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Quasi-planarDiazadithio and Diazadiseleno[8]circulenes: Synthesis, Structures and Properties

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Supporting Information

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General Information:

All reagents and solvents were reagent grade. Further purification and drying by standard methods were employed when necessary. The plates used for thin-layer chromatography (TLC) were E. Merck silica gel 60 F_{254} (0.24 nm thickness) precoated on aluminum plates, and then visualized under UV light (365 nm and 254 nm) or through staining with acidic ceric ammonium molybdate and subsequent heating. Column chromatography was performed using E. Merck silica gel (230-400 mesh). NMR spectra were recorded on a Bruker Advance III 400 spectrometer (400.19 MHz for ¹H and 100.62 MHz for ¹³C) and a Bruker 400 MHz NMR AV400Q (400.19 MHz for ¹H and 100.62 MHz for ¹³C) at room temperature. Mass spectra (ESI, EI and FAB) were obtained with a ThermoFinnigan MAT 95 XL spectrometer and determined at an ionized voltage of 70 eV unless otherwise stated.Melting points were measured on a digital melting point apparatus and were uncorrected. UV/vis absorption spectrum was recorded on a UV-Vis-NIR spectrophotometer. Fluorescence spectrum was obtained on a fluorescence spectrophotometer at 22 °C.

1. Experimental Section

Synthesis of N, N'-Di-n-butyl-tetraphenyleno[1,16-bcd:8,9-b'c'd']dipyrrole (6).

NaH (41.8 mg, 1.05 mmol of 60% dispersion in mineral oil) was added to a solution of 5 (313.5 mg,



0.95 mmol) in THF/DMF (1:1, 10 mL) at 0 °C. After stirring the mixture for 5 min, *n*-butyl bromide (135.6 mg, 0.99 mmol) was added, and the suspension was stirred for an additional 3 h. The reaction was diluted with dichloromethane (45 mL) and quenched with water (10 mL). The organic phase was washed with aqueous saturated NH₄Cl (20 mL), dried over Na₂SO₄ and concentrated under vacuum. Purification by column

chromatography (eluting with hexane/ethyl acetate 30:1) afforded **6** (230.9 mg, 55%) as a yellow solid.M.p.: 258.8-260.2 °C; IR (KBr): 2942, 2851, 1570, 1447, 1319, 1151, 1036 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.48 (t, *J* = 8.0 Hz, 4H), 7.33 (d, *J* = 8.0 Hz, 4H), 7.11 (d, *J* = 7.6 Hz, 4H), 4.32 (t, *J* = 7.4 Hz, 4H), 1.88-1.92 (m, 4H), 1.46-1.55 (m, 4H), 1.01 ppm (t, *J* = 7.4 Hz, 4H). ¹³C NMR (100 MHz, CDCl₃) δ 141.2, 137.5, 126.2, 124.3, 120.8, 107.4, 42.9, 30.9, 20.6, 13.9 ppm.HRMS (ESI) *m/z* calcd. for C₃₂H₃₀N₂Na [M+Na]⁺ 465.2301, found 465.2308.

Synthesis of2,4,5,7,9,11,12,14-Octabromo-*N*,*N*'-di-*n*-butyl-tetraphenyleno-[*1,16-bcd:8,9-b'c'd'*]dipyrrole (7).



To a 250 mL round-bottomed flask was placed $CHCl_3$ (100 mL), tetraphenylene**6**(663.4 mg, 1.5 mmol) and trifluoroacetic acid (0.5 mL). The mixture was stirred vigorously at 0 °C, and NBS (2.3 g, 12.6 mmol) was added in portions over 30 minutes. Then the reaction mixture was stirred at room temperature for 10 h, and poured into ice water (50 mL). The organic layer was separated, and the aqueous layer was extracted

with CH₂Cl₂ (30 mL × 3). The combined organic phase was washed with sat. NaHCO₃ and brine, and was dried over MgSO₄, filtered and the solvent was evaporated under reduced pressure. The residue was purified by column chromatography on silica gel (eluting with hexane) to give **7** (1.3 g, 82%) as a yellow solid. M.p> 300 °C; IR (KBr): 2934, 2864, 1536, 1452, 1364, 1253, 1049 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.88 (s, 4H), 5.18 (t, *J* = 7.8 Hz, 4H), 1.82-1.89 (m, 4H), 1.28-1.34 (m, 4H), 0.94 ppm (t, *J* = 7.4 Hz, 6H).¹³C NMR (100 MHz, CDCl₃) δ 138.5, 137.4, 132.3, 126.6, 118.6, 103.7, 44.8, 33.8, 19.7, 14.0 ppm. HRMS (ESI) *m/z* calcd. for C₃₂H₂₂Br₈N₂Na [M+Na]⁺ 1088.5142, found 1088.5147.

