

Supporting Information

Facile Synthesis of Graft Copolymers of Controlled Architecture. Copolymerization of Fluorinated and Non-Fluorinated Poly(dimethylsiloxane) Macromonomers with Trialkylsilyl Methacrylates using RAFT Polymerization

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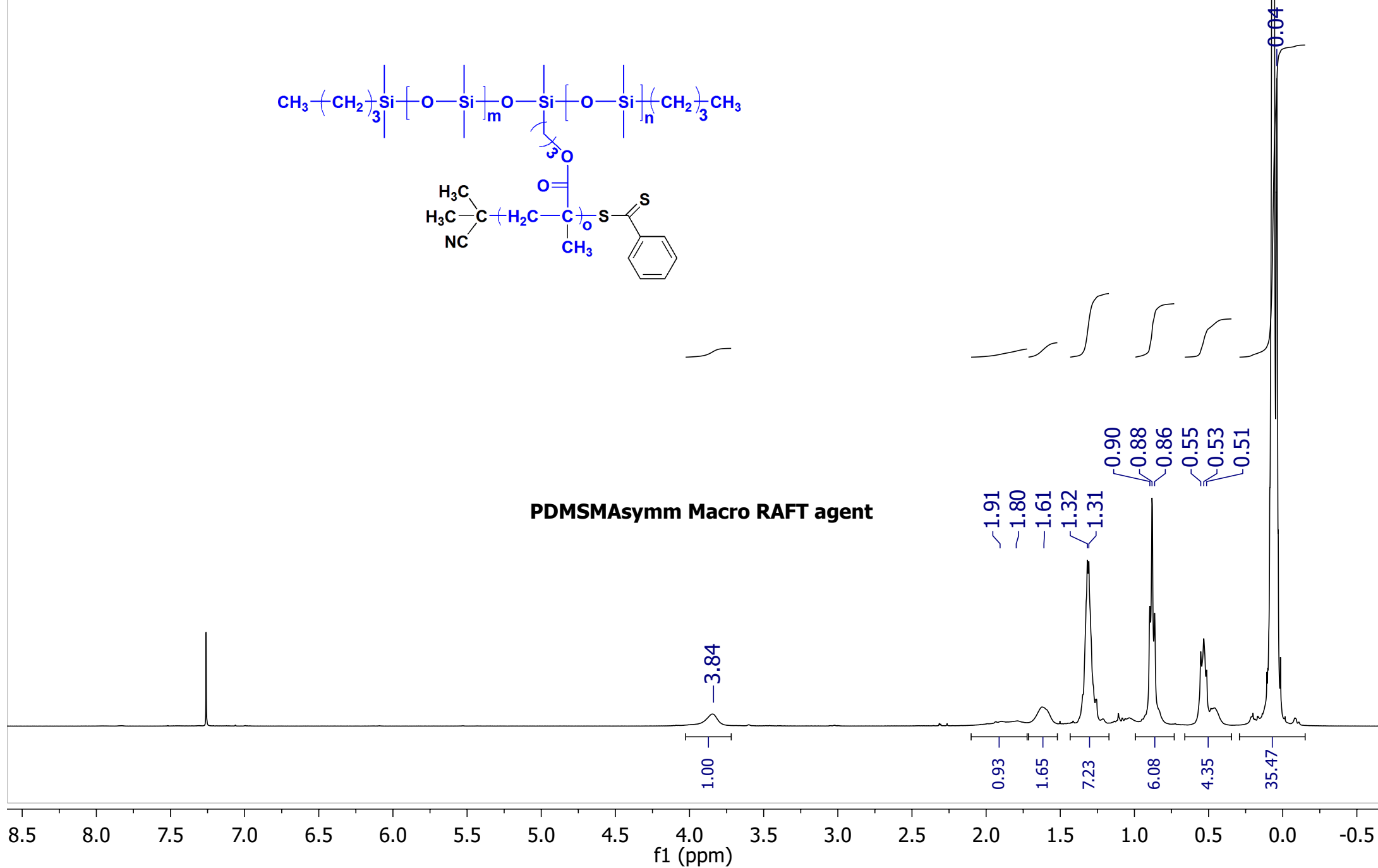
Figure 1. ^1H NMR spectrum of PDMSMA_{asymm} Macro RAFT agent in CDCl_3 

