

Supporting Information

Functionalized Microporous Organic Nanotube Networks as a New Platform for Highly Efficient Heterogeneous Catalysis

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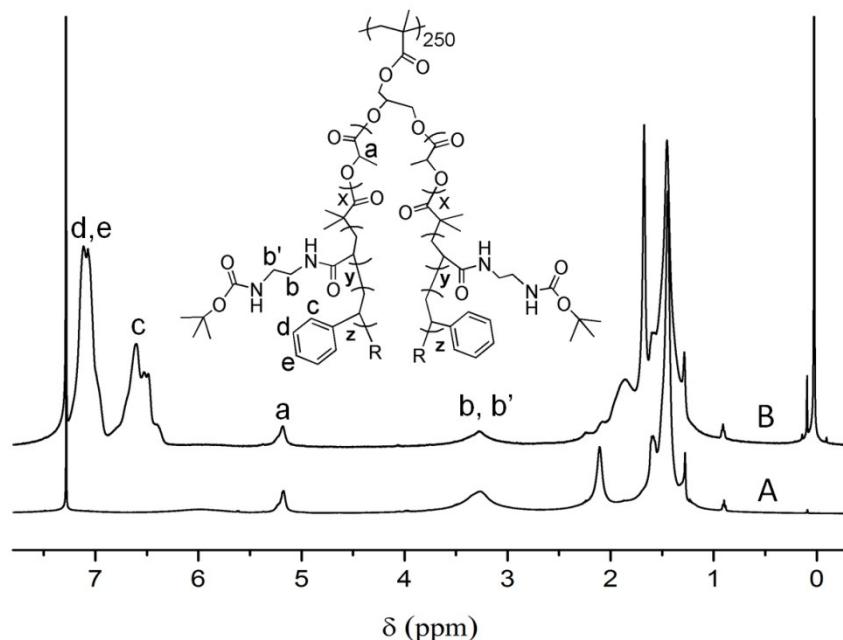


Figure S1. ^1H NMR (CDCl_3) spectra of Poly(GM-g-LA-PBAEAM) (A) and Poly(GM-g-LA-PBAEAM-g-St) (B).

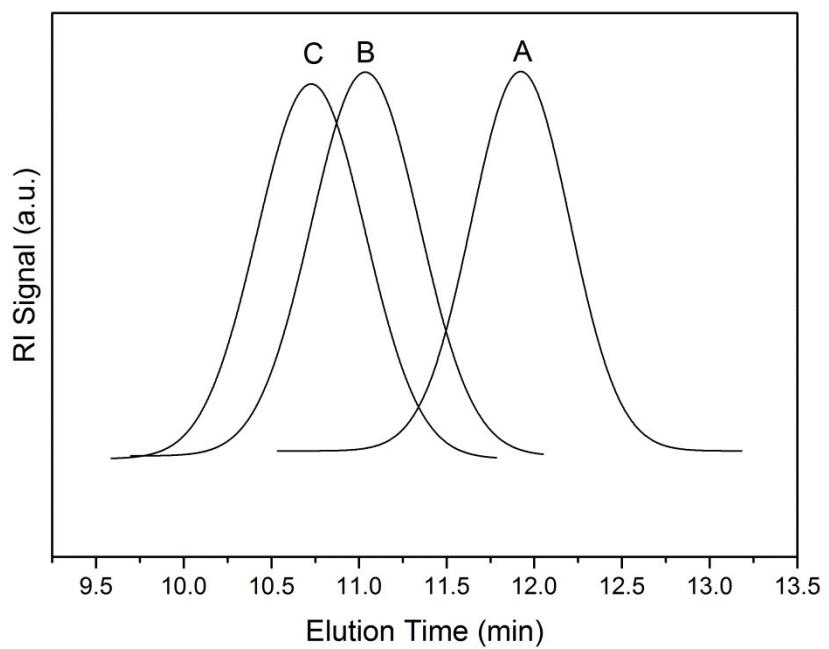


Figure S2. GPC characterization of Poly(GM-g-LA-TC) (A), Poly(GM-g-LA-PBAEAM) (B) and Poly(GM-g-LA-PBAEAM-g-St) (C).