Synthesis of 4,5,11,12-tetrabromo-*N*,*N*'-Di-*n*-butyl-tetraphenyleno[*1,16-bcd*:8,*9-b'c'd'*]dipyrrole (8).



To a solution of 2,4,5,7,9,11,12,14-octabromotetraphenylene **7** (536.5 mg, 0.5 mmol) in dry THF (150 mL) was added dropwise *n*-BuLi (1.31 mL, 2.1 mmol, 4.2 eq, 1.6 M in hexane) at -78 °C under N₂ and stirred for 2 h at the same temperature. Then H₂O (0.18 mL, 10.0 mmol, 20.0 eq) was added at -78 °C and stirred for 1 h at this same temperature. The resulting mixture was then allowed to warm to room temperature and stirred for another 2 h. The reaction was guenched by addition of aqueous

ammonium chloride solution (30 mL), the aqueous solution was extracted with dichloromethane (25 mL \times 3). The combined organic phases was washed with water(20 mL) and brine (20 mL), dried over MgSO₄, filtered and the solvent was evaporated under reduced pressure. The residue was purified by column chromatography on silica gel (eluting with hexane/ethyl acetate 30:1) to give the desired product **8**(169.7 mg, 45%) as a yellow solid. M.p>300 °C; IR (KBr): 2948, 2855, 1639, 1545, 1375,

1103 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.66 (d, J = 8.4 Hz, 4H), 7.20 (d, J = 8.4 Hz, 4H), 4.28-4.32 (m, 4H), 1.86-1.93 (m, 4H), 1.39-1.46 (m, 4H), 0.98 ppm (t, J = 7.2 Hz, 6H). ¹³C NMR (100 MHz, CDCl₃) δ 140.8, 134.0, 132.2, 123.6, 117.6, 109.2, 43.6, 31.2, 29.8, 20.7, 14.0 ppm. HRMS (ESI) m/z calcd. for C₃₂H₂₆N₂Br₄Na[M+Na]⁺ 776.8722, found 776.8725.

<u>Synthesis</u> of <u>4,5,11,12-tetrabromo-N,N'-di-*n*-butyl-2,7,9,14-tetrakis(trimethylsilyl)tetraphenyleno[*1,16-bcd:8,9-*<u>*b'c'd'*]dipyrrole (9).</u></u>



To a solution of 2,4,5,7,9,11,12,14-octabromotetraphenylene **7** (858.5 mg, 0.8 mmol) in dry THF (300 mL) was added dropwise *n*-BuLi (2.1 mL, 3.36 mmol, 4.2 eq, 1.6 M in hexane) at -78 °C and stirred for 2 h at the same temperature. Then Me₃SiCl (0.65 mL, 4.8 mmol, 6.0 eq) was added at -78 °C and stirred for 1 h at the same temperature. The resulting mixture was allowed to warm to room temperature and stirred for another 2 h. The reaction was quenched by addition of aqueous

ammonium chloride solution (30 mL). The aqueous solution was extracted with dichloromethane (25 mL × 3). The combined organic phases was washed with water(20 mL) and brine (20 mL), dried over MgSO₄, filtered and the solvent was evaporated under reduced pressure. The residue was purified by column chromatography on silica gel (eluting with hexane/ethyl acetate 60:1) to give the desired product **9**(541.8 mg, 65%) as a yellow solid. M.p> 300 °C; IR (KBr): 2947, 2856, 1639, 1385, 1285 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.70 (s, 4H), 4.38 (t, *J* = 6.4 Hz, 4H), 1.13-1.22 (m, 4H), 0.73-0.77 (m, 4H), 0.60 (t, *J* = 6.8 Hz, 6H), 0.51 ppm (s, 36H). ¹³C NMR (100 MHz, CDCl₃) δ 149.7, 139.1, 134.8, 127.3, 124.9, 119.0, 51.2, 30.3, 20.3, 14.2, 0.6 ppm. HRMS (ESI) *m*/*z* calcd. for C₄₄H₅₈N₂Si₄Br₄ Na[M+Na]⁺ 1065.0303, found 1065.0307.

