

Supporting information for:

**Post-polymerization functionalization of poly(ethylene oxide)-poly( $\beta$ -6-heptenolactone)  
diblock copolymers to tune properties and self-assembly**

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**General materials and procedures as well as some representative procedures can be found in the main manuscript**

**Synthesis of PEO<sub>45</sub>-*b*-PHEL<sub>45</sub>.** This polymer was synthesized as described for PEO<sub>45</sub>-*b*-PHEL<sub>23</sub> except that the following quantities were used:  $\beta$ -6-HEL (2.00 g, 15.9 mmol, 51 equiv.), [Al] (170 mg, 0.31 mmol, 1.0 equiv.), PEO (620 mg, 0.31 mmol, 1.0 equiv.), toluene (20 mL). Yield = 90%. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  1.69 – 1.71 (m, 92H), 2.04 – 2.07 (m, 97H), 2.47 – 2.62 (m, 92H), 3.38 (s, 3H), 3.64 (br s, 180H), 4.20 – 4.22 (m, 2H), 4.96 – 5.04 (m, 93H), 5.20-5.23 (m, 45H), 5.72-5.82 (m, 45H). M<sub>n</sub> based on <sup>1</sup>H NMR spectroscopy = 7040 g mol<sup>-1</sup>. SEC (THF): M<sub>n</sub> = 6630 g mol<sup>-1</sup>, M<sub>w</sub> = 7860 g mol<sup>-1</sup>, D = 1.19. FTIR: 2863, 1736, 1641 cm<sup>-1</sup>. T<sub>m</sub> = 29°C. T<sub>g</sub> = -59°C.

**Synthesis of PEO<sub>45</sub>-*b*-PHEL<sub>79</sub>.** This polymer was synthesized as described for PEO<sub>45</sub>-*b*-PHEL<sub>23</sub> except that the following quantities were used:  $\beta$ -6-HEL (2.20 g, 17.4 mmol, 92 equiv.), [Al] (106 mg, 0.19 mmol, 1.0 equiv.), PEO (387 mg, 0.19 mmol, 1.0 equiv.), toluene (15 mL). Yield = 83%. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  1.68 – 1.71 (m, 171H), 2.04 – 2.09 (m, 170H), 2.48 – 2.61 (m, 173H), 3.38 (s, 3H), 3.64 (br s, 180H), 4.20 – 4.23 (m, 2H), 4.97 – 5.03 (m, 158H), 5.21-5.22 (m, 81H), 5.74 – 5.80 (m, 76H). M<sub>n</sub> based on <sup>1</sup>H NMR spectroscopy = 10848 g mol<sup>-1</sup>. SEC (THF): M<sub>n</sub> = 12910 g mol<sup>-1</sup>, M<sub>w</sub> = 13340 g mol<sup>-1</sup>, D = 1.03. FTIR: 2924, 1737, 1642 cm<sup>-1</sup>. T<sub>m</sub> = 22°C. T<sub>g</sub> = -46°C.

**Synthesis of PEO<sub>45</sub>-*b*-PHEL<sub>31</sub>-TEG<sub>14</sub>.** The same procedure as for the synthesis of PEO<sub>45</sub>-*b*-PHEL<sub>21</sub>-octyl<sub>24</sub> was used except that the following reagents and quantities were used. Polymer =

PEO<sub>45</sub>-*b*-PHEL<sub>45</sub> (50.0 mg, 6.0  $\mu\text{mol}$ ); thiol = TEG-thiol (27.0 mg, 0.150 mmol); initiator = DMPA (1.92 mg, 8.0  $\mu\text{mol}$ ); solvent = toluene (1 mL). The polymer was purified by dialysis using 3500 MWCO regenerated cellulose membrane in DMF. Yield = 74%. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  1.39–1.42 (m, 30H), 1.56 – 1.60 (m, 55H), 1.69 – 1.73 (m, 66H), 2.05 – 2.09 (m, 68H), 2.51 – 2.57 (m, 118H), 2.69 (t, 27H,  $J$  = 7.0 Hz), 3.37 (s, 45H), 3.54–3.55 (m, 28H), 3.64 (bs, 275H), 4.20 – 4.22 (m, 2H), 4.96 – 5.03 (m, 62H), 5.18 – 5.21 (m, 45H), 5.73 – 5.80 (m, 29H).  $M_n$  based on <sup>1</sup>H NMR spectroscopy = 10549 g mol<sup>-1</sup>. SEC (THF):  $M_n$  = 7710 g mol<sup>-1</sup>,  $M_w$  = 8840 g mol<sup>-1</sup>,  $D$  = 1.15. FTIR: 2865, 1735, 1640.  $T_m$  = 29°C.  $T_g$  = -44°C.

**Synthesis of PEO<sub>45</sub>-*b*-CA<sub>45</sub>.** The same procedure as for the synthesis of PEO<sub>45</sub>-*b*-PHEL<sub>21-octyl<sub>24</sub> was used except that the following reagents and quantities were used. Polymer = PEO<sub>45</sub>-*b*-PHEL<sub>45</sub> (500 mg, 0.063 mmol); thiol = thioglycolic acid (936 mg, 8.82 mmol); initiator = DMPA (113 mg, 0.441 mmol); solvent = toluene (4 mL). The polymer was purified by dialysis using 3500 g mol<sup>-1</sup> MWCO regenerated cellulose membrane in DMF. Yield = 61%. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  1.29 – 1.32 (m, 95H), 1.50 – 1.52 (m, 189H), 2.44 – 2.54 (m, 188H), 3.13 (s, 88H), 3.27 (s, 4H), 3.54 (bs, 180H), 4.10 – 4.12 (m, 2H), 5.05 – 5.09 (m, 46H), 11.18 (br s, 41H).  $M_n$  based on <sup>1</sup>H NMR spectroscopy = 11180 g mol<sup>-1</sup>. FTIR: 3447, 2940, 2870, 1726, 1640 cm<sup>-1</sup>.  $T_g$  = -19°C.</sub>

**Synthesis of PEO<sub>45</sub>-*b*-PHEL<sub>20</sub>-CA<sub>25</sub>.** The same procedure as for the synthesis of PEO<sub>45</sub>-*b*-PHEL<sub>21-octyl<sub>24</sub> was used except that the following reagents and quantities were used. Polymer = PEO<sub>45</sub>-*b*-PHEL<sub>45</sub> (250 mg, 0.033 mmol); thiol = thioglycolic acid (62.0 mg, 0.673 mmol); initiator = DMPA (8.7 mg, 0.034 mmol); solvent = toluene (3 mL). The polymer was purified by</sub>

dialysis using 3500 g mol<sup>-1</sup> MWCO regenerated cellulose membrane in DMF. Yield = 58%. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>): δ 1.37 – 1.45 (m, 49H), 1.49 – 1.75 (m, 138H), 2.03 (m, 44H), 2.40 – 2.68 (m, 133H), 3.19 (m, 42H), 3.42 (s, 3H), 3.60 (br s, 180H), 4.16–4.18 (m, 2H), 4.89 – 5.03 (m, 42H), 5.18–5.22 (m, 44H), 5.66 – 5.79 (m, 20H). M<sub>n</sub> based on <sup>1</sup>H NMR spectroscopy = 9341 g mol<sup>-1</sup>. FTIR: 3463, 2933, 2869, 1732, 1644 cm<sup>-1</sup>. T<sub>g</sub> = -46°C.

**Synthesis of PEO<sub>45</sub>-*b*-CA<sub>11</sub>-PTX<sub>34</sub>.** In a flame-dried flask equipped with stir bar, PEO<sub>45</sub>-*b*-CA<sub>45</sub> (75.0 mg, 0.007 mmol, equiv.) was dissolved in dry DMF (5 mL). Paclitaxel (550 mg, 0.070 mmol, equiv.), EDC·HCl (147 mg, 0.710 mmol, equiv.) and DMAP (33 mg, 0.271 mmol, equiv.) were added and the reaction mixture was stirred overnight. The polymer was precipitated into cold ethanol and purified by dialysis using 6000-8000 g mol<sup>-1</sup> MWCO membrane in DMF. Yield = 49%. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>): δ 1.10 – 1.63 (m, 696H), 1.78 – 2.47 (m, 826H), 3.21 – 3.26 (m, 85H), 3.64 (br s, 190H), 3.73 – 3.79 (m, 34H), 4.11 – 4.18 (m, 34H), 4.22 – 4.28 (m, 33H), 4.32 – 4.45 (m, 30H), 4.88 – 4.98 (m, 35H), 5.08 – 5.25 (m, 45H), 5.50 – 5.56 (m, 36H), 5.61 – 5.67 (m, 41H), 5.90 – 5.60 (m, 34H), 6.13 – 6.24 (m, 40H), 6.25 – 6.31 (s, 29H), 7.27 – 7.71 (m, 513H), 8.08 – 8.14 (m, 76H). M<sub>n</sub> based on <sup>1</sup>H NMR spectroscopy = 39601 g mol<sup>-1</sup>. SEC (THF): M<sub>n</sub> = 9010 g mol<sup>-1</sup>, M<sub>w</sub> = 16920 g mol<sup>-1</sup>, D = 1.88. FTIR: 3447, 2940, 1726. T<sub>g</sub> = 131°C.

**Synthesis of PEO<sub>45</sub>-*b*-PHEL<sub>20</sub>-CA<sub>7</sub>-PTX<sub>18</sub>.** This polymer was synthesized as described above for PEO<sub>45</sub>-*b*-CA<sub>11</sub>-PTX<sub>34</sub> except that the following quantities were used: PEO<sub>45</sub>-*b*-PHEL<sub>20</sub>-CA<sub>25</sub> (50.0 mg, 0.005 mmol, 1.0 equiv.); paclitaxel (269 mg, 0.315 mmol, 63 equiv.); EDC·HCl (66 mg, 0.345 mmol, 69 equiv.); DMAP (17 mg, 0.140 mmol, 28 equiv.); DMF (4 mL). Yield = 56%. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>): δ 1.10 – 1.63 (m, 620H), 1.78 – 2.47 (m, 592H), 3.21 – 3.26

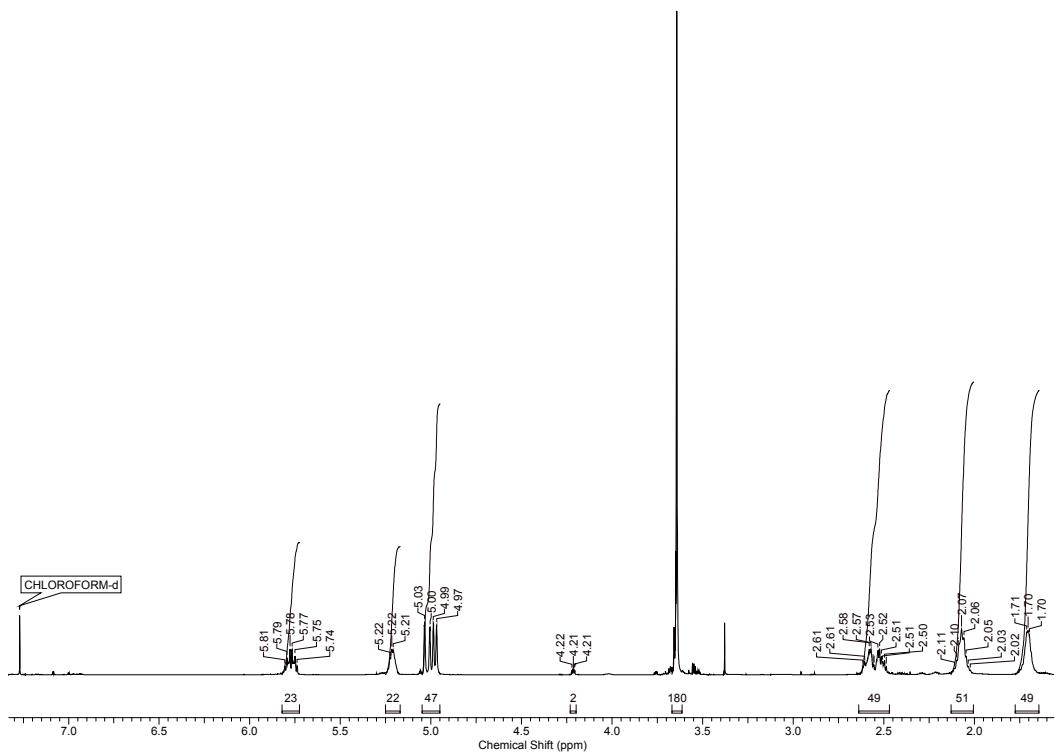
(m, 32H), 3.64 (br s, 180H), 3.73 – 3.79 (m, 18H), 4.11 – 4.18 (m, 19H), 4.22 – 4.28 (m, 16H), 4.32 – 4.45 (m, 17H), 4.88 – 4.98 (m, 59H), 5.08 – 5.25 (m, 44H), 5.50 – 5.56 (m, 18H), 5.61 – 5.67 (m, 45H), 5.90 – 5.60 (m, 16H), 6.13 – 6.24 (m, 17H), 6.25 – 6.31 (br s, 18H), 6.85 – 7.71 (m, 452H), 8.08 – 8.14 (m, 55H).  $M_n$  based on  $^1\text{H}$  NMR spectroscopy = 24387 g mol $^{-1}$ .  $M_n$  = 6750 g mol $^{-1}$ ,  $M_w$  = 8800 g mol $^{-1}$ ,  $D$  = 1.30. FTIR: 3459, 2940, 1726, 1242, 704 cm $^{-1}$ .  $T_g$  = 87°C.

**Synthesis of anhydride 2.** Dicyclohexylcarbodiimide (656 mg, 3.2 mmol, 0.6 equiv.) in dry  $\text{CH}_2\text{Cl}_2$  (5 mL) was added to a solution of 3-tritylsulfanyl-propionic acid (**1**) (2.0 g, 5.7 mmol, 1.0 equiv.) dissolved in dry  $\text{CH}_2\text{Cl}_2$  (15 mL). The reaction mixture was stirred at room temperature for 10 h and filtered to remove dicyclohexylurea byproduct. The resulting solution was then concentrated, washed with ethyl acetate and dried under vacuum to give **2** as a white solid. Yield = 86%.  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ ):  $\delta$  2.25 (4H, t,  $J$  = 7.3 Hz), 2.50 (4H, t,  $J$  = 7.3 Hz), 7.21 – 7.24 (6H, m), 7.30 (12H, t,  $J$  = 7.6 Hz), 7.50 (12H, d,  $J$  = 7.6 Hz).  $^{13}\text{C}$  NMR (125 MHz, DMSO):  $\delta$  167.4, 144.1, 129.1, 128.1, 126.8, 66.3, 33.8, 25.8. FTIR: 3069, 2939, 1819, 1701 cm $^{-1}$ . HRMS calcd for  $[\text{M}+\text{Na}]^+$   $\text{C}_{44}\text{H}_{38}\text{O}_3\text{S}_2\text{Na}$ : 701.2160; found: 701.2160.

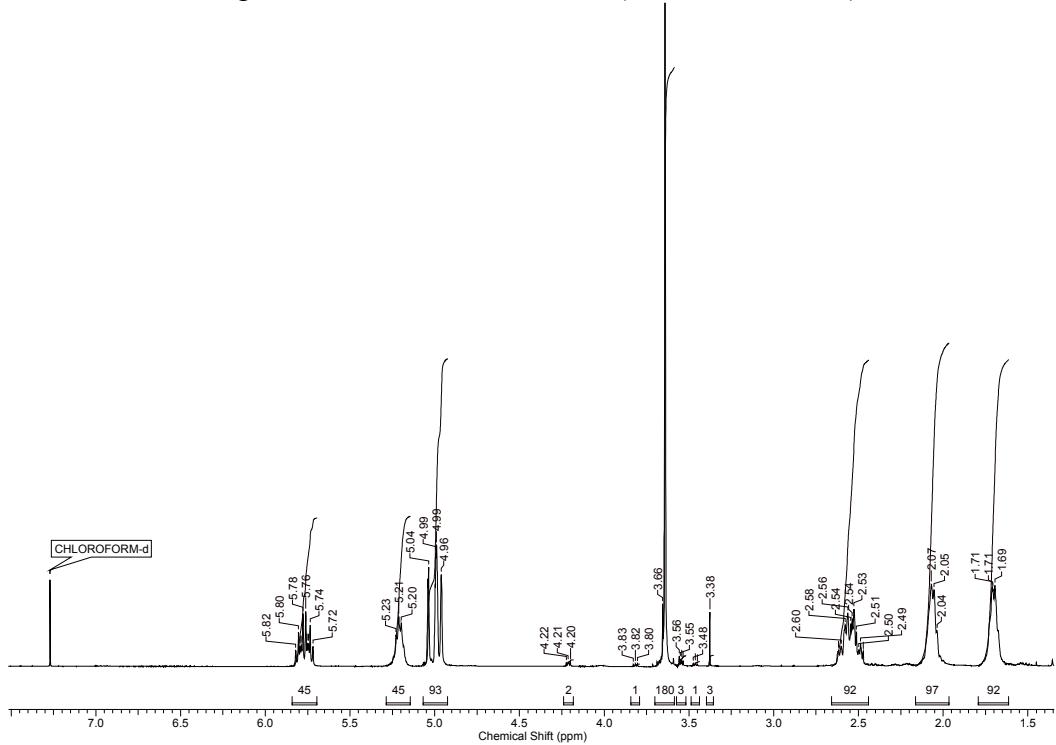
**Synthesis of rhodamine derivative 4.** A solution of rhodamine derivative **3**<sup>1</sup> (500 mg, 1.03 mmol, 1.0 equiv.) in dry THF (14 mL) was added dropwise to a solution of the anhydride **2** (350 mg, 0.516 mmol, 0.5 equiv.) dissolved in dry THF (10 mL). The reaction mixture was stirred overnight at room temperature. The solvent was evaporated and the product was purified by column chromatography using neutral alumina (EtOAc:Hexanes (1:1)). Yield = 36%.  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ ,  $\delta$ ): 1.17 (t, 12H  $J$  = 7.1 Hz), 2.06 (dd, 3H,  $J$  = 9.3, 5.7 Hz), 2.44 (t, 2H,  $J$  = 7.5 Hz), 2.98 – 3.03 (m, 2H), 3.27 – 3.38 (m, 10H), 6.24 (dd, 2H,  $J$  = 8.9, 2.5 Hz), 6.39 (d, 2H,  $J$  = 2.5 Hz), 6.41 (d, 2H,  $J$  = 8.9 Hz), 6.66 (t, 1H,  $J$  = 4.1 Hz), 7.10 (dd, 1H,  $J$  = 6.1, 1.6 Hz), 7.19

(t, 3H,  $J$  = 7.3 Hz), 7.41 (d, 6H,  $J$  = 7.5 Hz), 7.26 (t, 6H,  $J$  = 7.7 Hz), 7.44 – 7.49 (m, 2H), 7.89 (dd, 1H,  $J$  = 6.2, 2.0 Hz).  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ ):  $\delta$  170.85, 169.97, 153.98, 153.36, 149.02, 144.93, 132.86, 130.61, 129.73, 128.61, 128.06, 127.99, 127.95, 126.69, 124.04, 104.86, 97.81, 65.66, 66.71, 44.48, 40.71, 35.99, 27.92, 12.75. FTIR: 3081, 2966, 1721, 1615, 1514  $\text{cm}^{-1}$ . HRMS calcd for  $[\text{M}]^+$   $\text{C}_{52}\text{H}_{54}\text{N}_4\text{O}_3\text{S}$ : 814.3917; found: 814.3928.

