Electronic Supplementary Information (ESI)

Graphene encapsulated Fe₃O₄ nanorods to assemble a mesoporous hybrid composite as high-performance lithium-ion battery anode material

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Fig. S1 The XRD pattern (a) and SEM image (b) of α -Fe₂O₃ obtained via protocol without GO.



Fig. S2 Comparison of FT-IR spectra of the GO (black curve) and Fe_3O_4/rGO (red curve).



Fig. S3 SEM image of the commercial Fe_3O_4 powder.

Materials	Specific surface area (m ² g ⁻¹)	References
Fe ₃ O ₄ /rGO	152	This work
Conventional Fe ₃ O ₄	2	This work
powder		
Fe ₃ O ₄ /GNSs	52.84	1
Fe ₃ O ₄ @GS/GF	114.5	2
3D Graphene/Fe ₃ O ₄	95.22	3
Fe ₃ O ₄ NCs–GAs	118	4
Fe ₃ O ₄ decorated Graphene	130	5
ball		
Fe ₃ O ₄ -NS/G composites	121	6
Hollow Fe ₃ O ₄ /Graphene	132	7
Fe ₃ O ₄ /GNS	53	8
Fe ₃ O ₄ /C nanospindles	35.1	9
mesoporous Fe ₃ O ₄	133	10
nanocages		
Fe ₃ O ₄ /Helical Carbon	126	11
Nanofibers		
Fe ₃ O ₄	35.04	12
microspheres/Graphene		
Fe ₃ O ₄ Hollow Spheres	88.06	13
Fe ₃ O ₄ –RGO	81.67	14
Fe ₃ O ₄ /GS	83.9	15
Fe ₃ O ₄	92.2	16
Nanoflake/Graphene		
Fe ₃ O ₄ nanorods/Graphene	86	17
Fe ₃ O ₄ @Polypyrrole	69.63	18
Nanocages		

Table S1 Comparison of the BET specific surface areas of mesoporous Fe_3O_4/rGO composites with other Fe_3O_4 based composites.

Note: GNSs: graphene nanosheets; GS/GF: encapsulated with graphene; NCs: nanoclusters; GAs: graphene aerogels; NS: nanospheres.



Fig. S4 Pore size distribution of Fe_3O_4/rGO samples; inset: pore size distribution of bare Fe_3O_4 sample.



Fig. S5 Cyclic voltammograms for the first to fifth cycle of the bare Fe_3O_4 sample in a voltage range of 0.01-3.0 V at 0.1 mV s⁻¹.



Fig. S6 The cyclic performance of pure rGO anode at 100 mA g^{-1} .



Fig. S7 The XRD patterns of before and after cycles based on Fe_3O_4/rGO electrode material.



Fig. S8 SEM images of Fe₃O₄/rGO (a) and bare Fe₃O₄ (b) after 100 cycles at 100 mA g⁻¹.

Supporting references

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