

Supporting Information

A New Route to Achieve Side-Chain-Type Sulfone-Containing Polysiloxanes via Sulfide Oxidation-Induced Cleavage and Rearrangement of Si–O–Si Bonds with Fine Selectivity Toward Cyclosiloxanes

Zhiming Gou, Yujing Zuo, Jinwan Qi, Zhaoyue Li, and Shengyu Feng*

Key Laboratory of Special Functional Aggregated Materials & Key Laboratory of Colloid and Interface Chemistry (Shandong University), Ministry of Education; School of Chemistry and Chemical Engineering, Shandong University, Jinan 250100, P. R. China

Corresponding author: Shengyu Feng

Fax: +86-531-88564464;

Tel: +86-531-88364866;

E-mail: fsy@sdu.edu.cn

Experimental

Materials

2,2-Dimethoxy-2-phenylacetophenone (DMPA) and Oxone[®] were purchased from Aladdin Co. (China) and used as received. Tetramethyltetravinylcyclotetrasiloxane (D₄^{Vi}) was obtained as commercial products and used directly. Ethyl mercaptan and Methyl-mercaptopropionate were provided by Sigma–Aldrich and used as received. Tetrahydrofuran (THF) was purified according to routine procedure and distilled over sodium before use. Octavinylsilsesquioxane (OVS) was synthesized using the classic procedure.

Characterization and measurements

The reaction was irradiated by UV on a Spectroline Model SB–100P/FA lamp (365 nm, 100 w). UV intensity is 4500 μ W/cm² at a distance of 38 cm. Proton nuclear magnetic resonance (¹H NMR, ¹³C NMR, ²⁹Si NMR) spectra were recorded on a Bruker AVANCE 400 spectrometer at 25 °C using CDCl₃ as solvent and without tetramethylsilane as an interior label. Fourier transform infrared spectra (FT–IR) were recorded on a Bruker TENSOR27 infrared spectrophotometer using the KBr pellet technique within the 4000 cm⁻¹ to 400 cm⁻¹ region. High-resolution mass spectra were obtained using positive mode on Agilent Technologies 6510 Q-TOF LC-MS. The luminescence (excitation and emission) spectra of the samples were determined with a Hitachi F–4500 fluorescence spectrophotometer using a monochromated Xe lamp as an excitation source. Excitation and emission slits measured 5 nm and 2.5 nm, respectively. DSC measurements were studied using SDTQ 600 of TA Instruments. The heating and cooling temperature ramp rates were 10 °C/min. Thermogravimetric

analysis (TGA) was performed under N₂ using a TA SDTQ600 at a temperature range of room temperature to 800 °C with a heating rate of 10 °C/min.

Synthesis of **D4-1**

A cooled oven-dried 25 mL glass vessel was charged with D₄^{Vi} (0.86 g, 2.5 mmol), Ethyl mercaptan (0.62 g, 10 mmol), and DMPA (1 wt%, 0.02 g), and then capped. The vessel was placed under a 100 W UV light ($\lambda_{\text{max}} = 365$ nm). The reaction mixture was irradiated for 20 min with gentle stirring to yield **D4-1**. After the thiol-ene reaction, the product was purified by precipitation in methanol to eliminate photoinitiator. **D4-1** was obtained as a colorless liquid with a yield of 98%.

Data of **D4-1**:

¹H NMR (400 MHz, CDCl₃): 0.10-0.25 (SiCH₃), 0.86-0.97 (SiCH₂CH₂), 1.29-1.37 (SCH₂CH₃), 2.61-2.72 (CH₂SCH₂CH₂Si). ¹³C NMR (100.62 MHz, CDCl₃, ppm): $\delta =$ 0.58, (SiCH₃), 9.36 (SCH₂CH₃), 19.28, (-SiCH₂CH₂), 28.13, (SCH₂CH₂Si), 30.21, (SCH₂CH₃).

D4-2 was obtained as the same way to **D4-1**, yield 93%.

Data of **D4-2**:

¹H NMR (400 MHz, CDCl₃): 0.13-0.25 (SiCH₃), 0.93-1.01 (SiCH₂CH₂), 2.59-2.64 (-CH₂SCH₂-), 2.79-2.82 (-SCH₂CH₂COOCH₃), 3.71 (-COOCH₃). ¹³C NMR (100.62 MHz, CDCl₃, ppm): $\delta =$ 0.03 (-SiCH₃), 17.79 (-SiCH₂CH₂S-), 26.48 (-SiCH₂CH₂S-), 28.83 (-SCH₂CH₂COOCH₃), 34.47 (-CH₂COOCH₃), 51.68 (-COOCH₃), 171.23 (-COOCH₃).