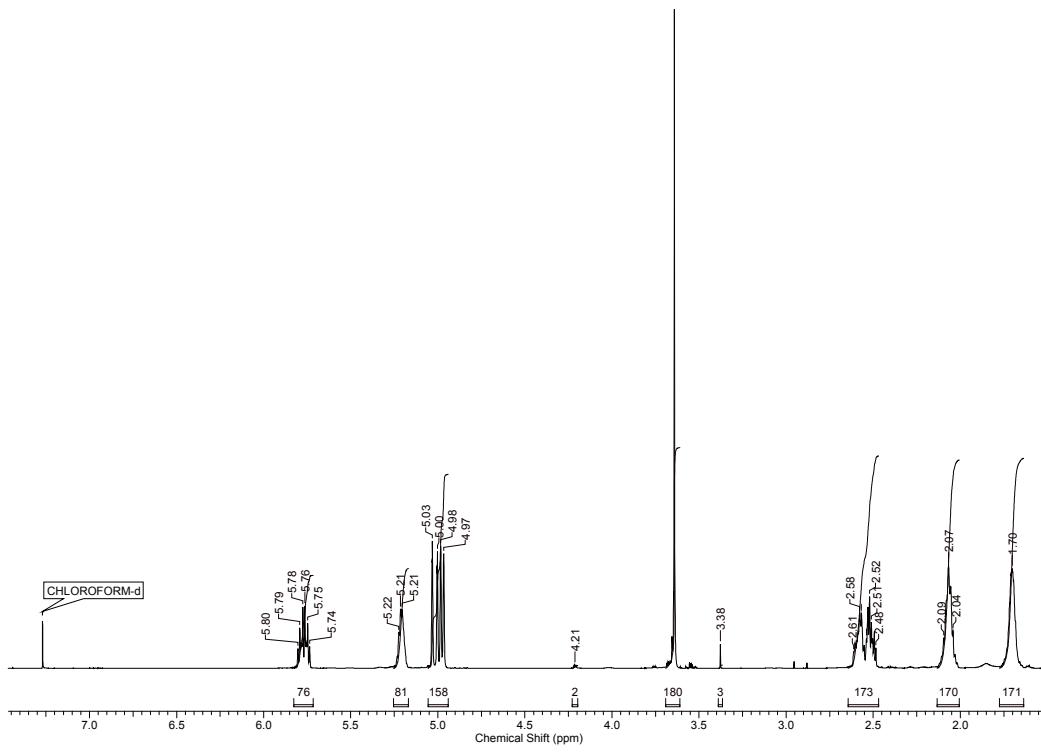
**Synthesis of  $\text{PEO}_{45}\text{-}b\text{-PHEL}_{40}\text{-RHD}_5$ .** To a solution of rhodamine derivative **4** (215 mg, 0.264 mmol, 1.0 equiv.) dissolved in dry  $\text{CH}_2\text{Cl}_2$  (1 mL) was added a solution of TFA: $\text{CH}_2\text{Cl}_2$  (1:1) (1 mL) dropwise at 0 °C. The reaction mixture was stirred at room temperature for 4 hr and monitored by TLC. The solvent was evaporated, and the resulting thiol product was used immediately without further purification. To a 10 mL Schlenk tube equipped with a stir bar, a mixture of  $\text{PEO}_{45}\text{-}b\text{-PHEL}_{45}$  (50.0 mg, 0.007 mmol, 1.0 equiv.), deprotected **4** (145 mg, 0.253 mmol, 36 equiv.) and azobisisobutyronitrile (AIBN) (8.0 mg, 0.050 mmol, 7.1 equiv.) in toluene (6 mL) were added and degassed by three cycles of freeze pump thaw. The reaction mixture was then stirred at 80 °C for 6 hours. The resulting polymer was purified by dialysis using 6000-8000 g mol<sup>-1</sup> MWCO membrane in DMF. Yield = 69%.  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ ):  $\delta$  1.15 (t, 65H,  $J$  = 7.0 Hz), 1.71 (br s, 94H), 2.02 – 2.07 (m, 103H), 2.41 (t, 14H,  $J$  = 7.3 Hz), 2.50 – 2.58 (m, 94H), 2.95 – 2.96 (m, 11H), 3.25 – 3.32 (m, 51H), 3.64 (br s, 180H), 4.22 (br s, 2H), 4.97 – 5.04 (m, 84H), 5.21 – 5.23 (m, 45H), 5.74 – 5.81 (m, 40H), 6.22 (dd, 9H,  $J$  = 8.8, 2.4 Hz), 6.39 (d, 19H,  $J$  = 9.4 Hz), 6.62 (br s, 4H), 7.07 (d, 6H,  $J$  = 7.0 Hz), 7.15 – 7.18 (m, 14H), 7.24 (t, 28H,  $J$  = 7.6 Hz), 7.38 (d, 28H,  $J$  = 7.3 Hz), 7.45 (quin, 12H,  $J$  = 6.9 Hz), 7.86 (d, 5H,  $J$  = 7.0 Hz).  $M_n$  based on  $^1\text{H}$  NMR spectroscopy = 11115 g mol<sup>-1</sup>. SEC (THF):  $M_n$  = 6300 g mol<sup>-1</sup>,  $M_w$  = 7260 g mol<sup>-1</sup>,  $D$  = 1.15. FTIR: 3081, 2930, 2866, 1739, 1721, 1640, 1616, 1518  $\text{cm}^{-1}$ .  $T_g$  = -33 °C.



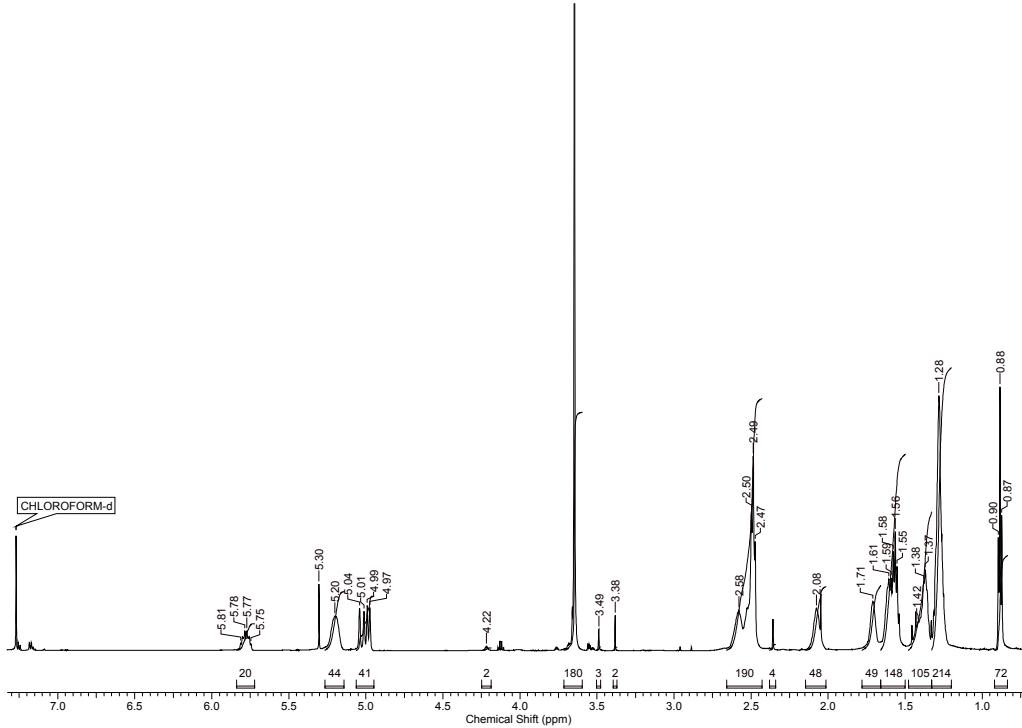
**Figure S1.**  $^1\text{H}$  NMR spectrum of PEO<sub>45</sub>-*b*-PHEL<sub>23</sub> (600 MHz, CDCl<sub>3</sub>).



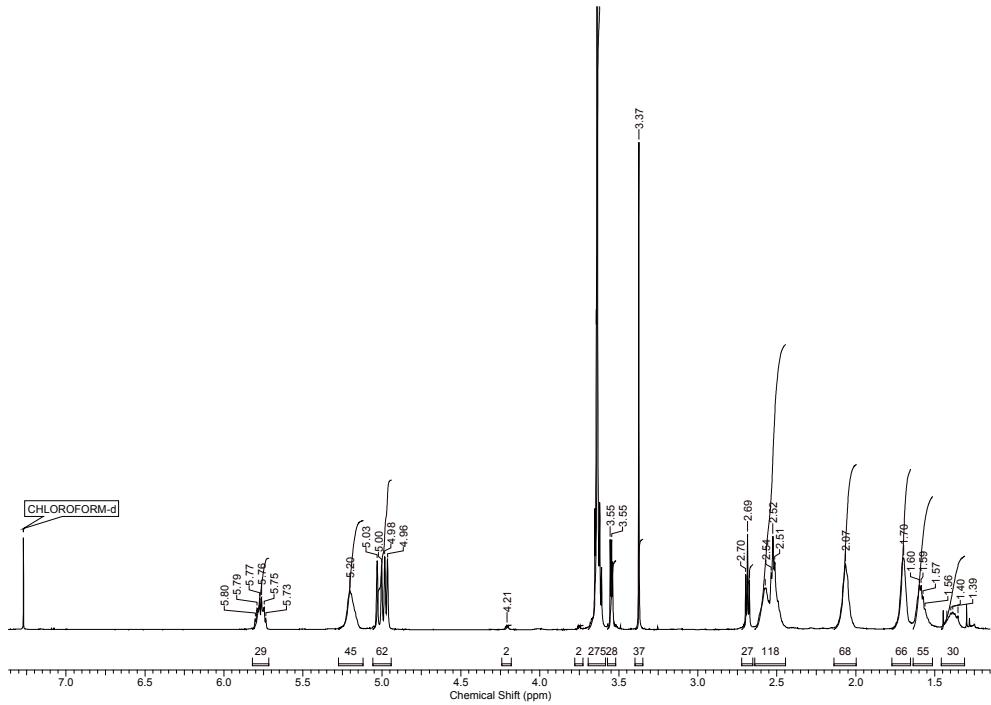
**Figure S2.**  $^1\text{H}$  NMR spectrum of PEO<sub>45</sub>-*b*-PHEL<sub>45</sub> (600 MHz, CDCl<sub>3</sub>).



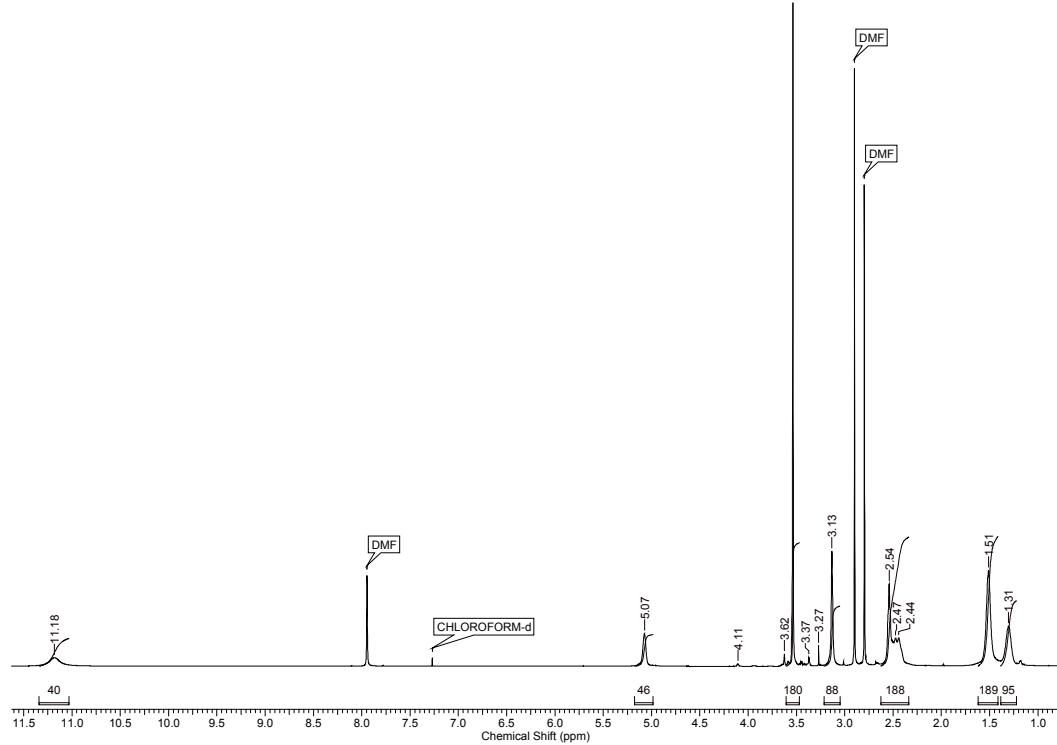
**Figure S3.**  $^1\text{H}$  NMR spectrum of PEO<sub>45</sub>-*b*-PHEL<sub>79</sub> (600 MHz, CDCl<sub>3</sub>).



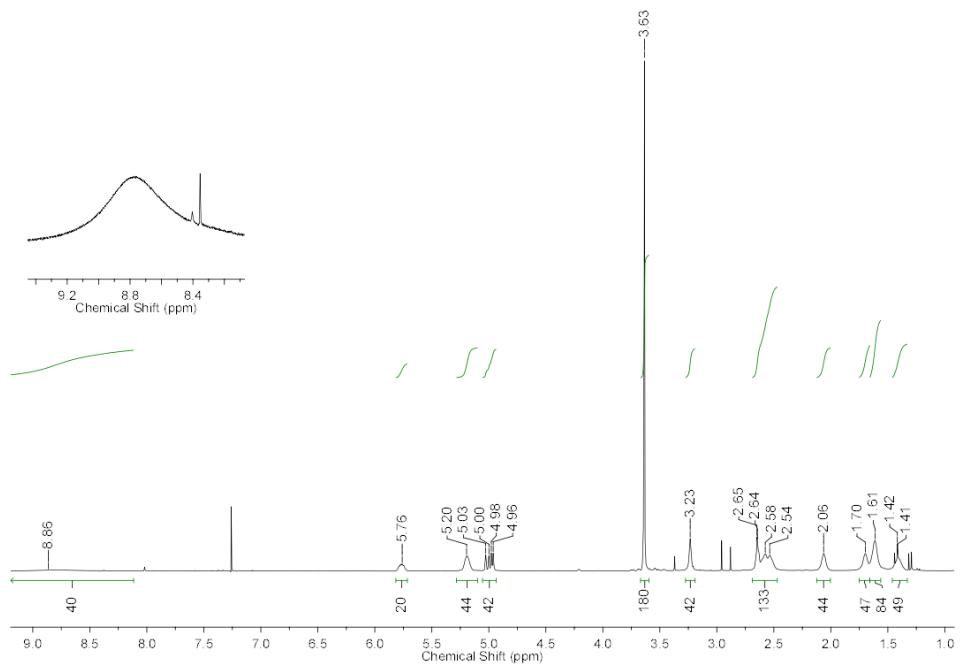
**Figure S4.**  $^1\text{H}$  NMR spectrum of PEO<sub>45</sub>-*b*-PHEL<sub>21</sub>-octyl<sub>24</sub> (600 MHz, CDCl<sub>3</sub>).



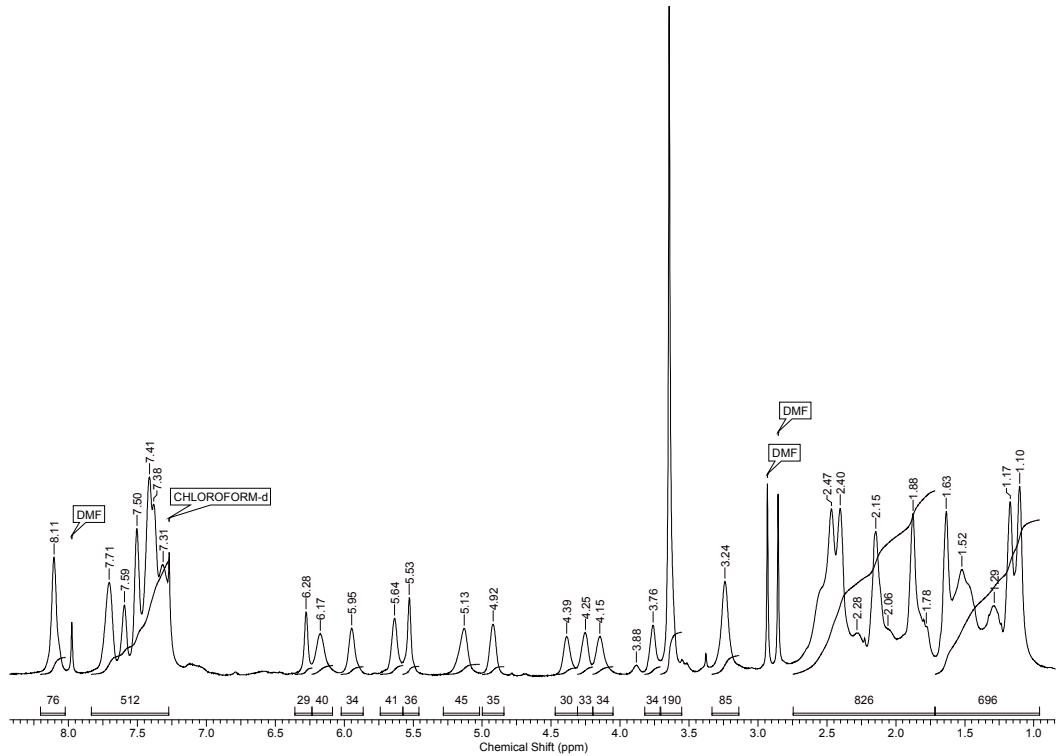
**Figure S5.**  $^1\text{H}$  NMR spectrum of PEO<sub>45</sub>-*b*-PHEL<sub>31</sub>-TEG<sub>14</sub> (600 MHz, CDCl<sub>3</sub>).



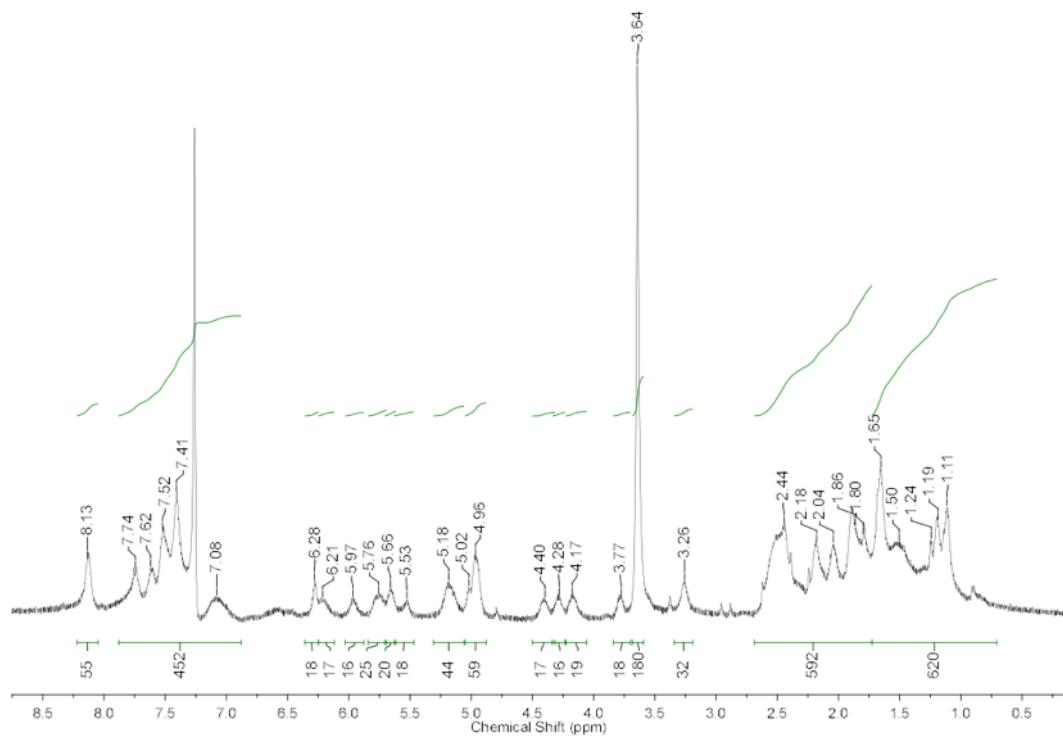
**Figure S6.**  $^1\text{H}$  NMR spectrum of PEO<sub>45</sub>-*b*-CA<sub>45</sub> (600 MHz, CDCl<sub>3</sub>).



