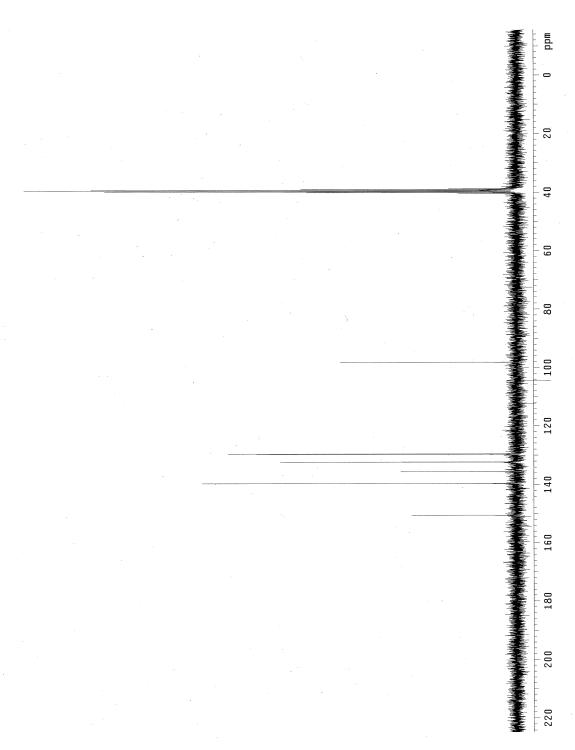
## $^{1}$ H NMR for 1-(2-iodophenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}$ J



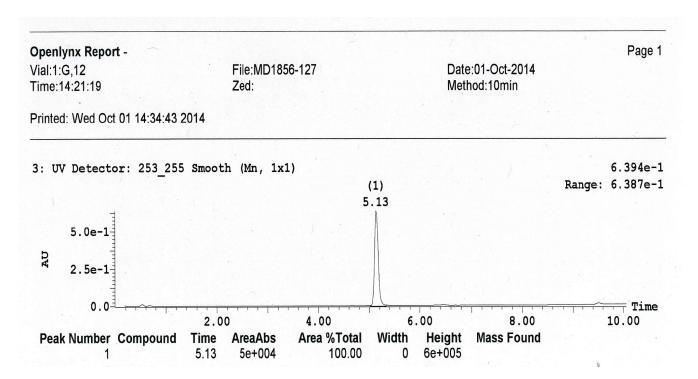


# $^{13}\mathrm{C}$ NMR for 1-(2-iodophenyl)-1,4-dihydro-5*H*-tetrazol-5-one $\underline{6}\mathrm{J}$

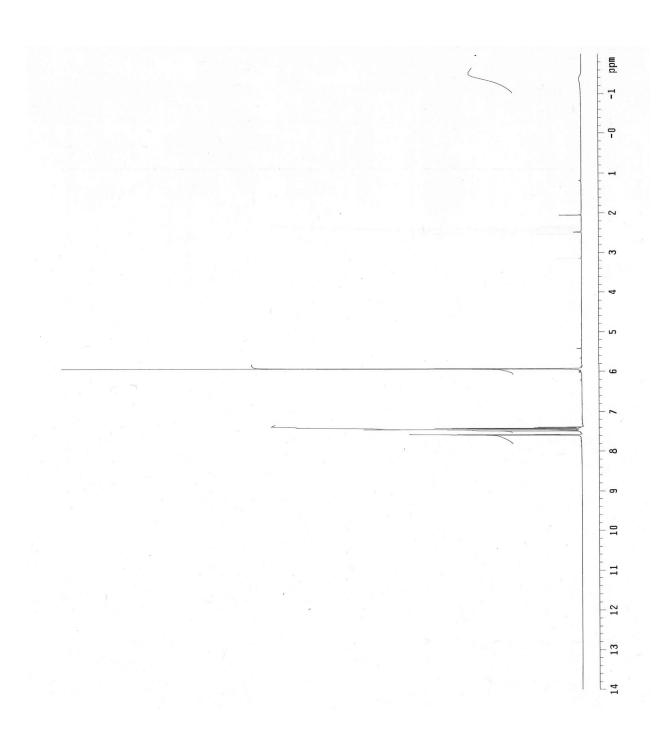




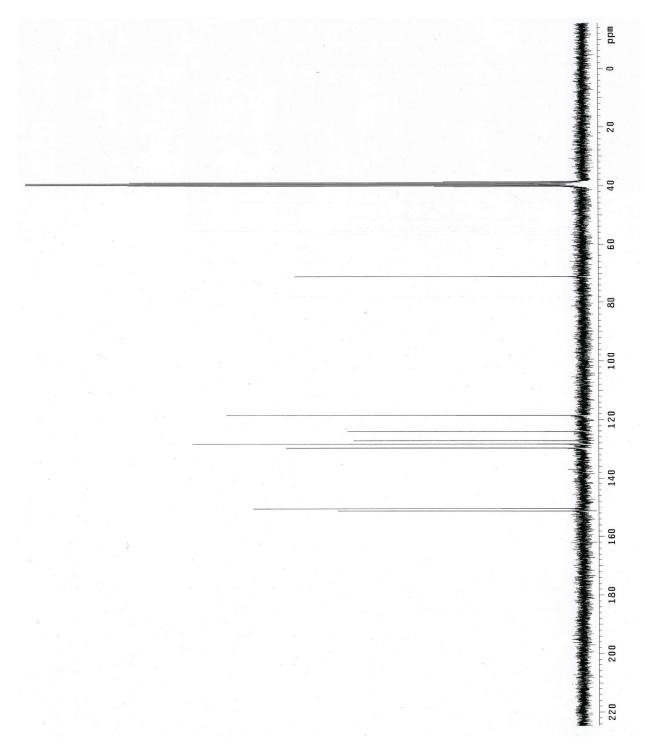
#### LC data for 1-((2,4-dichlorophenoxy)methyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6K}$



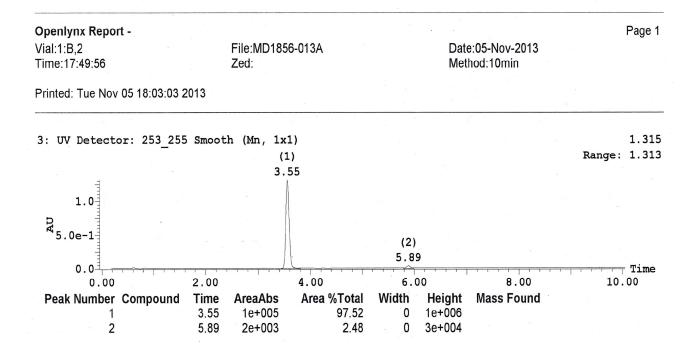
## $^1$ H NMR for 1-((2,4-dichlorophenoxy)methyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}\underline{K}$



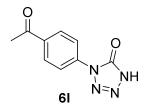
## $^{13}\mathrm{C}$ NMR for 1-((2,4-dichlorophenoxy)methyl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6</u>k

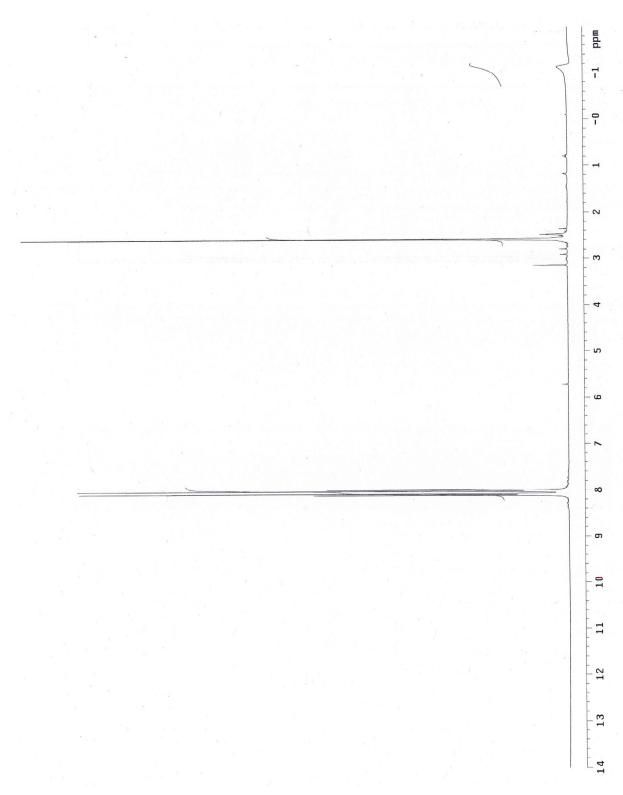


#### LC data for 1-(4-acetylphenyl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6</u>L

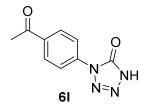


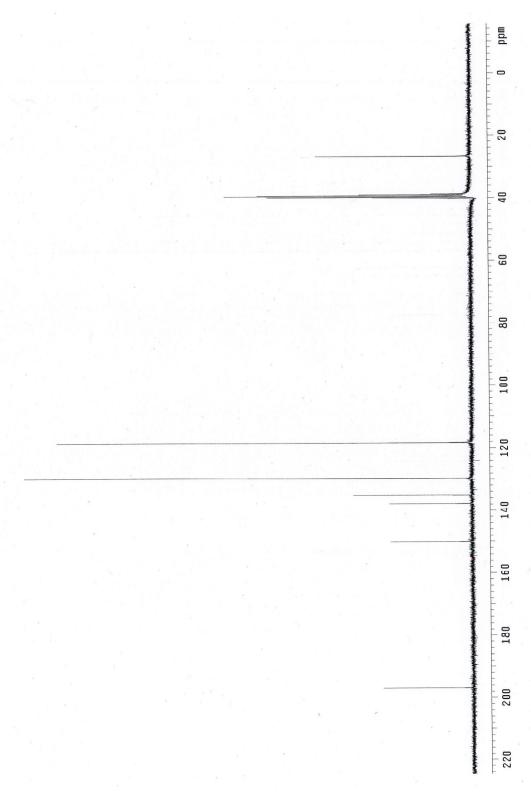
## $^{1}$ H NMR for 1-(4-acetylphenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}$ L