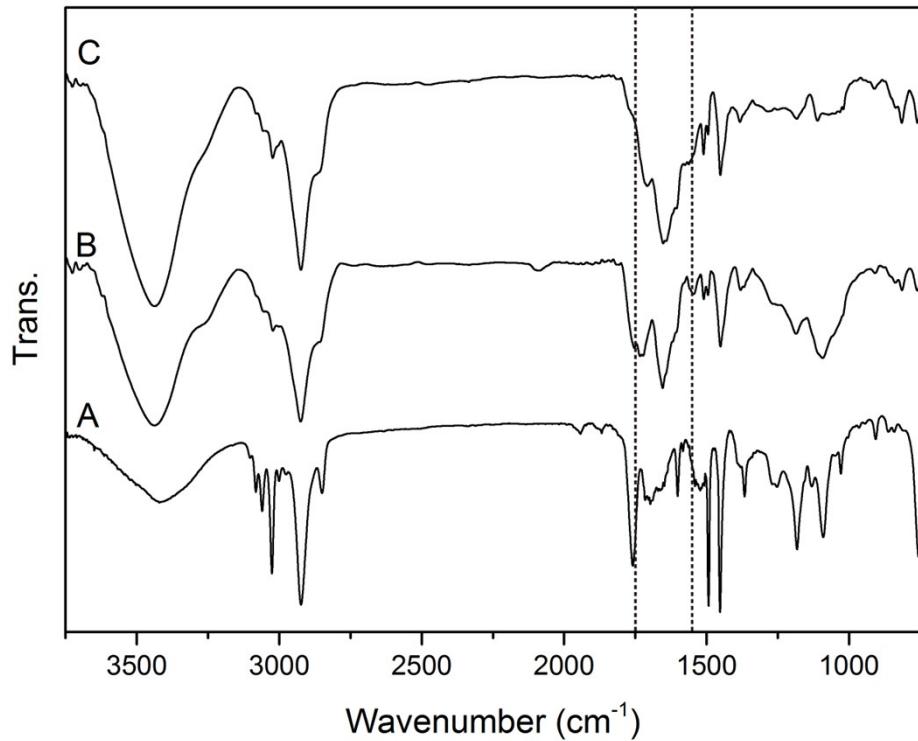


Figure S3. IR characterization of bottlebrush copolymer Poly(GM-g-LA-PBAEAM-g-St)(A), after cross-linked (B) and after hydrolysed (C).

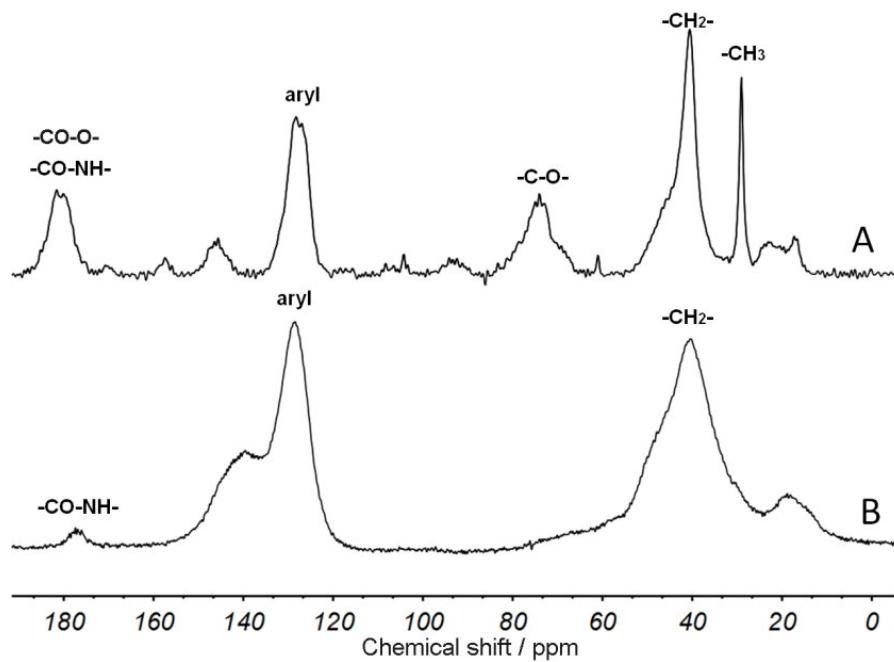


Figure S4. Solid state ^{13}C CP-MAS NMR spectra of the bottlebrush copolymer Poly(GM-g-LA-PBAEAM-g-St) (A) and the NH₂-MONNs (B).

