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Supporting Information

Redox Reaction between Benzyl Azides and Aryl aAzides: Concerted Synthesis of Aryl Nitriles and Anilines

Yongjin Kim, Young Ho Rhee* and Jaiwook Park*

Department of Chemistry, POSTECH (Pohang University of Science and Technology), 77

Cheongam-Ro Nam-Gu Pohang, Gyeongbuk, Korea 37673

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1. General Information

Air-sensitive manipulations were carried out with standard Schlenk techniques under argon atmosphere. Commercial chemicals used without further purification. Flash column chromatography was carried out on silica gel (230-400 mesh) as the stationary phase. ¹H and ¹³C NMR spectra were recorded with Bruker AVANCE III 300MHZ FT-NMR spectrometer and chemical shift are given in δ ppm. ¹H NMR spectra were referenced to tetramethylsilane (TMS, 0 ppm). ¹³C NMR spectra were referenced to CDCl₃ (77.23 ppm) as an internal standard. Infrared spectra were recorded on a Shimadzu IR-470 spectrometer with NaCl pellet. Mass spectral data were obtained from the Korea Basic Science Institute (Daegu) on a Jeol JMS 700 high resolution mass spectrometer. Ruthinum complex **1**^[1] and **11**^[2] were synthesized according to the literature procedure.

2. Synthesis of Benzyl Azides

2-a. Synthesis of benzyl azides (2a~2m)

A solution of benzyl bromide (or benzyl chloride) (3.0 mmol) and sodium azide (2.0 equiv, 6.0 mmol) in DMF (20 mL) was stirred for 12 h at room temperature. After completion of the reaction, the mixture was dilluted in H₂O (30 mL) and extracted with diethyl ether (3 x 20 mL). The oragnic layer was washed with water (2 x 50 mL) and brine (50 mL), dried over anhydrous sodium sulfate, concentrated under reduced pressure. The residue was purified by column chromatography on silica gel to afford the benzyl azides.

MeO N₃ 1-(azidomethyl)-4-methoxybenzene (2a)^[3]

Yield: 99%; Colorless liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.25-7.22 (m, 2H), 6.92-6.89 (m, 2H), 4.26 (s, 2H), 3.81 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ = 159.8, 129.9, 127.6, 114.4, 55.5, 54.6.

N₃ 1-(azidomethyl)-4-methylbenzene (2b)^[3]

Yield: 93%; Colorless liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.23-7.16 (m, 4H), 4.27 (s, 2H), 2.35 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ = 138.3, 132.5, 129.7, 128.5, 54.8, 21.3.



1-(azidomethyl)-4-(trifluoromethyl)benzene (2c)^[4]

Yield: 82%; Colorless liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.66-7.63 (m, 2H), 7.64-7.43 (m, 2H), 4.43 (s, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 139.6, 128.5, 126.1, 126.05, 126.0, 125.95, 122.4, 54.3.

NC NC 4-(azidomethyl)benzonitrile (2d)^[3]

Yield: 89%; Colorless liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.70-7.67 (m, 2H), 7.45-7.43 (m, 2H), 4.45 (s, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 141.0, 132.8, 128.7, 118.6, 112.4, 54.2.

N₃ 1-(azidomethyl)-2-methylbenzene (2e)^[3]

Yield: 96%; Colorless liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.28-7.17 (m, 5H), 4.33 (s, 2H), 2.36 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ = 137.0, 133.6, 130.9, 129.5, 128.8, 126.4, 53.2, 19.2.

N₃ 1-(azidomethyl)-3-methylbenzene (2f)^[3]

Yield: 95%; Colorless liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.29-7.24 (m, 1H), 7.16-7.09 (m, 3H), 4.29 (s, 2H), 2.36 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ = 138.8, 135.5, 129.3, 129.1, 128.9, 125.5, 55.0, 21.6.

(azidomethyl)benzene (2g)^[3]

Yield: 95%; Pale yellow liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.40-7.27 (m, 5H), 4.30 (s, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 135.6, 129.0, 128.5, 128.4, 54.9.



Yield: 99%; Colorless liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.38-7.33 (m, 2H), 7.27-7.23 (m, 2H), 4.31 (s, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 134.4, 134.1, 129.7, 129.2, 54.2.



Yield: 93%; Colorless liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.53-7.49 (m, 2H), 7.21-7.19 (m, 2H), 4.30 (s, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 134.6, 132.2, 130.0, 122.6, 54.3.