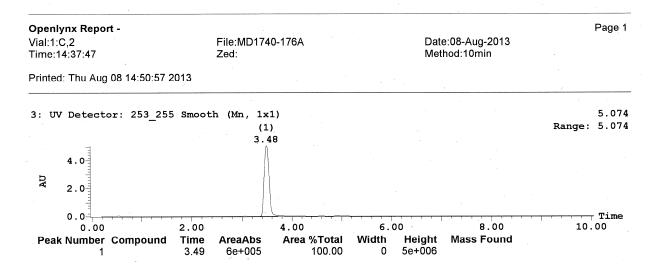


## $^{13}\mathrm{C}$ NMR for 1-(4-acetylphenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6L}$

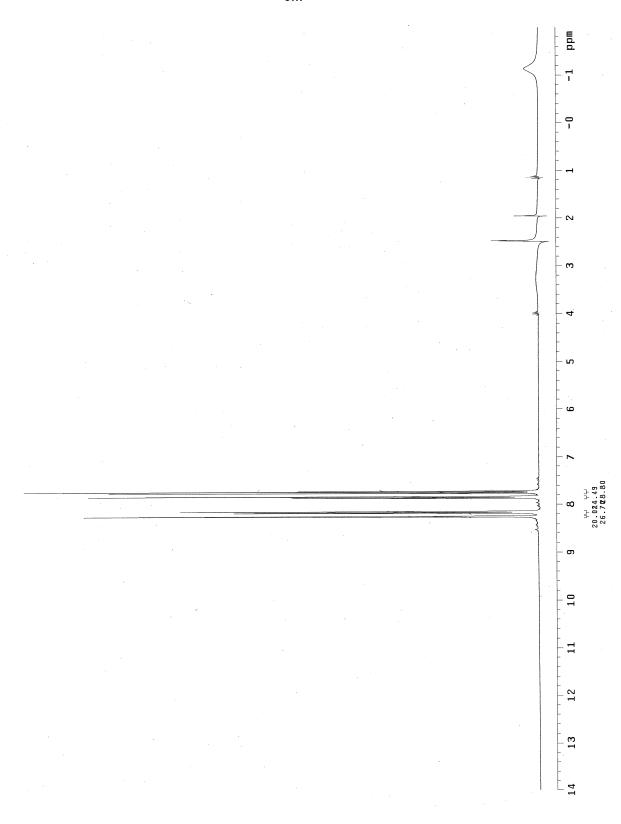


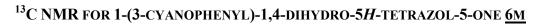


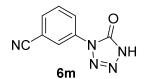
#### LC data for 1-(3-cyanophenyl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6m</u>

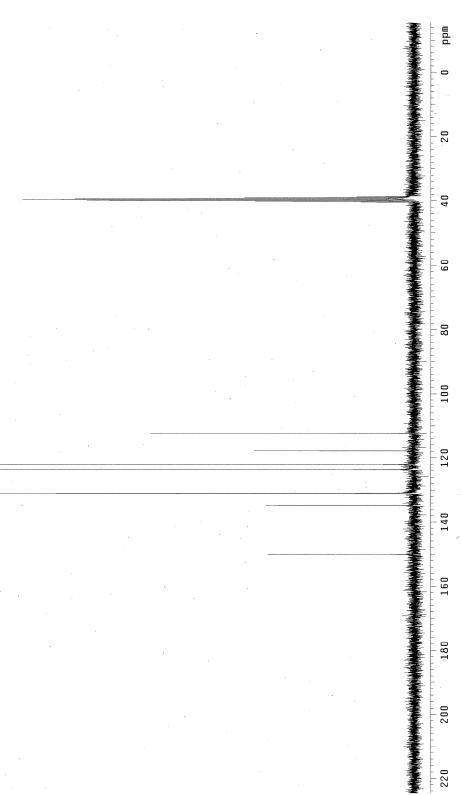


## $^{1}$ H NMR for 1-(3-cyanophenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}$ M



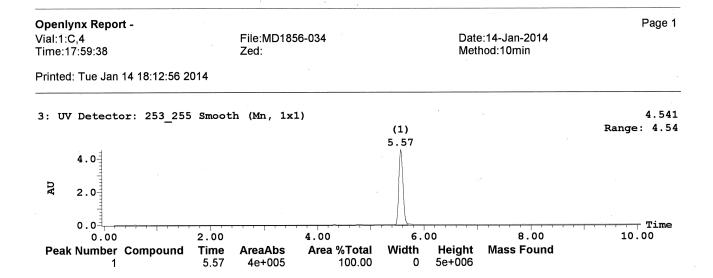




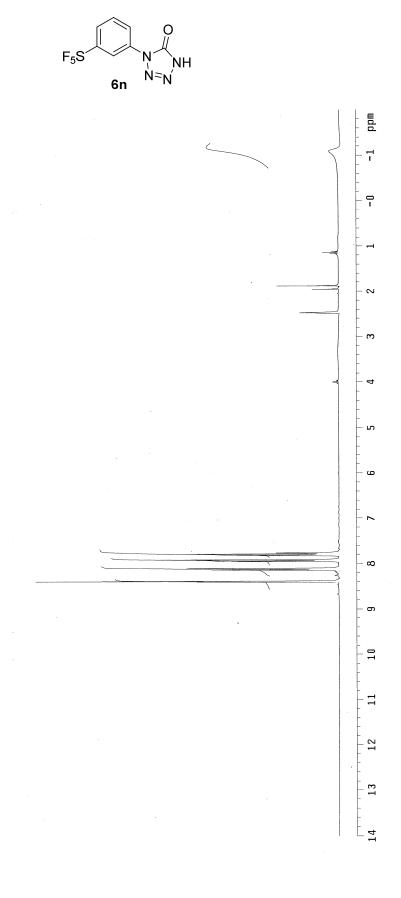


#### LC data for 1-(3-(pentafluorosulfanyl)phenyl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6</u>N

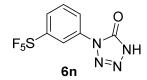
$$F_5$$
S  $N = N$ 

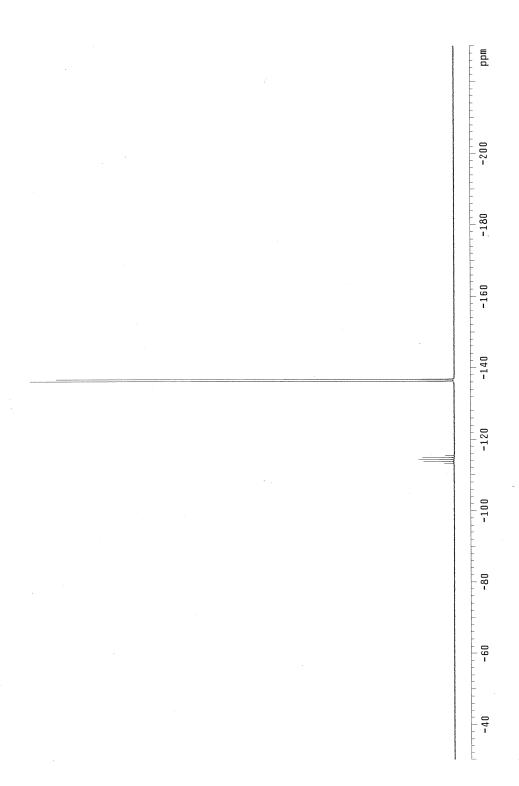


## $^1\mathrm{H}$ NMR for 1-(3-(pentafluorosulfanyl)phenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6\mathrm{N}}$

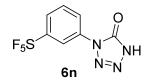


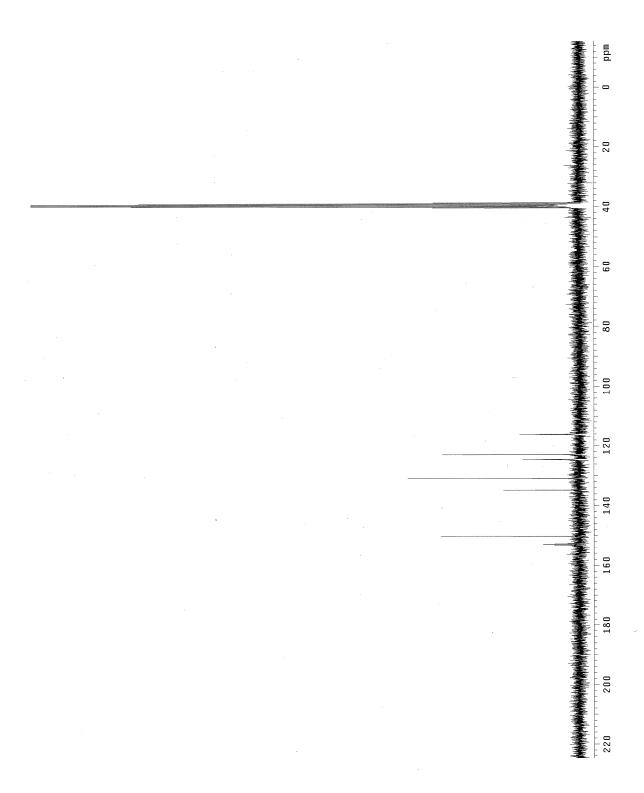
## $^{19}\mathrm{F}$ NMR for 1-(3-(pentafluorosulfanyl)phenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}\mathrm{N}$





## $^{13}\mathrm{C}$ NMR for 1-(3-(pentafluorosulfanyl)phenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}\mathrm{N}$





