

Electronic Supplementary Information

ABAB-type Phthalocyanines Simultaneously Bearing Electron Donating and Electron Accepting Groups. Synthesis, Spectroscopy, and Structure

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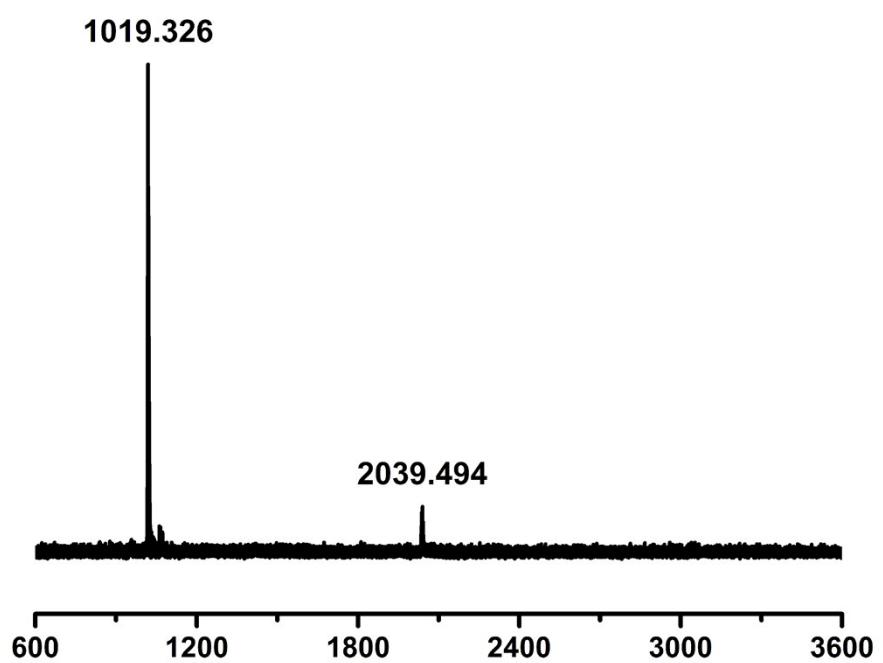


Fig. S1. MALDI-TOF mass spectra of **1**.

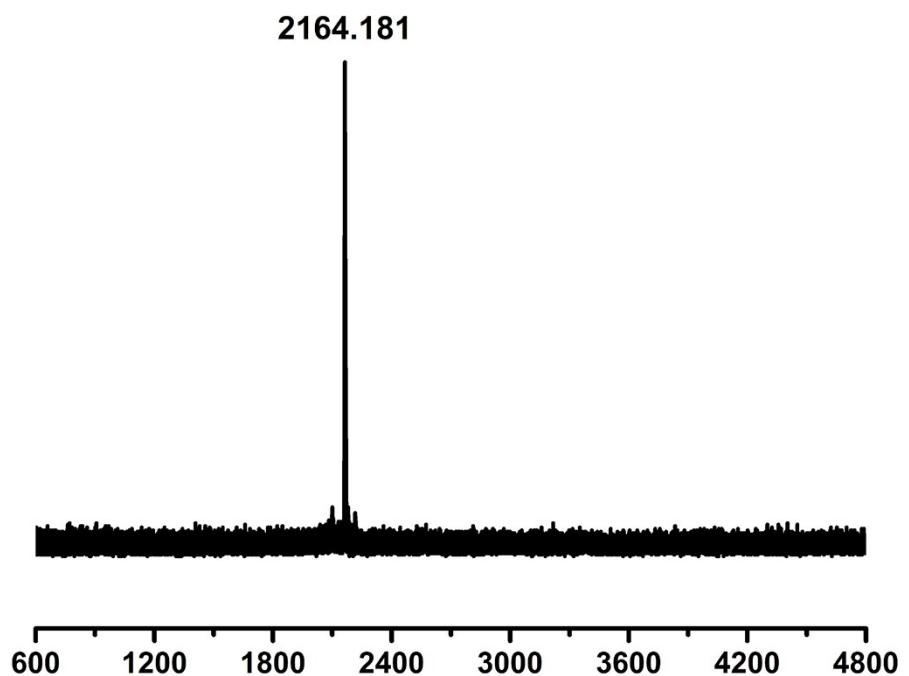


Fig. S2. MALDI-TOF mass spectra of **2**.