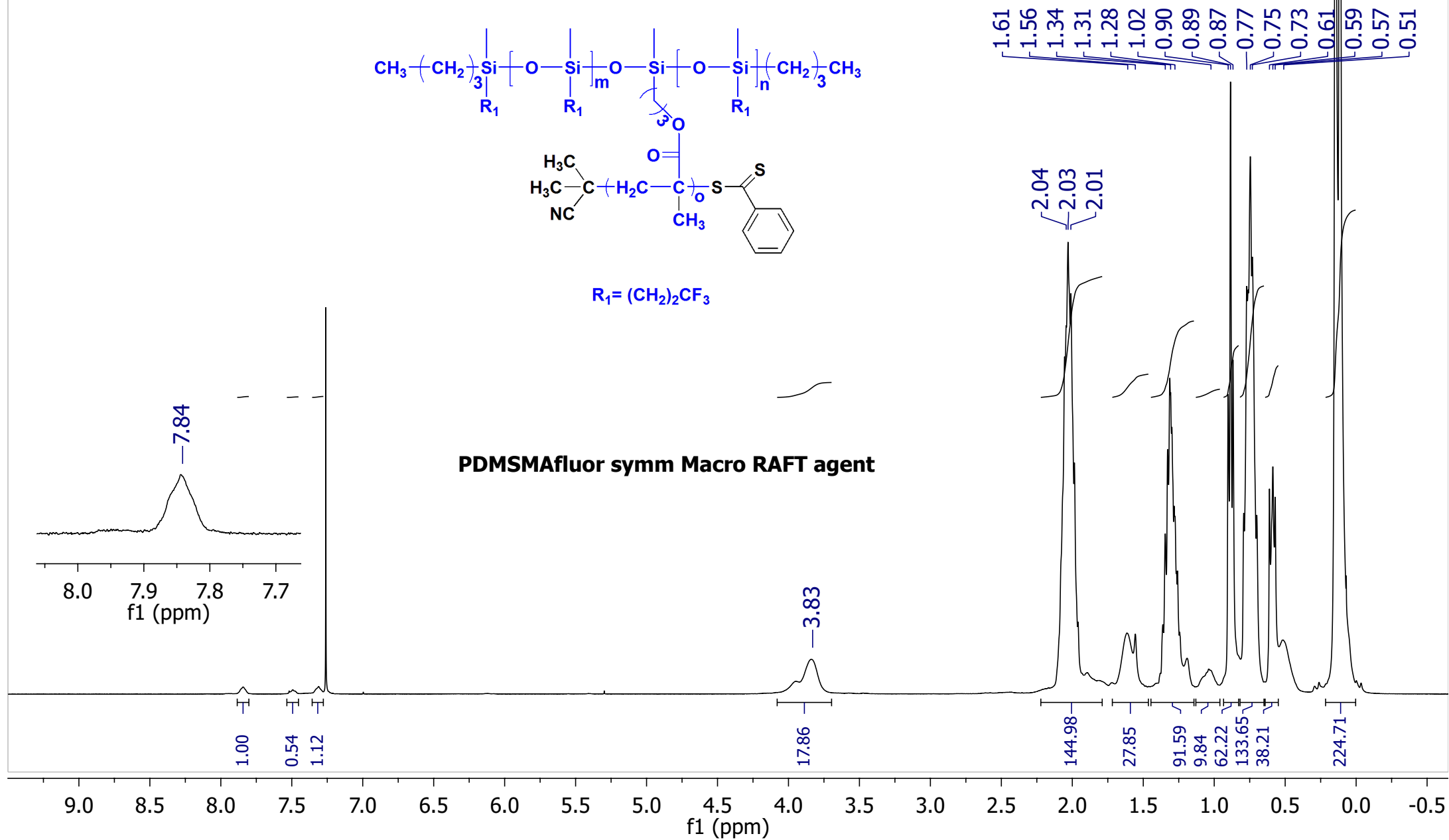
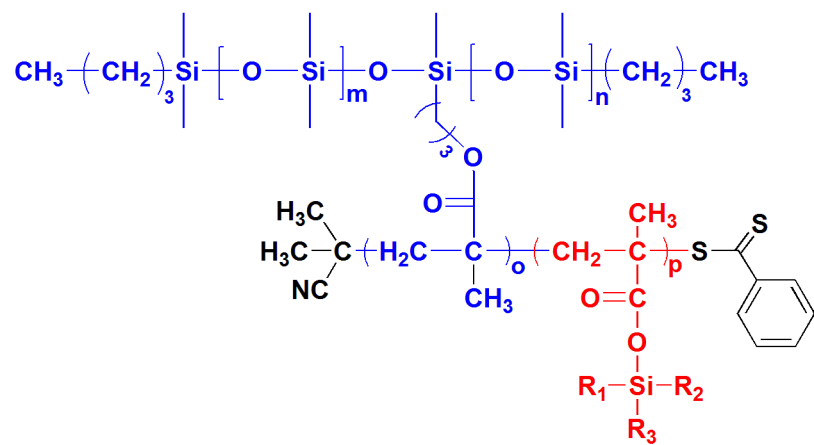
Figure 2. ^1H NMR spectrum of PDMSMAfluor symm Macro RAFT agent in CDCl_3 

Figure 3. ^1H NMR spectrum of p(PDMSMA_{symm})-b-p(TBDMSMA) in CDCl_3



$\text{R}_1 = \text{R}_2 = \text{Me}$; $\text{R}_3 = \text{t-Bu}$ TBDMSMA

p(PDMSMA_{symm})-b-p(TBDMSMA)