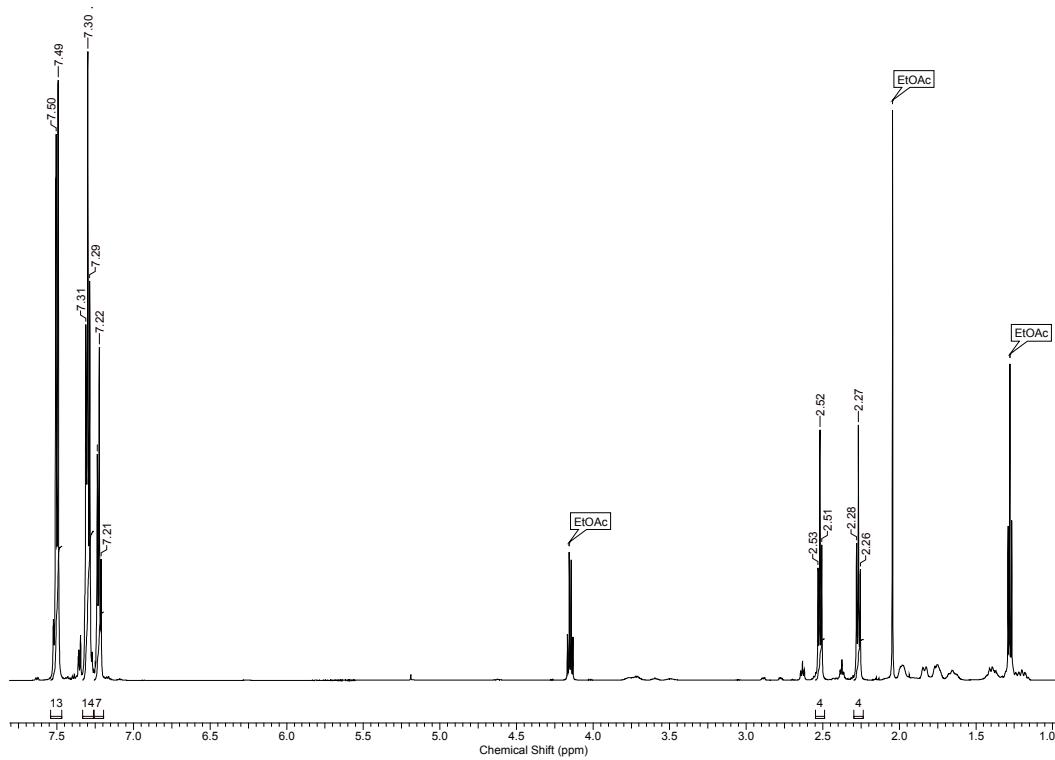
**Figure S7.** <sup>1</sup>H NMR spectrum of PEO<sub>45</sub>-*b*-PHEL<sub>20</sub>-CA<sub>25</sub> (600 MHz, CDCl<sub>3</sub>).



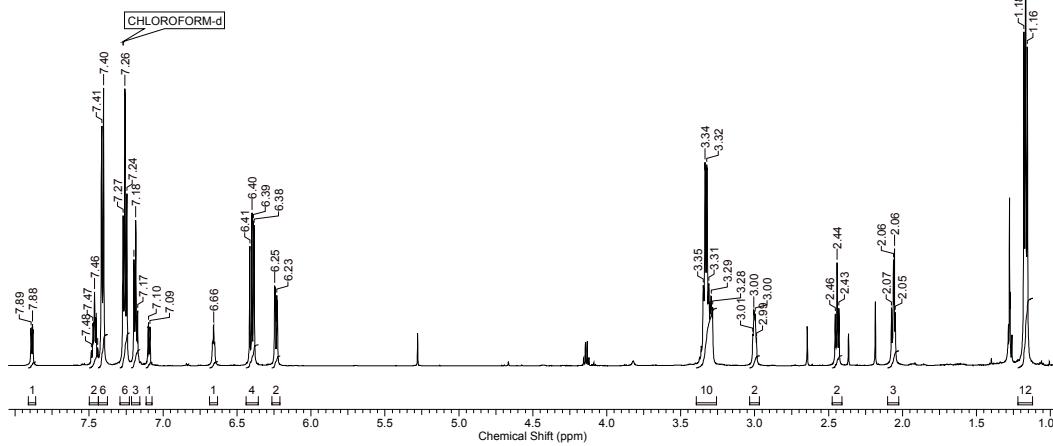
**Figure S8.** <sup>1</sup>H NMR spectrum of PEO<sub>45</sub>-*b*-CA<sub>11</sub>-PTX<sub>34</sub> (600 MHz, CDCl<sub>3</sub>).



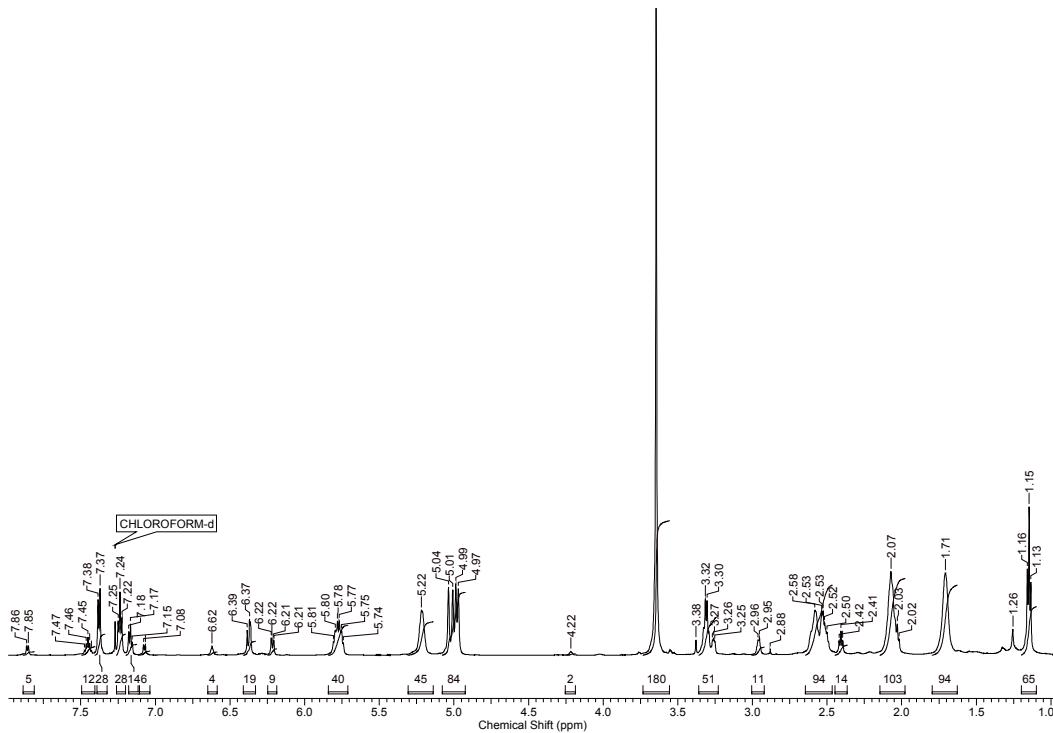
**Figure S9.**  $^1\text{H}$  NMR spectrum of  $\text{PEO}_{45}\text{-}b\text{-PHEL}_{20}\text{-CA}_7\text{-PTX}_{18}$  (600 MHz,  $\text{CDCl}_3$ ).



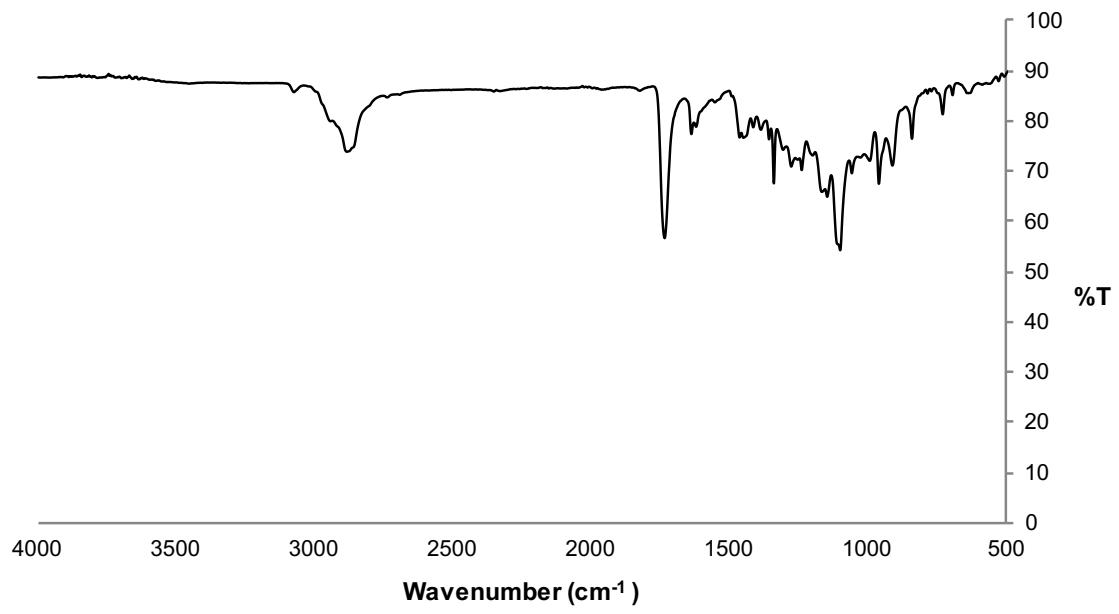
**Figure S10.**  $^1\text{H}$  NMR spectrum of **2** (600 MHz,  $\text{CDCl}_3$ ).



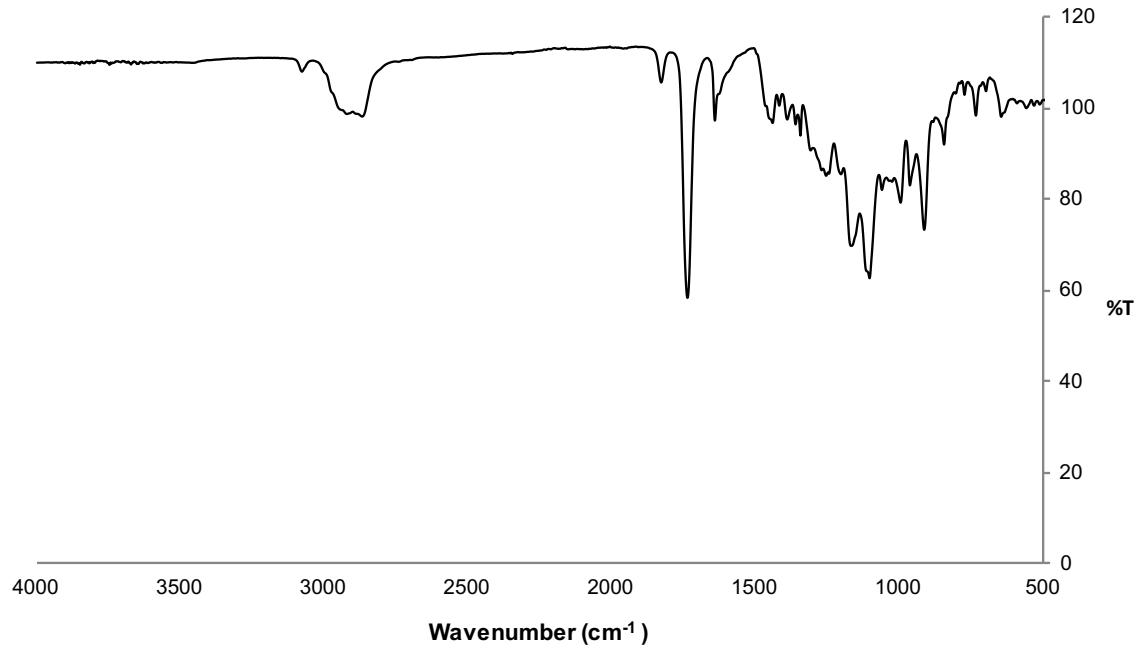
**Figure S11.**  $^1\text{H}$  NMR spectrum of **4** (600 MHz,  $\text{CDCl}_3$ ).



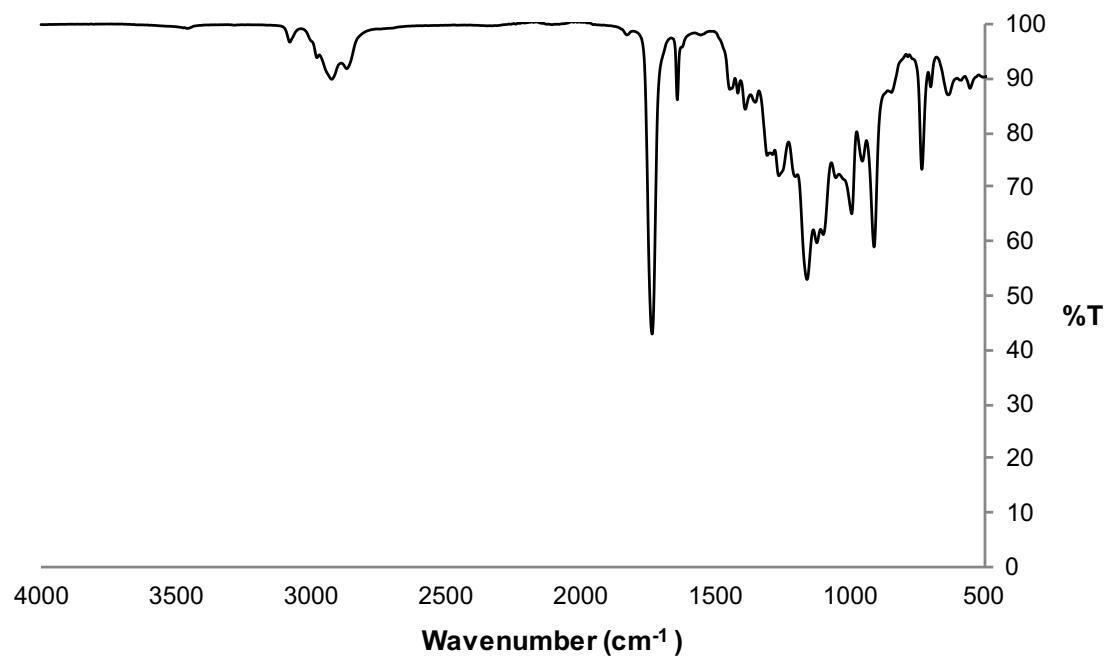
**Figure S12.**  $^1\text{H}$  NMR spectrum of PEO<sub>45</sub>-b-PHEL<sub>40</sub>-RHD<sub>5</sub> (600 MHz, CDCl<sub>3</sub>).



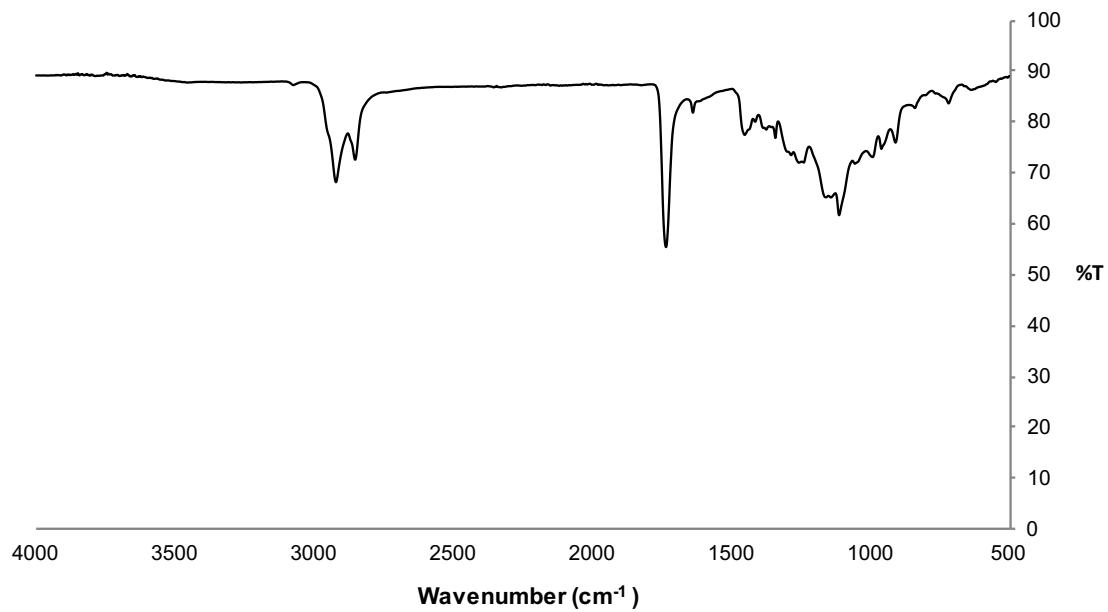
**Figure S13.** ATR-FTIR spectrum of PEO<sub>45</sub>-*b*-PHEL<sub>23</sub>.



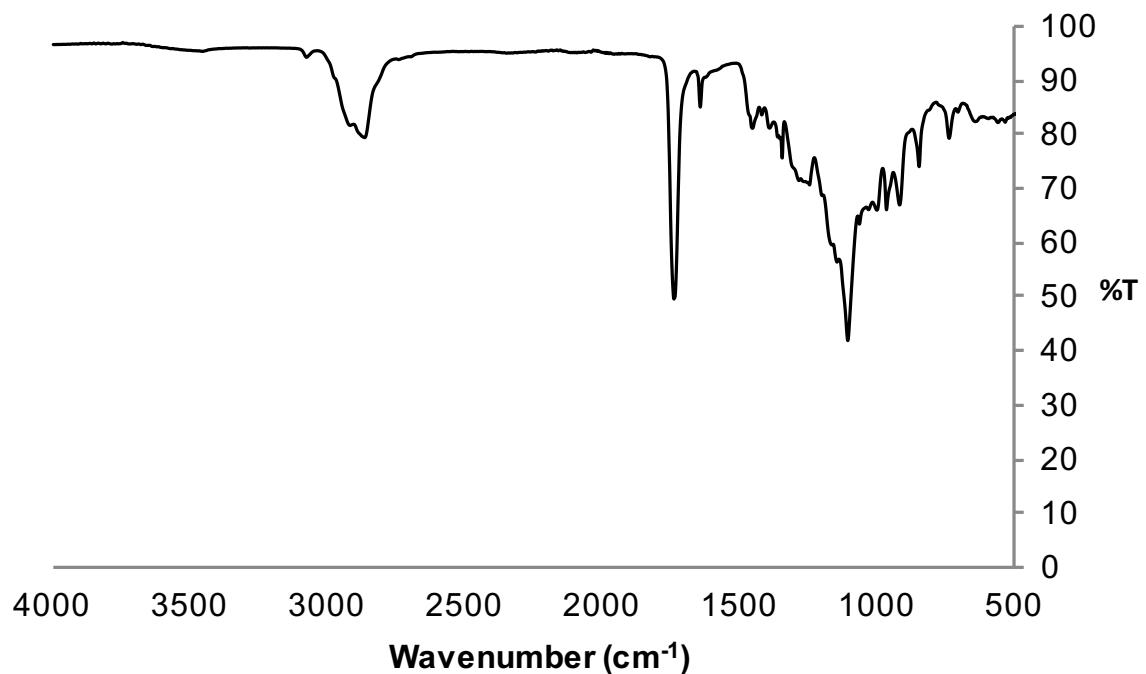
**Figure S14.** ATR-FTIR spectrum of PEO<sub>45</sub>-*b*-PHEL<sub>45</sub>.



