

## Supporting Information

### **Manganese Protoporphyrin IX Reconstituted Myoglobin Capable of Epoxidation of C=C Bond with Oxone®**

**Yuan-Bo Cai,<sup>a</sup> Si-Yu Yao,<sup>a</sup> Mo Hu<sup>b</sup>, Xiaoyun Liu<sup>\*b</sup>, and Jun-Long Zhang<sup>\*a</sup>**

*<sup>a</sup> Beijing National Laboratory for Molecular Sciences, State Key  
Laboratory of Rare Earth Materials Chemistry and Applications, College  
of Chemistry and Molecular Engineering, Peking University, Beijing  
100871, P. R. China.*

Fax: (+86) 10-62767034

E-mail: [zhangjunlong@pku.edu.cn](mailto:zhangjunlong@pku.edu.cn)

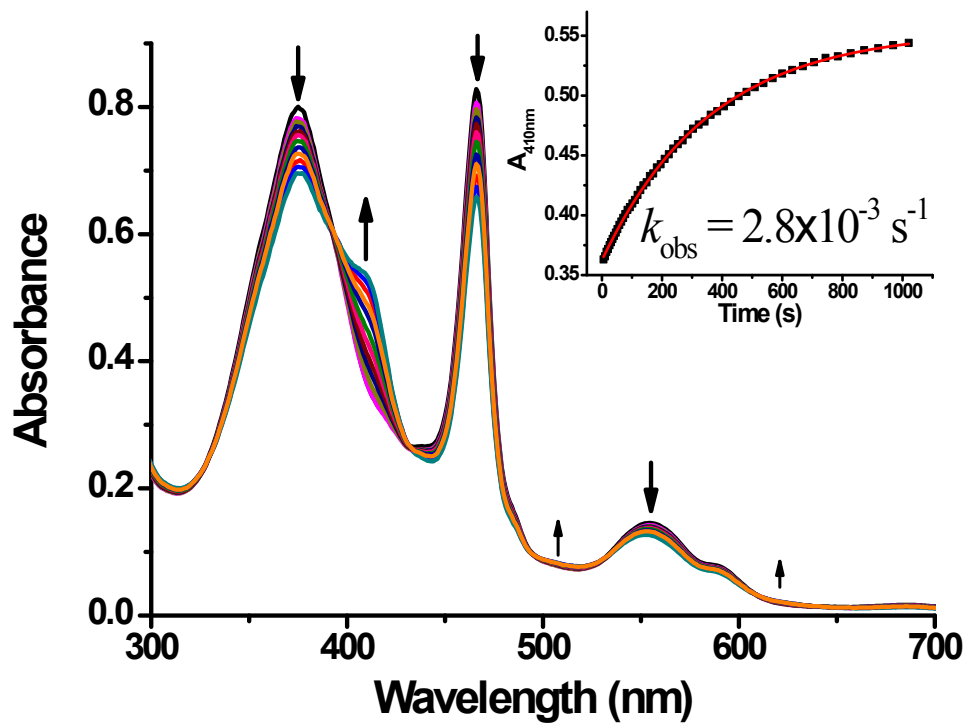
*<sup>b</sup> Institute of Analytical Chemistry and Synthetic and Functional  
Biomolecules Center, College of Chemistry and Molecular Engineering,  
Peking University, Beijing 100871, P. R. China*

Fax: (+86) 10-62759813

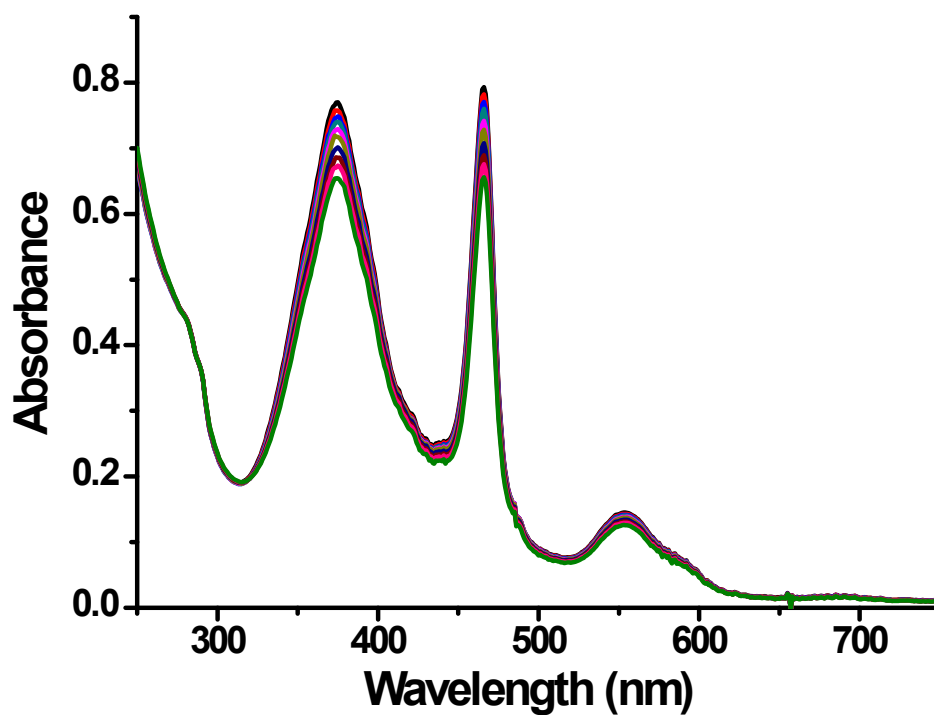
E-mail: [xiaoyun.liu@pku.edu.cn](mailto:xiaoyun.liu@pku.edu.cn)

