

Supporting Information

Bisimino-functionalized dibenzo[a,c]acridines as highly conjugated pincer frameworks for palladium(II): synthesis, characterization and catalytic performance in Heck coupling

Kaiser Mahmood,^a Erlin Yue,^a Wenjuan Zhang,^{a,b} Gregory A. Solan,^{a,c} Tongling Liang ^a and Wen-Hua Sun^{a,d,*}

^aKey Laboratory of Engineering Plastics and Beijing National Laboratory for Molecular Sciences, Institute of Chemistry, Chinese Academy of Sciences, Beijing 100190, China

^bSchool of Materials Science and Engineering, Beijing Institute of Fashion Technology, Beijing 100029, China

^cDepartment of Chemistry, University of Leicester, University Road, Leicester LE1 7RH, UK

^dUniversity of Chinese Academy of Sciences, Beijing 100049, China

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1. Additional catalytic testing

Table S1a. Screening 4-iodotoluene/styrene using **Pd1** as catalyst^a

<chem>Cc1ccc(I)cc1</chem> + <chem>C=Cc1ccccc1</chem> $\xrightarrow[\text{Base, Solvent, Heat}]{\text{Pd1 (0.002 mol\%)}}$ <chem>Cc1ccc(C=Cc2ccccc2)cc1</chem>						
Entry	Base	Solvent	Temp. (°C)	Time (h)	Conv. (%) ^b	TON ^c
1	Na ₂ CO ₃	DMF	140	8	96	48000

^a Reaction conditions: 4×10^{-5} mmol **Pd1**, 2.0 mmol 1-iodo-4-methylbenzene, 2.4 mmol styrene, 2.2 mmol base, 4.0 mL solution. ^b Determined by GC. ^c TON: mol product/mol Pd.

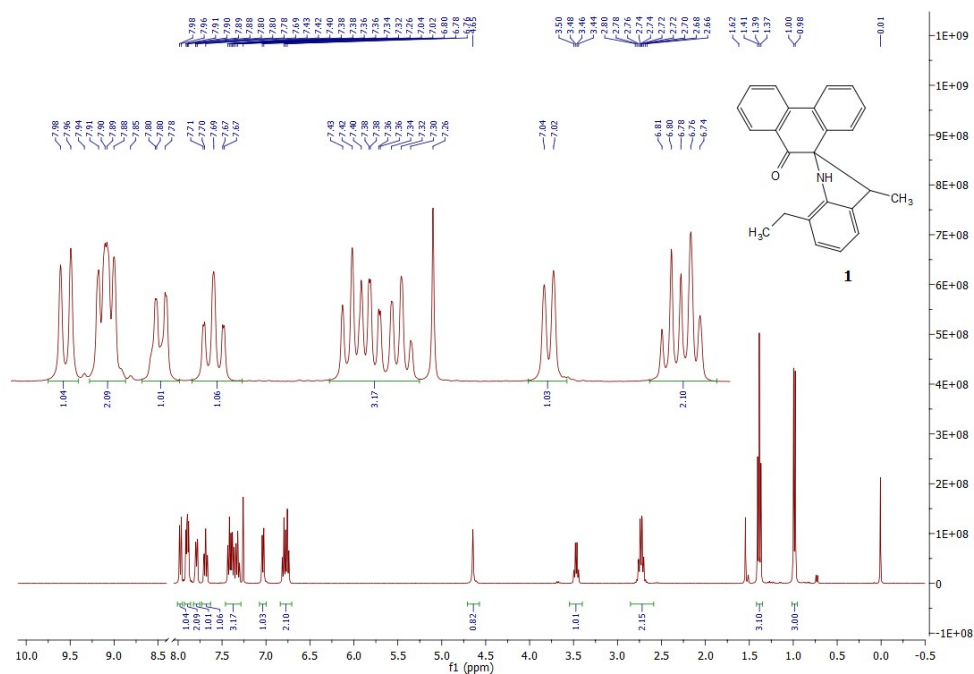
Table S1b. Screening 4-iodobenzene/styrene using **Pd1** as catalyst; exploring loading and reaction duration effects

<chem>c1ccc(I)cc1</chem> + <chem>C=Cc1ccccc1</chem> $\xrightarrow[\text{Base, Solvent, Heat}]{\text{Pd1 (0.001 - 0.002 mol\%)}}$ <chem>c1ccc(C=Cc2ccccc2)cc1</chem>						
Entry	Base	Solvent	Temp. (°C)	Time (h)	Conv. (%) ^b	TON ^c
2 ^d	Na ₂ CO ₃	DMF	140	8	88	88000
3 ^e	Na ₂ CO ₃	DMF	140	24	92	46000

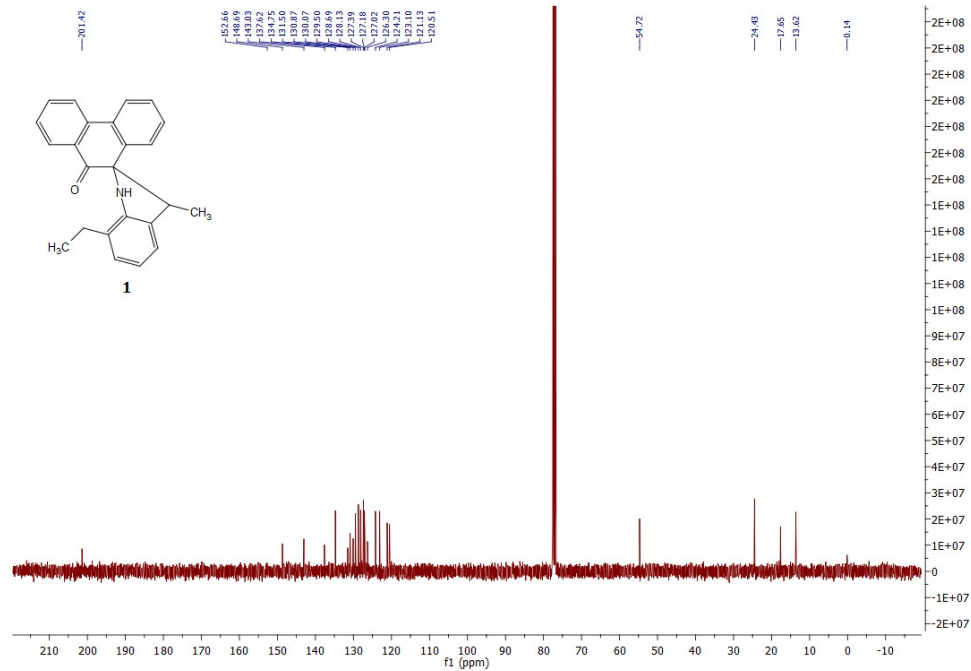
Reaction conditions: ^d 2×10^{-5} mmol **Pd1** (0.001 mol%), 2.0 mmol iodobenzene. ^e 4×10^{-5} mmol **Pd1** (0.002 mol%), 2.0 mmol iodobenzene.