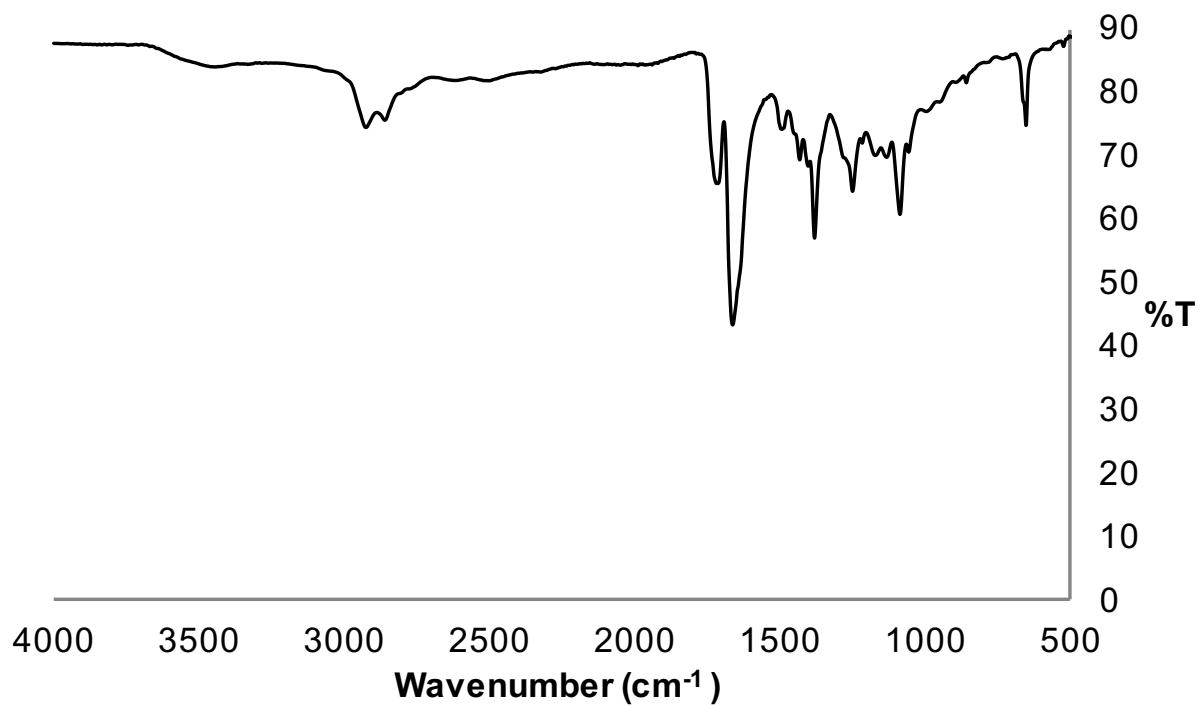
**Figure S15.** ATR-FTIR spectrum of PEO<sub>45</sub>-*b*-PHEL<sub>79</sub>.



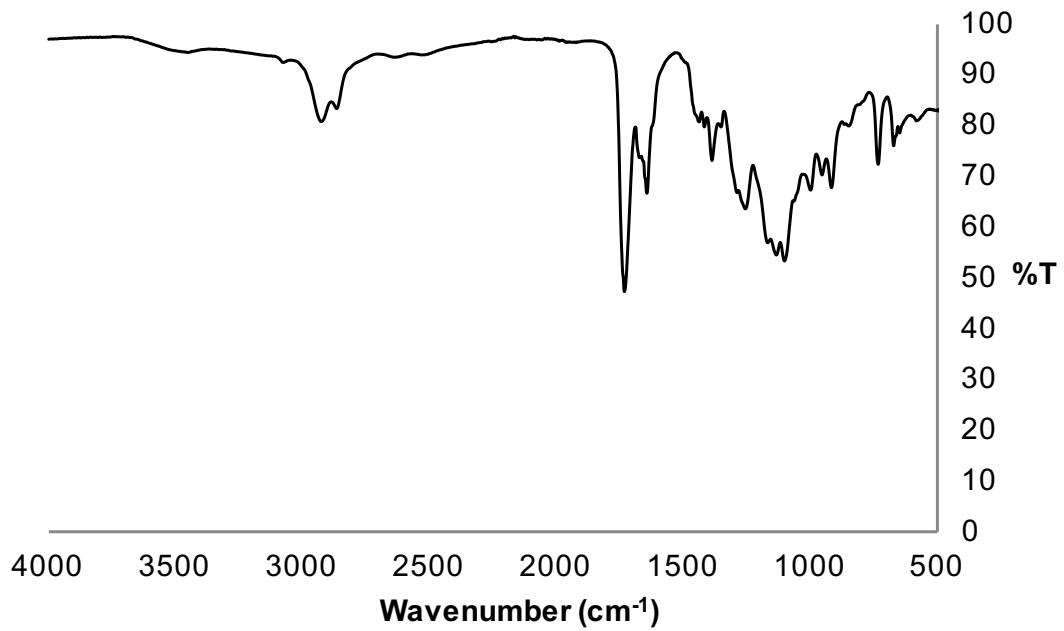
**Figure S16.** ATR-FTIR spectrum of PEO<sub>45</sub>-*b*-PHEL<sub>21</sub>-octyl<sub>24</sub>.



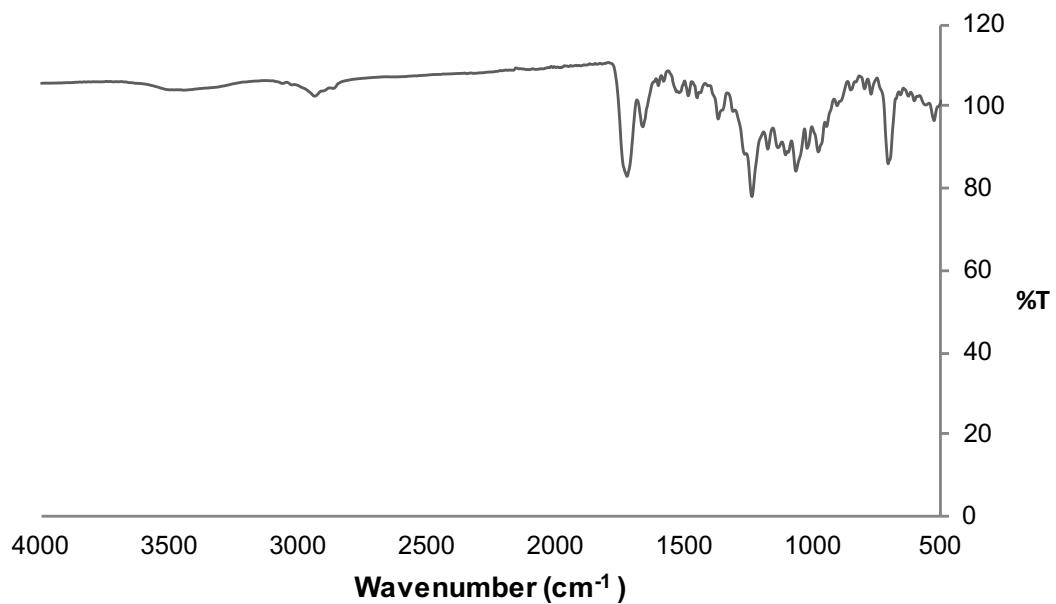
**Figure S17.** ATR-FTIR spectrum of PEO<sub>45</sub>-*b*-PHEL<sub>31</sub>-TEG<sub>14</sub>.



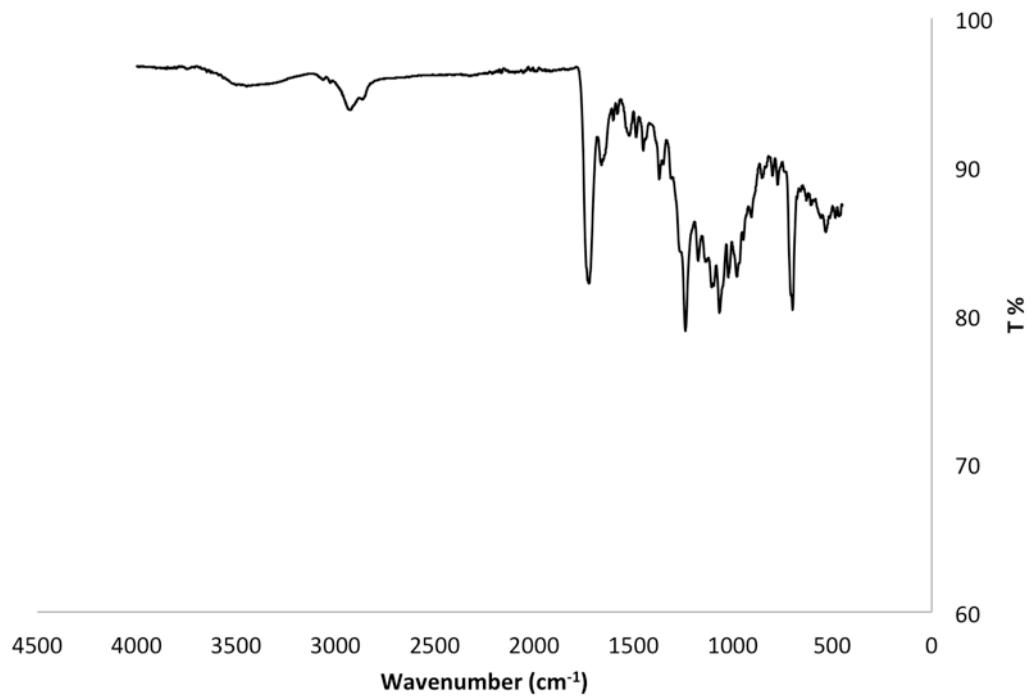
**Figure S18.** ATR-FTIR spectrum of PEO<sub>45</sub>-*b*-CA<sub>45</sub>.



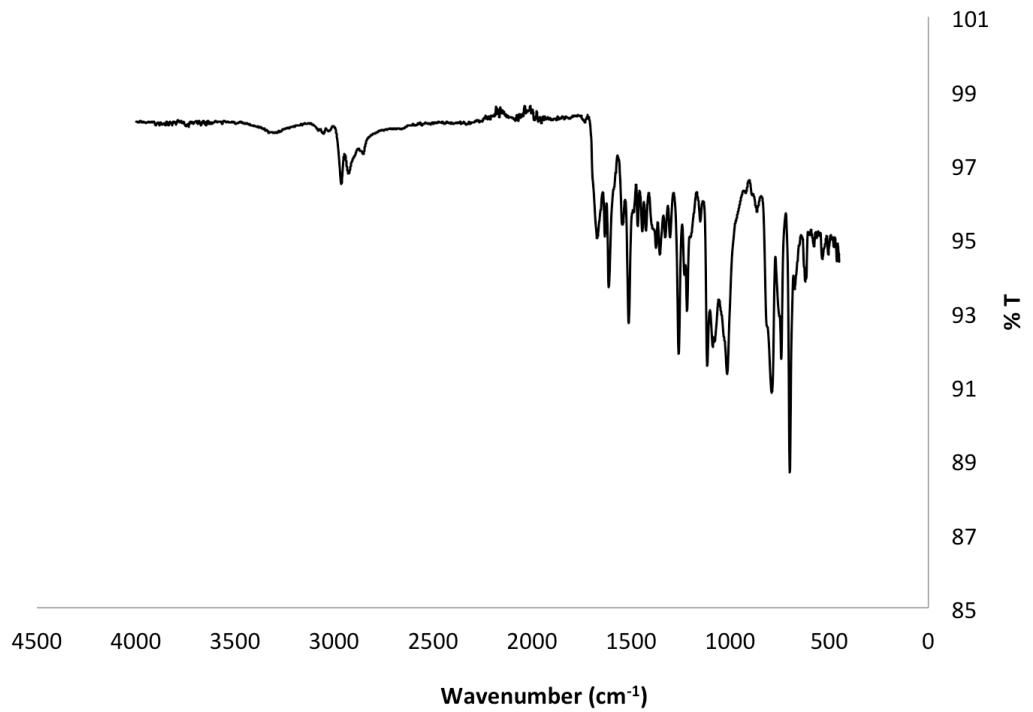
**Figure S19.** ATR-FTIR spectrum of  $\text{PEO}_{45}\text{-}b\text{-PHEL}_{20}\text{-CA}_{25}$ .



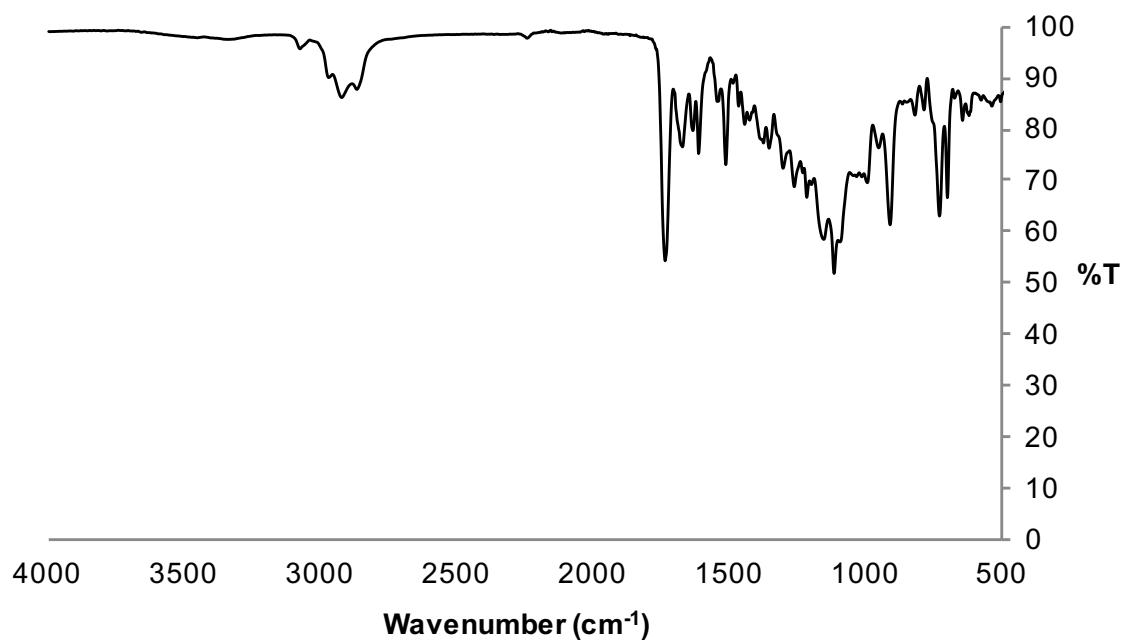
**Figure S20.** ATR-FTIR spectrum of  $\text{PEO}_{45}\text{-}b\text{-CA}_{11}\text{-PTX}_{34}$ .



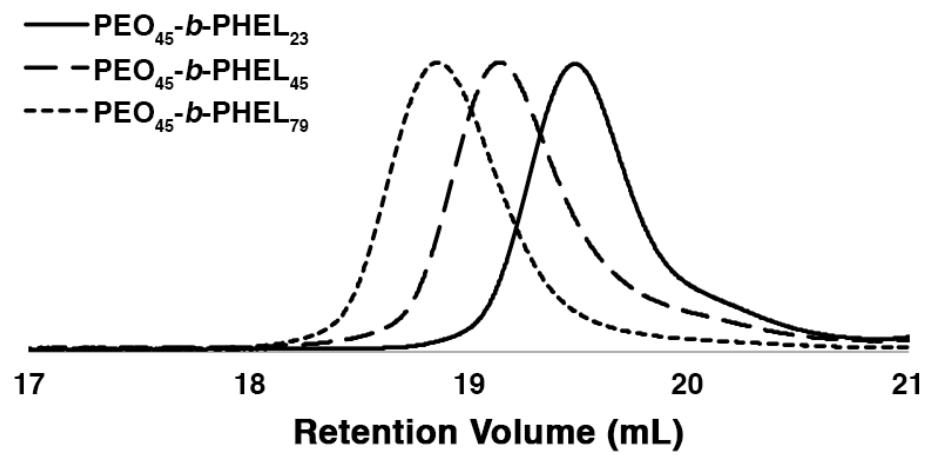
**Figure S21.** ATR-FTIR spectrum of  $\text{PEO}_{45}\text{-}b\text{-PHEL}_{20}\text{-CA}_7\text{-PTX}_{18}$ .



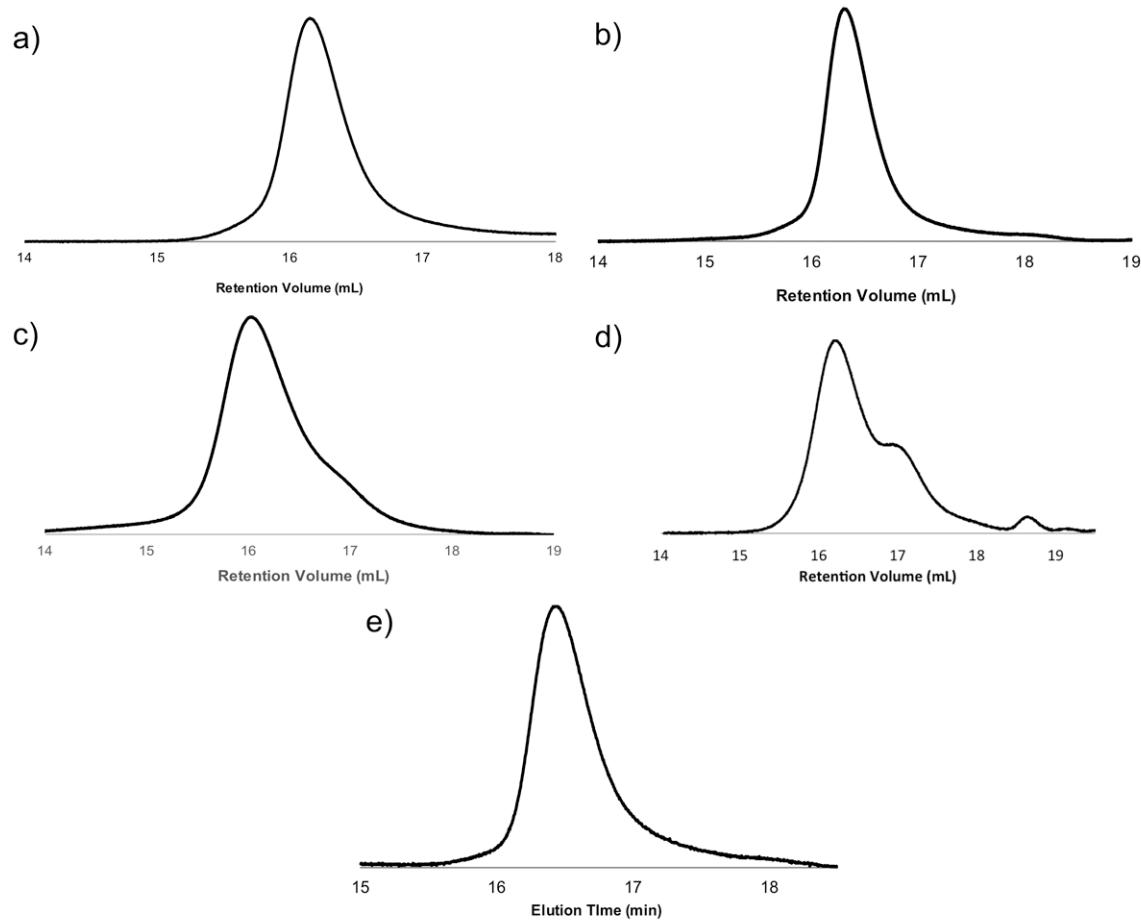
**Figure S22.** ATR-FTIR spectrum of 4.