#### LC data for 1-(4-(thiomethyl)phenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{60}$

Openlynx Report -

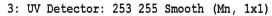
Page 1

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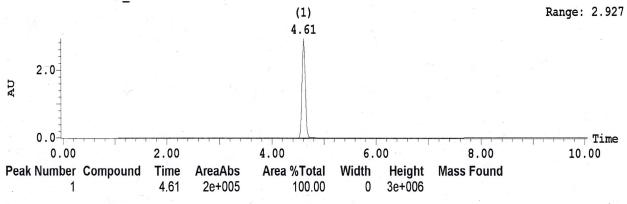
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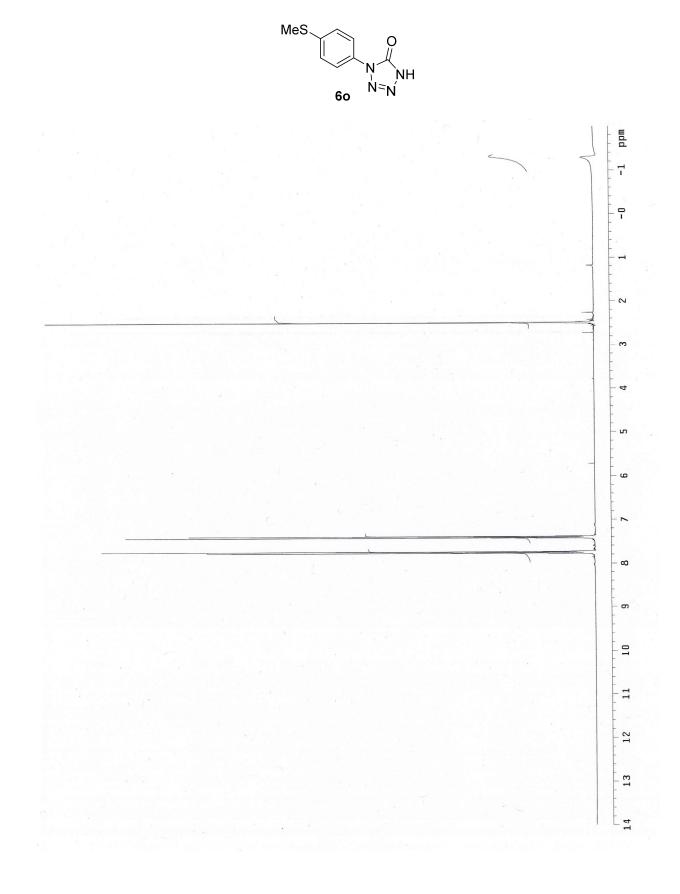
Printed: Mon Nov 04 15:14:55 2013



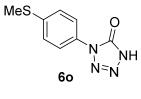
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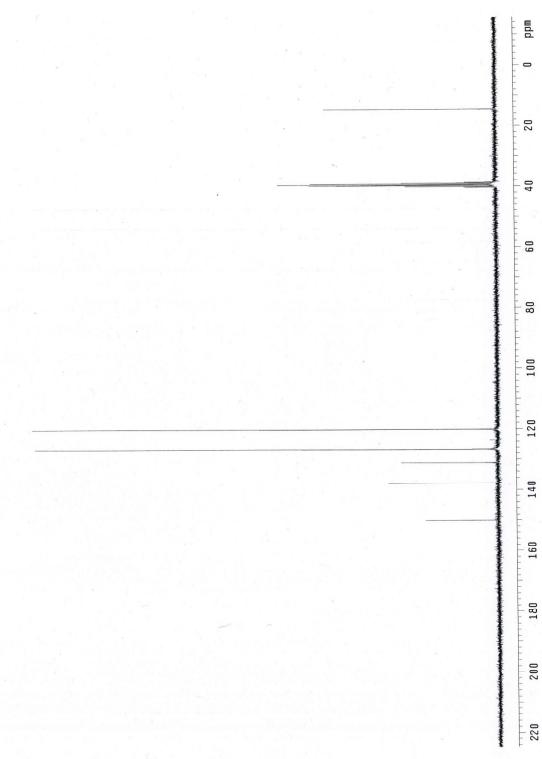


## $^{1}$ H NMR for 1-(4-(Thiomethyl)phenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{60}$

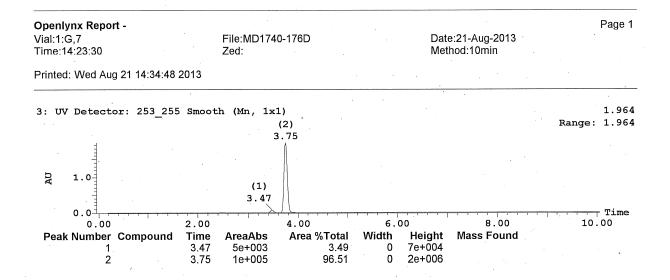


# $^{13}\mathrm{C}$ NMR for 1-(4-(Thiomethyl)Phenyl)-1,4-dihydro-5*H*-tetrazol-5-one $\underline{60}$

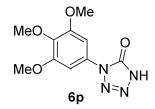


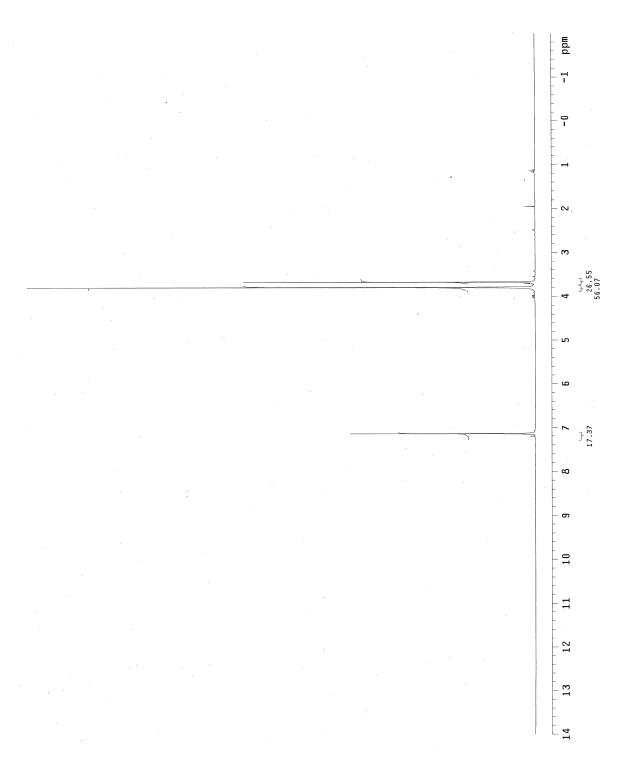


#### LC data for 1-(3,4,5-trimethoxtphenyl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6P</u>

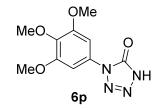


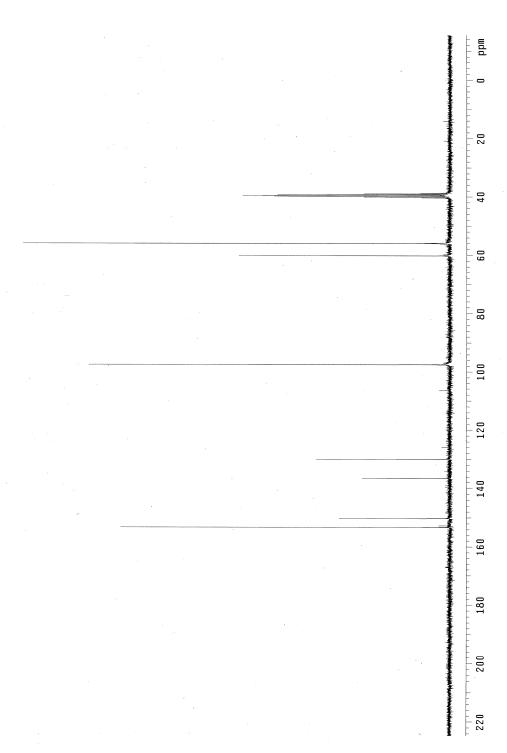
## $^1\mathrm{H}$ NMR for 1-(3,4,5-trimethoxtphenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6P}$