Synthesis



of dithiophene[2',3',4', 5':4,5;2'',3'',4'',5'':12,13]tetraphenyleno[1,16-bcd:8,9-b'c'd']dipyrrole, 3,4,8,18-tetra(trimethylsilyl)-4,10-dihydro-4,10-dibutyl (10).

To a solution of tetraphenylene**9** (208.4 mg, 0.2 mmol) in dry THF (60 mL) was added dropwise *n*-BuLi (0.6 mL, 0.96 mmol, 4.8 eq, 1.6 M in hexane) at -78 °C under N₂, and the mixture was stirred for 3 h at the

same temperature. Then S powder (96.0 mg, 3.0 mmol, 15.0 eq) was added in one portion at -78° C and stirred for 2 h at the same temperature. The resulting mixture was allowed to warm to room temperature and stirred overnight. The reaction was quenched by addition of aqueous ammonium chloride solution (20 mL), and the aqueous solution was extracted with dichloromethane (15 mL × 3). The combined organic phases were dried over MgSO₄, filtered and the solvent was evaporated under reduced pressure. The residue was purified by column chromatography on silica gel (eluting with hexane/ethyl acetate 20:1) to give **10**(118.5 mg, 75%) as a yellow solid. M.p>300 °C; IR (KBr): 2947, 2876, 1634, 1450, 1351, 1254, 1171 cm⁻¹;¹H NMR (400 MHz, CDCl₃) δ 8.29 (s, 4H), 4.74 (t, *J* = 7.2 Hz, 4H), 0.88-0.99 (m, 4H), 0.81-0.89 (m, 4H), 0.65 (s, 36H), 0.41-0.49 ppm (m, 8H). ¹³C NMR (100 MHz, CDCl₃) δ 147.3, 133.9, 129.7, 129.4, 123.1, 120.6, 53.6, 29.9, 20.1, 14.1, 1.3 ppm. HRMS (ESI) *m/z* calcd. for C₄₄H₅₈S₂N₂Si₄Na [M+ Na]⁺813.3011, found 813.3015.

Synthesis	of	diselenine[2',3',4',
5':4,5;2'',3'',4'',5'':12,13]te	traphenyleno[<i>1,16-bcd:8,9-b'c'd'</i>]dipyrrole	e <u>,</u>
3,4,8,18-tetra(trimethylsily)	l)-4,10-dihydro-4,10-dibutyl (11).	



To a solution of tetraphenylene**9** (312.6 mg, 0.3 mmol) in dry THF (100 mL) was added dropwise *n*-BuLi (0.9 mL, 1.44 mmol, 4.8 eq, 1.6 M in hexane) at -78 °C under N₂, and the mixture was stirred for 3 h at the same temperature. Then Se powder (355.3 mg, 4.5 mmol, 15.0 eq) was added at -78 °C and stirred for 2 h at the same temperature. The resulting mixture was allowed to warm to room temperature and stirred overnight. The reaction was quenched by addition of aqueous

ammonium chloride solution (30 mL), the aqueous solution was extracted with dichloromethane (20 mL \times 3). The combined organic phases was washed with water(20 mL) and brine (20 mL), dried over MgSO₄, filtered and the solvent was evaporated under reduced pressure. The residue was purified by column chromatography on silica gel (eluting with hexane/ethyl acetate 20:1) to give the desired product **11**(207.3 mg, 78%) as an orange solid. M.p>300 °C; IR (KBr): 2945, 2858, 1351, 1261, 1167 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.32 (s, 4H), 4.71 (t, *J* = 7.2 Hz, 4H), 0.92-0.99 (m, 4H), 0.84-0.91 (m, 4H), 0.65 (s, 36H), 0.49-0.61 ppm (m, 6H). ¹³C NMR (100 MHz, CDCl₃) δ 147.9, 134.4, 133.0, 132.9, 122.8, 122.7, 52.78, 29.8, 20.1, 14.1, 1.4ppm.HRMS (ESI) *m*/*z* calcd. for C₄₄H₅₈Se₂N₂Si₄Na [M + Na]⁺909.1899, found 909.1905.

2. X-Ray Crystal Structure Analyses

Single crystals of 7(clear yellow), **10** (clear yellow) and **11**(clear orange) were obtained by slow evaporation of their CH₂Cl₂ solution at room temperature. The single crystal X-ray diffraction was conducted on Agilent SuperNova diffractometer. The crystal structure was solved by direct methods *SIR2004* and refined by fullmatrixleast-squares on *F*2 by means of *SHELXL97*. The calculated positions of thehydrogen atoms were included in the final refinement.