## Table of Contents

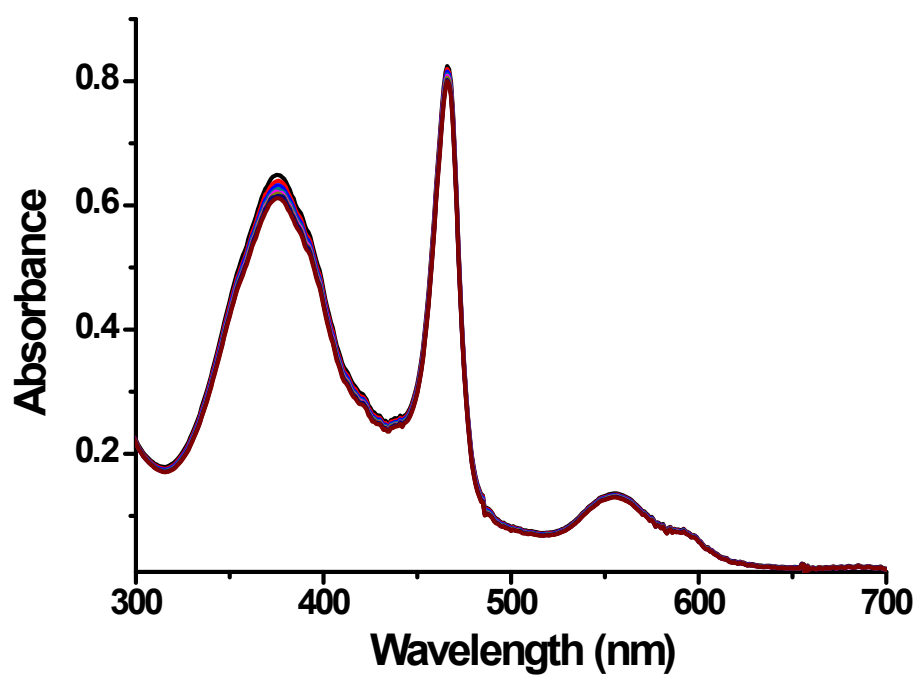
1	Kinetic studies of His3 Mn <sup>III</sup> Mb reacted with 20 equiv KHSO <sub>5</sub> .....	3
2.	Kinetic studies of F43H Mn <sup>III</sup> Mb reacted with 200 equiv KHSO <sub>5</sub> .....	4
3.	Kinetic studies of L29H Mn <sup>III</sup> Mb reacted with 200 equiv KHSO <sub>5</sub> .....	5
4.	Kinetic studies of H64F Mn <sup>III</sup> Mb reacted with 200 equiv KHSO <sub>5</sub> .....	6
5	Kinetic studies of F43H/H64F Mn <sup>III</sup> Mb reacted with 200 equiv KHSO <sub>5</sub> .....	7
6	Kinetic studies of L29H/H64F Mn <sup>III</sup> Mb reacted with 200 equiv KHSO <sub>5</sub> .....	8
7	Kinetic studies of L29H/F43H/H64F Mn <sup>III</sup> Mb reacted with 200 equiv KHSO <sub>5</sub> ...	9
8	ESI-MS of apo His3 Mb with 20 equiv KHSO <sub>5</sub> .....	10
9	LC/MS peaks of oxidized/unoxidized peptide HPGDFGADAQGAMNK.....	11
10	Fragmentation patterns of peptide HPGDFGADAQGAM(Ox)NK.....	13
11	<sup>1</sup> H NMR of styrene epoxidation catalyzed by His3 Mn <sup>III</sup> Mb.....	14
12	Conversion and yield of styrene epoxidation at different pHs.....	15



