

## Supporting Information

### **Sporulaminals A and B: a Pair of Unusual Epimeric Spiroaminal Derivatives from a Marine-Derived Fungus *Paraconiothyrium sporulosum* YK-03**

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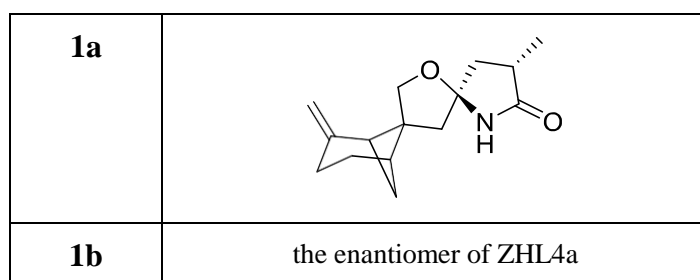
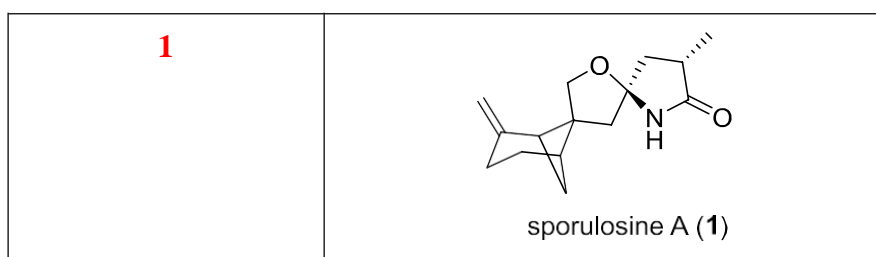
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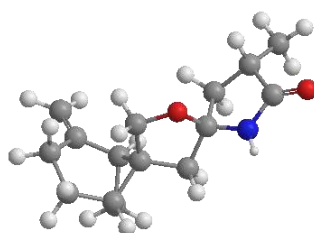
## Computational methods

Relative configurations of **1** and **2** were deduced from their NOESY data.

Compound **1**: Stochastic conformational searches were firstly conducted under MMFF94 force field for **1a** (2*S*, 4*R*, 6*S*, 7*S*, 9*S*), which gave two conformers. Their difference was that the orientation of methyl of the lactam ring was different. Conformers were optimized at the B3LYP/6-31G(d) basis set level in methanol and the frequencies were calculated by further time-dependent density functional theory (TDDFT) method, showing that these two conformers combined into one stable conformer, which was identical to the conformer of X-ray crystallographic analysis of compound **1**.

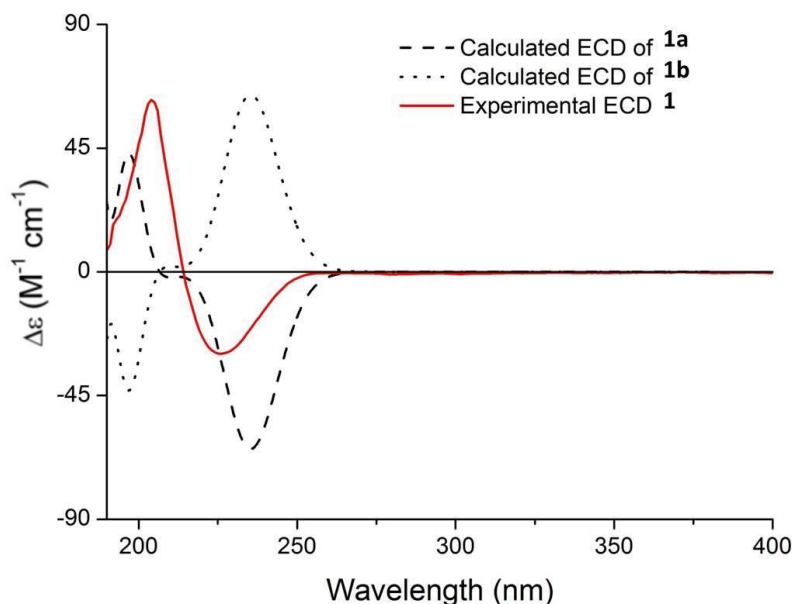


Stable conformer of **1a** in MeOH:



Using the conformers at the B3LYP/6-31G(d) basis set level in methanol, forty excitation states at the B3LYP/6-31G(d) basis set level were calculated, peak stretcher

was 0.4eV, and finally the calculation results were Boltzmann averaged to yield the depicted electronic circular dichroism (ECD) spectra of **1**, which was identical to the calculation results at the cam-B3LYP/6-31G(d) basis set level. All calculations were performed by Gaussian 09 program package (Version C.01).



**Figure S1.** Calculated and experimental ECD spectra of **1**

Compound **2**: The Spartan 14.0 (Wavefunction Inc., Irvine, CA, USA) search using molecular mechanics MMFF was performed for **2a** (2S, 4S, 6S, 7S, 9S), which gave 100 conformers. The low-energy conformers of **2a** accounting for more than 5% Boltzmann distribution were further optimized successively in the gas phase by semi-empirical method in Gaussian 09 program package, which were reoptimized and analysed frequency, orderly, using the density functional theory (DFT) at the B3LYP/6-31G (d, p) level and the same way in the methanol, resulted in no imaginary frequencies. Solvent effects were taken into consideration by using the conductor polarizable continuum model (CPCM). The conformers of **2a** were calculated electronic circular dichroism (ECD) by the time-dependent density functional theory (TD-DFT) method at the B3LYP/6-31++G (d, p) level with the CPCM model in methanol solution. The overall calculated ECD curves of **2a** were generated by Boltzmann weighting of their selected low-energy conformers using SpecDis 1.51 with  $\sigma = 0.12\text{eV}$  at -8nm shift.

**Table S1** Energy analysis of **2a**

Label	MMFF		
	E(kJ/mol)	rel. E(kJ/mol)	Boltzmann Dist.
2a-1	-3.64	0.00	0.554
2a-2	-3.10	0.54	0.446

**Table S2** Computational methods for ECD of **2a****2a-1** Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	3.143606	0.845033	-0.593609
2	6	0	3.707414	-0.453570	-0.006880
3	6	0	2.519355	-1.429386	-0.107791
4	6	0	1.257225	-0.538567	-0.174019
5	8	0	0.762872	-0.243800	1.146697
6	6	0	0.057513	-1.134857	-0.929951
7	6	0	-1.179354	-0.630413	-0.167682
8	6	0	-0.626302	-0.589884	1.256840
9	6	0	4.213674	-0.208949	1.424057
10	8	0	3.774911	1.863295	-0.856479
11	7	0	1.797321	0.678435	-0.760576
12	6	0	-1.859827	0.647478	-0.805980
13	6	0	-2.734428	1.312146	0.228141
14	6	0	-3.875824	0.432442	0.720212
15	6	0	-3.560220	-1.085279	0.608303
16	6	0	-2.515529	-1.359407	-0.490538
17	6	0	-2.699397	-0.351880	-1.660416
18	6	0	-2.521355	2.555817	0.669831
19	1	0	4.547707	-0.773735	-0.631175
20	1	0	2.445190	-2.121968	0.733439
21	1	0	2.586965	-2.016359	-1.028736
22	1	0	0.104279	-2.228231	-0.858487
23	1	0	0.083475	-0.869985	-1.988894
24	1	0	-0.730035	-1.576943	1.733942
25	1	0	-1.107321	0.155316	1.895873

26	1	0	3.390022	0.084999	2.081852
27	1	0	4.966713	0.584213	1.434554
28	1	0	4.665631	-1.119197	1.829102
29	1	0	1.211617	1.477602	-0.966837
30	1	0	-1.216944	1.366308	-1.323570
31	1	0	-4.163319	0.703939	1.740336
32	1	0	-4.751058	0.647924	0.094038
33	1	0	-4.482312	-1.632475	0.383808
34	1	0	-3.195903	-1.476625	1.565247
35	1	0	-2.450380	-2.427085	-0.726831
36	1	0	-3.716593	-0.059078	-1.935641
37	1	0	-2.164792	-0.662446	-2.560487
38	1	0	-1.721706	3.171927	0.266835
39	1	0	-3.144061	3.002567	1.440813

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**2a-2** Standard orientation:

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	3.143606	0.845033	-0.593609
2	6	0	3.707413	-0.453570	-0.006880
3	6	0	2.519355	-1.429386	-0.107792
4	6	0	1.257225	-0.538567	-0.174020
5	8	0	0.762872	-0.243800	1.146697
6	6	0	0.057512	-1.134858	-0.929951
7	6	0	-1.179354	-0.630413	-0.167682
8	6	0	-0.626302	-0.589884	1.256840
9	6	0	4.213674	-0.208949	1.424057
10	8	0	3.774910	1.863295	-0.856478
11	7	0	1.797321	0.678435	-0.760577
12	6	0	-1.859827	0.647478	-0.805981
13	6	0	-2.734428	1.312146	0.228141
14	6	0	-3.875823	0.432442	0.720213
15	6	0	-3.560219	-1.085279	0.608304
16	6	0	-2.515529	-1.359407	-0.490538
17	6	0	-2.699398	-0.351881	-1.660415
18	6	0	-2.521354	2.555817	0.669830
19	1	0	4.547707	-0.773735	-0.631175
20	1	0	2.445190	-2.121968	0.733438
21	1	0	2.586966	-2.016358	-1.028736
22	1	0	0.104279	-2.228232	-0.858487
23	1	0	0.083475	-0.869986	-1.988894

24	1	0	-0.730036	-1.576942	1.733942
25	1	0	-1.107322	0.155317	1.895872
26	1	0	3.390022	0.084999	2.081852
27	1	0	4.966712	0.584213	1.434554
28	1	0	4.665632	-1.119197	1.829102
29	1	0	1.211616	1.477601	-0.966837
30	1	0	-1.216944	1.366308	-1.323572
31	1	0	-4.163318	0.703940	1.740337
32	1	0	-4.751058	0.647924	0.094039
33	1	0	-4.482312	-1.632475	0.383810
34	1	0	-3.195902	-1.476624	1.565248
35	1	0	-2.450380	-2.427085	-0.726830
36	1	0	-3.716594	-0.059079	-1.935640
37	1	0	-2.164793	-0.662447	-2.560487
38	1	0	-1.721705	3.171927	0.266833
39	1	0	-3.144059	3.002568	1.440812

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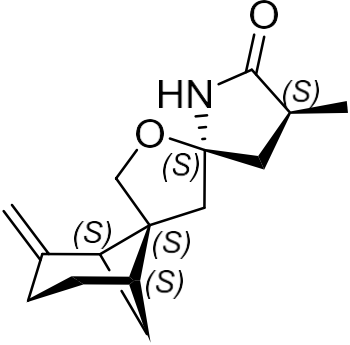
## References

[1] Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, J. A., Jr.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V.N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, J. M.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V.G.; Voth, G. A.; Salvador, P.; Dannenberg, J.J.; Dapprich, S.; Daniels, A.D.; Farkas, O.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J. Gaussian 09, Revision C1; Gaussian, Inc.: Wallingford, CT, 2010. [4]. Bruhn, T.; Hemberger, Y.; Schaumlöffel, A.; Bringmann, G. *Spec Dis*, version 1.51, University of Würzburg, Germany, 2010.