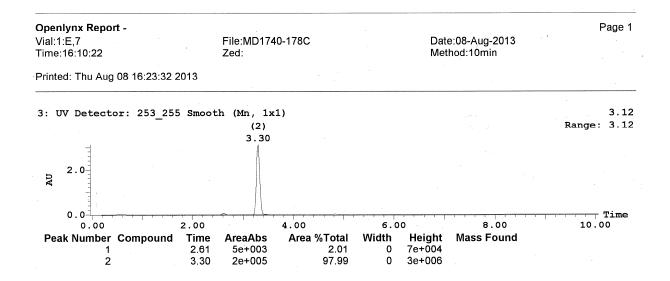


### $^{13}$ C NMR for 1-(3,4,5-trimethoxtphenyl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6P}$

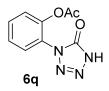


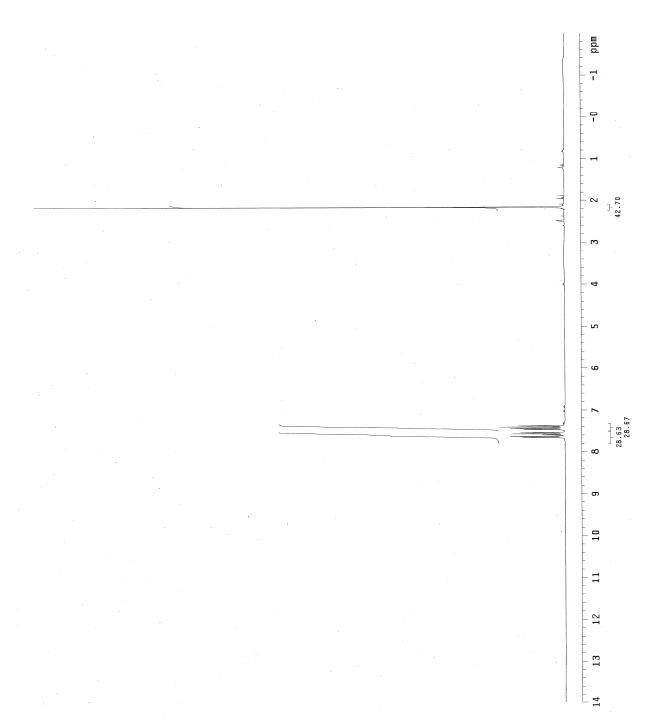


### LC data for 2-(5-0x0-4,5-dihydro-1H-tetrazol-1-yl)phenyl acetate $\underline{60}$

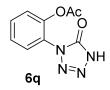


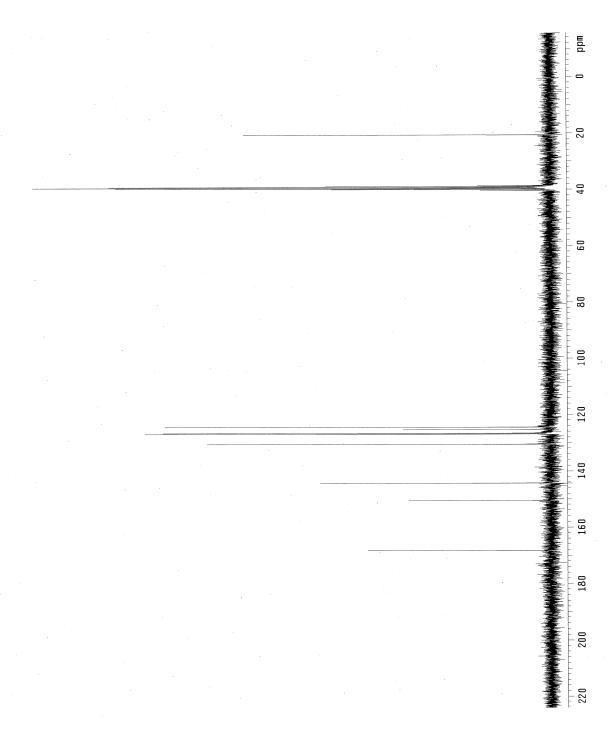
### $^1\mathrm{H}$ NMR for 2-(5-0x0-4,5-dihydro-1H-tetrazol-1-yl)phenyl acetate $\underline{6Q}$



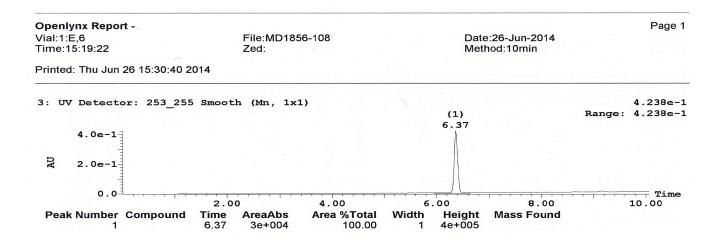


### $^{13}\mathrm{C}$ NMR for 2-(5-0x0-4,5-dihydro-1H-tetrazol-1-yl)phenyl acetate $\underline{60}$

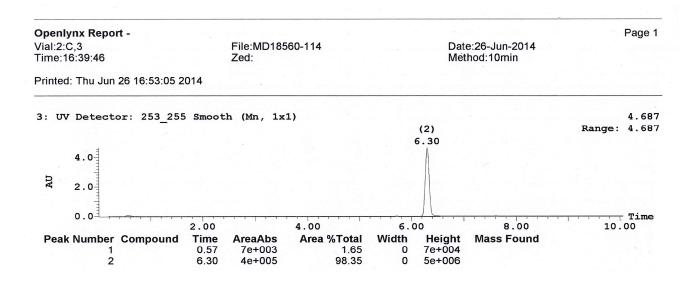




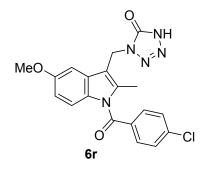
## LC data for 1-((1-(4-chlorobenzoyl)-5-methoxy-1H-indol-3-yl)methyl)-1,4-dihydro-5H-tetrazol-5-one 6R

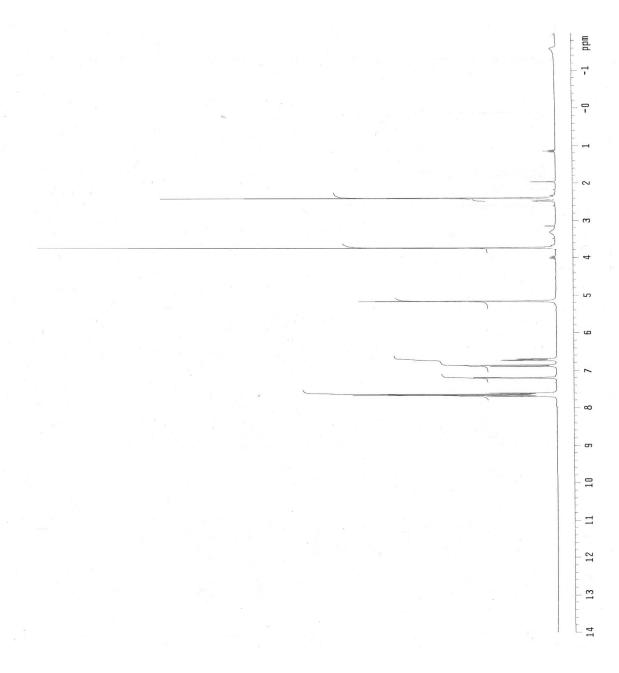


#### LC DATA FOR COMMERCIAL INDOMETHACIN (AS A COMPARISON TO COMPOUND 6R ABOVE)

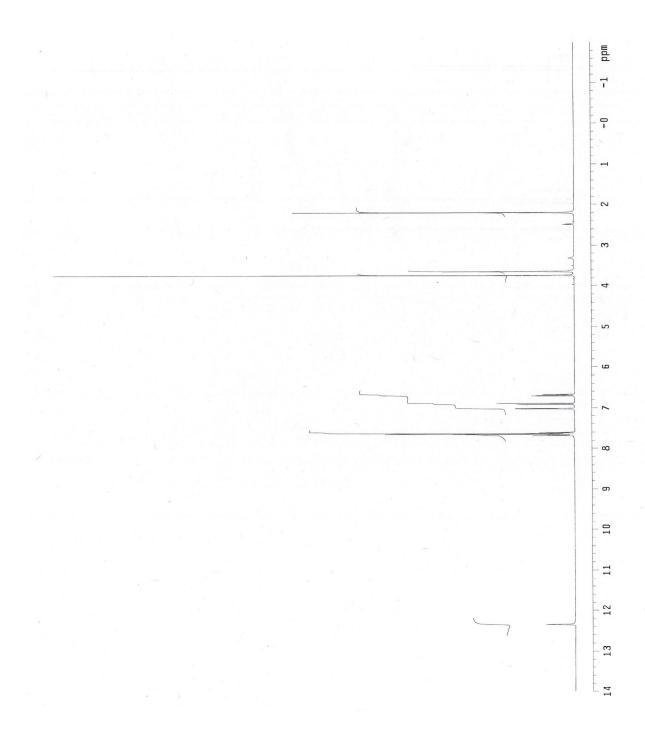


# $^{1}$ H NMR FOR 1-((1-(4-CHLOROBENZOYL)-5-METHOXY-1*H*-INDOL-3-YL)METHYL)-1,4-DIHYDRO-5*H*-TETRAZOL-5-ONE <u>6R</u>

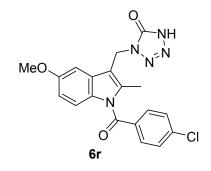


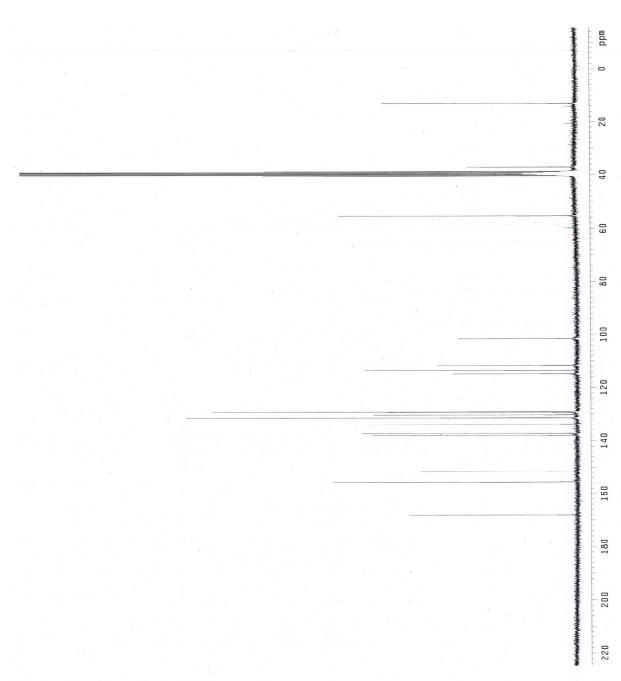


## $^1\mathrm{H}$ NMR for commercial Indomethacin (as a comparison to compound $\underline{6\mathrm{r}}$ above)

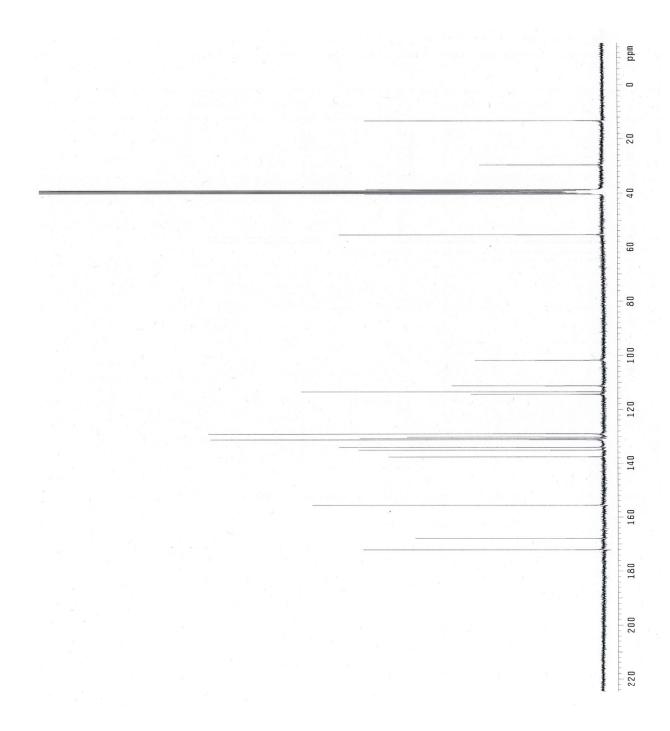


# $^{13}\mathrm{C}$ NMR for 1-((1-(4-chlorobenzoyl)-5-methoxy-1*H*-indol-3-yl)methyl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6r</u>

