

Electronic Supplementary Information for

CoS₂ Nanoparticles–Graphene Hybrid as Cathodes Catalysts for Aprotic Li-O₂ Batteries

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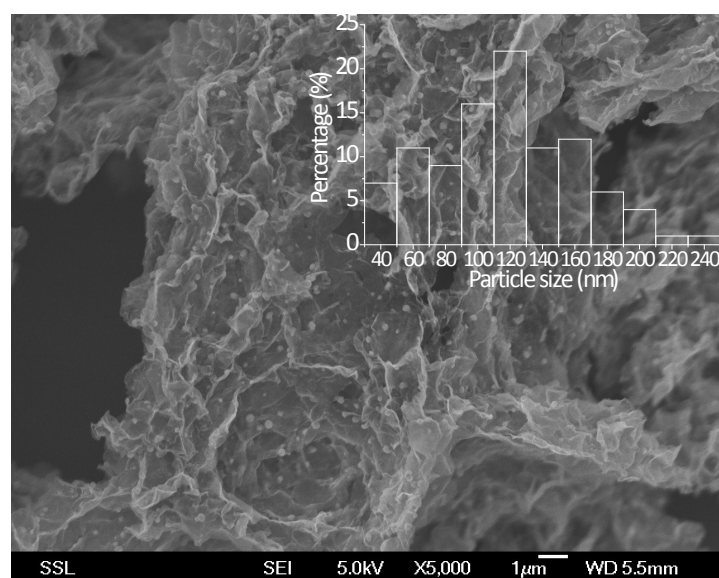


Fig. S1 Low-resolution SEM image of CoS₂/RGO hybrid and the corresponding particle size distribution in the inset.

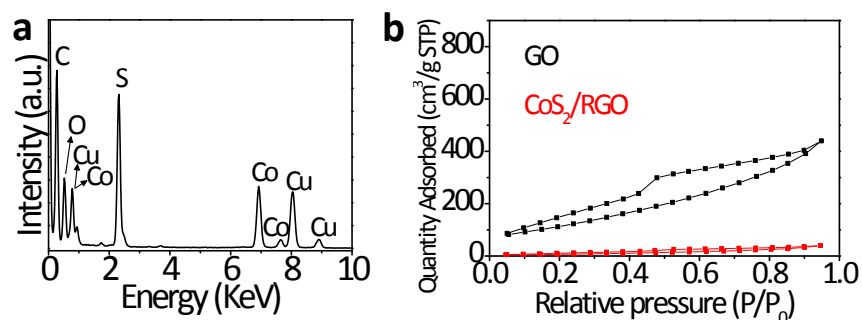


Fig. S2 EDX spectra (a) and N₂ adsorption/desorption isotherms (b) of the CoS₂/RGO hybrid. In (b), the N₂ adsorption/desorption isotherms of the RGO was also presented.

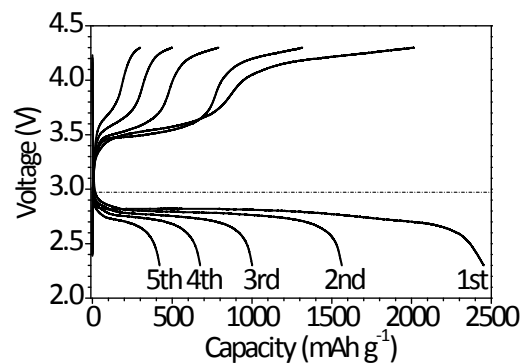


Fig. S3 Full discharge–charge curves of the Li-O₂ cell with CoS₂/RGO cathode and LiClO₄-DMSO electrolyte at a current density of 0.1 A g⁻¹.