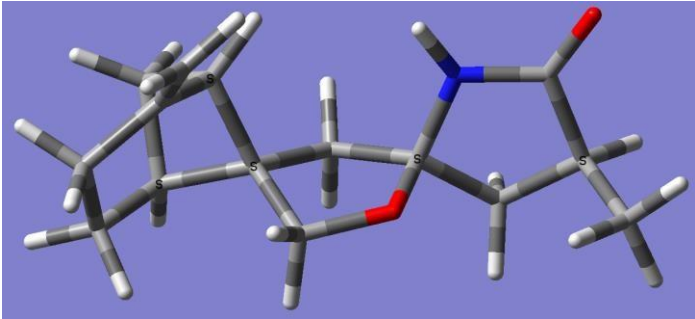
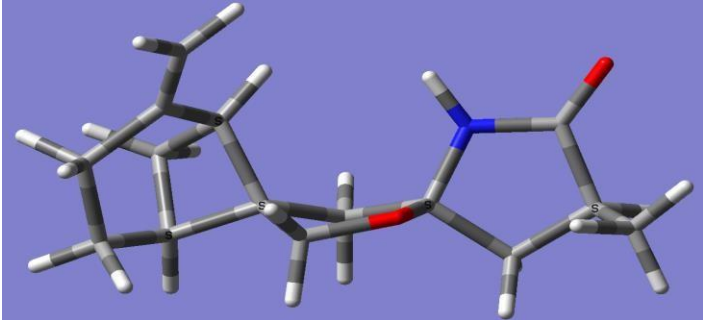


[2] Bruhn, T.; Schaumlöffel, A.; Hemberger, Y.; Bringmann, G. Quantifying the Comparison of Calculated and Experimental Electronic Circular Dichroism Spectra, *Chirality* 2013, 25, 243–249.

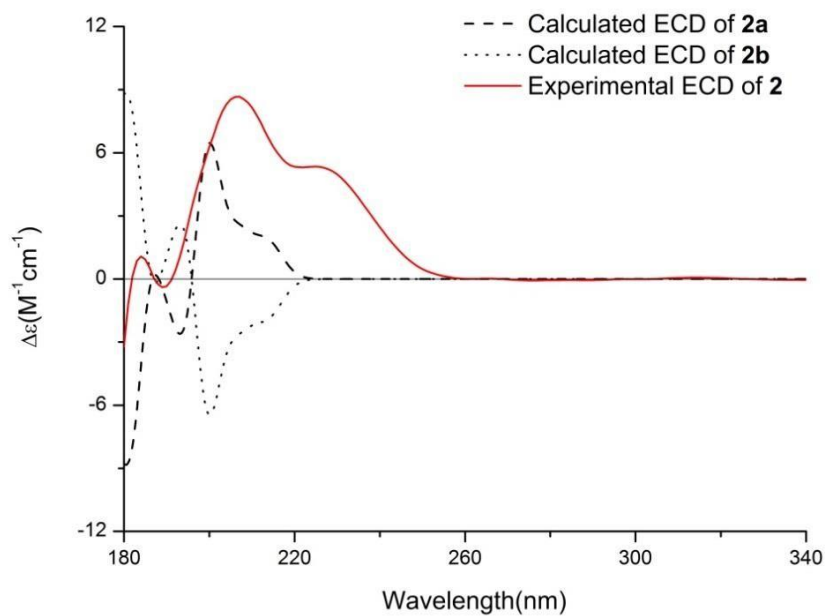
**Table S3.** 2D Structures of **2a**

label	structure
2a	

**Table S4** B3LYP/6-31++G (d, p) optimized lowest energy 3D conformers of **2a**.

label	conformer	Boltzmann weighting factors
2a-1		50.00
2a-2		50.00

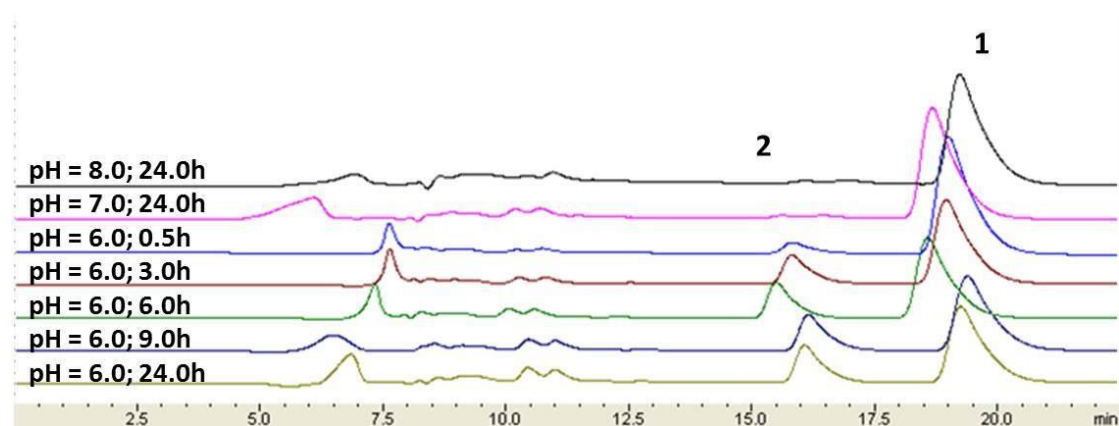
**Figure S1** Calculated and experimental ECD spectra of **2a**.



**Figure S2.** Calculated and experimental ECD spectra of **2**

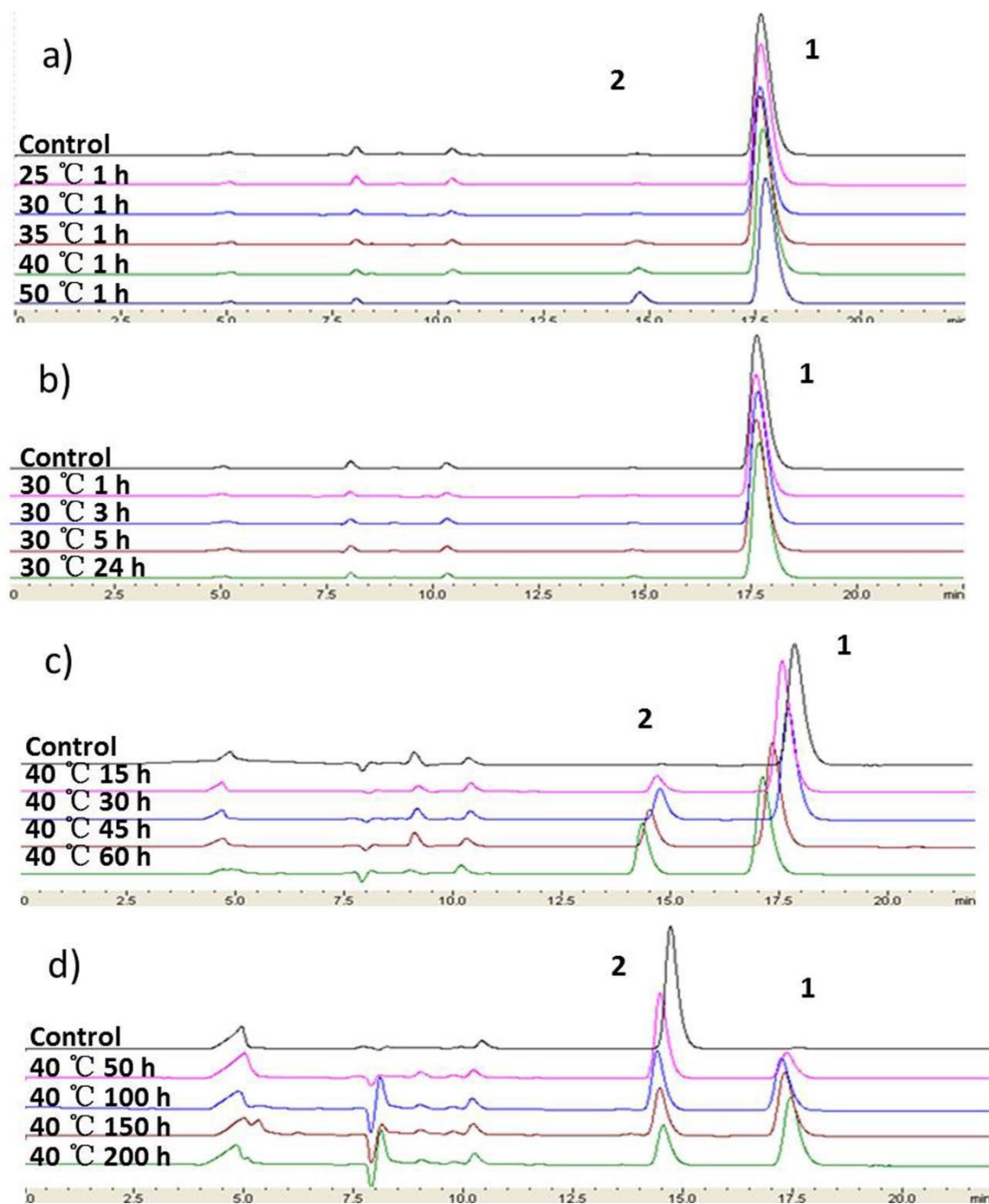
*pH-, H<sub>2</sub>O- and temperature-dependent epimerization effect*

pH-dependent epimerization effect:



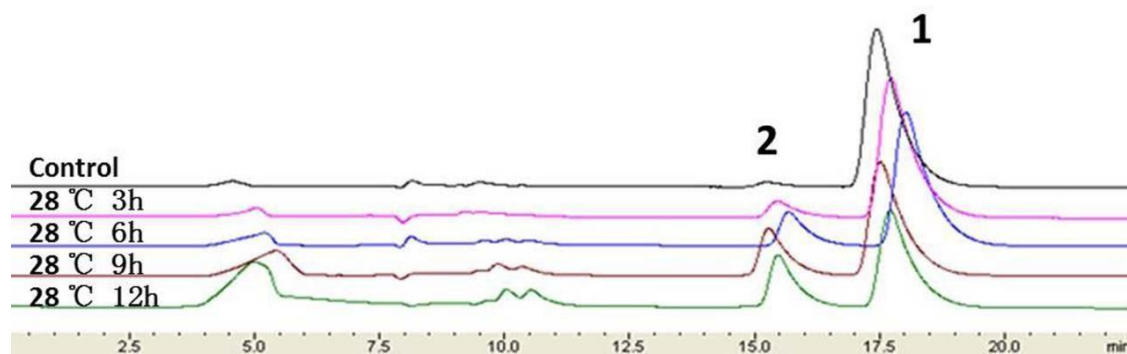
**Figure S3.** Dynamic chiral HPLC spectra of compound **1** dissolved in anhydrous ethanol at pH = 8.0, 7.0, 6.0.