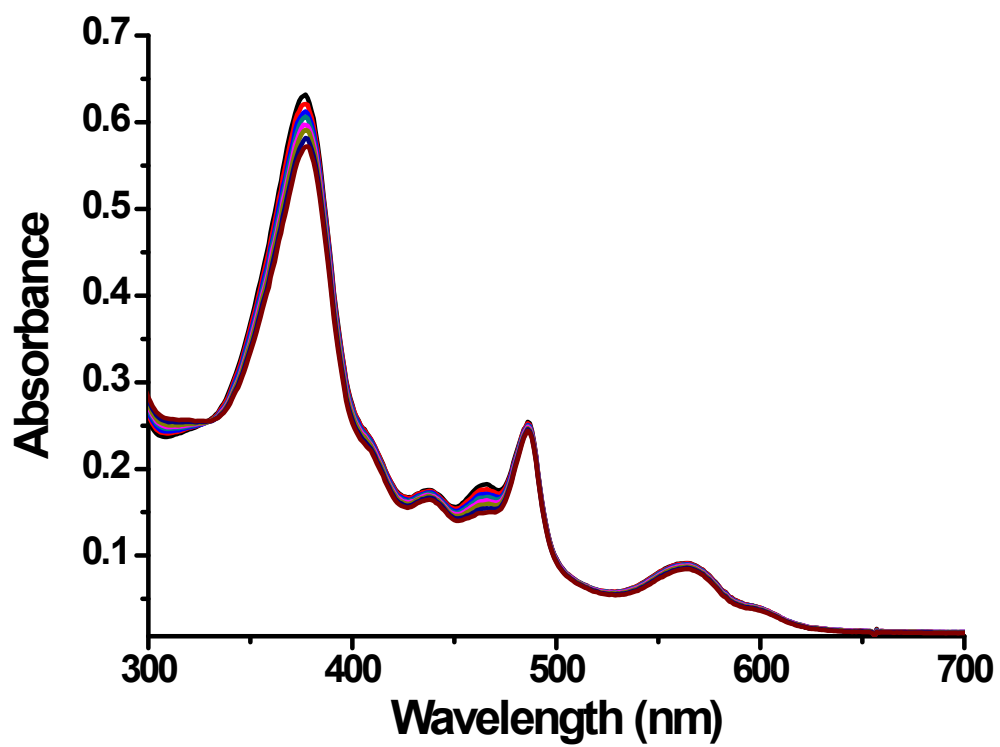
**Figure S1** Kinetic studies of 30  $\mu\text{M}$  His3 Mn<sup>III</sup>Mb reacted with 20 equiv KHSO<sub>5</sub> in PBS buffer (pH=7.4). Inset: Time course plots of absorbance at 410 nm, and corresponding rate constants of Mn=O formation.



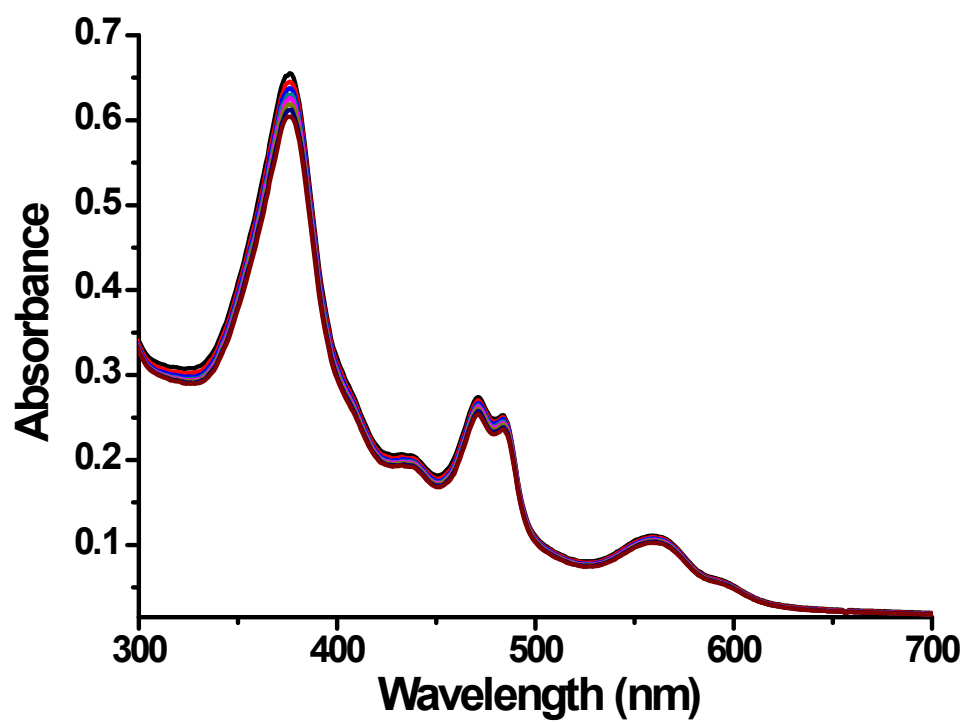
**Figure S2** Kinetic studies of 30  $\mu\text{M}$  F43H  $\text{Mn}^{\text{III}}$ Mb reacted with 200 equiv  $\text{KHSO}_5$  in PBS buffer (pH=7.4).



