

A solution to achieve good reusability of MNPs Fe₃O₄-supported (S)-Diphenylprolinoltrimethylsilyl ether catalysts in asymmetric Michael reaction

Tao Wu,^a Dandan Feng,^a Bing Xie^{*b} and Xuebing Ma^{*a}

Table of contents

1. Elemental analysis.....	2
2. TGA.....	2
3. N ₂ adsorption-desorption isotherm.....	3
4. ¹ H and ¹³ C NMR spectra of intermediates and Jørgensen-Hayashi catalyst 5.....	5
5. Characterization of asymmetric Michael reaction products.....	9
6. Reusability of Fe ₃ O ₄ /PVP@SiO ₂ /ProTMS.....	24
7. TEM image of 10 th -recycled Fe ₃ O ₄ /PVP@ SiO ₂ /ProTMS.....	26

1. Elemental analysis

Table s1. The elemental analysis of supported Jørgensen–Hayashi catalysts

Entry	Catalyst	C (%)	H (%)	N (%)	S (%)	Loaded ProTMS (% , mmol/g)	Loaded MPTMS (% , mmol/g)
1	Fe ₃ O ₄ /PVP	0.01	0.23	0.05	/	/	/
2	Fe ₃ O ₄ @SiO ₂ /ProTMS	31.51	4.10	1.02	6.15	37.81, 0.73	8.9, 1.18
3	Fe ₃ O ₄ /PVP@SiO ₂ /ProTMS	31.56	3.22	0.9	5.42	33.15, 0.64	8.0, 1.06
4	Fe ₃ O ₄ /PVP@SiO ₂ /ProTMS ^a	34.27	3.93	1.76	5.62	/	/

^a 10th-recycled Fe₃O₄/PVP@SiO₂/ProTMS

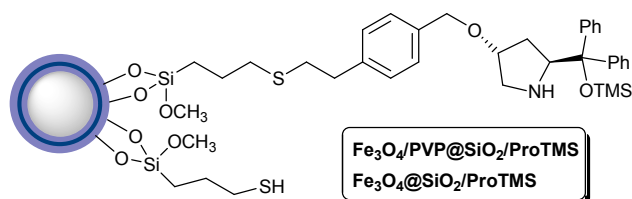
The calculation of loaded MPTMS, ProTMS and Jørgensen–Hayashi catalyst **5**:

In Fe₃O₄@SiO₂/ProTMS and Fe₃O₄/PVP@SiO₂/ProTMS catalysts, only Jørgensen–Hayashi catalyst **5** contained nitrogen element, the loading capacities of Jørgensen–Hayashi catalyst **5** was equal to that of ProTMS, and could be calculated according to elemental analysis (1.02% and 0.90%) as followed.

In Fe₃O₄@SiO₂/ProTMS: $1.02 \div (14 \times 100) \times 1000 = 0.73$ (mmol g⁻¹)

In Fe₃O₄/PVP@SiO₂/ProTMS: $0.9 \div (14 \times 100) \times 1000 = 0.64$ (mmol g⁻¹)

Furthermore, Jørgensen–Hayashi catalyst **5** was immobilized by the radical addition of sulfydryl (-SH) to C=C double bond to obtain ProTMS. It is confirmed that nitrogen in ProTMS has the same content as sulphur, and the total contents of sulphur (6.15% and 5.42%) are the sum of sulfide (-S-) and thiol (-SH).



Then, the total contents of sulphur:

In Fe₃O₄@SiO₂/ProTMS: $6.15 \div (32 \times 100) \times 1000 = 1.92$ (mmol g⁻¹)

In Fe₃O₄/PVP@SiO₂/ProTMS: $5.42 \div (32 \times 100) \times 1000 = 1.69$ (mmol g⁻¹)

The contents of free MPTMS unreacted with Jørgensen–Hayashi catalyst **5**:

In Fe₃O₄@SiO₂/ProTMS: $1.92 - 0.73 = 1.19$ (mmol g⁻¹)

In Fe₃O₄/PVP@SiO₂/ProTMS: $1.69 - 0.64 = 1.05$ (mmol g⁻¹)

The molar ratios of free MPTMS to ProTMS:

In Fe₃O₄@SiO₂/ProTMS: $1.19 / 0.73 = 1.63$

In Fe₃O₄/PVP@SiO₂/ProTMS: $1.05 / 0.64 = 1.64$

2. TGA

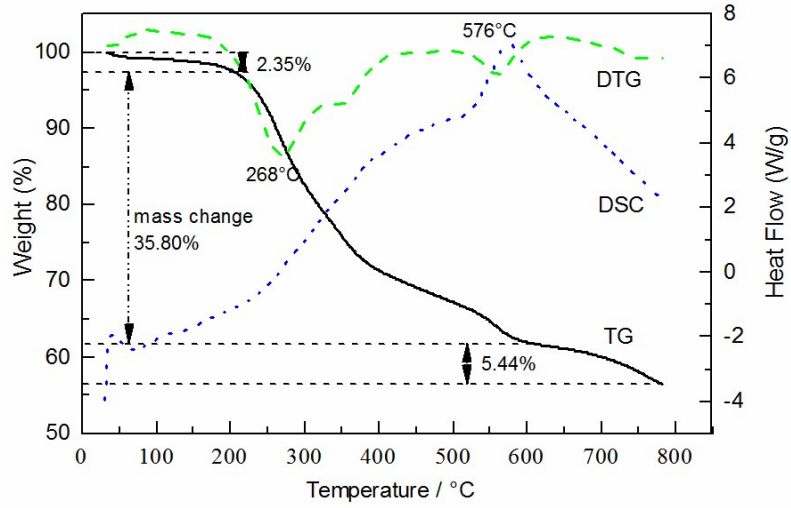


Fig. s1 TGA of Fe₃O₄@SiO₂/ProTMS

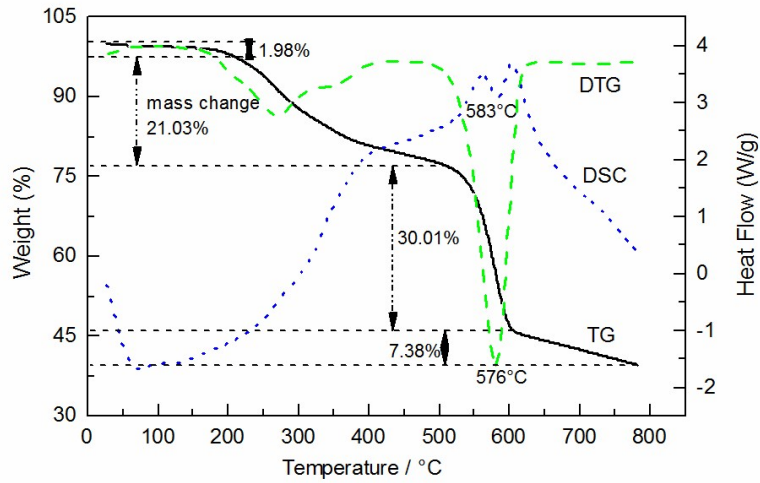


Fig. s2. TGA of Fe₃O₄/PVP@SiO₂/ProTMS

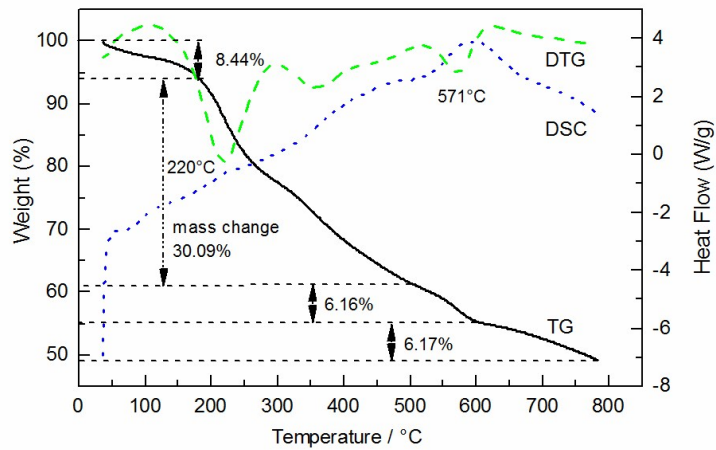


Fig. s3. TGA of 10th-recycled Fe₃O₄/PVP@SiO₂/ProTMS

3. N₂ adsorption-desorption isotherm

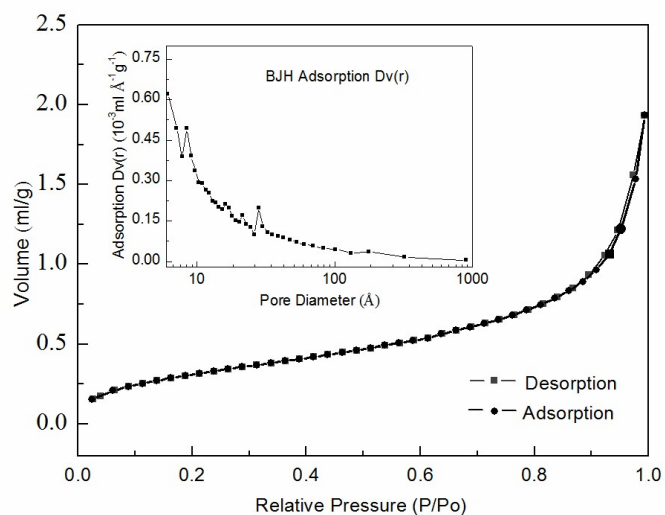


Fig. s4. N₂ adsorption-desorption isotherm and pore distribution of Fe₃O₄

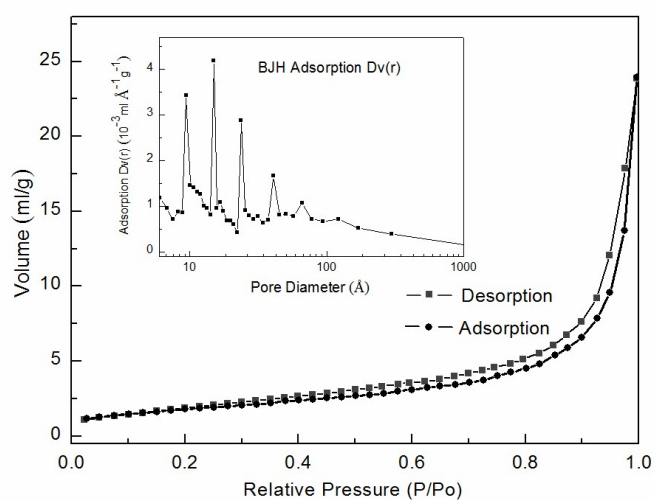


Fig. s5. N₂ adsorption-desorption isotherm and pore distribution of Fe₃O₄/PVP

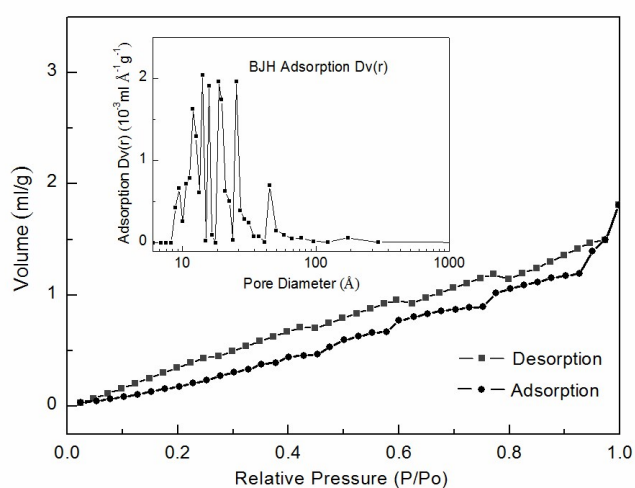


Fig. s6. N₂ adsorption-desorption isotherm and pore distribution of Fe₃O₄@SiO₂/ProTMS

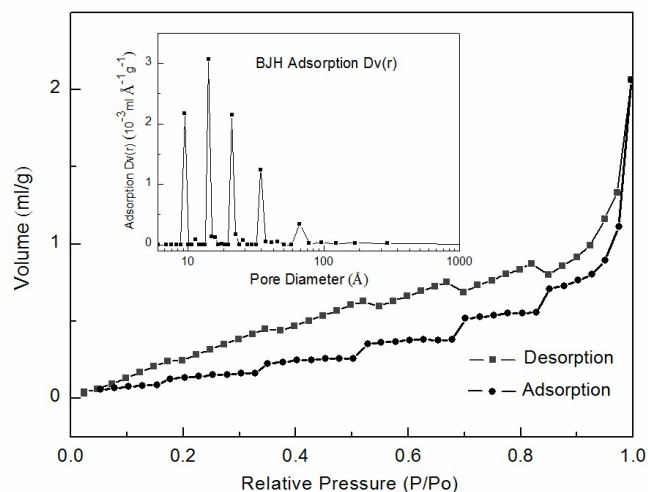


Fig. s7. N₂ adsorption-desorption isotherm and pore distribution of Fe₃O₄/PVP@SiO₂/ProTMS

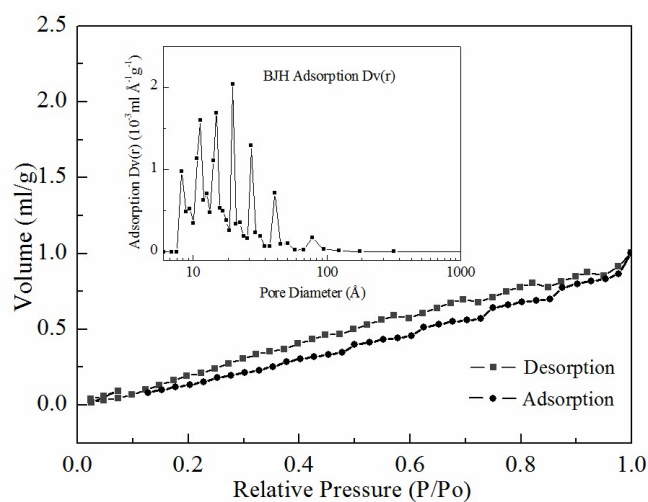


Fig. s8. N₂ adsorption-desorption isotherm and pore distribution of 10th-recycled Fe₃O₄/PVP@SiO₂/ProTMS