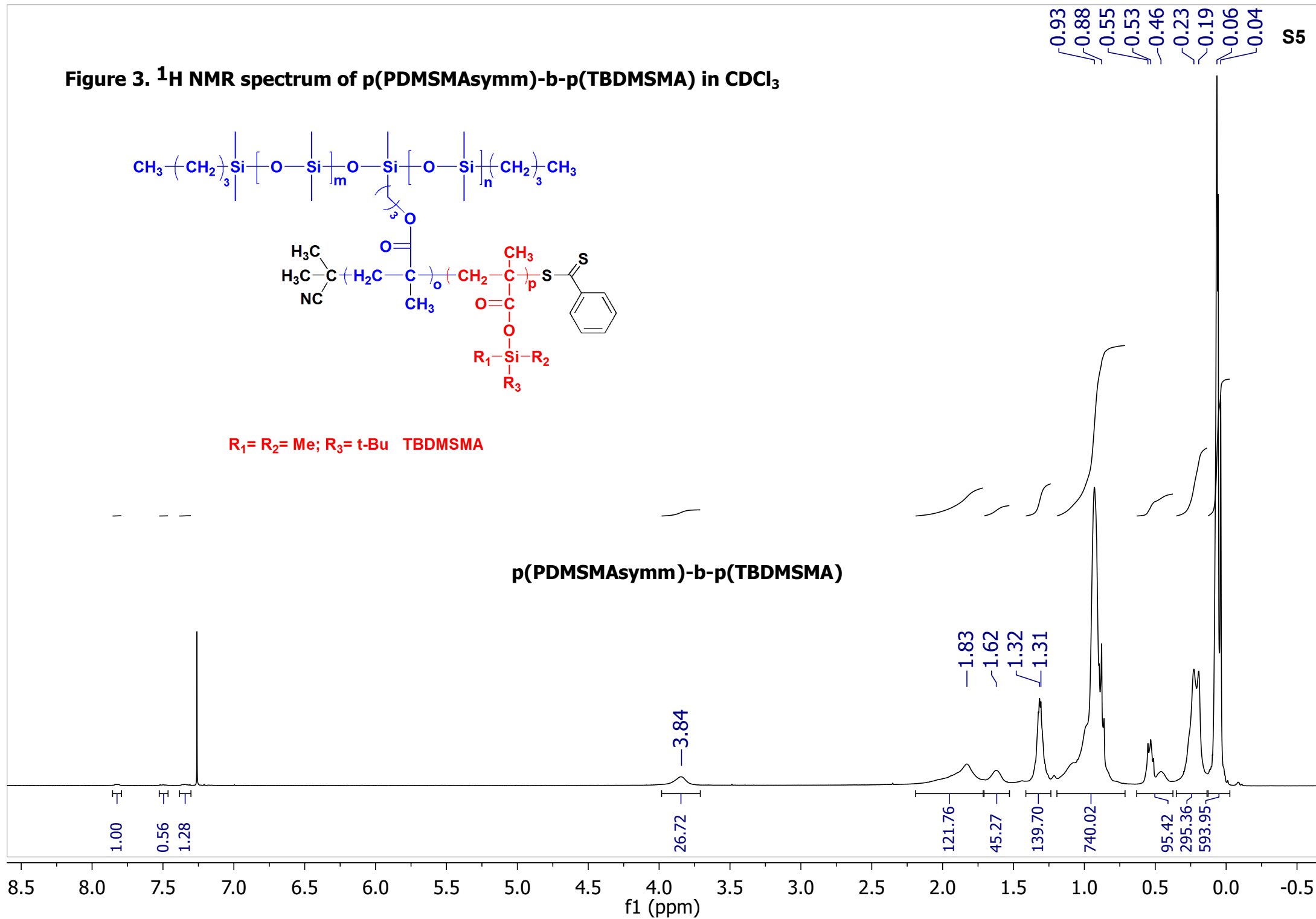


Figure 4. ^1H NMR spectrum of p(PDMSMAfluor symm)-b-p(TBDMSMA) in CDCl_3

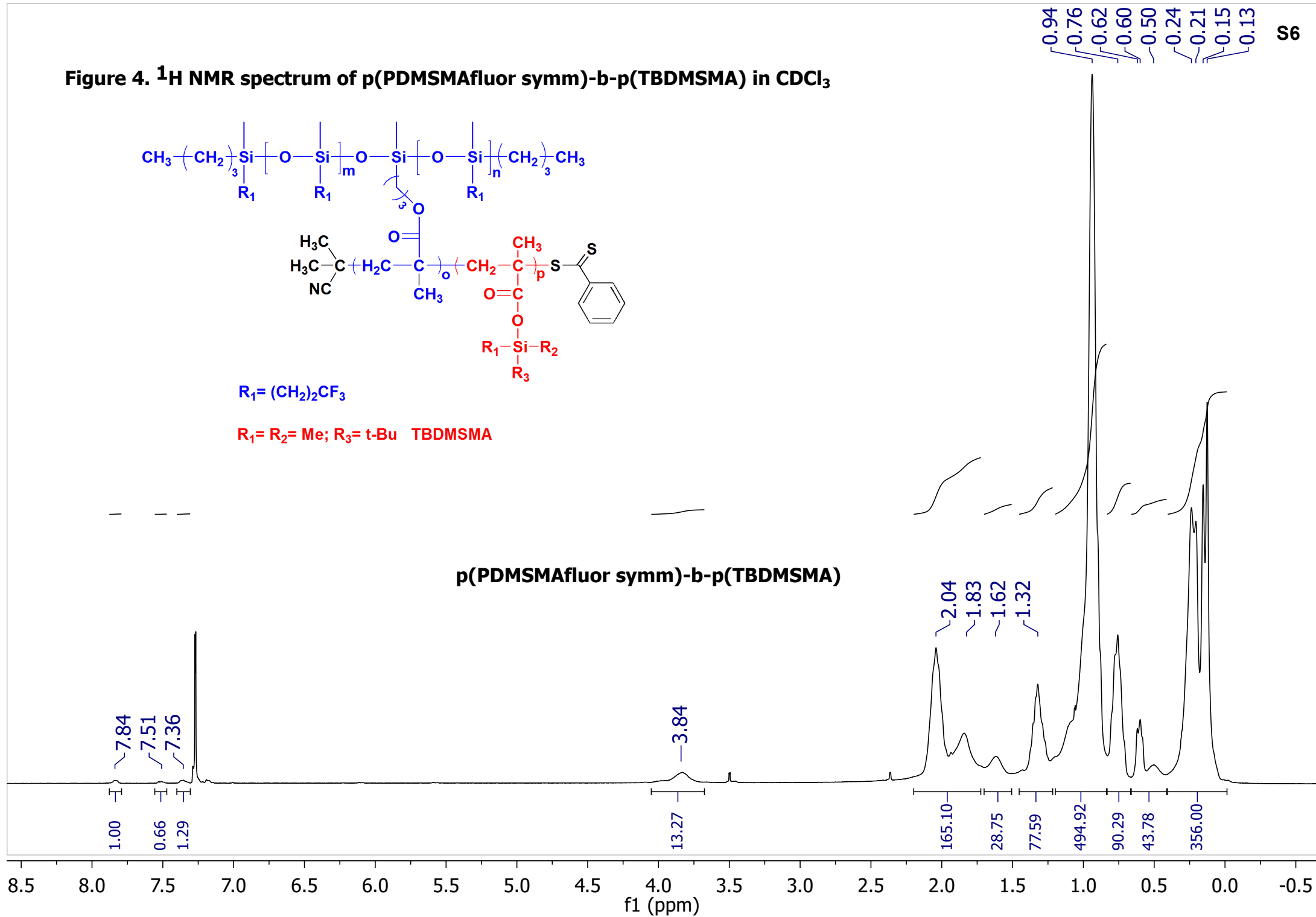
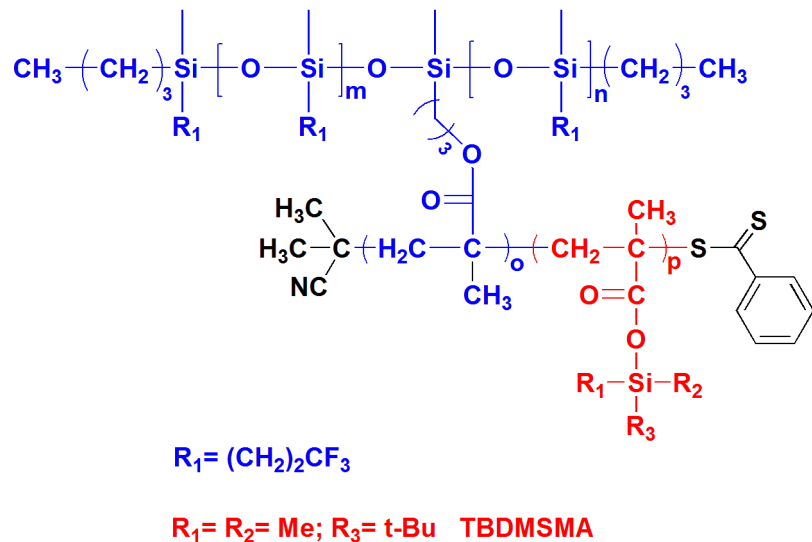
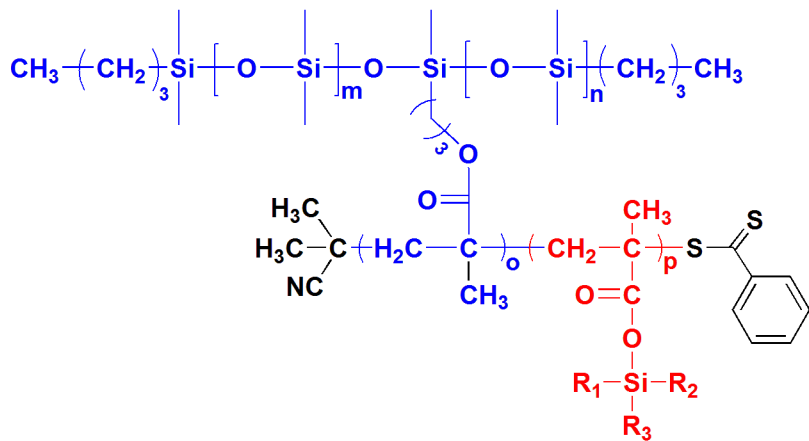