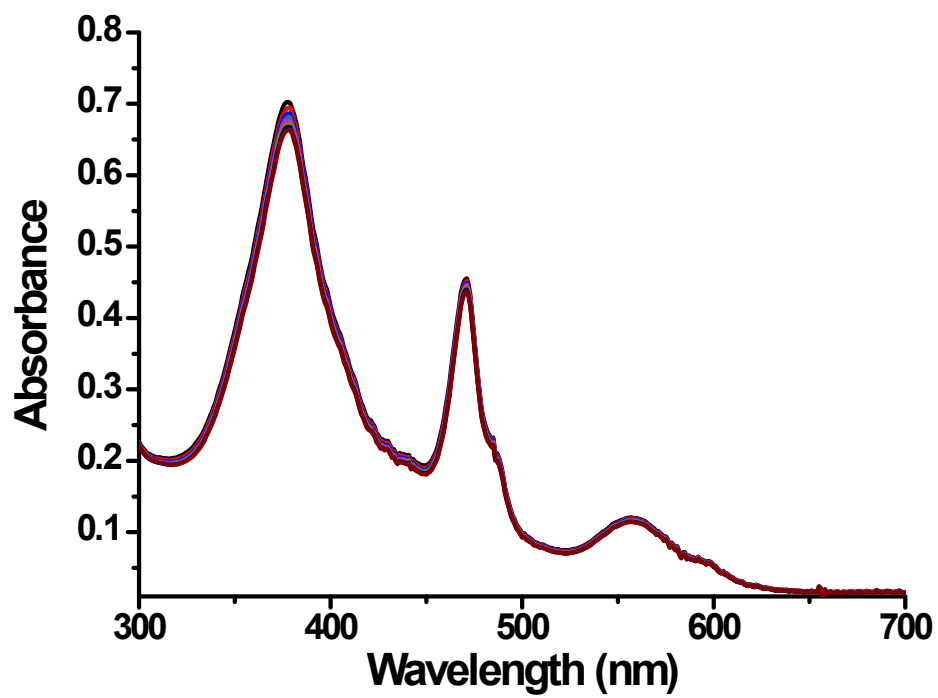
**Figure S3** Kinetic studies of 30 μM L29H Mn<sup>III</sup>Mb reacted with 200 equiv KHSO<sub>5</sub> in PBS buffer (pH=7.4).



**Figure S4** Kinetic studies of 30  $\mu\text{M}$  H64F  $\text{Mn}^{\text{III}}$ Mb reacted with 200 equiv  $\text{KHSO}_5$  in PBS buffer (pH=7.4).

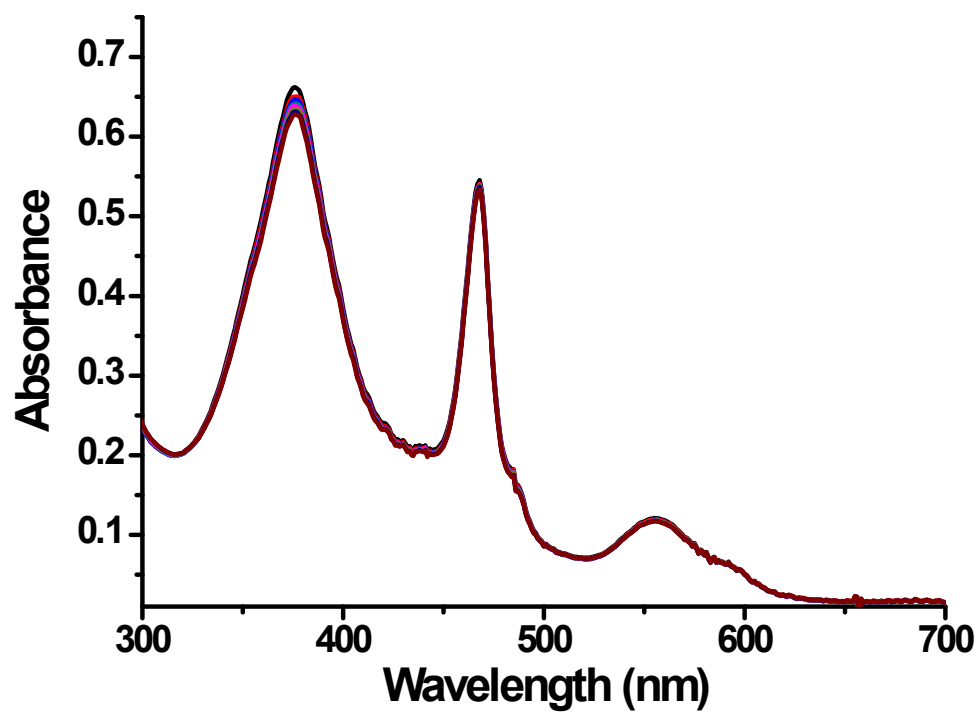


**Figure S5** Kinetic studies of 30  $\mu\text{M}$  F43H/H64F Mn<sup>III</sup>Mb reacted with 200 equiv KHSO<sub>5</sub> in PBS buffer (pH=7.4).