2. ¹H and ¹³C NMR spectra for all organic and inorganic compounds

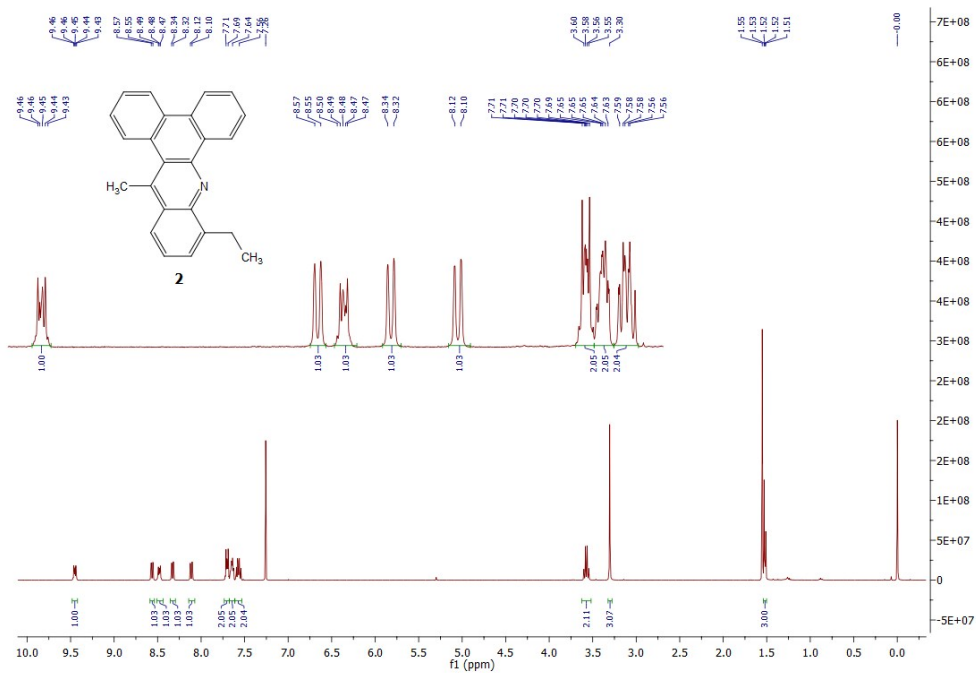
¹H NMR spectrum of **1**



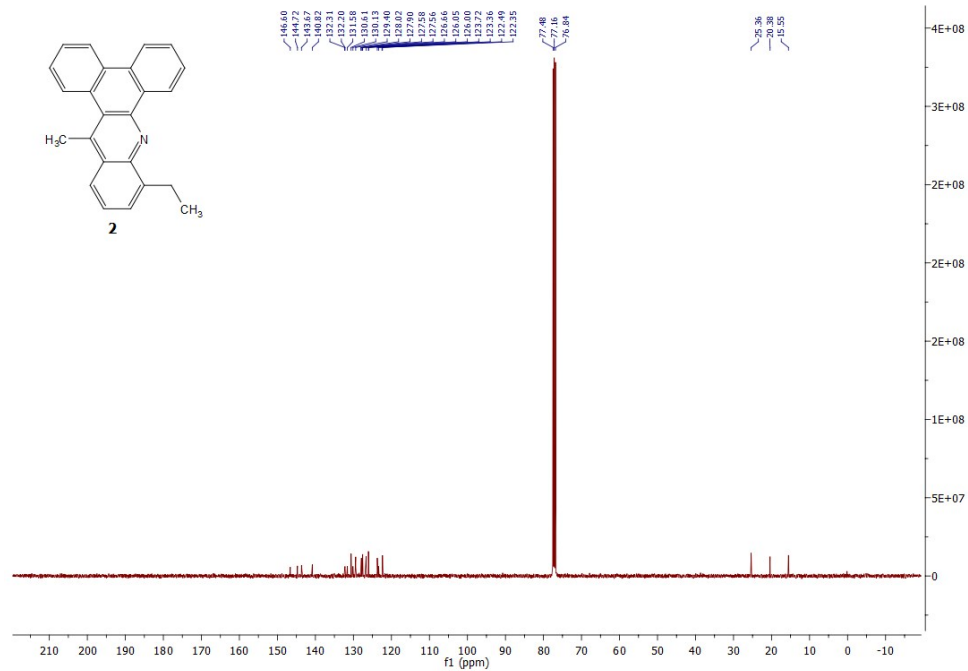
