BiOBr nanodots decorated hierarchical Bi₂WO₆ fabricating p-n heterostructure with enhanced photo-induced electric performance and photocatalytic degradation

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3. Result and discussion

3.1. Characterization of catalysts



Fig.S1 plots of $(\alpha hv)^2$ versus energy (hv) for the band gap energy of Bi_2WO_6 and various

BiOBr/Bi2WO6 composites

Fig. S1 shows that the band gaps of $BiOBr/Bi_2WO_6$ composites were about 2.4-2.5 eV. Therefore, when the composite material system was irradiated with visible light, the Bi_2WO_6 and BiOBr in the composite photocatalysts both absorbed photons, as well as excited electron and hole pairs.



Fig.S2 SEM images of prepared photocatalysts (a) lower magnification SEM images of 15
%BiOBr/Bi₂WO₆ Bi₂WO₆; (b) 5 %BiOBr/Bi₂WO₆; (c) 15 %BiOBr/Bi₂WO₆; (d) 25
%BiOBr/Bi₂WO₆

The morphologies of 15 %BiOBr/Bi₂WO₆ Bi₂WO₆ were observed using lower magnification SEM, which was better to observe overall morphology. As shown in Fig.S2a, the 15 %BiOBr/Bi₂WO₆ Bi₂WO₆ composites exhibited the similar morphology with the pure Bi₂WO₆, indicating that the deposition of BiOBr nanodots did not affect the flower-like microspheres of the Bi₂WO₆. Meanwhile, the BiOBr nanodots were not observed due to the lower magnification. The 5 %BiOBr/Bi2WO6, 15 %BiOBr/Bi2WO6 and 25 %BiOBr/Bi2WO6 composites was observed as shown in Fig. S2(b, c, d). It was clearly observed that the BiOBr nanodots were uniformly dispersed on the surface of the Bi2WO6 nanosheets, and did not affect the flower-like microspheres of the Bi₂WO₆. Fig. S1b displays the SEM image of the 5 %BiOBr/Bi₂WO₆ composites, where the BiOBr assemble on the surface of the Bi₂WO₆ microsphere with low distribution density. Meanwhile, the density of BiOBr nanodots increased gradually with the enhancement of BiOBr content. From the Fig. S2c, the BiOBr nanodots were homogeneously coated on the surface of Bi2WO6 with an intimately contact. The nanosheet building blocks of Bi₂WO₆ microsphere acted as clapboards to separate BiOBr and could endow a higher surface-tovolume ratio and more reactive sites to the photocatalysts, which was favorable for its photocatalytic activity. From the 25 %BiOBr/Bi₂WO₆ composite (as seen in Fig. S2d), the overloading BiOBr tended to agglomerate to some extent and cover the surface of Bi_2WO_6 .

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Photocatalysts	Bi ₂ WO ₆	5%	15%	25%
		BiOBr/Bi2WO6	BiOBr/Bi2WO6	BiOBr/Bi2WO6
Surface	9.23	13.56	17.45	11.78
area[m ² g ⁻¹]				

Table S3The specific surface area of as-prepared composite catalysts.

The specific surface area is an important factor to determine the adsorptive capability of the photocatalysts towards organic contaminant. From the results of the nitrogen sorption tests (as seen in Table S1), the specific surface area of pure Bi₂WO₆, 5 %BiOBr/Bi₂WO₆, 15 %BiOBr/Bi₂WO₆ and 25 %BiOBr/Bi₂WO₆ was calculated to be 9.23 m²/g, 13.56 m²/g, 17.45 m²/g and 11.78 m²/g, respectively. It is widely accepted that the larger specific surface areas could provide more reactive center sites and absorb more reactants, resulting in an improved photocatalytic activity. The 15 %BiOBr/Bi₂WO₆ possessed the larger specific surface areas than

that of 5 %BiOBr/Bi₂WO₆ and 25 %BiOBr/Bi₂WO₆ composite, which could explained as that the BET of the composites are increased with the content of BiOBr increased and suitable BiOBr content could be well dispersed on the surface of Bi₂WO₆ microsphere. The overloading content BiOBr nanodots were thought to agglomerate (seen in Fig. S1d) and shade the active sties on the surface of Bi₂WO₆, resulting in a lower degradation rates.



Fig.S3. Transient photocurrent response for 5 %BiOBr/Bi₂WO₆ and 15 %BiOBr/Bi₂WO₆ 25 %BiOBr/Bi₂WO₆ omposites

The photo-responses of 5% BiOBr/Bi₂WO₆, 15% BiOBr/Bi₂WO₆ and 25% BiOBr/Bi₂WO₆ samples performed under several on/off sunlight irradiation cycles at a particular time period shown in Fig. S3. It was noted that the 15 % BiOBr/Bi₂WO₆ composites exhibited the highest photocurrent intensity of the three samples, which was about 1.49 and 1.78 times than that of 5% BiOBr/Bi₂WO₆ and 25% BiOBr/Bi₂WO₆, indicating that the recombination of electrons and holes was greatly inhibited and the separation of photogenerated charges, and thus improving the photocatalytic activity.