Petal-like graphene-Ag composites with highly exposed active edge sites was designed

and constructed for electrochemical determination of metronidazole

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Supplementary materials:



Fig. S1 (a) Plot of Q-t curve of the GR-Ag/GCE and the p-GR-Ag/GCE in 1 mM K_3 [Fe(CN)₆] containing 0.1 M KCl; (b) Plot of Q-t^{1/2} curve of the GR-Ag/GCE and the p-GR-Ag/GCE.



Fig. S2 (a) The histograms of the cathodic peak current of 300 μ M metronidazole at different electrodes; (b) The histograms of the cathodic peak potential of 300 μ M metronidazole at different electrodes.



Fig. S3 Plots of peak current versus volume of the dropped suspension.



Fig. S4 (a) Cyclic voltammograms of 300 μ M metronidazole at the p-GR-Ag/GCE in PBS (pH 6.0) buffer solution with different scan rates (10, 30, 50, 80, 110, 140, 170, 200, 230, 260, 290, 320, 350, 380, and 410 mV s⁻¹); **(b)** Plot of cathodic peak current *versus* square root of scan rates.



Fig. S5 The LSVs of 200 μ M metronidazole in PBS (pH 6.0) buffer solution at five different p-GR-Ag/GCE.



Fig. S6 The LSV responses of 50 μ M metronidazole in PBS (pH 6.0) buffer solution at the p-GR-Ag/GCE by successive measurements.



Fig. S7 The variation in the response current of 100 μ M metronidazolein PBS (pH 6.0) buffer solution at the p-GR-Ag/GCE for 10 days.

Table S1

Interferents	Response with respect to metronidazole (%)
Li ₂ SO ₄	2.18
LiClO ₄	0.34
CaCl ₂	3.62
Na ₂ HPO ₄	2.49
KH ₂ PO ₄	2.64
glucose	1.09
fructose	4.56
maltose	0.62
phenacetin	4.37
uric acid	3.92
ascorbic acid	3.96

The response of the p-GR-Ag/GCE to the interferents.