1-(azidomethyl)-4-tert-butylbenzene (2j)^[5]

Yield: 95%; Colorless liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.42-7.39 (m, 2H), 7.26-7.24 (m, 2H), 4.30 (s, 2H), 1.32 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ = 151.6, 132.6, 128.2, 126.0, 54.8, 34.8, 31.5.

2-(azidomethyl)naphthalene (2k)^[6]

Yield: 99%; White solid; ¹H NMR (300 MHz, CDCl₃) δ = 7.87-7.81 (m, 3H), 7.76 (s, 1H), 7.52-7.46 (m, 2H), 7.43-7.39 (m, 1H), 4.48 (s, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 133.5, 133.3, 133.0, 129.0, 128.1, 128.0, 127.4, 126.7, 126.5, 126.1, 55.2.

MeO₂C

Methyl 4-(azidomethyl)benzoate (2l)^[3]

Yield: 94%; Pale yellow liquid; ¹H NMR (300 MHz, CDCl₃) δ = 8.07-8.04 (m, 2H), 7.41-7.38 (m, 2H), 4.42 (s, 2H), 3.92 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ = 166.8, 140.6, 130.3, 128.1, 54.5, 52.4.

N₃ 1-(azidomethyl)-4-vinylbenzene (2m)^[7]

Yield: 90%; Yellow liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.44-7.41 (m, 2H), 7.29-7.25 (m, 2H), 6.72 (dd, *J* = 10.9 Hz, *J* = 17.6 Hz, 1H), 5.77 (d, *J* = 17.7 Hz, 1H), 5.28 (d, *J* = 10.9 Hz, 1H), 4.32 (s, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 137.9, 136.4, 135.0, 128.7, 126.9, 114.7, 54.8.

2-b. Synthesis of 4-(azidomethyl)benzaldehyde (2n)

4-(azidomethyl)benzaldehyde was synthesized accroding to the literature procedure.^[8] 4-(bromomethyl)benzonitrile (3.0 mmol) in dry toluene (25 mL), diisobutylaluminum hydride (1.5 equiv, 7.5 mmol, 1.0 M in dichloromethane) was added dropwise at 0 °C then stirred for 3 h at room temperature. After consumption of starting material, 10% HCl (25 mL) was added dropwise at 0 °C and stirred for 1 h at 0 °C. The reaction mixture was extracted with diethyl ether (3 x 30 mL), wash with H₂O (2 x 50 mL) and brine (50 mL), dried over anhydrous sodium sulfate, concentrated under reduced pressure to afford 4-(bromomethyl)benzaldehye. The crude mixture was directly used for next step without purification. A solution of 4-(bromomethyl)benzaldehyde and sodium azide (2.0 equiv, 6.0 mmol) in DMF (20 mL) was stirred for 12 h at room temperature. After completion of the reaction, the mixture was dilluted in H₂O (30 mL) and extracted with diethyl ether (3 x 20 mL). The oragnic layer was washed with water (2 x 50 mL) and brine (50 mL), dried over anhydrous sodium sulfate, concentrated under reduced pressure. The residue was purified by column chromatography on silica gel to afford 4-(azidomethyl)benzaldehyde 2n

0 4-(azidomethyl)benzaldehyde (2n)^[8]

Yield: 98% (2 steps); Pale yellow liquid; ¹H NMR (300 MHz, CDCl₃) δ = 10.0 (s, 1H), 7.92-7.89 (m, 2H), 7.51-7.48 (m, 2H), 4.46 (s, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 191.8, 142.3, 136.4, 130.4, 128.7, 54.5.

2-c. Synthesis of (4-(azidomethyl)phenoxy)(tert-butyl)dimethylsilane (20)

