Electronic Supplementary Information

Adsorption of berberine hydrochloride onto mesoporous carbons with tunable pore size

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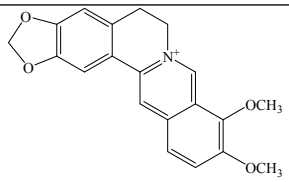
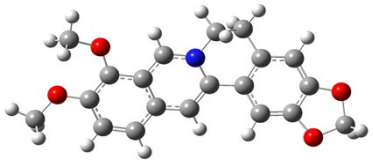
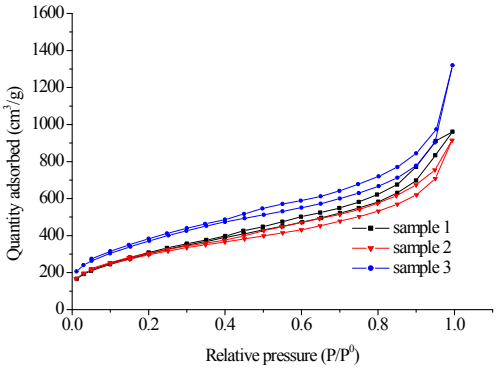
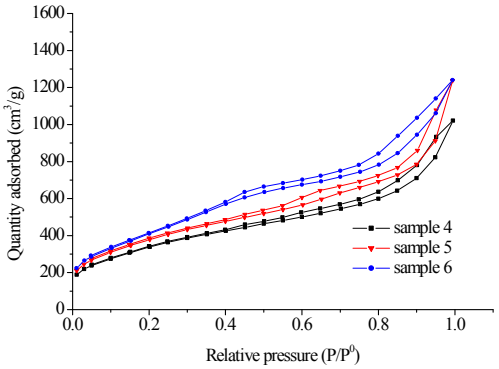
Table S1 The molecular structure, optimized molecular structure model and molecular size of berberine hydrochloride

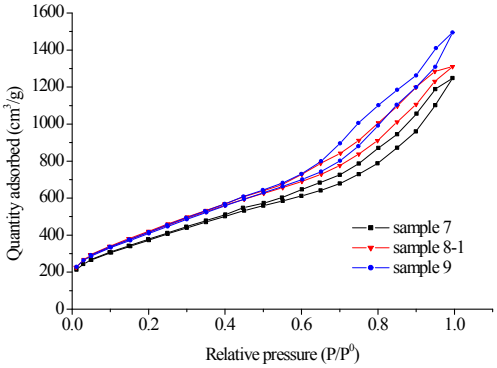
Substance	Molecular Structure	Optimized molecular structure model	Molecular size
Berberine hydrochloride			1.47 nm × 0.66 nm × 0.32 nm



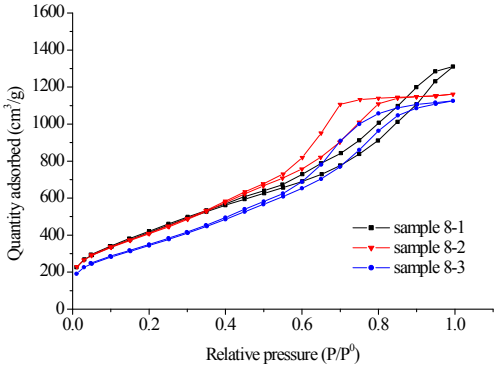
(a)



(b)



(c)



(d)

Fig. S1 Nitrogen adsorption-desorption isotherms of the mesoporous carbons at 77K: (a) sample 1-3 with the molar composition of P123/phenol at 0.021; (b) sample 4-6 with the molar composition of P123/phenol at 0.033; (c) sample 7, 8-1 and 9 with the molar composition of P123/phenol at 0.044; (d) sample 8-1, 8-2 and 8-3 with the carbonization temperature at 900, 800 and 700 ^{o}C, respectively.