Crystallographic data for 7: $C_{32}H_{22}Br_8N_2$ (1065.53), Triclinic, P-1, $0.40 \times 0.30 \times 0.20 \text{ mm}^3$, a = 10.8503(5), b = 12.4427(5), c = 12.9897(5) Å, a = 100.3(2), $\beta = 91.3(2)$, $\gamma = 108.6(2)^\circ$, T= 296 K, V = 1628.82(12) Å³, Z = 2,35914 reflections collected. The final $R_I = 0.0405$ ($I > 2\sigma(I)$), and the final $R_I = 0.0558$ (for all data). The goodness of fit on F² was 1.028



Figure S1.Crystal structure of 7.

Crystallographic data for **10**: C₄₄H₅₈N₂S₂Si₄ (791.40), Monoclinic, P2(1)/n, $0.40 \times 0.30 \times 0.20 \text{ mm}^3$, a = 14.2494(6), b = 14.1245(6), c = 22.4876(9) Å, a = 90.0, $\beta = 98.800(2)$, $\gamma = 90.0^\circ$, T = 296 K, V = 4472.7(3) Å³, Z = 4,53972 reflections collected. The final R_I = 0.0569 ($I > 2\sigma(I)$), and the final R_I = 0.0919 (for all data). The goodness of fit on F² was 1.023.



Figure S2. Crystal structure of 10.

Crystallographic data for **11**: C₄₄H₅₈N₂Se₂Si₄(885.20), Monoclinic, P2(1)/n, $0.40 \times 0.30 \times 0.20 \text{ mm}^3$, a = 14.2771(2), b = 14.2280(2), c = 22.6224(4) Å, a = 90.0, $\beta = 99.4270(10)$, $\gamma = 90.0^\circ$, T = 296(2) K, V = 4533.33(12) Å³, Z = 4,57800reflections collected. The final $R_I = 0.0471$ ($I > 2\sigma(I)$), and the final $R_I = 0.1114$ (for all data). The goodness of fit on F² was 1.010.



Figure S3. Crystal structure of 11.

3. Photophysical Study

UV-vis absorption spectra were recorded on a Varian CARY 1E UV-vis spectrophotometer.

Fluorescence spectra were taken on a Hitachi F-4500 spectrofluorometer.

The fluorescence quantum yields (Φ) of **10** and **11** were determined by comparing the photoluminescence integrated intensities (area) and absorbance intensities with anthracene, a reference compound with $\Phi = 0.28$.¹ The quantum yield was calculated by using the following equation:²

$$\Phi(\text{sample}) = \left(\frac{F_{\text{sample}}}{F_{\text{ref}}}\right) \left(\frac{A_{\text{ref}}}{A_{\text{sample}}}\right) \left(\frac{\eta_{\text{sample}}}{\eta_{\text{ref}}}\right)^2 \Phi_{(\text{ref})}$$

Where *F* is the integrated intensities (area under emission peak), *A* is the absorbance, η is the refractive index and Φ is the quantum yield, the subscript ref refers to reference.

Anthracene was dissolved in EtOH ($\eta = 1.36$) and heterocycle [8]circulene **10**, **11** were dissolved in CH₂Cl₂ ($\eta = 1.42$). All samples were measured by using UV-Vis spectrometer to obtain the absorbance, and the absorbance values were kept below 0.05 at the excitation wavelength in order to minimize re-absorption effects. The samples were then measured by photoluminescence spectrometer in order to get the photoluminescence emission intensity at the excitation wavelength (356 nm).

Sample	Integrated intensities (F)	Absorbance intensity at 356 nm (A)	Refractive index (η)	Quantum yield (Φ)
Anthracene	6214.14	0.038	1.36	0.28
10	553.63	0.018	1.42	0.057
11	183.29	0.017	1.42	0.019

Table S1 Quantum yields of 10 and 11.



Figure S4. The absorption spectra of 10 and 11 recorded in CH₂Cl₂ at 20 °C.

4. Electrochemical Study

The cyclic voltammetry was performed in a solution of anhydrous CH₃CN with 0.1 M tetrabutylammoniumhexafluorophosphate (Bu₄NBF₄) as supporting electrolyte, at a scan rate of 50mV/s. A platinum bead was used as a working electrode, a platinum wire was used as an auxiliary electrode, and a silver wire was used as a psedo-reference. Ferrocene/ferrocenium(FeCp₂⁺/FeCp₂⁰) was used as an internal standard, and potentials were recorded versus FeCp₂⁺/FeCp₂^o, temperature, 20 °C.