Temperature-dependent epimerization effect:

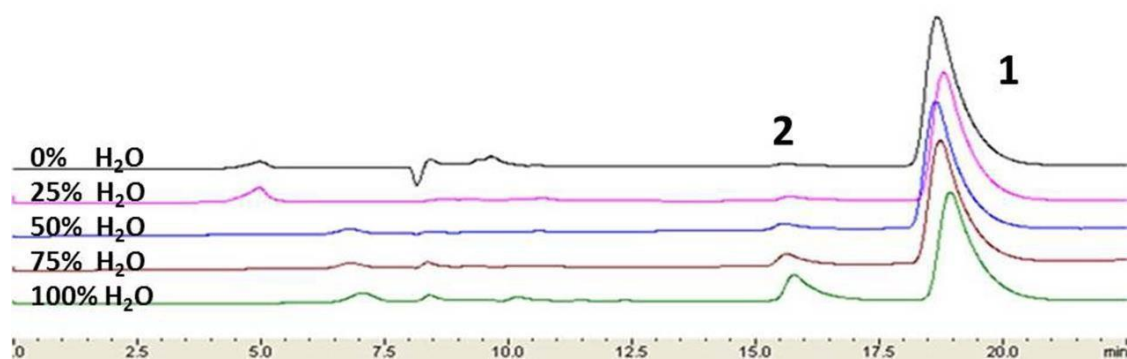


**Figure S4.** Dynamic chiral HPLC spectra: a) **1** dissolved in anhydrous ethanol heating at 25, 30, 35, 40, 50 °C; b) **1** dissolved in anhydrous ethanol heating at 30 °C; c) **1** dissolved in anhydrous ethanol heating at 40 °C; d) **2** dissolved in anhydrous ethanol heating at 40 °C

H<sub>2</sub>O-dependent epimerization effect:

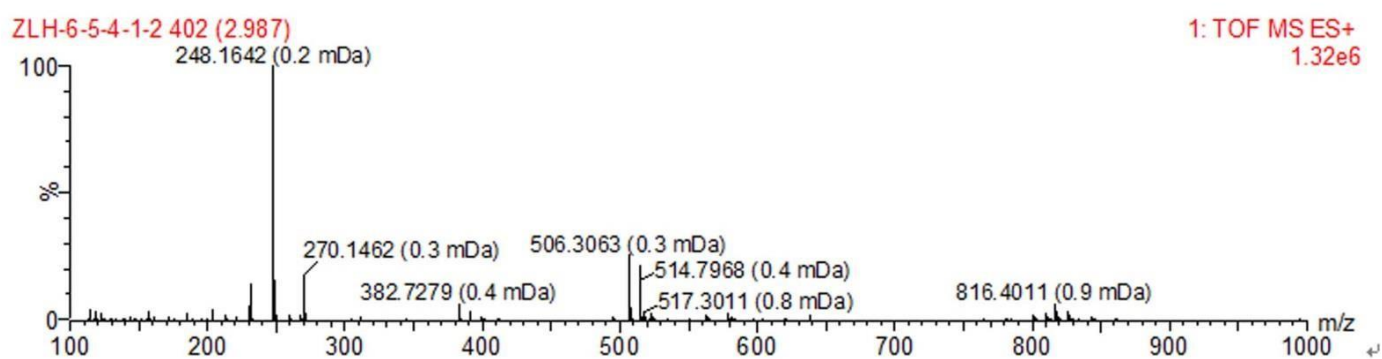
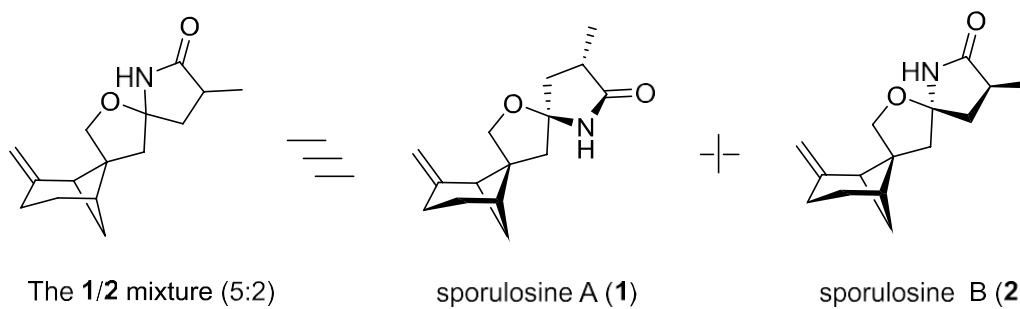


**Figure S5.** Dynamic chiral HPLC spectra of **1** dissolved in H<sub>2</sub>O heating at 28 °C



**Figure S6.** Chiral HPLC chromatogram of **1** dissolved in the mixed solution of Ethanol and H<sub>2</sub>O heating at 30 °C for 4h

**Figure S7. HR ESI-TOF MS spectrum for the mixture of 1 and 2 in CH<sub>3</sub>OH**



m/z	248.1642	C <sub>15</sub> H <sub>22</sub> NO <sub>2</sub>	[M+H] <sup>+</sup>
m/z	270.1462	C <sub>15</sub> H <sub>21</sub> NO <sub>2</sub> Na	[M+Na] <sup>+</sup>

