### **Supporting Information**

#### Trityl-based alkoxyamines as potential NMP controllers and spin-labels.

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Biradical	<i>Т</i> , К	<i>J</i> , G	Δ <i>J</i> , G	biradical lwpp <sup>a,b</sup> , G	free trityl lwpp <sup>a,c</sup> , G	free nitroxide lwpp <sup>a,c</sup> , G	<i>a</i> <sub>n</sub> , G	a <sub>p</sub> , G
1•	273	55	30	1.5	0.45 <sup>d</sup>			
	283	61	24	1.2		1.5	15.4 <sup>e</sup>	_
	293	62	25.5	1.2				
	303	63	30	1.2				
	313	67	30	1				
	323	70	30	1				
	333	70	40	0.8				
	343	75	40	0.8				
	353	75	40	0.8				
3•	283	160	130	1	0.45 <sup>d</sup>			
	293	160	130	1		2	13.6	44.9 <sup>f</sup>
	303	170	130	1				
	313	180	150	1				
	323	200	150	1				
	333	220	150	1				

Table 1SI. Simulation parameters of EPR spectra for Figures 2SI and 3SI

<sup>a</sup> lwpp – peak-to-peak line width; <sup>b</sup> Lorentzian line shape; <sup>c</sup> Gaussian line shape.  ${}^{d}g = 2.00281$ .  ${}^{e}g = 2.00595$ .  ${}^{f}g = 2.0058$ 



**Figure 1SI.** Kinetics of alkoxyamine homolysis:  $\Diamond - 1$  at 398K;  $\Box - 2$  at 353K;  $\circ - 3$  at 373K;  $\Delta - 4$  at 373K, star - 5 at 373K

**Figure 2SI.** The temperature dependence of the EPR signal from the reaction mixture of *2* (after thermolysis at 373K for 5h): full line – experimental spectrum; dotted line – simulated spectrum











**Figure 3SI**. The temperature dependence of the EPR signal from the reaction mixture of *3* (after thermolysis at 373K for 5h): full line – experimental spectrum; dotted line – simulated spectrum





Figure S4.1 <sup>1</sup>H-NMR spectral data for alkoxyamine 6

<sup>13</sup>C{<sup>1</sup>H} NMR spectra (CDCl<sub>3</sub>  $\delta_C$  = 77.0 ppm) Bruker AV-400 (100.61 MHz) of **6** 



Figure S4.2 <sup>13</sup>C-NMR spectral data for alkoxyamine 6

## Figure 5SI. HR MS, <sup>1</sup>H-,<sup>13</sup>C-NMR and FT IR spectral data for alkoxyamine 7



Figure S5.1 HR MS (ESI) spectrum of alkoxyamine 7: calcd. for  $C_{17}H_{33}O_4N$  [M<sup>+</sup>] 315.2404, found 315.2408





Figure S5.2 <sup>1</sup>H-NMR spectral data for alkoxyamine 7



Figure S5.3 <sup>13</sup>C-NMR spectral data for alkoxyamine 7



Figure S5.4 FT IR (KBr) spectrum of alkoxyamine 7

## Figure 6SI. HR MS(ESI), EPR, and FT IR spectral data for trityl 1



Figure S6.1.1 MS(ESI) spectrum of trityl 1: calcd. for C<sub>61</sub>H<sub>70</sub>NO<sub>9</sub>S<sub>12</sub> [M<sup>-</sup>] 1344.1699, found 1344.167



Figure S6.1.2 MS(ESI) spectrum of trityl 1: calcd.  $C_{61}H_{71}NO_9S_{12}$  for [M+H<sup>+</sup>] 1345.1777, found 1345.169



**Figure S6.2.** EPR for 0.5 mM deoxygenated solution in DCM: multiplet,  $\alpha_H$ =9.4 µT, linewidth (Gauss) 8.7 µT, *g* = 2.00280



Figure S6.3 FT IR (KBr) spectrum of trityl 1





**Figure S7.1.1.** HR MS (ESI) spectrum of trityl **2**: calcd. for C<sub>61</sub>H<sub>76</sub>NO<sub>11</sub>S<sub>12</sub> [M<sup>-</sup>] 1382.2067, found 1382.210



**Figure S7.1.2.** HR MS (ESI) spectrum of trityl **2**: calcd. for C<sub>61</sub>H<sub>77</sub>NO<sub>11</sub>S<sub>12</sub> [M+H<sup>+</sup>] 1383.2145, found 1383.217