Figure S5.Cyclic voltammogram of 10 and 11 recorded from a solution in CH₃CN



Figure S6. Cyclic voltammogram of 10 recorded from a solution in CH₃CN with varying scan rate (mV/s).



Figure S7. Cyclic voltammogram of 11 recorded from a solution in CH₃CN with varying scan rate (mV/s).

5. Theoretical calculation

All calculations were carried out using the Gaussian 09 software package at the density functional theory (DFT) level with the Becke's three-parameter hybrid exchange functional and the Lee-Yang-Parr correlation functional (B3LYP) employing the 6-311++G(d,p) basis set for all atoms, and the results correlated well with the crystal bond lengths and bond angles. The shape and energy of the frontier molecular orbitals of **7**, **8** and **9** were calculated using the same method tacking into account the CH₂Cl₂ solvent effect using the polarized continuum model (PCM).



10 X= S, 11 X= Se Figure S8.NICS values calculated for 10and 11.

	Co	mpound 1	0	Cor	1	
species	-2	0	+2	-2	0	+2
А	-1.1	-8.6	25.7	0.4	-7.6	31.9
В	-7.5	-7.6	49.1	-7.7	-7.2	44.6
С	-4.7	-8.7	54.5	-5.5	-8.1	51.0
D	-13.1	7.7	32.5	-13.1	7.5	29.8

Table S2. NICS values calculated for 10 and 11.



Figure S9. Aromatic regions of the ¹H NMR spectra of 9, 10 and 11 in CDCl₃ at room temperature.



Figure S10. Calculated frontier molecular orbitals of 10.



Figure S11. Calculated frontier molecular orbitals of 11.

Table S2Coordinate data sets and absolute energies for DFT optimized complexes at B3LYP/6-311++G(d,p).

10-3776.3183166	a.u.		C 3.10883500	-1.13376900	-0.18247000
S -0.00000500	-4.30771700	-0.08539200	C 3.57857900	-2.46990600	-0.24955700
S 0.00003700	4.29984600	-0.24094100	C 2.57729000	-3.42879200	-0.19355800
Si -5.30838700	0 -3.17027500	-0.67184800	C1.23121100	-3.07503100	-0.13399700
Si -5.30758100	3.13131000	-0.80266900	C 0.73386400	-1.74976100	-0.17518300
Si 5.30761900	3.13125700	-0.80290100	C 1.73733500	-0.73375500	-0.19496400
Si 5.30835700	-3.17033200	-0.67211400	C -5.05587800	-4.83404800	-1.54055300
N-3.92859600	-0.00347700	-0.14830900	C -6.35827500	-3.53275200	0.86403700
N 3.92862900	0 -0.00351900	-0.14849300	C -6.25461300	-2.07321200	-1.88493200
C -1.23121000	-3.07501800	-0.13394000	C -5.08715200	0.01056000	0.76244500
C -2.57729600	-3.42876400	-0.19343700	C -4.69908700	0.14687900	2.24209900
C -3.57857700	-2.46986700	-0.24938600	C -5.91035200	0.09664200	3.18093100
C -3.10881600) -1.13373600	-0.18232400	C -5.52313800	0.24275400	4.65573400
C -1.73731200	-0.73373600	-0.19488400	C -6.13967600	2.03788400	-2.09714500
C -0.73385200	-1.74975400	-0.17515000	C -6.44071800	3.42837200	0.68882300
C -1.73761800	0.72448900	-0.22246100	C-5.07600100	4.83075000	-1.60527200
C -3.10903000	1.12476000	-0.22666500	C 5.07602100	4.83068900	-1.60551800
C -3.57852200	2.45827700	-0.33991800	C 6.44080600	3.42833100	0.68855200
C -2.57726400	3.41838000	-0.31742500	C 6.13967100	2.03781600	-2.09739200
C -1.23110800	3.06633700	-0.24539100	C 5.08723100	0.01050400	0.76220200
C -0.73385800	1.74053300	-0.23923000	C 4.69924200	0.14680900	2.24187700
C 1.23117000	3.06632400	-0.24544800	C 5.91055600	0.09657300	3.18064500
C 2.57732700	3.41835200	-0.31754300	C 5.52342000	0.24266600	4.65547100
C3.57857300	2.45823900	-0.34008200	C 6.25452300	-2.07328400	-1.88526000
C 3.10907200	1.12472700	-0.22680900	C 6.35833300	-3.53281600	0.86370900
C1.73765600	0.72447000	-0.22254100	C 5.05578100	-4.83410400	-1.54080200
C 0.73390600	1.74052600	-0.23926400	H -2.81961500	-4.48314500	-0.24314600