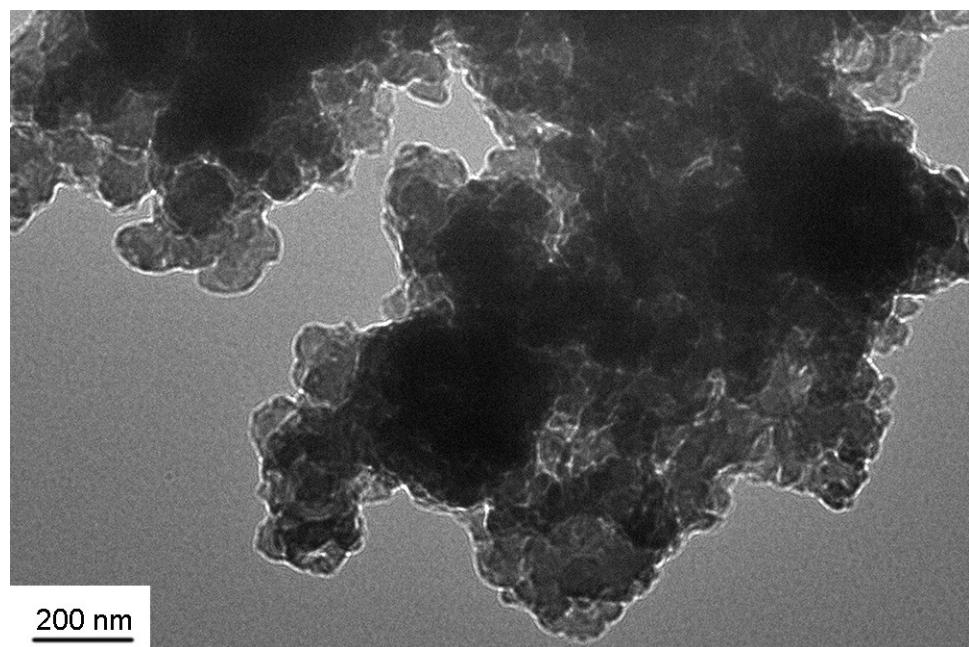


Figure S5.TEM image of the NH₂-OONNs.

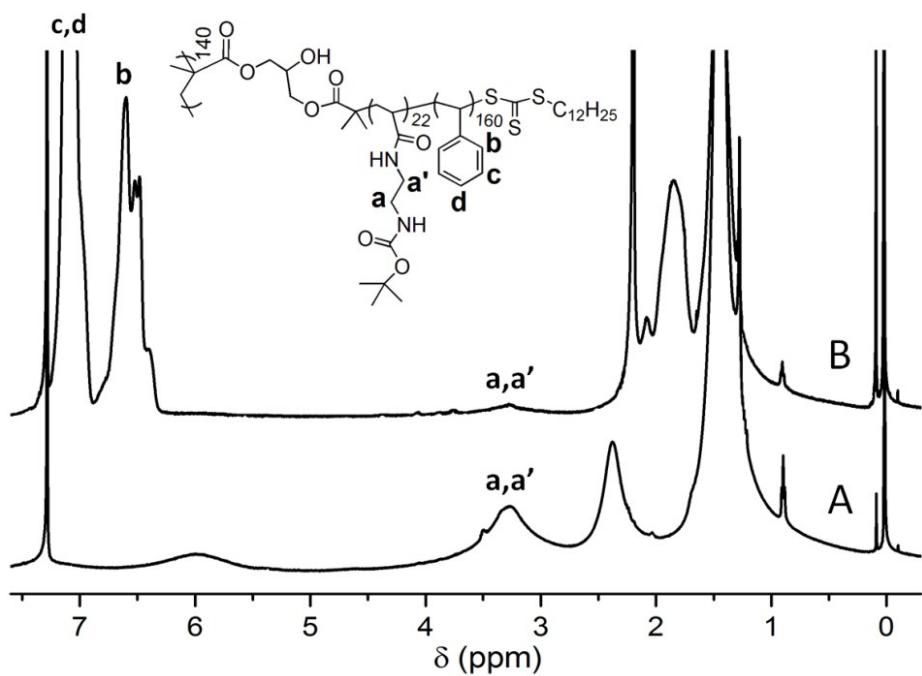


Figure S6. ¹HNMR (CDCl_3) spectra of Poly(GM-g-PBAEAM) (A) and Poly(GM-g-PBAEAM-g-St) (B).