D4-3 was obtained as the same way to **D4-1**, yield 90%.

Data of **D4-3**:

^1H NMR (400 MHz, CDCl_3): 0.13-0.25 (SiCH_3), 0.90-0.95 (SiCH_2CH_2), 2.65-2.68 ($\text{SiCH}_2\text{CH}_2\text{SCH}_2-$), 2.70-2.83 ($-\text{SCH}_2\text{CH}_2\text{COOH}$), 11.64 ($-\text{COOH}$). ^{13}C NMR (100.62 MHz, CDCl_3 , ppm): δ = 0.05 ($-\text{SiCH}_3$), 18.10 ($-\text{SiCH}_2\text{CH}_2\text{S}$), 25.45 ($-\text{SiCH}_2\text{CH}_2\text{S}$), 26.60 ($-\text{SCH}_2\text{CH}_2\text{COOH}$), 34.15 ($-\text{CH}_2\text{COOH}$), 172.17 ($-\text{COOH}$).

Data of **T1**:

^1H NMR (400 MHz, CDCl_3): 0.004 (s, 18H, SiCH_3), 0.58-0.63 (m, 4H, SiCH_2CH_2), 1.55-1.64 (m, 4H, SiCH_2CH_2), 2.55-2.58 (t, 4H, $-\text{SiCH}_2\text{CH}_2\text{CH}_2\text{SCH}_2-$), 2.72 (s, 4H, $-\text{SiCH}_2\text{CH}_2\text{CH}_2\text{SCH}_2-$).

Synthesis of POSS-1

In cold glass vessel, OVS (1.0g, 1.58 mmol), Ethyl mercaptan (0.79 g, 12 mmol) and 2 wt% of DMPA were dissolved in 10 ml THF. The mixture was irradiated with stirring for 15 min with UV light at room temperature and then concentrated under reduced pressure. The crude product was washed by CH_3OH and the residual solvent was removed under vacuum overnight at room temperature. POSS-1 was obtained as a white powder with a yield of 96%.

Data of **POSS-1**:

^1H NMR (400 MHz, CDCl_3 , ppm): δ = 1.06-1.15 (t, $-\text{SiCH}_2-$, 16H), 1.35-1.39 (t, SCH_2CH_3 , 24H), 2.63-2.70 (t, $-\text{SiCH}_2\text{CH}_2-$, 16H), 2.71-2.77 (t, $-\text{SCH}_2\text{CH}_3$, 16H), ^{13}C NMR (100 MHz, CDCl_3 , ppm): 12.41 ($-\text{SiCH}_2-$), 16.80 (SCH_2CH_3), 26.04 ($-\text{SiCH}_2\text{CH}_2\text{S}$), 27.09 ($-\text{SCH}_2\text{CH}_3$). ^{29}Si NMR (79 Hz, CDCl_3 , ppm): δ = -68.30.

POSS-2 was obtained as the same way to **D4-1**, yield 93%.

Data of **POSS-2**:

^1H NMR (400 MHz, CDCl_3 , ppm): δ = 1.01-1.06 (t, $-\text{SiCH}_2-$, 16H), 2.60-2.65 (m, $-\text{SiCH}_2\text{CH}_2\text{SCH}_2-$, 32H), 2.79-2.82 (t, $-\text{SCH}_2\text{CH}_2\text{COOCH}_3$, 16H), 3.70 (s, $-\text{COOCH}_3$, 16H), ^{13}C NMR (100 MHz, CDCl_3 , ppm): 12.87 ($-\text{SiCH}_2-$), 26.04 ($-\text{SiCH}_2\text{CH}_2\text{S}-$), 33.09 ($-\text{SiCH}_2\text{CH}_2\text{S}-$), 52.40 ($-\text{SCH}_2\text{CH}_2\text{CO}-$), 52.68 ($-\text{COOCH}_3$), 172.67 ($-\text{COOCH}_3$). ^{29}Si NMR (79 Hz, CDCl_3 , ppm): δ = -68.54.

Synthesis of **P1-ox.** (post-oxidation of **D4-1**)

D4-1 (7.40g, 5.00 mmol) was dissolved in 20 mL THF, set under an argon atmosphere and cooled with an ice bath. Subsequently, 8.00 g oxone (26.0 mmol) was added in and the reaction mixture was stirred for 12 hours slowly warming to room temperature. Another portion of oxone (6.0 mmol) was added and the reaction mixture was stirred for further 12 hours. Afterwards, reaction mixture was filtered, 100 mL trichloromethane was added and the reaction mixture was washed with saturated NaCl solution (3×40 mL). The aqueous phase was again extracted with trichloromethane and the combined organic layers were dried over Na_2SO_4 . The solution was concentrated and the product was purified by precipitation in petroleum ether. **P1-ox** was obtained as a viscous liquid with a yield of 55%.