Table s2. The mesoporous properties of as-synthesized sample ^a

Entry	Catalyst	Surface Area [m ² /g] ^b	Average Pore Diameter [Å] ^c	Pore Volume [10 ⁻³ cc/g] ^d
1	Fe ₃ O ₄	7.91	51.9	20.53
2	Fe ₃ O ₄ /PVP	65.88	117.2	385.9
3	Fe ₃ O ₄ @SiO ₂ /ProTMS	3.34	75.69	12.65
4	Fe ₃ O ₄ /PVP@SiO ₂ /ProTMS	3.92	157.1	30.77
5	Fe ₃ O ₄ /PVP@SiO ₂ /ProTMS ^e	16.86	25.00	21.07

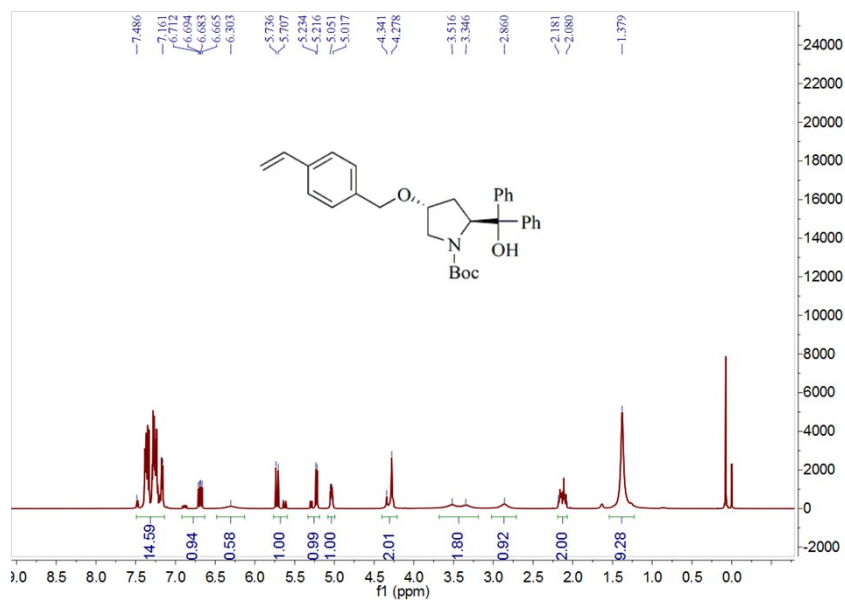
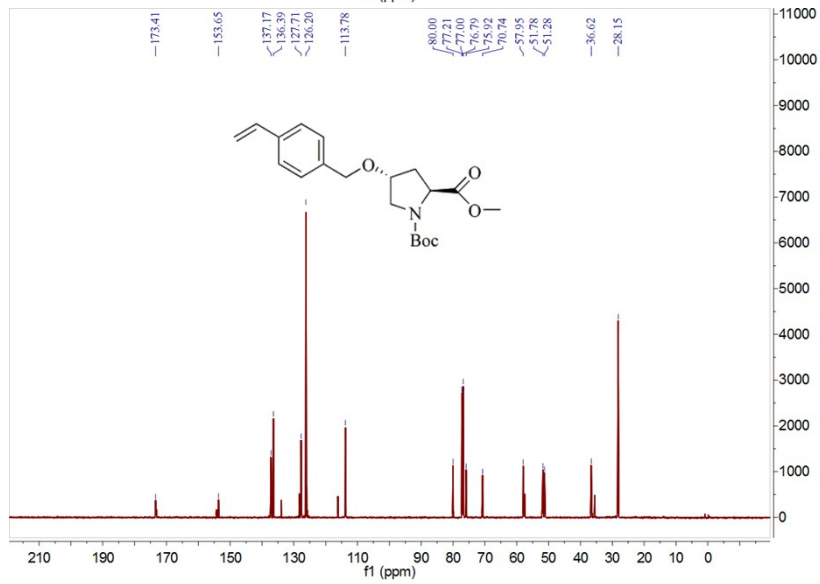
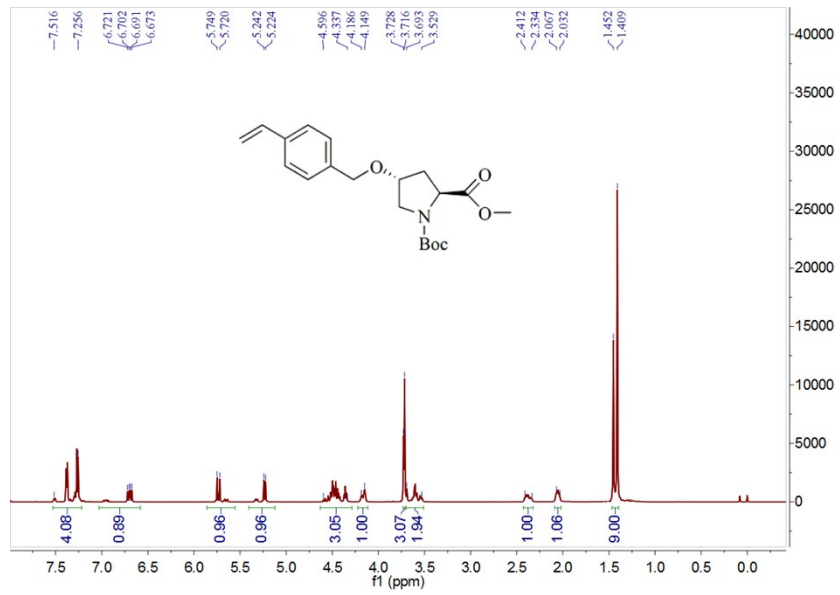
^a The sample was degassed at 100 °C for 5 h.

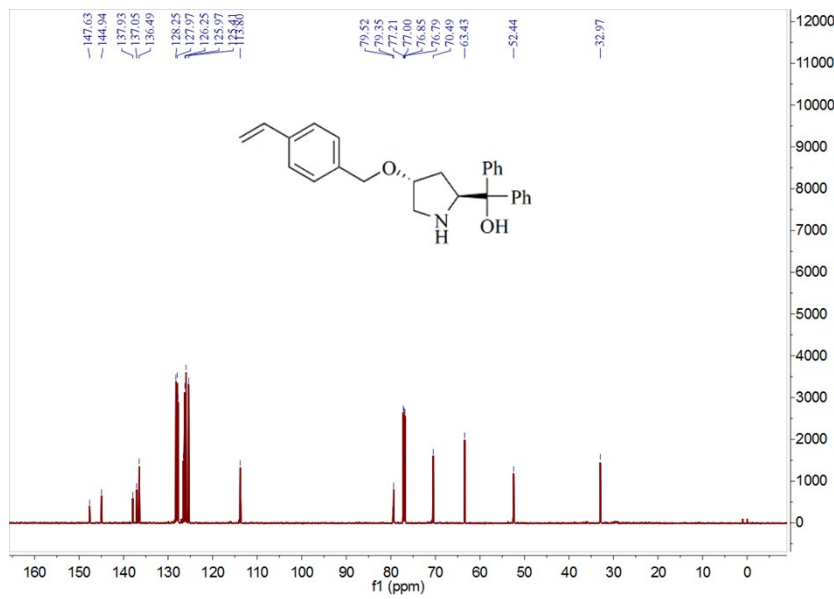
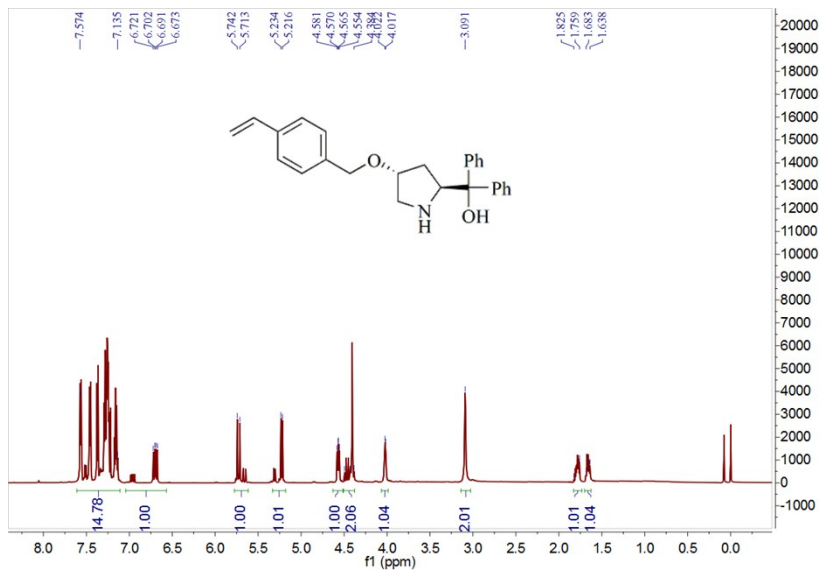
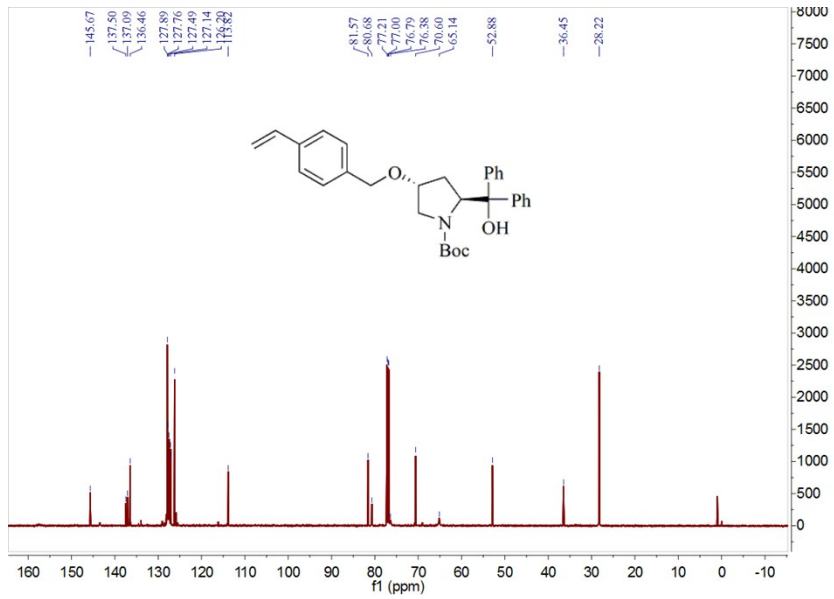
^b Based on multipoint BET method.

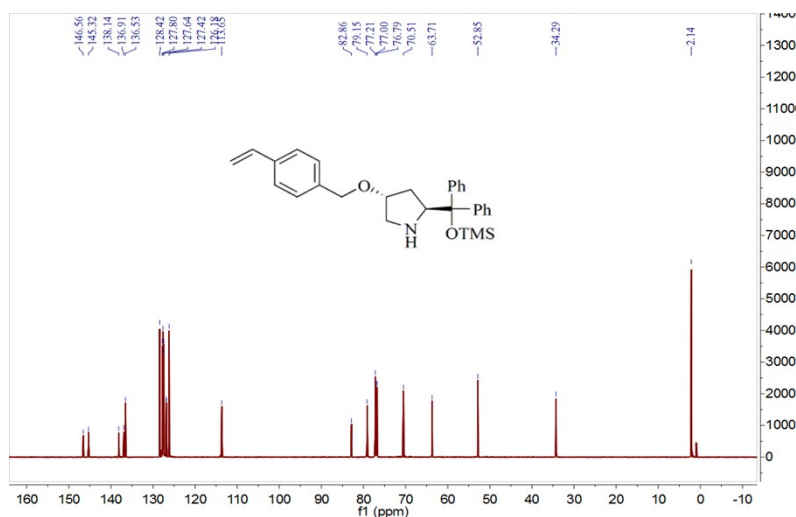
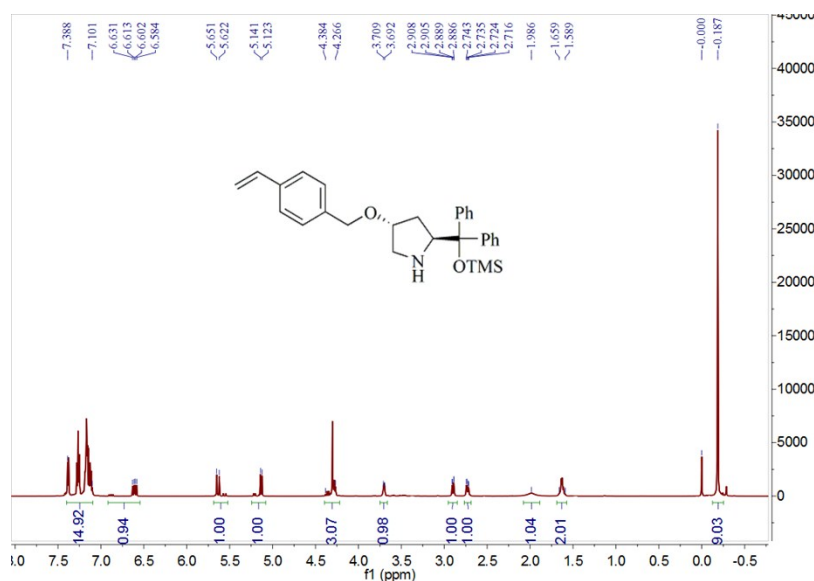
^c Based on the desorption data using BJH method. ^d Based on the desorption data of BJH method.

^e 10th-recycled Fe₃O₄/PVP@SiO₂/ProTMS

4. ¹H and ¹³C NMR spectra of intermediates and Jørgensen-Hayashi catalyst 5







5. Characterization of asymmetric Michael reaction products

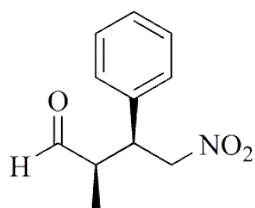
Racemic Michael adducts were prepared using pyrrolidine as catalyst according to the same procedure. Nitroalkenes derivatives were synthesized according to the reported literature. [1-3]

[1] Gao S. H., et. al., *Org. Lett.*, **2006**, 8, 2373-2376.

[2] Trost B. M., et. al., *J. Am. Chem. Soc.*, **2008**, 130, 2438-2439.

