

—Supporting Information—

Two in One: Use of Azide Functionality for Controlled Photo-crosslinking and Click-modification of Polymer Microspheres

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Synthesis of Chloromethyl-functional Microspheres

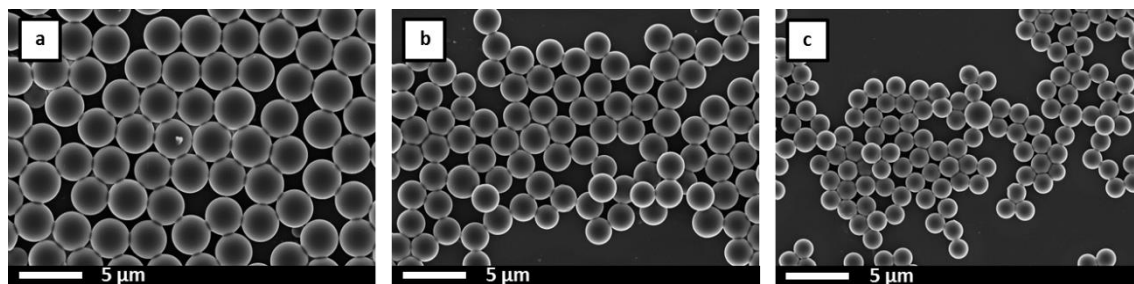


Figure S1. SEM images of 4-vinylbenzyl chloride (VBC)-functional microspheres with VBC feed contents of (a) 10 wt-%; (b) 50 wt-%; (c) 100 wt-% showing well-defined microspheres.

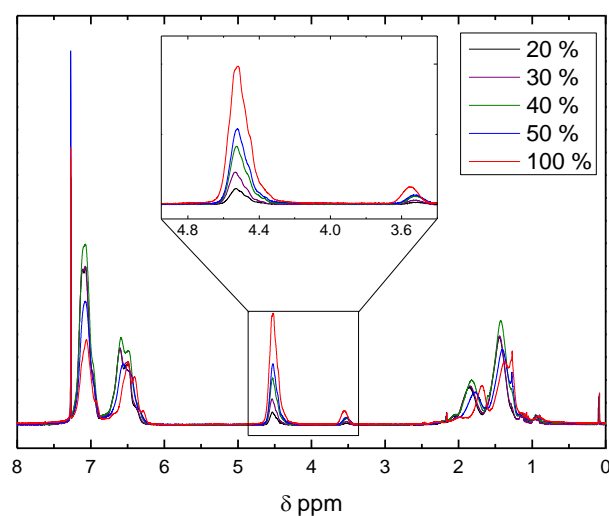


Figure S2. ^1H NMR spectra of microspheres (dissolved in CDCl_3) containing 4-vinylbenzyl chloride in varying amounts (20, 30, 40, 50, 100 wt-%). $\delta/\text{ppm} = 7.30\text{--}6.20$ (Ar- H); 4.51 ($-\text{CH}_2\text{Cl}$); 3.51 ($-\text{CH}_2\text{OEt}$); 2.10–0.90 (backbone).

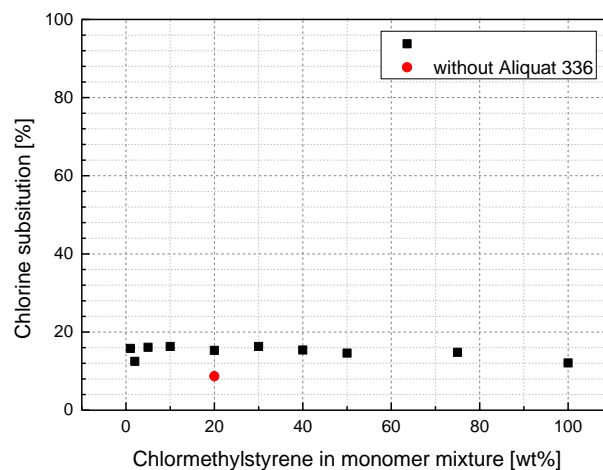


Figure S3. Degree of chlorine substitution by ethanol for different 4-vinylbenzyl chloride contents in poly(styrene-*co*-4-vinylbenzyl chloride) microspheres.