Figure 5. ¹H NMR spectrum of p(PDMSMA_{symm})-b-p(TIPSMA) in CDCl₃



R₁ = R₂ = R₃ = i-Pr TIPSMA

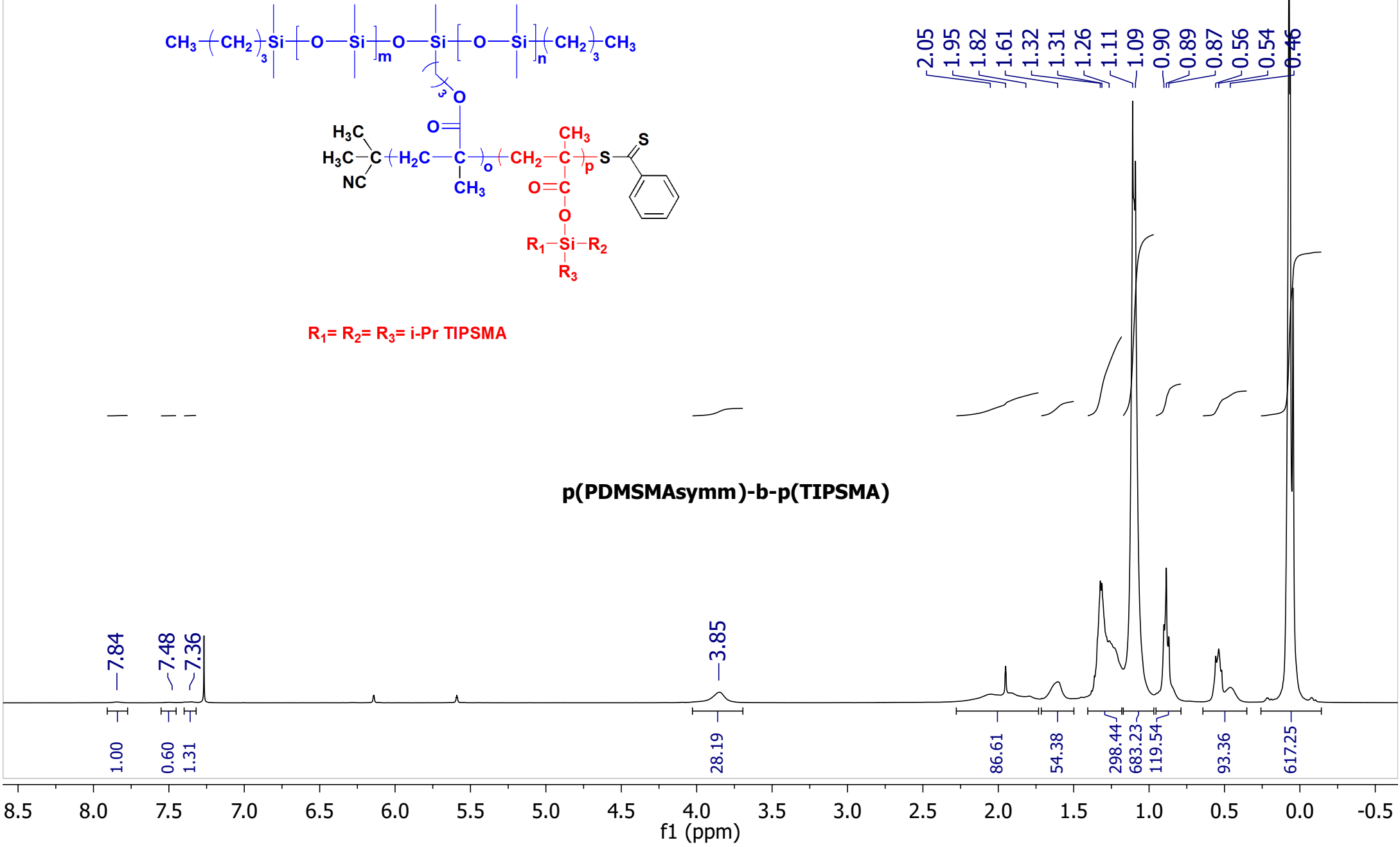
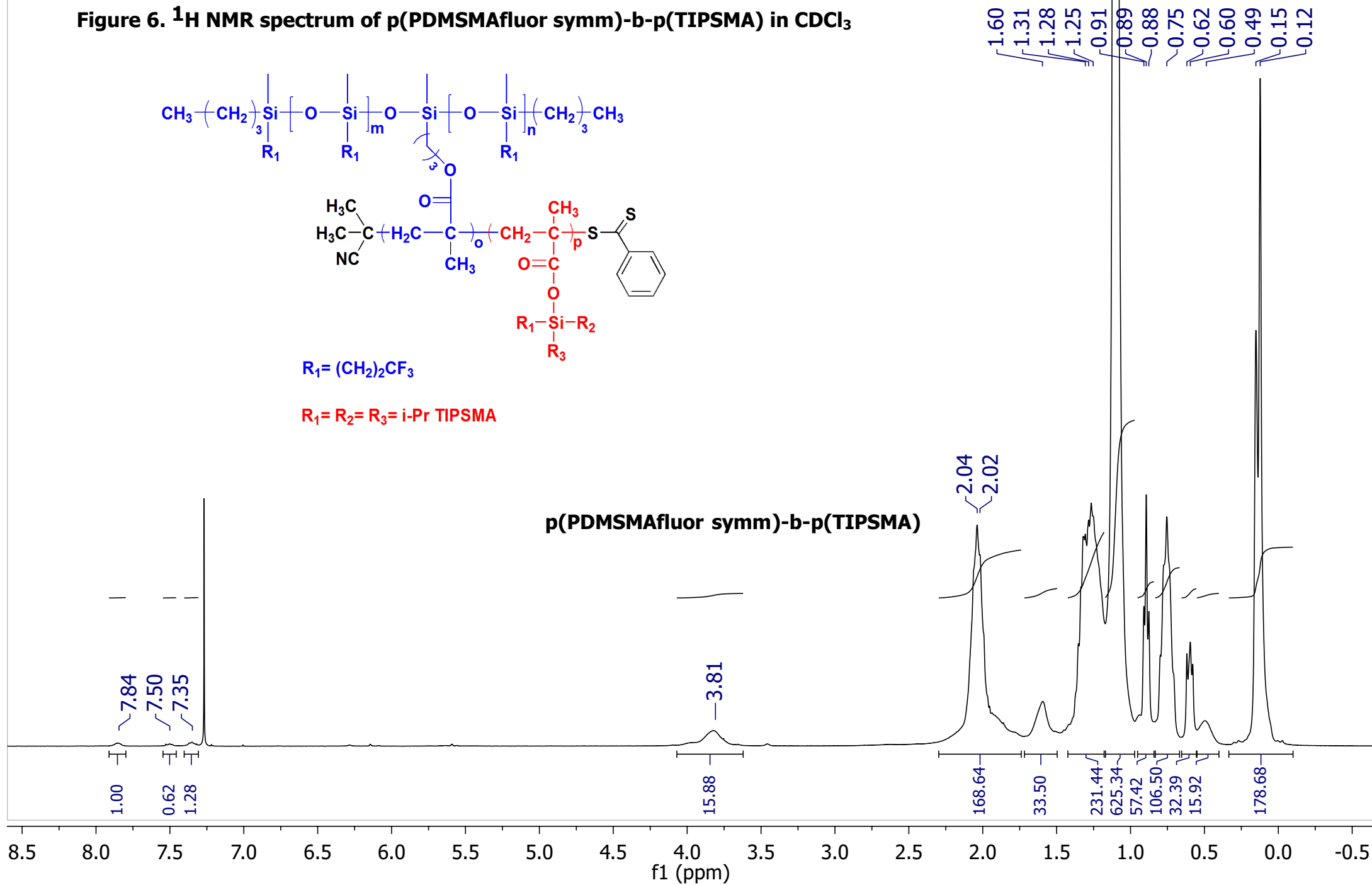
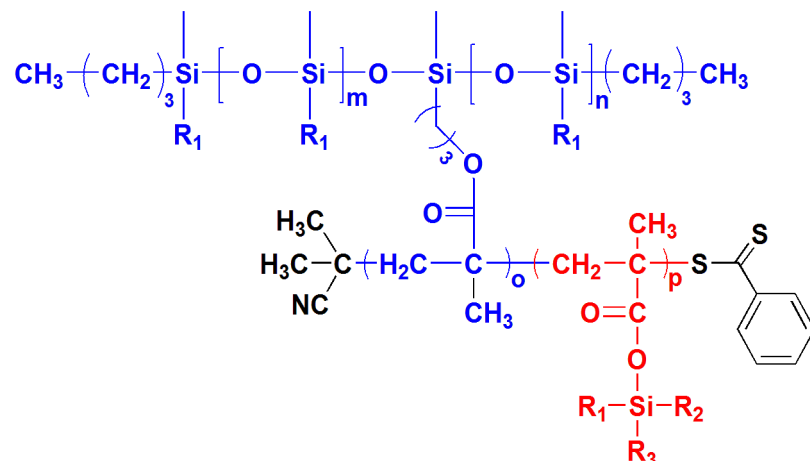