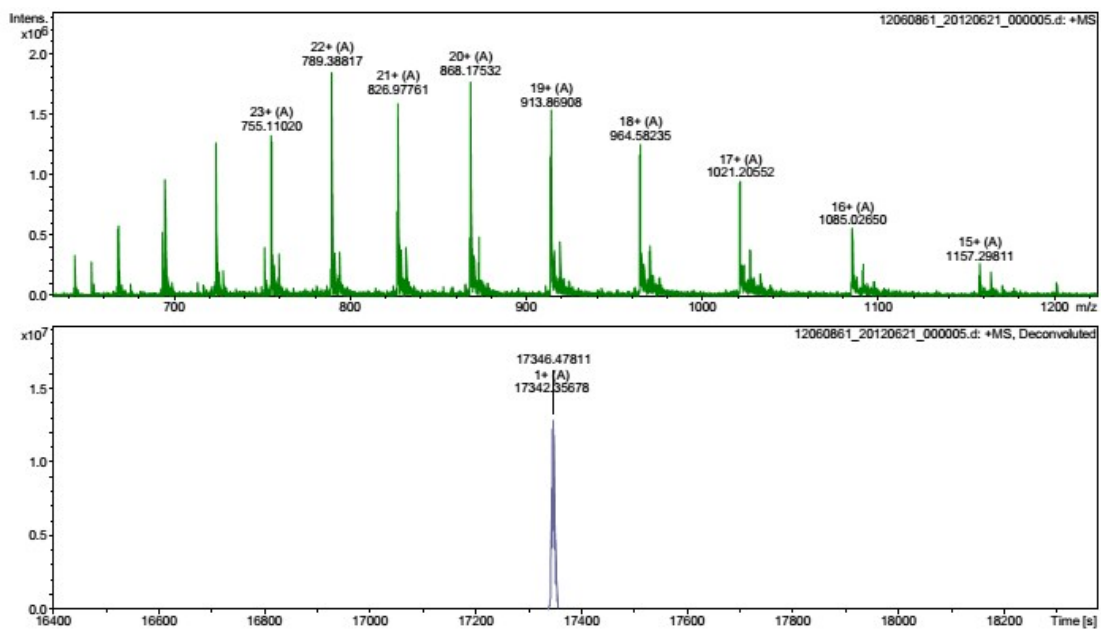


**Figure S6** Kinetic studies of 30  $\mu\text{M}$  L29H/H64F  $\text{Mn}^{\text{III}}\text{Mb}$  reacted with 200 equiv  $\text{KHSO}_5$  in PBS buffer (pH=7.4).





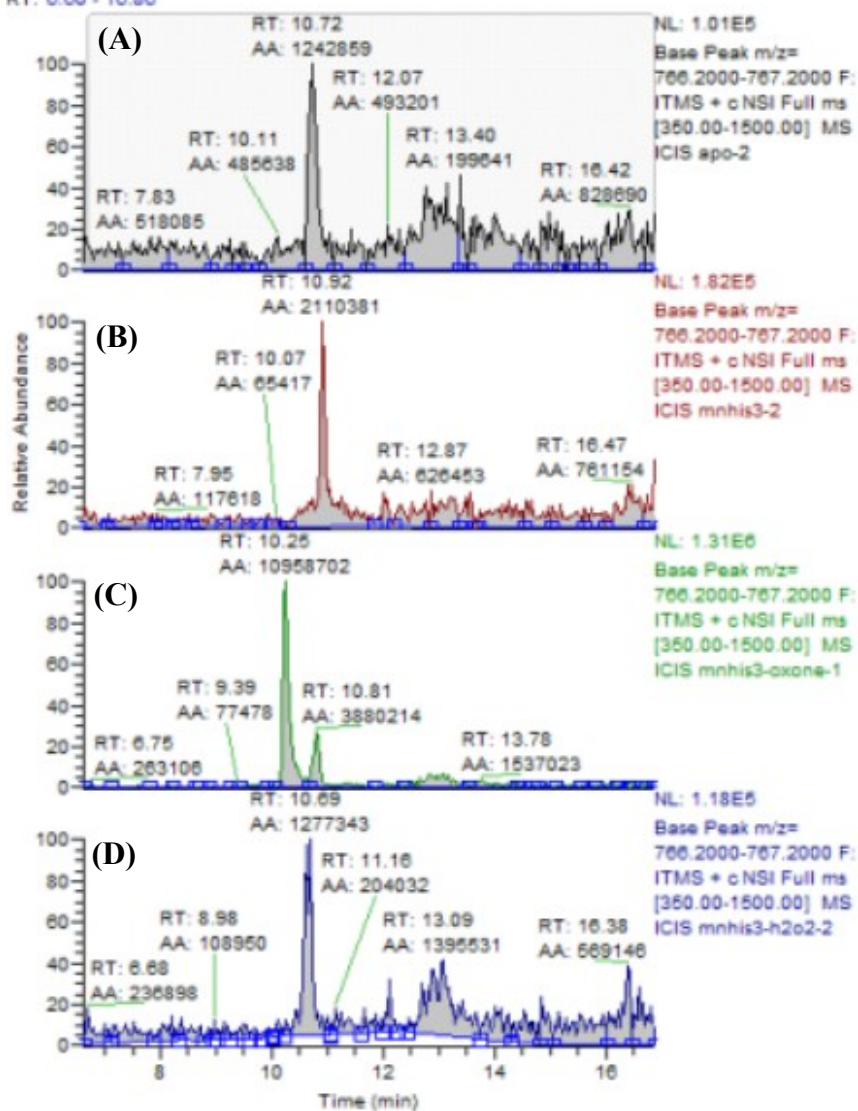
**Figure S7** Kinetic studies of 30  $\mu\text{M}$  L29H/F43H/H64F  $\text{Mn}^{\text{III}}$ Mb reacted with 200 equiv  $\text{KHSO}_5$  in PBS buffer (pH=7.4).