**Figure S8.  $^1\text{H}$  NMR (600 MHz,  $\text{DMSO-}d_6$ ) spectrum for the mixture of 1 and 2**

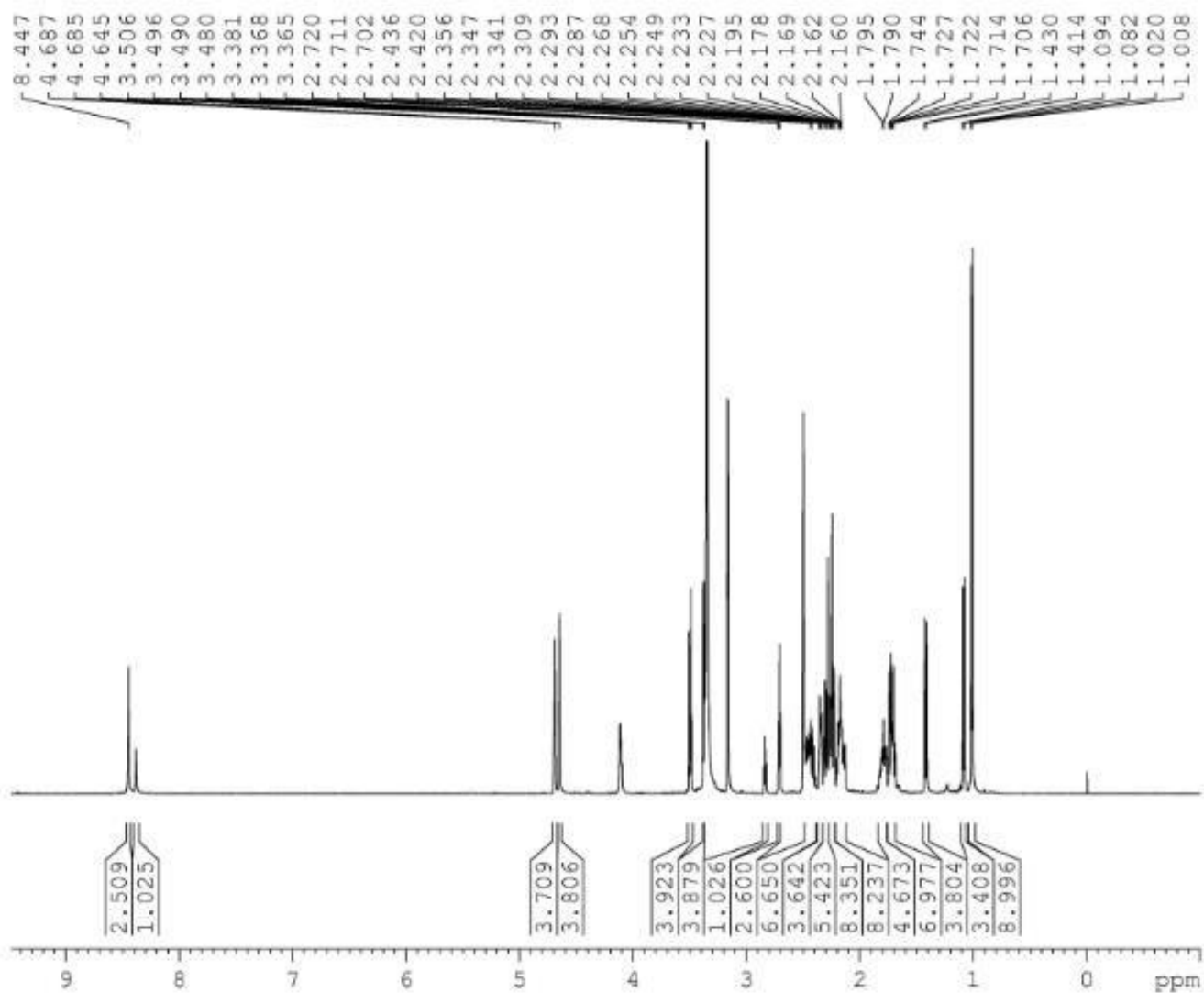
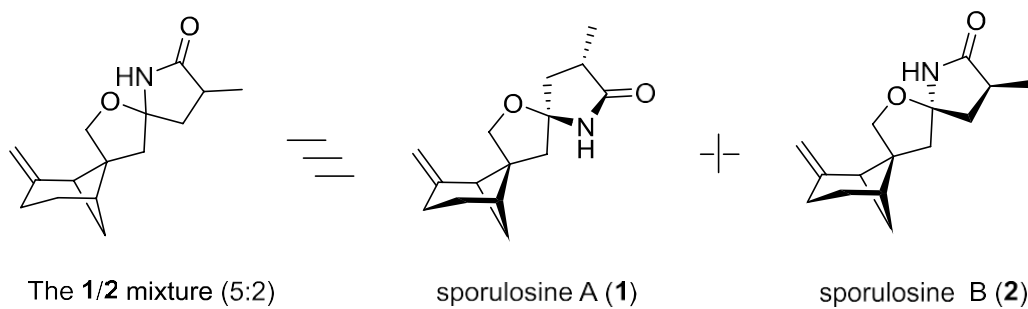
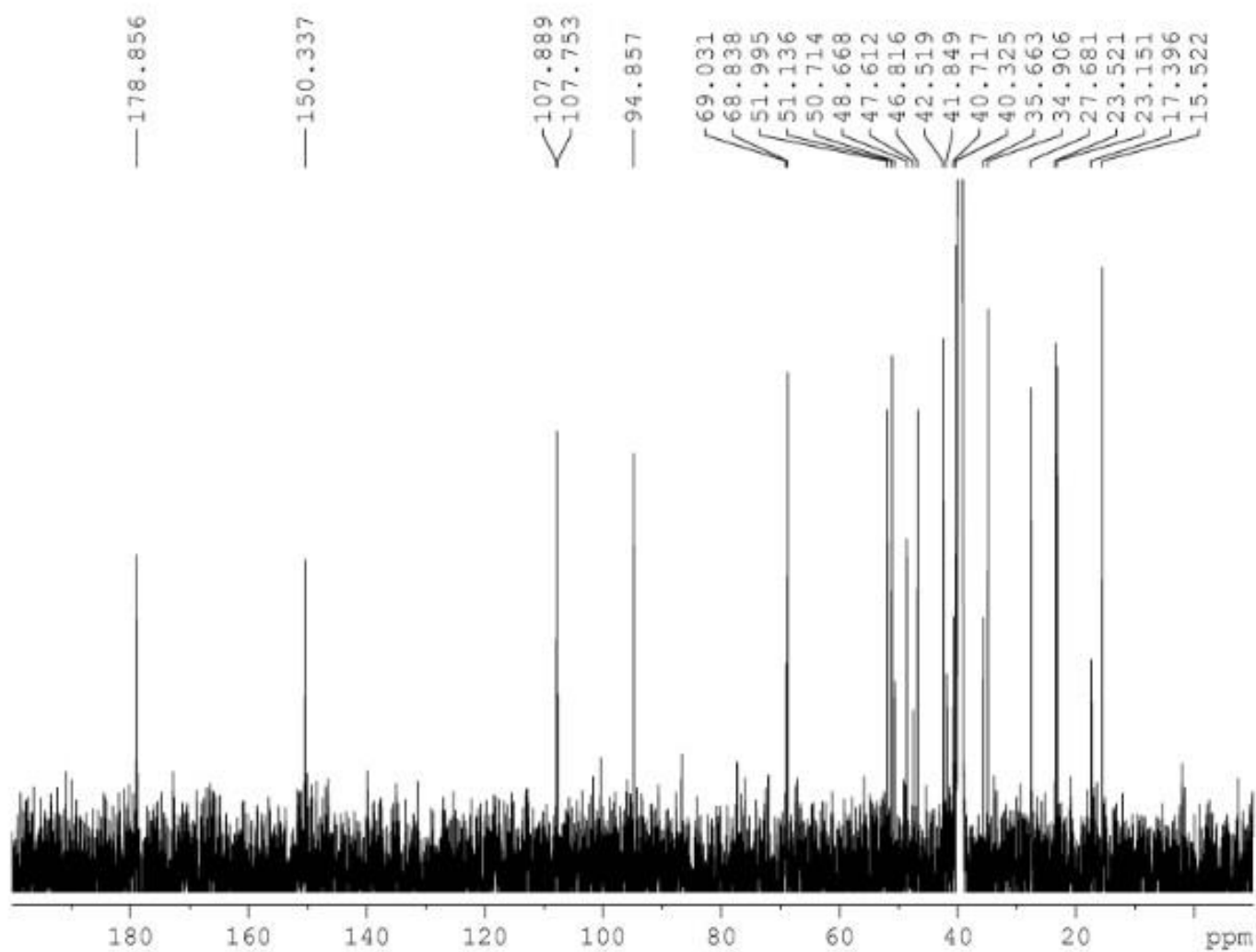
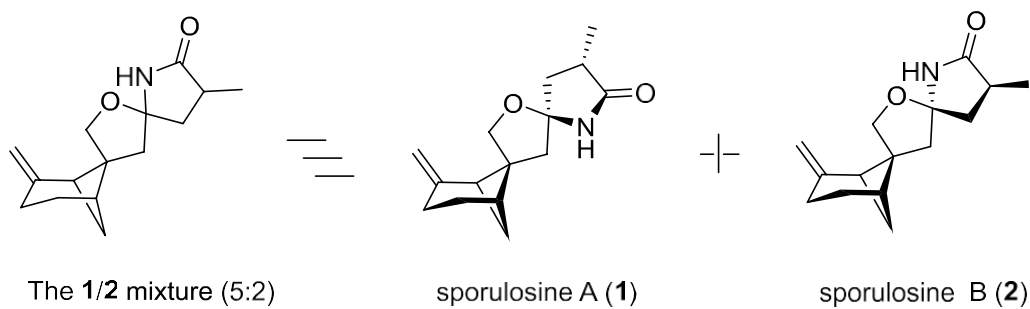
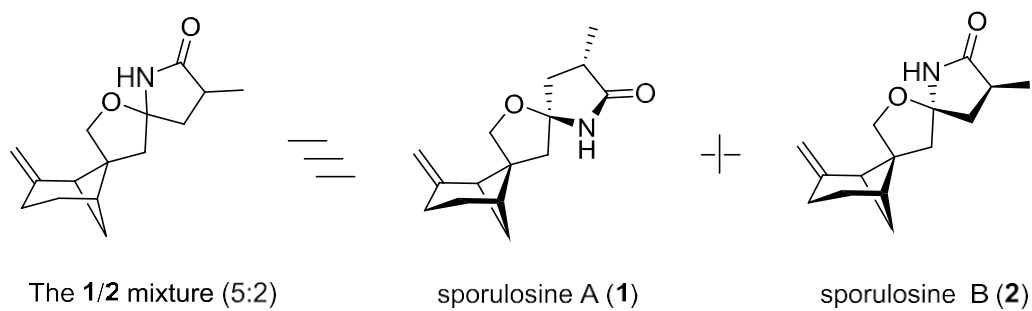


Figure S9.  $^{13}\text{C}$  NMR (100 MHz,  $\text{DMSO-}d_6$ ) spectrum for the mixture of 1 and 2



**Figure S10. HSQC (600 MHz, DMSO-*d*<sub>6</sub>) spectrum for the mixture of 1 and 2**



AV-600-HSQC  
Sample:

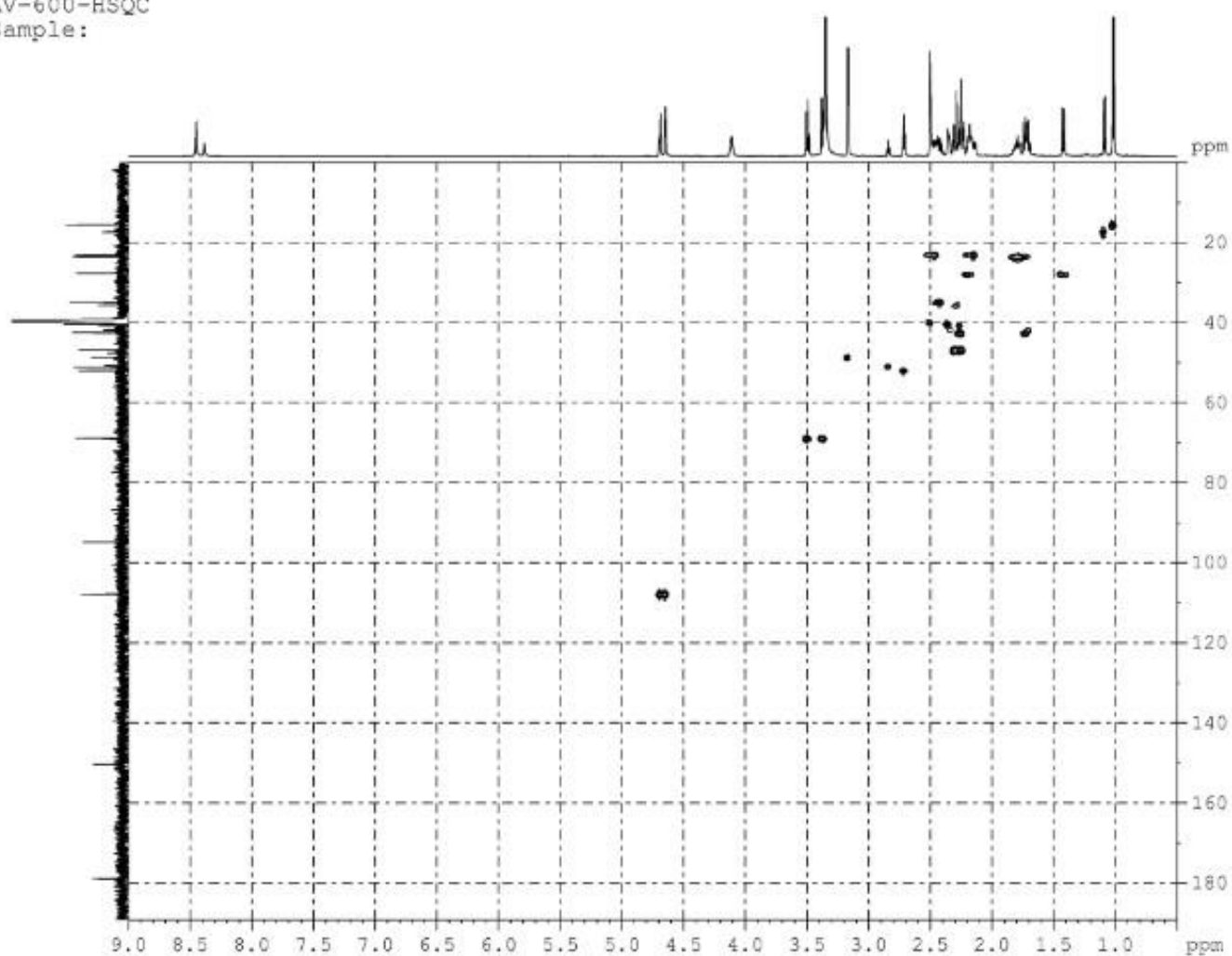
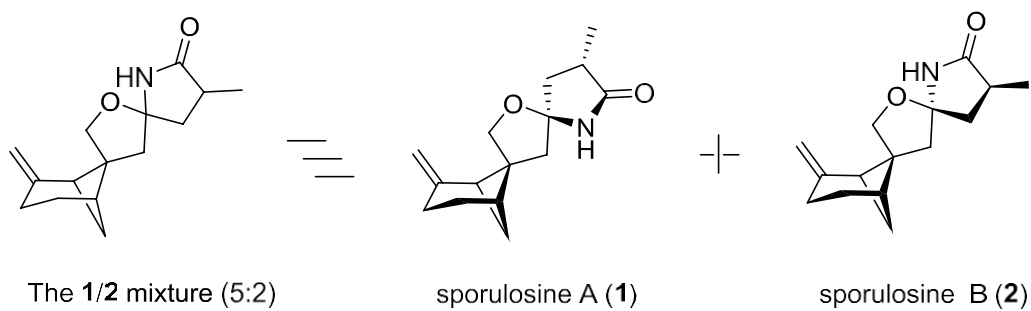




Figure S11. HMBC (600 MHz, DMSO-*d*<sub>6</sub>) spectrum for the mixture of 1 and 2



AV-600-HMBC  
Sample:

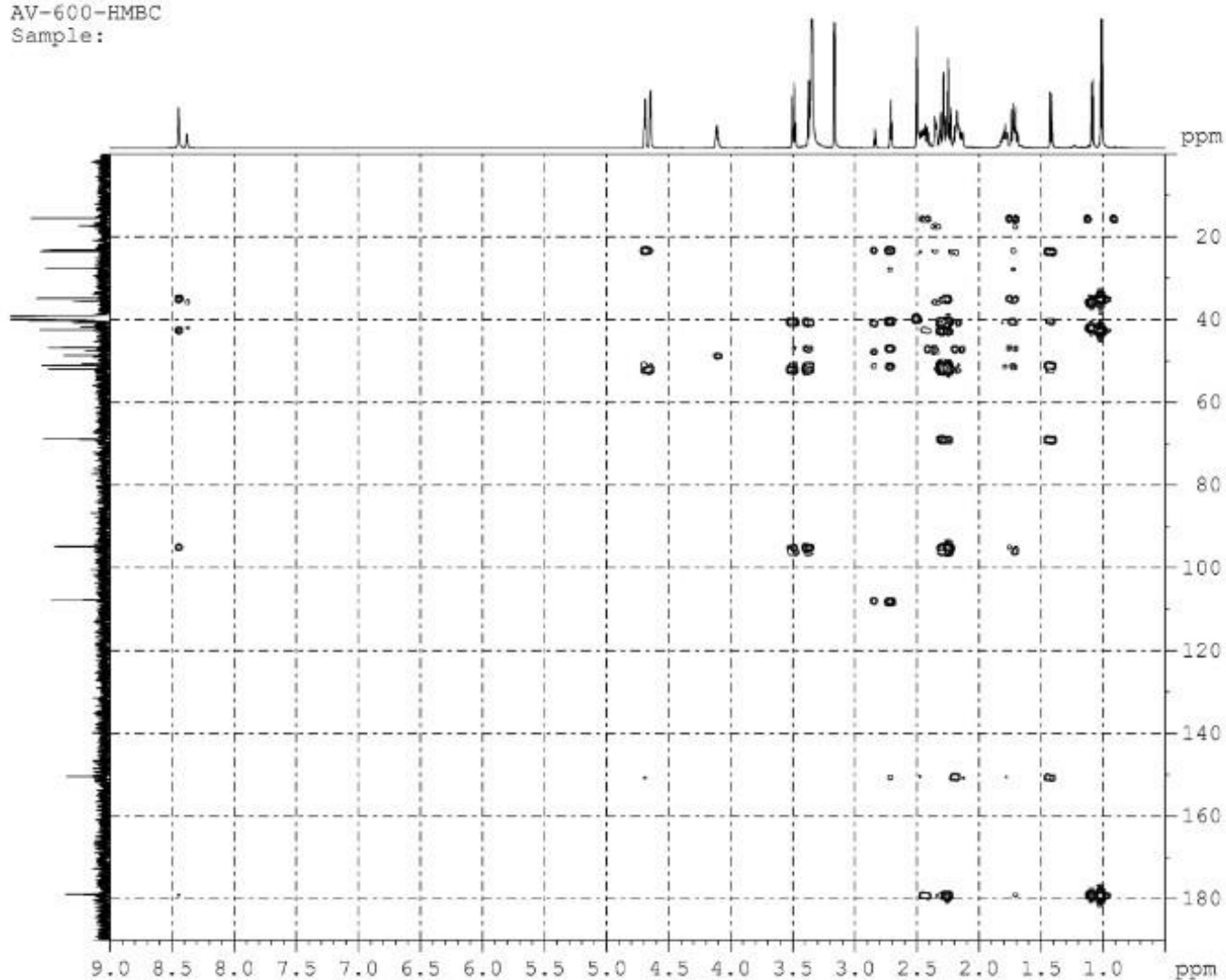


Figure S12. NOESY (600 MHz, DMSO-*d*<sub>6</sub>) spectrum for the mixture of 1 and 2

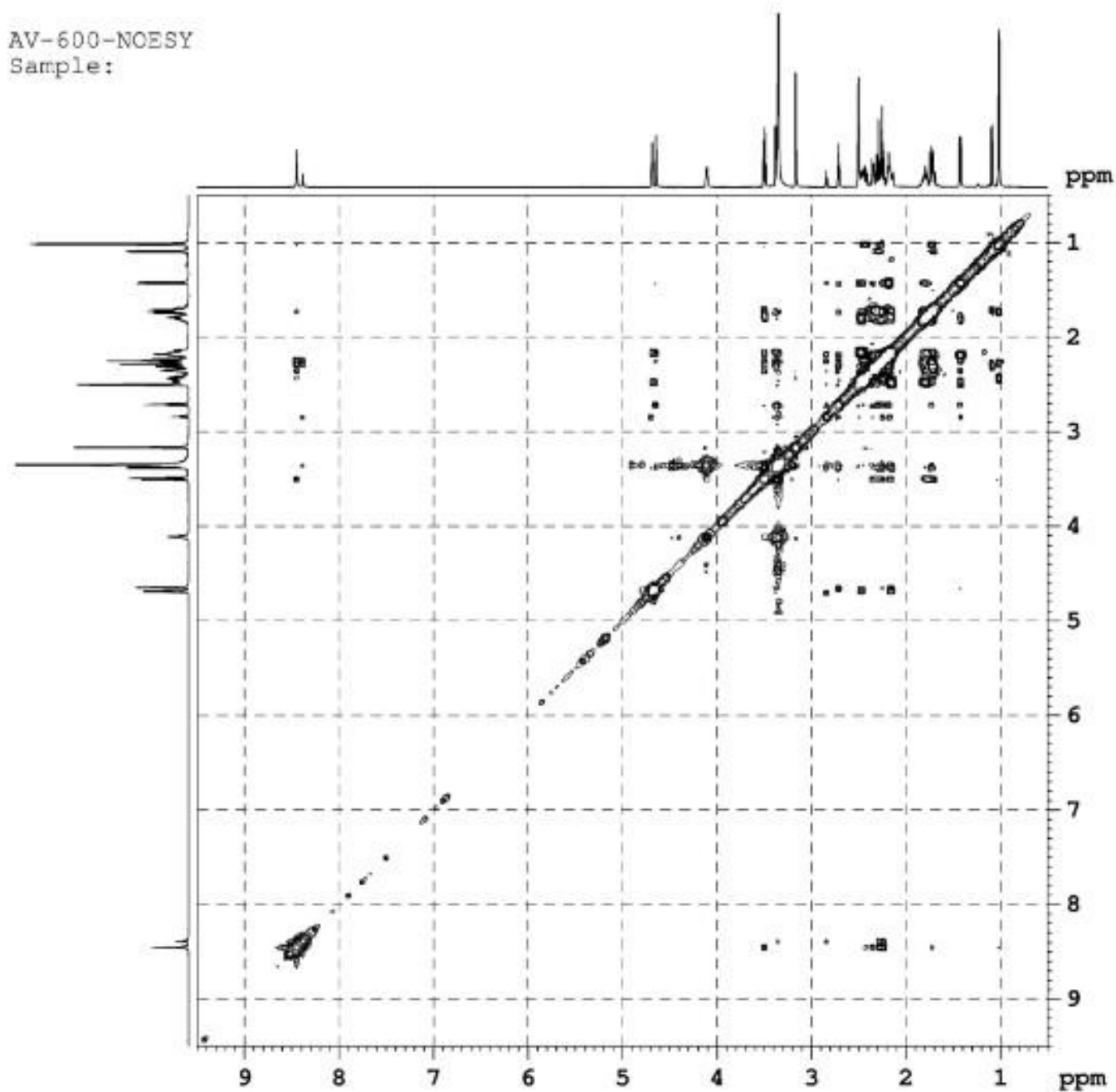
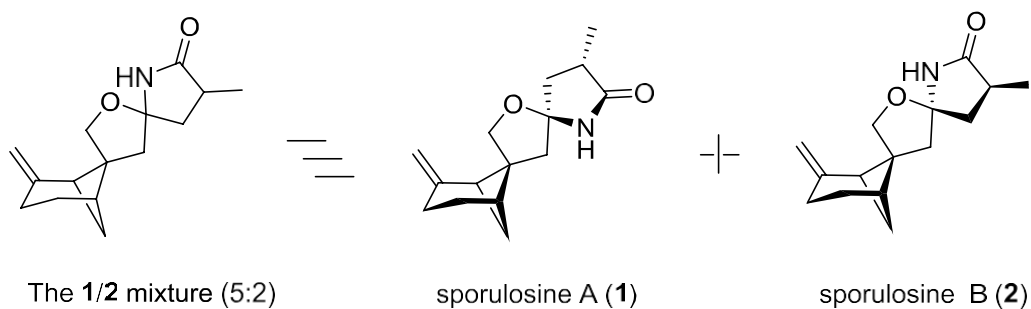
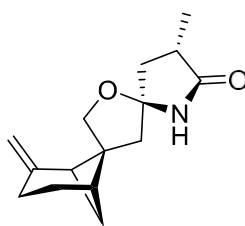


Figure S13. HR ESI-TOF MS spectrum of compound 1 in CH<sub>3</sub>OH



sporulosine A (1)

### Mass Spectrum Molecular Formula Report

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 Sample Name ZLH-6-D  
 Comment

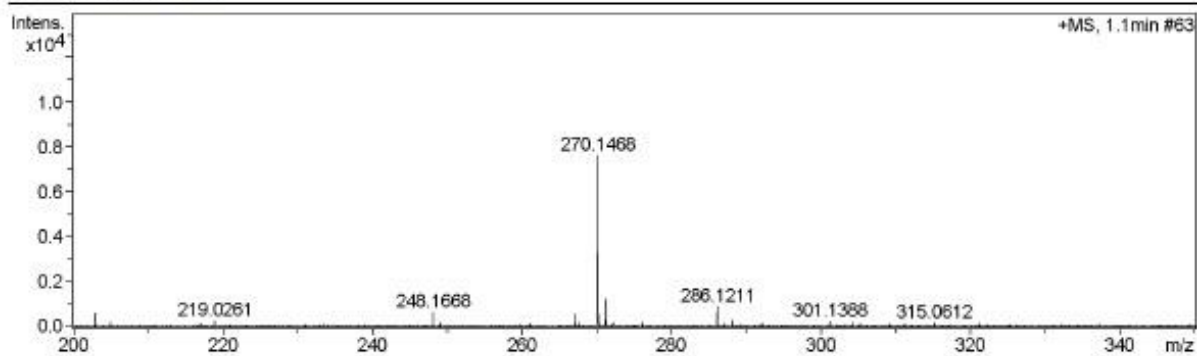
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 Operator Bruker Customer  
 Instrument / Ser# micrOTOF-Q 125

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Scan End	3000 m/z	Set Collision Cell RF	400.0 Vpp	Set Divert Valve	Source

