# Polymeric Nanocapsules Templated on Liquid Cores as Efficient Photoreactors

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## General characterization of polymers



Figure S1 FT-IR spectra of PAH, PAH-graft-HTEMPO and PAH-graft-PVN.

Several bands characteristic for aromatic groups could be noticed in the FT-IR spectrum of PAH-*graft*-PVN: C-H out-of-plane bending (856 cm<sup>-1</sup>, 893 cm<sup>-1</sup>, 953 cm<sup>-1</sup>), C=C stretching (1452 cm<sup>-1</sup>), C-H stretching (2921 cm<sup>-1</sup>).

The FT-IR calibration was performed to calculate the content of VN in PAH-*graft*-PVN using the calibration curve shown in Figure S1. FT-IR spectra of series of blends of PAH and VN mixed at different weight ratios were recorded and integrated within 834 cm<sup>-1</sup>-865 cm<sup>-1</sup> range characteristic for C-H bending in aromatic rings.



**Figure S2** Calibration curve showing the dependence of the integrals of the band in the range 834 cm<sup>-1</sup>- 865 cm<sup>-1</sup> versus the content of VN in the PAH/VN blends.

**Table S1** Number-weighted particle size of PAH-*graft*-PVN at pH= 3 and pH= 7 obtained from DLS measurements before and after heating.

Sample	Before heating		After heating	
	pH=3	pH=7	pH=3	pH=7
Diameter (nm)	75±7	102±6	79±5	94±3



**Figure S3** SEM image of the copolymer aggregates deposited from the aqueous solution at pH=3 (a) and histogram of their size distribution (b).



**Figure S4** SEM image of the copolymer aggregates deposited from the aqueous solution at pH=7 (a) and histogram of their size distribution (b).

## Fluorescence studies of PAH-graft-PVN



Figure S5 Normalized steady-state emission spectra of PAH-*graft*-PVN solution (0.1 g/L) in 0.1 M NaCl at pH=3 and pH=7 ( $\lambda_{ex} = 280$  nm).

### Capsules templated on liquid cores



Figure S6 Photographs of the capsules stabilized by PAH-graft-PVN on toluene cores at pH=3 and pH=7.



Figure S7 TEM images of the capsules with C18/ferrocene cores deposited from their aqueous suspension.



**Figure S8** Steady-state emission spectra of capsules stabilized by PAH-*graft*-PVN at pH=3 and pH=7 excited at 280 nm.

## Photophysical properties of Pe in PAH-graft-PVN aggregates and capsules



Figure S9 Steady-state emission spectra of perylene ( $\lambda_{ex} = 410 \text{ nm}$ ) and naphthalene ( $\lambda_{ex} = 280 \text{ nm}$ ) in toluene solutions.



**Figure S10** Excitation spectrum of perylene dissolved in toluene (c=10<sup>-5</sup>M,  $\lambda_{em}$  = 460 nm).

The efficiency of energy transfer was calculated using the following ratiometric equation:

$$E_{rel} = \frac{I_A}{I_A + I_D}$$

where  $I_A$  and  $I_D$  correspond to the total acceptor (A) and donor (D) fluorescence intensities. The values of  $I_A$  and  $I_D$  were determined as appropriate integrals from the deconvoluted spectra of the aggregates or capsules (Fig. 8) into the isolated A (naphthalene) and D (perylene) components.

#### Photooxidation of solubilized and encapsulated Pe



**Figure S11** Steady-state emission spectra of Pe ( $\lambda_{ex} = 410 \text{ nm}$ ) solubilized in PAH-*graft*-PVN after various time of irradiation in Rayonet photoreactor.



**Figure S12** Normalized fluorescence intensity ( $\lambda_{ex} = 410 \text{ nm}$ ,  $\lambda_{em} = 450 \text{ nm}$ ) for perylene solubilized in PAHgraft-PVN after various times of irradiation in Rayonet photoreactor.



Figure S13 Steady-state emission spectra of Pe ( $\lambda_{ex} = 410 \text{ nm}$ ) encapsulated in the capsules after various times of irradiation in Rayonet photoreactor.



**Figure S14** Normalized fluorescence intensity ( $\lambda_{ex} = 410 \text{ nm}$ ,  $\lambda_{em} = 450 \text{ nm}$ ) for perylene encapsulated in the capsules after various times of irradiation in Rayonet photoreactor.

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