^{13}C NMR spectrum of **1**



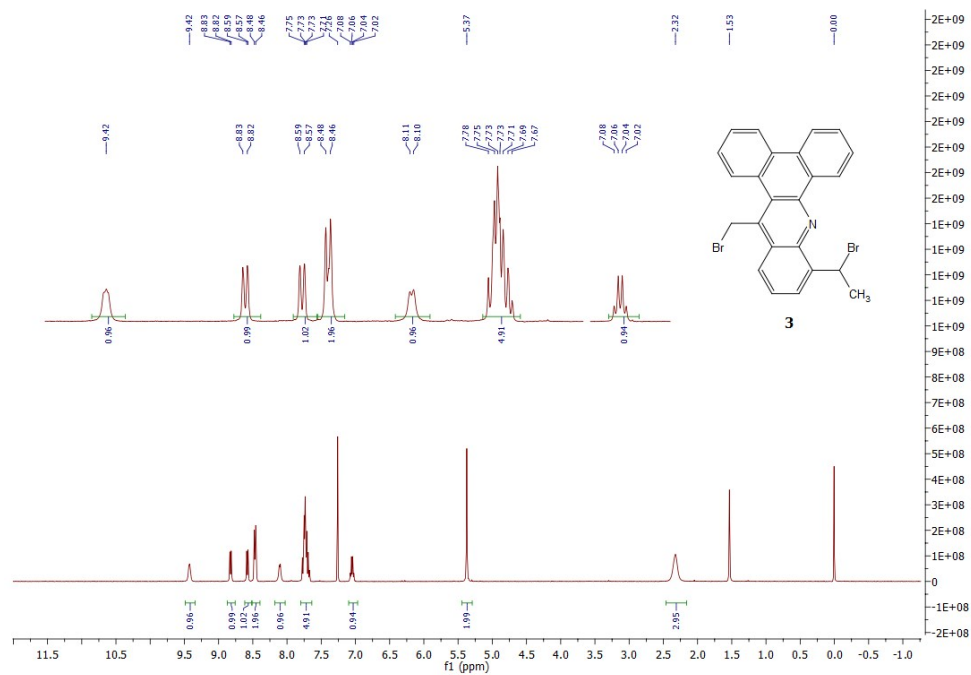
^1H NMR spectrum of **2**

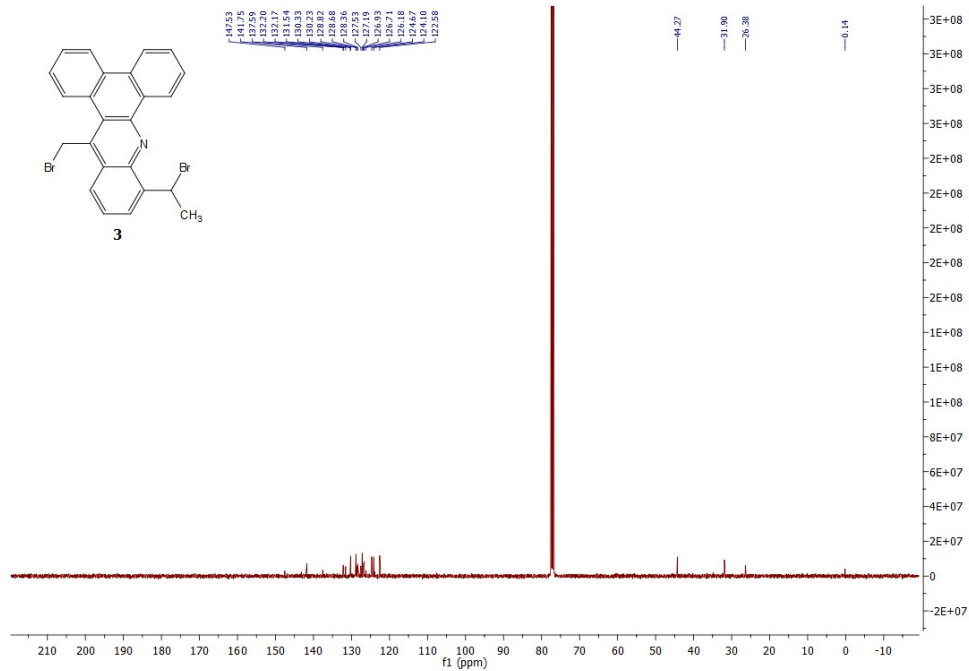
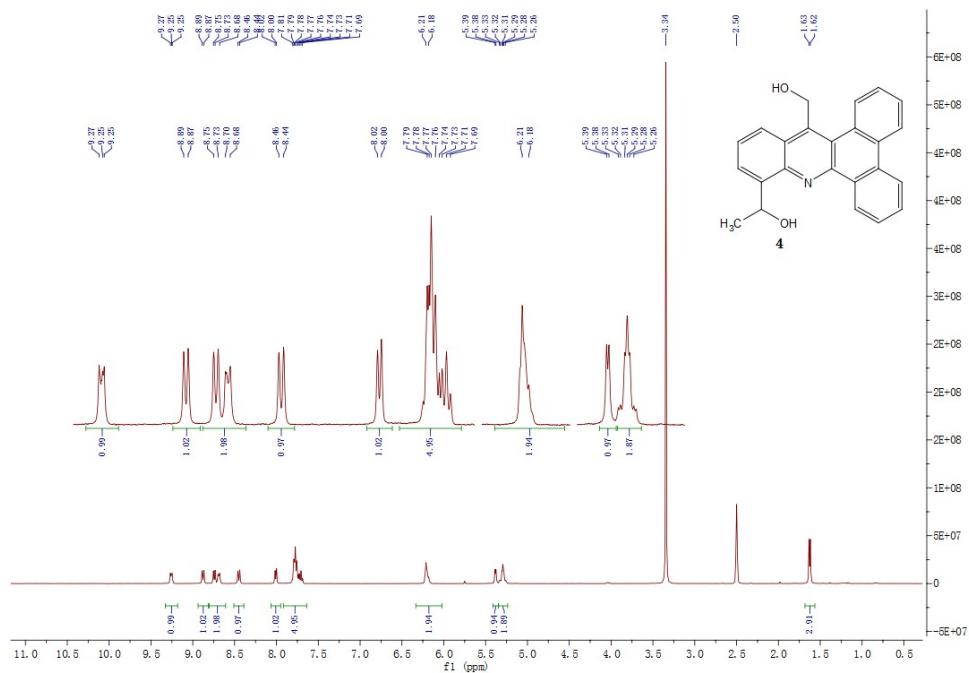