**Figure S23.** ATR-FTIR spectra of PEO<sub>45</sub>-*b*-PHEL<sub>40</sub>-RHD<sub>5</sub>.



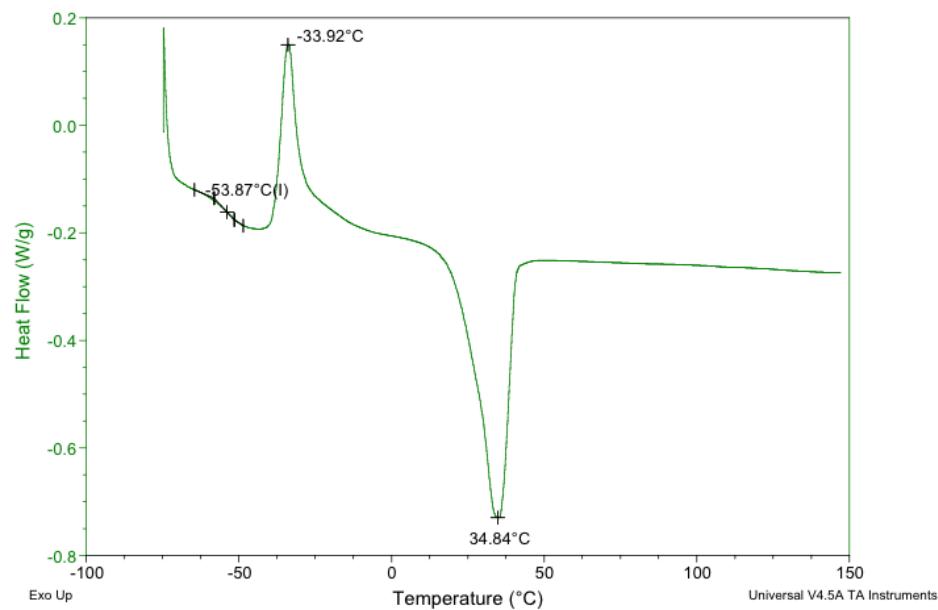
**Figure S24.** SEC traces for PEO<sub>45</sub>-*b*-PHEL<sub>23</sub>, PEO<sub>45</sub>-*b*-PHEL<sub>45</sub> and PEO<sub>45</sub>-*b*-PHEL<sub>79</sub> (THF, refractive index detection).



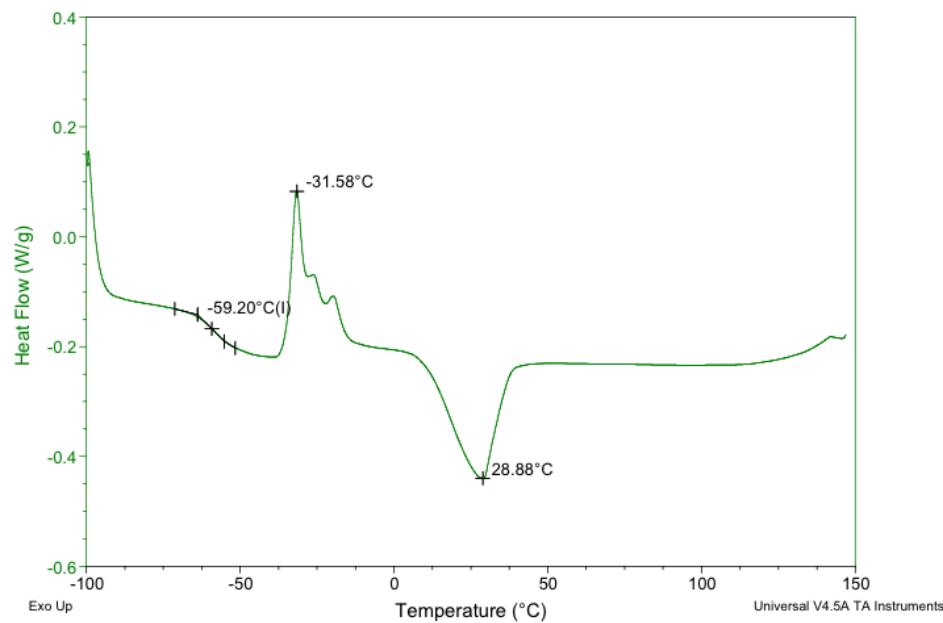
**Figure S25.** SEC traces (THF, refractive index detection) for: a) PEO<sub>45</sub>-*b*-PHEL<sub>21</sub>-octyl<sub>24</sub>, b) PEO<sub>45</sub>-*b*-PHEL<sub>31</sub>-TEG<sub>14</sub>, c) PEO<sub>45</sub>-*b*-CA<sub>11</sub>-PTX<sub>34</sub>, d) PEO<sub>45</sub>-*b*-PHEL<sub>20</sub>-CA<sub>7</sub>-PTX<sub>18</sub>, e) PEO<sub>45</sub>-*b*-PHEL<sub>40</sub>-RHD<sub>5</sub>.

**Table S1.** Summary of the onset degradation temperatures ( $T_o$ ) for PEO-*b*-PHEL and its derivatives as measured by TGA.

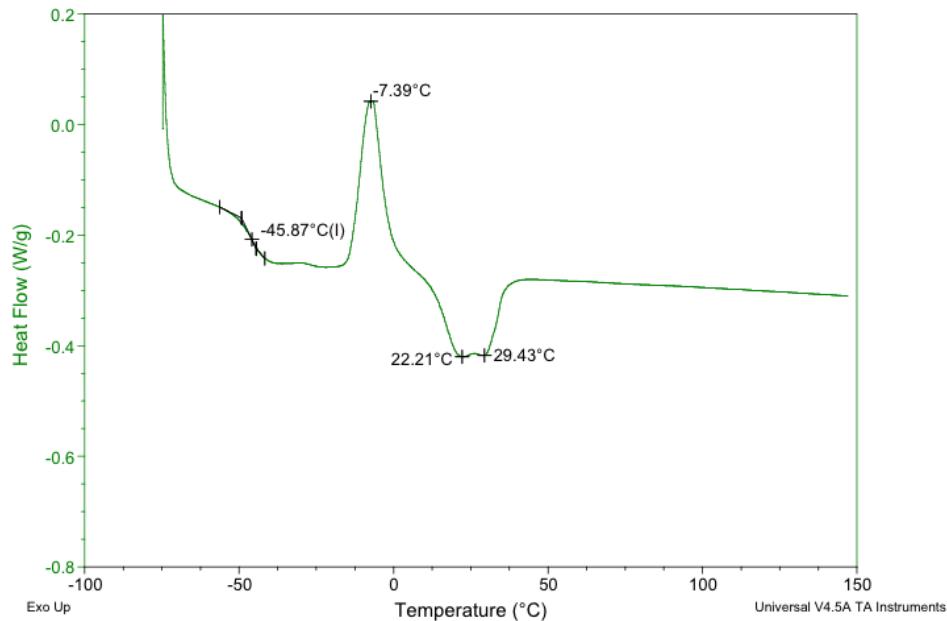
Polymer	$T_o$ (°C)
PEO <sub>45</sub> - <i>b</i> -PHEL <sub>23</sub>	262
PEO <sub>45</sub> - <i>b</i> -PHEL <sub>45</sub>	257
PEO <sub>45</sub> - <i>b</i> -PHEL <sub>79</sub>	262
PEG <sub>45</sub> - <i>b</i> -PHEL <sub>21</sub> -octyl <sub>24</sub>	256
PEG <sub>45</sub> - <i>b</i> -PHEL <sub>31</sub> -TEG <sub>14</sub>	254
PEG <sub>45</sub> - <i>b</i> -CA <sub>45</sub>	246
PEG <sub>45</sub> - <i>b</i> -CA <sub>11</sub> -PTX <sub>34</sub>	221
PEG <sub>45</sub> - <i>b</i> -PHEL <sub>20</sub> -CA <sub>7</sub> -PTX <sub>18</sub>	219
PEG <sub>45</sub> - <i>b</i> -PHEL <sub>40</sub> -RHD <sub>5</sub>	247



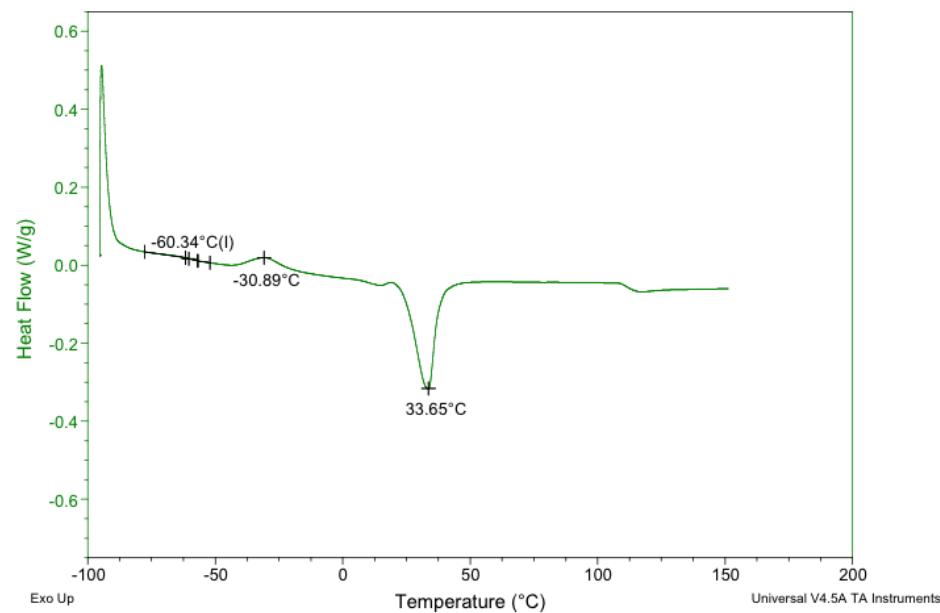
**Figure S26.** DSC trace for PEO<sub>45</sub>-*b*-PHEL<sub>23</sub> (obtained from third heating cycle).



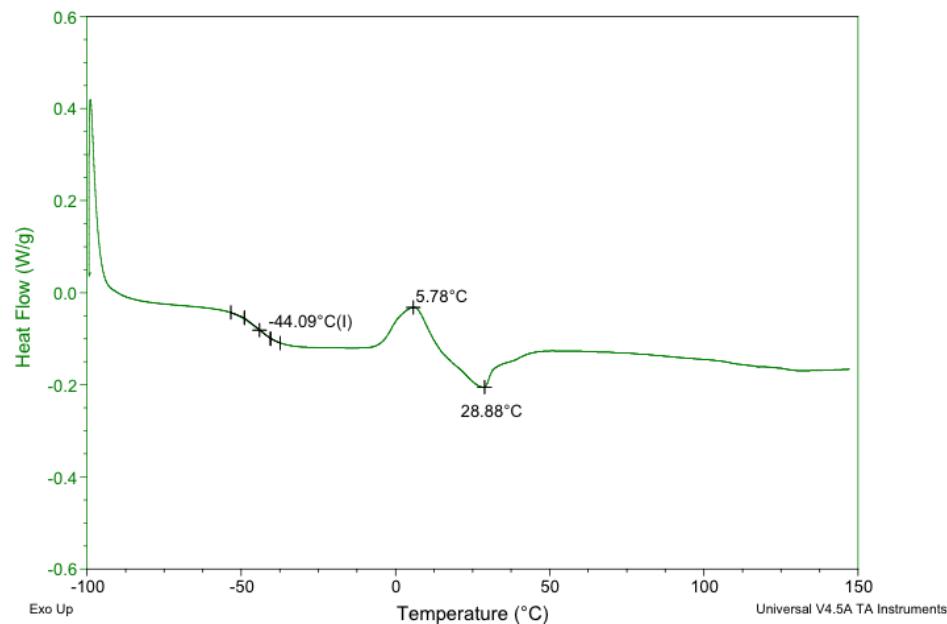
**Figure S27.** DSC trace for PEO<sub>45</sub>-*b*-PHEL<sub>45</sub> (obtained from third heating cycle).



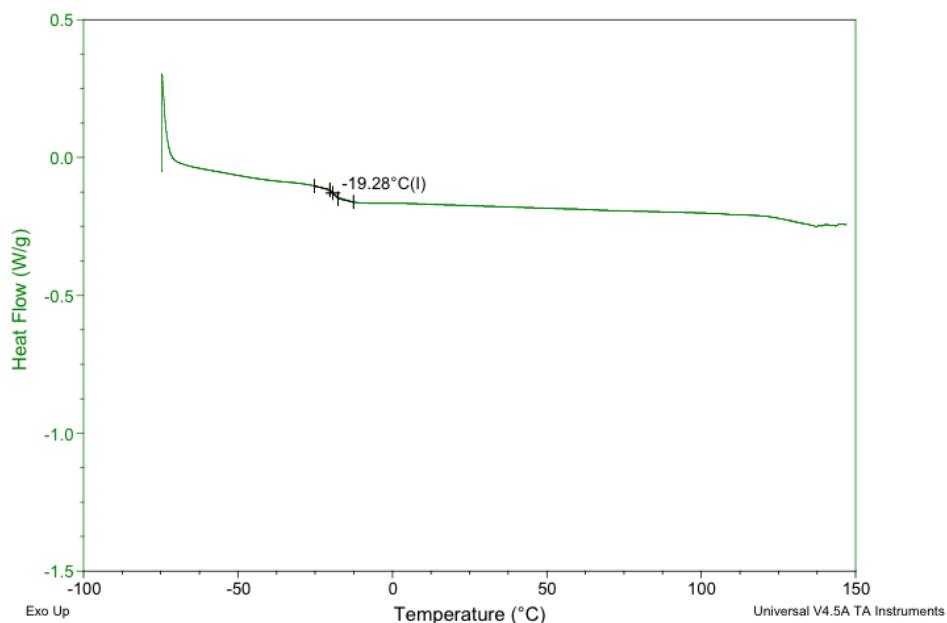
**Figure S28.** DSC trace for  $\text{PEO}_{45}\text{-}b\text{-PHEL}_{79}$  (obtained from third heating cycle).



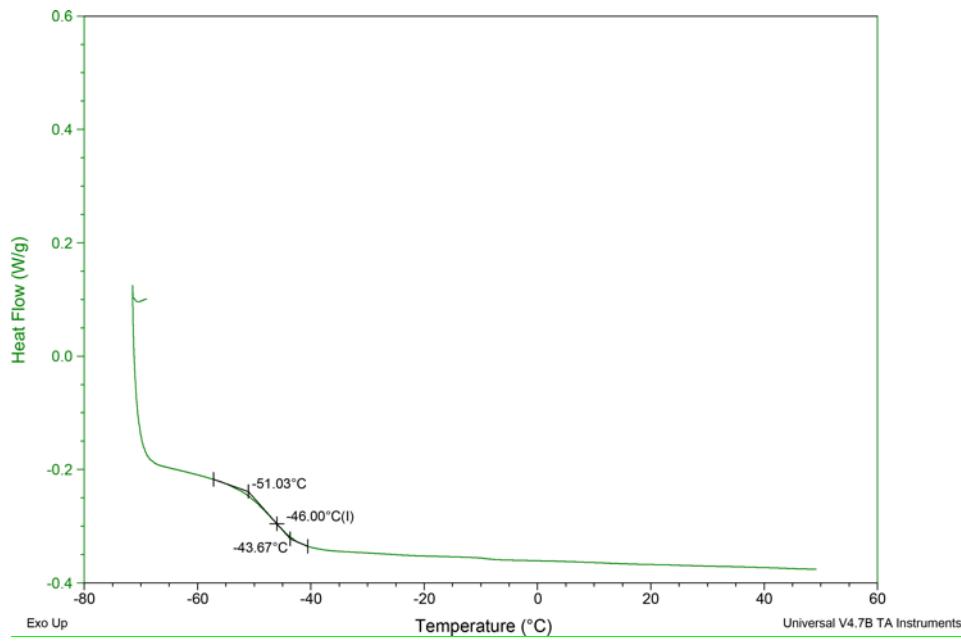
**Figure S29.** DSC trace for  $\text{PEO}_{45}\text{-}b\text{-PHEL}_{21}\text{-octyl}_{24}$  (obtained from fourth heating cycle).



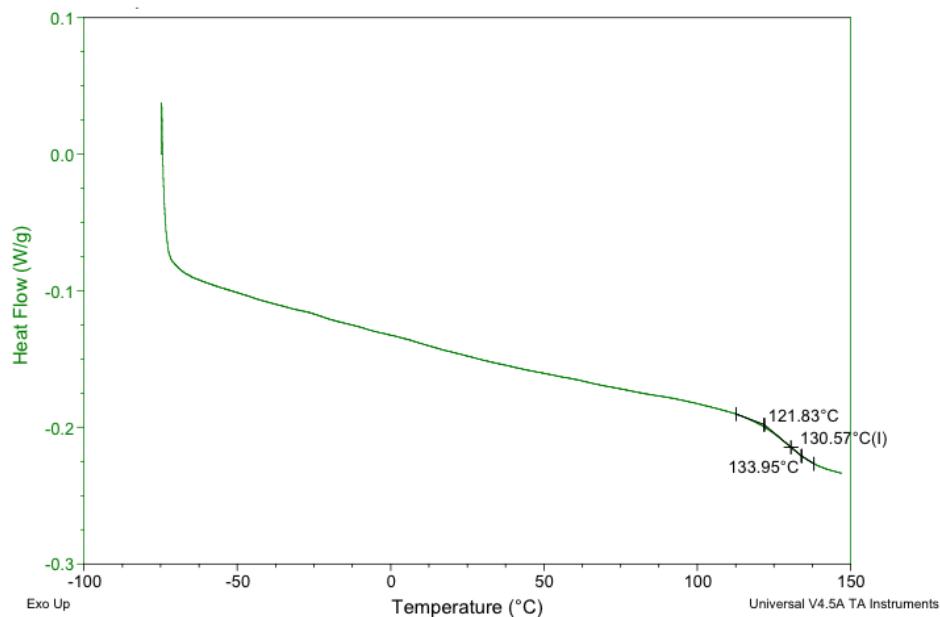
**Figure S30.** DSC trace for PEO<sub>45</sub>-*b*-PHEL<sub>31</sub>-TEG<sub>14</sub> (obtained from fourth heating cycle).