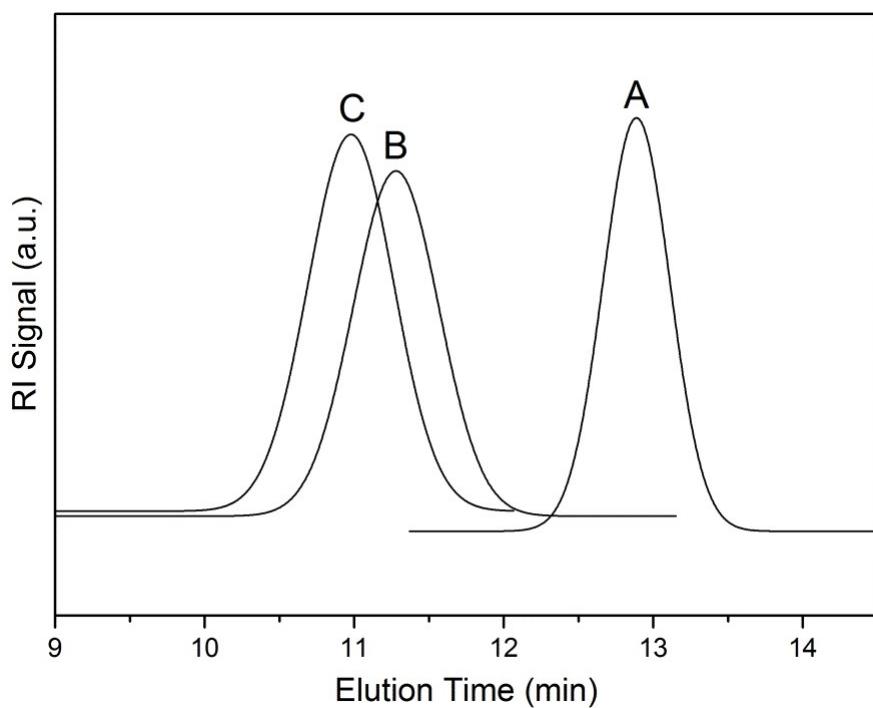


Figure S7. GPC characterization of Poly(GM-g-TC) (A), Poly(GM-g-PBAEAM) (B) and Poly(GM-g-PBAEAM-g-St) (C).

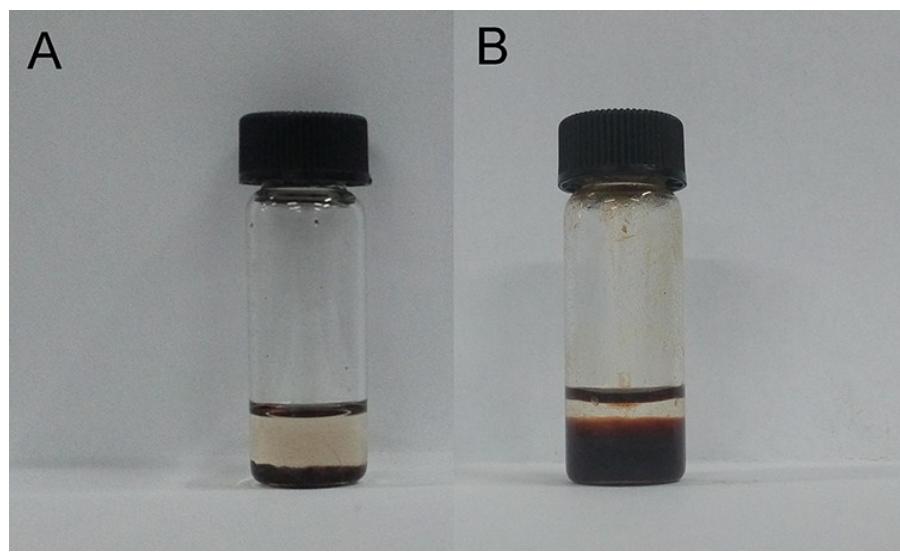


Figure S8. Photographs of the NH₂-MONNs stirred in toluene at 80°C for 0 min (A) and 45 min (B). These photographs indicate that the NH₂-MONNs are remarkably swollen after stirring in toluene solvent, which is attributed to the flexible framework.