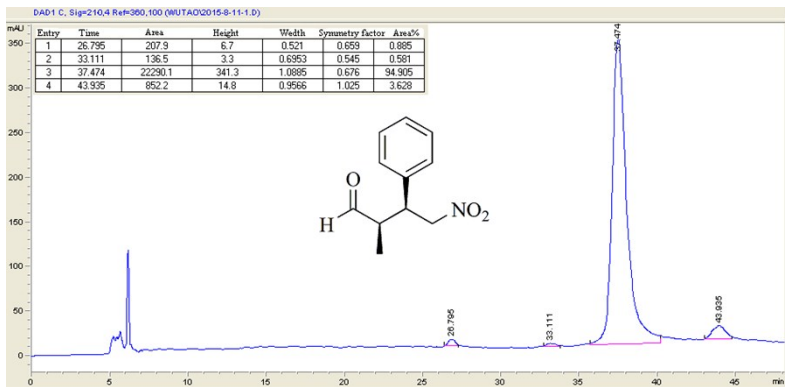
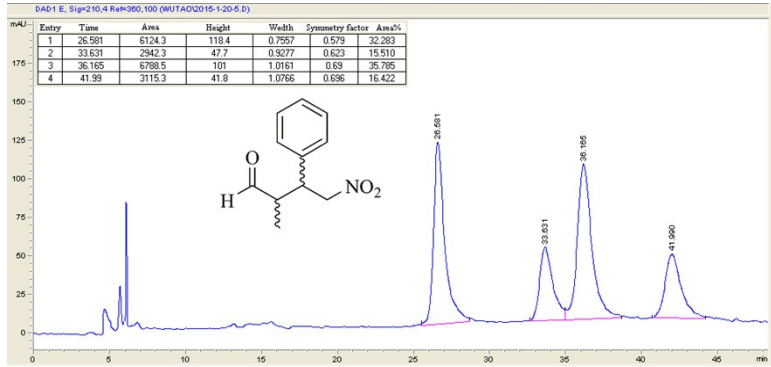
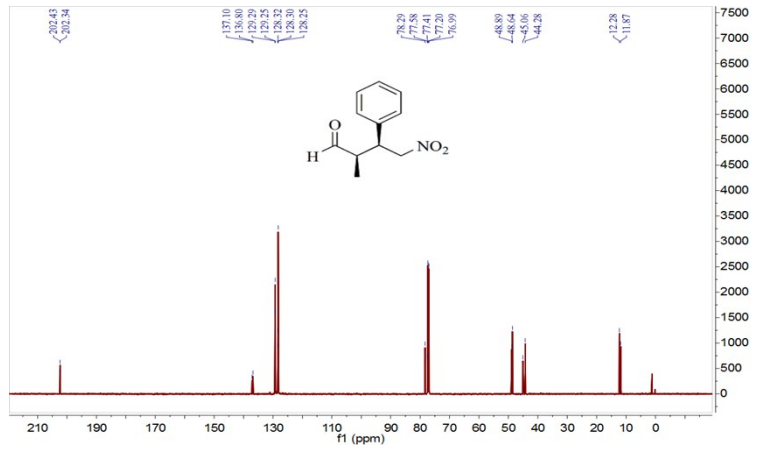
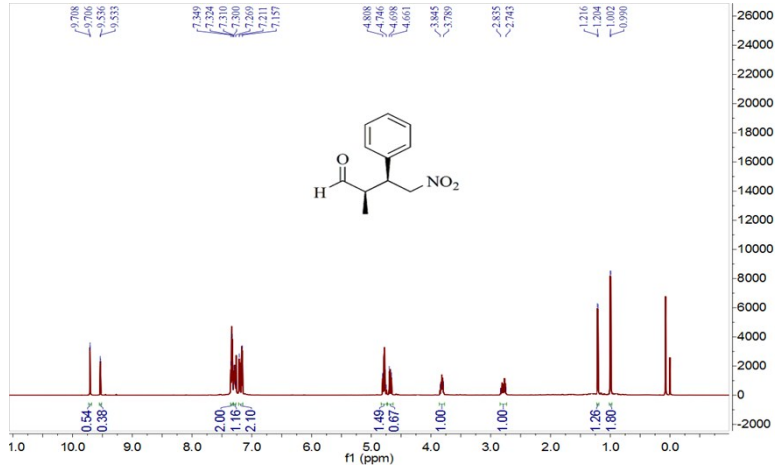
[3] Lopchuk J. M., et. al., *Org. Lett.*, **2013**, 15, 5218-5221.

(2R, 3S)-2-methyl-4-nitro-3-phenylbutanal: 96:4 dr, 98.2% ee, HPLC on Daicel Chiralpak OD-H column:

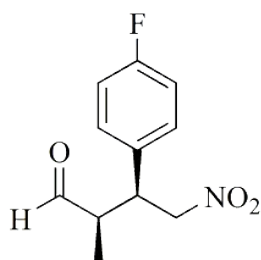


heptane/*i*-PrOH = 8/2, flow rate 1.0 mL min⁻¹, λ = 210 nm: t_R = 26.8 min (minor), t_R = 37.5 min (major); ¹H NMR (600 MHz, CDCl₃, TMS, major diastereomer): δ 1.00 (d, J = 7.3 Hz, 3H, CH₃), 2.74–2.84 (m, 1H, CHCH₃), 3.79–3.85 (m, 1H, CHCH₂), 4.77–4.81 (m, 2H, CH₂), 7.16–7.21 (m, 2H, Ar-*H*), 7.27–7.30 (m, 1H, Ar-*H*), 7.32–7.35 (m, 2H, Ar-*H*), 9.71 (d, J = 1.5 Hz, 1H, CHO); ¹³C NMR (150 MHz, CDCl₃, major diastereomer): δ 12.3 (CH₃),

44.3 (CHCH₂), 48.6 (CHCH₃), 78.3 (CH₂), 128.2, 128.3, 129.2, 129.3, 136.8 (Ph), 202.3 (C=O).



(2R, 3S)-3-(4-fluorophenyl)-2-methyl-4-nitrobutanal: 89:11 dr, 98.2% ee, HPLC on Daicel Chiralpak AD-H



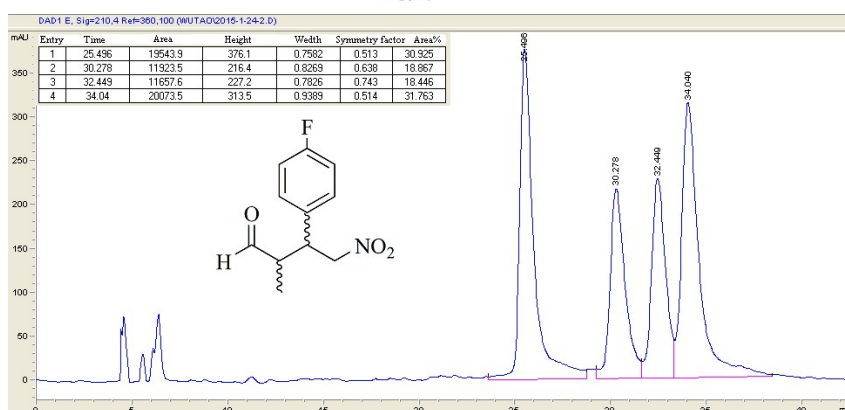
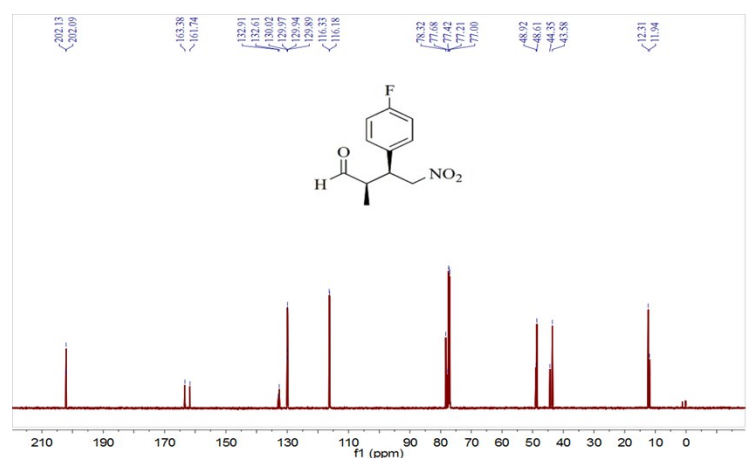
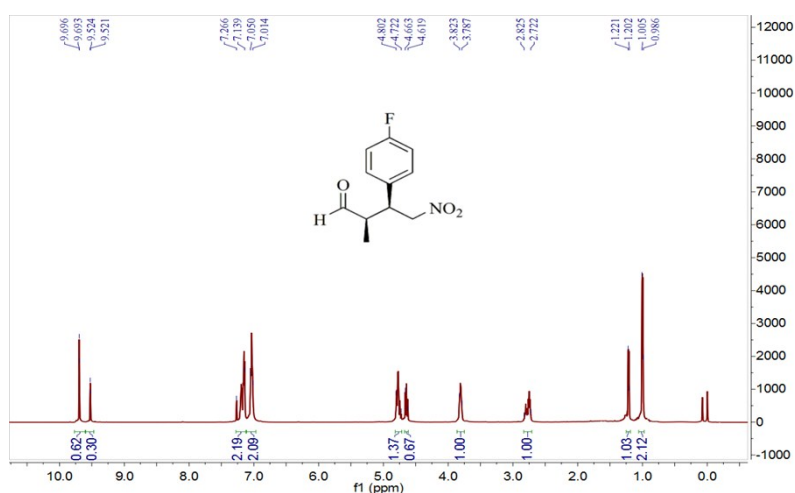
column: heptane/*i*-PrOH = 19/1, flow rate 1.0 mL min⁻¹, λ = 210 nm: t_R = 26.4 min

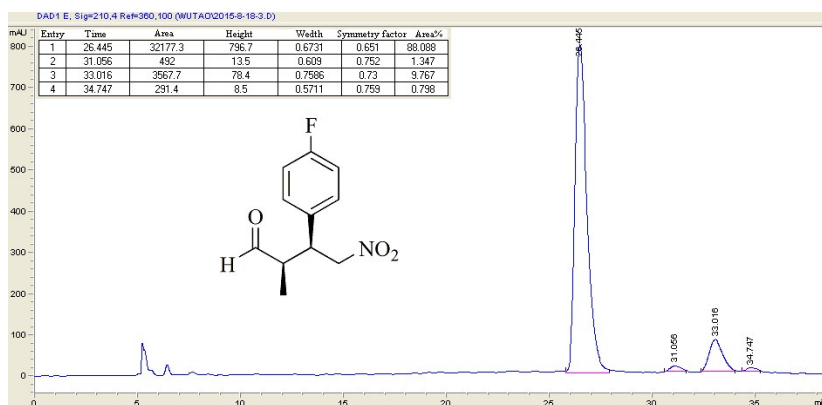
(major), t_R = 34.7 min (minor); ¹H NMR (600 MHz, CDCl₃, TMS, major diastereomer): δ

0.99–1.01 (m, 3H, CH₃), 2.72–2.83 (m, 1H, CHCH₃), 3.79–3.82 (m, 1H, CHCH₂), 4.72–4.80 (m, 2H, CH₂), 7.01–7.05 (m, 2H, Ar-H), 7.14–7.27 (m, 2H, Ar-H), 9.69 (s, 1H, CHO);

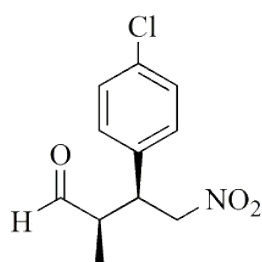
¹³C NMR (150 MHz, CDCl₃, major diastereomer): δ 12.3 (CH₃), 43.6 (CHCH₂), 48.6

(CHCH₃), 78.3 (CH₂), 116.2, 116.3, 129.9, 130.0, 132.6 (Ph), 202.1 (C=O).





(2R, 3S)-3-(4-chlorophenyl)-2-methyl-4-nitrobutanal: 87:13 dr, 97.0% ee, HPLC on Daicel Chiralpak AD-H



column: heptane/*i*-PrOH = 19/1, flow rate 1.0 mL min⁻¹, λ = 210 nm: t_R = 27.1 min

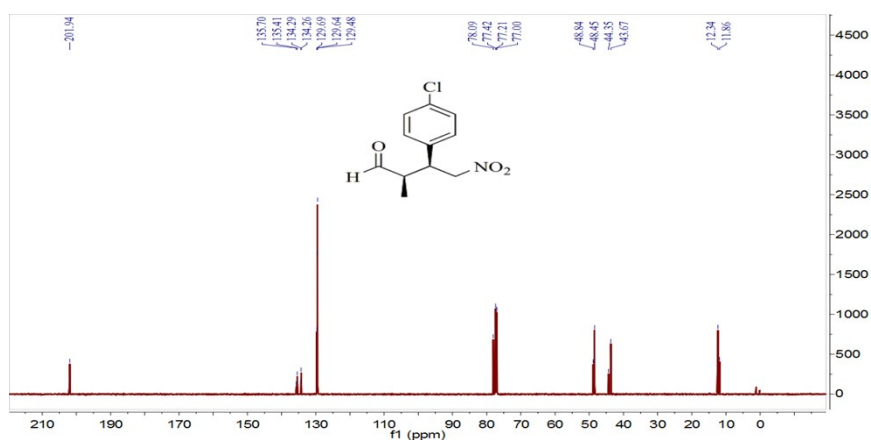
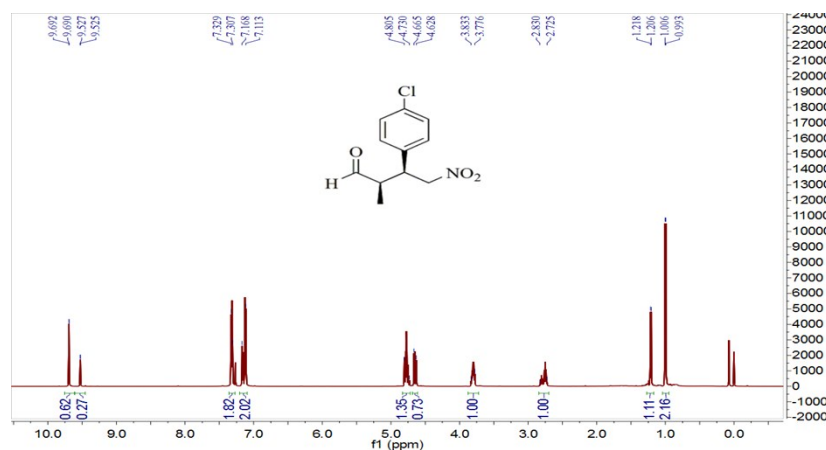
(major), t_R = 36.2 min (minor); ¹H NMR (600 MHz, CDCl₃, TMS, major diastereomer): δ

