Electronic Supporting Information

3 Biological fluorination from the sea: Discovery of a SAM-dependent nucleophilic fluorinating

- 4 enzyme from the marine-derived bacterium Streptomyces xinghaiensis NRRL B24674
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19 **1. Plasmid construction**

20 Gene sequence of the fluorinase was deduced from the whole genome of Streptomyces xinghaiensis 21 NRRL B-24674. The Streptomyces xinghaiensis fluorinase (flA) and other fluorinase genes were all 22 codon-optimised and synthesised by Shanghai Generay Biotech Co., Ltd. The synthetic gene flA was 23 then cloned into a pET-28a(+) vector with two restriction sites EcoRI/HindIII located in the upstream 24 and downstream of *flA* gene respectively.

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26 2. Site-directed mutagenesis

27 The overlap PCR method was used to introduce D16E, Y77A, T80S, F156A, S158A mutants as 28 previously described with slight modifications.¹ The amplified PCR product was then cloned into the 29 pET-28a(+) vector to generate pET-28a(+)-flA mutants respectively, which were all sequenced and 30 confirmed by Beijing BGI company. The primers used in site-directed mutagenesis were all listed in 31 Tab S2.

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33 3. Protein expression and purification

34 The plasmids were transformed into *E.coli* BL21(DE3) using the heat shock method. Simply, a mixture 35 of chemically competent bacteria and plasmid DNA was placed at 42°C for 90 s and then placed back on ice. E. coli BL21(DE3) cells, transformed with the synthetic fluorinase plasmids were grown in 36 37 Luria broth containing 100 mg/mL kanamycin at 37°C until the cell density reached an absorbance at 38 ~ 0.6 at 600 nm. The culture was then cooled down on ice for 30 min. Over-expression was induced 39 with 0.01 mM isopropylthiogalactoside (IPTG) and continued the incubation at 16°C for around 30 h. 40 Cells were harvested by centrifugation at $6000 \times g$ for 15 min at 4°C. The cell pellets were re-suspended in a lysis buffer (25 mM Tris-HCl, pH 8.0, 300 mM NaCl, 2.5 mM β-mercaptoethanol) 41 42 and lysed by sonication. Cell debris was removed by centrifugation (8000 × g, 20 min, 4°C). The 43 supernatant was filtered and loaded onto a 5 mL pre-packed His-Trap FF column (GE Healthcare), 44 which was connected to a AKTA Purifier (GE Healthcare). Protein-bound Ni²⁺ beads were washed with 45 washing buffer (25 mM Tris-HCl, pH 8.0, 300 mM NaCl, 50 mM imidazole) and eluted with elution 46 buffer (25 mM Tris-HCl, pH 8.0, 300 mM NaCl, 200 mM imidazole). The eluted protein was dialysed 47 against buffer (50 mM Tris-HCl, pH 7.0) and concentrated using a Millipore concentrator. The flow 48 rates of AKTA Purifier for sample loading, washing and elution were all 1 mL/min. The protein 49 concentration was measured based on Lambert-Beer Law and the extinction coefficient was determined

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- 50 using the ExPAsy ProtParam tool.
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52 4. Kinetics assay of fluorinase

53 For determining the kinetics of fluorinase for SAM, the fluorinase (14.0 mM) was incubated at various 54 concentrations of SAM (from 0 to 800 µM) and a saturating concentration of KF (200 mM) in 20 mM 55 phosphate buffer (pH 7.8) at various time points. The reaction temperature was 37°C. Similarly, for 56 determining the kinetics of fluorinase for KF, the fluorinase (15.5 μ M) was incubated at various 57 concentrations of KF (from 0 to 30 mM) and a saturating concentration of SAM (500 μ M) in 20 mM 58 sodium phosphate buffer (pH 7.8) at various time points. All reactions were quenched by addition of 59 200 μ L 10% trichloroacetic acid (TCA). All reactions were centrifuged (14000 × g, 10 min at 4°C) to 60 remove the precipitated protein. The samples were further filtered using a 0.22 µM Waterman filter 61 membrane and subjected to HPLC analysis. The level of 5'-FDA production was determined against a 62 standard curve of 5'-FDA. All experiments were performed in triplicate. Kinetic parameters were 63 obtained by least-square fitting of the initial velocity against substrate concentrations based on 64 Michaelis-Menten equation using GraphPad Prism 5.

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66 5. Verification of fluorinase

57 Streptomyces xinghaiensis fluorinase activity was tested at 37°C by monitoring 5'-FDA production using analytical HPLC. Assays were performed at 37°C for 10 min in 100 μ L of 20 mM sodium phosphate buffer (pH 7.8), 400 μ M SAM, 0.5 mg/mL purified protein, 200 mM KF. Reactions were quenched by 200 μ L 10% trichloroacetic acid (TCA) and then centrifuged (14000 × g, 10 min) to obtain the supernatant. The supernatant was further filtered using a 0.22 μ M Waterman filter and subjected to analytical HPLC or LC-MS analysis.