Data of **P1-ox**:

^1H NMR (400 MHz, CDCl_3): 0.24-0.35 (SiCH_3), 1.16-1.20 (SiCH_2CH_2), 1.30-1.34 (SCH_2CH_3), 2.79-2.94 ($\text{SO}_2\text{CH}_2\text{CH}_2\text{Si}$), 2.94–3.18 ($\text{SO}_2\text{CH}_2\text{CH}_3$). ^{13}C NMR (100.62 MHz, CDCl_3 , ppm): δ = -0.58, (SiCH_3), 6.44, (SCH_2CH_3), 25.49, ($-\text{SiCH}_2\text{CH}_2$), 46.55, ($\text{SO}_2\text{CH}_2\text{CH}_2\text{Si}$), 67.80, ($\text{SO}_2\text{CH}_2\text{CH}_3$).

P2-ox was obtained using the same procedure to **P1-ox** as a viscous liquid with a yield of 48%.

Data of **P2-ox**:

^1H NMR (400 MHz, CDCl_3): 0.26 (SiCH_3), 1.14-1.17 (SiCH_2CH_2), 2.89-2.91 ($-\text{SiCH}_2\text{CH}_2\text{SO}_2-$), 3.01-3.06 ($-\text{SO}_2\text{CH}_2\text{CH}_2\text{COOCH}_3$), 3.34-3.39, ($-\text{SO}_2\text{CH}_2\text{CH}_2\text{COOCH}_3$), 3.75 ($-\text{COOCH}_3$). ^{13}C NMR (100.62 MHz, CDCl_3 , ppm): δ = 0.90 ($-\text{SiCH}_3$), 25.49 ($-\text{SiCH}_2\text{CH}_2\text{SO}_2-$), 26.72 ($-\text{CH}_2\text{COOCH}_3$), 51.68 ($-\text{SiCH}_2\text{CH}_2\text{SO}_2-$), 67.79 ($-\text{SO}_2\text{CH}_2\text{CH}_2\text{COOCH}_3$), 76.94 ($-\text{COOCH}_3$), 171.03 ($-\text{COOCH}_3$).

Data of **P3-ox**:

^1H NMR (400 MHz, CDOD_3): 0.26-0.28 (SiCH_3), 1.12-1.29 (SiCH_2CH_2), 2.79-2.82 ($-\text{SiCH}_2\text{CH}_2\text{SO}_2-$), 3.11-3.12 ($-\text{SO}_2\text{CH}_2\text{CH}_2\text{COOCH}_3$), 3.30-3.41, ($-\text{SO}_2\text{CH}_2\text{CH}_2\text{COOH}$). ^{13}C NMR (100.62 MHz, CDOD_3 , ppm): δ = 0.95 ($-\text{SiCH}_3$), 24.35 ($-\text{SiCH}_2\text{CH}_2\text{SO}_2-$), 31.12 ($-\text{CH}_2\text{COOH}$), 56.38 ($-\text{SiCH}_2\text{CH}_2\text{SO}_2-$), 62.51 ($-\text{SO}_2\text{CH}_2\text{CH}_2\text{COOH}$), 170.18 ($-\text{COOCH}$).

Data of **M1**:

^1H NMR (400 MHz, CDCl_3): 0.04 (s, 18H, SiCH_3), 0.62-0.67 (m, 4H, SiCH_2CH_2), 1.85-1.93 (m, 4H, SiCH_2CH_2), 3.09-3.13 (t, 4H, $-\text{SiCH}_2\text{CH}_2\text{CH}_2\text{SO}_2\text{CH}_2-$), 3.44 (s, 4H, $-\text{SiCH}_2\text{CH}_2\text{CH}_2\text{SO}_2\text{CH}_2-$). ^{13}C NMR (100.62 MHz, CDOD_3 , ppm): δ = -1.85 ($-\text{SiCH}_3$), 15.97 ($-\text{SiCH}_2\text{CH}_2\text{SO}_2-$), 17.13 ($-\text{SiCH}_2\text{CH}_2\text{CH}_2\text{SO}_2-$), 44.72 ($-\text{SiCH}_2\text{CH}_2\text{CH}_2\text{SO}_2-$), 57.73 ($-\text{SiCH}_2\text{CH}_2\text{CH}_2\text{SO}_2\text{CH}_2-$). MS-ESI: 409.1388 [$\text{M} + \text{Na}$] $^+$.