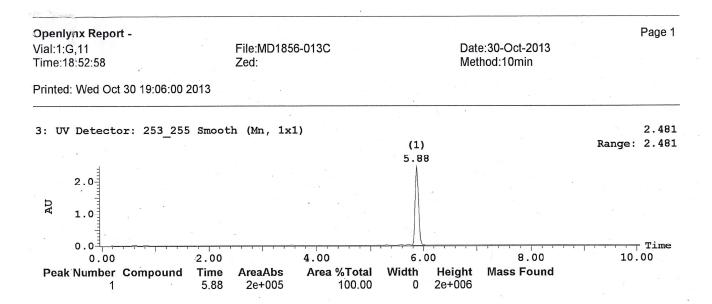




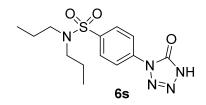
### $^{13}$ C NMR FOR COMMERCIAL INDOMETHACIN (AS A COMPARISON TO COMPOUND $\underline{6R}$ ABOVE)

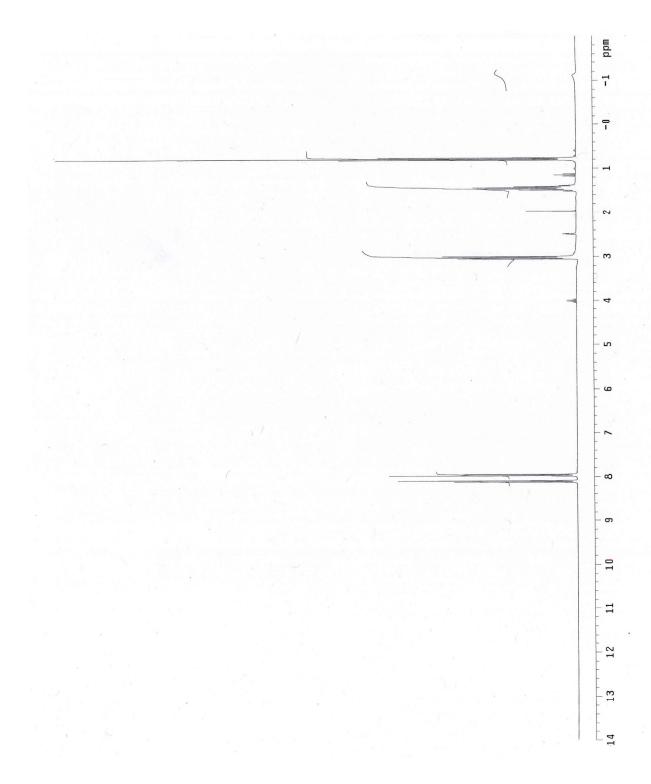


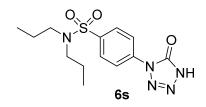
### LC data for 4-(5-0x0-4,5-dihydro-1H-tetrazol-1-yl)dipropylbenezenesulfonamide $\underline{68}$

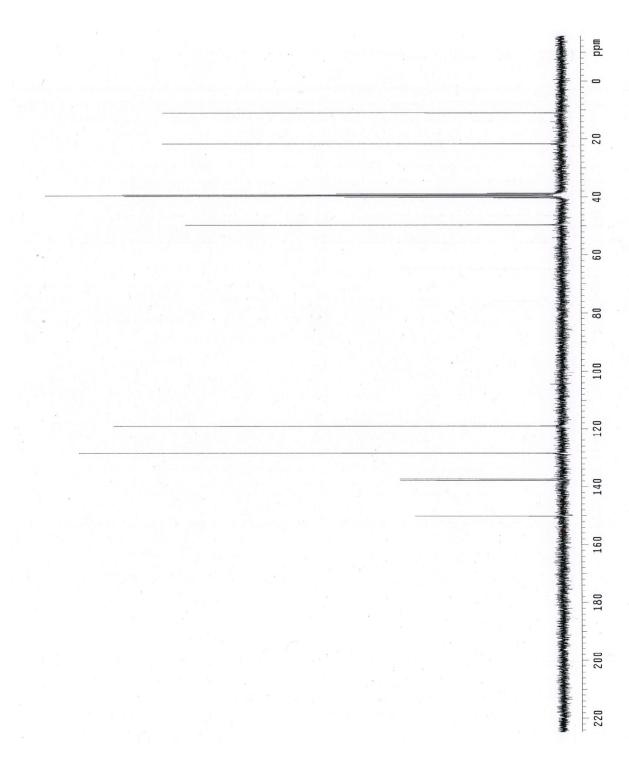


### $^1$ H NMR for 4-(5-0x0-4,5-dihydro-1H-tetrazol-1-yl)dipropylbenezenesulfonamide $\underline{68}$

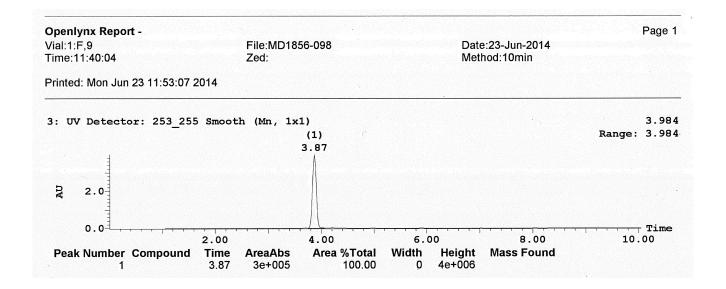




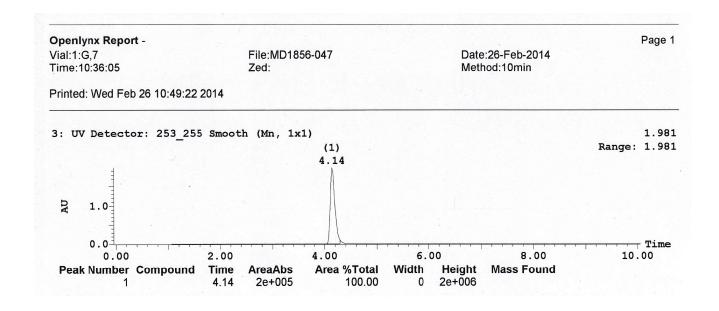




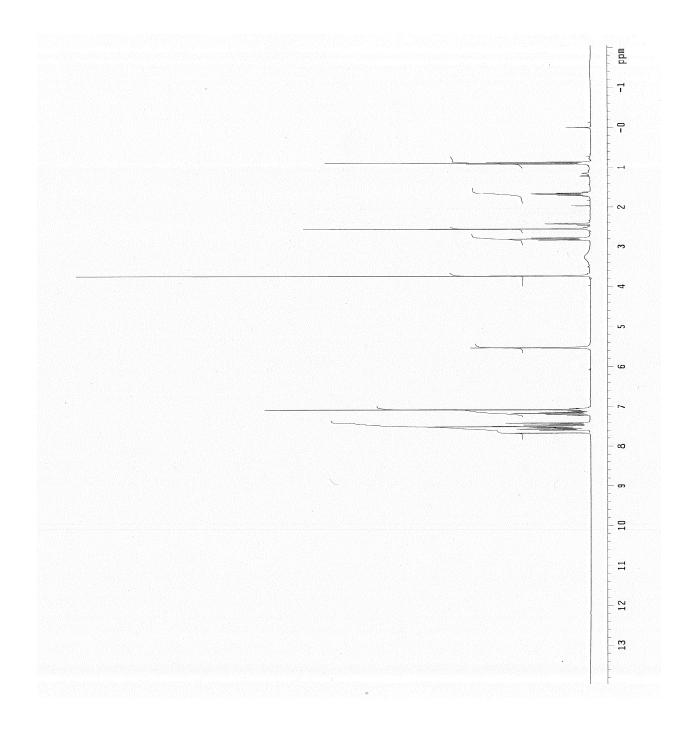
## LC data for 1-(4'-((1,7-dimethyl-2'-propyl-1H,3'H-[2,5'-dibenzo[D]imidazol]-3'-yl)methyl-[1,1'-biphenyl-2-yl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6}\underline{T}$



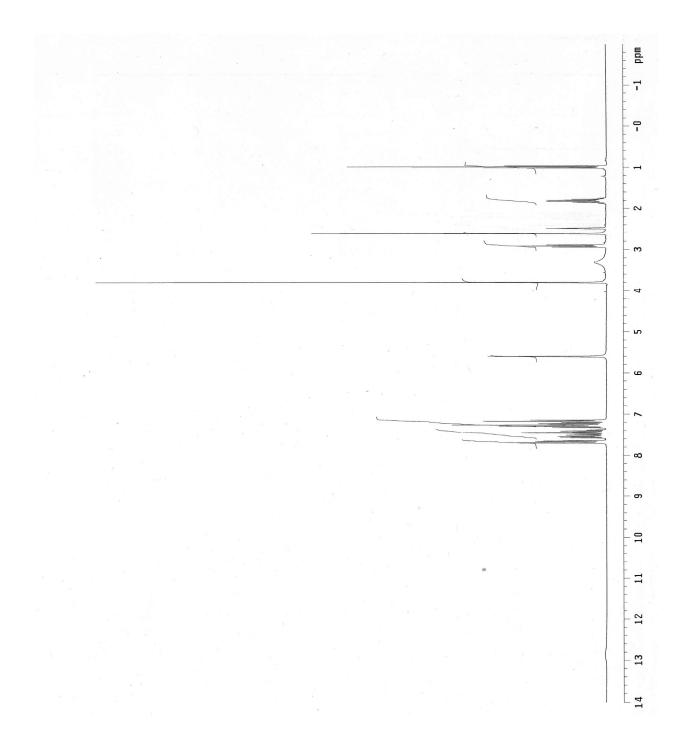
#### LC DATA FOR COMMERCIAL TELMISARTAN (AS A COMPARISON TO COMPOUND 6T ABOVE)