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74 6. Verification of chlorinase activity

Assays were performed as same as previously described for verification of fluorinase, but additional L-amino acid oxidase (Sigma-Aldrich) was added in the reaction mixture with concentration at 0.5 mg/mL for the purpose of pushing the reaction forward. Reactions were incubated at 37° C for 60 to 180 min, quenched by addition of 200 µL 10% trichloroacetic acid (TCA). The supernatant was obtained by centrifugation, which was filtered and subjected to HPLC or LC-MS analysis.

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81 7. HPLC and LC-MS identification for 5'-FDA

82 The supernatant was subjected to a 1260 infinity HPLC (Agilent Technologies) or a 6120 Quadrupole 83 LC-MS (Agilent Technologies) analysis on an analytic ZORBAX SB-C18 column (4.6×150 mm, 5 84 μm). The mobile phases for HPLC were composed of water containing 0.05% trifluoroacetic acid (TFA) 85 and CH₃CN containing 0.05% TFA; while the buffer for LC-MS was exactly the same apart from the 86 fact that the both water and CH₃CN buffer were void of TFA. The gradient applied for both HPLC and 87 LC-MS was shown as follows: 5% CH₃CN (with 0.05% trifluoroacetic acid) for 2 min, 5% to 25% 88 CH₃CN over 10 min and 25% CH₃CN for 2 min, 25% to 50% CH₃CN over 4 min and 50% CH₃CN for 89 2 min, 50% to 95% CH₃CN over 3 min and 95% CH₃CN for 2 min, 95% to 5% CH₃CN over 3 min and 90 5% CH₃CN for 2 min at a flow rate of 1 mL/min under 260 nm UV spectral monitoring. 91

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92 8. Circular dichroism

93 Circular dichroism was executed with protein concentration at 1.5 mg/mL in 20 mM sodium phosphate

- buffer (pH 7.8) using MOS-450 circular dichroism instrument (BioLogic) at ambient temperature.
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96 9. Analysis of protein stability under different conditions by thermal shift assay

- 97 Thermal shift analysis of protein stability was carried out in the presence of SYPRO Orange dye using 98 an Applied Biosystems StepOneTM Real-Time PCR (Life Technologies). Assays were performed in a 99 25 μ L reaction containing 3 μ M fluorinase, 3 μ M SYPRO Orange dye, in the presence and absence of 3 100 μ M relevant small molecule compounds. The ROX channel was chosen for fluorescence recording. 101 When the fluorinase melted (experienced thermal denaturation), it was able to give rise enhanced 102 fluorescence; thus, a graph of fluorescence intensity *vs*. temperature was plotted to calculate T_m 103 (melting temperature) values under different conditions. All experiments were performed in triplicate.
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105 **10. pH, temperature and metal ion effect on the activity of fluorinase**

106 In order to find optimal pH value, a reaction mixture (100 µL) containing 200 mM KF, 0.4 mM SAM 107 in 20 mM sodium acetate (pH 4.0-5.5), sodium phosphate (pH 6.0-7.0), Tris-HCl (pH 7.5-9.0), or 108 Gly-sodium hydroxide (pH 9.5-10.0) buffer was prepared. The reaction was at 37°C for 10 min with 109 fluorinase at 0.5 mg/mL. The supernatant of each reaction was then subjected to HPLC analysis. Effect 110 of temperature on fluorinase activity was performed in 20 mM Tris-HCl buffer (pH 7.8) and at varied 111 temperature. To measure the metal ion effects on the activity, a group of 100 µL reaction mixtures 112 containing 200 mM KF, 0.4 mM SAM in 20 mM Tris-HCl (pH 7.8) buffer were mixed with 1 mM 113 different divalent metal chloride salts (Mg²⁺, Mn²⁺, Fe²⁺, Cu²⁺, Zn²⁺, Ca²⁺, Ni²⁺ and Sn²⁺) respectively 114 or 1 mM EDTA (to remove the potential associated metal ions of the reaction). No metal ions nor 115 EDTA was added in the 'control' reaction. The relative fluorinase activity was reflected by the 116 production of 5'-FDA, which can be monitored by HPLC and quantified based on a standard curve of 117 5'-FDA. All experiments were performed in triplicate.

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119 11. Protein stability at room temperature

120 The purified fluorinase was maintained at 25°C for 0 day, 1 day, 2 days, 3 days and 4 days respectively. 121 Then assays were executed at 37°C for 10 min in 100 μ L reaction containing 50 mM Tris-HCl buffer 122 (pH 7.0), 400 μ M SAM, 200 mM KF and 0.5 mg/mL fluorinase. Reactions were quenched by addition 123 of 200 μ L 10% trichloroacetic acid. The samples were further processed for HPLC analysis. The 124 enzyme activity can be reflected by the production of 5'-FDA. All experiments were performed in 125 triplicate.





Fig. S1. A comparison of fluorinase, chlorinase and duf-62 enzymes.