POSS-1' was obtained using the same procedure to P1-ox as a white powder with a yield of 50%.

Data of **POSS-1'**:

^1H NMR (400 MHz, CDCl_3 , ppm): δ = 1.24-1.28 (t, $-\text{SiCH}_2-$, 16H), 1.41-1.45 (t, SCH_2CH_3 , 24H), 3.05-3.11 (m, $-\text{SiCH}_2\text{CH}_2\text{SCH}_2$, 32H). ^{13}C NMR (100 MHz, CDCl_3 , ppm): 14.27 ($-\text{SiCH}_2-$), 17.93 ($\text{SO}_2\text{CH}_2\text{CH}_3$), 48.31 ($-\text{SiCH}_2\text{CH}_2\text{SO}_2-$), 50.13 ($-\text{SO}_2\text{CH}_2\text{CH}_3$). ^{29}Si NMR (79 Hz, CDCl_3 , ppm): δ = -68.80 ppm. MS-ESI: 1406.9970 $[\text{M} + \text{Na}]^+$, 1385.0198 $[\text{M} + \text{H}]^+$.

POSS-2' was obtained using the same procedure to P1-ox as a light yellow liquid with a yield of 57%.

Data of **POSS-2'**:

^1H NMR (400 MHz, CDCl_3 , ppm): δ = 1.25-1.29 (t, $-\text{SiCH}_2-$, 16H), 2.86-2.90 (m, $-\text{SiCH}_2\text{CH}_2\text{SO}_2-$, 16H), 3.11-3.15 (m, $-\text{SO}_2\text{CH}_2\text{CH}_2\text{COOCH}_3$, 16H), 3.35-3.38 (m, $-\text{SO}_2\text{CH}_2\text{CH}_2\text{COOCH}_3$, 16H), 3.73 (s, $-\text{COOCH}_3$, 16H). ^{13}C NMR (100 MHz, CDCl_3 , ppm): 14.81 ($-\text{SiCH}_2-$), 39.01 ($-\text{SiCH}_2\text{CH}_2\text{SO}_2-$), 45.55 ($-\text{SO}_2\text{CH}_2\text{CH}_2\text{CO}-$), 49.07 ($-\text{SO}_2\text{CH}_2\text{CH}_2\text{CO}-$), 51.87 ($-\text{COOCH}_3$), 172.68 ($-\text{COOCH}_3$). ^{29}Si NMR (79 Hz, CDCl_3 , ppm): δ = -68.61. MS-ESI: 1868.0706 $[\text{M} + \text{NH}_4]^+$.