4-(azidomethyl)phenol was synthesized accroding to the literature procedure.^[9] A mixture of 4-hydroxy benzylalcohol (5.0 mmol), sodium azide (1.2 equiv, 10.0 mmol) and triphenylphosphine (1.0 equiv, 5.0 mmol) in a mixture of CCl₄ and DMF (1:4, 10 mL) was stirred at 90 °C reflux for 5 h. After the completion of the reaction, the reaction mixture extracted with diethyl ether (3 x 20 mL). The oragnic layer was washed with water (2 x 50 mL) and brine (50 mL), dried over anhydrous sodium sulfate, concentrated under reduced pressure. The residue was purified by column chromatography on silica gel (1:5 EtOAc:n-hexane) to afford the 4-(azidomethyl)phenol as pale yellow liquid (65%). To a solution of 4- (azidomethyl)phenol (2.0 mmol) and imidazole (2.5 equiv, 5.0 mmol) in DMF (10 mL), tert-butyldimethylsilyl chloride (TBSCl) (1.2 equiv, 2.4 mmol) was added and stirred at room temeparture for 12 h. After the completion of the reaction, the reaction mixture was quenched with H₂O, extracted with diethyl ether (3 x 20 mL). The oragnic layer was washed with water (2 x 50 mL) and brine (50 mL), dried over anhydrous sodium sulfate, concentrated under reduced pressure. The residue was purified by column chromatography on silica gel (n-hexane) to afford (4-(azidomethyl)phenox))(tert-butyl)dimethylsilane **2q**.

TMSO (4-(azidomethyl)phenoxy)(*tert*-butyl)dimethylsilane (20)

Yield: 67% (2 steps); Colorless liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.19-7.15 (m, 2H), 6.86-6.82 (m, 2H), 4.26(s, 2H), 0.98 (s, 9H), 0.20 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ =

156.0, 129.9, 128.3, 120.6, 54.7, 25.9, 18.4, -4.2; IR(NaCl): $\nu = 3050$, 2957, 2931, 2886, 2859, 2097, 1610, 1512 cm⁻¹; HRMS (FAB): m/z calcd. for C₁₃H₂₁N₃OSi: 263.1454; found: 263.1453.

3. Synthesis of Aryl azides

To a solution of methyl 4-aminobenzoate (7.0 mmol) in water (40 mL), aqueous hydrochloric acid (35%, 12 mL) is added. The mixture was cooled to 0 °C then a solution of sodium nitrite (1.1 equiv, 7.7 mmol) in water (10 mL) is added. After stirring for 20 min at 0 °C. Sodium azide (1.2 equiv, 8.4 mmol) was added portionwise. The reaction mixture was stirred at room temperature for 4 h. After completion of the reaction, the mixture was extracted with diethyl ether (3 x 40 mL). The oragnic layer was washed with saturated NaHCO₃ aqueous solution (100 mL), H₂O (100 mL) and brine (100 mL), dried over anhydrous sodium sulfate, concentrated under reduced pressure. The residue was purified by column chromatography on silica gel to afford the aryl azides.

N₃ CO₂Me Methyl 4-azidobenzoate (4a)^[10]

Yield: 90%; Pale yellow soild; ¹H NMR (300 MHz, CDCl₃) δ = 8.05-8.01 (m, 2H), 7.09-7.04 (m, 2H), 3.91 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ = 166.5, 144.9, 131.6, 126.9, 119.0, 52.3.

N₃ Azidobenzene (4b)^[10]

Yield: 85%; Pale yellow liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.38-7.31 (m, 2H), 7.16-7.10 (m, 1H), 7.05-7.00 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 140.2, 130.0, 125.1, 119.2.

N₃ OMe 1-azido-4-methoxybenzene (4c)^[10]

Yield: 93%; Brown solid; ¹H NMR (300 MHz, CDCl₃) δ = 6.96-6.92 (m, 2H), 6.90-6.86 (m, 2H), 3.78 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ = 157.2, 132.5, 120.2, 115.3, 55.7.

4. Synthesis of nitriles and anilines

The ruthenium catalyst **1** (2.0 mol%, 0.005 mmol) was added to a flame-dried J-Young flask filled with argon. The dry THF (1.0 mL) and benzyl azide (0.25 mmol) were added under a stream of argon. The reaction mixture was stirred for 2 h at room temperature with a household 30W fluorescent light. After the accumulation of N-H aldimine, aryl azide (1.0 equiv, 0.25 mmol) was added then stirred at 70 °C for 2 h. After completion of the reaction, solvent was removed under reduced pressure. The residue was purified by column chromatography on silica gel to afford corresponding nitrile and aniline.



Yield: 98%; White solid; ¹H NMR (300 MHz, CDCl₃) δ = 7.61-7.58 (m, 2H), 6.98-6.94 (m, 2H), 3.86 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ = 163.0, 134.2, 119.4, 114.9, 104.2, 55.7.