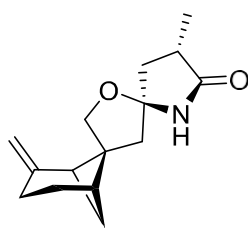
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Formula, max.					
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Nitrogen Rule	yes	Electron Configuration	both	Maximum	3
Filter H/C Ratio	yes	Minimum	0		
Estimate Carbon	yes				



Sum Formula	Sigma	m/z	Err [ppm]	Mean Err [ppm]	Err [mDa]	rdb	N Rule	e <sup>-</sup>
C 15 H 21 N 1 Na 1 O 2	0.007	270.1465	-1.42	-1.09	-0.38	5.50	ok	even

**Figure S14.  $^1\text{H}$  NMR (600 MHz,  $\text{DMSO-}d_6$ ) spectrum of compound 1**



sporulosine A (1)

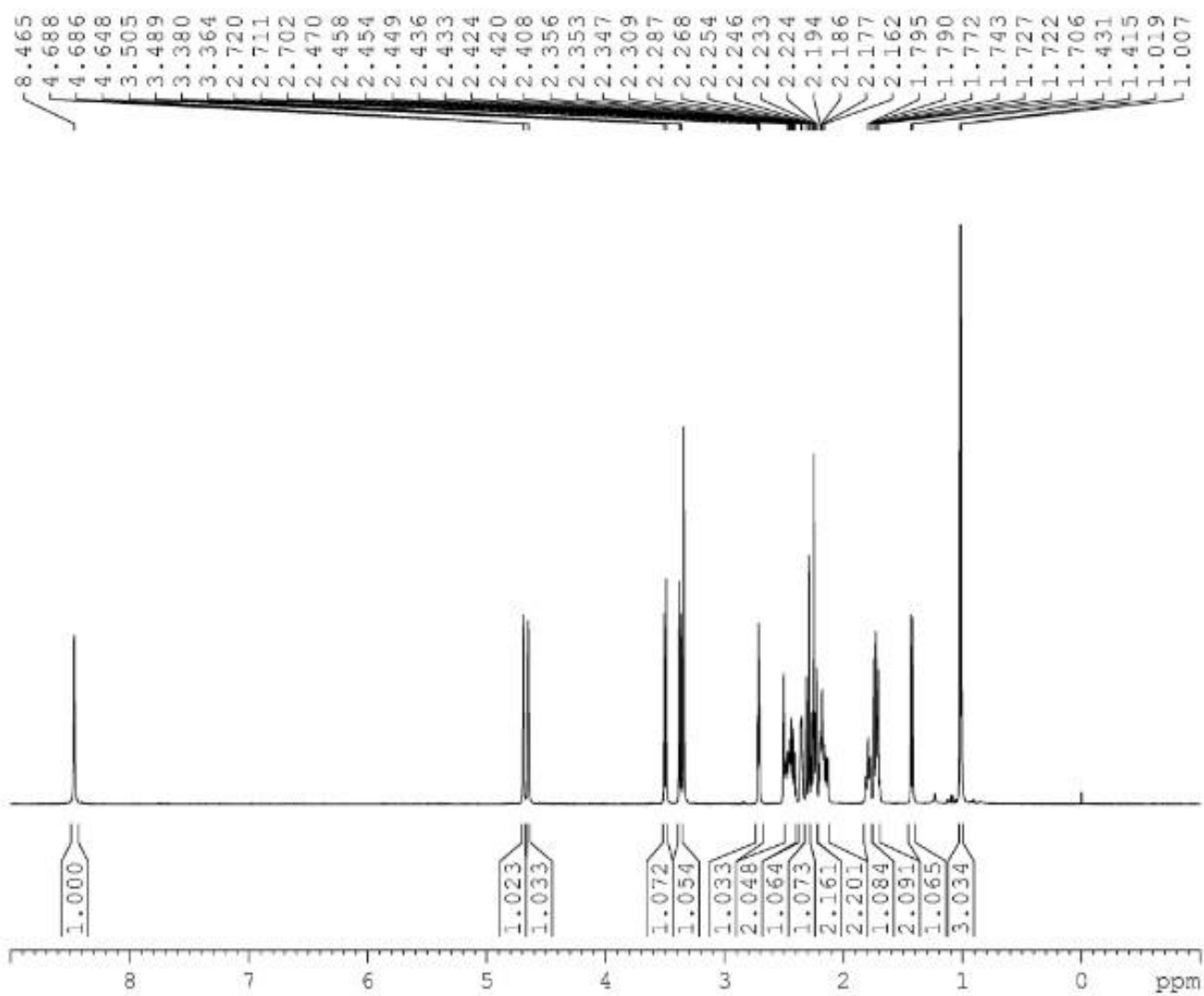
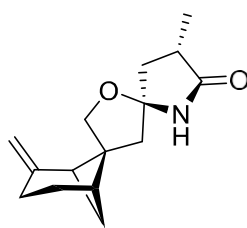


Figure S15.  $^{13}\text{C}$  NMR (150 MHz,  $\text{DMSO-}d_6$ ) spectrum of compound 1



sporulosine A (1)

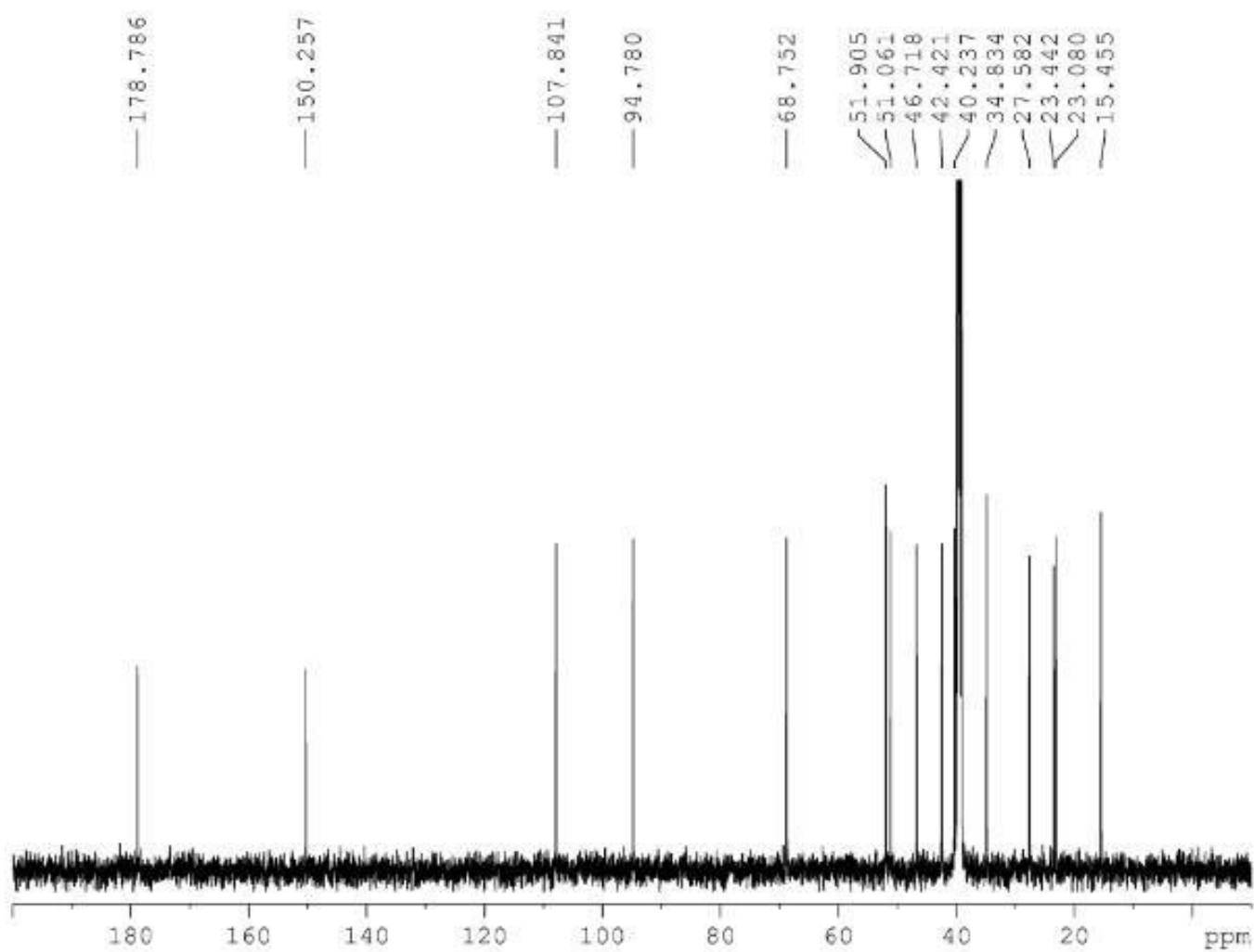
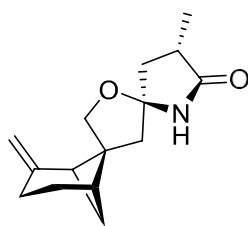
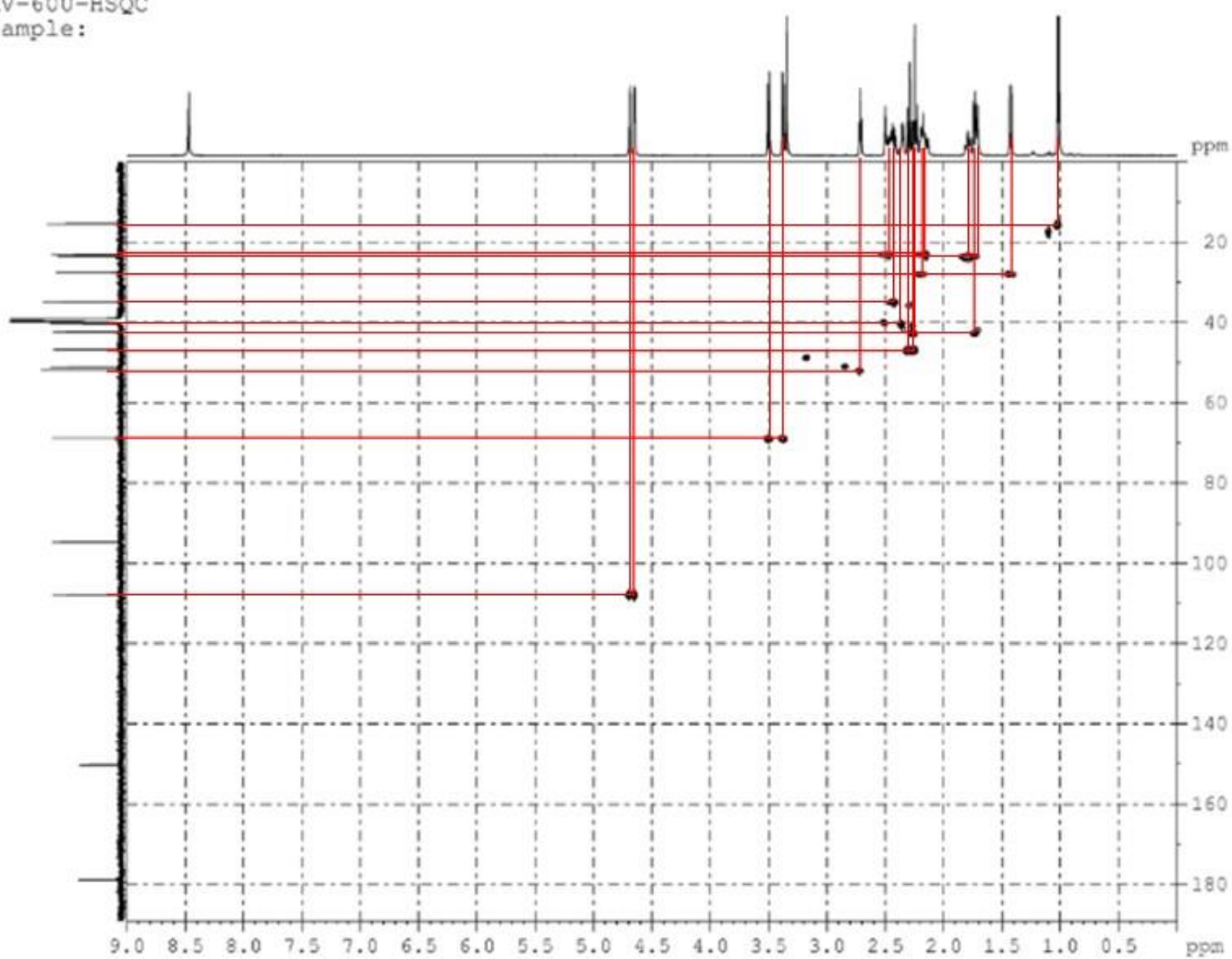