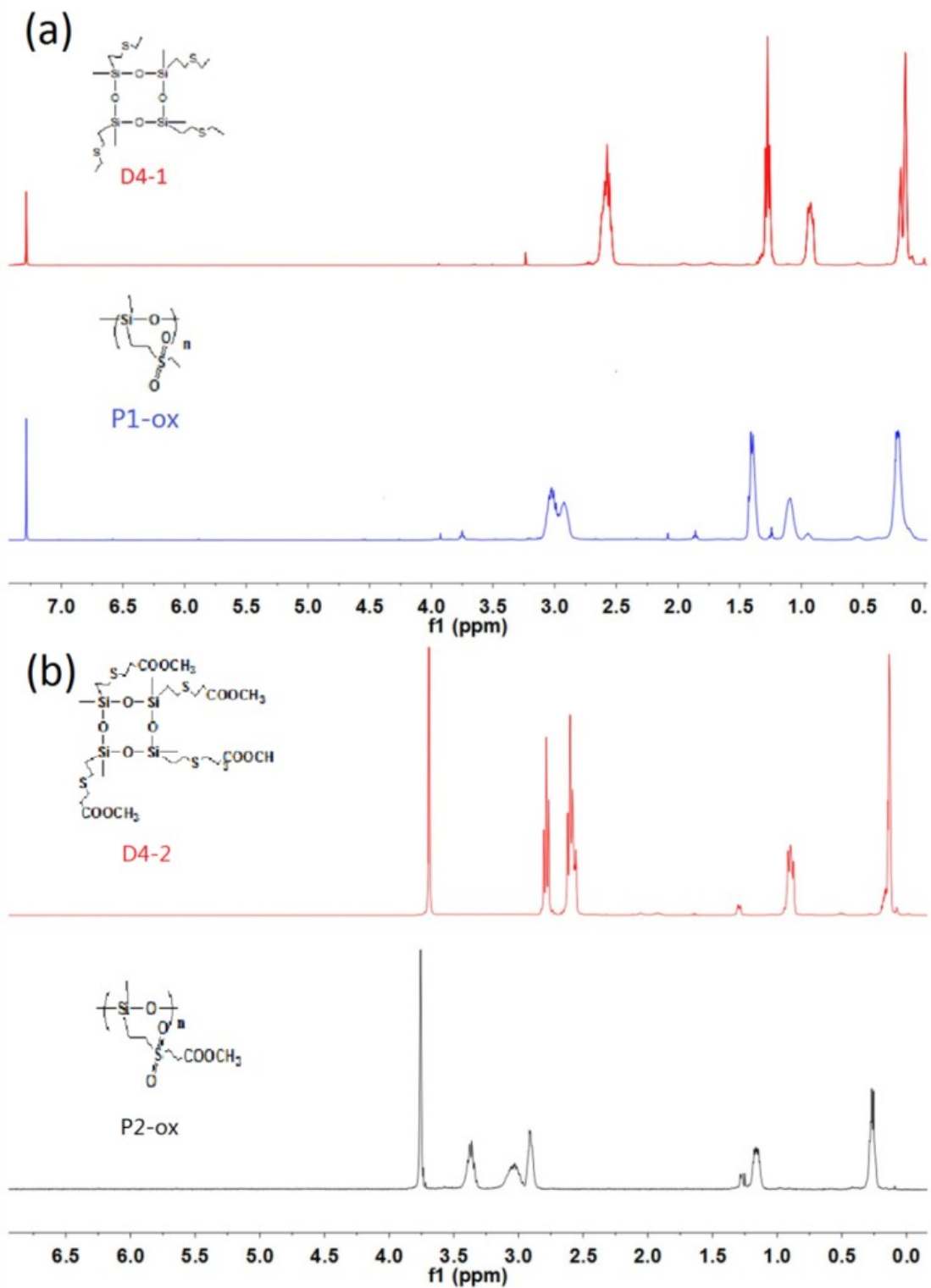


Figure S1. ^1H NMR spectra of **D4-1** and **P1-ox** (a) and **D4-2** and **P2-ox** (b).

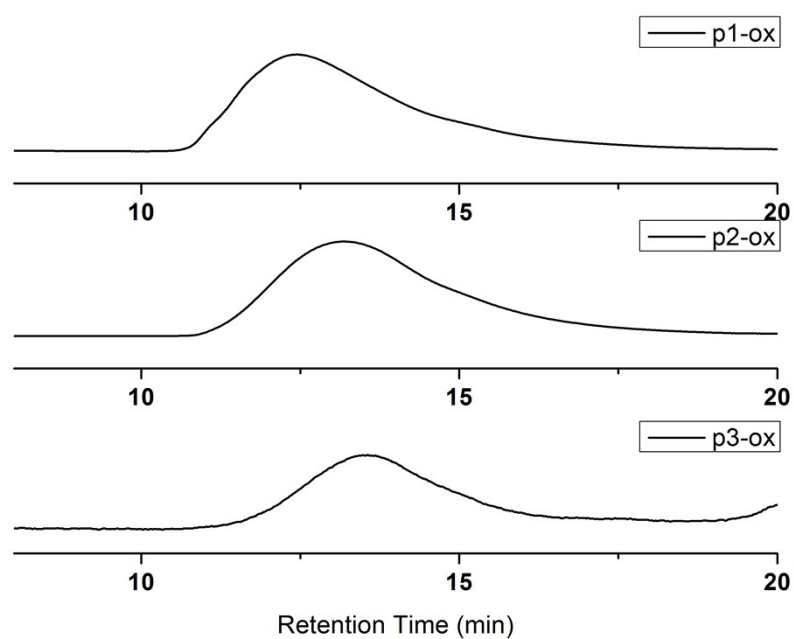


Figure S2. GPC traces of **P1-ox**, **P2-ox** and **P3-ox**.

