Electronic Supplementary Information (ESI)

Cu(I)-Catalyzed Amino-Yne Click Polymerization

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Table S1. Crystal data and structure refinement of monomer 1

Tuble 51. Crystal und sit detail e remement of monomer 1				
Empirical formula	$C_{33}H_{24}O_4$			
Formula weight	484.52			
Temperature	293 K			
Wavelength	0.71073 Å			
Crystal system	triclinic			
Space group	P-1			
Unit cell dimensions	a = 8.8102(8) Å	$\alpha = 105.048(8)^{\circ}$.		
	b = 12.1290(11) Å	$\beta = 91.812(7)^{\circ}$.		
	c = 13.7804(11) Å	$\gamma = 110.568(8)^{\circ}$.		
Volume	$1319.1(2) \text{ Å}^3$			
Z	2			
Density (calculated)	1.220 mg/m^3			
Absorption coefficient	0.080 mm ⁻¹			
F(000)	508			
Crystal size	$0.45 \times 0.36 \times 0.28 \text{ mm}^3$			
Theta range for data collection	3.11 to 25.35°.			
Index ranges	-10<=h<=8, -14<=k<=13, -14<=l<=16			
Reflections collections	8271			
Independent reflections	4828 [R(int) = 0.0256]			
Completeness to theta = 66.50°	99.57%			
Absorption correction	Semi-empirical from equivalents			
Max. and min. transmission	1.00000 and 0.88392			
Refinement method	Full-matrix least-squares on F ²			
Data / restraints / parameters	4828 / 0 / 337			
Goodness-of-fit on F ²	1.019			
Final R indices [I>2sigma(I)]	R1 = 0.0520, $wR2 = 0.1227$			
R indices (all data)	R1 = 0.0865, $wR2 = 0.1478$			
Largest diff. peak and hole	0.325 and -0.304 e.Å-3			

Table S2. Crystal data and structure refinement of model compound 4.

Tuble 52. Crystal data and service remement of model compound			
Empirical formula	$C_{45}H_{38}N_2O_4$		
Formula weight	670.77		
Temperature	293 K		
Wavelength	0.71073 Å		
Crystal system	orthorhombic		
Space group	P b c n		
Unit cell dimensions	$a = 12.0454(6) \text{ Å}$ $\alpha = 90.00^{\circ}.$		
	$b = 19.9342(12) \text{ Å}$ $\beta = 90.00^{\circ}.$		
	$c = 15.6271(7) \text{ Å}$ $\gamma = 90.00^{\circ}$.		
Volume	$3752.3(3) \text{ Å}^3$		
Z	4		
Density (calculated)	1.187 mg/m^3		
Absorption coefficient	0.076 mm ⁻¹		
F(000)	1416		
Crystal size	$0.40 \times 0.30 \times 0.26 \text{ mm}^3$		
Theta range for data collection	2.96 to 25.35°.		
Index ranges	-14<=h<=12, -21<=k<=24, -18<=l<=17		
Reflections collections	22198		
Independent reflections	3433 [R(int) = 0.0473]		
Completeness to theta = 66.50°	99.65%		
Absorption correction	Semi-empirical from equivalents		
Max. and min. transmission	1.00000 and 0.95672		
Refinement method	Full-matrix least-squares on F ²		
Data / restraints / parameters	3433 / 0 / 232		
Goodness-of-fit on F ²	1.015		
Final R indices [I>2sigma(I)]	R1 = 0.0455, $wR2 = 0.1089$		
R indices (all data)	R1 = 0.0777, $wR2 = 0.1261$		
Largest diff. peak and hole	0.154 and -0.129 e.Å-3		

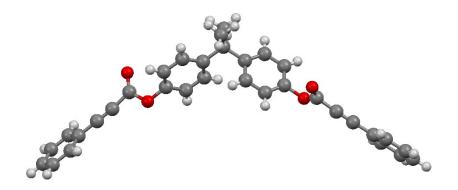


Figure S1. Single crystal structure (CCDC 1457164) of monomer 1.