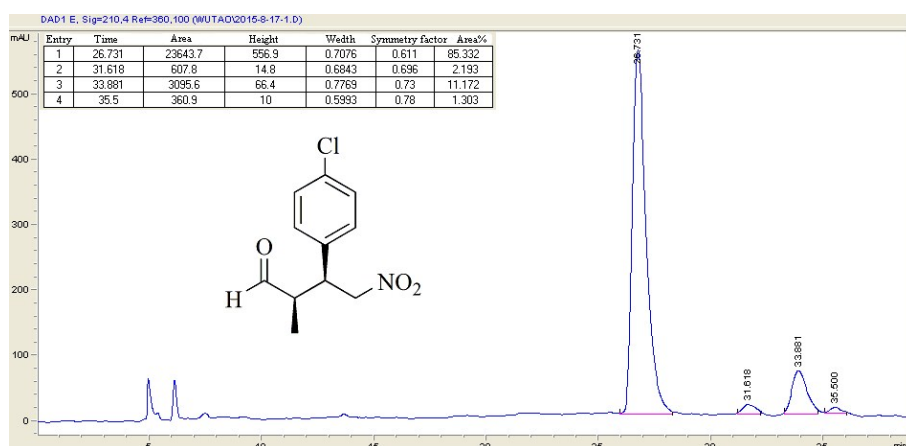
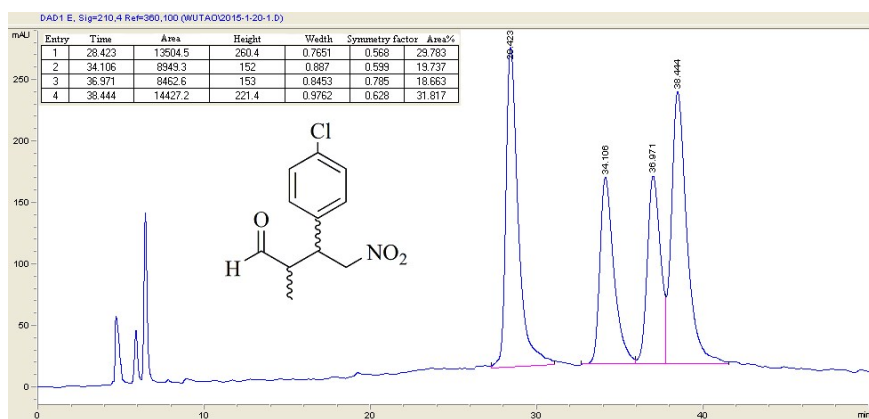
1.00 (d, *J* = 7.3 Hz, 3H, CH₃), 2.73–2.83 (m, 1H, CHCH₃), 3.78–3.83 (m, 1H, CHCH₂),

4.73–4.81 (m, 2H, CH₂), 7.11–7.17 (m, 2H, Ar-H), 7.31–7.33 (m, 2H, Ar-H), 9.69 (d, *J* =

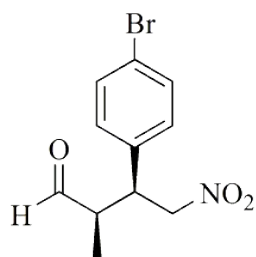
1.4 Hz, 1H, CHO); ¹³C NMR (150 MHz, CDCl₃, major diastereomer): δ 12.3 (CH₃), 43.7

(CHCH₂), 48.4 (CHCH₃), 78.1 (CH₂), 129.5, 129.6, 129.7, 134.3, 135.4 (Ph), 201.9 (C=O).



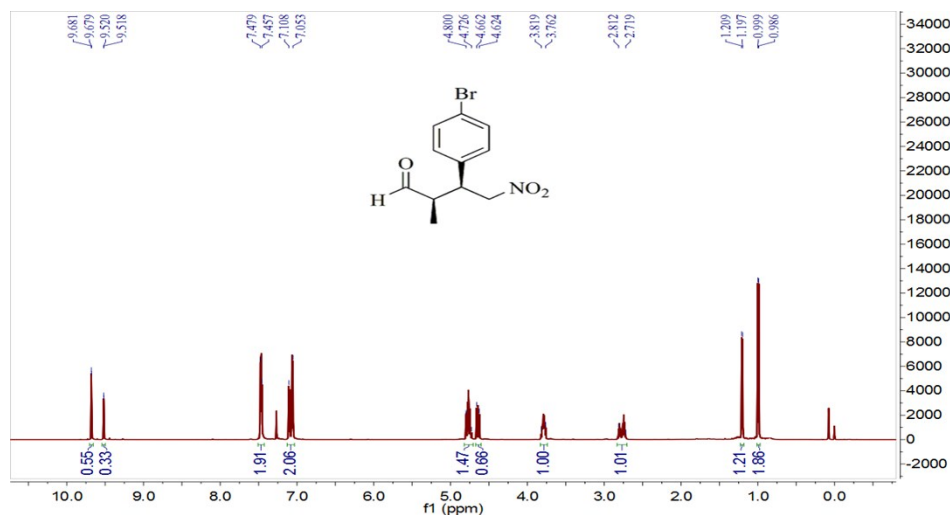


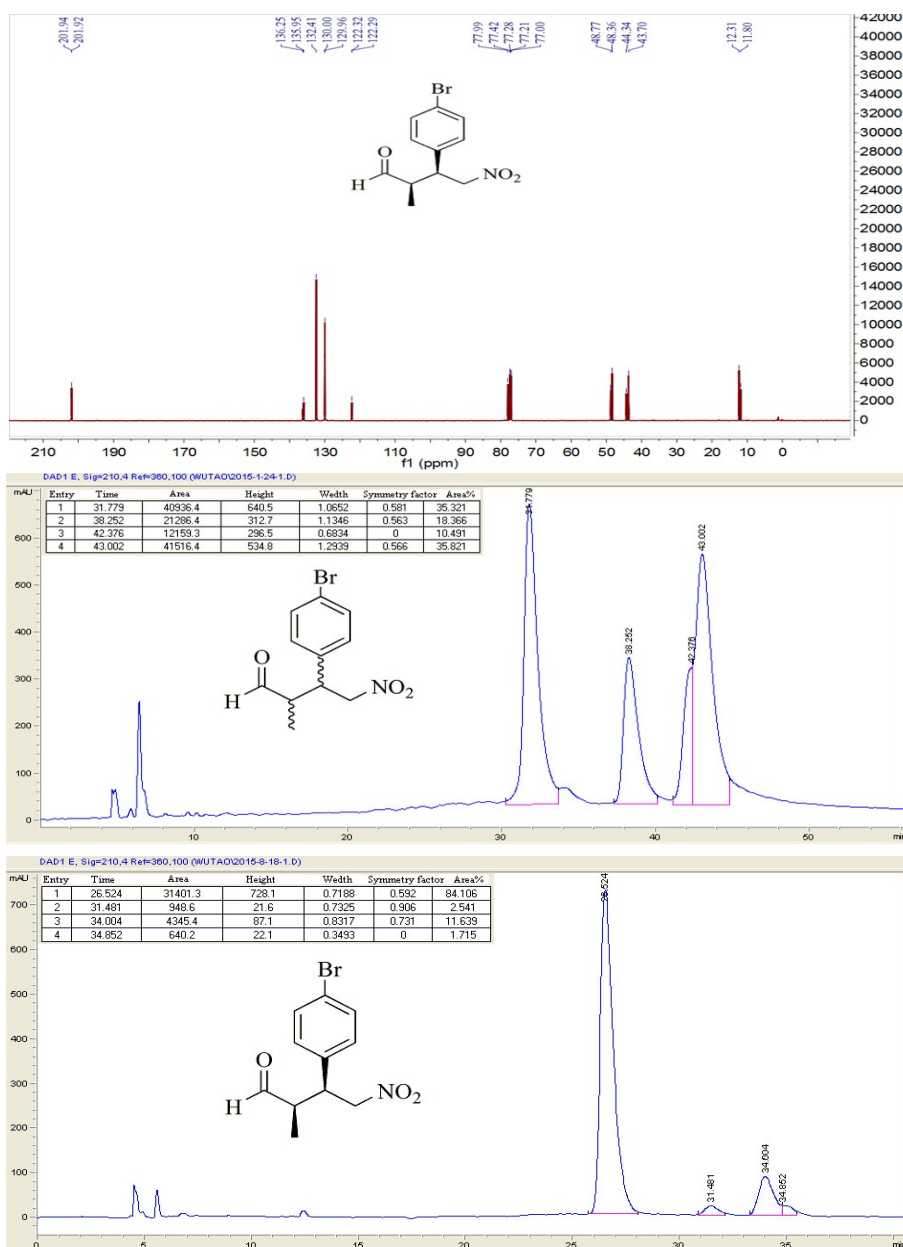
(2R, 3S)-3-(4-bromophenyl)-2-methyl-4-nitrobutanal: 86:14 dr, 96.0% ee, HPLC on Daicel Chiralpak AD-H



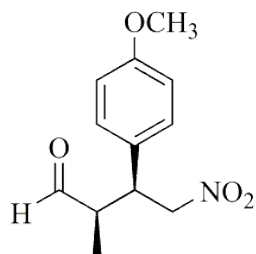
column: heptane/*i*-PrOH = 19/1, flow rate 0.9 mL min⁻¹, λ = 210 nm: t_R = 26.5 min (major), t_R = 34.9 min (minor); ¹H NMR (600 MHz, CDCl₃, TMS, major diastereomer): δ 0.99 (d, J = 7.3 Hz, 3H, CH₃), 2.72–2.81 (m, 1H, CHCH₃), 3.76–3.81 (m, 1H, CHCH₂), 4.73–4.80 (m, 2H, CH₂), 7.05–7.11 (m, 2H, Ar-H), 7.46–7.48 (m, 2H, Ar-H), 9.68 (d, J = 1.3 Hz, 1H, CHO); ¹³C NMR (150 MHz, CDCl₃, major diastereomer): δ 12.3 (CH₃), 43.7 (CHCH₂),

48.4 (CHCH₃), 78.0 (CH₂), 122.3, 129.9, 130.0, 132.4, 135.9 (Ph), 201.9 (C=O).

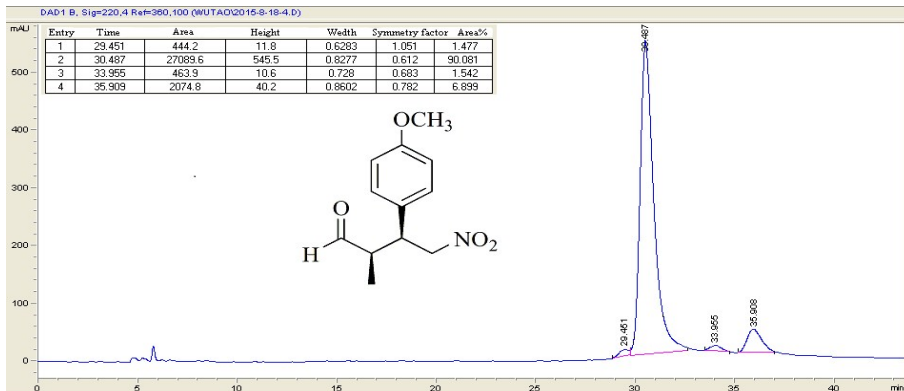
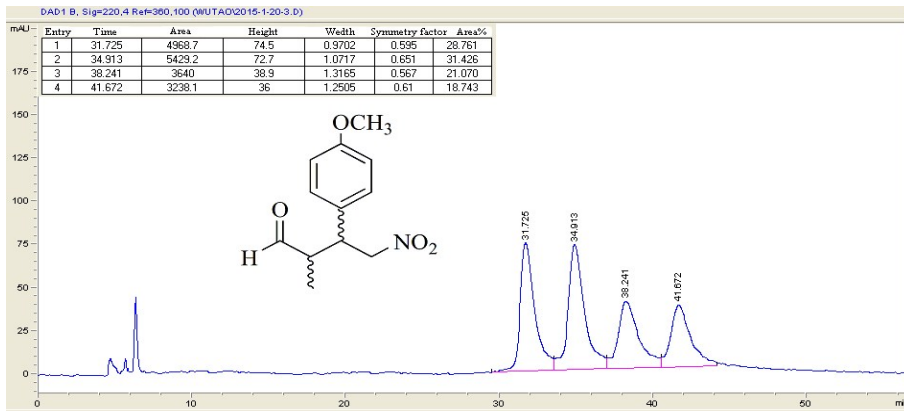
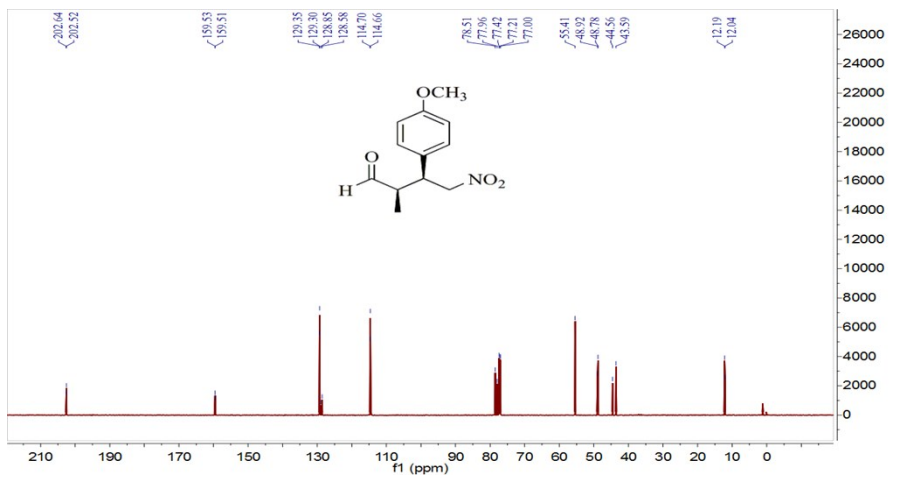
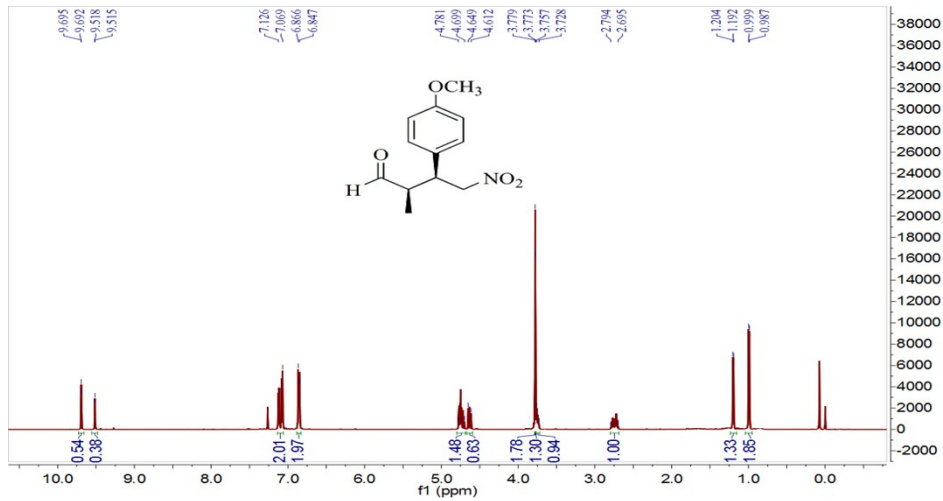