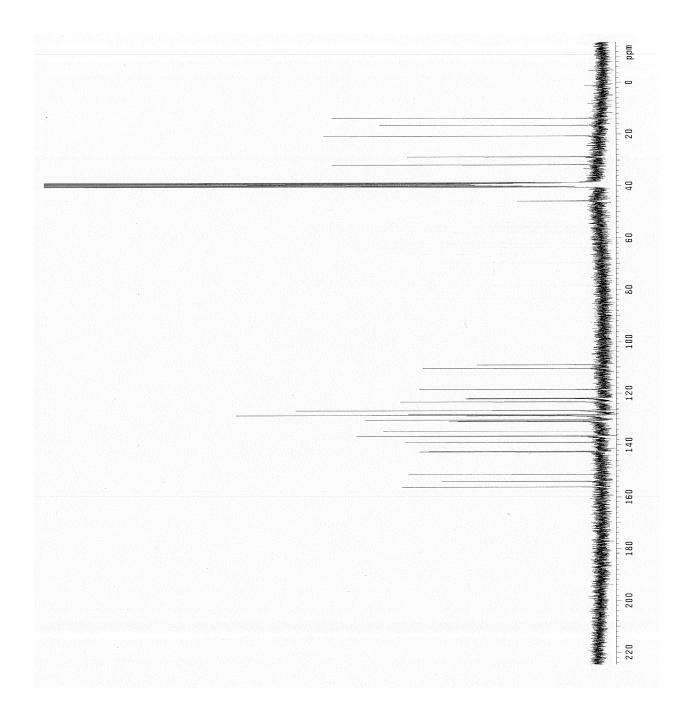
 $^{1}$ H NMR FOR 1-(4'-((1,7-dimethyl-2'-propyl-1H,3'H-[2,5'-dibenzo[D]imidazol]-3'-Yl)methyl-[1,1'-biphenyl-2-yl)-1,4-dihydro-5H-tetrazol-5-one  $\underline{6}\underline{T}$ 



### $^1$ H NMR FOR COMMERCIAL TELMISARTAN (AS A COMPARISON TO COMPOUND $\underline{6T}$ ABOVE)

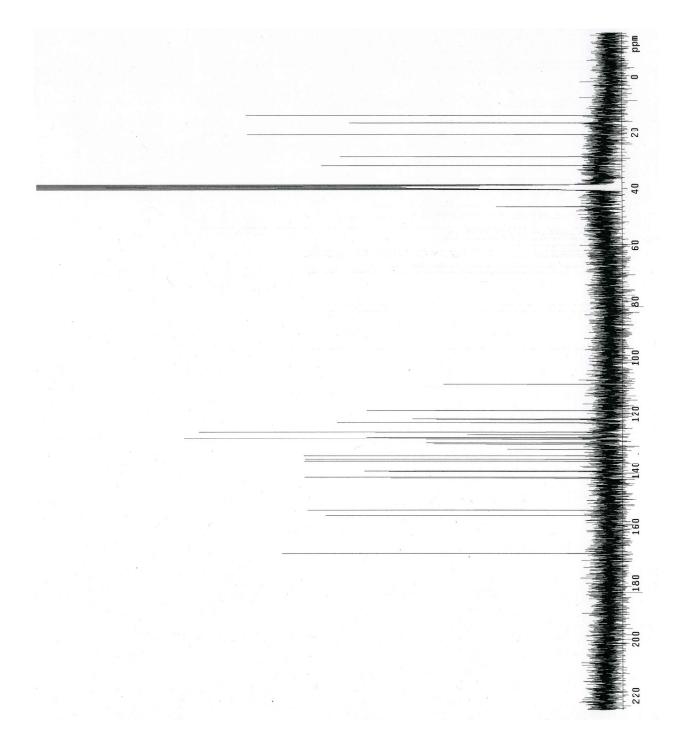


 $^{13}\mathrm{C}$  NMR for 1-(4'-((1,7-dimethyl-2'-propyl-1*H*,3'*H*-[2,5'-dibenzo[*d*]imidazol]-3'-Yl)methyl-[1,1'-biphenyl-2-yl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6T</u>

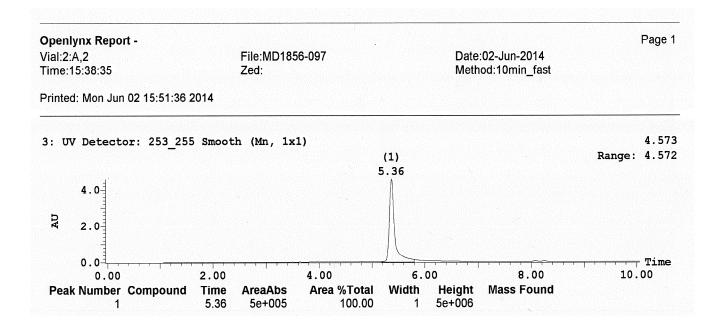


## $^{13}\mathrm{C}$ NMR for commercial Telmisartan (as a comparison to compound $\underline{6T}$ above)

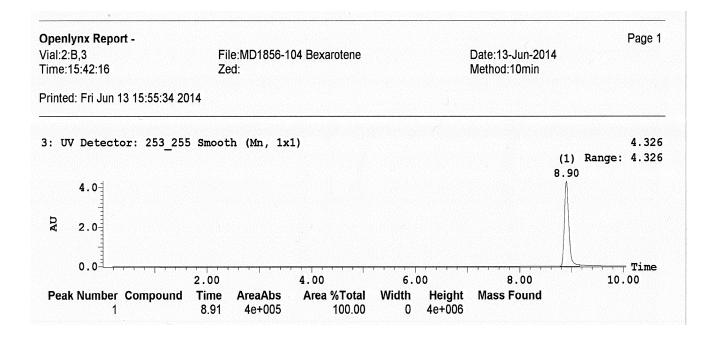




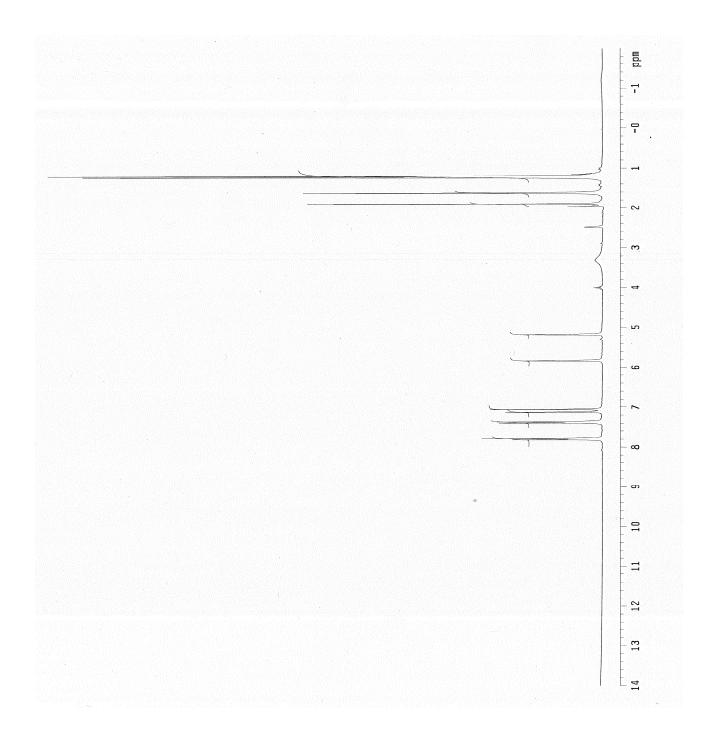
## LC data for 1-(4-(1-(3,5,5,8,8-pentamethyl-5,6,7,8-tetrahydronaphthalen-2-YL)VINYL)PHENYL)-1,4-DIHYDRO-5H-TETRAZOL-5-ONE $\underline{6U}$



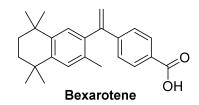
#### LC DATA FOR COMMERCIAL BEXAROTENE (AS A COMPARISON TO COMPOUND <u>6U</u> ABOVE)

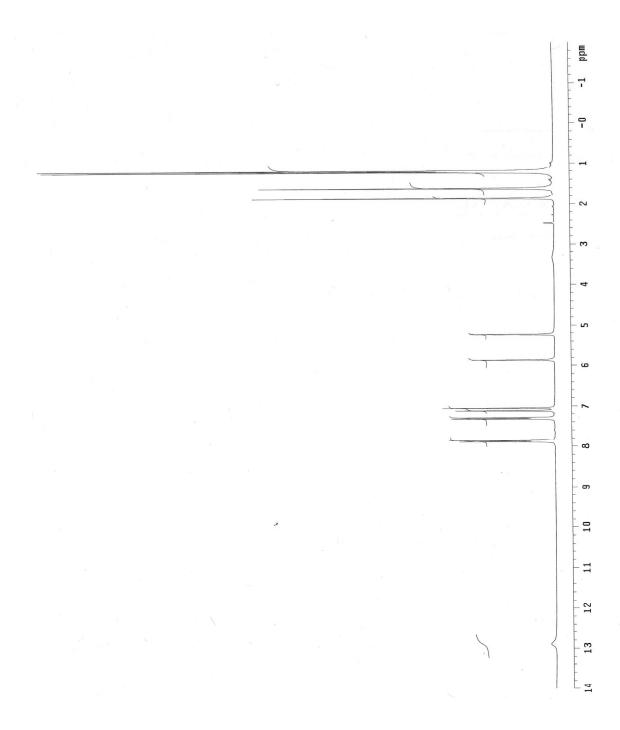


 $^{1}$ H NMR for 1-(4-(1-(3,5,5,8,8-pentamethyl-5,6,7,8-tetrahydronaphthalen-2-yl)vinyl)phenyl)-1,4-dihydro-5*h*-tetrazol-5-one  $\underline{6}\underline{0}$ 