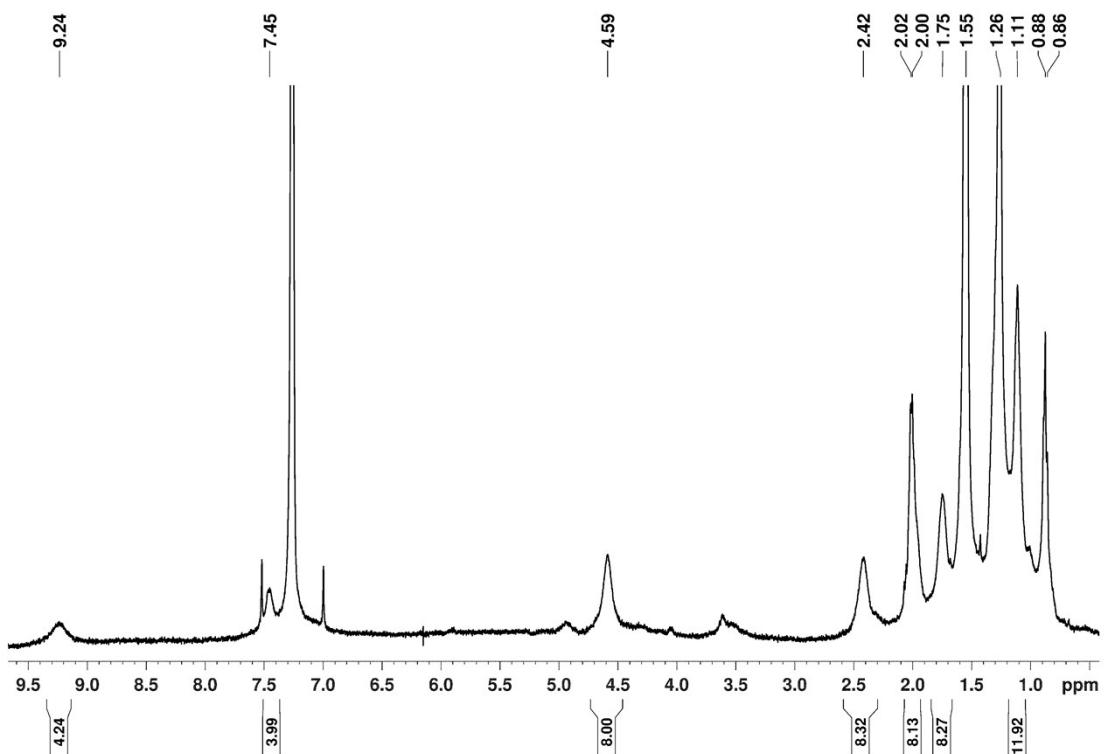


Fig. S3. ^1H NMR spectrum of $\text{Mg}(\text{DADAPc})$ (**1**) recorded in CDCl_3 at room temperature.