Azidomethyl-functional Microsphere Characterization

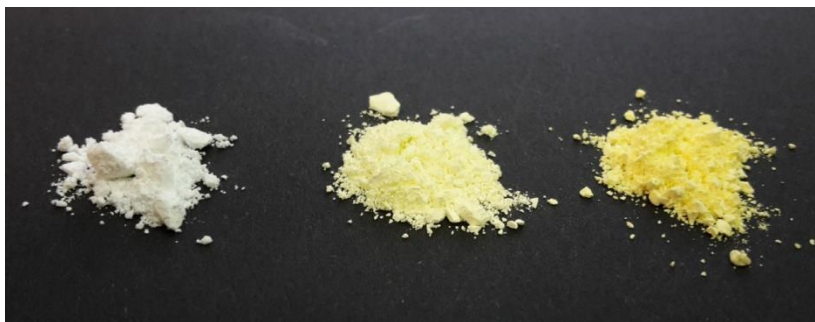


Figure S4. Photographs of microsphere samples MS₁ (left), MS₅₀ (middle), and MS₁₀₀ (right) showing powders of white to yellow color depending on VBA content.

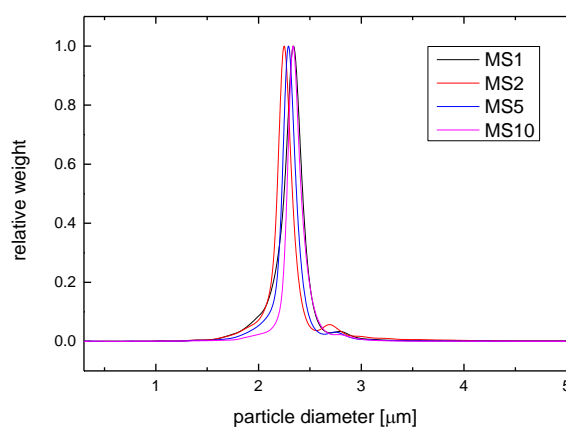


Figure S5. Disc centrifuge results of samples MS₁ (black), MS₂ (red), MS₅ (blue), and MS₁₀ (pink) giving particle diameters around 2.5 μm.

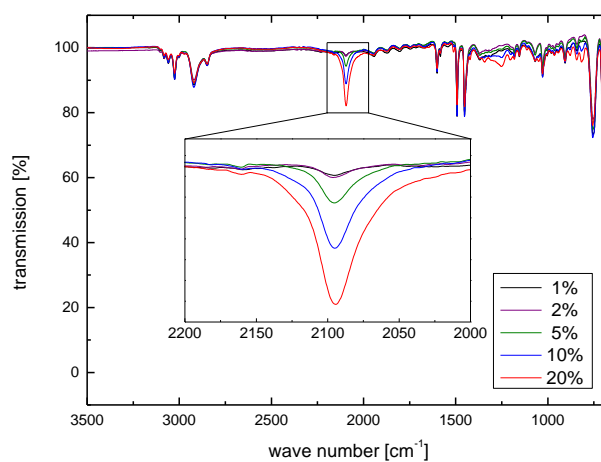


Figure S6. IR spectra of P(S-*co*-VBA) microspheres with VBA feed ranging from 1–20 wt-% with the N=N=N asymmetric stretch band inset

