

Electronic supplementary information

Theoretical Studies on Oxidation-Switchable Second-Order Nonlinear Optical Responses of Metallosalen-Keggin Polyoxometalate Derivatives

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Content

1. The highest occupied molecular orbital (HOMO) of **Pd-salen-POM**, **Ni-salen-POM** and **Co-salen-POM**.
2. Molecular orbitals of **Ni-salen-POM** and **oxo-Ni-salen-POM** involved in the crucial excited states.
3. Molecular orbitals of **Co-salen-POM** and **oxo-Co-salen-POM** involved in the crucial excited states.
4. The relative energies (eV) of the **Pd-salen-POM**, **oxo-Pd-salen-POM**, **Ni-salen-POM**, **oxo-Ni-salen-POM**, **Co-salen-POM** and **oxo-Co-salen-POM** compounds for several spin configurations have been calculated by different functionals.
5. The transition energies ΔE (eV), maximum absorption wavelengths λ_{\max} (nm), oscillator strengths f , and major contribution of all compounds obtained by PBE0 functional.

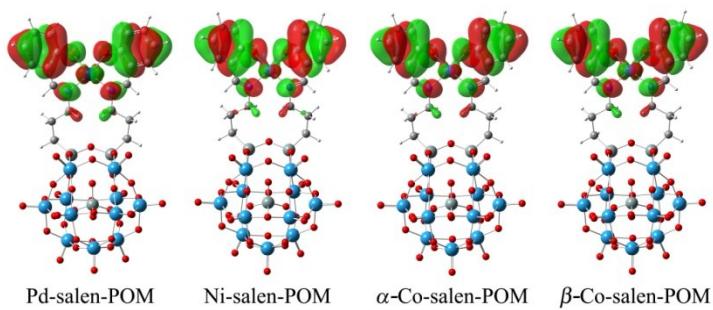


Figure S1. The highest occupied molecular orbitals (HOMO) of **Pd-salen-POM**, **Ni-salen-POM** and **Co-salen-POM**.

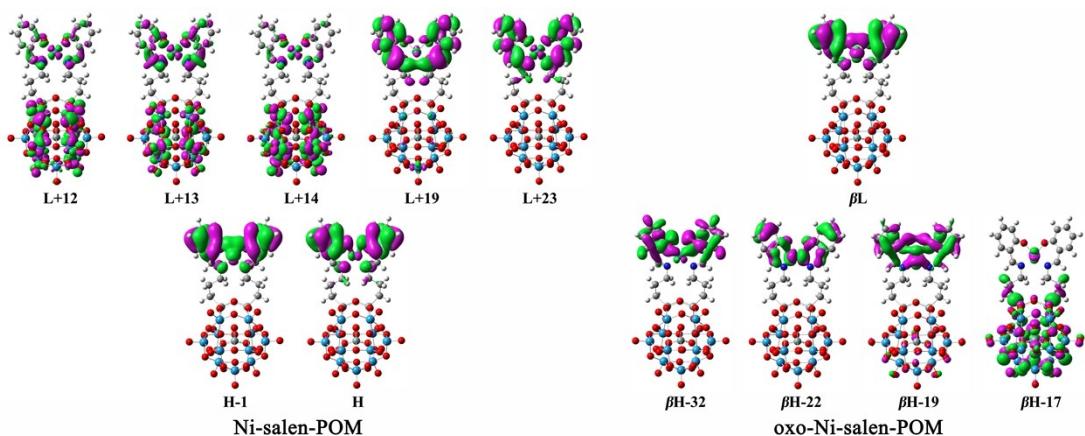


Figure S2. Molecular orbitals of **Ni-salen-POM** and **oxo-Ni-salen-POM** involved in the crucial excited states.