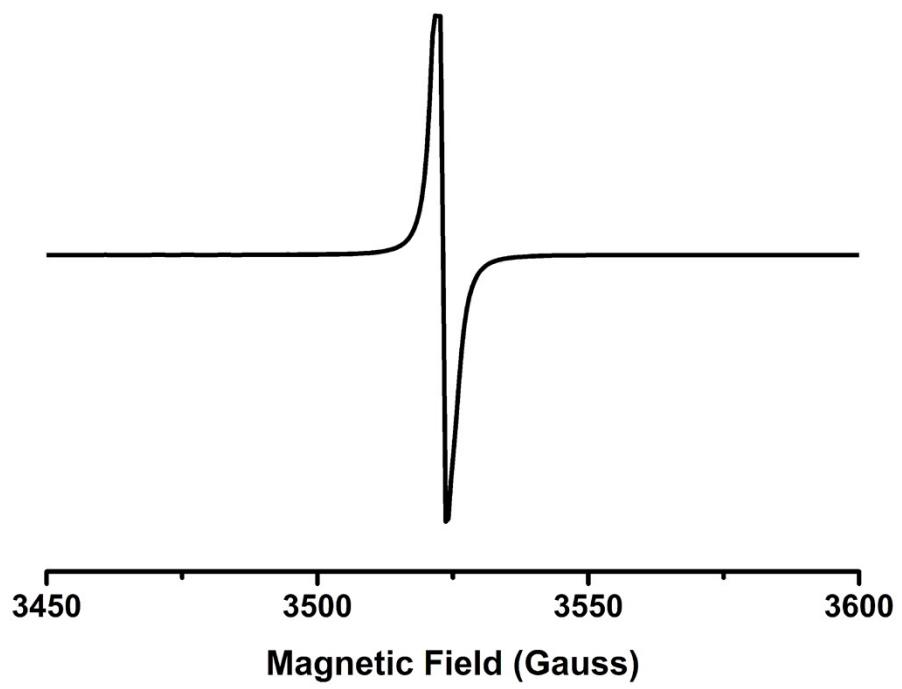


Fig. S4. Powder ESR spectrum of **2** at room temperature.

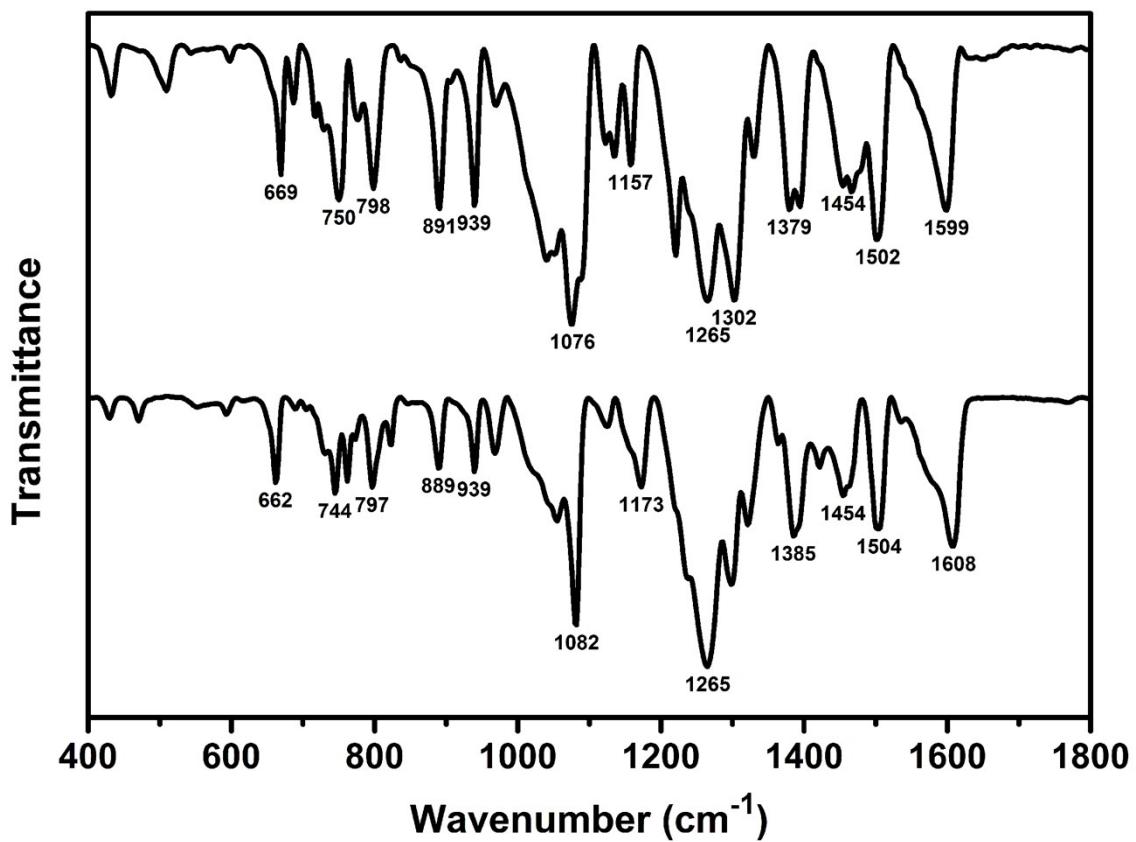


Fig. S5. IR spectra of **1** and **2** in the region of 400-1800 cm⁻¹ with 2 cm⁻¹ resolution.