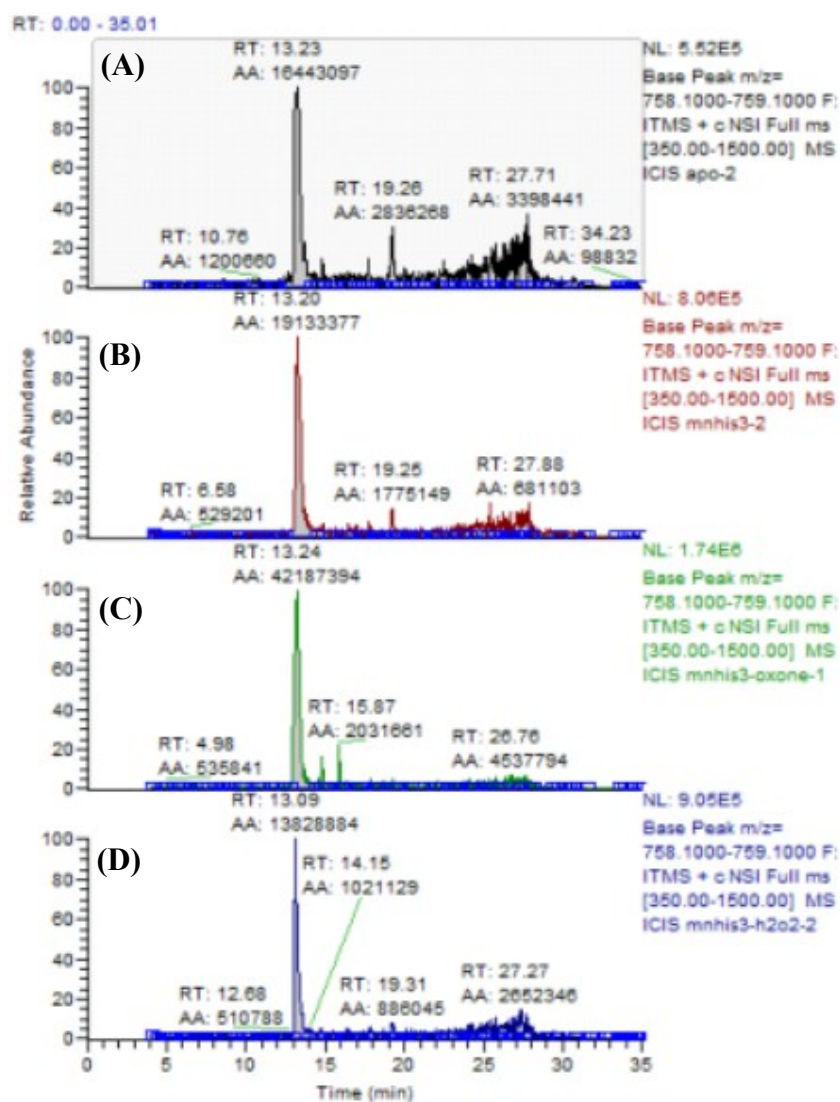
(b)