Figure 6. ^1H NMR spectrum of p(PDMSMAfluor symm)-b-p(TIPSMA) in CDCl_3



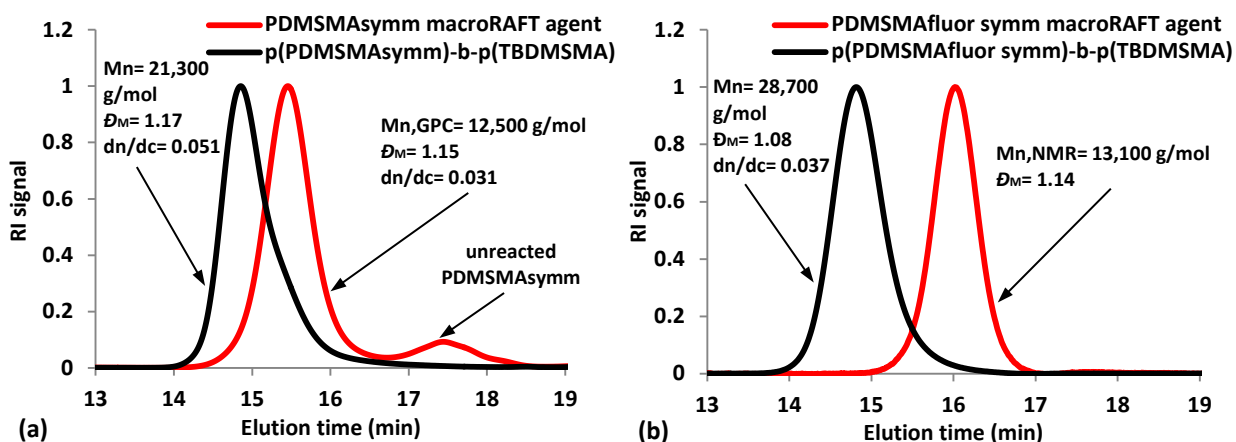


Figure S7. (a) TD-SEC traces of PDMSsymm macro RAFT agent ($M_n = 12\,500$ g/mol and $\bar{D}_M = 1.15$) and p(PDMSAsymm)-b-p(TBDMSMA) block copolymer ($M_n = 21\,300$ g/mol and $\bar{D}_M = 1.17$) obtained by *in situ* chain extension of macro RAFT agent with TBDMSMA monomer at 70°C in toluene. For the determination of $M_{n,TD-SEC}$ crude PDMSsymm macro RAFT agent was used. (b) TD-SEC traces of PDMSfluor symm macro RAFT agent ($M_{n,NMR} = 13\,100$ g/mol and $\bar{D}_M = 1.14$) and p(PDMSAfluor symm)-b-p(TBDMSMA) block copolymer ($M_n = 28\,700$ g/mol and $\bar{D}_M = 1.08$) obtained by *in situ* chain extension with TBDMSMA at 70°C in toluene. For the determination of $M_{n,NMR}$ precipitated PDMSfluor symm macro RAFT agent was used.