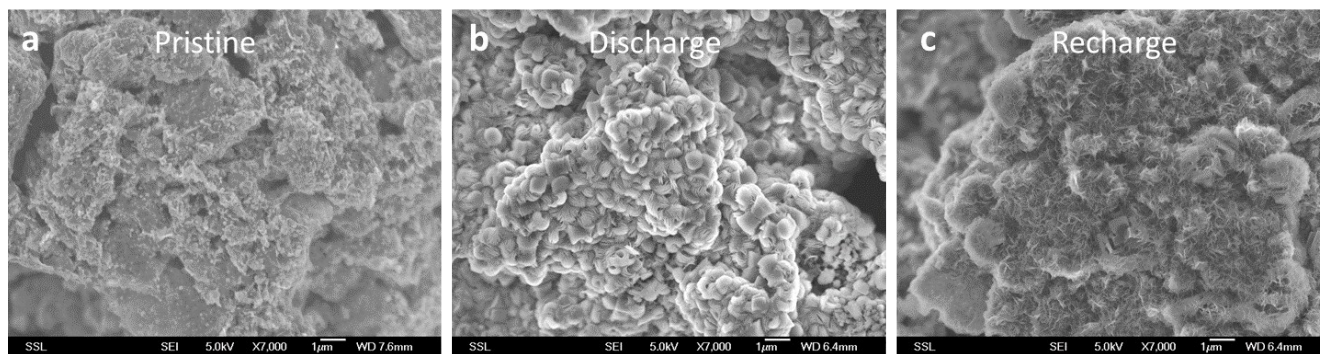


Fig. S4 Low-resolution SEM images for the pristine electrode (a), after discharge (b) and after recharge (c) for clear comparison at the 0.1 A g⁻¹ rate and first full discharge-charge states.