### $^1$ H NMR FOR COMMERCIAL BEXAROTENE (AS A COMPARISON TO COMPOUND $\underline{6}\underline{0}$ ABOVE)

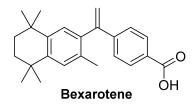


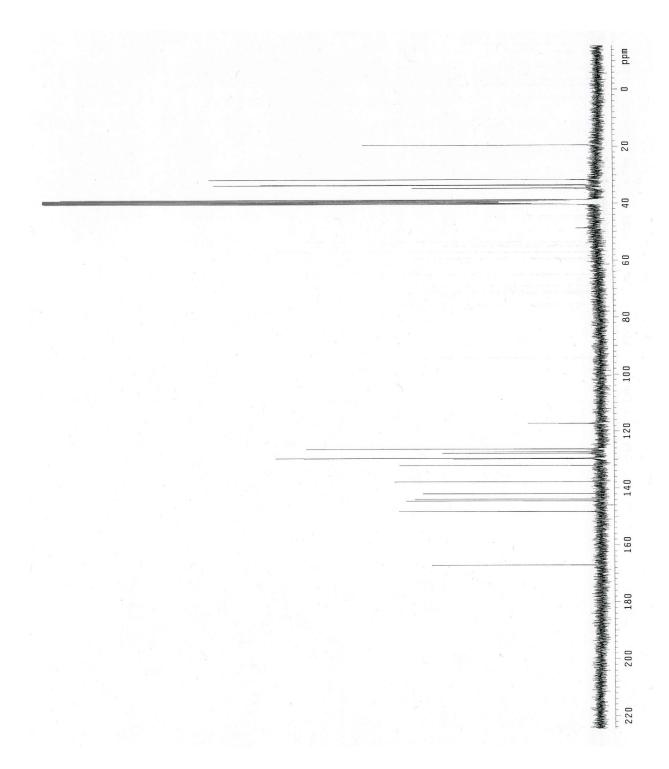


 $^{13}\mathrm{C}$  NMR for 1-(4-(1-(3,5,5,8,8-pentamethyl-5,6,7,8-tetrahydronaphthalen-2-yl)vinyl)phenyl)-1,4-dihydro-5*h*-tetrazol-5-one <u>6u</u>

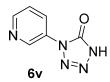


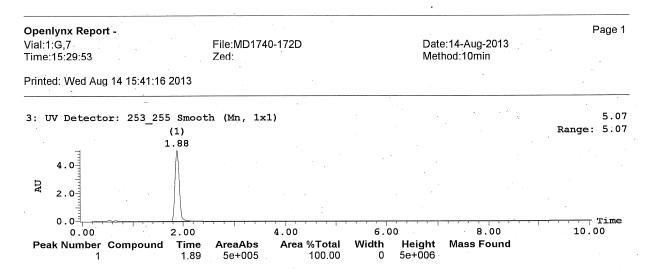
## $^{13}\mathrm{C}$ NMR for commercial Bexarotene (as a comparison to compound $\underline{6}\underline{\mathrm{U}}$ above)

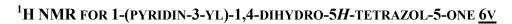


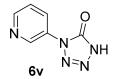


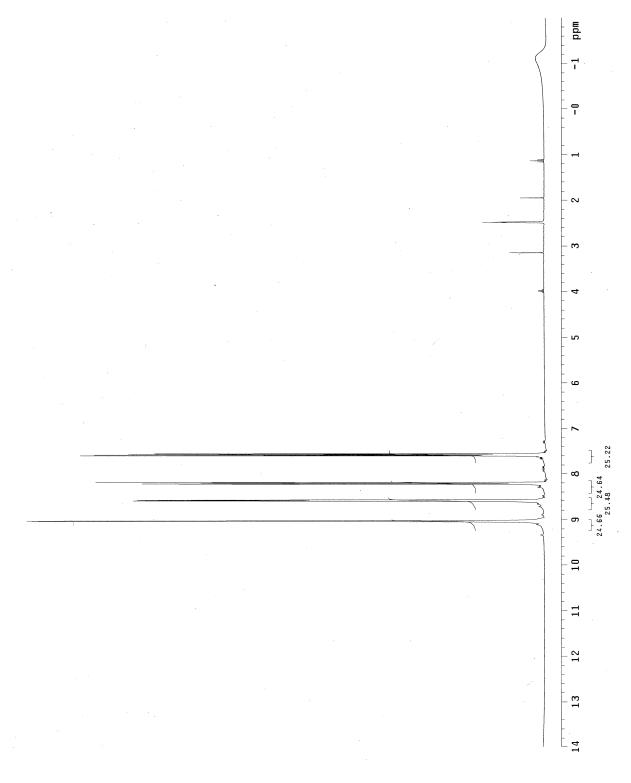
### LC data for 1-(pyridin-3-yl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6v</u>

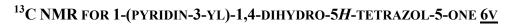


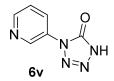


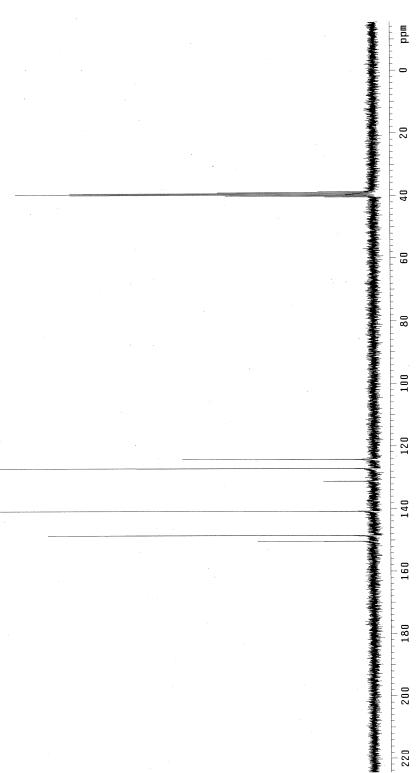




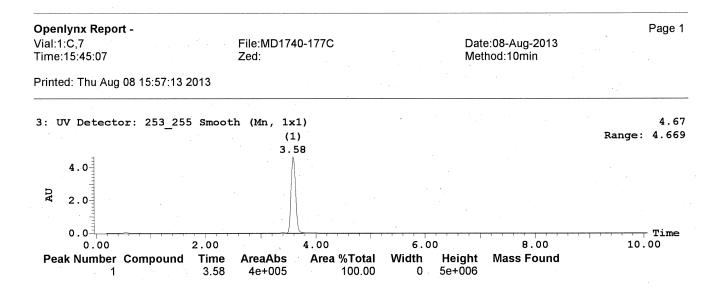




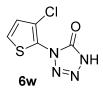


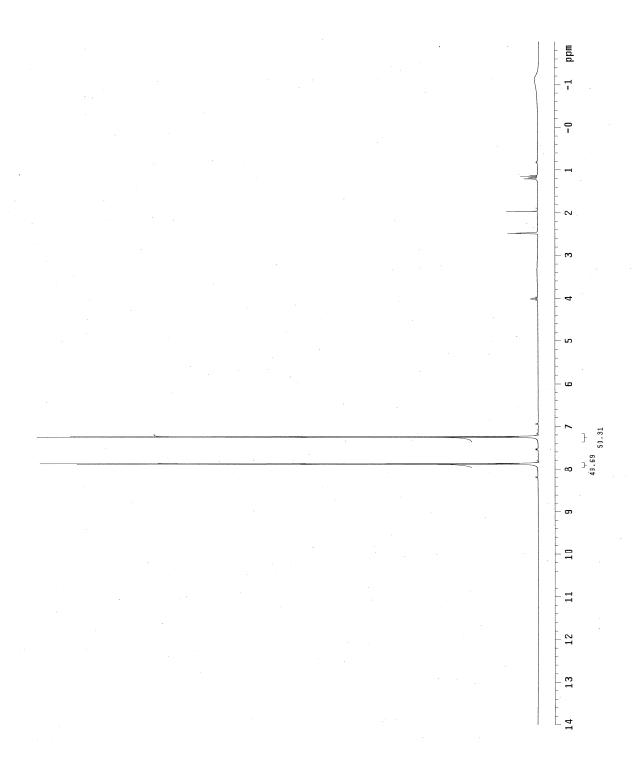


### LC data for 1-(3-chlorothiophen-2-yl)-1,4-dihydro-5*H*-tetrazol-5-one <u>6w</u>

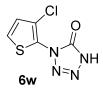


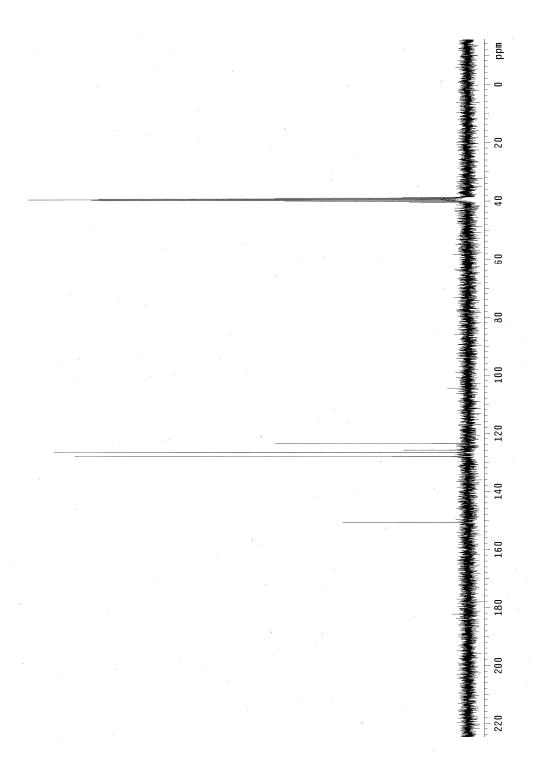
## $^1\mathrm{H}$ NMR for 1-(3-chlorothiophen-2-yl)-1,4-dihydro-5H-tetrazol-5-one $\underline{6\mathrm{W}}$