Determination of global kinetic constants of trialkylsilyl methacrylates and PDMS macromonomers

The global kinetic constants (k_g) of polymerization of trialkylsilyl methacrylates and PDMSMA macromonomers were determined using an overall kinetics law of the following form (Equation 1)¹:

$$V_p = -\frac{d[M]}{dt} = k_p \times [M] \times [P_n] \quad (1)$$

where k_p is the propagation rate constant, $[M]$ is the monomer molar concentrations, and $[P_n]$ is the radical concentration in the reaction media. In the quasi-stationary state, characterized by a constant concentration of radical species in the reaction medium, the amount of the initiated radicals is equal to the amount of the terminated radicals as shown in Equation 2:

$$2 \times f \times k_d \times [A] = 2 \times k_t \times [P_n] \times [P_n] \quad (2)$$

where k_d is the rate constant of decomposition of the initiator (initiation rate constant), f is the efficiency of the initiator, $[A]$ and $[A_0]$ are the initial and instantaneous concentrations of the initiator, and k_t is the termination rate constant. Therefore, it follows:

$$[P_n] = \sqrt[2]{(f \times k_d \times [A])/k_t} = \sqrt[2]{(f \times k_d \times [A_0] \times e^{-k_d t})/k_t} \quad (3)$$

By substituting Equation (3) in Equation (1), it follows:

$$V_p = -\frac{d[M]}{dt} = k_p \times [M] \times \sqrt[2]{f \times k_d \times [A_0] \times e^{-k_d t}/k_t} \quad (4)$$

If the initiator concentration is considered constant, the integration of the Equation (4) gives:

$$\ln \frac{[M_0]}{[M]} = k_p \times \sqrt[2]{f \times k_d \times [A_0]/k_t} \quad (5)$$

The values of global rate constants of RAFT polymerization ($k_g = k_p \times \sqrt[2]{f \times k_d/k_t}$) were therefore calculated from the slope of the each straight line in Figure S8.

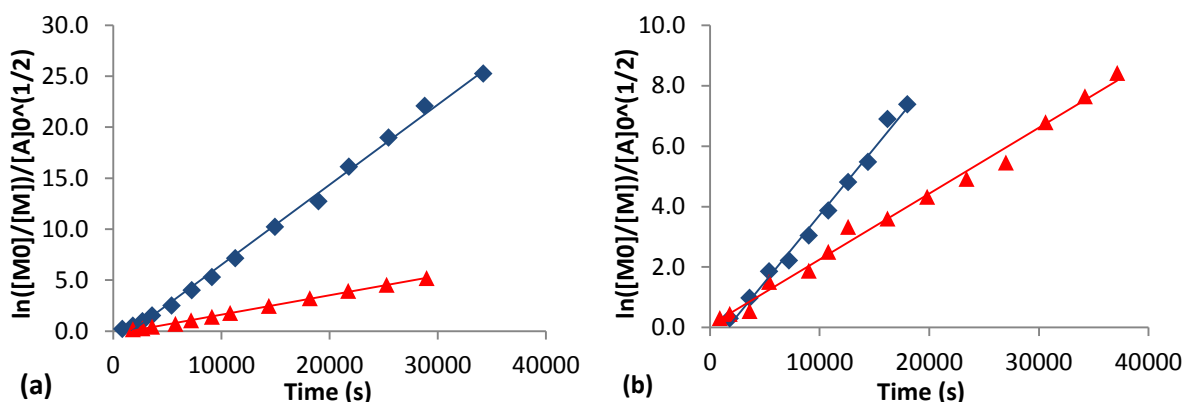


Figure S8. (a) Plots of $\ln([M_0]/[M])/[A]_0^{1/2}$ versus polymerization time for TBDMSMA (\blacklozenge) and TIPSMA (\blacktriangle) polymerization in toluene at 70°C ([CPDB]/[AIBN]= 5/1): $k_g(\text{TBDMSMA})= 8.0 \times 10^{-4} \text{ L mol}^{-1/2} \text{ s}^{-1}$ ($R^2= 0.997$); $k_g(\text{TIPSMA})= 2.0 \times 10^{-4} \text{ L mol}^{-1/2} \text{ s}^{-1}$ ($R^2= 0.999$). (b) Plots of $\ln([M_0]/[M])/[A]_0^{1/2}$ versus polymerization time for symmetrical PDMSMA (\blacklozenge), and fluorinated symmetrical PDMSMA (\blacktriangle) polymerization in toluene at 70°C ([CPDB]/[AIBN]= 2/1): $k_g(\text{PDMSMA symm})= 4.0 \times 10^{-4} \text{ L mol}^{-1/2} \text{ s}^{-1}$ ($R^2= 0.991$); $k_g(\text{PDMSMA fluor symm})= 2.0 \times 10^{-4} \text{ L mol}^{-1/2} \text{ s}^{-1}$ ($R^2= 0.991$).

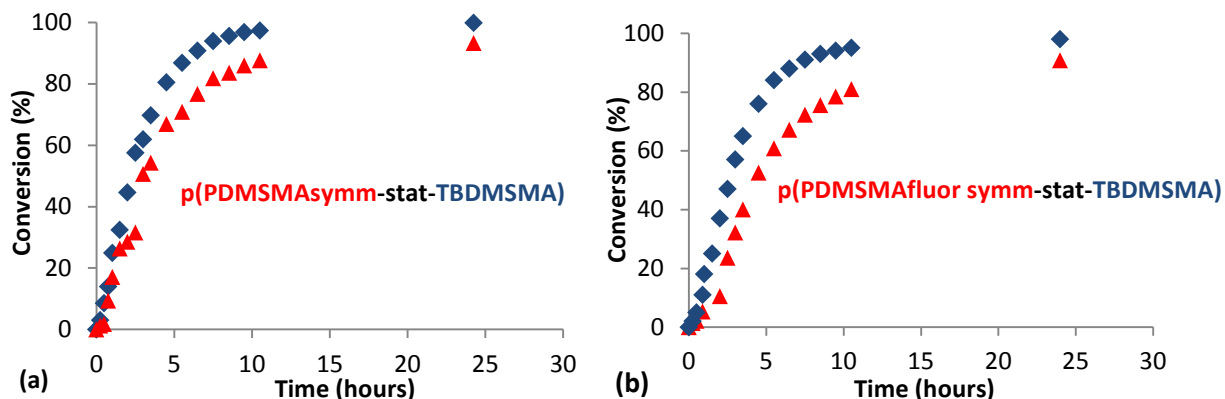


Figure S9. (a) Plot of monomer conversion versus time for RAFT copolymerization of TBDMSMA (\blacklozenge) and symmetrical PDMSMA macromonomer (PDMSMA_{symm}, \blacktriangle), in the presence of CPDB at 70°C in toluene. (b) Plot of monomer conversion *versus* time for RAFT copolymerization of TBDMSMA (\blacklozenge) and symmetrical fluorinated PDMSMA macromonomer (PDMSMA_{fluor symm}, \blacktriangle), in the presence of CPDB at 70°C in toluene.