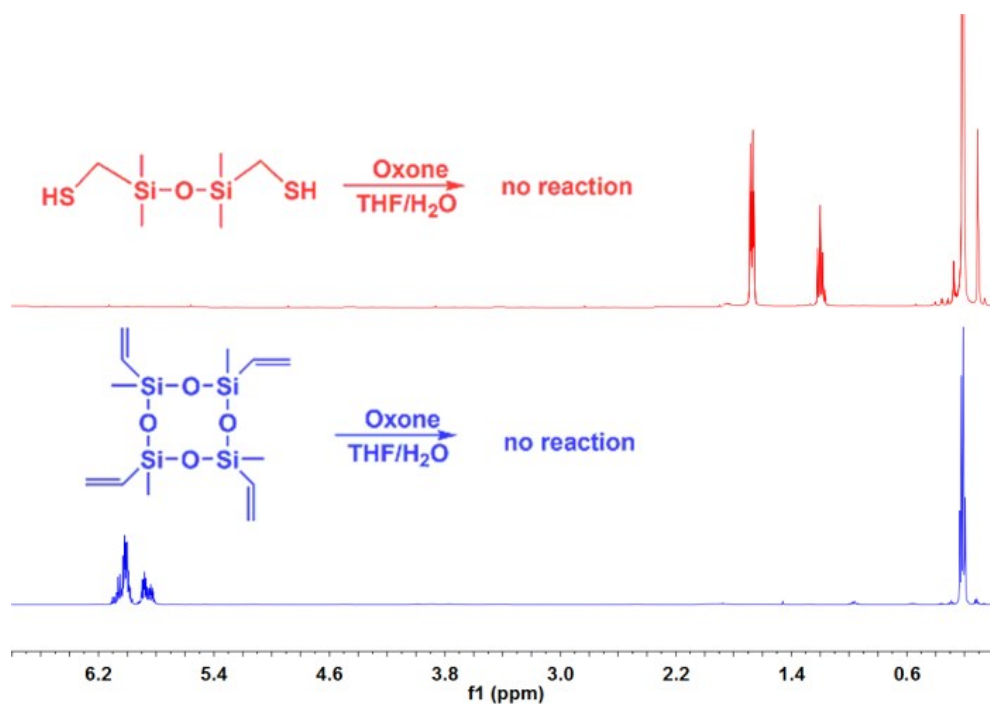
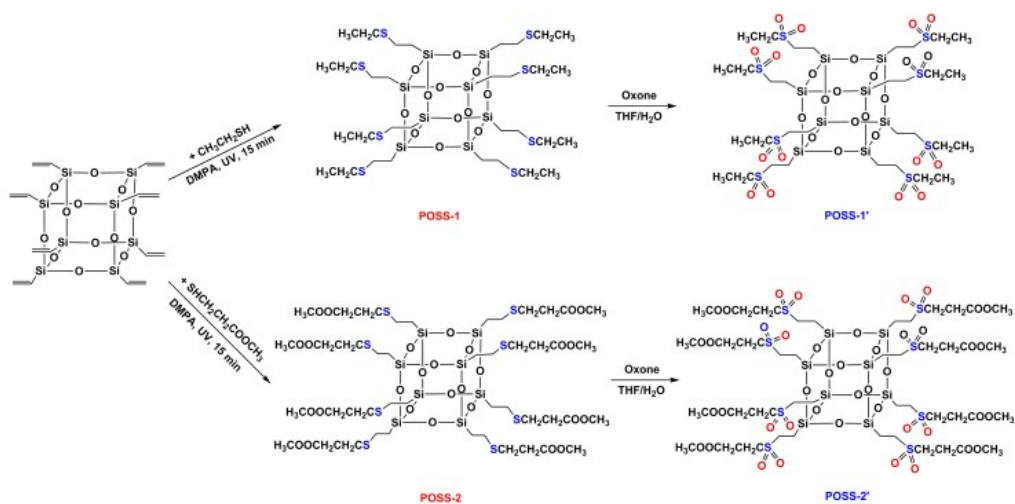


Figure S3. Control experiments of MM^{SH} and D_4^{Vi} , and the ^1H NMR spectra of the products obtained after oxidation.



Scheme S1. Synthesis routes of **POSS-1**, **POSS-2**, **POSS-1'**, and **POSS-2'**.