^{13}C NMR spectrum of **2**

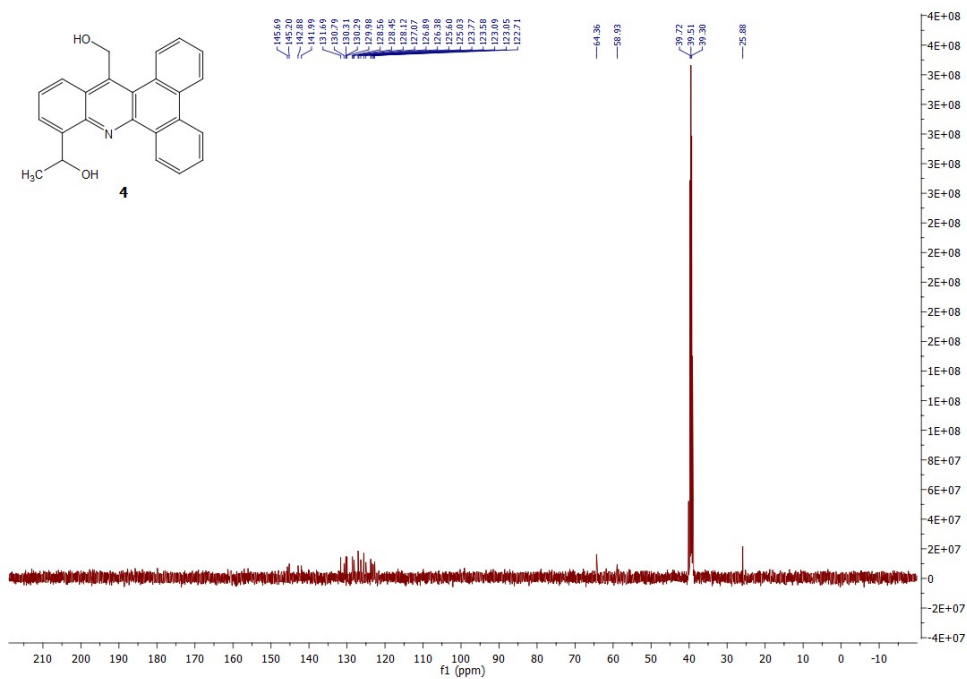


^1H NMR spectrum of **3**

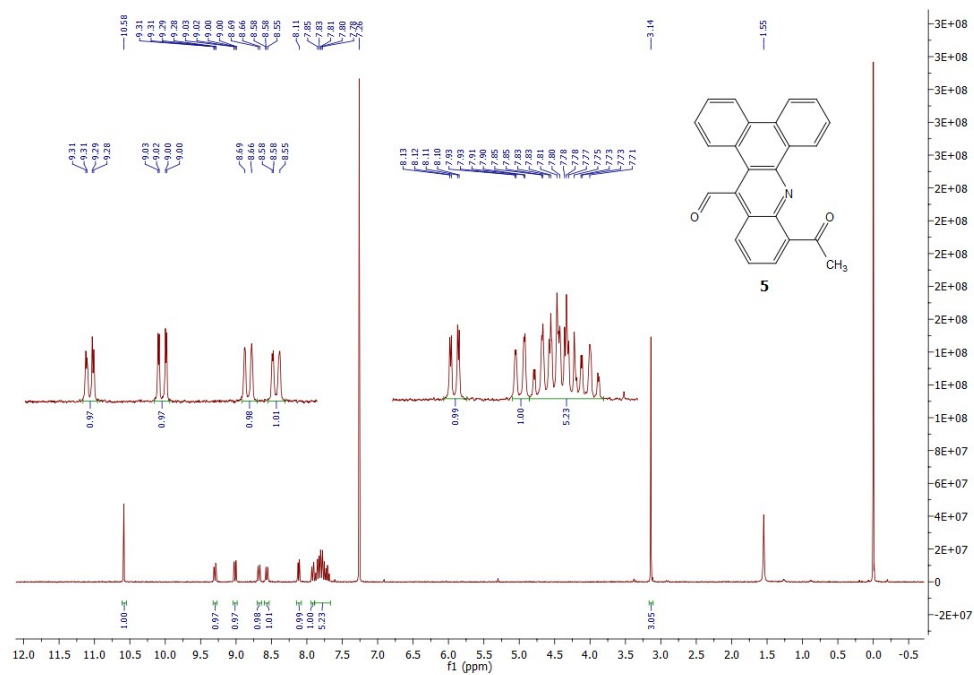


¹³C NMR spectrum of **3**¹H NMR spectrum of **4**

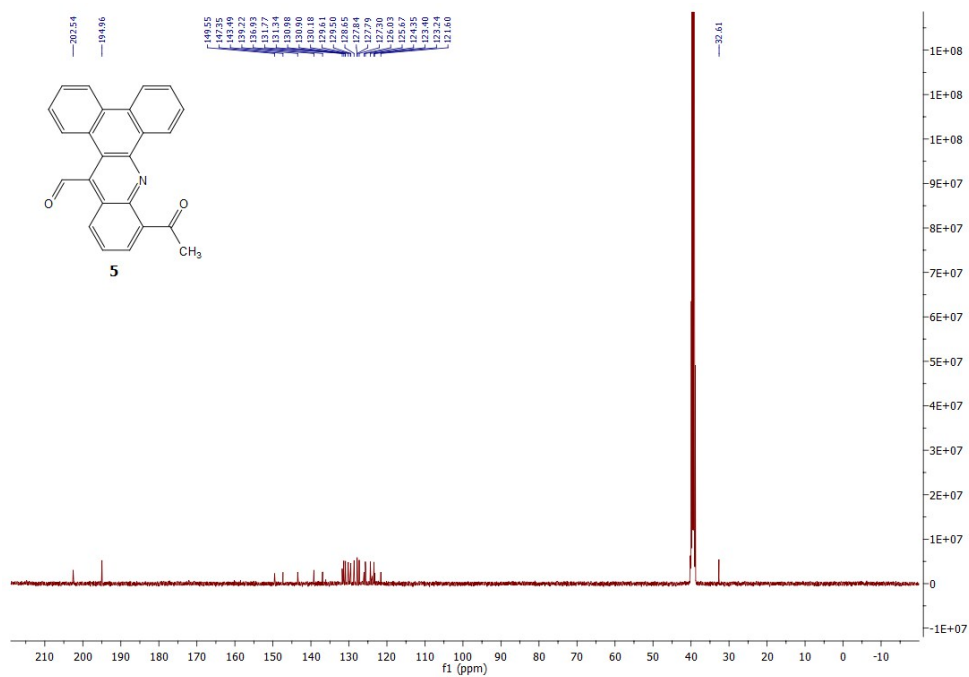
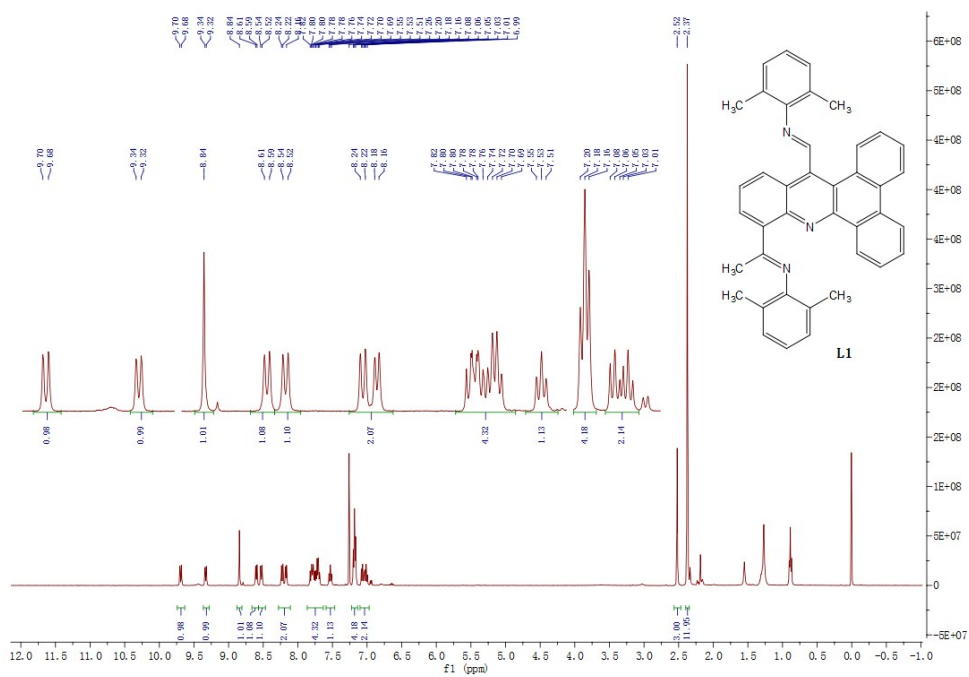
¹³C NMR spectrum of **4**