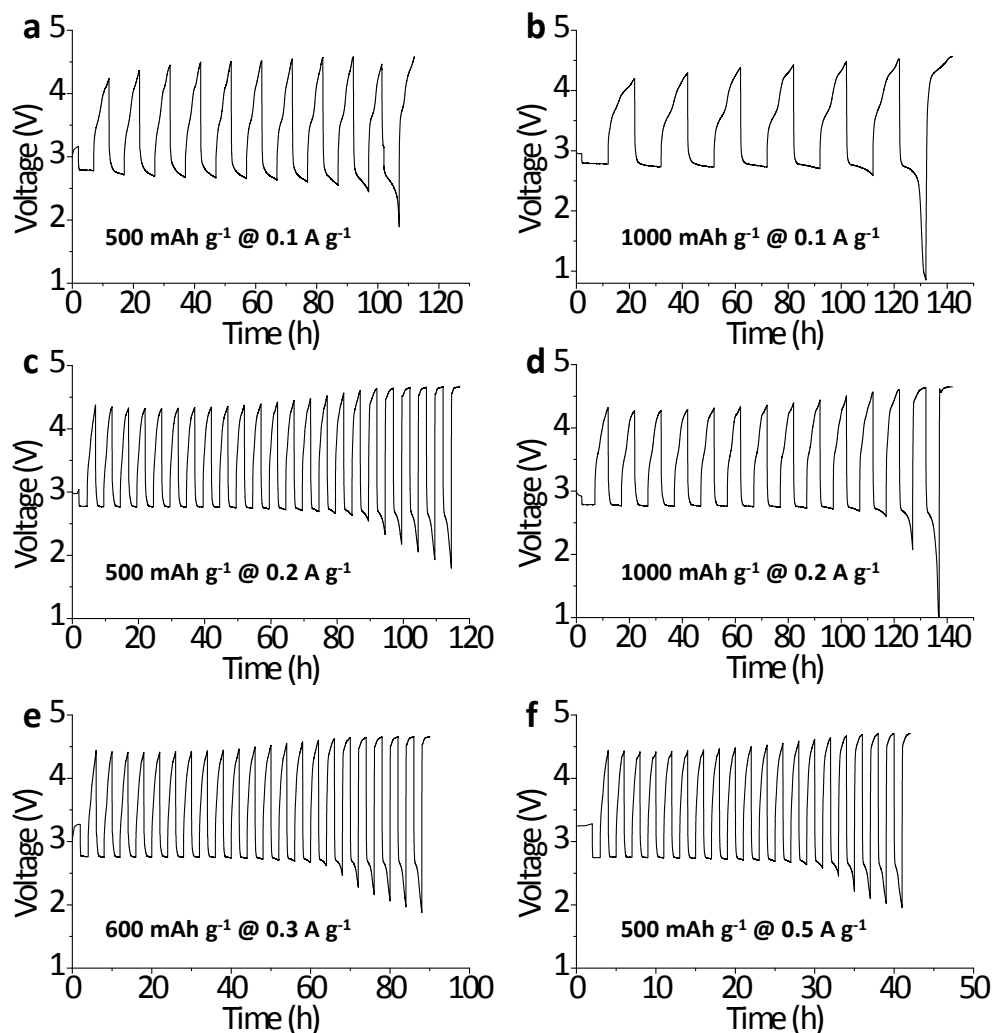


Fig. S5 The performances of the Li-O₂ cells with CoS₂/RGO hybrid cathodes and LiClO₄-DMSO electrolytes at different current densities (A g⁻¹) and different limited capacities. (a) 500 mAh g⁻¹ @ 0.1 A g⁻¹. (b) 1000 mAh g⁻¹ @ 0.1 A g⁻¹. (c) 500 mAh g⁻¹ @ 0.2 A g⁻¹. (d) 1000 mAh g⁻¹ @ 0.2 A g⁻¹. (e) 600 mAh g⁻¹ @ 0.3 A g⁻¹. (f) 500 mAh g⁻¹ @ 0.5 A g⁻¹.

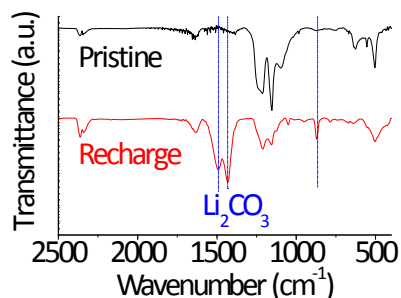


Fig. S6 FTIR spectrum of CoS₂/RGO hybrid at pristine and after 10th charge states.

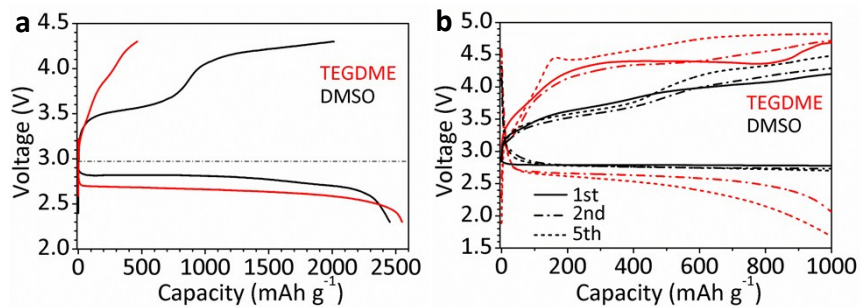


Fig. S7 The performances of the Li-O₂ cells with CoS₂/RGO cathodes and LiCF₃SO₃-TEGDME electrolytes in comparison with those of the LiClO₄-DMSO electrolytes based Li-O₂ cells. (a) First full discharge-charge curves at the rate of 0.1 A g⁻¹. (b) Discharge-charge curves at the rate of 0.1 A g⁻¹ and a limited capacity of 500 mAh g⁻¹.

In comparison with the LiClO₄-DMSO electrolyte based Li-O₂ cell, the Li-O₂ cell with LiCF₃SO₃-TEGDME electrolyte demonstrated a higher ORR overpotential (~0.28 V) upon discharge at a current density of 0.1 A g⁻¹ (Fig. S7a). Especially, no obvious charge platform was observed and a small charge capacity was obtained. This meant that the cell with LiCF₃SO₃-TEGDME electrolyte owned a very weak capability of oxidizing the Li₂O₂ upon charge, leading the lower round-trip efficiency (the ratio of discharge to charge voltage). As a result, the cycle stability was also unsatisfied (Fig. S7b).