H -2.81953100	4.47033300	-0.40334500	H -4.37821100	4.79912700	-2.44627100	
H 2.81960000	4.47030300	-0.40347300	H -4.72698500	5.59283900	-0.90289500	
H 2.81959500	-4.48317500	-0.24327900	H -6.04355200	5.16949900	-1.99057600	
H -6.02556400	-5.19590000	-1.89804600	H 6.04356100	5.16942800	-1.99085900	
H -4.64995000	-5.60553000	-0.88070700	H 4.72703200	5.59278900	-0.90313800	
H -4.39412300	-4.74812700	-2.40671500	H 4.37820200	4.79905800	-2.44649200	
H -7.25619500	-4.08632200	0.56939200	H 6.81121000	2.52169500	1.17011400	
H -6.68646500	-2.64184500	1.40355100	H 5.92076000	4.01817600	1.44957300	
H -5.80127900	-4.15863300	1.56783500	H 7.31329900	4.00596100	0.36506500	
H -7.10565300	-2.63436700	-2.28476200	H 6.30229500	1.01075700	-1.77069900	
H -5.62125900	-1.78506900	-2.72841700	H 7.11071500	2.46132400	-2.37408100	
H -6.64866400	-1.16014600	-1.43705100	H 5.52539500	1.99753600	-3.00197100	
H -5.64610600	-0.90840800	0.61509900	H 5.75553400	0.81649500	0.47030100	
H -5.75546900	0.81654800	0.47057000	H 5.64617800	-0.90846200	0.61481700	
H -4.00165500	-0.65731100	2.50064400	H 4.15930600	1.08812700	2.38717600	
H -4.15914900	1.08820000	2.38736400	H 4.00182700	-0.65738600	2.50045200	
H -6.44530500	-0.84989500	3.03939900	H 6.61841000	0.88975400	2.91246200	
H -6.61822600	0.88981500	2.91277600	H 6.44550900	-0.84995800	3.03907500	
H -6.40239200	0.20068700	5.30434300	H 6.40270900	0.20059700	5.30403200	
H -5.01910700	1.19661500	4.83855400	H 4.84217400	-0.55566600	4.96516100	
H -4.84187100	-0.55557200	4.96539600	H 5.01939300	1.19652300	4.83832800	
H -5.52542800	1.99761200	-3.00174400	H 6.64863200	-1.16023100	-1.43740200	
H -7.11072800	2.46139300	-2.37380000	H 5.62111700	-1.78512000	-2.72869800	
H -6.30228800	1.01082200	-1.77045700	H 7.10552100	-2.63446000	-2.28515300	
H -7.31321000	4.00602400	0.36537300	H 6.68657400	-2.64191100	1.40319600	
H -5.92064000	4.01819100	1.44984200	H 7.25622600	-4.08640400	0.56901000	
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Н	4.64988200	-5.60558100	-0.88093300	С	3.41058500	-1.77516700	1.68874200
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н	4.39398000	-4.74817700	-2.40692800	С	1.55062900	-0.51008100	0.50626800
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11-	7782.9605851a.	u.		С	1.80714900	-1.55686800	-1.54108500
Se	-1.46441500	0.51894400	-3.95471900	С	1.73695700	-1.88292900	-2.91627800
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н	-4.56907500	1.84521900	1.03012600	н	1.51229800	-4.06861400	1.01572900
н	-3.89968700	1.89328300	2.65301100	н	0.67984700	-3.98421000	-0.52836100
н	-6.50360200	3.27074800	1.79777500	н	2.59678200	-6.23593200	0.28825800
н	-5.83690200	3.29278900	3.41864700	Н	1.71959300	-6.14999900	-1.22691300
н	-7.63109300	1.56557900	3.23166300	н	0.54995300	-7.66674300	0.37291600
н	-6.07962400	0.78576600	3.55923600	н	-0.46743700	-6.26367900	0.02853800
н	-6.75032700	0.76396800	1.92602300	н	0.41931000	-6.35762500	1.55256500
н	-0.95260100	6.69998600	2.30925100	н	4.96919000	-3.37139800	-2.62812000
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н	1.39531100	-4.70905800	-4.63625800				

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6. NMR spectra















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Bruker Advance III 400

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