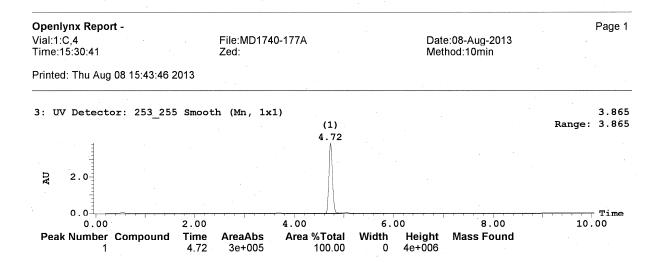


### $^{13}\mathrm{C}$ NMR for 1-(3-chlorothiophen-2-yl)-1,4-dihydro-5*H*-tetrazol-5-one $\underline{6\mathrm{W}}$

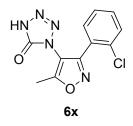


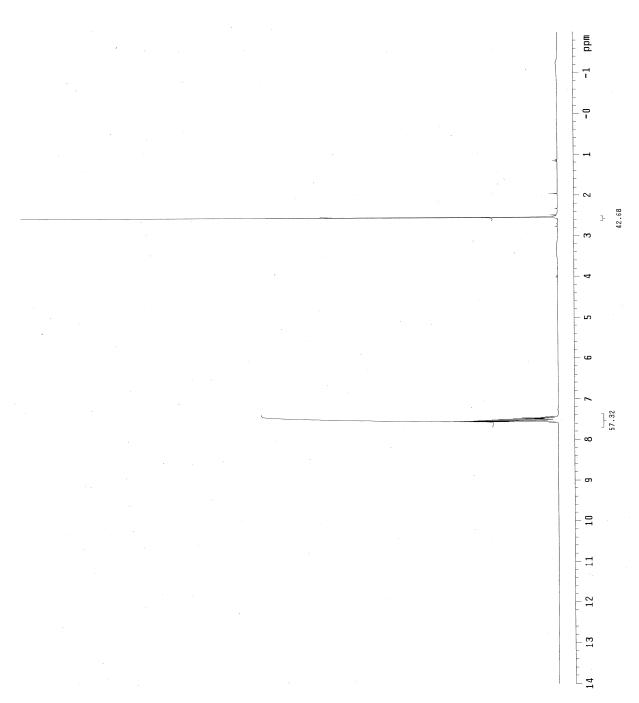


## LC data for 1-(3-(2-chlorophenyl)5-methylisoxazol-4-yl)-1,4-dihydro-5H-tetrazol-5-one 6x

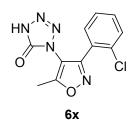


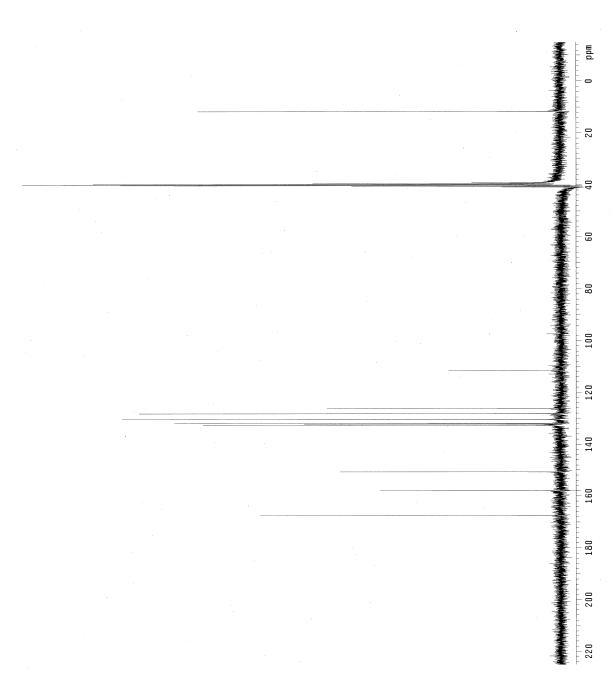
 $^1\mathrm{H}$  NMR for 1-(3-(2-chlorophenyl)5-methylisoxazol-4-yl)-1,4-dihydro-5*H*-tetrazol-5-one  $\underline{6x}$ 





 $^{13}\mathrm{C}$  NMR for 1-(3-(2-chlorophenyl)5-methylisoxazol-4-yl)-1,4-dihydro-5*H*-tetrazol-5-one  $\underline{6x}$ 





### LC data for methyl (E)-3-(5-0x0-4,5-dihydro-1H-tetrazol-1-yl)acrylate $\underline{6}\underline{Y}$

$$MeO \longrightarrow N \longrightarrow NH$$

Openlynx Report -Page 1 Vial:2:F,1 Time:15:28:10 File:MD1856-017B Date:04-Nov-2013 Zed: Method:10min Printed: Mon Nov 04 15:41:14 2013 3: UV Detector: 253\_255 Smooth (Mn, 1x1) 3.329 Range: 3.329 2.60 2.0 AU 0.0 Time 2.00 4.00 6.00 8.00 10.00 0.00

Area %Total

100.00

Width

0

Height

3e+006

Mass Found

Peak Number Compound

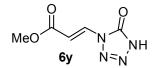
Time

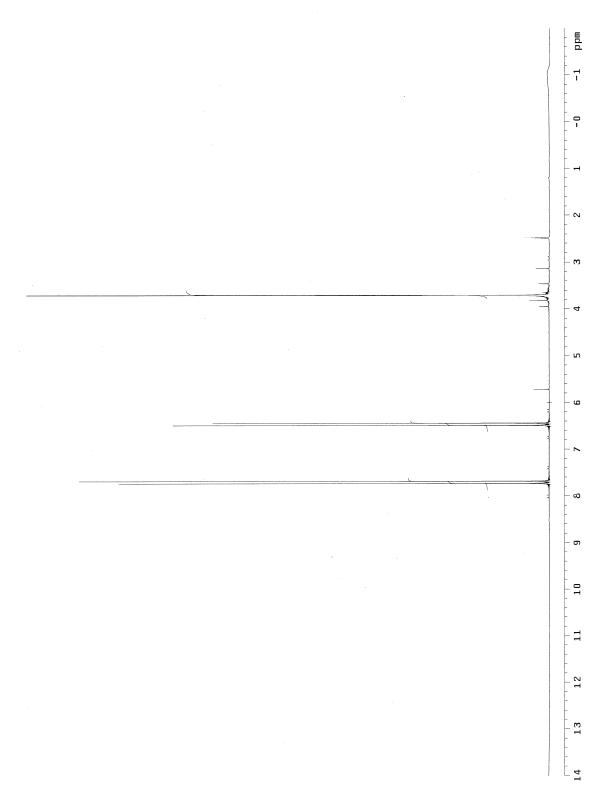
2.60

AreaAbs

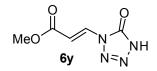
2e+005

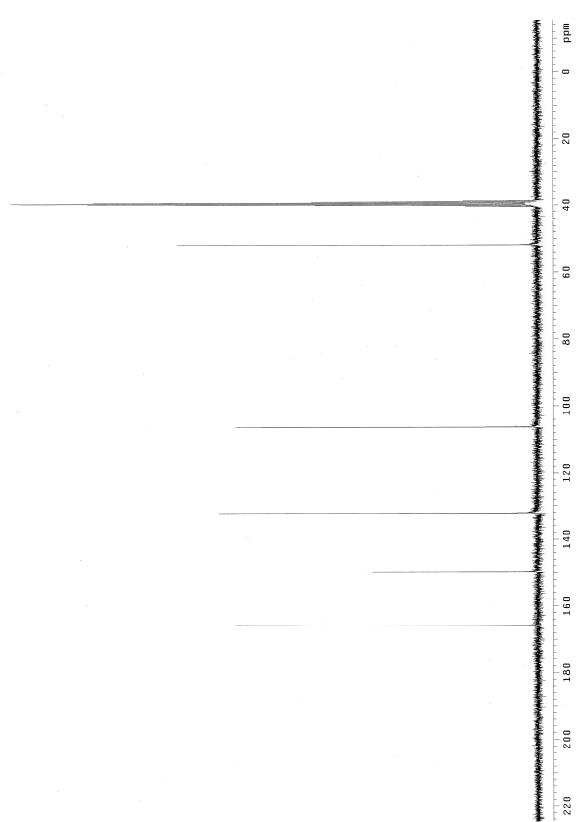
### $^1$ H NMR for methyl (*E*)-3-(5-0x0-4,5-dihydro-1*H*-tetrazol-1-yl)acrylate $\underline{6}$ W





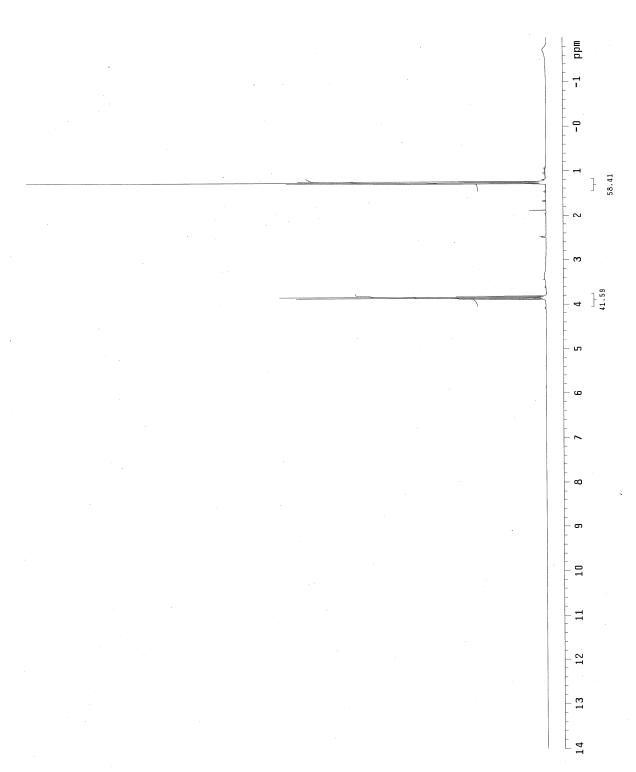
## $^{13}\mathrm{C}$ NMR for methyl (E)-3-(5-0x0-4,5-dihydro-1H-tetrazol-1-yl)acrylate $\underline{6\mathrm{W}}$





### $^{1}$ H NMR for 1-ethyl-1,4-dihydro-5H-tetrazol-5-one $\underline{6z}$





## $^{13}\mathrm{C}$ NMR for 1-ethyl-1,4-dihydro-5*H*-tetrazol-5-one $\underline{6Z}$