Figure S16. The HSQC correlations of compound 1 in HSQC (600 MHz, DMSO-*d*<sub>6</sub>) spectrum of the 1/2 mixture

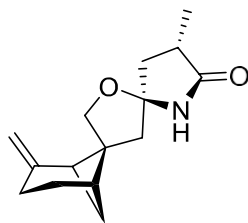


sporulosine A (1)

AV-600-HSQC  
Sample:



**Figure S17. The HMBC correlations of compound 1 in HMBC (600 MHz, DMSO-*d*<sub>6</sub>) spectrum of the 1/2 mixture**



sporulosine A (1)

AV-600-HMBC  
Sample:

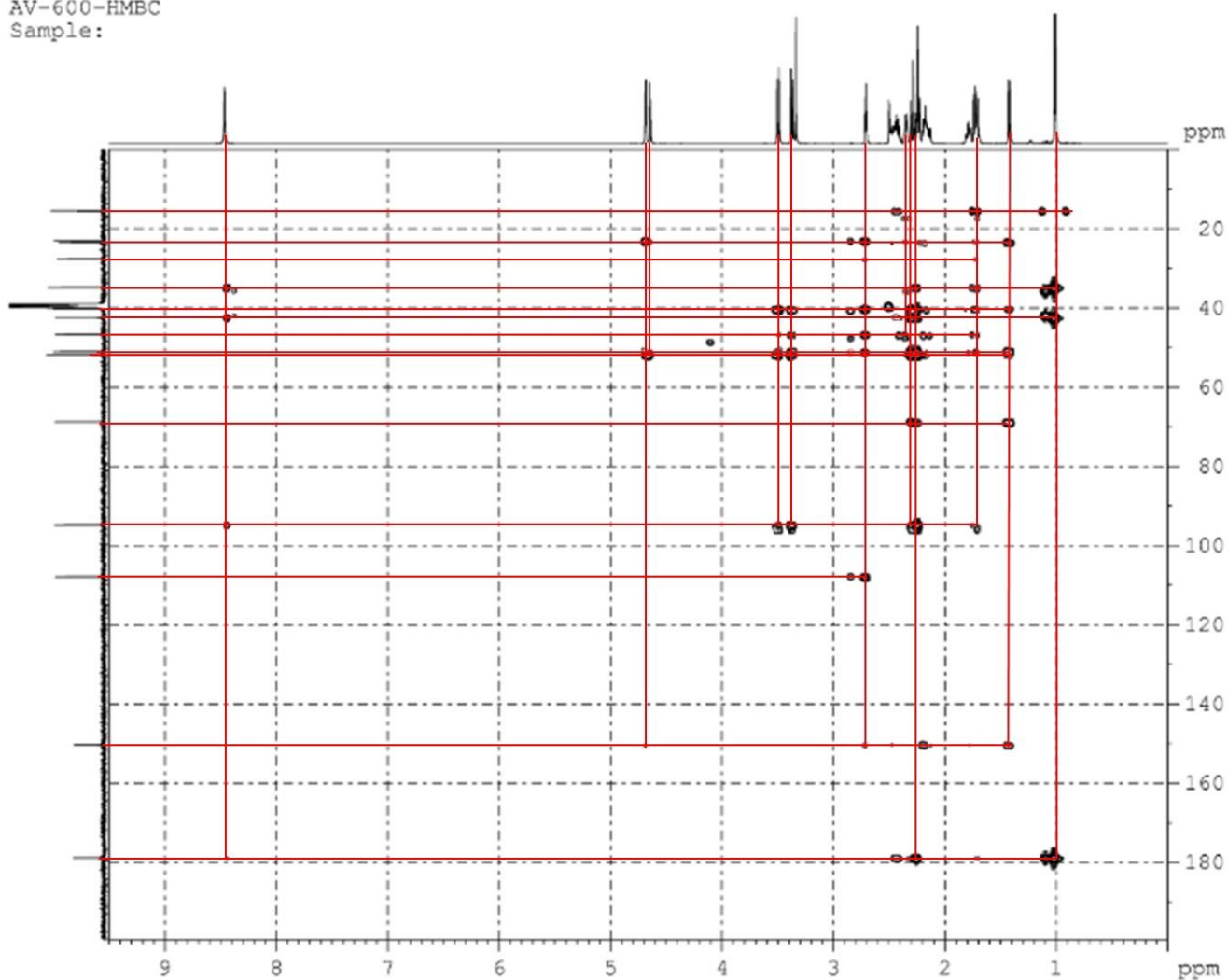
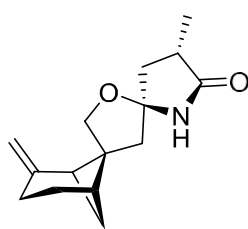


Figure S18. NOESY (600 MHz, DMSO-*d*<sub>6</sub>) spectrum of compound 1



sporulosine A (1)

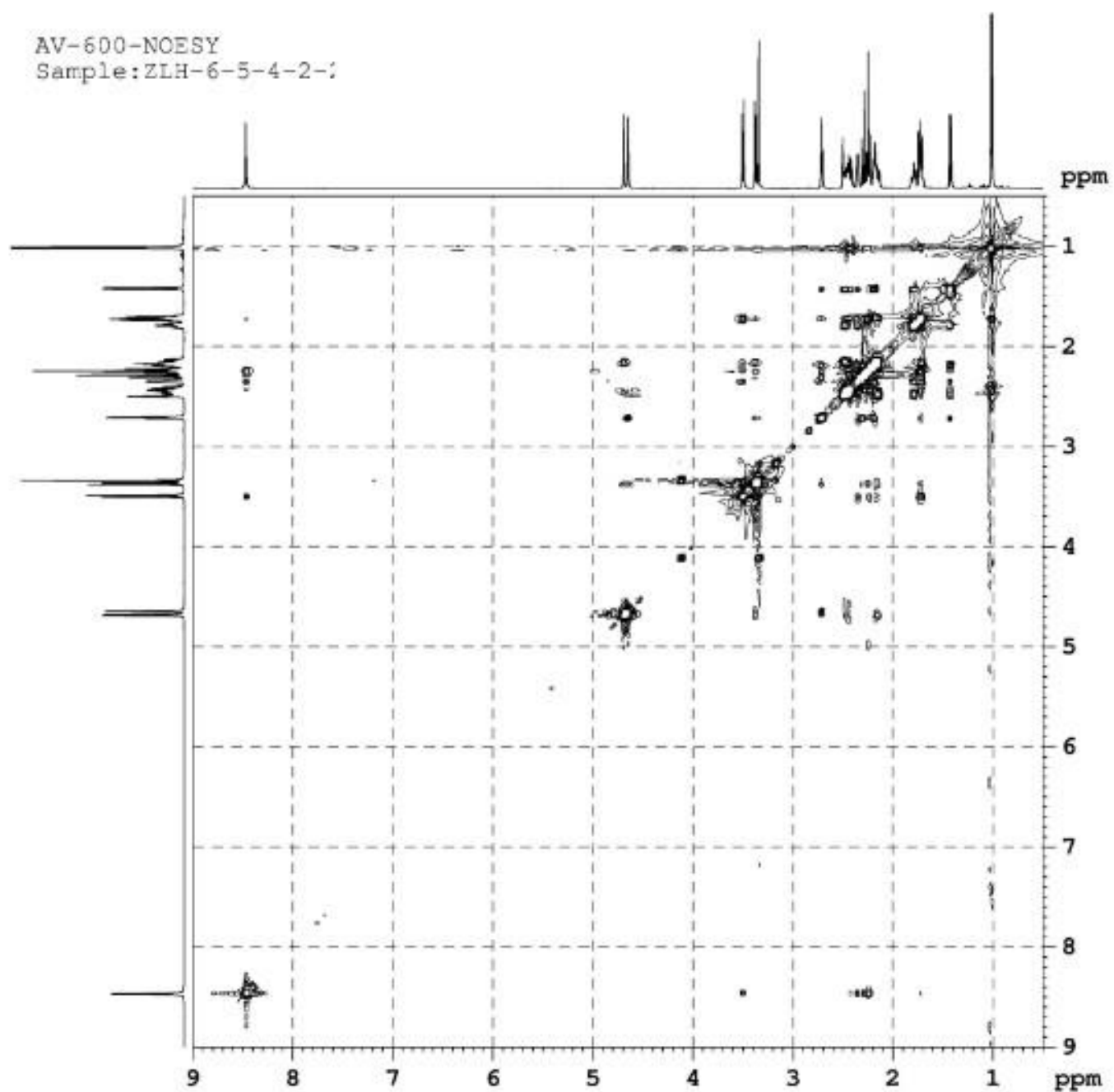
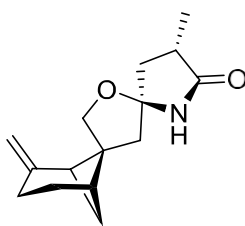




Figure S19. IR spectrum of compound 1



sporulosine A (1)