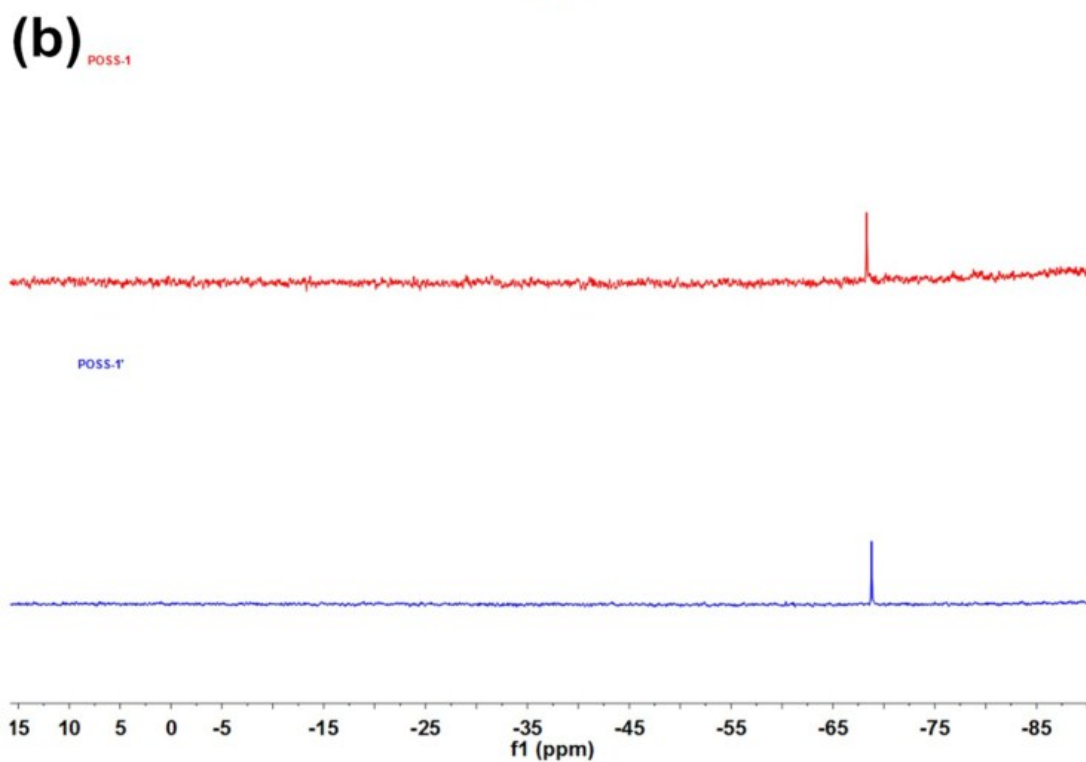
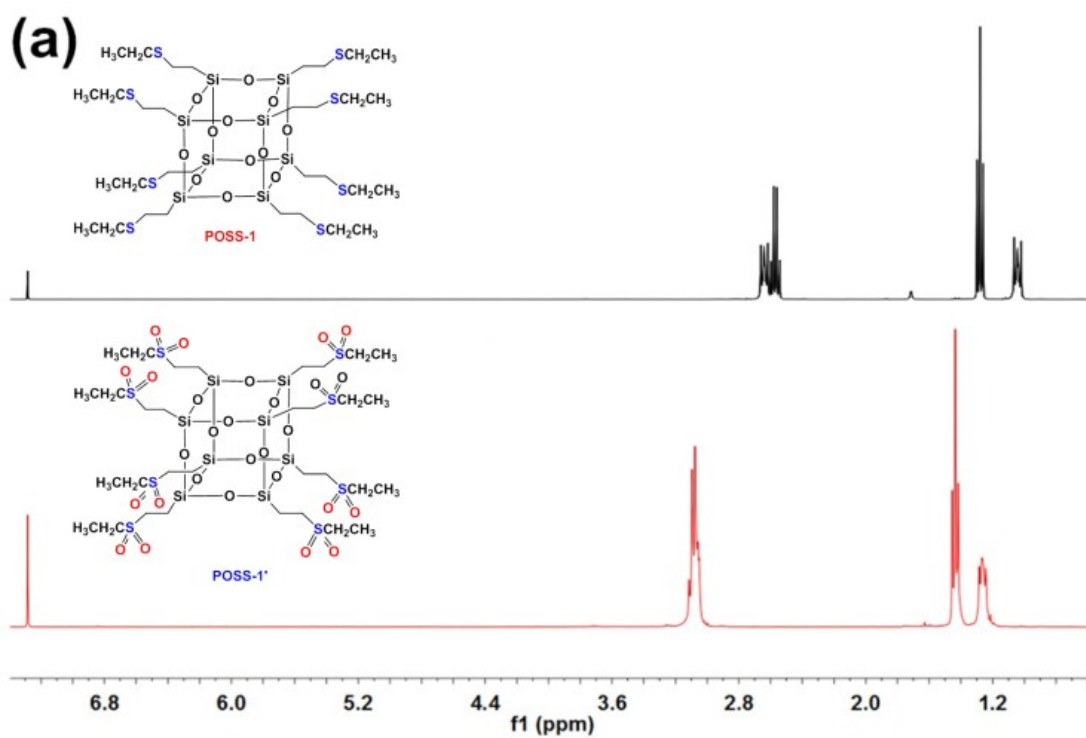


Figure S4. ¹H NMR spectra of **POSS-1** and **POSS-1'** (a) and ²⁹Si NMR spectra of **POSS-1** and **POSS-1'** (b).