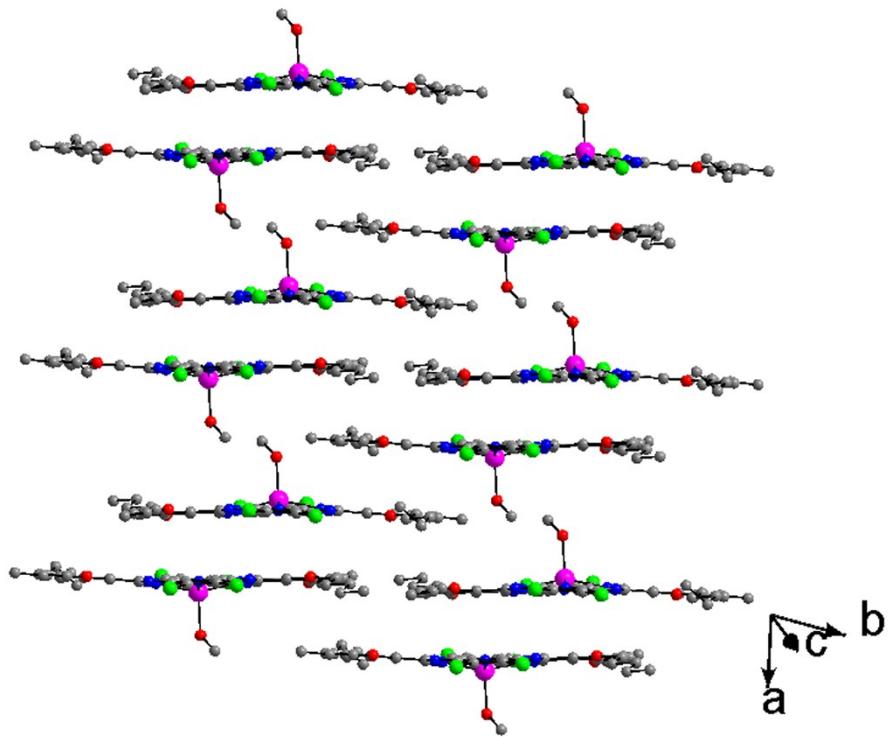


Fig. S6. Molecular packing in single crystals of **1** with hydrogen atoms omitted for clarity.