**Figure S8** ESI-MS of apo His3 Mb with 20 equiv KHSO<sub>5</sub> after reacting for 3 min.

RT: 6.60 - 16.86

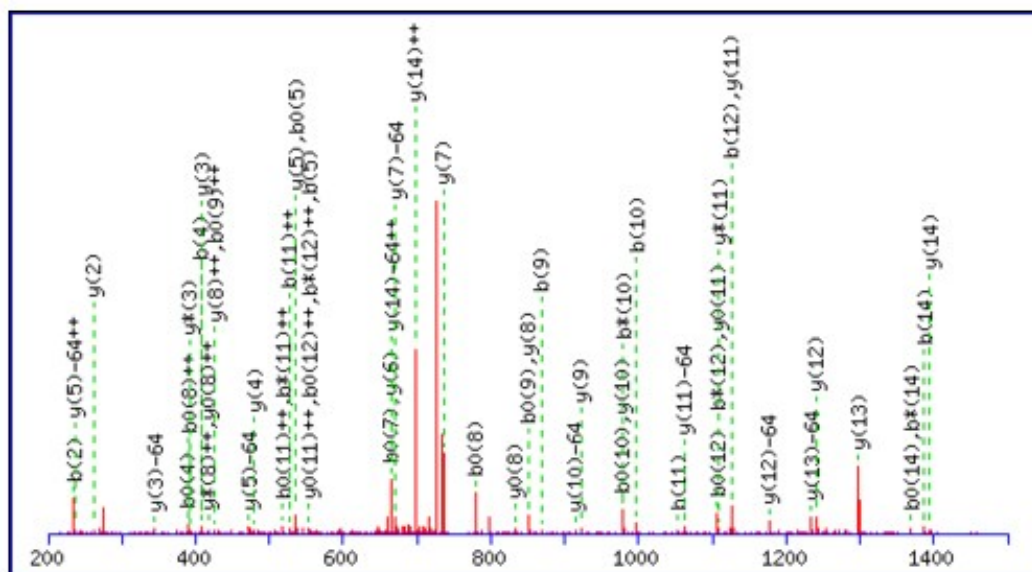


(a)



(b)

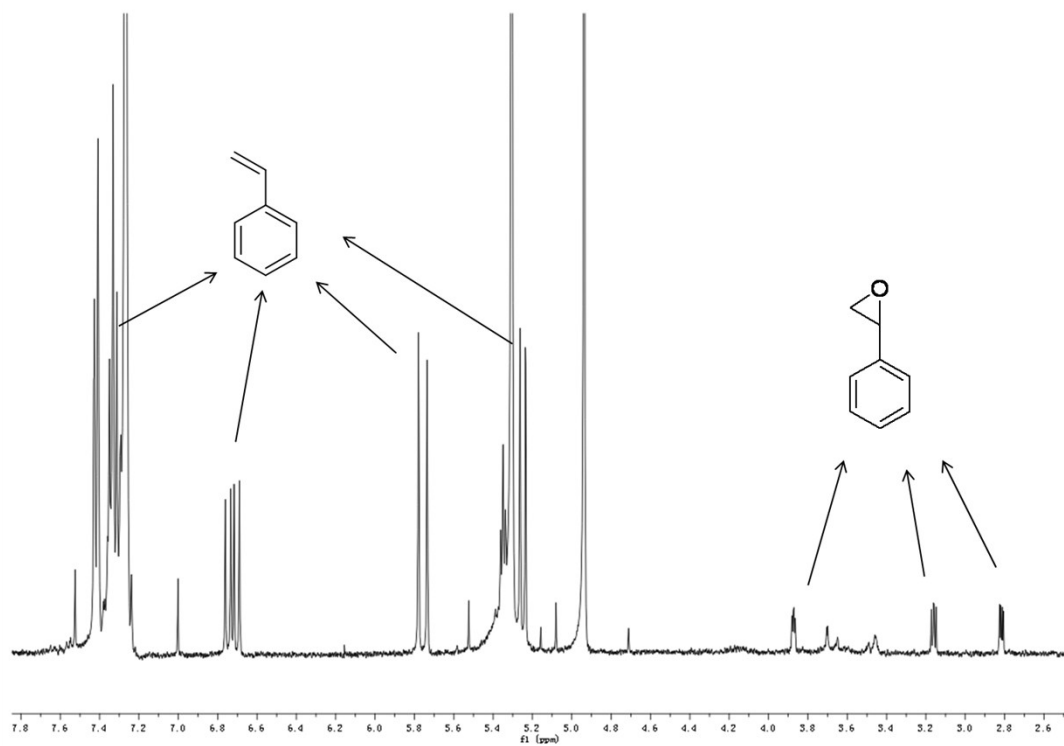
**Figure S9.** Retention times and peak areas of (a) oxidized and (b) unoxidized peptide HPGDFGADAQGAMNK in LC/MS. In each graph, A-D represent apo-His3 Mb with Oxone<sup>®</sup>, His3 Mn<sup>III</sup>Mb without any oxidants, His3 Mn<sup>III</sup>Mb with Oxone<sup>®</sup>, and His3 Mn<sup>III</sup>Mb with H<sub>2</sub>O<sub>2</sub>, respectively.