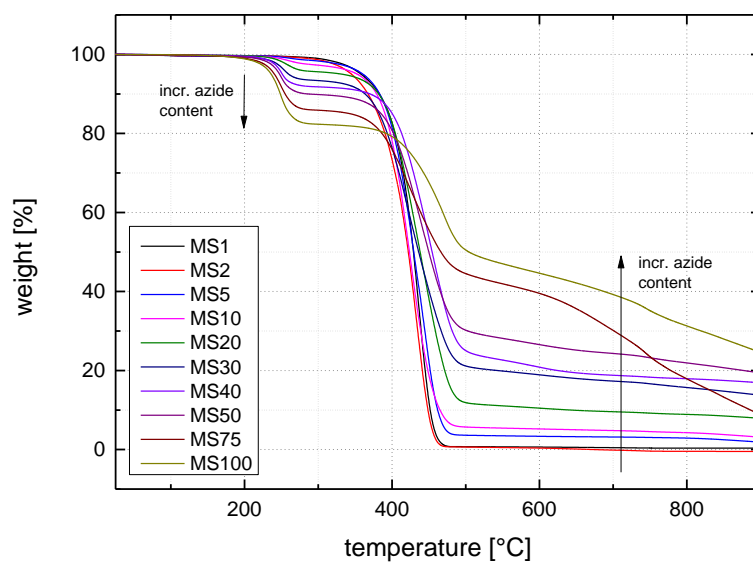


Figure S7. Thermogravimetric analysis of azide-functional microspheres showing the azide-to-nitrene reaction around 250 °C and the decomposition of the polymer material as of ~ 350 °C.

Photo-crosslinking

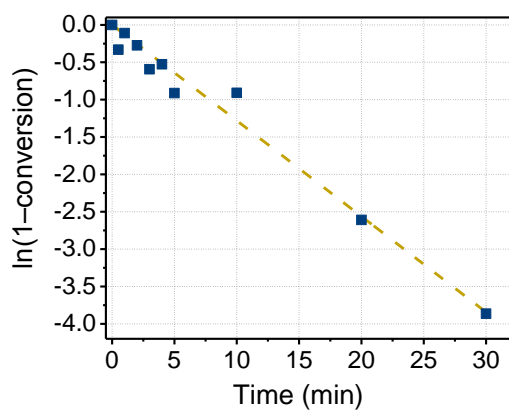


Figure S8. First order kinetic fit of the IR-determined azide conversion during the irradiation of sample MS₁₀₀.

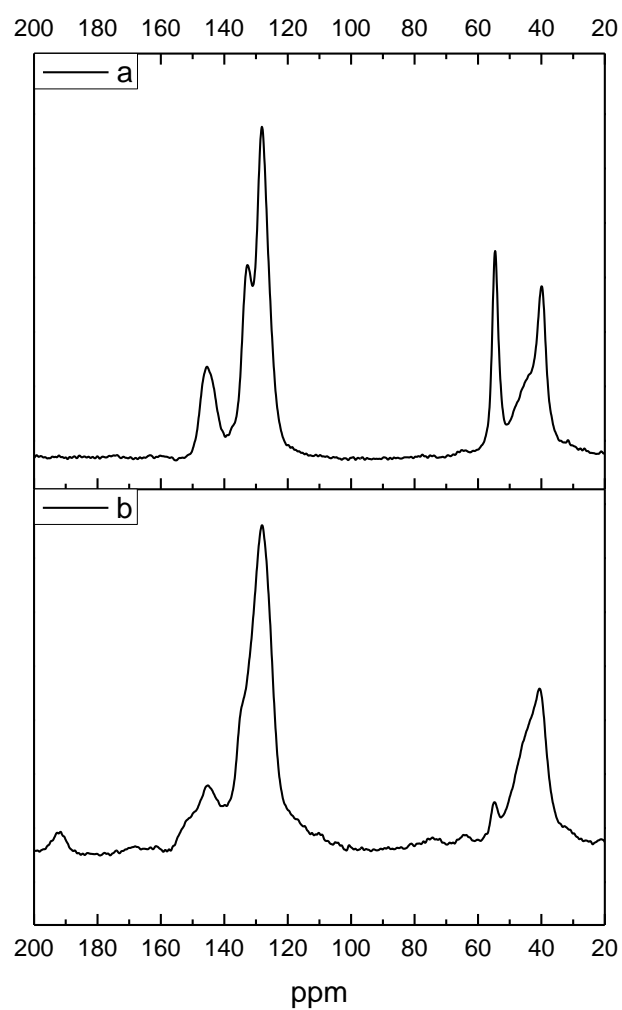


Figure S9. ^{13}C solid state NMR spectra of microspheres MS_{100} before (top) and after (bottom) exhaustive photo-crosslinking.