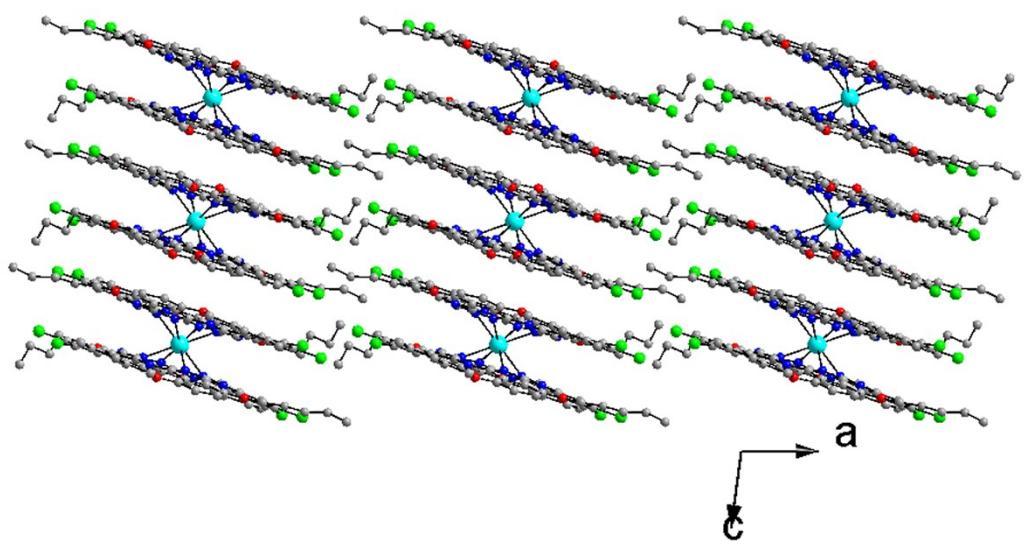


Fig. S7. Molecular packing in single crystals of **2** with hydrogen atoms omitted for clarity [Lu turquoise, C grey, N blue, O red and Cl green].

Table 1. Analytical and mass spectroscopic data for **1** and **2**.

Compound	Yield (%)	[M+H] ⁺ (<i>m/z</i>) ^[a,b]	Analysis (%) ^[a]		
			C	H	N
MgPc(ABAB) (1) ^[c]	6.2	1019.3 (1019.3)	60.29(60.56)	5.18(5.37)	10.81(10.66)
Lu ^{III} [Pc(ABAB)] ₂ (2) ^[d]	58	2164.2 (2164.5)	55.12(55.22)	4.64(4.63)	9.91(9.81)

[a] Calculated values given in parentheses. [b] By MALDI-TOF mass spectrometry. The value corresponds to the most abundant isotopic peak of the protonated molecular ion [M+H]⁺. [c] Contain 1 equiv. of solvated CH₃OH. [d] Contain 1 equiv. of solvated CHCl₃.

Table S2. Electronic absorption data for **1** and **2** in CHCl₃. ·

Compound	$\lambda_{\max}/\text{nm} (\log \epsilon)$		
1	348(4.75)	658(4.48)	726 (5.04)
2	330(5.18)	718(5.13)	1725(4.20)

Table S3. Structural data for **1** and **2**.

	1	2
average M-N(isoindole) bond distance [Å]	2.036	2.376
M-N ₄ (isoindole) plane distance [Å]	0.456	1.337
interplanar distance [Å]	3.379 ^[a]	2.674
average dihedral angle ϕ between N ₄ mean plane and isoindole ring bearing the two pentyloxy chains [°]	3.36	9.07
average dihedral angle ϕ between N ₄ mean plane and isoindole ring bearing the two chloride atoms [°]	3.66	6.52
average twist angle [°] ^[b]	-----	42.30
the nearest M...M distance	5.795	10.308

[a] Defined as the distance between two N₄ mean plane of one MgPc(ABAB) π dimer.

[b] Defined as the rotation angle of one macrocycle away from the eclipsed conformation of the two macrocycles.

Table S4. Crystallographic data for **1** and **2**.

	1	2
Molecular formula	C ₅₄ H ₆₀ N ₈ O ₆ Cl ₄ Mg	C ₁₀₄ H ₁₀₄ N ₁₆ Cl ₈ Lu
<i>M</i>	1083.21	2164.60
Crystal system	Triclinic	Monoclinic
Space group	<i>P</i> -1	<i>C</i> 2/c
<i>a</i> /Å	11.3936(12)	36.9471(4)
<i>b</i> /Å	15.0984(10)	18.5389(2)
<i>c</i> /Å	16.4187(15)	14.4165(2)
α°	83.898(7)	90
β°	89.844(8)	97.5320(10)
γ°	69.577(8)	90
<i>U</i> /Å ³	2630.1(4)	9789.5(2)
<i>Z</i>	2	4
F(000)	1136	4444
<i>D_c</i> /Mg m ⁻³	1.368	1.469
μ/mm^{-1}	2.637	4.476
Data collection range/ ^o	3.14 to 62.50	3.97 to 68.00
Reflections measured	13042	18042
Independent reflections	8233 (Rint = 0.0207)	8827 (Rint = 0.0317)
Parameters	665	622
<i>R</i> ₁ [<i>I</i> >2σ(<i>I</i>)]	0.0452	0.0334
<i>wR</i> ₂ [<i>I</i> >2σ(<i>I</i>)]	0.1206	0.0878
Goodness of fit	1.051	1.033