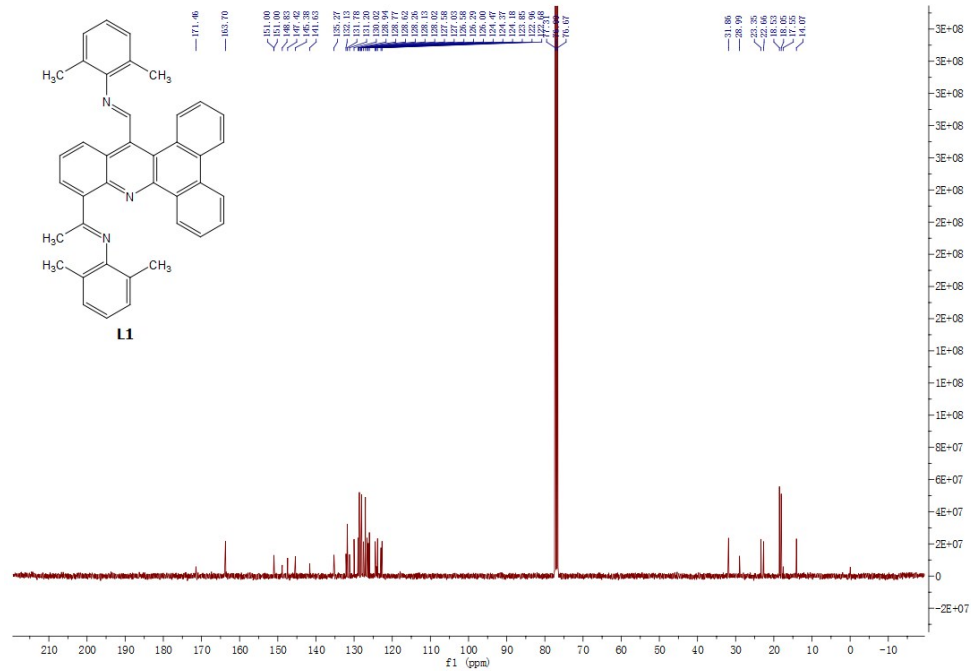


¹H NMR spectrum of **5**

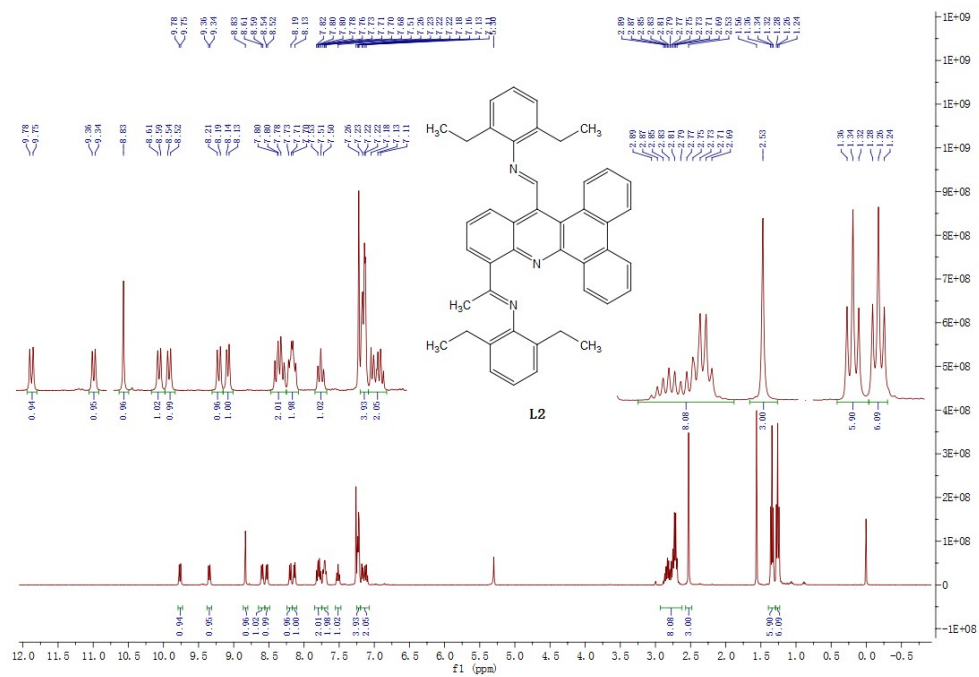


¹³C NMR spectrum of **5**

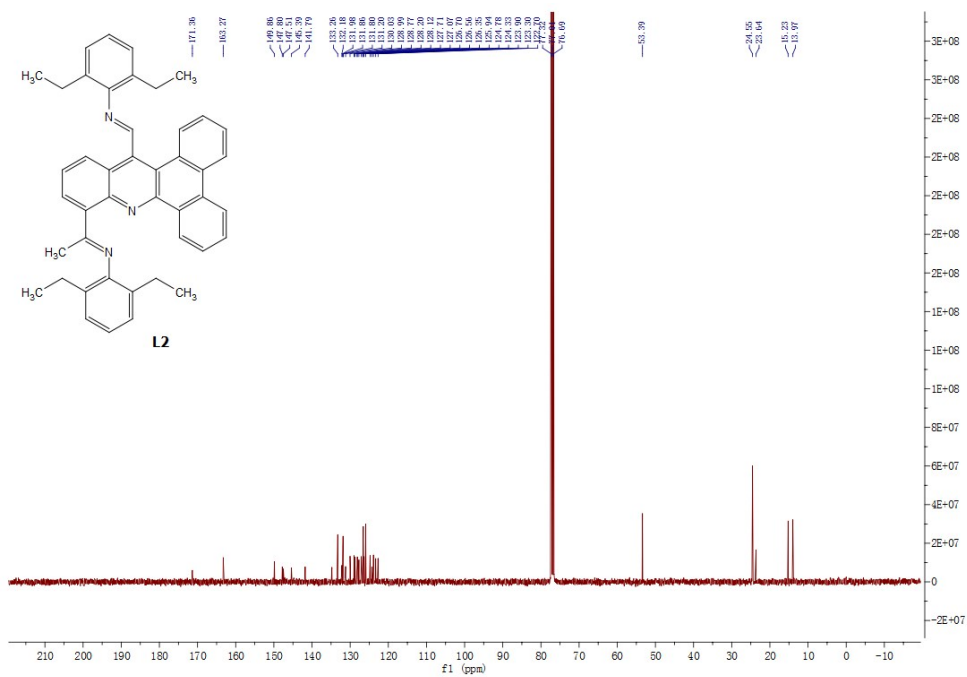
¹H NMR spectrum of **L1**

^{13}C NMR spectrum of **L1**

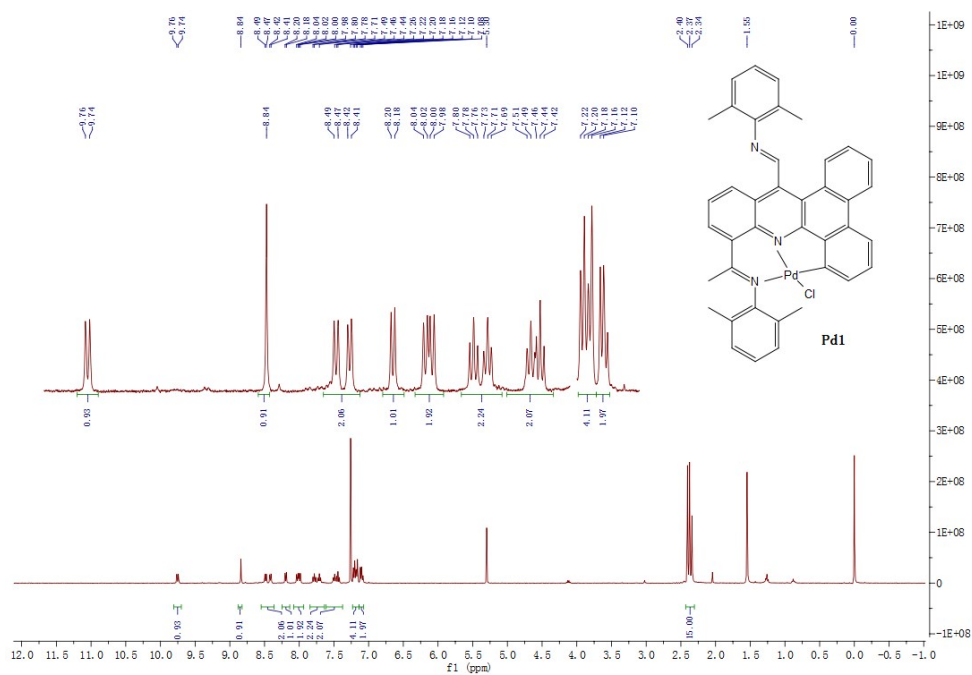
¹H NMR spectrum of L2

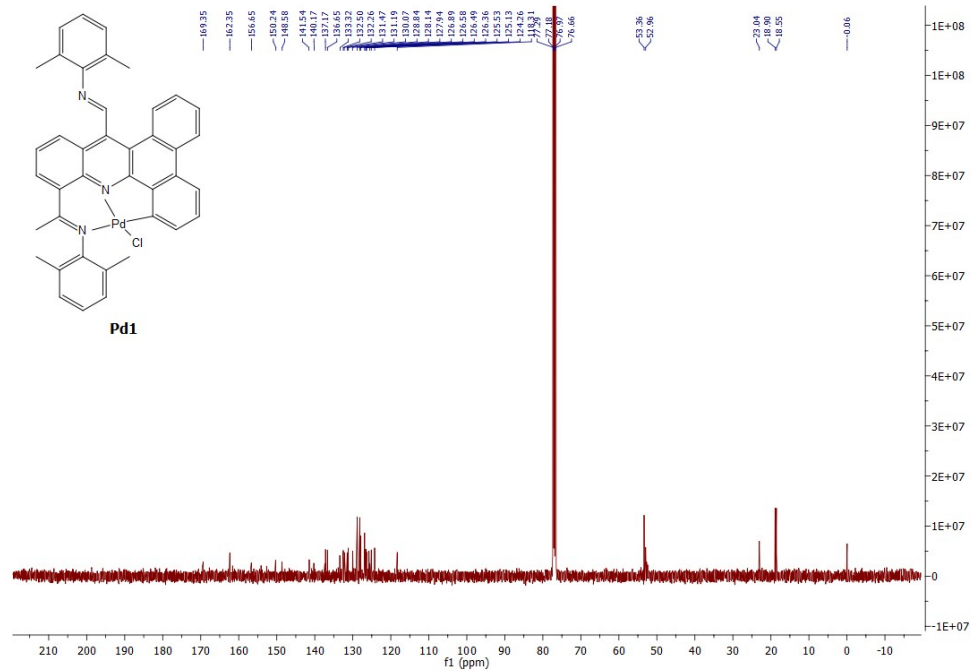


^{13}C NMR spectrum of **L2**

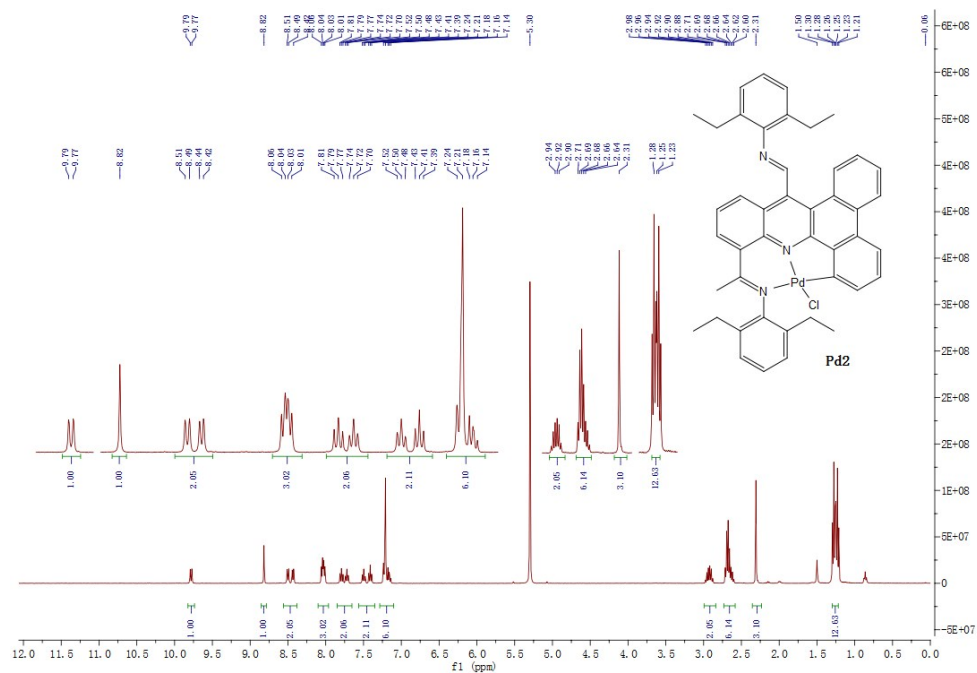


^1H NMR spectrum of **Pd1**

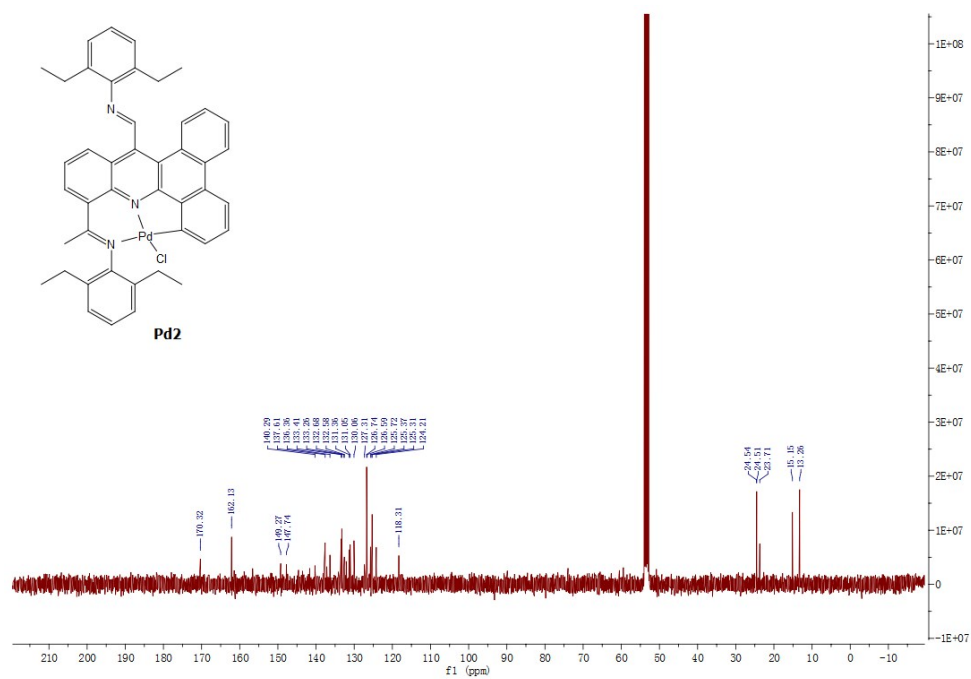


^{13}C NMR spectrum of **Pd1**

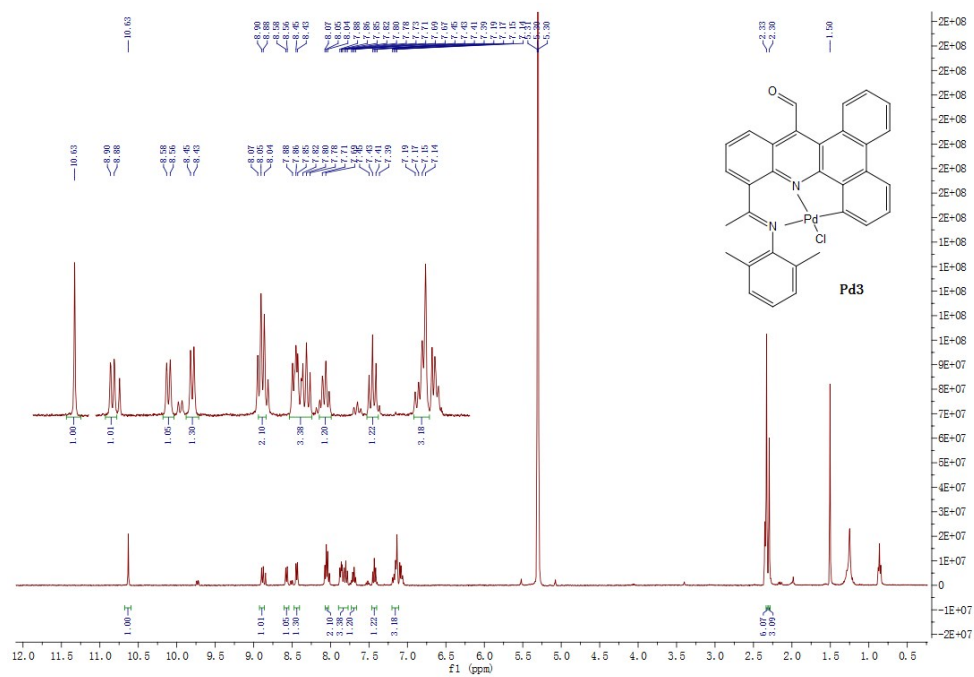
¹H NMR spectrum of **Pd2**



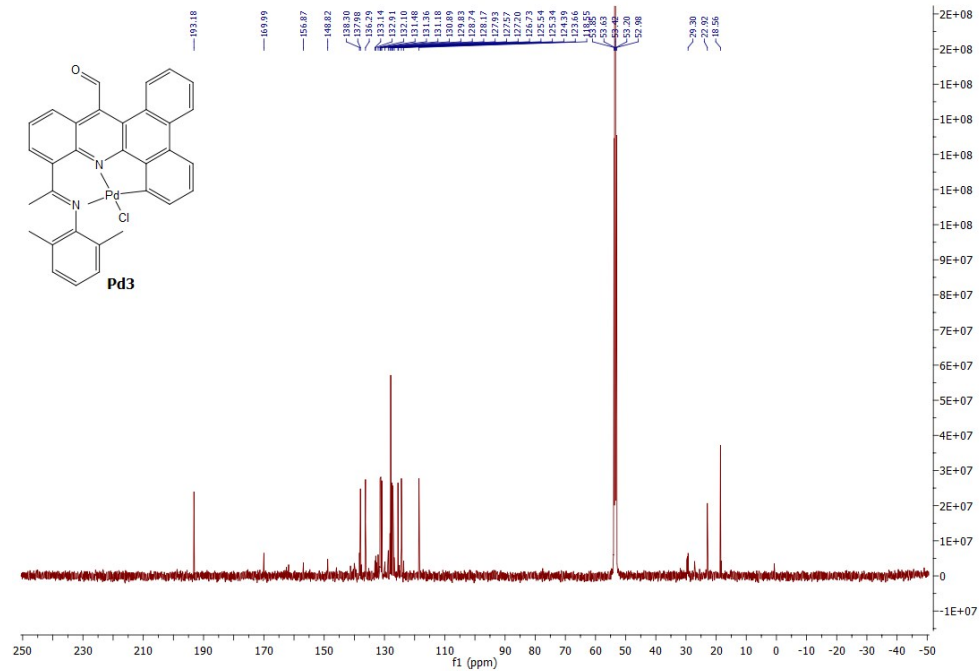