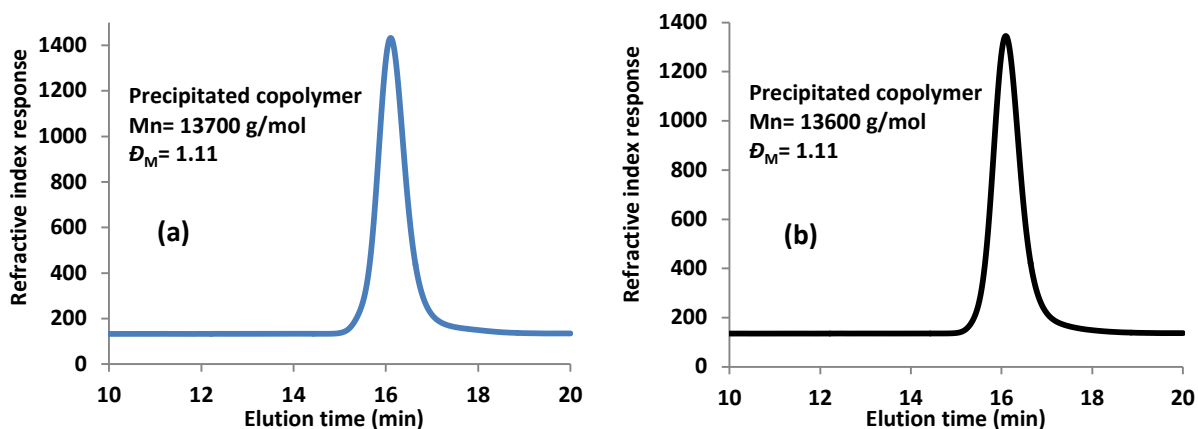


Figure S10. TD-SEC traces of statistical copolymers obtained by CPDB-mediated RAFT copolymerization of TBDMSMA and either PDMS_{symm} or PDMSMA_{fluor symm} macromonomers at 70°C in toluene. (a) p(PDMSMA_{symm}-stat-TBDMSMA) (Mn= 13 700 g/mol and Đ_M= 1.11). (b) p(PDMSMA_{fluor symm}-stat-TBDMSMA) (Mn= 13 600 g/mol and Đ_M= 1.11).

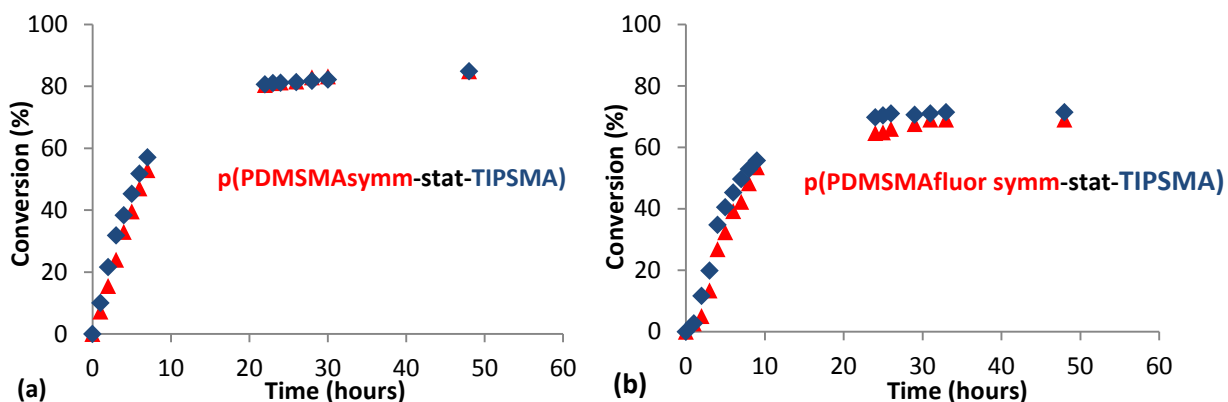


Figure S11. (a) Plot of monomer conversion *versus* time for RAFT copolymerization of TIPSMA (\blacklozenge) and symmetrical PDMSMA macromonomer (\blacktriangle) (PDMSMA_{symm}), in the presence of CPDB at 70°C in toluene. (b) Plot of monomer conversion *versus* time for RAFT copolymerization of TIPSMA (\blacklozenge) and symmetrical fluorinated PDMSMA macromonomer (PDMSMA_{fluor symm}, (\blacktriangle)), in the presence of CPDB at 70°C in toluene.

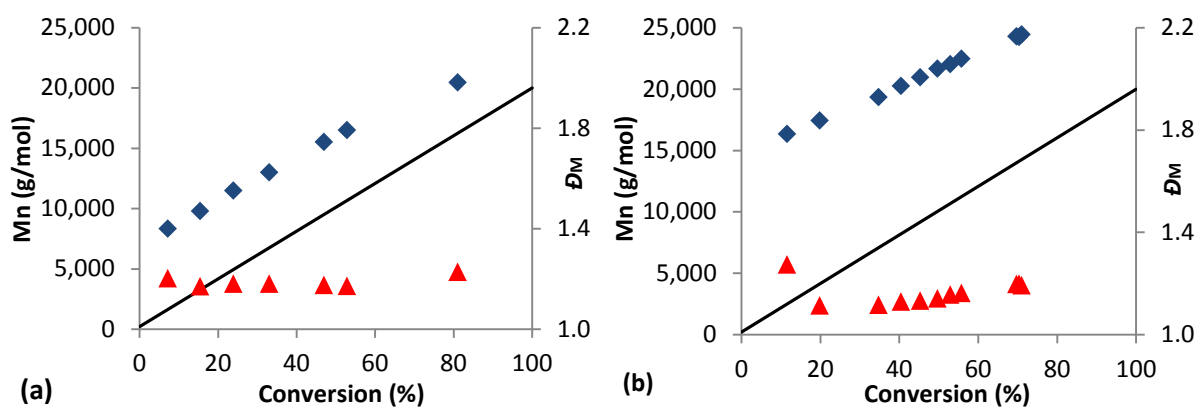


Figure S12. Evolution of the number-average molar mass M_n (\blacklozenge) and dispersity D_M (\blacktriangle) with monomer conversion in the statistical copolymerization of TIPSMA and either PDMSMA_{symm} (a) or PDMSMA_{fluor symm} (b) at 70°C mediated by CPDB.