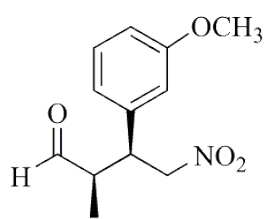
(2R,3S)-3-(4-methoxyphenyl)-2-methyl-4-nitrobutanal: 92:8 dr, 96.7% ee, HPLC on Daicel Chiralpak OD-H



column: heptane/*i*-PrOH = 8/2, flow rate 1.0 mL min⁻¹, λ = 220 nm: t_R = 29.5 min (minor), t_R = 30.5 min (major); ¹H NMR (600 MHz, CDCl₃, TMS, major diastereomer): δ 0.98 (d, *J* = 7.2 Hz, 3H, CHCH₃), 2.70–2.79 (m, 1H, CHCH₃), 3.73–3.76 (m, 1H, CHCH₂), 3.78 (s, 3H, OCH₃), 4.70–4.78 (m, 2H, CH₂), 6.85–6.87 (m, 2H, Ar-*H*), 7.07–7.13 (m, 2H, Ar-*H*), 9.69 (d, *J* = 1.6 Hz, 1H, CHO); ¹³C NMR (150 MHz, CDCl₃, major diastereomer): δ 12.2 (CHCH₃), 43.6 (CHCH₂), 48.8 (CHCH₃), 55.4 (OCH₃), 78.5 (CH₂), 114.6, 114.7, 128.6, 129.3, 159.5 (Ph), 202.5 (C=O).



(2R,3S)-3-(3-methoxyphenyl)-2-methyl-4-nitrobutanal: 89:11 dr, 95.8% ee, HPLC on Daicel Chiralpak OJ-H



column: heptane/*i*-PrOH = 8/2, flow rate 1.0 mL min⁻¹, λ = 220 nm: t_R = 55.2 min (minor),

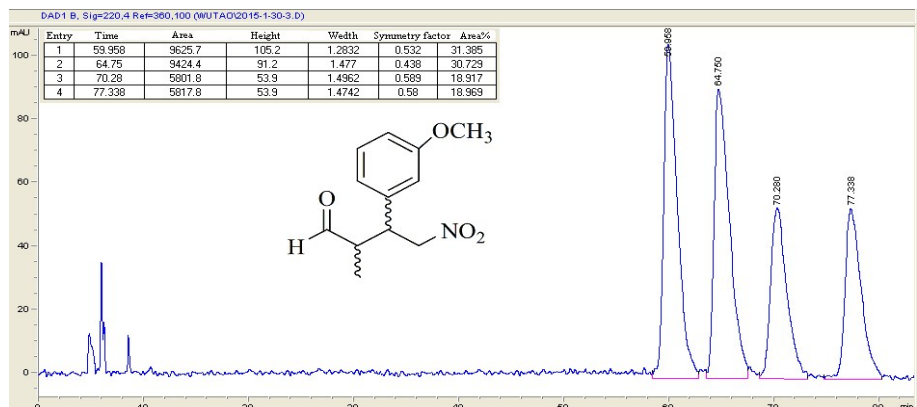
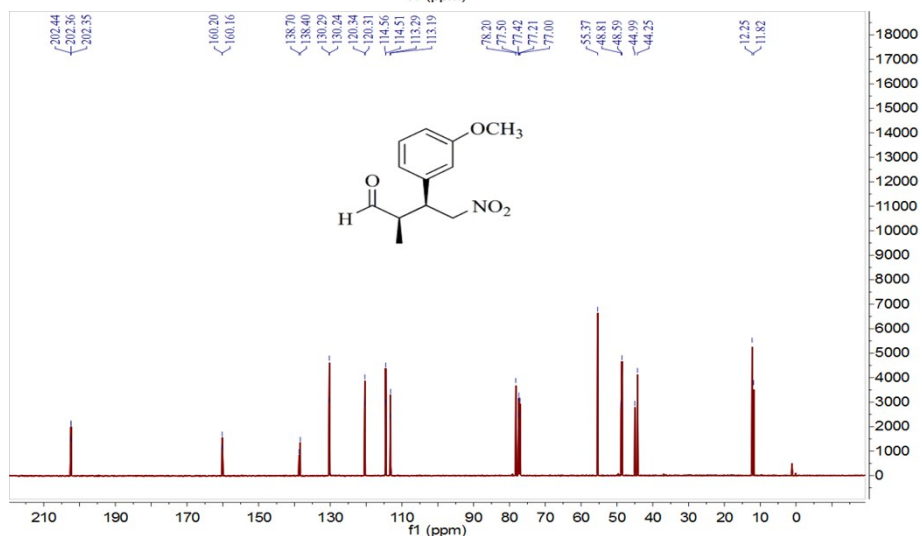
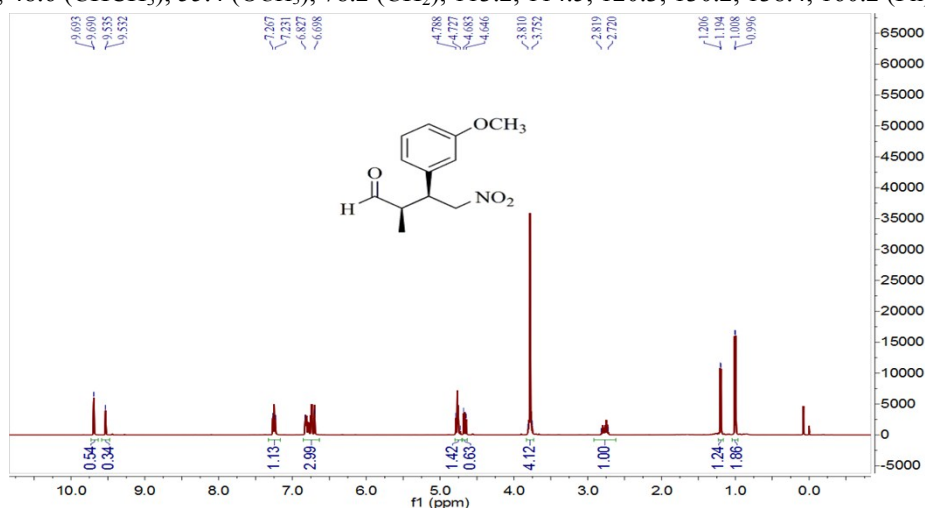
t_R = 58.3 min (major); ¹H NMR (600 MHz, CDCl₃, TMS, major diastereomer): δ 1.00 (d,

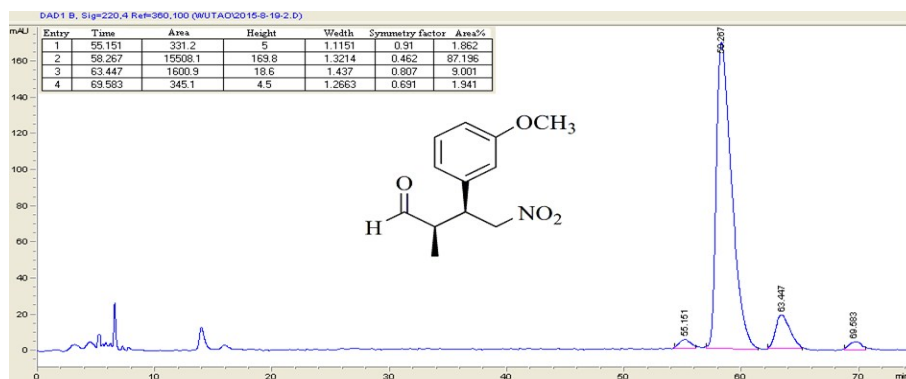
J = 7.3 Hz, 3H, CHCH₃), 2.72–2.82 (m, 1H, CHCH₃), 3.75–3.81 (m, 4H, CHCH₂, OCH₃),

4.73–4.79 (m, 2H, CH₂), 7.70–7.83 (m, 3H, Ar-*H*), 7.23–7.27 (m, 1H, Ar-*H*), 9.69 (d, *J* =

1.6 Hz, 1H, CHO); ¹³C NMR (150 MHz, CDCl₃, major diastereomer): δ 12.3 (CHCH₃),

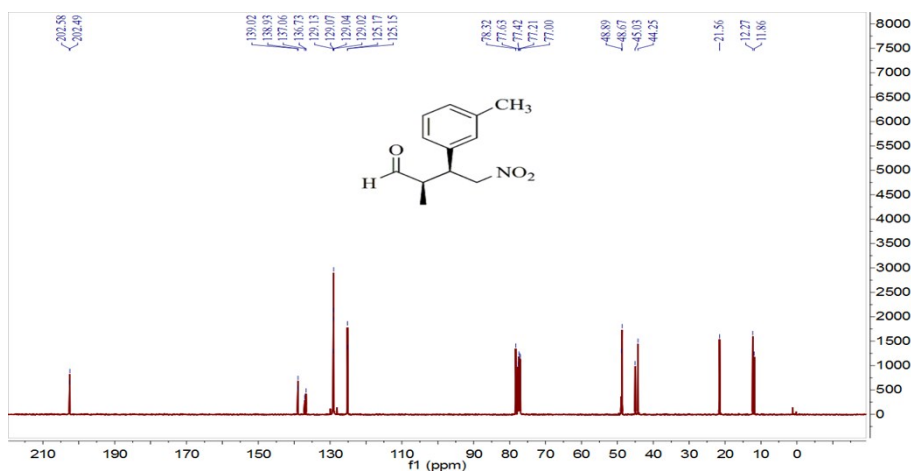
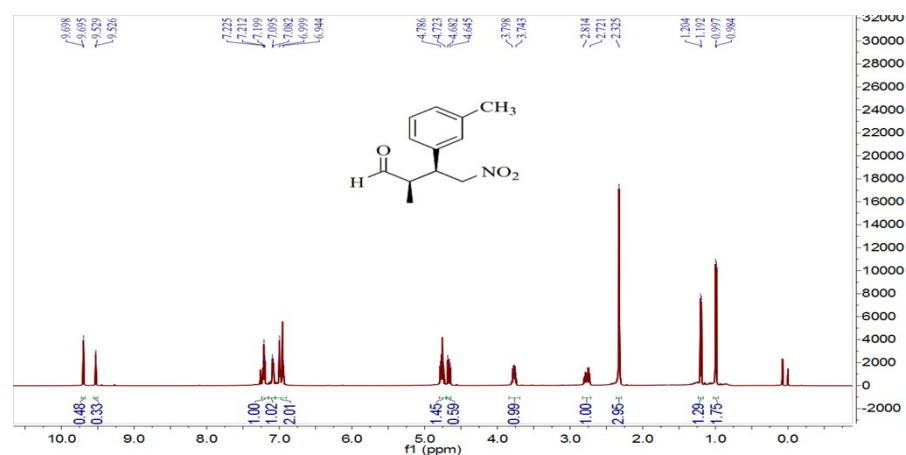
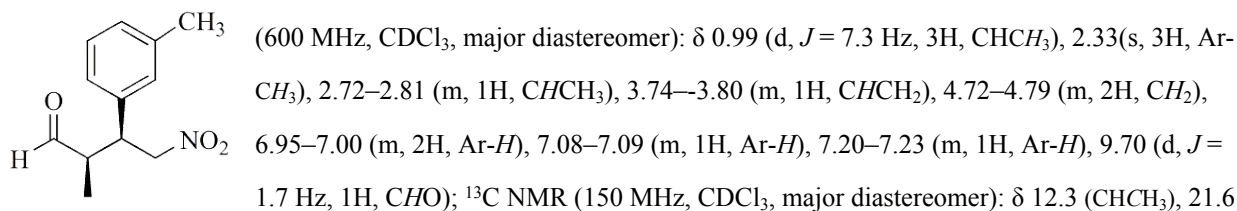
44.3 (CHCH₂), 48.6 (CHCH₃), 55.4 (OCH₃), 78.2 (CH₂), 113.2, 114.5, 120.3, 130.2, 138.4, 160.2 (Ph), 202.3 (C=O).