¹³C NMR spectrum of **Pd2**



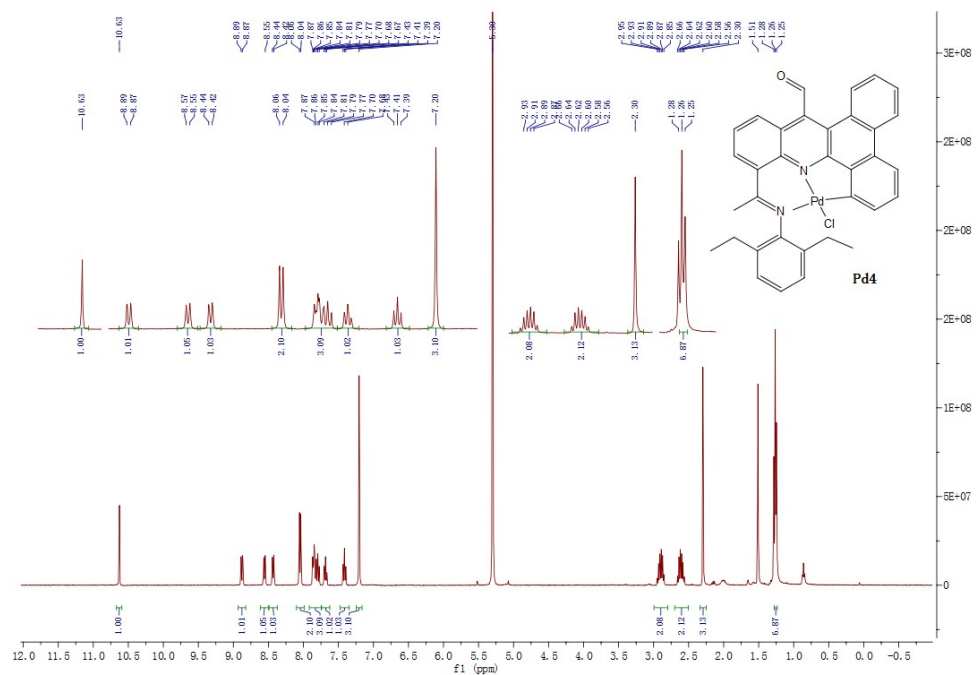
^1H NMR spectrum of **Pd3**



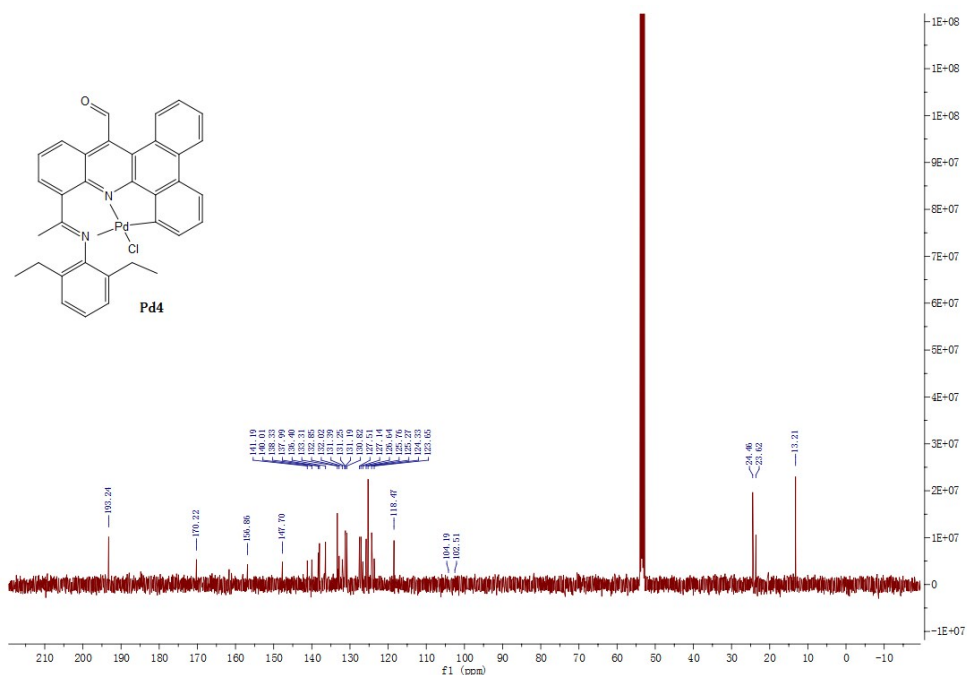
^{13}C NMR spectrum of **Pd3**



^1H NMR spectrum of **Pd4**



¹³C NMR spectrum of **Pd4**



3. Crystal structure description, data and structural refinements for **1** and **2**

Suitable crystals for single crystal X-ray diffraction studies of **1** and **2** were grown by layering a dichloromethane solution of **1** and **2** with heptane. ORTEP diagrams of **1** and **2** are shown in Figures S1 and S2; selected bond distances and angles are collected in the corresponding figure caption. The structure of spiro-[7-ethyl-3-methylindoline-2,10-phenanthren-9-one] **1** consists of a six- and five-membered ring systems which are linked by chiral center C13; both enantiomers are present within this centrosymmetric space group. The six-membered ring further contains two linked aryl groups and a ketone unit that are essentially co-planar with each other. By contrast the more strained five-membered ring is completed by a secondary amine and a tertiary carbon which are 1,2-linked by an ethyl-substituted aryl ring. The C14-O1 bond distance of 1.225(3) Å is consistent with a ketonic carbon-oxygen double bond.