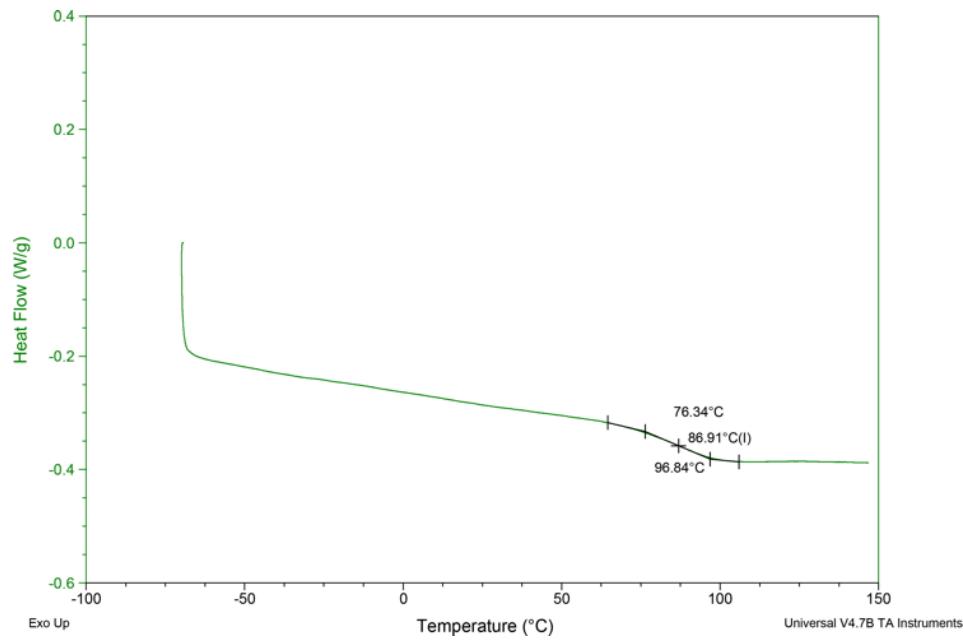
**Figure S31.** DSC trace for PEO<sub>45</sub>-*b*-CA<sub>45</sub> (obtained from third heating cycle).



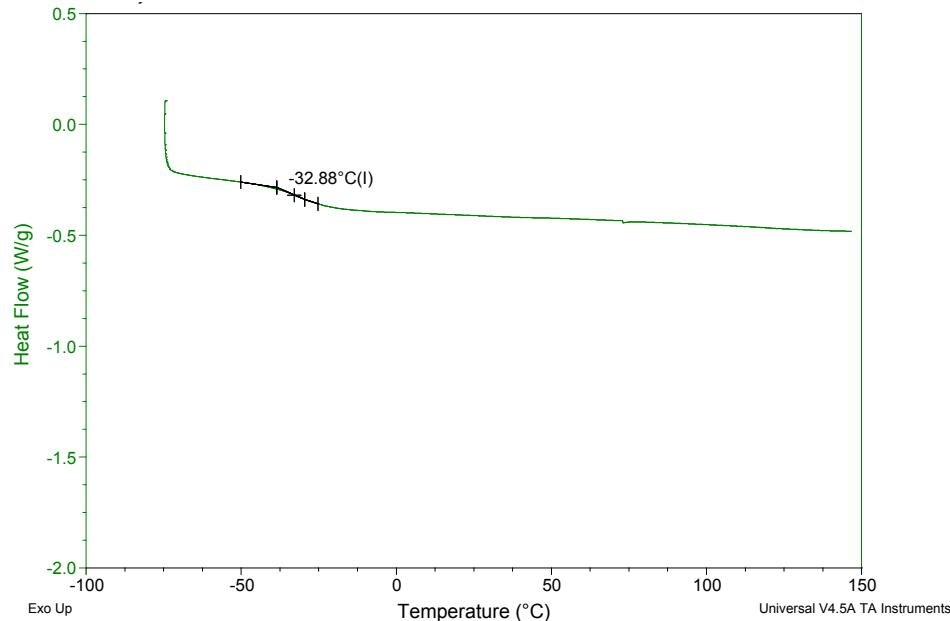
**Figure S32.** DSC trace for  $\text{PEO}_{45}\text{-}b\text{-PHEL}_{20}\text{-CA}_{25}$  (obtained from third heating cycle).



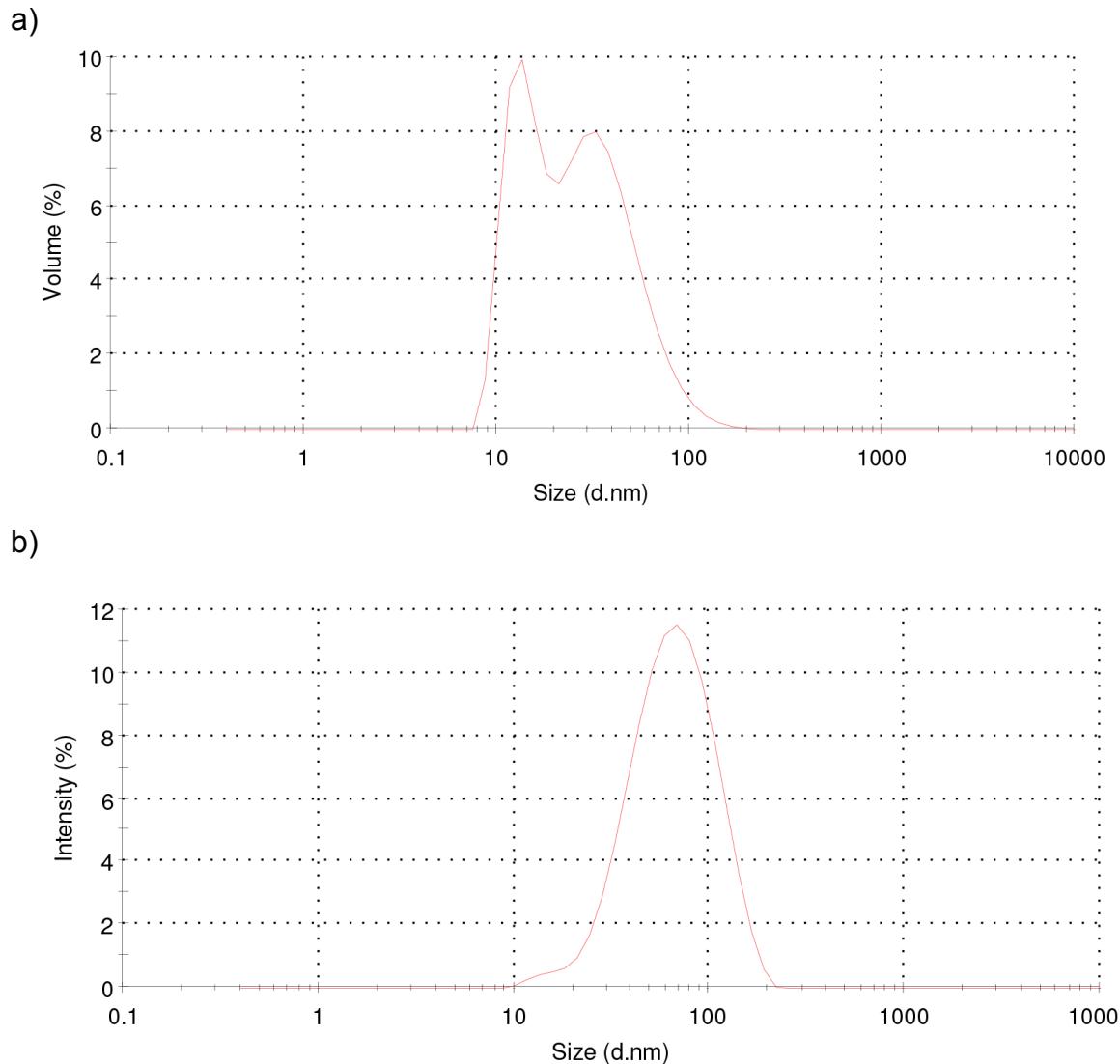
**Figure S33.** DSC trace for  $\text{PEO}_{45}\text{-}b\text{-CA}_{11}\text{-PTX}_{34}$  (obtained from third heating cycle).



**Figure S34.** DSC trace for  $\text{PEO}_{45}\text{-}b\text{-PHEL}_{20}\text{-CA}_7\text{-PTX}_{18}$  (obtained from third heating cycle).

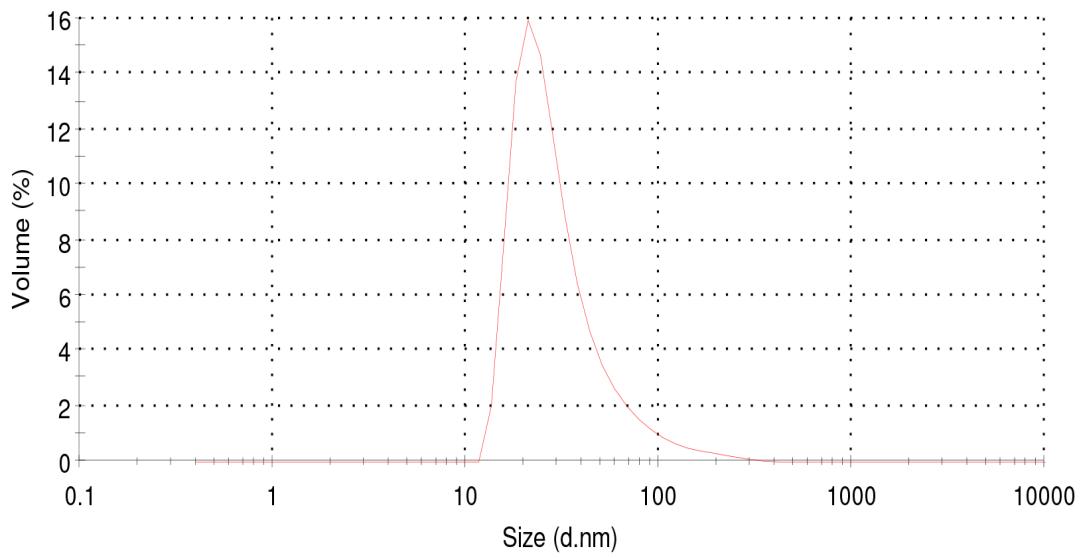


**Figure S35.** DSC trace for  $\text{PEO}_{45}\text{-}b\text{-PHEL}_{40}\text{-RHD}_5$  (obtained from third heating cycle).

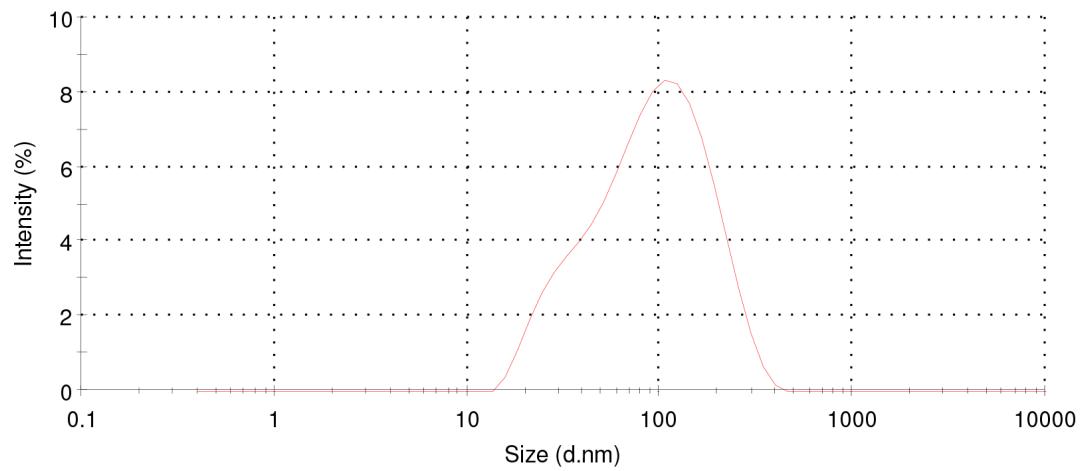


**Figure S36.** a) Volume and b) intensity distribution of hydrodynamic diameter measured by DLS for PEO<sub>45</sub>-*b*-PHEL<sub>23</sub>.

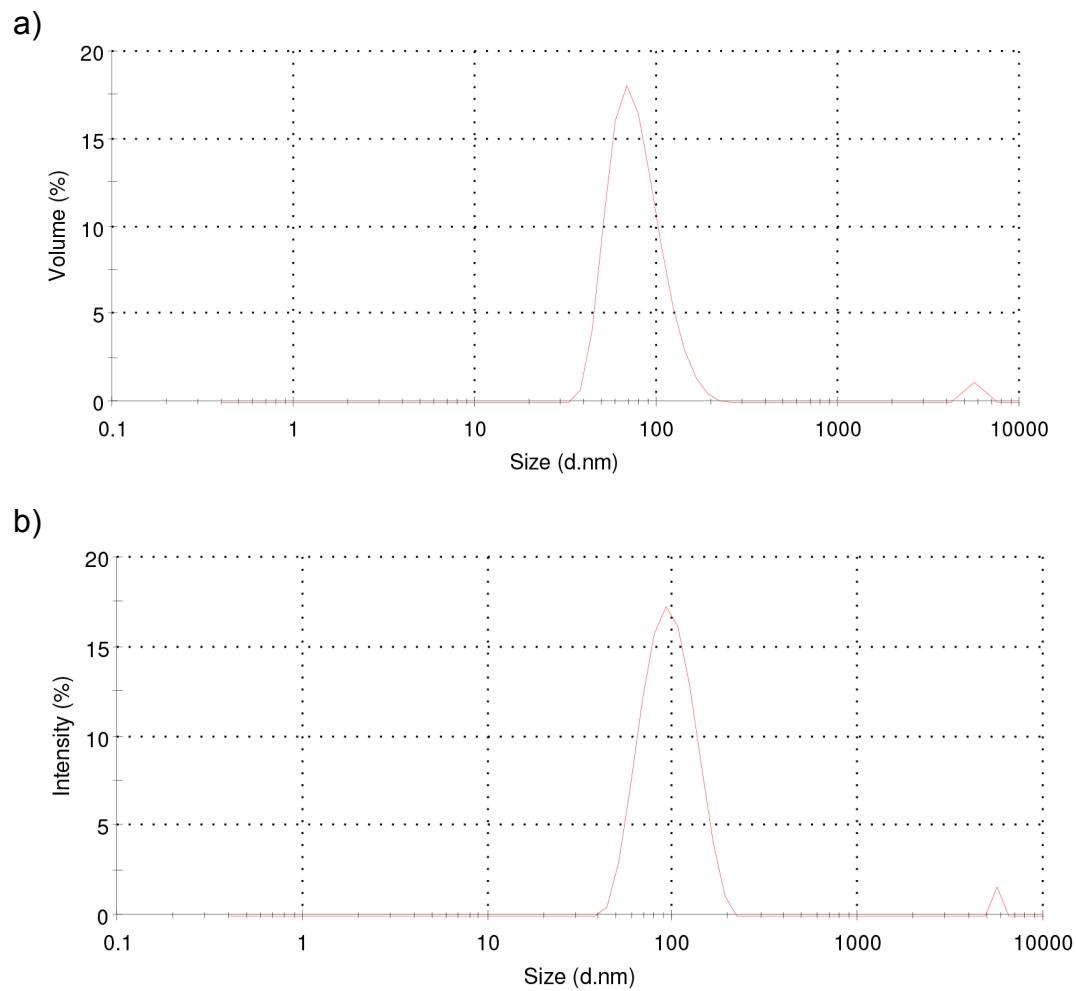
a)



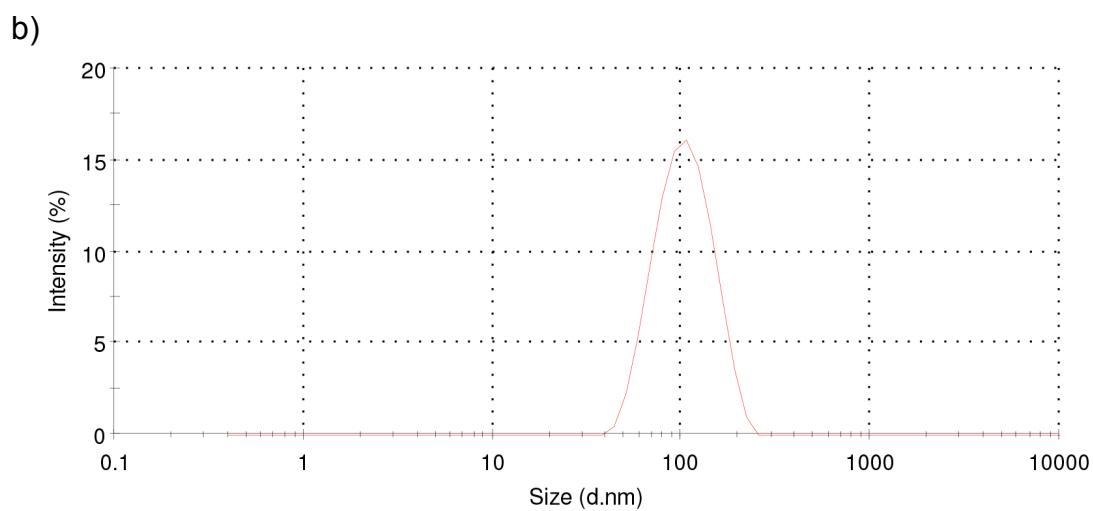
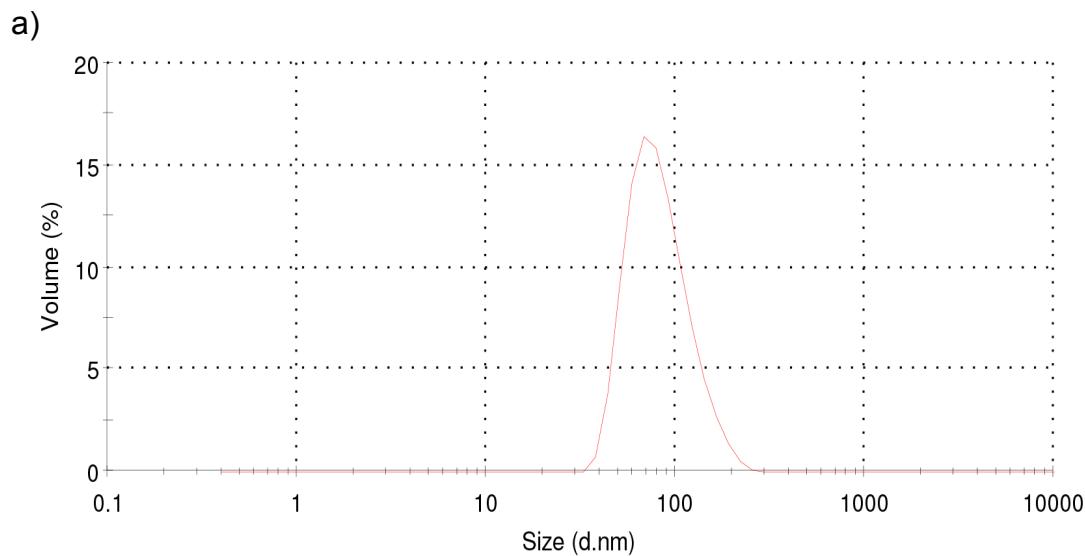
b)