Fig. S2. Amino acid alignment of the known fluorinases and other related proteins generated by
Jalview.² S.cat, S.MA-37, S.xin, Actinoplanes and N.bra denoted the fluorinases from Streptomyces
cattleya, Streptomyces sp. MA37, Streptomyces xinghaiensis NRRL B-24674, Actinoplanes sp.
N902-109 and Nocardia brasiliensis respectively.³ Chlorinase is from the marine actinomycete
Salinospora tropica.⁴ SAM hydroxide adenosyltransferase (duf-62) from Pyrococcus horikoshii.⁵ The
diagnostic characteristic of fluorinases to chlorinase and duf-62 enzymes is the boxed 22 amino acid
loop.

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139 Fig. S3. Phylogenetic tree of some functionally related enzymes in terms of protein sequences

- 140 generated by MEGA software.⁶
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143 Fig. S4. SDS-PAGE showing the over-expression of fluorinases by BL21(DE3). 1: no IPTG; 2:

144 IPTG induction; 3: insoluble fraction of over-expression; 4: souble fraction of over-expression. The

145 over-expressed fluorinase was indicated by an arrow.



- Fig. S5. SDS-PAGE showing purified fluorinase. 1: loaded purified sample; 2: over-loaded sample.
- The purified fluorinase was indicated by an arrow.





- Fig. S7. CD spectra of wild type and mutated fluorinases to monitor secondary structures.



Fig. S8. Wild type and mutated S. Xinghaiensis fluorinase. A: D16E; B: T80A; C: Y77A; D:T80S; E: F156A; F: S158A.



Fig. S9. Melting profile of wild-type, mutant and small molecule complexed fluorinases in 50 mM Tris-HCl buffer (pH 7.0).



Fig. S10. Initial generation rate of 5'-FDA vs [SAM] for fluorinases from different sources.

Samples	Primers
WT	F:5'-GGAATTCATGTCTGCGGACCCGACCCAG-3'
	R:5'-CCCAAGCTTAGTTGGTTTCAACACGAACTTTC -3'
D16E	F _m :5'-GGAATTCATGTCTGCGGACCCGACCCAGCGCCCGATC
	ATTGGCTTCATGTCTGAACTGGGCACTACCGACGACTCCG-3'
¥77A	F _m :5'-GCGACCACCGCCCCGGCGACCGG-3'
	R _m :5'-TCAGTACCGGTCGCCGGGGGGGGGGGGGGGGGG
T80S	F _m :5'-CTACCCGGCGAGCGGTACTGAAACC-3'
	R _m :5'-TACGGGTTTCAGTACCGCTCGCCGG-3'
F156A	F _m :5'-CGGAACCGACTGCATATTCTCGTGAAATGG-3'
	R _m :5'-CGCAACCATTTCACGAGAATATGCAGTCGG-3'
S158A	F _m :5'-CGACTTTCTATTCTCGTGAAATGGTTGCGATCC-3'
	R _m :5'-CCATTTCACGAGCATAGAAAGTCGGTTCCGGACG-3'
T80A	F _m :5'-CTACCCGGCGGCCGGTACTGAAACC-3'
	Rm:5'-TACGGGTTTCAGTACCGGCCGCCGG-3'

Tab. S1. All primers used for generation of mutant fluorinases.

Samples	T _m (°C)
WT (50 mM Tris-HCl, pH 7.0)	50.9 ± 1.35
WT + Adenosine (50 mM Tris-HCl, pH 7.0)	55.5 ± 0.80
WT + 5'-FDA (50 mM Tris-HCl, pH 7.0)	54.6 ± 1.20
D16E (50 mM Tris-HCl, pH 7.0)	Not able to determine
Y77A (50 mM Tris-HCl, pH 7.0)	41.1 ± 0.39
T80S (50 mM Tris-HCl, pH 7.0)	40.5 ± 2.31
T80A (50 mM Tris-HCl, pH 7.0)	Not able to be determine
F156A (50 mM Tris-HCl, pH 7.0)	49.0 ± 1.03
S158A (50 mM Tris-HCl, pH 7.0)	41.3 ± 0.41

Tab. S2. Calculated T_m values of samples pictured on Fig. S9.

Samples	T _m (°C)	
WT (50 mM Tris-HCl, pH 7.0)	50.5 ± 0.65	
WT (50 mM Tris-HCl, pH 7.5)	44.3 ± 0.42	
WT (50 mM Tris-HCl, pH 8.0)	43.3 ± 1.19	
WT (50 mM Tris-HCl, pH 8.5)	46.5 ± 1.13	
WT (100 mM phosphate buffer, pH 6.0)	43.3 ± 0.30	
WT (100 mM phosphate buffer, pH 6.5)	43.5 ± 0.24	
WT (100 mM phosphate buffer, pH 7.0)	44.2 ± 0.55	

177 Tab. S3. Calculated T_m values of *Streptomyces xinghaiensis* fluorinase in different buffers.

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