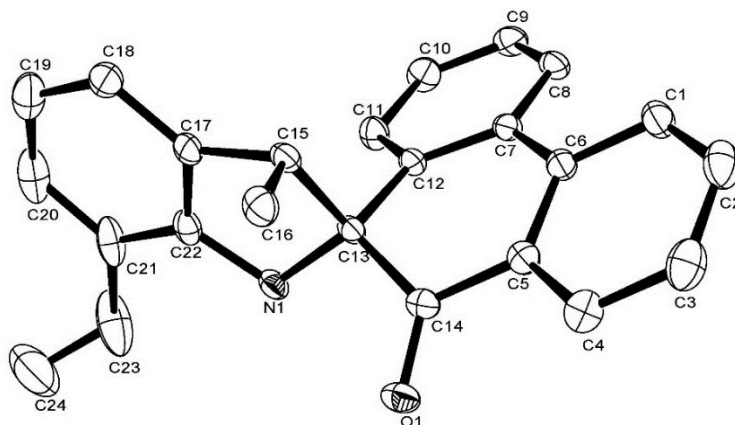


Figure S1. ORTEP diagram of **1** with thermal ellipsoids set at 50% probability level; all hydrogen atoms are omitted for clarity. Selected bond lengths (Å) and angles (°): O(1)-C(14) 1.225(3), N(1)-C(13) 1.462(3), N(1)-C(22) 1.392(3), C(14)-C(13) 1.519(4), C(15)-C(13) 1.585(3), C(12)-C(13) 1.534(3), N(1)-C(13)-C(14) 111.45(19), N(1)-C(13)-C(12) 113.4(2), C(14)-C(13)-C(12) 110.4(2), N(1)-C(13)-C(15) 102.45(19), C(14)-C(13)-C(15) 110.4(2) C(12)-C(13)-C(15) 108.42(19).

The structure of 10-ethyl-14-methyldibenzo[a,c]acridine **2** consists of a slightly twisted dibenzo[a,c]acridine polyaromatic system in which the two fused benzo rings are not perfectly planar (dihedral angle *ca.* 18.0°) with respect to the pyridine ring N1-C14-C13-C15-C17-C22. At the 10- and 15-positions are located the ethyl and methyl groups, respectively.

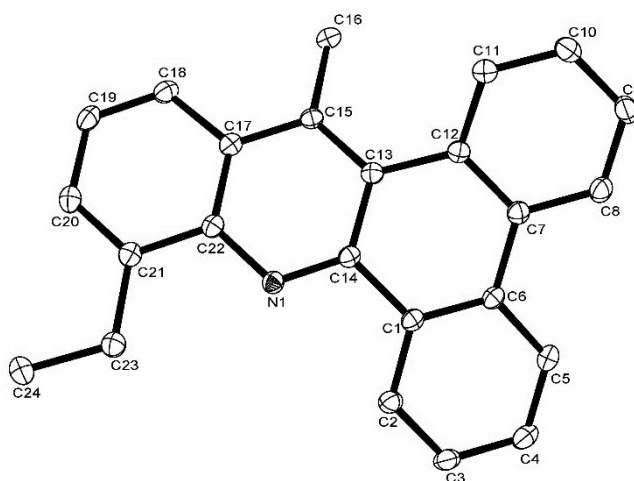


Figure S2. ORTEP diagram of **2** with thermal ellipsoids set at 50% probability level; all hydrogen atoms are omitted for clarity. N(1)-C(14) 1.324(2), N(1)-C(22) 1.364(2), C(22)-C(21) 1.434(3), C(14)-C(1) 1.479(3), C(1)-C(2) 1.401(3), C(21)-C(23) 1.508(3), C(10)-C(11) 1.378(3), N(1)-C(14)-C(1) 116.44(17), C(2)-C(1)-C(14) 120.05(18), C(22)-C(21)-C(23) 118.54(17)

Table S1. Crystal data and structure refinements for **1** and **2**

	1	2
Empirical formula	C ₂₄ H ₂₁ NO	C ₂₄ H ₁₉ N
Formula weight	339.42	321.40
Temperature/K	173(2)	173(2)
Wavelength/ Å	0.71073	0.71073
Crystal system	monoclinic	orthorhombic
Space group	<i>P</i> 2 ₁ / <i>c</i>	<i>Pbca</i>
a/ Å	9.2943(19)	11.335(2)
b/ Å	18.714(4)	7.4991(15)
c/ Å	20.557(4)	37.869(8)
Alpha/°	90.00	90.00
Beta/°	95.79(3)	90.00
Gamma/°	90.00	90.00
Volume/ Å ³	9.2943(19)	3218.9(11)
Z	8	8
D _{calcd} /(g·cm ⁻³)	1.268	1.326
μ/mm ⁻¹	0.077	0.076
<i>F</i> (000)	1440.0	1360.0
Crystal size/mm	0.375 × 0.261 × 0.083	0.289 × 0.115 × 0.031
θ Range (°)	4.54 to 50	5.6 to 54.96
Limiting indices	-11 ≤ h ≤ 11, -21 ≤ k ≤ 22, -24 ≤ l ≤ 24	-10 ≤ h ≤ 14, -9 ≤ k ≤ 9, -28 ≤ l ≤ 48
No. of rflns collected	19024	11343
No. unique rflns	6173	3672
R(int)	0.0365	0.0472
No. of params	473	228
Completeness to θ	98.6	99.4
Goodness of fit on F ²	1.038	1.172
Final R indices [I > 2 Σ (I)]	R ₁ = 0.0700, wR ₂ = 0.1381	R ₁ = 0.0693, wR ₂ = 0.1375
R indices (all data)	R ₁ = 0.0781, wR ₂ = 0.1456	R ₁ = 0.0799, wR ₂ = 0.1441
Largest diff. peak, and hole/(e Å ⁻³)	0.65/-0.57	0.29/-0.22

4. Perspective views of Pd2 and Pd3

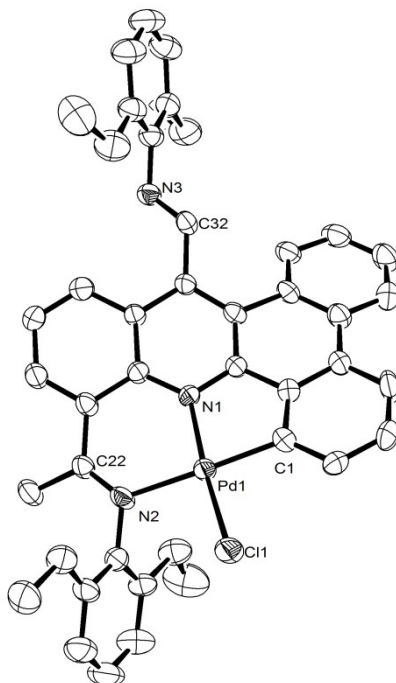


Figure S3. ORTEP diagram of **Pd2** with thermal ellipsoids set at 50% probability level; all hydrogen atoms are omitted for clarity. Selected bond length (Å) and angles (°) are given in Table 2.

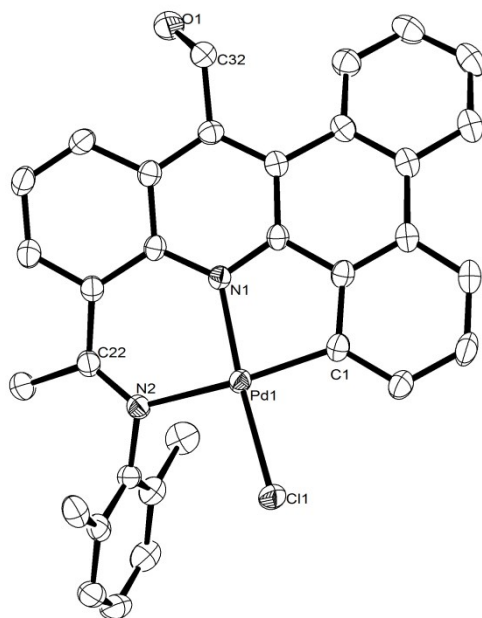


Figure S4. ORTEP diagram of **Pd3** with thermal ellipsoids set at 50% probability level; all hydrogen atoms are omitted for clarity. Selected bond length (Å) and angles (°) are given in Table 2.