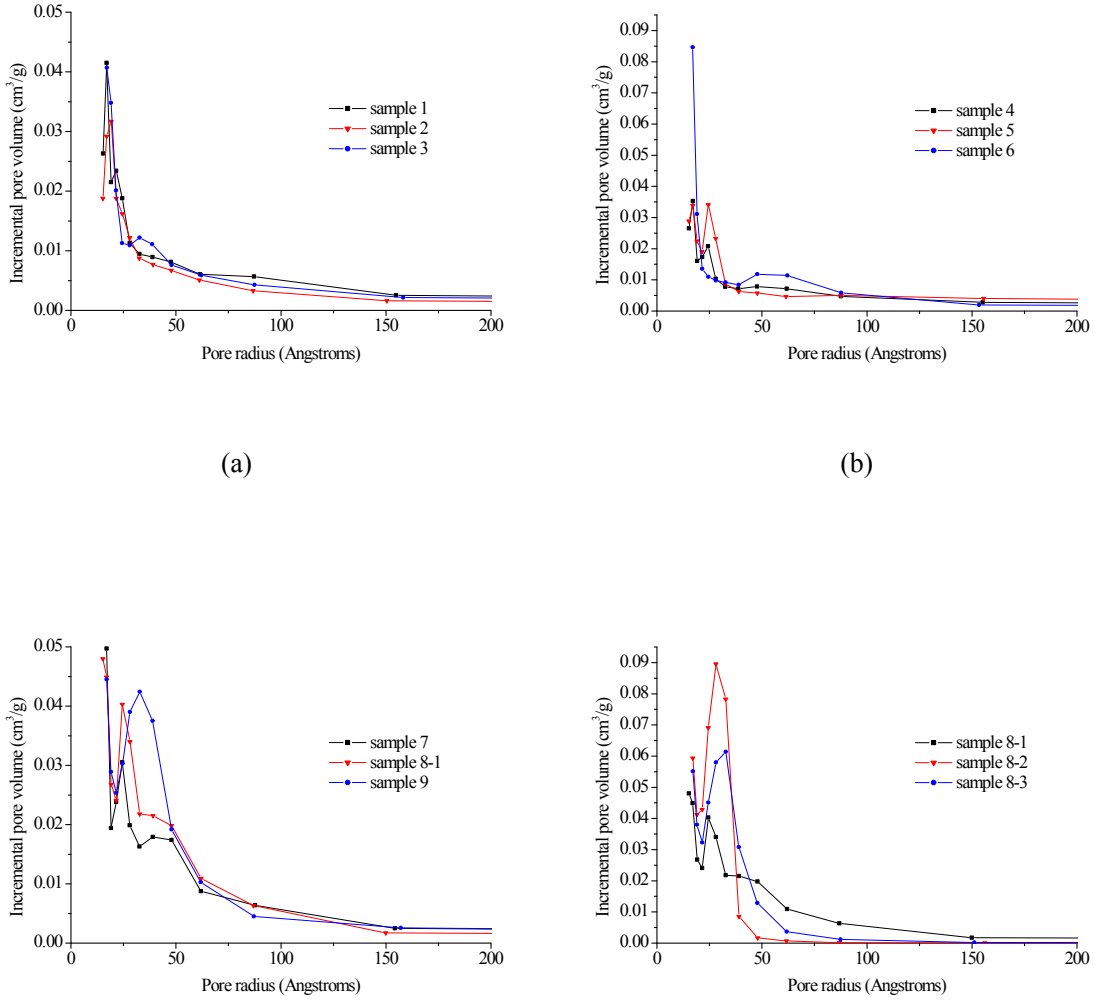


Fig. S2. pore diameter distributions of the mesoporous carbons: (a) sample 1-3 with the molar composition of P123/phenol at 0.021; (b) sample 4-6 with the molar composition of P123/phenol at 0.033; (c) sample 7, 8-1 and 9 with the molar composition of P123/phenol at 0.044; (d) sample 8-1, 8-2 and 8-3 with the carbonization temperature at 900, 800 and 700 ^{o}C, respectively.