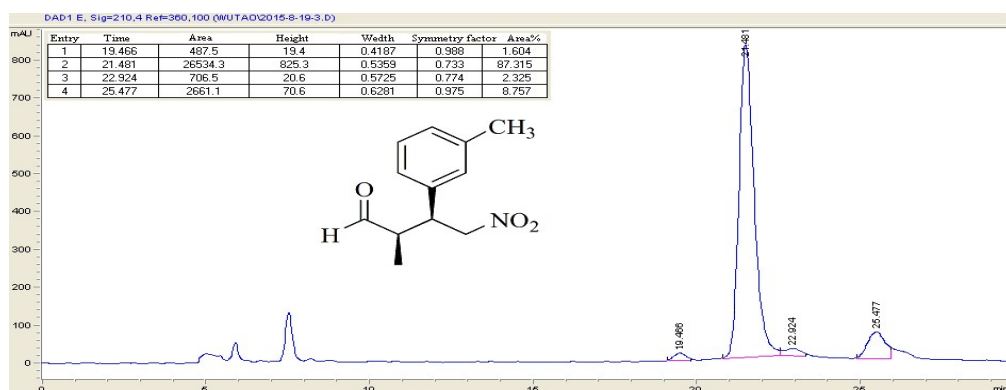
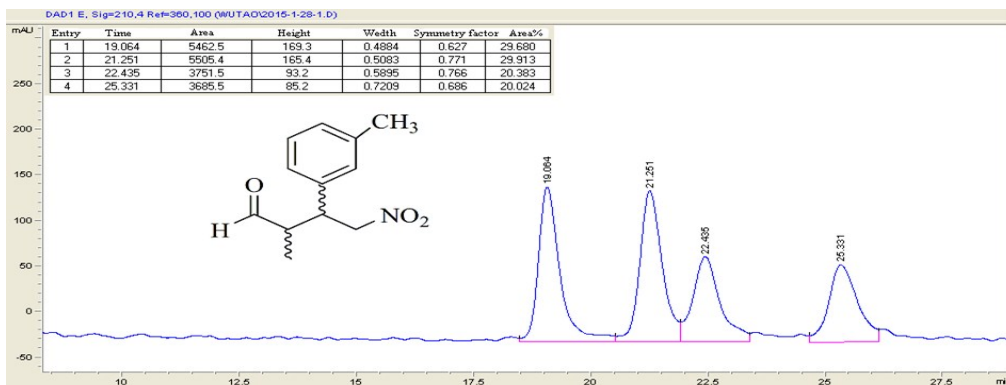




(2R,3S)-2-methyl-4-nitro-3-m-tolylbutanal: 89:11 dr, 96.4% ee, HPLC on Daicel Chiralpak OD-H column:

heptane/*i*-PrOH = 8/2, flow rate 1.0 mL min⁻¹, λ = 210 nm: t_R = 19.5 min (minor), t_R = 21.5 min (major); ¹H NMR





(2R,3S)-3-(2-chlorophenyl)-2-methyl-4-nitrobutanal: 88:12 dr, 96.1% ee, HPLC on Daicel Chiralpak AD-H

column: heptane/*i*-PrOH 19/1, flow rate 1.0 mL min⁻¹, λ = 210 nm: t_R = 20.4 min (major),

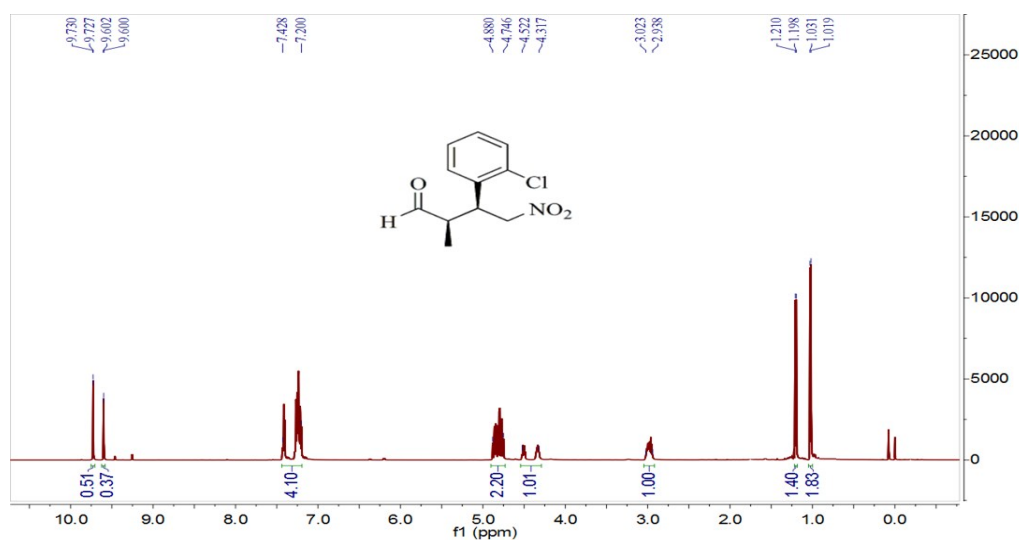
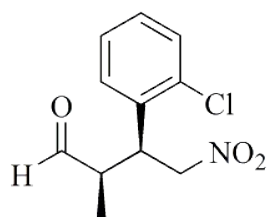
t_R = 23.9 min (minor); ¹H NMR (600 MHz, CDCl₃, TMS, major diastereomer): δ 1.03 (d,

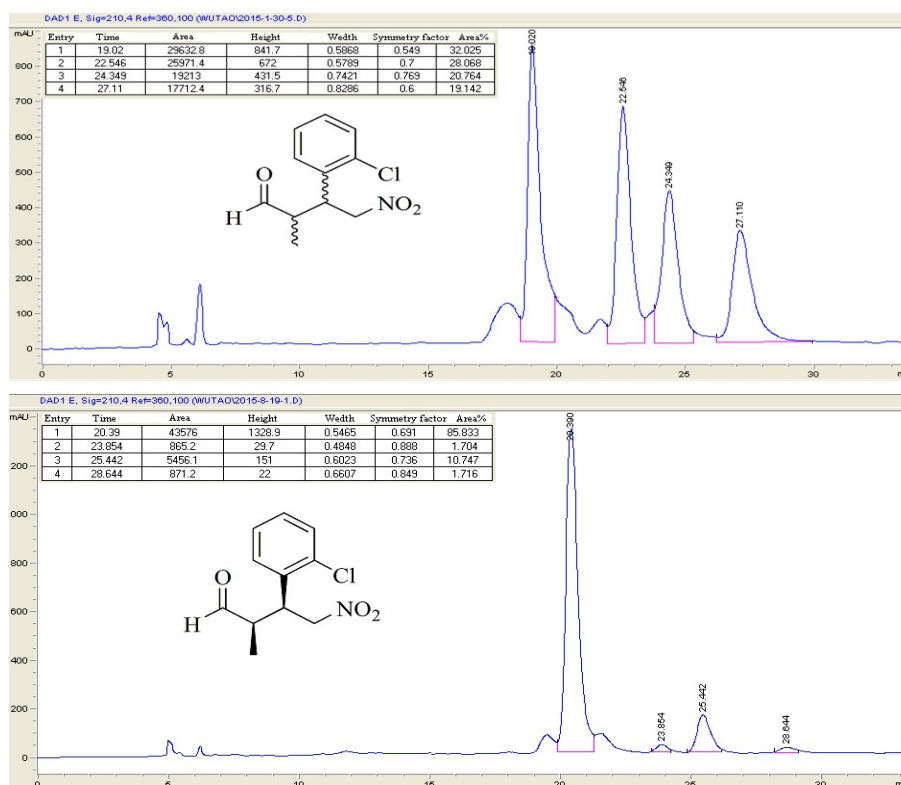
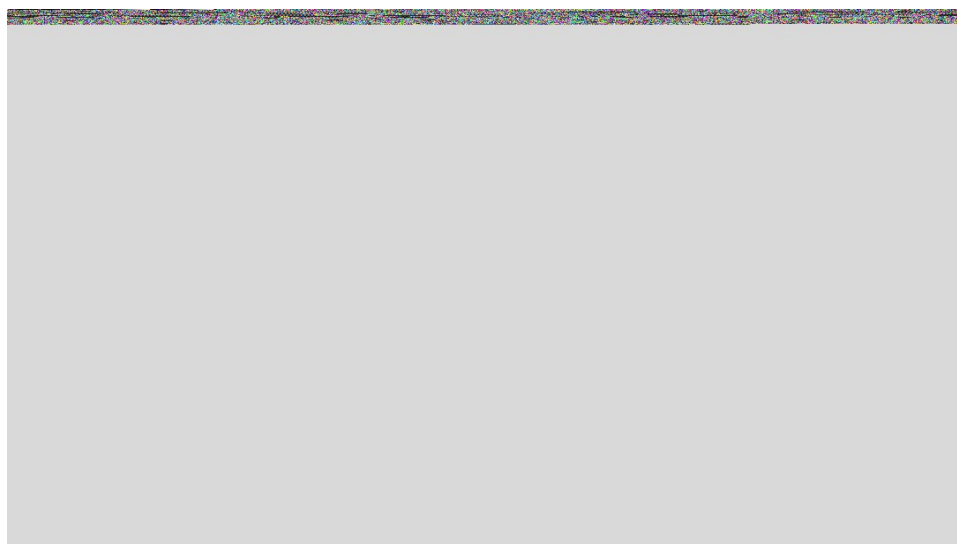
J = 7.4 Hz, 3H, CH₃), 2.94–3.02 (m, 1H, CHCH₃), 4.32–4.52 (m, 1H, CHCH₂), 4.75–4.88

(m, 2H, CH₂), 7.20–7.43 (m, 4H, Ar-H), 9.73 (d, J = 1.4 Hz, 1H, CHO); ¹³C NMR (150

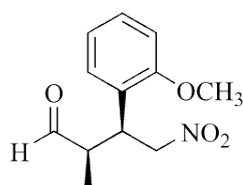
MHz, CDCl₃, major diastereomer): δ 12.4 (CH₃), 40.1 (CHCH₂), 48.0 (CHCH₃), 76.9

(CH₂), 127.6, 128.5, 129.4, 130.7, 134.3, 134.8 (Ph), 202.0 (C=O).





(2R, 3S)-3-(2-methoxyphenyl)-2-methyl-4-nitrobutanal: 92:8 dr, 95.7% ee, HPLC on Daicel Chiralpak AS-H

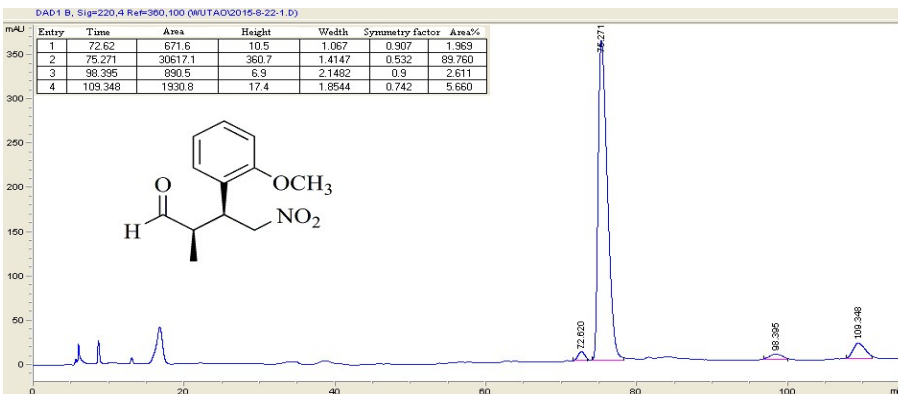
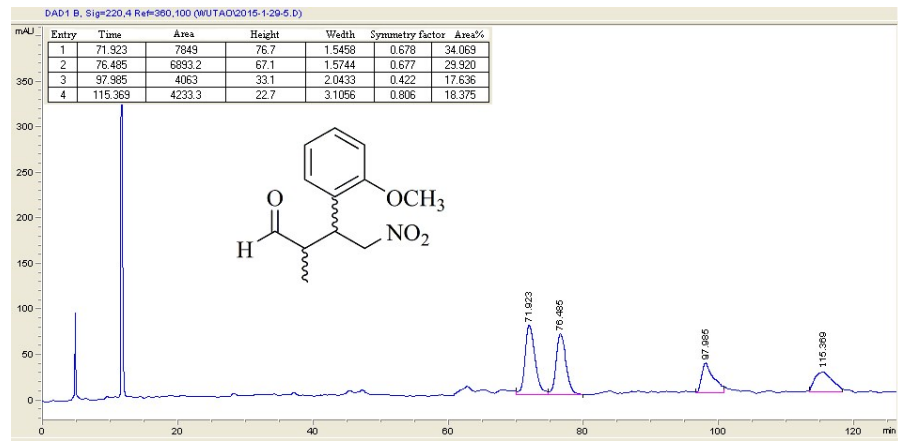
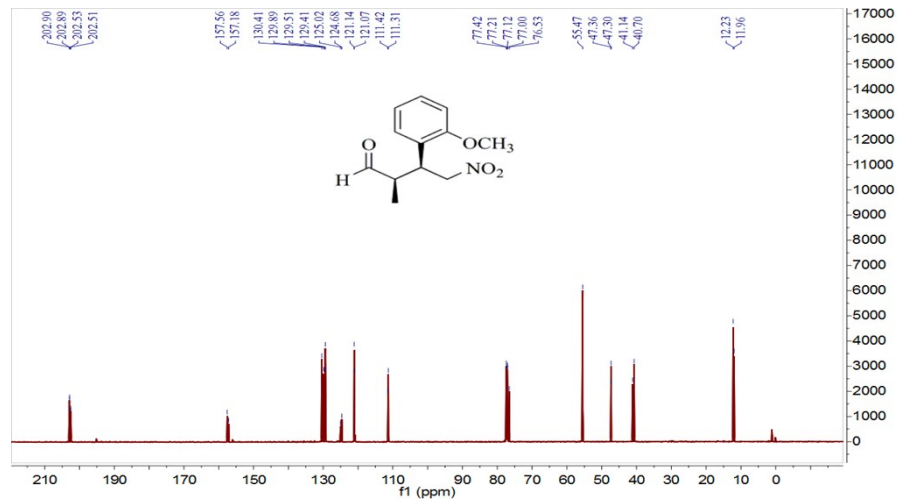
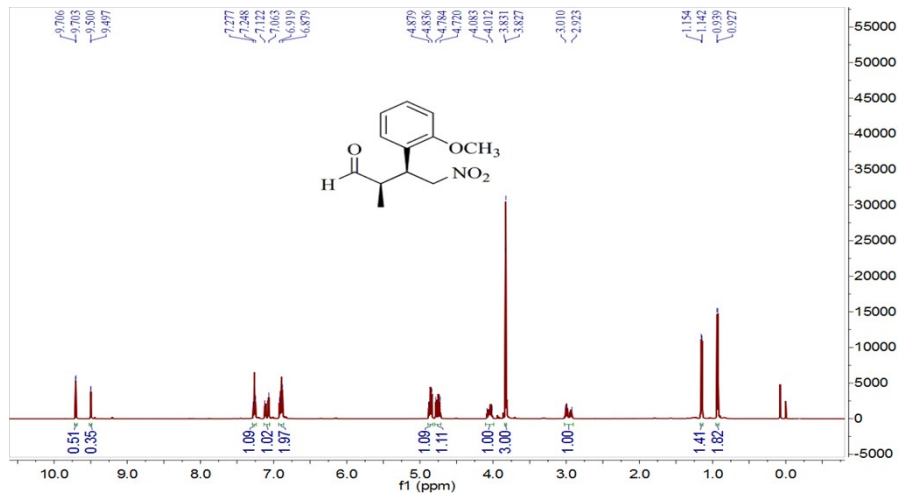


column: heptane/*i*-PrOH = 98/2, flow rate 0.9 mL min⁻¹, λ = 220 nm: t_R = 72.6 min (minor),

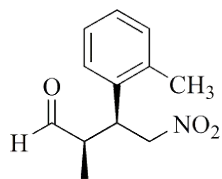
t_R = 75.3 min (major); ¹H NMR (600 MHz, CDCl₃, TMS, major diastereomer): δ 0.93 (d, J = 7.3 Hz, 3H, CHCH₃), 2.92–3.01 (m, 1H, CHCH₃), 3.83 (s, 3H, OCH₃), 4.02–4.07 (m, 1H, CHCH₂), 4.72–4.88 (m, 2H, CH₂), 6.88–6.92 (m, 2H, Ar-H), 7.06–7.12 (m, 1H, Ar-H),

7.25–7.28 (m, 1H, Ar-H), 9.70 (d, J = 1.8 Hz, 1H, CHO); ¹³C NMR (150 MHz, CDCl₃,

major diastereomer): δ 12.2 (CHCH₃), 40.7 (CHCH₂), 47.3 (CHCH₃), 55.5 (OCH₃), 77.1 (CH₂), 111.3, 121.1, 124.7, 129.4, 130.4, 157.5 (Ph), 202.9 (C=O).