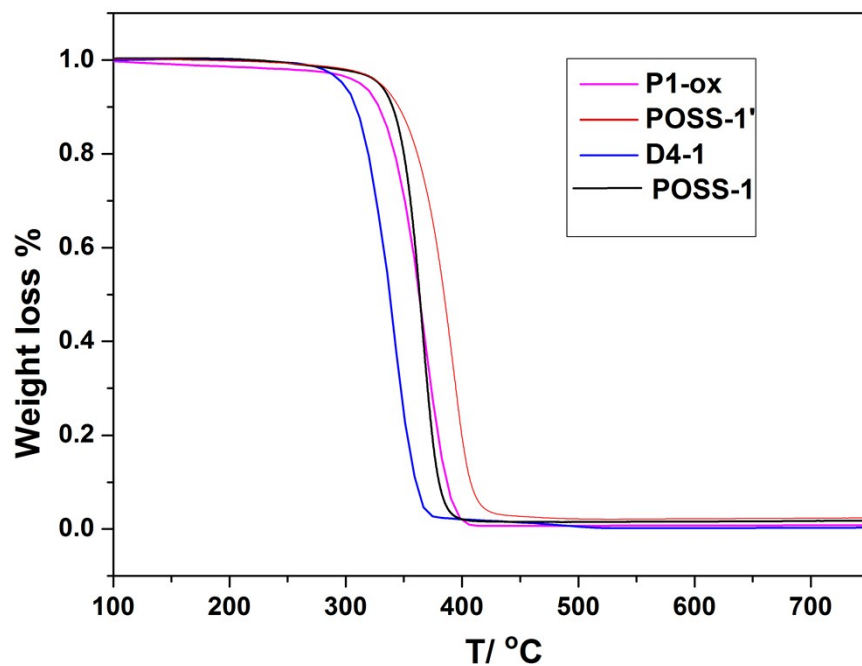


Figure S5. TGA curves of the obtained moleculars.

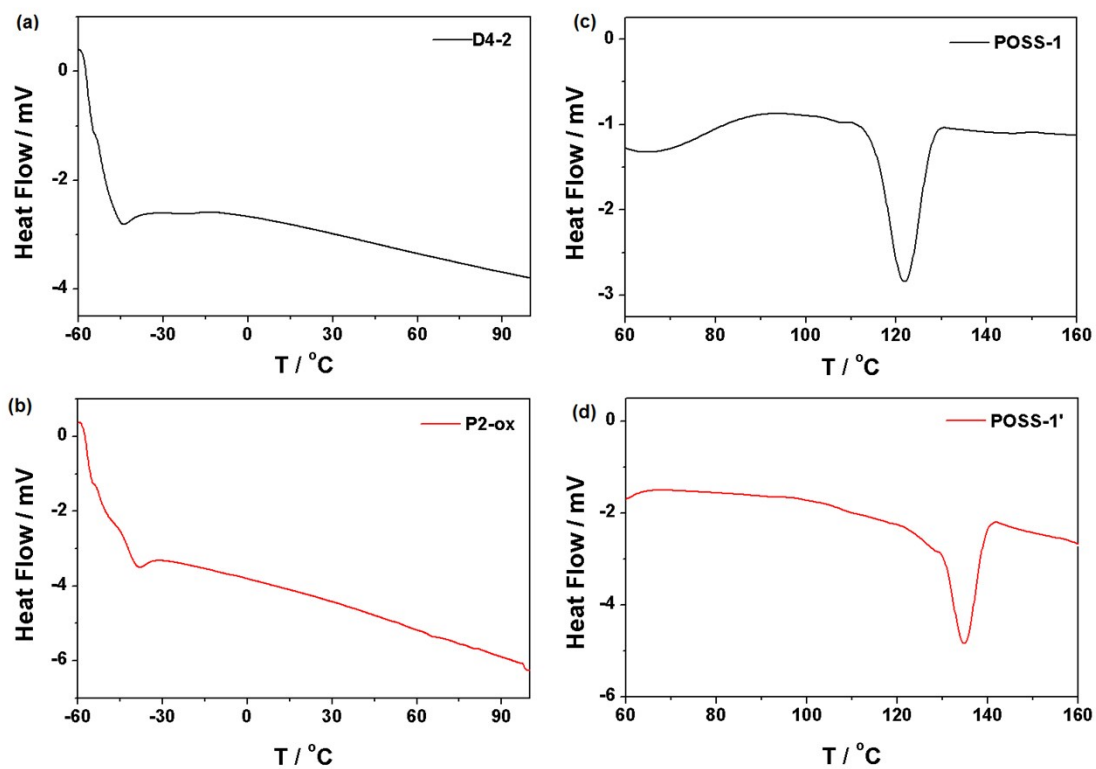


Figure S6. DSC curves of **D4-2** (a), **P2-ox** (b), **POSS-1** (c), and **POSS-1'**(d).