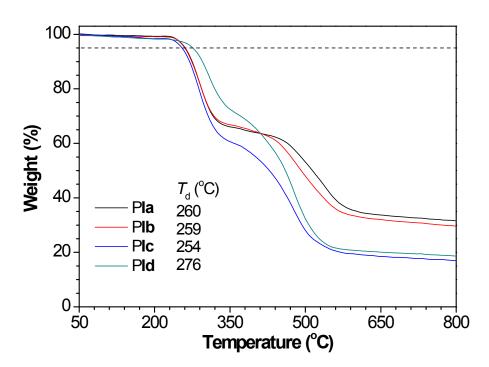


Figure S2. TGA curves of PIa-PId recorded under nitrogen at a heating rate of 10 °C/min.

Table S3. The weight loss in the first decomposition stage.

polymers	experimental values (%)	calculated values (%)
P Ia	34.2	33.1
P Ib	33.3	33.0
PIc	39.5	38.0

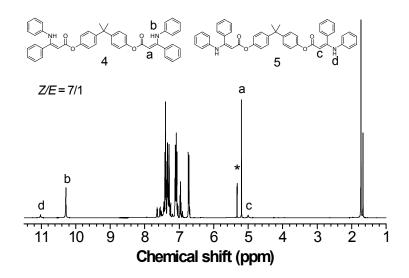


Figure S3 ¹H NMR spectra of crude product of model reaction in dichloromethane- d_2 . The solvent peaks are marked with asterisks.

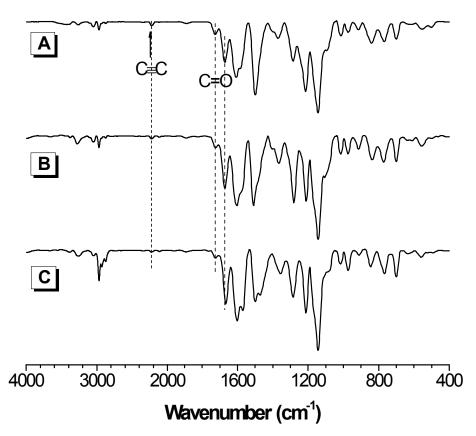


Figure S4. FT-IR spectra of (A) PIb, (B) PIc and (C) PId.

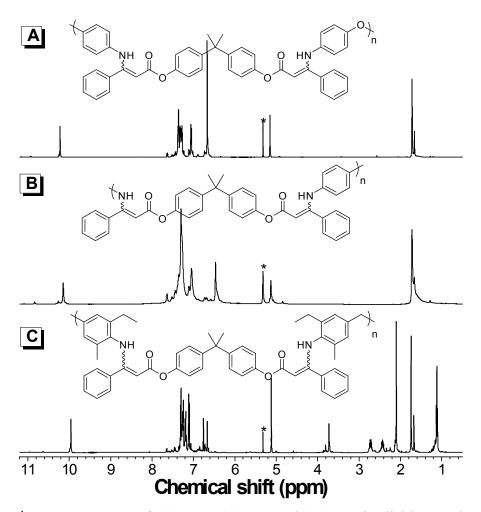


Figure S5. ¹H NMR spectra of (A) P**Ib**, (B) P**Ic** and (C) P**Id** in dichloromethane- d_2 . The solvent peaks are marked with asterisks.

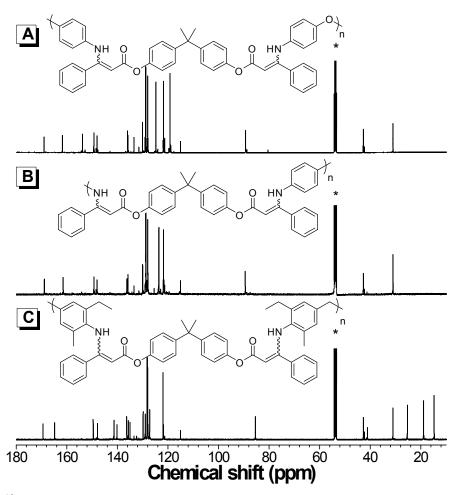


Figure S6. ¹³C NMR spectra of (A) P**Ib**, (B) P**Ic** and (C) P**Id** in dichloromethane- d_2 . The solvent peaks are marked with asterisks.