(2R, 3S)-2-methyl-4-nitro-3-*o*-tolylbutanal: 90:10 dr, 96.6% ee, HPLC on Daicel Chiralpak OD-H column:



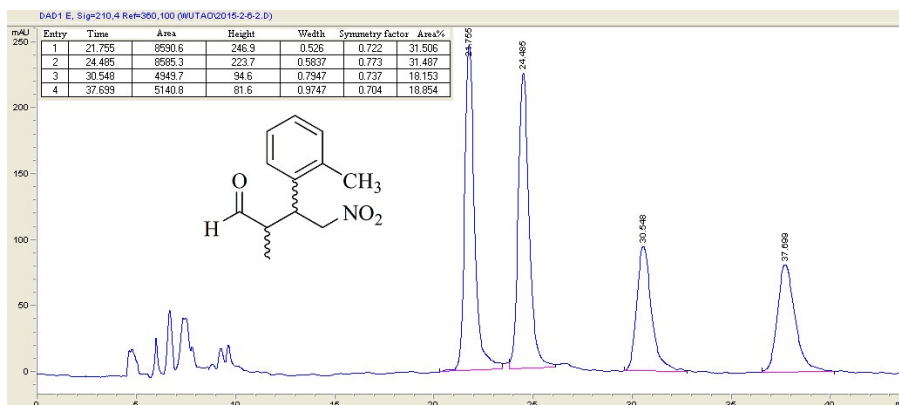
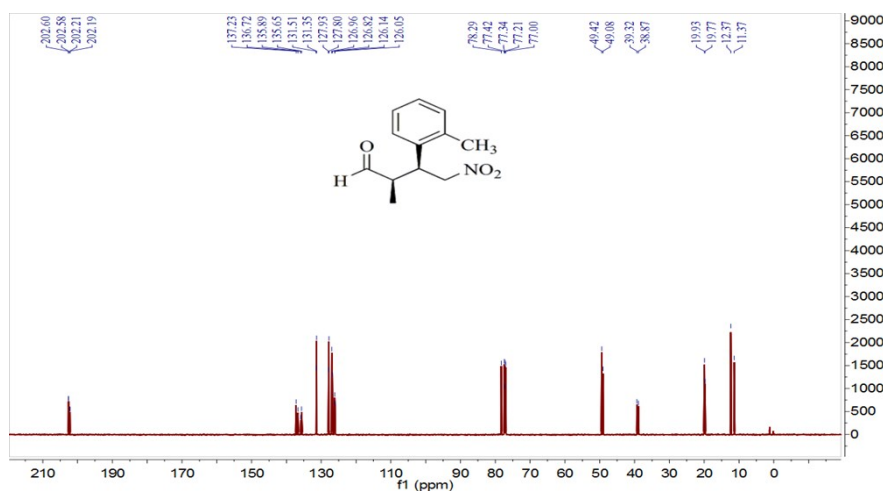
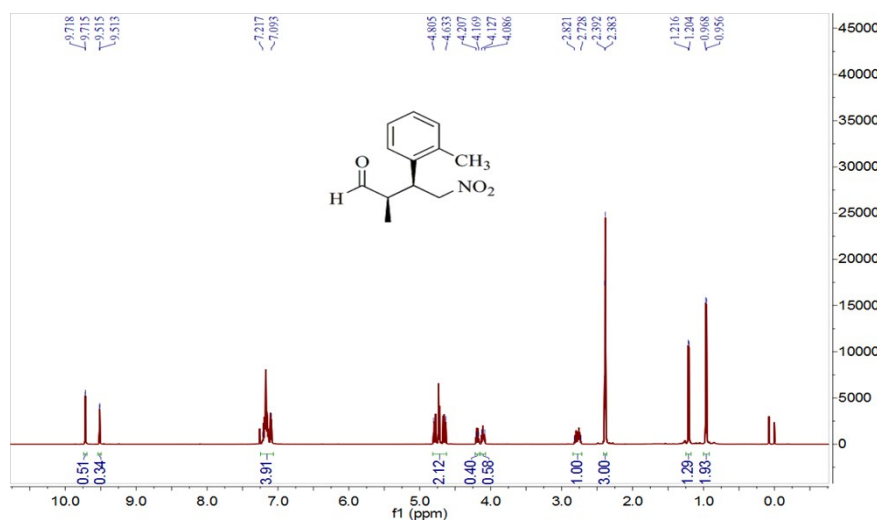
heptane/*i*-PrOH = 8/2, flow rate 1.0 mL min⁻¹, λ = 210 nm: t_R = 21.5 min (minor), t_R = 25.2 min (major); ¹H NMR (600 MHz, CDCl₃, TMS, major diastereomer): δ 0.96 (d, *J* = 7.3 Hz,

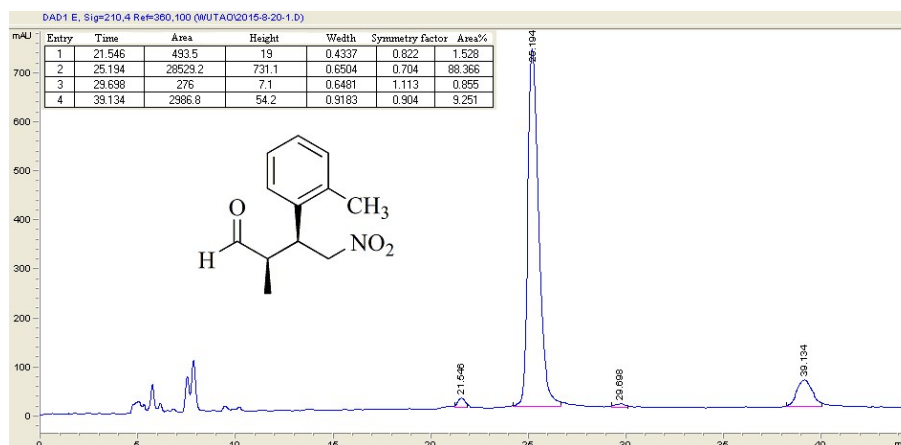
3H, CHCH₃), 2.39 (s, 3H, ArCH₃), 2.73–2.82 (m, 1H, CHCH₃), 4.09–4.13 (m, 1H,

CHCH₂NO₂), 4.63–4.81 (m, 2H, CH₂NO₂), 7.09–7.22 (m, 4H, ArH), 9.72 (d, *J* = 1.9 Hz, 1H,

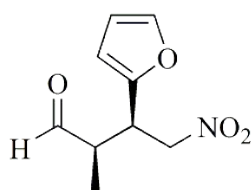
CHO); ¹³C NMR (150 MHz, CDCl₃, major diastereomer): δ 12.4 (CHCH₃), 19.9 (Ar-CH₃), 38.9 (CHCH₂), 49.4

(CHCH₃), 78.3 (CH₂), 126.1, 126.9, 127.8, 131.3, 133.6, 137.2 (Ph), 202.6 (C=O).





(2R, 3R)-3-(furan-2-yl)-2-methyl-4-nitrobutanal: 81:19 dr, 96.4% ee, HPLC on Daicel Chiralpak AS-H column:



heptane/i-PrOH = 9/1, flow rate 0.8 mL min⁻¹, λ = 214 nm: t_R = 38.2 min (minor), t_R = 41.5

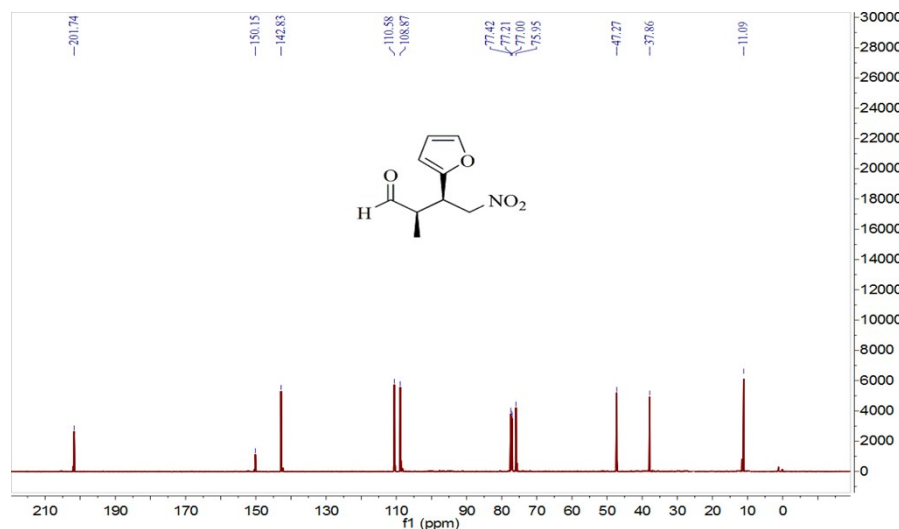
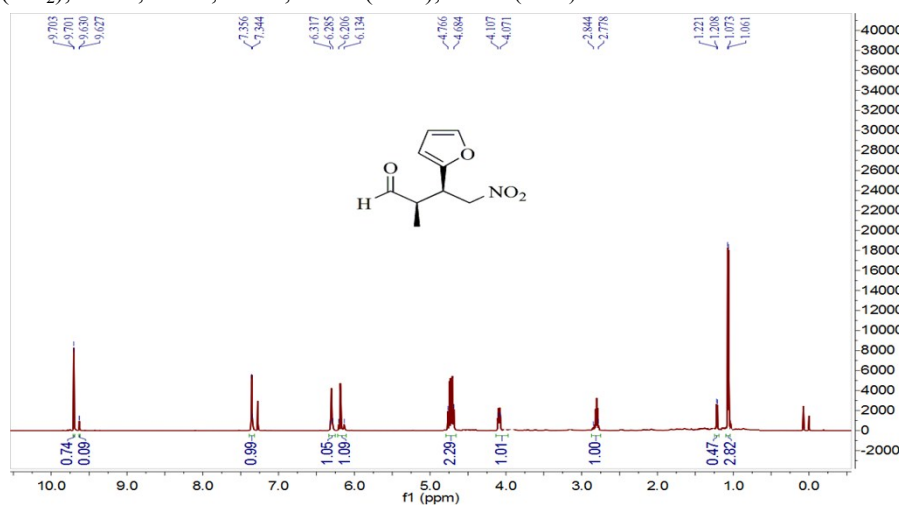
min (major); ¹H NMR (600 MHz, TMS, CDCl₃, major diastereomer): δ 1.07 (d, J = 7.3 Hz,

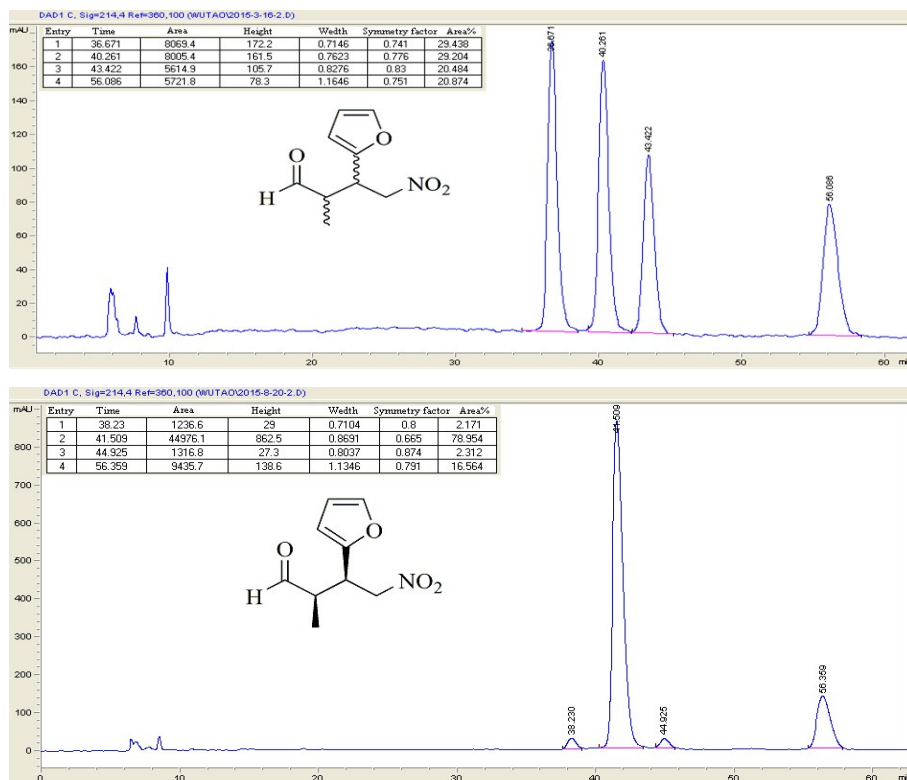
3H, CH₃), 2.78–2.84 (m, 1H, CHCH₃), 4.07–4.11 (m, 1H, CHCH₂), 4.68–4.77 (m, 2H,

CH₂), 6.13–6.21 (m, 1H, =CH), 6.29–6.32 (m, 1H, =CH), 7.34–7.36 (m, 1H, =CH), 9.70 (d,

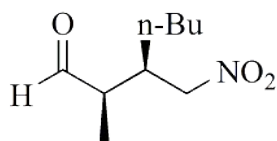
J = 0.8 Hz, 1H, CHO); ¹³C NMR (150 MHz, CDCl₃, major diastereomer): δ 11.1 (CH₃), 37.9 (CHCH₂), 47.3

(CHCH₃), 76.0 (CH₂), 108.9, 110.6, 142.8, 150.2 (=CH), 201.7 (C=O).