4-methylbenzonitrile (5b)^[12]

Yield: 99%; ¹H NMR (300 MHz, CDCl₃, in crude mixture with an internal standard (CH₂Br₂)) $\delta = 7.55-7.52$ (m, 2H), 7.28-7.25 (m, 2H), 2.41 (s, 3H).

F₃C CN 4-(trifluoromethyl)benzonitrile (5c)^[12]

Yield: 98%; White solid; ¹H NMR (300 MHz, CDCl₃) δ = 7.82 (d, *J* = 8.3 Hz, 2H), 7.77 (d, *J* = 8.5 Hz, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 135.0, 134.5, 132.9, 126.44, 126.40, 126.34, 126.29, 125.1, 121.4, 117.6, 116.3.



Yield: 98%; White solid; ¹H NMR (300 MHz, CDCl₃) δ = 7.80 (s, 4H); ¹³C NMR (75 MHz, CDCl₃) δ = 133.0, 117.2, 116.9.



Yield: 99%; ¹H NMR (300 MHz, CDCl₃, in crude mixture with an internal standard (CH₂Br₂)) $\delta = 7.60-7.57$ (m, 1H), 7.50-7.44 (m, 1H), 7.32-7.23 (m, 2H), 2.54 (s, 3H).



Yield: 99%; ¹H NMR (300 MHz, CDCl₃, in crude mixture with an internal standard (CH₂Br₂)) δ = 7.46-7.31 (m, 4H), 2.38 (s, 3H).

Yield: 90%; ¹H NMR (300 MHz, THF-d8, in crude mixture with an internal standard (CH₃NO₂)) δ = 7.71-7.68 (m, 2H), 7.65-7.59 (m, 1H), 7.52-7.47 (m, 2H).

CI CN 4-chlorobenzonitrile (5h)^[11]

Yield: 99%; White solid; ¹H NMR (300 MHz, CDCl₃) δ = 7.63-7.59 (m, 2H), 7.49-7.45 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 139.7, 133.5, 129.8, 118.1, 110.9.



Yield: 99%; White solid; ¹H NMR (300 MHz, CDCl₃) δ = 7.66-7.62 (m, 2H), 7.55-7.51 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 133.6, 132.8, 128.1, 118.2, 111.4.



Yield: 99%; Colorless liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.60-7.57 (m, 2H), 7.50-7.46 (m, 2H), 1.33 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ = 156.8, 132.2, 126.4, 119.3, 109.5, 35.5, 31.1.

CN 2-naphthonitrile (5k)^[11]

Yield: 96%; White solid; ¹H NMR (300 MHz, CDCl₃) δ = 8.22 (s, 1H), 7.92-7.87 (m, 3H), 7.67-7.57 (m, 3H); ¹³C NMR (75 MHz, CDCl₃) δ = 134.8, 134.3, 132.4, 129.4, 129.2, 128.6, 128.2, 127.8, 126.5, 119.4, 109.6.



Yield: 99%; White solid; ¹H NMR (300 MHz, CDCl₃) δ = 8.16-8.13 (m, 2H), 7.76-7.74 (m, 2H), 3.97 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ = 165.6, 134.1, 132.4, 130.3, 118.2, 116.6, 52.9.

4-vinylbenzonitrile (5m)^[15]

Yield: 96%; Colorless liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.61 (d, *J* = 8.2 Hz, 2H), 7.48 (d, *J* = 8.2 Hz, 2H), 6.72 (dd, *J* = 17.6, 10.9 Hz, 1H), 5.87 (d, *J* = 17.6 Hz, 1H), 5.45 (d, *J* = 10.9 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ = 142.1, 135.6, 132.6, 126.9, 119.1, 117.9, 111.3.

O 4-formylbenzonitrile (5n)^[12]

Yield: 90%; White solid; ¹H NMR (300 MHz, CDCl₃) δ = 10.10 (s, 1H), 8.03-7.99 (m, 2H), 7.87-7.84 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 190.8, 138.9, 133.1, 130.1, 117.9, 117.8.



Yield: 99%; Colorless liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.56-7.51 (m, 2H), 6.91-6.87 (m, 2H), 0.98 (s, 9H), 0.23 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ = 159.9, 134.2, 121.0, 119.4, 104.8, 25.7, 18.4, -4.2.



