Supporting Information

A strategy for constructing anti-adhesion surfaces based on interfacial thiol-ene photoclick chemistry between DOPA derivatives with a catechol anchor group and zwitterionic betaine macromolecules

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Synthesis of N,N'-Bis(acryloyl)cystamine (BAC)

N,N'-Bis(acryloyl)cystamine (BAC) was synthesized similar to the method established by E. Elisa *et al.* Cystamine dihydrochloride (CADH) (11.60 g, 0.05 mol) was added to a four-necked, 500 mL flask containing deionized water (50 mL) which was cooled to 0 - 5 °C. Then acryloyl chloride (AC) dissolved in dichloromethane solution (10 mL, 10 mol/L) and an aqueous NaOH solution (20 mL, 10 mol/L) were added simultaneously and dropwise under stirring over a total time span of 1 h. Meanwhile, the temperature of the reaction system was kept at 0 - 5 °C. After the addition was completed, the reaction mixture was stirred for more than 2 h at room temperature and filtrated. The filtrate was extracted three times with 100 mL of dichloromethane and the organic phases were dried over MgSO₄ and filtrated. Finally, the solvent was removed in a rotary evaporator at 40 °C, so obtaining the white powdery solid product. Yield: 60 %.

FTIR (KBr pellet, cm⁻¹): 3252 (N–H, stretch); 3067 (=C–H, stretch); 2925, 2855 (–CH₃, –CH₂–, stretch); 1667 (C

=O, stretch); 1653 (C=C, stretch); 1556 (N–H, in-plane bend); 1466 (–CH₂–, deformation); 1405 (–CH₃, deformation), 1311 (C–H, in-plane bend); 1254 (–CH₃, deformation); 696 (N–H, out-of-plane bend). (ESI, Fig. S1 $^{+}$)



Fig. S1 FTIR spectrum of BAC.

Synthesis of N,N-dimethyl-N-(3-methacrylamidopropyl)-N-(3-sulfopropyl) ammonium betaine (DMAPMAPS)

N,N-dimethyl-N-(3-methacrylamidopropyl)-N-(3-sulfopropyl) ammonium betaine (DMAPMAPS) was synthesized by the ring-opening reaction between DMAPMA and 1,3-PS. A mixture of 1,3-PS (13.18 g, 0.11 mol) and acetone (25 mL) was added dropwise into a three-necked flask containing a mixture of DMAPMA (17.02 g, 0.10 mol) and acetone (25 mL) under stirring over a total time span of 0.5 h. Then the reaction mixture was stirred for more than 24 h at room temperature and filtrated. The residue was washed and vacuum dried at 40 °C, and then the final white powdery product was obtained. Yield: 94%.

FTIR (KBr pellet, cm⁻¹): 3320 (N–H, stretch); 3030 (=C–H, stretch); 2976, 2928, 2877 (–CH₃, –CH₂–, stretch); 1653 (C=O, stretch); 1611 (C=C, stretch); 1543 (N–H, in-plane bend); 1484 (N⁺–C, stretch); 1193 (S=O, deformation); 1034 (S=O, deformation). (ESI, Fig. S2[†])

¹H-NMR δ (400 MHz, D₂O, ppm): 5.66 ppm (s, 1H, CHH=CH–); 5.42 ppm (s, 1H, CHH=CH–); 3.40 ~ 3.44 ppm (t, 2H, SO₃-CH₂-CH₂-CH₂-); 3.29 ~ 3.33 ppm (m, 4H, -NH-CH₂-); 3.05 ppm (s, 6H, -N(CH₃)₂-); 2.90 ~ 2.94 ppm (t, 2H, -CH₂-SO₃); 2.11 ~ 2.19 ppm (m, 2H, -CH₂-CH₂-SO₃); 1.96 ppm (s, 2H, -CO-C(CH₃)=). (ESI, Fig. S3†)











Fig. S4 FTIR spectra of DMA and DOPA.



10.0 9.5 9.0 8.5 8.0 7.5 7.0 6.5 6.0 5.5 5.0 4.5 4.0 3.5 3.0 2.5 2.0 1.5 1.0 0.5 0.0 Chemical shift (ppm)

Fig. S5 ¹H-NMR spectrum of DMA.



Fig. S6 1 H-NMR spectra of poly (DMAPMAPS-*co*-BAC) sample P₁ (a) and P₃ (b).

Proton peak	Chemical shift	Integral area	
	(ppm)	\mathbf{P}_1	P ₃
а	0.96	1.70	1.63
c, e	3.51	1.00	1.00
d, b, k	2.01	2.32	2.17
f	3.15	1.08	3.96
g, 1	3.39	3.57	1.16
h	2.23	1.13	1.14
i	3.01	1.26	1.38
m, j	2.72	0.14	0.07

Table S1 The data on integral areas of ${}^{1}H$ NMR spectra of P_{1} and P_{3} .