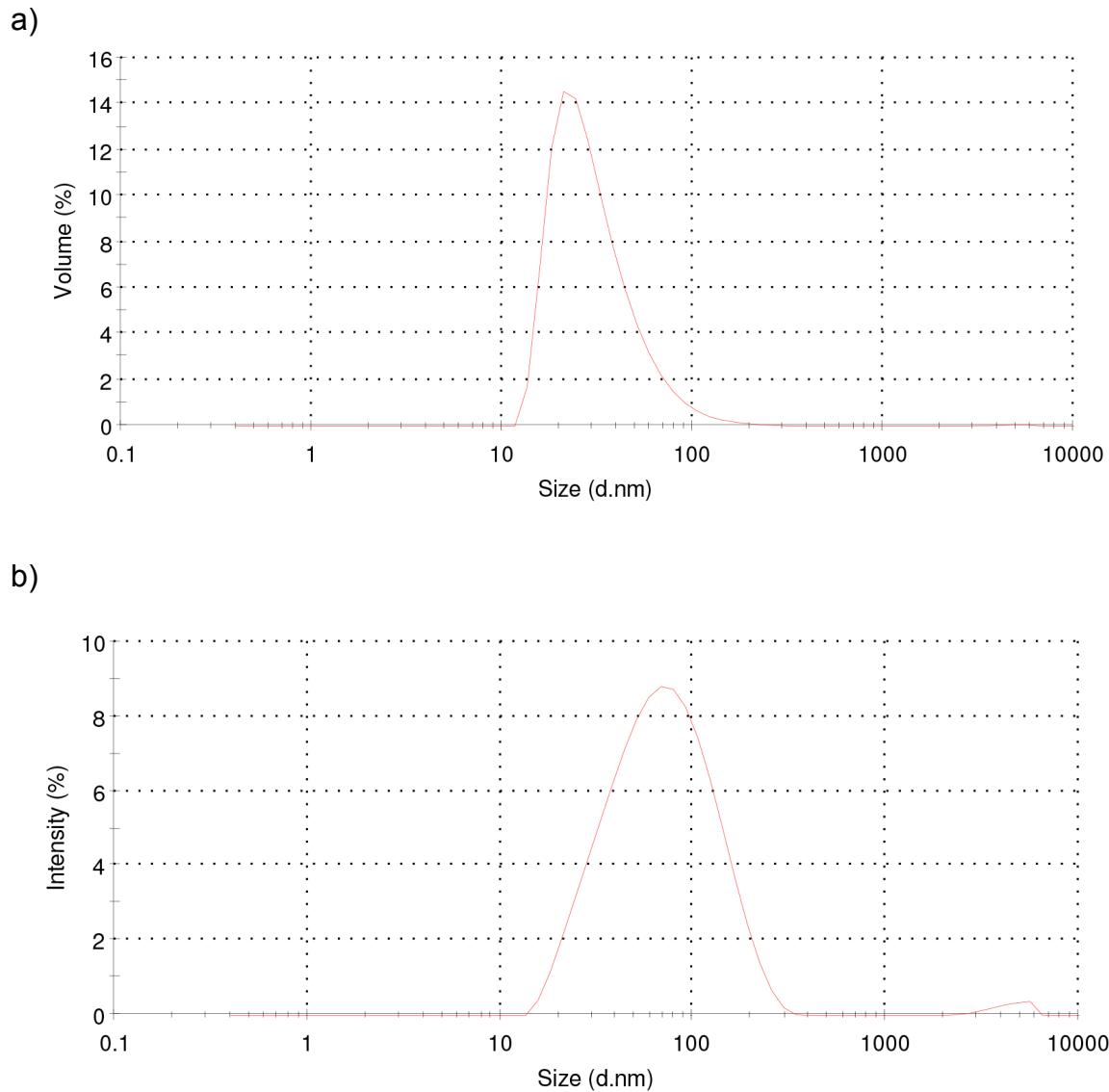
**Figure S37.** a) Volume and b) intensity distribution of hydrodynamic diameter measured by DLS for PEO<sub>45</sub>-*b*-PHEL<sub>45</sub>.



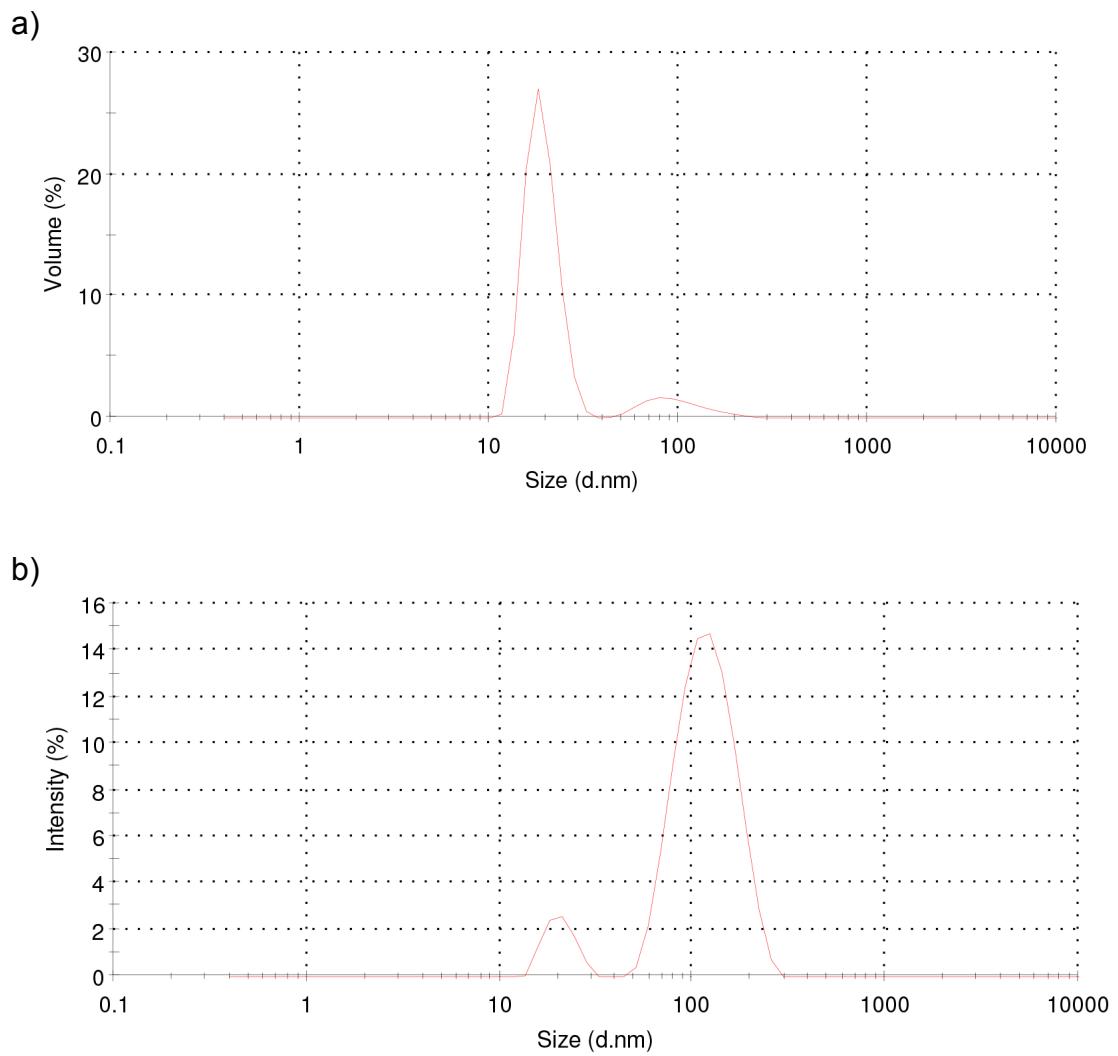
**Figure S38.** a) Volume and b) intensity distribution of hydrodynamic diameter measured by DLS for  $\text{PEO}_{45}-b\text{-PHEL}_{79}$ .



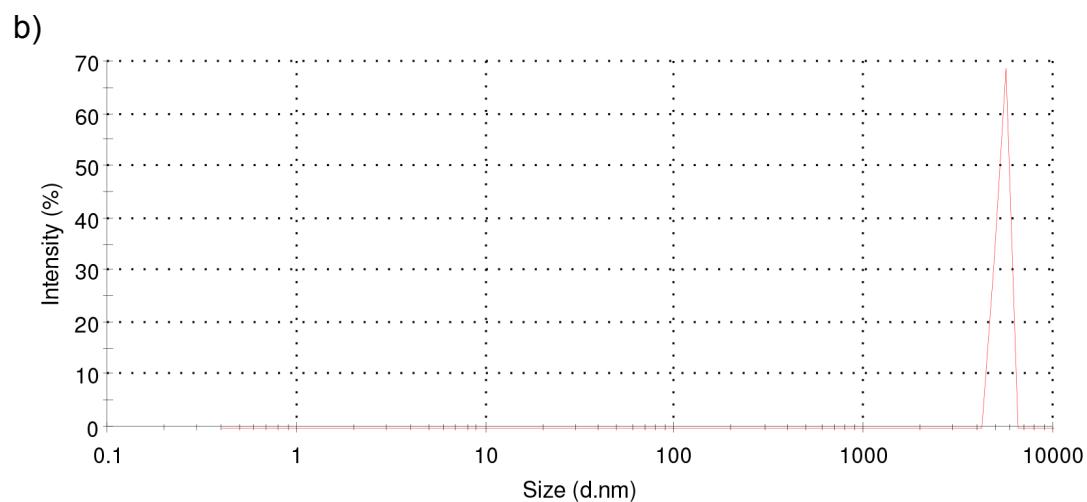
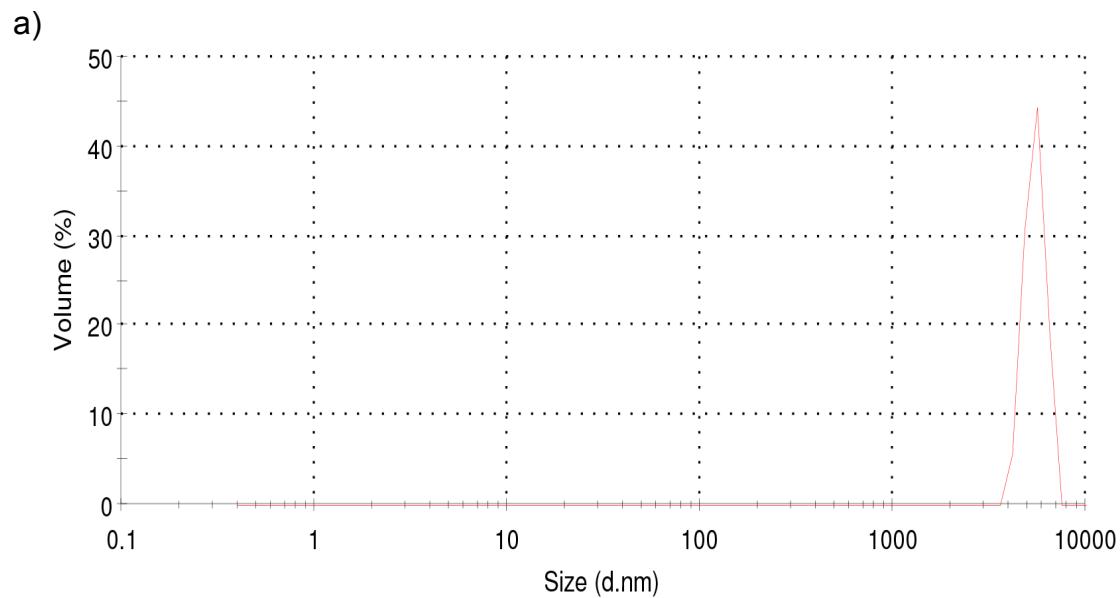
**Figure S39.** a) Volume and b) intensity distribution of hydrodynamic diameter measured by DLS for  $\text{PEG}_{45}-b\text{-PHEL}_{21}\text{-octyl}_{24}$ .



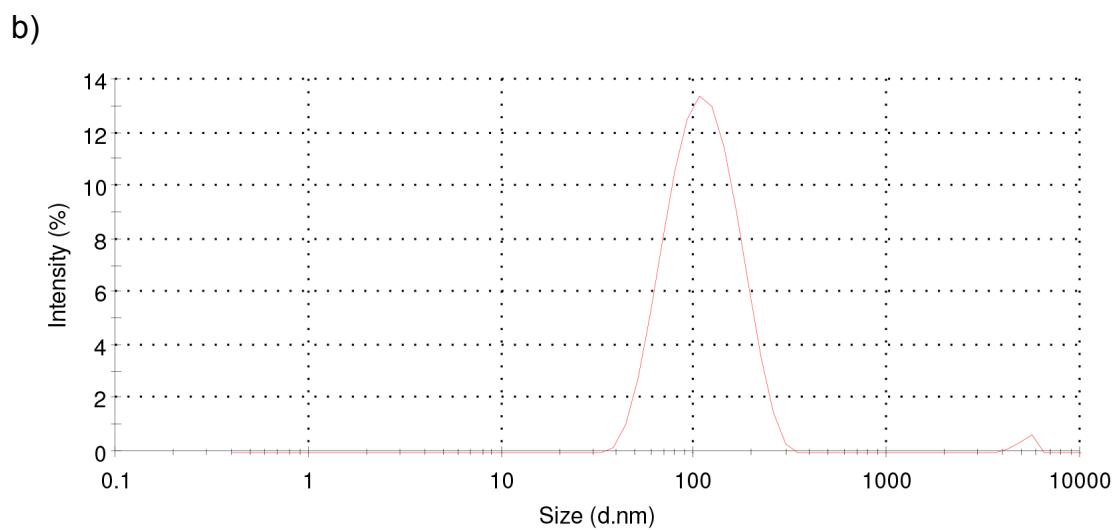
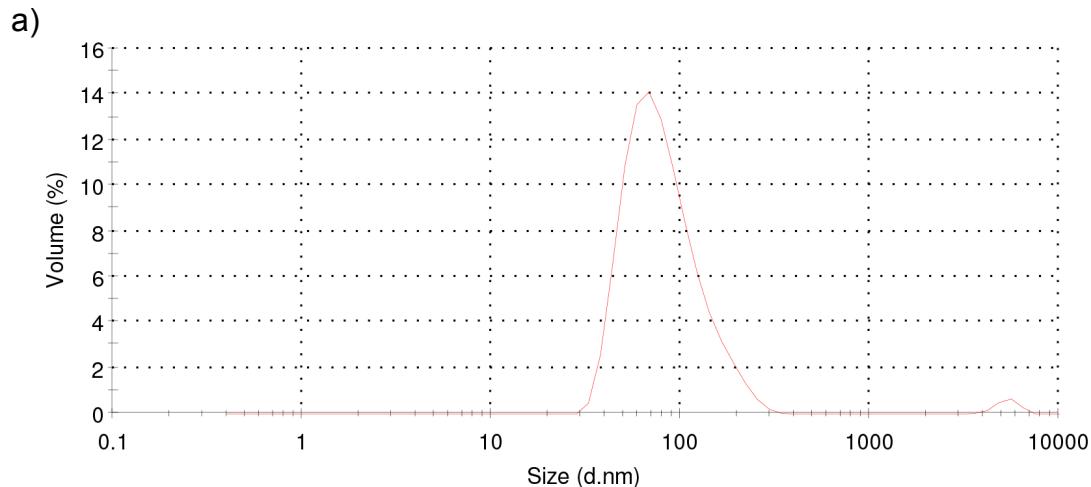
**Figure S40.** a) Volume and b) intensity distribution of hydrodynamic diameter measured by DLS for PEG<sub>45</sub>-*b*-PHEL<sub>31</sub>-TEG<sub>14</sub>.



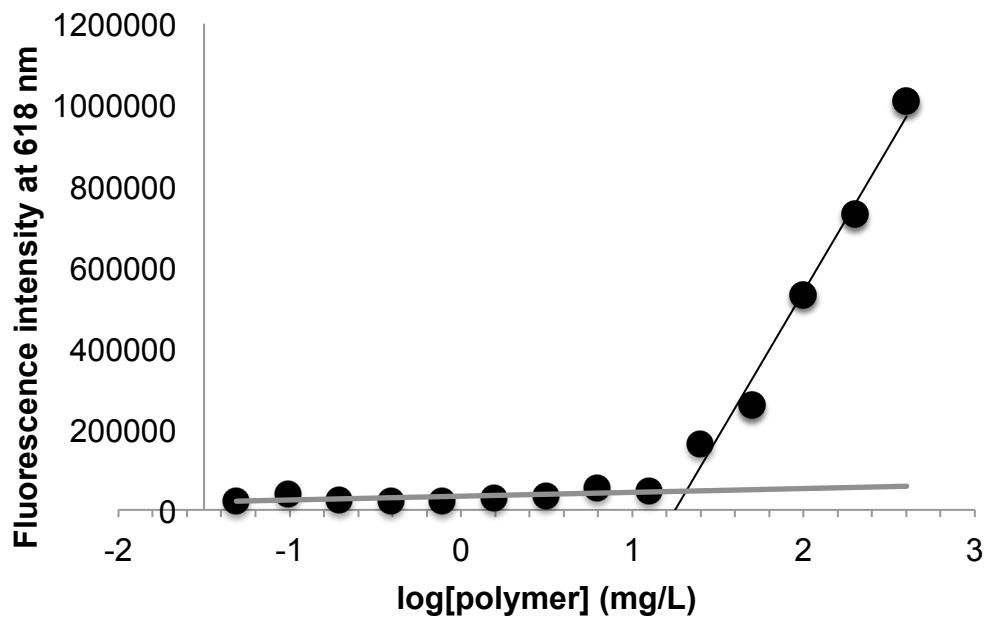
**Figure S41.** a) Volume and b) intensity distribution of hydrodynamic diameter measured by DLS for PEG<sub>45</sub>-*b*-PHEL<sub>20</sub>-CA<sub>25</sub>.



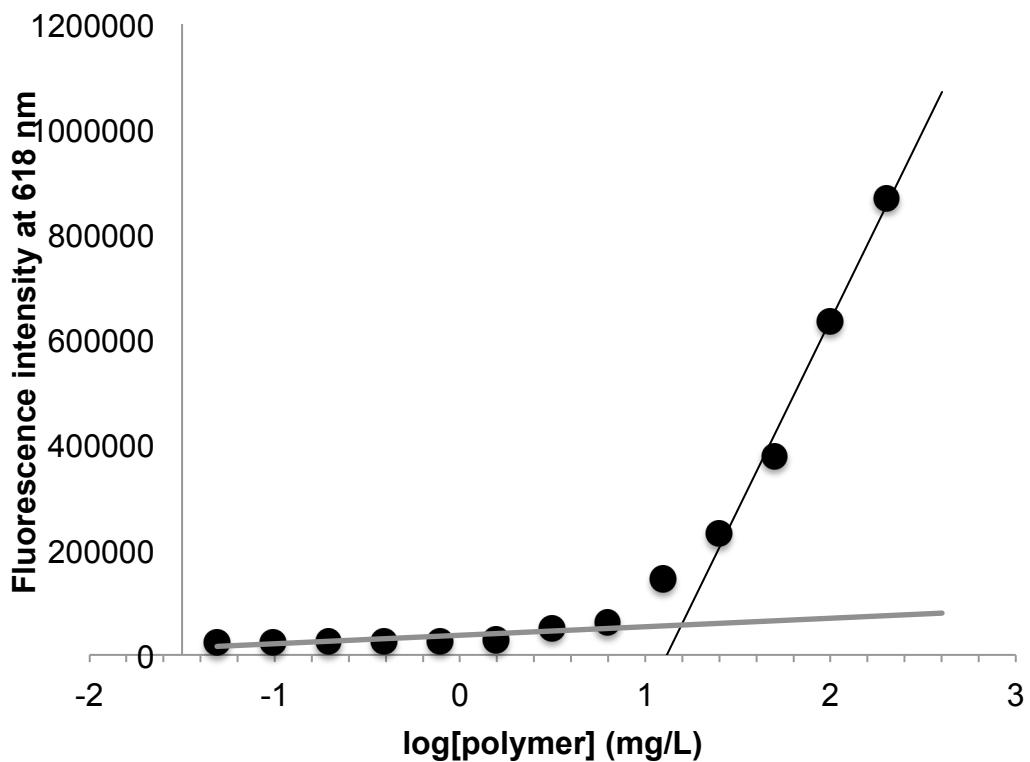
**Figure S42.** a) Volume and b) intensity distribution of hydrodynamic diameter measured by DLS for PEG<sub>45</sub>-*b*-PHEL<sub>20</sub>-CA<sub>7</sub>-PTX<sub>18</sub>.