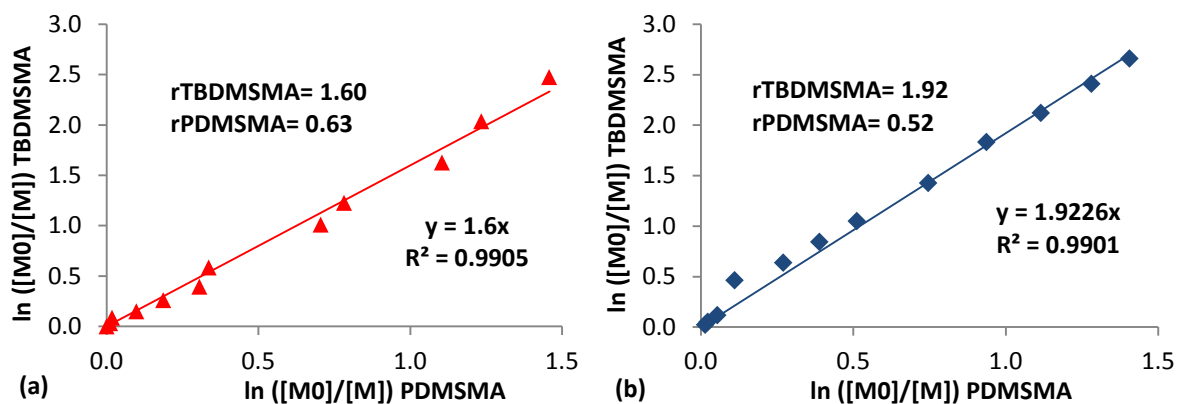


Figure S13. Jaacks plot for the RAFT polymerization of TBDMSMA and PDMSMA_{symm} (a), and PDMSMA_{fluor symm} (b) in toluene at 70°C.

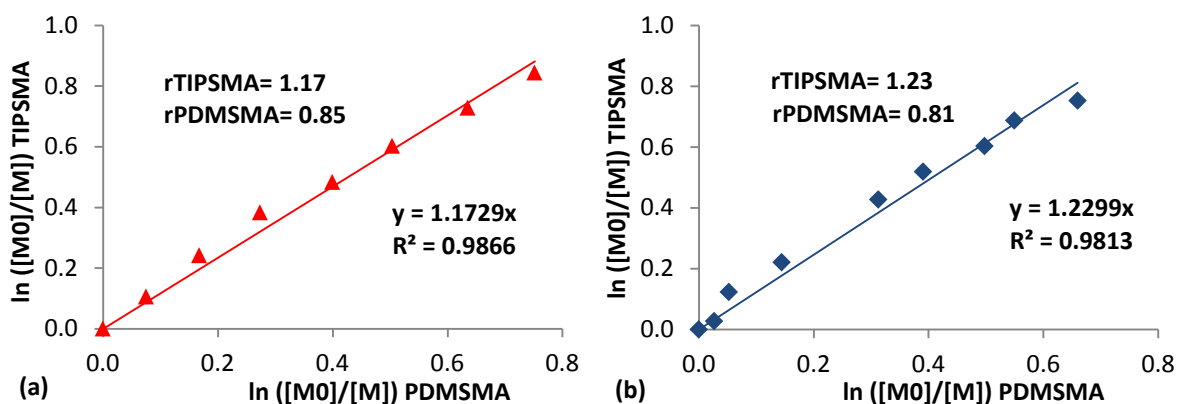


Figure S14. Jaacks plot for the RAFT polymerization of TIPSMA and PDMSMA_{symm} (a), and PDMSMA_{fluor symm} (b) in toluene at 70°C.

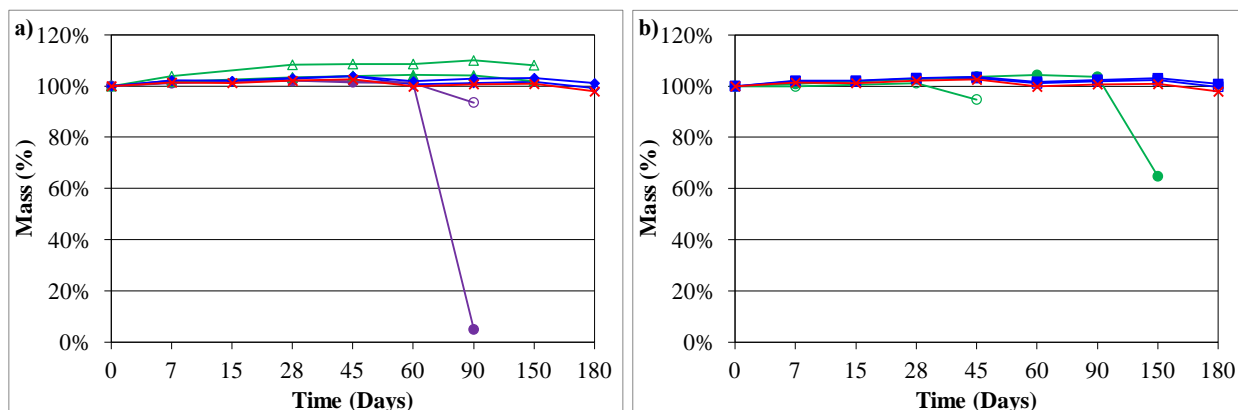


Figure S15. Evolution of the mass with immersion time in artificial seawater. **(a)** Coatings composed of statistical graft copolymers (\diamond) p(PDMSMA_{asymm-stat-TIPSMA}) 50/50, (\blacklozenge) p(PDMSMA_{fluor symm-stat-TIPSMA}) 50/50, (\circ) p(PDMSMA_{asymm-stat-TBDMSMA}) 20/80, (\bullet) p(PDMSMA_{fluor symm-stat-TBDMSMA}) 20/80, (\triangle) p(PDMSMA_{asymm-stat-TBDMSMA}) 50/50, (\blacktriangle) p(PDMSMA_{fluor symm-stat-TBDMSMA}) 50/50 **(b)** coatings composed of diblock graft copolymers (\square) p(PDMSMA_{asymm})-b-pTIPSMA 50/50, (\blacksquare) p(PDMSMA_{fluor symm})-b-pTIPSMA 50/50, (\circ) p(PDMSMA_{asymm})-b-pTBDMSMA 50/50, (\bullet) p(PDMSMA_{fluor symm})-b-pTBDMSMA 50/50. The film-forming hydrolyzable homopolymer pTIPSMA (\times) is used as reference.

References

- 1) Matyjaszewski, K.; Davis, T. P. Handbook of Radical Polymerization; John Wiley & Sons Inc., 2002.