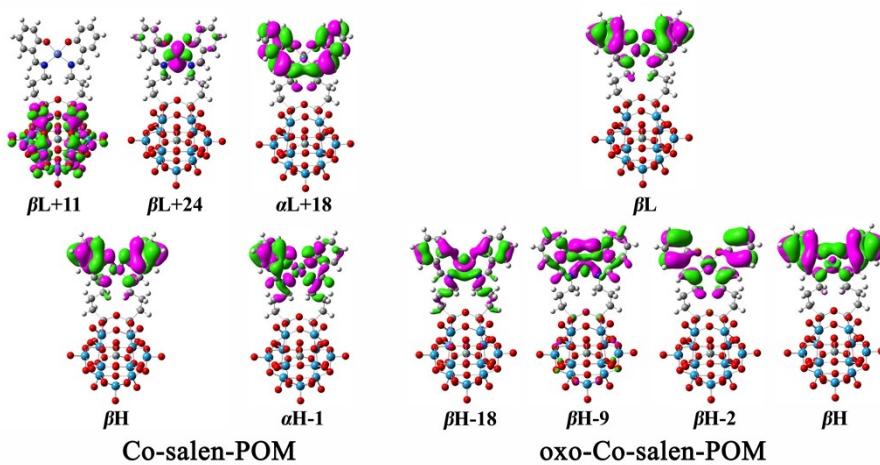


Figure S3. Molecular orbitals of **Co-salen-POM** and **oxo-Co-salen-POM** involved in the crucial excited states.

Table S1. The relative energies ΔE (eV) of the **Pd–salen–POM**, **oxo–Pd–salen–POM**, **Ni–salen–POM**, **oxo–Ni–salen–POM**, **Co–salen–POM** and **oxo–Co–salen–POM** compounds for several spin configurations have been calculated by different functionals.

Compounds	spin configuration	B3LYP (eV)	M06-2X (eV)	CAM-B3LYP (eV)	LC-BLYP (eV)	PBE0 (eV)
Pd–salen–POM	1	0	0	0	0	0
	3	1.980	2.069	2.141	2.070	2.102
oxo–Pd–salen–POM	2	0	0	0	0	0
	4	1.628	1.384	0.788	1.470	1.551
Ni–salen–POM	1	0	0	0	0	0
	3	0.801	3.089	2.649	0.211	0.879
oxo–Ni–salen–POM	2	0	0	0	0	0
	4	0.235	0.277	0.390	0.396	0.032
Co–salen–POM	2	0.190	0.301	0.270	0.381	0.096
	4	0	0	0	0	0
oxo–Co–salen–POM	1	0.886	1.185	1.372	1.878	1.300
	3	0	0	0	0	0
	5	0.246	0.143	0.122	0.513	0.117

Table S2. The transition energies ΔE (eV), maximum absorption wavelengths λ_{\max} (nm), oscillator strengths f , and major contribution of all compounds obtained by PBE0 functional.

Compounds	Excited state	ΔE (eV)	λ_{\max} /nm	f	Major contribution
Pd–salen–POM	S24	3.518	352.4	0.148	H→L+21 (34%), H-1→L+19 (31%)
			382.0 ^[a]		H→L+7 (8%)
Ni–salen–POM	S33	3.641	340.6	0.200	H-1→L+19 (65%), H→L+23 (16%)
	S32	3.618	342.7	0.037	H-1→L+13 (29%), H-1→L+12 (18%)
			370.0 ^[a]		H-1→L+14 (12%), H→L+19 (10%)
Co–salen–POM	S60	3.431	361.4	0.027	β H→ β L+24 (59%), β H→ β L+11 (7%)
			358.0 ^[a]		α H-1→ α L+18 (5%)
oxo–Pd–salen–POM	S4	1.539	805.4	0.027	β H-1→ β L (98%)
oxo–Ni–salen–POM	S19	2.655	467.1	0.035	β H-38→ β L (41%), α H-1→ α L+4 (12%)
					α H-39→ α L+4 (8%)
oxo–Co–salen–POM	S13	1.740	712.4	0.019	β H-19→ β L (82%), β H-17→ β L (5%)
	S15	2.010	616.8	0.049	β H-32→ β L (82%), β H-22→ β L (6%)
	S10	1.603	773.6	0.106	β H-18→ β L (45%), β H-9→ β L (23%)
					β H→ β L (712%), β H-2→ β L (11%)

[a] The experimental data of maximum absorption wavelengths.