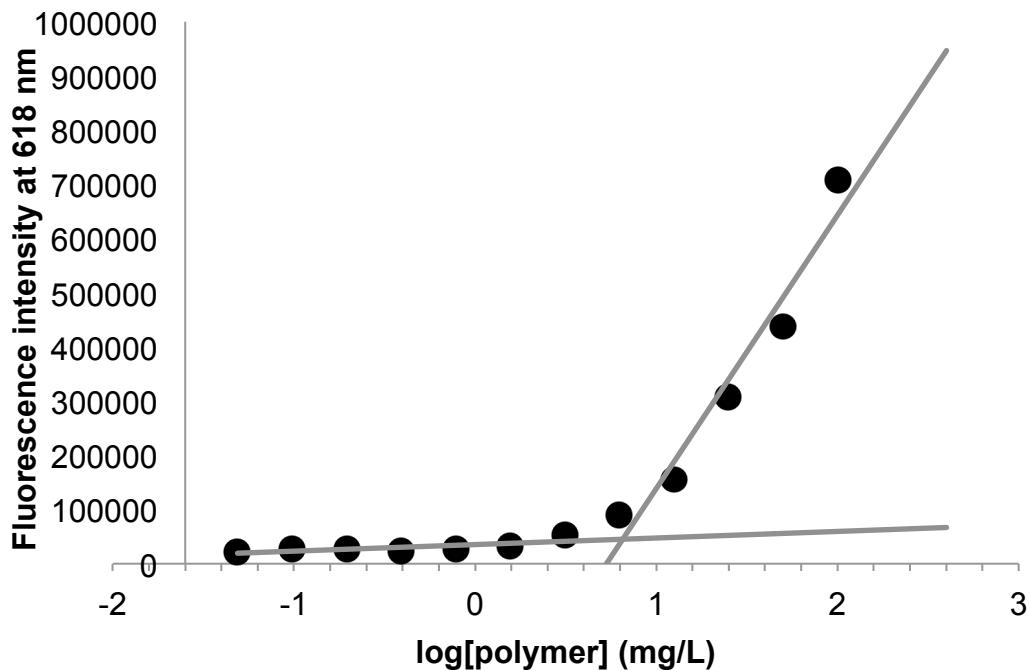
**Figure S43.** a) Volume and b) intensity distribution of hydrodynamic diameter measured by DLS for PEG<sub>45</sub>-*b*-PHEL<sub>40</sub>-RHD<sub>5</sub>.



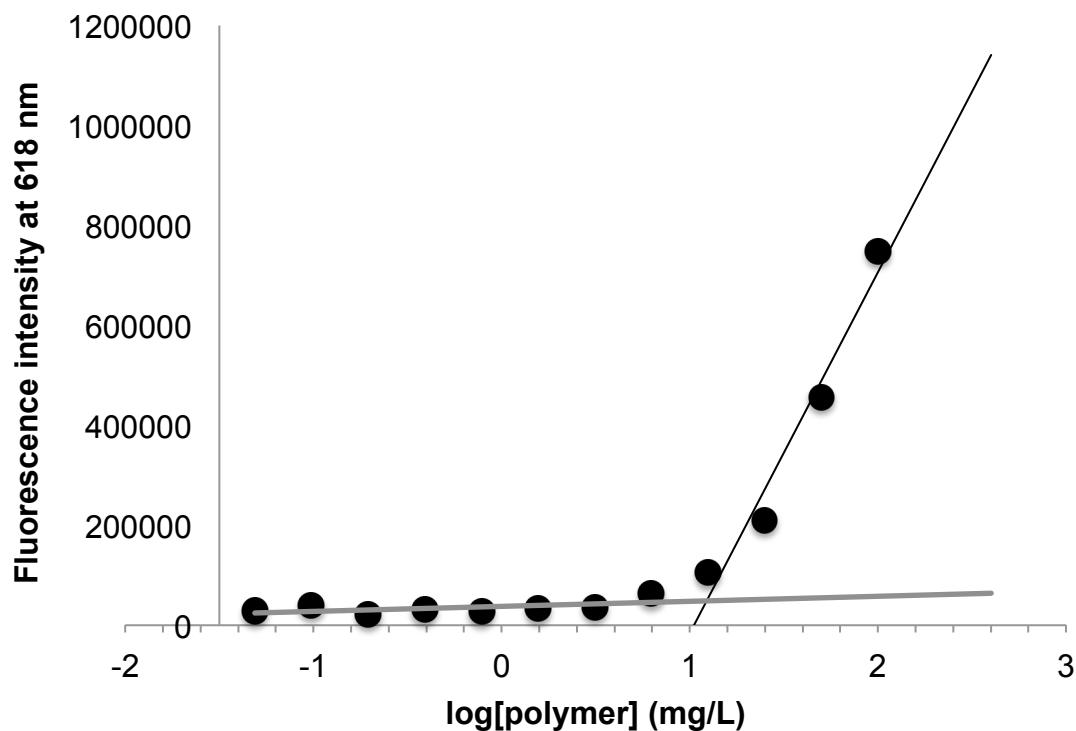
**Figure S44.** Fluorescence intensity of nile red versus log(concentration of  $\text{PEO}_{45}\text{-}b\text{-PHEL}_{23}$ ). The intersection of the two linear regions provides the CAC.



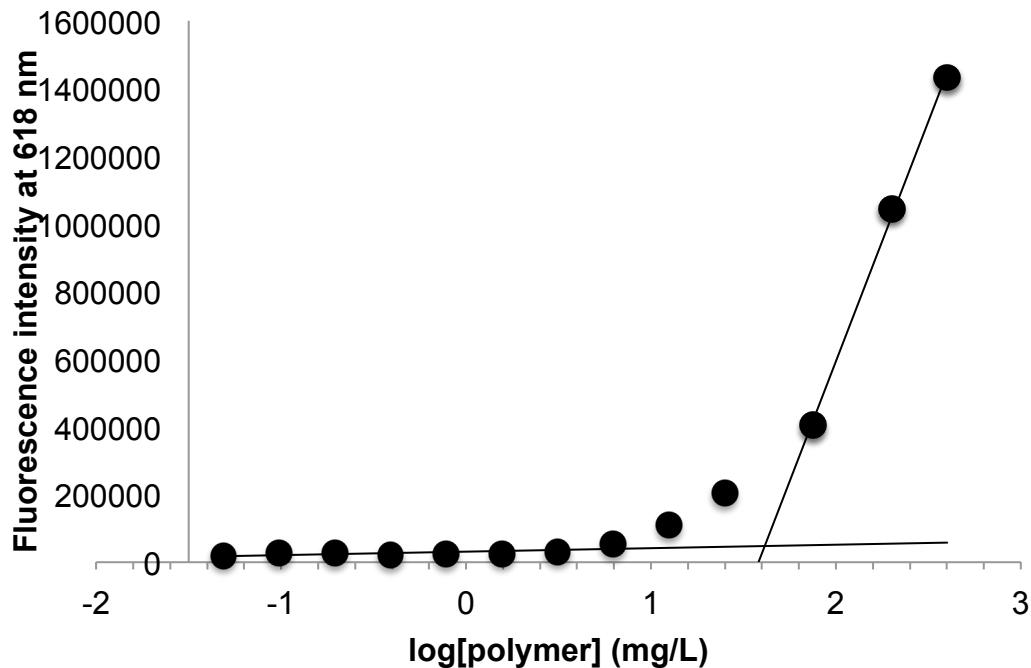
**Figure S45.** Fluorescence intensity of nile red versus log(concentration of  $\text{PEO}_{45}\text{-}b\text{-PHEL}_{45}$ ). The intersection of the two linear regions provides the CAC.



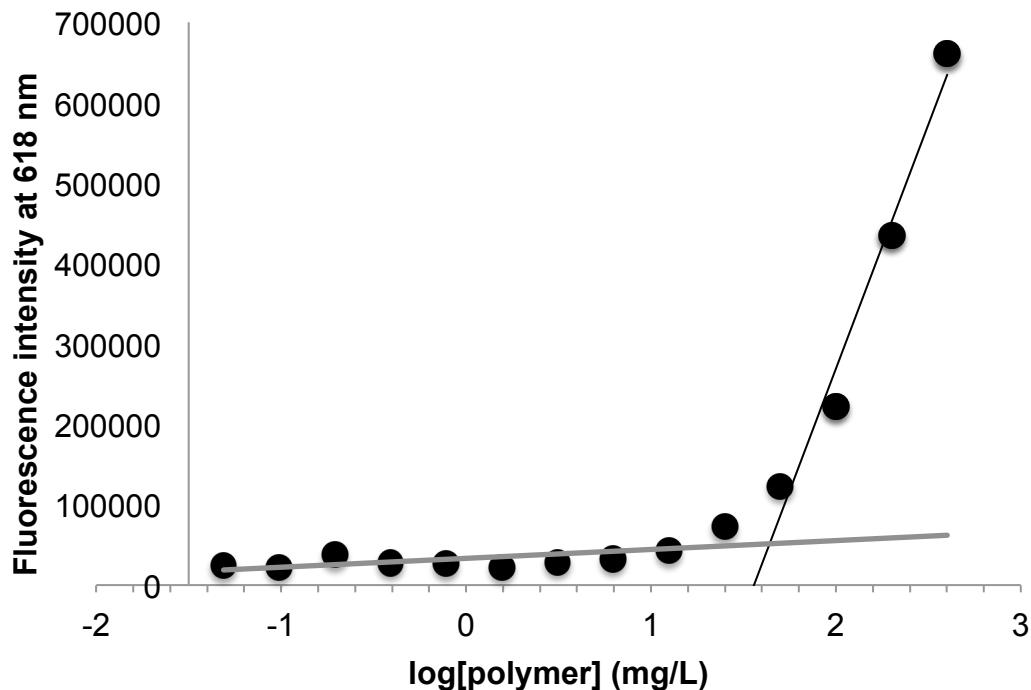
**Figure S46.** Fluorescence intensity of nile red versus log(concentration of  $\text{PEO}_{45}\text{-}b\text{-PHEL}_{79}$ ). The intersection of the two linear regions provides the CAC.



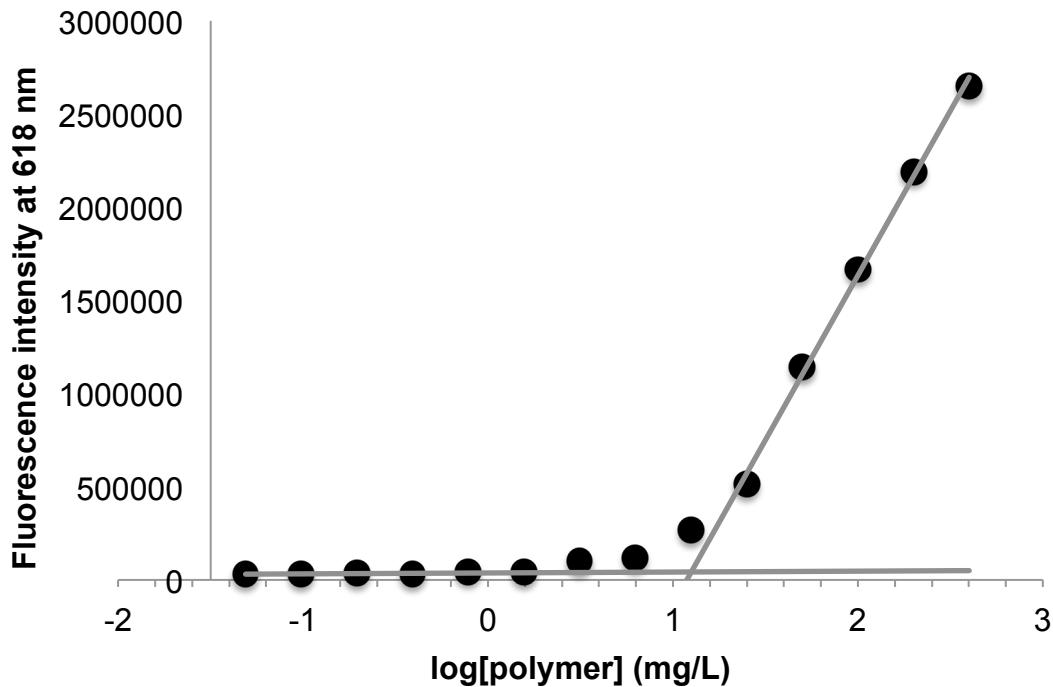
**Figure S47.** Fluorescence intensity of nile red versus log(concentration of  $\text{PEO}_{45}\text{-}b\text{-PHEL}_{21\text{octyl}}_{24}$ ). The intersection of the two linear regions provides the CAC.



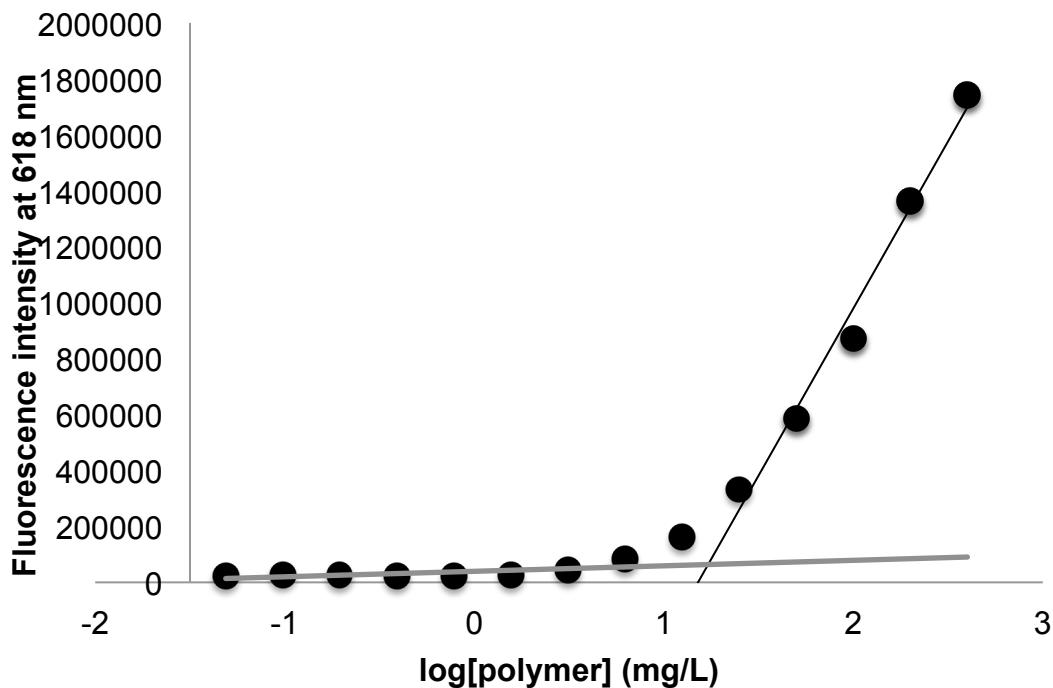
**Figure S48.** Fluorescence intensity of nile red versus  $\log(\text{concentration of PEO}_{45}\text{-}b\text{-PHEL}_{31}\text{-TEG}_{14})$ . The intersection of the two linear regions provides the CAC.



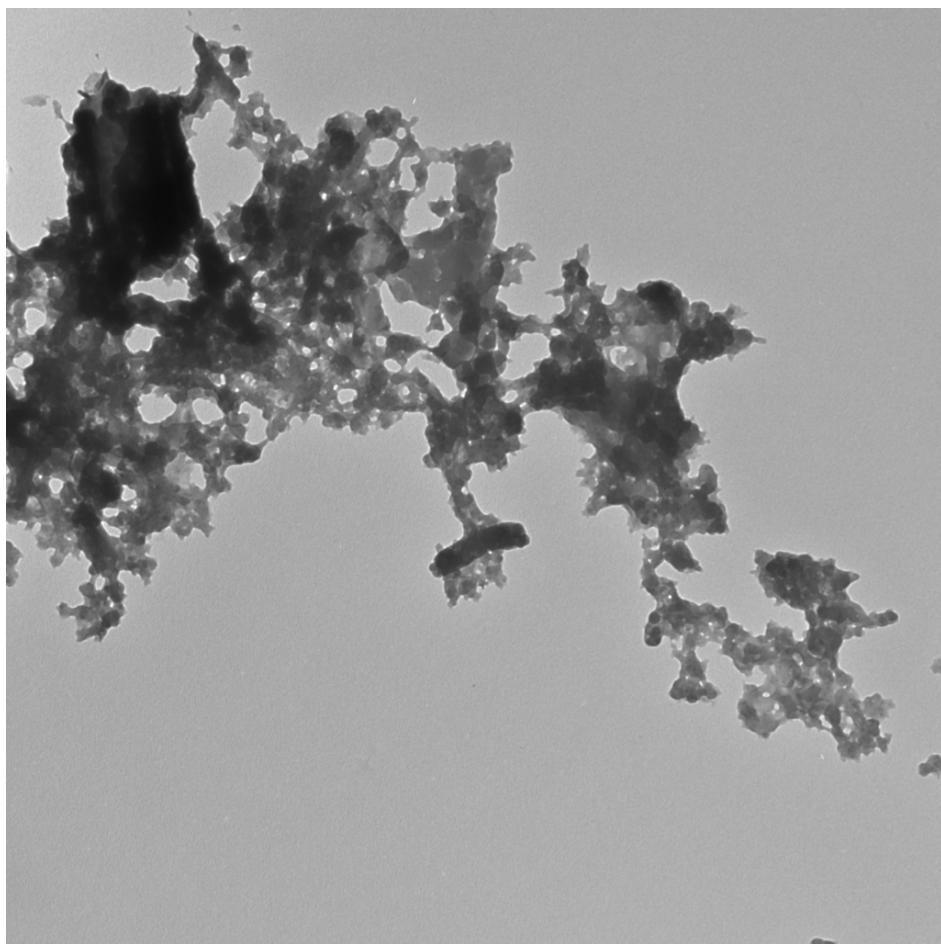
**Figure S49.** Fluorescence intensity of nile red versus  $\log(\text{concentration of PEO}_{45}\text{-}b\text{-PHEL}_{20}\text{-CA}_{25})$ . The intersection of the two linear regions provides the CAC.



**Figure S50.** Fluorescence intensity of nile red versus  $\log(\text{concentration of PEO}_{45}\text{-}b\text{-PHEL}_{20}\text{-CA}_7\text{-PTX}_{18})$ . The intersection of the two linear regions provides the CAC.



**Figure S51.** Fluorescence intensity of nile red versus  $\log(\text{concentration of PEO}_{45}\text{-}b\text{-PHEL}_{40}\text{-RHD}_5)$ . The intersection of the two linear regions provides the CAC.



H-1.tif

H

Print Mag: 26500x @ 7.0 in

16:27 09/23/16

TEM Mode: Imaging

500 nm

HV=80.0kV

Direct Mag: 13500x

**Figure S52.** TEM image of assemblies formed from  $\text{PEO}_{45}-b\text{-PHEL}_{20}\text{-CA}_7\text{-PTX}_{18}$  upon solvent exchange from THF/water.

## References

1. X. Zhang, Y. Shiraishi and T. Hirai, *Org. Lett.*, 2007, **9**, 5039-5042.