## Fluorescence staining of salicylaldehyde azine and applications in the determination of potassium *tert*-butoxide

Jinlai Yang,<sup>a</sup> Jian Rui,<sup>a</sup> Xu Xu,<sup>ab</sup> Yiqin Yang,<sup>c</sup> Jun Su,<sup>a</sup> Haijun Xu,<sup>ab</sup> Yunyun Wang,<sup>a</sup> Nan Sun,<sup>a</sup> Shifa Wang\*<sup>ab</sup>

## Supplementary data

\*Corresponding author: Dr. and Prof. Shifa Wang

College of Chemical Engineering, Nanjing Forestry University, Nanjing, Jiangsu

210037, People's Republic of China

Jiangsu Key Lab of Biomass-based green Fuels and Chemicals, Nanjing, Jiangsu

- 210037, People's Republic of China
- Phone: +86-25-85427812; Fax: +86-25-85427812
- Email: wangshifa65@163.com

empirical formula	C <sub>14</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub>
formula weight	240.26
crystal system	monoclinic
space group	P 21/n
a [Å]	8.5168 (8)
b [Å]	6.3070 (6)
c [Å]	12.2581(10)
α [deg]	90
β [deg]	113.498 (5)
γ [deg]	90
volume [Å <sup>3</sup> ]	603.85 (9)
Z	2
Dc [g/cm <sup>-3</sup> ]	1.321
F (000)	252
T [K]	293 (2)
wavelength [Å]	0.71073
absorption coefficient [mm <sup>-1</sup> ]	0.090
completeness to theta	27.48 (99.3%)

 Table S1
 Crystal data, diffraction data, and refinement data of compound 1.

Table S2 Determination potassium tert-but	oxide concentration	
Sample	1	2
Content (%)	>97%	>97%
Weight (mg)	12.79	25.88
Volume of solution (ml)	100	100
Dosage (ml)	1	1
Dilution factor	10	10
Peak fluorescence intensity (a.u.)	204.39	322.61
Weight calculation (mg)	12.51	25.86
Content calculation (%)	97.81	99.91
RSD (%)	1.5	



Fig. S1 TGA traces of compound 1 under a  $N_2$  atmosphere heated at 10 °C /min.



Fig. S2 Bright-field transmission images (the first layer images), confocal fluorescence microscopic images (the second layer images) of paper (a), pledget (b), cotton (c) dyed by compound 1 ( $1.0 \times 10^{-3}$  M) and a mixture of compound 1 and KBr powder ( $1.0 \times 10^{-6}$  mol/g, d). Third layer images are confocal fluorescence microscopic images of a, b, c and d without compound 1.



Fig. S3 Application of anti-counterfeiting paper with compound 1 under sunlight (left) and 365 nm UV light (right).



Fig. S4 Application of indicator icon with compound 1.



Fig. S5 Fluorescence intensity of compound 1 ( $1 \times 10^{-4}$  M) in different pH HEPES buffers ( $\lambda_{ex385 \text{ nm}}$ ).



Fig. S6 Photographs of pledget (a) and filter paper (b) dyed by compound 1 without (right) and with (left) water washing for 5 min under 365 nm light.



Fig. S7 Photographs of pledget (a) and filter paper (b) dyed by compound 1 without (right) and with (left) ethanol washing for 5 min under 365 nm light.



Fig. S8 Photographs of pledget (a) and filter paper (b) dyed by compound 1 without (right) and with (left) sunlight treatment for 120 h.



Fig. S9 UV-vis absorption spectra (a) and fluorescence spectra (b) of compound 1 ( $1 \times 10^{-4}$  M) in different solvents ( $\lambda_{ex385 nm}$ ).



Fig. S10 UV-vis absorption spectral change (a) and fluorescence spectral change (b) of compound 1 in ethanol  $(1.0 \times 10^{-4} \text{ M})$  with different molar ratios of TFA and compound 1 ( $\lambda_{ex385 \text{ nm}}$ ).



Fig. S11 Mass spectrometry of compound 1 without (upper) and with (lower) equivalent of potassium *tert*-butoxide measured by LC-MS.