**Figure S7.2.** EPR for 0.5 mM deoxygenated solution in DCM: multiplet,  $\alpha_H$ =9.4 µT, linewidth (Gauss) 8.5 µT, g = 2.00280



Figure S7.3 FT IR (KBr) spectrum of trityl 2

Figure 8SI. HR MS(ESI), EPR, and FT IR spectral data for trityl 3



**Figure S8.1.1** HR MS (ESI) spectrum of trityl **3**: calcd. for C<sub>65</sub>H<sub>81</sub>NO<sub>12</sub>PS<sub>12</sub> [M<sup>-</sup>] 1482.2145, found 1482.251.



**Figure S8.1.2.** HR MS (ESI) spectrum of trityl **3**: calcd. for C<sub>65</sub>H<sub>82</sub>NO<sub>12</sub>PS<sub>12</sub> [M+H<sup>+</sup>] 1483.2223, found 1483.212.



**Figure S8.1.3.** HR MS (ESI) spectrum of trityl **3**: calcd. for C<sub>65</sub>H<sub>85</sub>N<sub>2</sub>O<sub>12</sub>PS<sub>12</sub> [M+NH<sub>4</sub><sup>+</sup>] 1500.2489, found 1500.244; calcd. for C<sub>65</sub>H<sub>81</sub>NNaO<sub>12</sub>PS<sub>12</sub> [M+Na<sup>+</sup>] 1505.2043, found 1505.201.



**Figure S8.2.** EPR for 0.5 mM deoxygenated solution in DCM: multiplet,  $\alpha_{\rm H}$ =9.4 µT, linewidth (Gauss) 8.7 µT, *g* = 2.00280



Figure S8.3 FT IR (KBr) spectrum of trityl 3

#### Figure 9SI. HR MS(ESI), EPR, and FT IR spectral data for trityl 4



**Figure S9.1.1.** HR MS (ESI) spectrum of trityl **4**: calcd. for C<sub>65</sub>H<sub>83</sub>N<sub>2</sub>O<sub>11</sub>PS<sub>12</sub> [M+H<sup>+</sup>] 1482.2383, found 1482.229



**Figure S9.1.2.** HR MS (ESI) spectrum of trityl **4**: calcd. for C<sub>65</sub>H<sub>86</sub>N<sub>3</sub>O<sub>11</sub>PS<sub>12</sub> [M+NH<sub>4</sub><sup>+</sup>] 1499.2648, found 1499.257; calcd. for C<sub>65</sub>H<sub>82</sub>N<sub>2</sub>NaO<sub>11</sub>PS<sub>12</sub> [M+Na<sup>+</sup>] 1504.2202, found 1504.218.



**Figure S9.2.** EPR for 0.5 mM deoxygenated solution of trityl **4** in DCM: multiplet,  $\alpha_{\rm H}$ = 10.6  $\mu$ T, linewidth (Gauss) 7.3  $\mu$ T, *g* = 2.00281



Figure S9.3 FT IR (KBr) spectrum of trityl 4

# Figure 10SI. HR MS(ESI), EPR, and FT IR spectral data for trityl 5



**Figure S10.1.1.** HR MS (ESI) spectrum of trityl **5**: calcd. for C<sub>64</sub>H<sub>81</sub>N<sub>2</sub>O<sub>12</sub>PS<sub>12</sub> [M+H<sup>+</sup>] 1484.2176, found 1484.210; calcd. for C<sub>64</sub>H<sub>80</sub>N<sub>2</sub>NaO<sub>12</sub>PS<sub>12</sub> [M+Na<sup>+</sup>] 1506.1995, found 1506.189.



**Figure S10.2.** EPR for 0.5 mM deoxygenated solution of **5** in DCM: multiplet (splitting on 8 protons in groups of 3H, 3H, 2H),  $\alpha_{H1}$ = 9.0  $\mu$ T,  $\alpha_{H2}$ = 9.5  $\mu$ T,  $\alpha_{H3}$ = 9.8  $\mu$ T (respectively), linewidth (Gauss) 8.6  $\mu$ T, *g* = 2.00281



Figure S10.3 FT IR (KBr) spectrum of trityl 5



Figure 11SI. Livingness of polymer initiated with 1