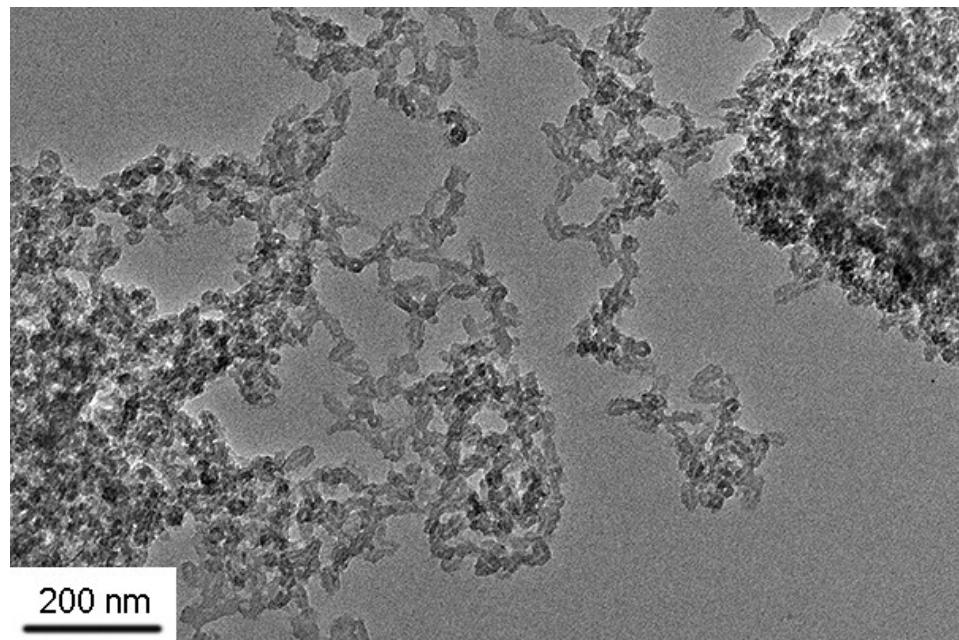


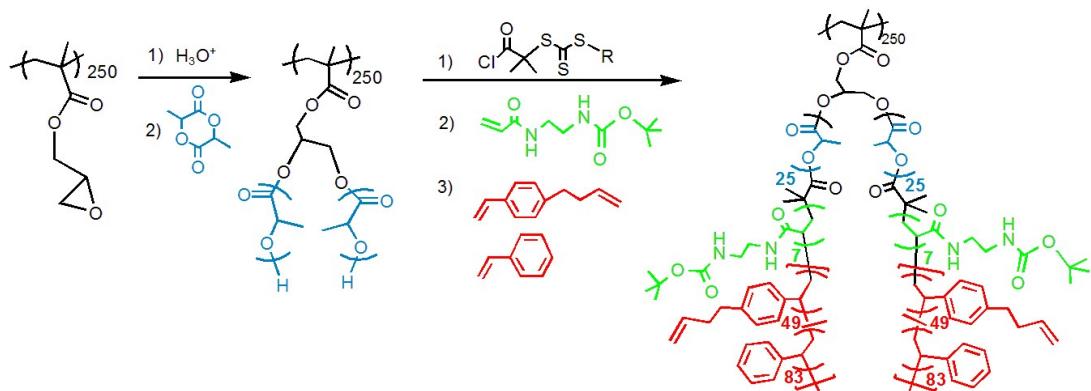
Figure S9. TEM image of the NH₂-MONNs after recycled.

Table S1. Element analysis of NH₂-MONNs (**1**) and the NH₂-OONNs (**2**).

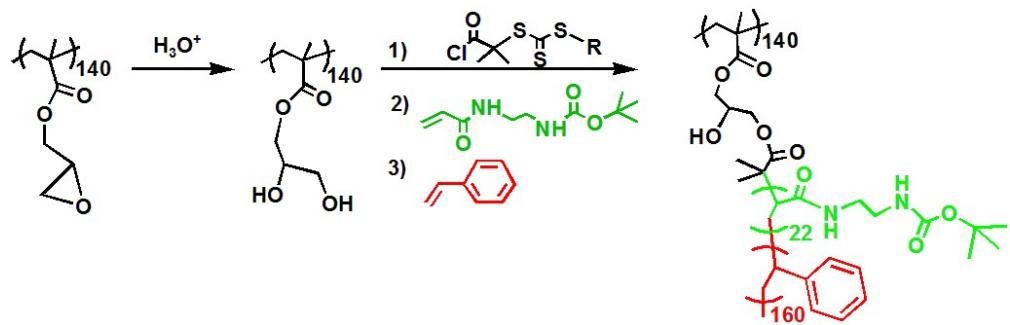
NO.	Wight [mg]	C/N Ratio	Content [%]	Peak Area
1	1.5110	151.7	N: 0.491	231
			C: 74.42	29402
			H: 7.225	8957
2	1.2980	116.6	N: 0.625	253
			C: 72.92	24752
			H: 7.307	7779

Table S2. Textural parameters of the NH₂-MONNs (**1**) and the NH₂-OONNs (**2**).

Materials	BET	Surface	Pore	Volume	Micropore	Mesopore/Macropore
	Area (m ² /g)	(cm ³ /g)		Volume	Volume	
1	936	1.67		0.17	1.50	
2	32	0.27		--	0.27	



Scheme S1. Synthesis route of the NH₂-OONNs precursor.



Scheme S2. Synthesis route of the NH_2 -bottlebrushes precursor.