Yield: 99%; White solid; ¹H NMR (300 MHz, CDCl₃) δ = 7.87-7.82 (m, 2H), 6.65-6.61 (m, 2H), 4.09 (br s, 2H), 3.85 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ = 167.4, 151.1, 131.8, 119.9, 114.0, 51.8.



Yield: 99%; Yellow liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.18-7.11 (m, 2H), 6.76-7.73 (m, 1H), 6.65-6.63 (m, 2H), 3.58 (br s, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 146.6, 129.5, 118.7, 115.3.

NH₂ OMe 4-methoxyaniline (6c)^[17]

Yield: 99%; Brown solid; ¹H NMR (300 MHz, CDCl₃) δ = 6.75-6.70 (m, 2H), 6.63-6.57 (m, 2H), 3.71 (s, 3H), 3.41 (br s, 2H); ¹³C NMR (75 MHz, CDCl₃) δ = 152.8, 140.1, 116.4, 114.9, 55.8.

5. Synthesis of amidine 7

The ruthenium catalyst **1** was added to a flame-dried J-Young flask filled with argon. The dry THF (1.0 mL) and **2a** (0.25 mmol) were added under a stream of argon. The reaction mixture was stirred for 2 h at room temperature with a household 30W fluorescent light. After the accumulation of **3a**, **4a** (1.0 equiv, 0.25 mmol) was added then stirred at 70 °C for 2 h. The reaction mixture was cooled to room temperature then NaH (1.5 equiv, 0.375 mmol) is added at 0 °C and stirred at room temperature for 12h. After completion of the reaction, the reaction mixture was quenched with water at 0 °C, extracted with diethyl ether (3 x 5 mL). The oragnic layer was washed with H₂O (2 x 30 mL), brine (30 mL), dried over anhydrous sodium sulfate, concentrated under reduced pressure. The residue was recrystallized at -20°C (n-hex:dichloromethane = 2:1) to afford amidine **7**.



4-methoxy-N-phenylbenzimidamide (7)^[19]

Yield: 92%; White solid; ¹H NMR (300 MHz, CDCl₃) δ = 7.78 (d, *J* = 8.5 Hz, 2H), 7.33 (t, *J* = 7.7 Hz, 2H), 7.06-7.01 (m, 1H), 6.97-6.90 (m, 4H); ¹³C NMR (75 MHz, CDCl₃) δ = 161.7, 154.9, 149.6, 129.6, 128.6, 123.1, 122.0, 113.9, 55.6.

6. Mechanistic investigation

6-a. The reaction using deuterated benzyl azide

6-a-1. Synthesis of 1-(azidomethyl-d2)-4-methylbenzene 8



To a solution of methyl 4-methylbenzoate (5.0 mmol) THF (10 mL), lithium aluminum deteride (1.2 equiv, 6.0 mmol) was added at 0 °C. The reaction mixture was stirred at room temperature for 3 h. After completion of the reaction, the mixture was quenched with 1 N HCl (10 mL), extracted with ethyl acetate (3 x 10 mL). The oragnic layer was washed with H₂O (30 mL) and brine (30 mL), dried over anhydrous sodium sulfate, concentrated under reduced pressure. The crude residue was directly used for next step without further purification. To a solution of *p*-tolylmethanol-*d2* and diphenyl phosphoryl azide (DPPA) (2.0 equiv, 10 mmol) in toluene (15 mL), Diazabicyclo[5.4.0]undec-7-ene (DBU) (2.0 equiv, 10 mmol) was added dropwise at 0 °C. The reaction mixture was stirred at room temperature for 12 h and quenched with water (30 mL). The solution was extracted with dichloromethane (3 x 20 mL). The organic was washed with H₂O (2 x 30 mL) brine (30 mL), dried over anhydrous sodium sulfate, concentrated under reduced pressure. The residue was purified by flash column chromatography to afford the deuterated azide **8**.

D D N₃ 1-(azidomethyl-*d2*)-4-methylbenzene (8)

Yield: 60% (2 steps); Pale yellow liquid; ¹H NMR (300 MHz, CDCl₃) δ = 7.26-7.18 (m, 4H), 2.37 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ = 138.3, 132.3, 129.6, 128.4, 54.46, 54.29, 54.11, 53.94, 21.3; IR(NaCl): ν = 3026, 3001, 2924, 2868, 2495, 2106, 1907, 1615, 1516 cm⁻¹; HRMS (EI): m/z calcd. for C₈H₇D₂N₃: 149.0922; found: 149.0921.