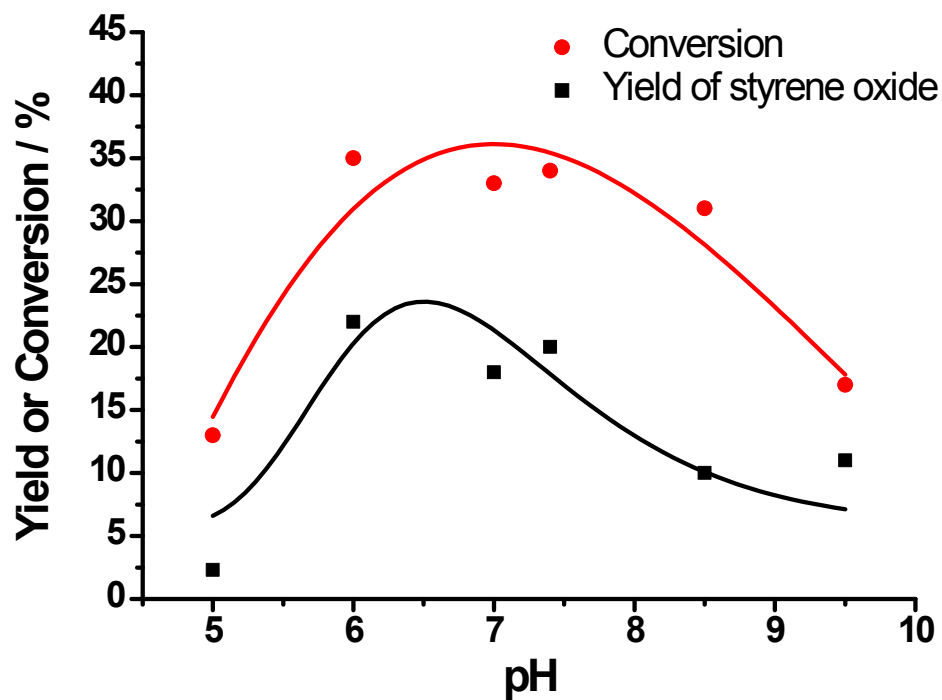
Average mass of neutral peptide Mr(calc): 1531.6057  
 Variable modifications:  
 M13 : Oxidation (M), with neutral losses 0.0000(shown in table), 64.1069  
 Ions Score: 112 Expect: 6.3e-012  
 Matches : 57/204 fragment ions using 74 most intense peaks ([help](#))

#	b	b <sup>++</sup>	b <sup>+</sup>	b <sup>+++</sup>	b <sup>0</sup>	b <sup>0++</sup>	Seq.	y	y <sup>++</sup>	y <sup>+</sup>	y <sup>+++</sup>	y <sup>0</sup>	y <sup>0++</sup>	#
1	138.1467	69.5770					H							15
2	<b>235.2619</b>	118.1346					P	<b>1395.4738</b>	<b>698.2406</b>	1378.4433	689.7253	1377.4585	689.2329	14
3	292.3132	146.6603					G	<b>1298.3586</b>	649.6830	1281.3281	641.1677	1280.3433	640.6753	13
4	<b>407.4006</b>	204.2040			389.3853	195.1964	D	<b>1241.3073</b>	621.1573	1224.2768	612.6421	1223.2920	612.1497	12
5	<b>554.5745</b>	277.7909			536.5592	268.7833	F	<b>1126.2199</b>	563.6136	1109.1894	555.0984	1108.2046	554.6060	11
6	611.6258	306.3166			593.6105	297.3090	G	<b>979.0460</b>	490.0267	962.0155	481.5114	961.0307	481.0190	10
7	682.7037	341.8555			664.6884	332.8479	A	<b>921.9947</b>	461.5010	904.9642	452.9858	903.9794	452.4934	9
8	797.7911	399.3992			779.7758	390.3916	D	<b>850.9168</b>	425.9621	833.8863	417.4468	832.9015	416.9544	8
9	<b>868.8690</b>	434.9382			850.8537	425.9306	A	<b>735.8294</b>	368.4184	718.7989	359.9031			7
10	<b>996.9982</b>	499.0028	979.9677	490.4875	978.9829	489.9952	Q	<b>664.7515</b>	332.8794	647.7210	324.3642			6
11	<b>1054.0495</b>	527.5284	1037.0190	519.0132	1036.0342	518.5208	G	<b>536.6223</b>	268.8148	519.5918	260.2996			5
12	<b>1125.1274</b>	563.0674	1108.0969	554.5521	1107.1121	554.0598	A	<b>479.5710</b>	240.2892	462.5405	231.7739			4
13	1272.3229	636.6651	1255.2924	628.1499	1254.3076	627.6575	M	<b>408.4931</b>	204.7502	391.4626	196.2350			3
14	<b>1386.4255</b>	693.7164	1369.3950	685.2012	1368.4102	684.7088	N	<b>261.2976</b>	131.1525	244.2671	122.6372			2
15							K	147.1950	74.1012	130.1645	65.5859			1

**Figure S10.** Fragmentation patterns of peptide from HPGDFGADAQGAM(Ox)NK the digestion of His3 Mn<sup>III</sup>Mb reacted with KHSO<sub>5</sub>.



**Figure S11.** <sup>1</sup>H NMR after the reaction of styrene and Oxone<sup>®</sup> catalyzed by His3 Mn<sup>III</sup>Mb.



**Figure S12.** Conversion of epoxidation (red) and yield of styrene oxide (black) at different pHs. Conversion was calculated by  $1 - (\text{unreacted styrene} / \text{total added styrene})$ . Yield was based on styrene oxide generated / total added styrene.