Electronic Supplementary Information

Robust Porous Polymers Enabled by a Fast Trifluoroacetic Acid Etch with Improved Selectivity for Polylactide

Amrita Sarkar[†], and Morgan Stefik^{†*}

[†] Department of Chemistry and Biochemistry, University of South Carolina, Columbia, South Carolina 29208, United States



Fig. S1 The **as-made PS-***b***-PLA** 2D SAXS pattern consisting of isotropic rings. Color scale corresponds to the log of the X-ray intensity.



Fig. S2 (a) SAXS of **as-made** (2 mm thick) as compared to **NaOH-etch** (2mm thick) PS-*b*-PLA film. The samples were indexed for LAM symmetry with peaks indicated at $q/q^*=1$, 2, 3, 4, and 5. The scattering data were offset vertically for clarity. (b) Top-view SEM image of sample **NaOH-etch**.



Fig. S3 Heteronuclear single quantum coherence spectroscopy of sample TFA-etch.



Fig. S4 Heteronuclear single quantum coherence spectroscopy of PS-OH macroinitiator.



Fig. S5 Heteronuclear single quantum coherence spectroscopy of sample **NaOH-etch**. New and unidentified carbons resulting from the etch are indicated by arrows.



Fig. S6 The **as-made** samples resulted from a combination of solvent casting (a) and vacuum oven annealing (b). Samples were etched by immersion in a TFA solution (c) placed in an oven (d). The etched films were rinsed with water (e) and immersed in methanol (f) before drying in a vacuum oven.



Fig. S7 TFA vapor etching was conducted with the **as-made** sample and etchant solution placed adjacent within a closed jar (a). The entire setup was placed within an oven (b). The etched film was rinsed with water (c) and immersed in methanol (d) before drying in vacuum oven.



Fig. S8 PLA etch kinetics using the vapor from an 8M TFA solution.

PS-*b***-PLA2** Synthesis:

PS-*b*-PLA2 was synthesized following a similar procedure as PS-*b*-PLA. The PS-OH was synthesized using reagent ratio of [Styrene]:[Initiator]:[Me₆TREN]:[Cu(I)]:[Sn(II)] = 300:1:0.105:0.005:0.1. After 12 hrs of polymerization the PS-OH of 13.9 kg/mol and dispersity of 1.22 was obtained. The PS-*b*-PLA of 27.8 kg/mol with a molar mass dispersity (Đ) of 1.23 was obtained after 15 minutes of polymerization using a reagent ratio of [**PS-OH**]:[Lactide]:[DBU]:[Thiourea]=1:150:1.34:1.34. Sample **HEX-As-Made** was made using the already described annealing procedures. Sample **HEX-TFA-Etch** was prepared using immersion in 1M TFA solution for 27 hrs, following the already described procedures.

Sample	M _{n, PS}	M _{n, PLA}	Total M _n	Ð ^a	f ^v _{PS} ^d	f ^v _{PLA} ^d	Morphology ^e
	(kg mol ⁻¹) ^a	(kg mol ⁻¹) ^b	(kg mol ⁻¹) ^c				
PS-b-PLA2	13.9	13.9	27.8	1.23	0.54	0.46	HEX

^aobtained from GPC analysis, ^bobtained from ¹H NMR analysis, ^cobtained from a+b ^dvolume fractions (f') calculated using densities^{21,36} at 110°C, PS= 1.02 g/cm³, PLA = 1.18 g/cm³ ^eMorphology determined by SAXS.



Fig. S9 (a) ¹H NMR spectra of (i) **HEX-As-Made** and (ii) **HEX-TFA-etch** dissolved in CDCl₃. 100% of PLA was removed selectively (PLA peak disappeared at 5.19 ppm) by immersion in 1M TFA solution for 26 hrs. (b) SAXS of **HEX-As-Made** and **HEX-TFA-etch** demonstrate preservation of a hexagonal morphology. The principal peak (q*) in both scattering profile is centered at 0.25 nm⁻¹. The samples were indexed for HEX symmetry with peaks indicated at $q/q^*=1$, $\sqrt{3}$, and $\sqrt{4}$. The scattering data were offset vertically for clarity. (c-d) Top-view SEM images of sample **HEX-TFA-etch** with a measured pore size of 16.2 ±2.5 nm.