6-a-2. Synthesis of nitrile and methyl-4-amino-d2-benzoate

The ruthenium catalyst 1 (2.0 mol%, 0.0025 mmol) was added to a flame-dried NMR reaction J-young tube and filled with argon. The THF-d8 (0.6 mL) and 8 (0.125 mmol) were added under a stream of argon. The reaction mixture was stirred for 2 h at room temperature with a household 30W fluorescent light. After the accumulation of N-H aldimine, 4a (1.0 equiv, 0.125 mmol) was added then stirred at 70 °C for 2 h.

ND₂ CO₂Me Methyl 4-amino(d-2)benzoate (10)

¹H NMR (300 MHz, THF-d8, in crude mixture) δ = 7.73-7.68 (m, 2H), 6.55-6.51 (m, 2H), 3.73 (3H).

6-a-3. Synthesis of nitrile and methyl-4-amino-d2-benzoate for deuterium NMR

The ruthenium catalyst 1 (2.0 mol%, 0.0025 mmol) was added to a flame-dried NMR reaction J-young tube and filled with argon. The THF (0.6 mL) and 8 (0.125 mmol) were added under a stream of argon. The reaction mixture was stirred for 2 h at room temperature with a household 30W fluorescent light. After the accumulation of N-H aldimine, aryl azide (1.0 equiv, 0.125 mmol) was added then stirred at 70 °C for 2 h. The yield of methyl 4-amino-*d2*-benzoate 10 was determined by deterium NMR using THF-d8 as an internal standard.

ND₂ CO₂Me Methyl 4-amino(*d*-2)benzoate (10)

²D NMR (500 MHz, THF-d8, in crude mixture with an internal standard (THF-d8)) $\delta = 5.16$ (s, 2D)

6-b. The reaction using Ru-H 12 as a catalyst

To a solution of the ruthenium chloride **11** (0.02 mmol) in THF (1.0 mL), tributyltin hydride (0.02 mmol) was added. The reaction mixture was stirred for 3 h at room temperature with a 12W blue LED to generate the ruthenium hydride **12** (The generation of **12** was checked by ¹H and ¹³C NMR using THF-d8 as a NMR solvent.^[1]). After generation of **12**, **2a** (0.25 mmol) and **4a** (0.25 mmol) were added and stirred at 70 °C for 16 h under illumination with a household 30 W fluorescent light. After completion of the reaction, solvent was removed under reduced pressure. The yield of **5a** and **6a** was determined by ¹H NMR using dibromomethane as an internal standrad.



Ruthenium hydride 12^[1]

¹H NMR (300 MHz, THF-d8, in crude mixture) $\delta = 7.32-7.27$ (m, 4H), 7.23-7.11 (m, 9H), 6.95 (s, 1H), 6.72-6.65 (m, 3H), 2.04 (s, 6H), -10.08 (s, 1H); ¹³C NMR (75 MHz, THF-d8, in crude mixture) $\delta = 202.9$, 148.4, 133.0, 132.9, 129.7, 128.4, 128.0, 118.6, 113.9, 110.0, 106.3, 102.2, 11.3.

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8. NMR Data

8-a. NMR spectrum of nitriles ¹H and ¹³C NMR spectrum of **5**a



¹H spectrum of **5b (crude)**



¹H and ¹³C NMR spectrum of **5**c



$^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectrum of $\mathbf{5d}$



¹H NMR spectrum of **5e (crude)**



¹H NMR spectrum of **5f (crude)**









¹H and ¹³C NMR spectrum of 5i





¹H and ¹³C NMR spectrum of **5**j





 $^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectrum of 5k





 $^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectrum of **51**





¹H and ¹³C NMR spectrum of **5m**

| 7.596 7.596 7.494 7.264 | 6.775 6.775 6.738 6.716 | -5.845 | 5.465 -5.428 | -0.000 |
|----------------------------------|----------------------------------|--------|-----------------|--------|
| NK L | N/A | M | V. | |





¹H and ¹³C NMR spectrum of **5n**





¹H and ¹³C NMR spectrum of **50**





¹H and ¹³C NMR spectrum of **7**





¹H and ¹³C NMR spectrum of 8





¹H NMR spectrum of **5b** and **10**



 $^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectrum of 12 and tin mixture