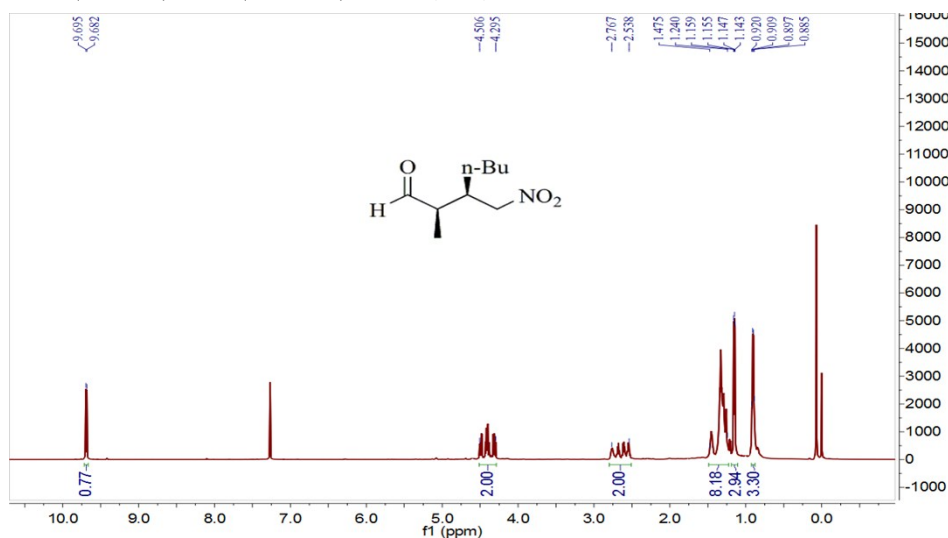


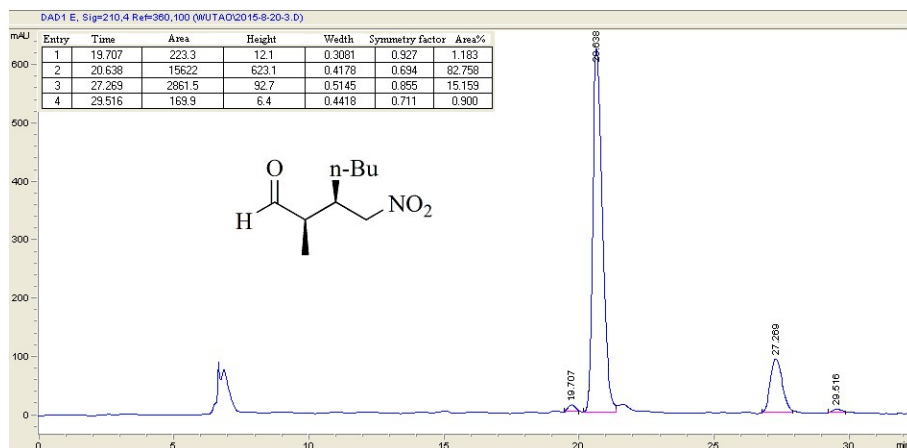
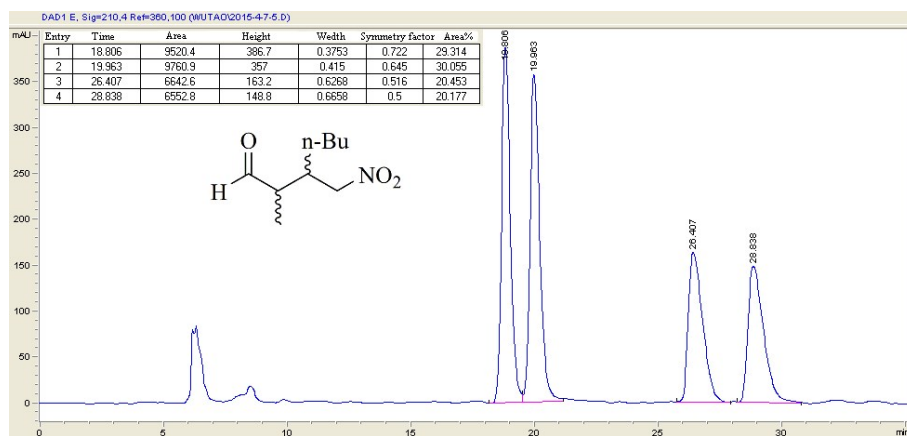
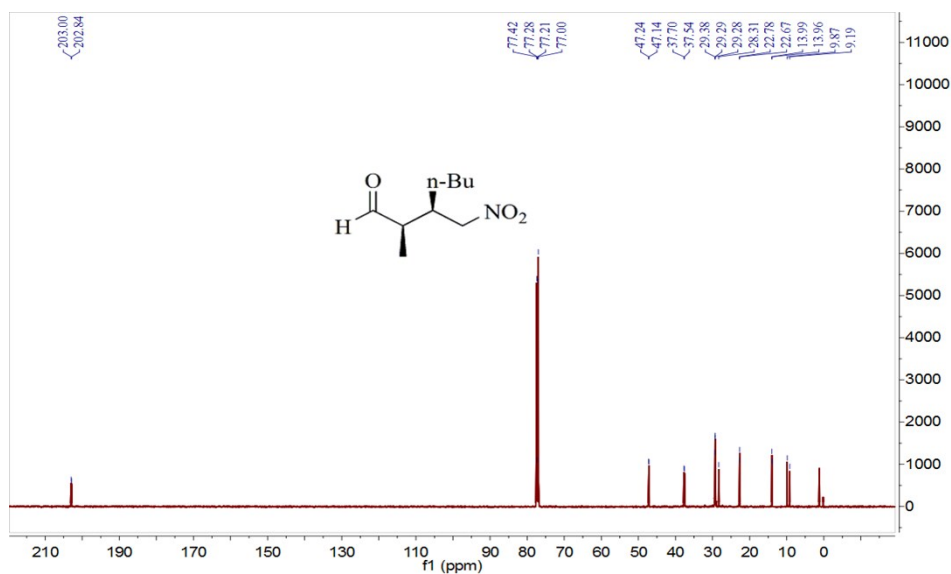
(2R, 3R)-3-n-butyl-2-methyl-4-nitrobutanal: 84:16 dr, 97.2% ee, HPLC on Daicel Chiralpak AS-H column:



heptane/*i*-PrOH = 19/1, flow rate 0.8 mL min⁻¹, λ = 210 nm: t_R = 19.7 min (minor), t_R = 20.6 min (major); ¹H NMR (600 MHz, CDCl₃, TMS, major diastereomer): δ 0.90 (dd, *J* = 7.0, 13.9 Hz, 3H, CHCH₃), 1.15 (dd, *J* = 2.8, 7.3 Hz, 3H, CH₂CH₃), 1.24–1.48 (m, 6H, CH₂CH₂CH₂), 2.54–2.77 (m, 2H, CHCH₃, CHCH₂NO₂), 4.29–4.51 (m, 2H, CH₂NO₂), 9.69 (d, *J* = 7.7 Hz, 1H, CHO);

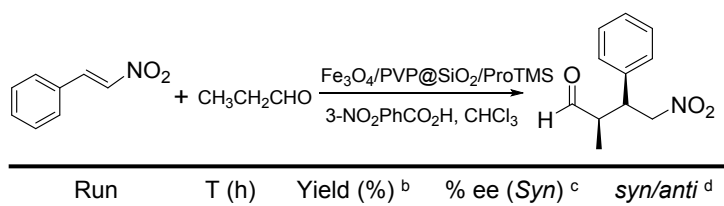
¹³C NMR (150 MHz, CDCl₃, major diastereomer): δ 9.9 (CH₃), 14.0 (CH₂CH₃), 22.7, 28.3, 29.3 (CH₂CH₂CH₂), 37.7 (CHCH₂NO₂), 47.1 (CHCH₃), 77.3 (CH₂NO₂), 203.0 (C=O).





6. Reusability of Fe₃O₄/PVP@ SiO₂/ProTMS

Table s3. Recycling experiments of Fe₃O₄/PVP@SiO₂/ProTMS in Michael addition reaction ^a



1	8	>99	98	96/4
2	12	>99	98	96/4
3	18	96	98	96/4
4	18	95	98	96/4
5	24	93	98	96/4
6	24	89	98	96/4
7	24	88	98	96/4
8	48	86	98	96/4
9	48	81	97	96/4
10	48	77	98	96/4

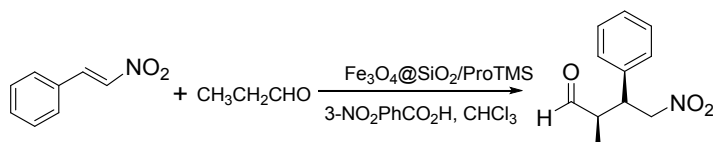
^a Reaction conditions: *trans*- β -Nitrostyrene (40 mg, 0.27 mmol), propionaldehyde (94 mg, 1.62 mmol), Fe₃O₄/PVP@ProTMS (30 mg, 7 mol %), 3-NO₂PhCO₂H (10 mol %), CHCl₃ (2.0 mL), 0 °C.

^b Isolated yield.

^c Determined by ¹H NMR.

^d Determined by chiral HPLC of crude.

Table s4. Recycling experiments of Fe₃O₄@SiO₂/ProTMS in Michael addition reaction ^a



Run	T (h)	Yield (%) ^b	%ee (<i>Syn</i>) ^c	<i>syn/anti</i> ^d
1	15	93	98	93/7
2	24	89	98	92/8
3	35	85	98	91/9
4	48	81	98	90/10
5	48	70	98	90/10
6	48	56	98	89/11

^a Reaction conditions: *trans*- β -Nitrostyrene (40 mg, 0.27 mmol), propionaldehyde (94 mg, 1.62 mmol), Fe₃O₄@ProTMS (26 mg, 7 mol %), 3-NO₂PhCO₂H (10 mol %), CHCl₃ (2.0 mL), 0 °C.

^b Isolated yield.

^c Determined by ¹H NMR.

^d Determined by chiral HPLC of crude.

7. TEM image of $\text{Fe}_3\text{O}_4/\text{PVP}@/\text{SiO}_2/\text{ProTMS}$ and 10th-recycled $\text{Fe}_3\text{O}_4/\text{PVP}@/\text{SiO}_2/\text{ProTMS}$

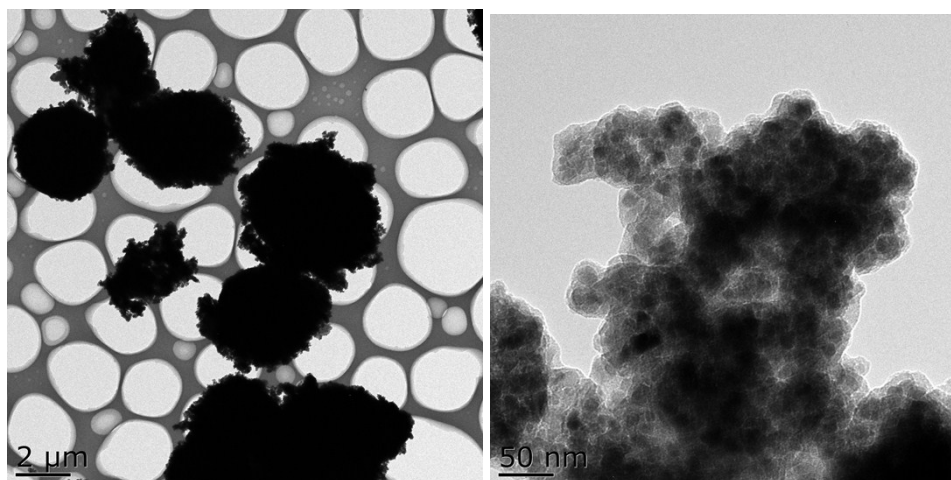


Fig. s9. TEM image of fresh $\text{Fe}_3\text{O}_4/\text{PVP}@/\text{SiO}_2/\text{ProTMS}$

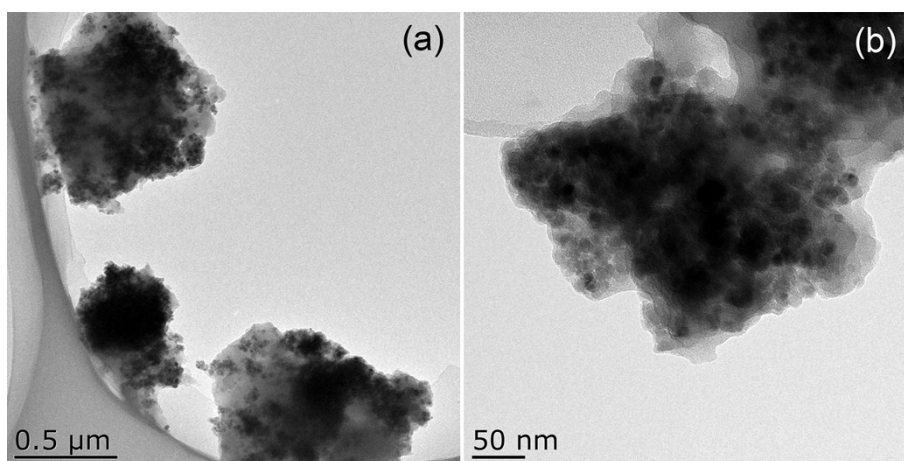


Fig. s10. TEM image of 10th-recycled $\text{Fe}_3\text{O}_4/\text{PVP}@/\text{SiO}_2/\text{ProTMS}$