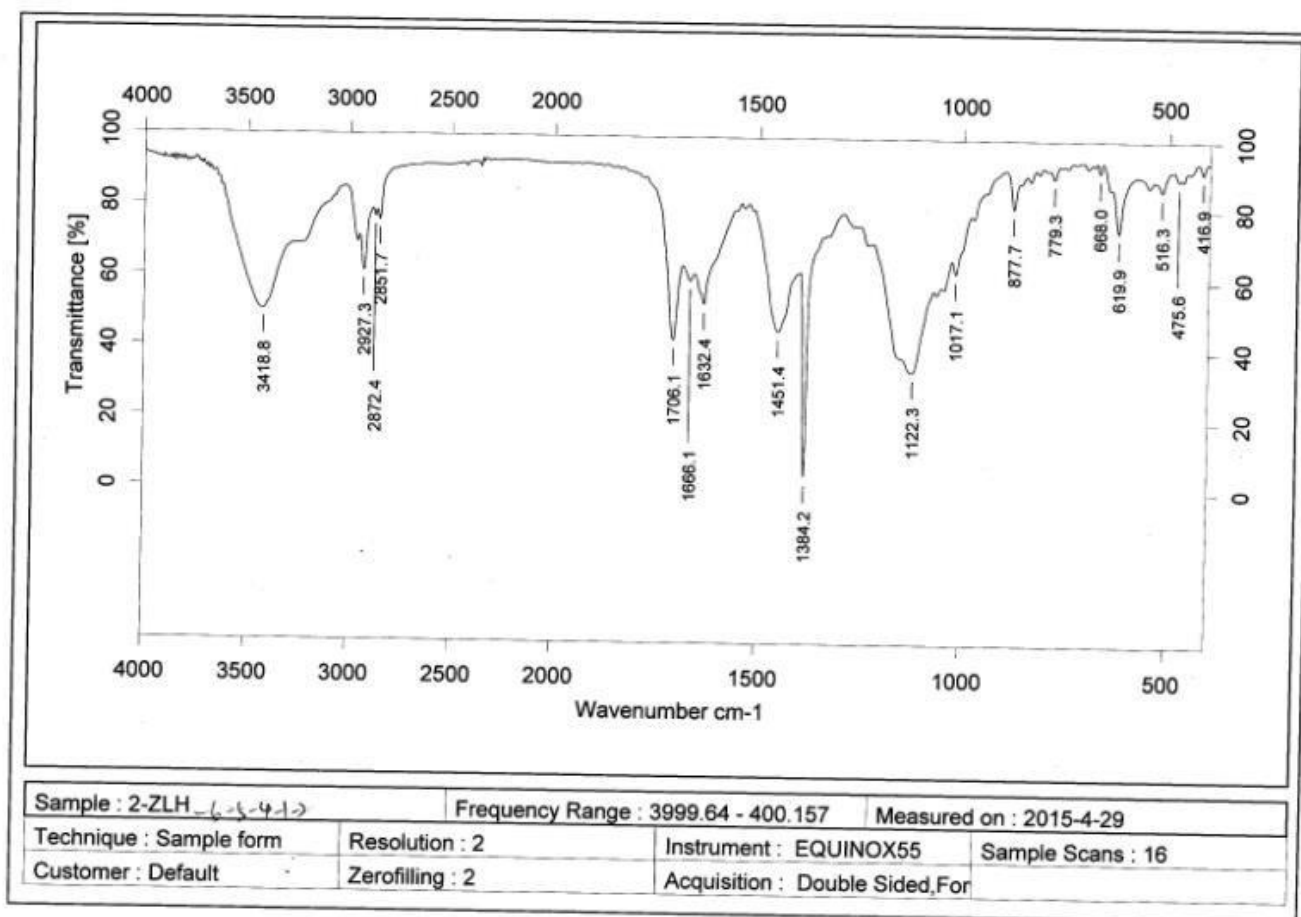
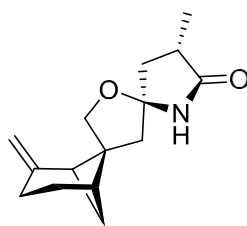
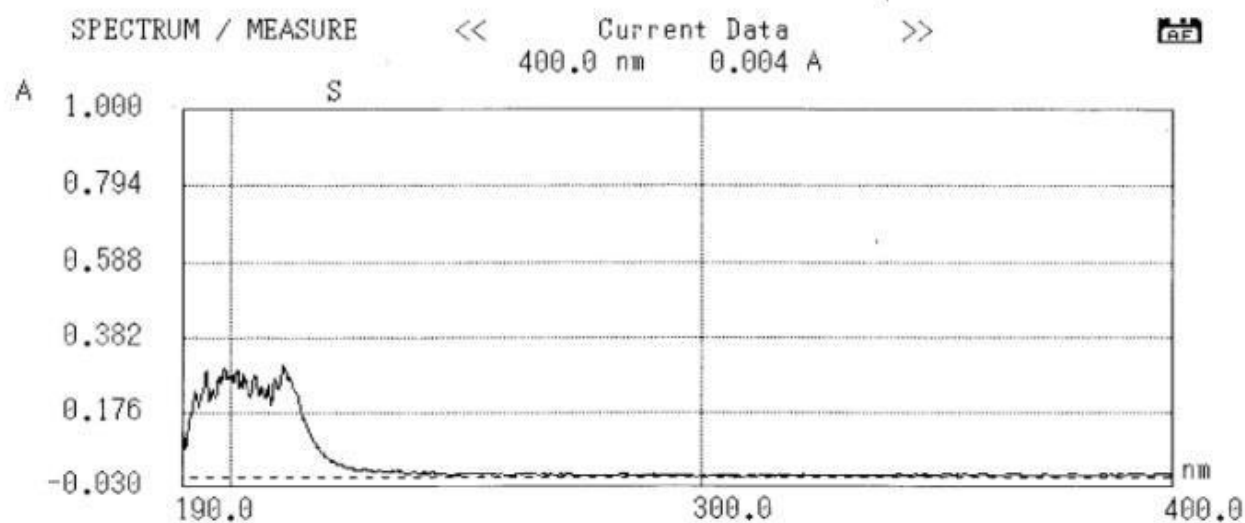


Figure S20. UV spectrum of compound 1



sporulosine A (1)

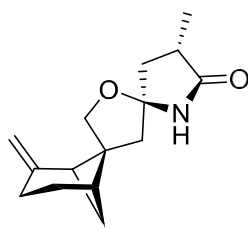


> Set sample then press [Start Scan].

⊞Enlarge/Reduce File (SP) Rec. Output Set Parameters Start Scan

NO.	ABSCISSA	PEAK	HEIGHT	ABSCISSA	VALLEY	HEIGHT
1	211.0	0.3096	0.0816	208.6	0.1987	-0.0726
2	205.2	0.2783	0.0807			
3	192.4	0.2341	0.0690			

**Figure S21. CD spectrum of compound 1**



sporulosine A (**1**)

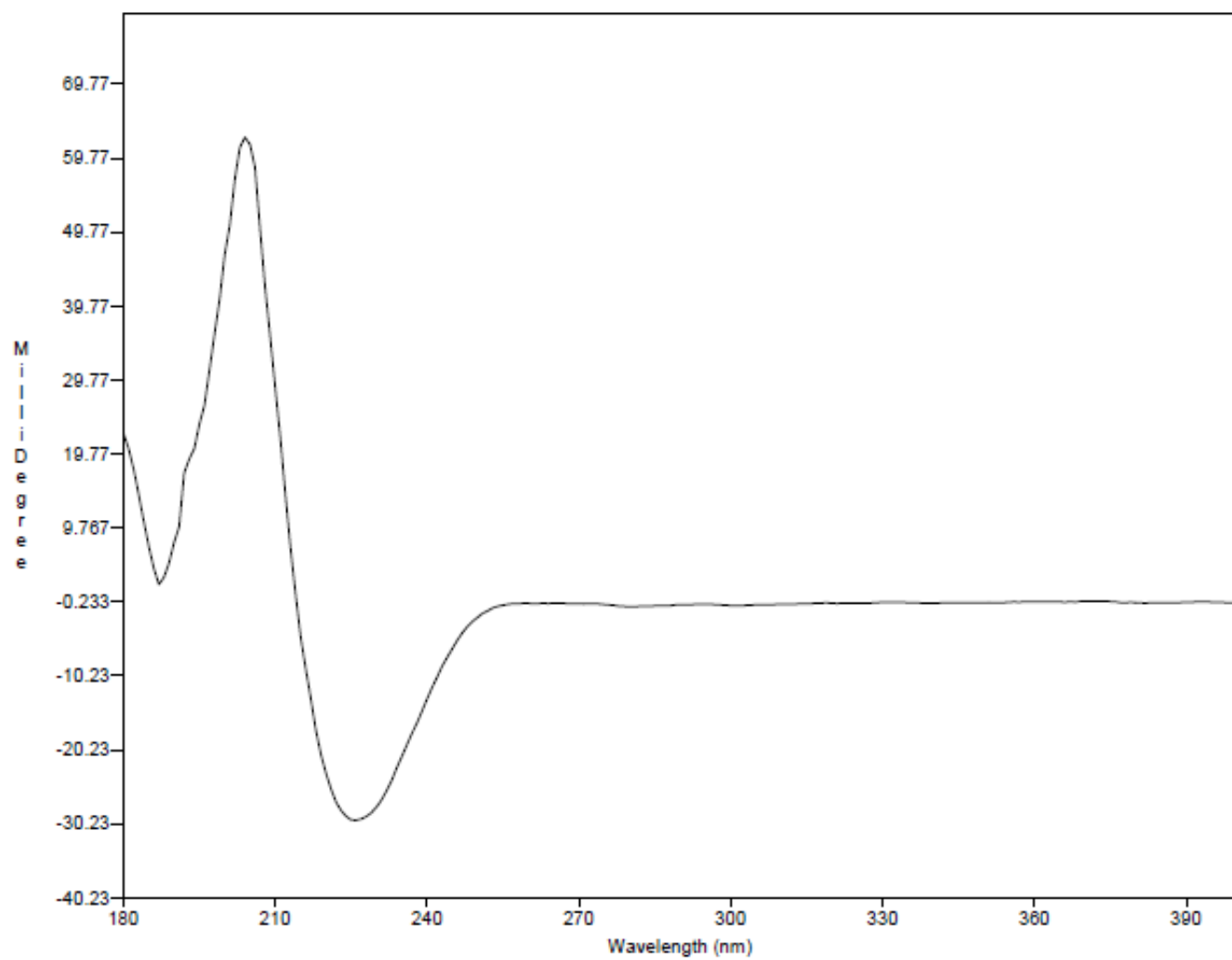
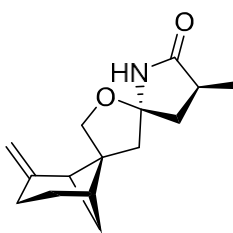


Figure S22. HR ESI-TOF MS spectrum of compound 2 in CH<sub>3</sub>OH



sporulosine B (2)

### Mass Spectrum Molecular Formula Report

#### Analysis Info

Analysis Name D:\Data\20151230CEYANG\ZLH-6-X.d  
 Method tune\_wide\_pos.m  
 Sample Name ZLH-6-X  
 Comment

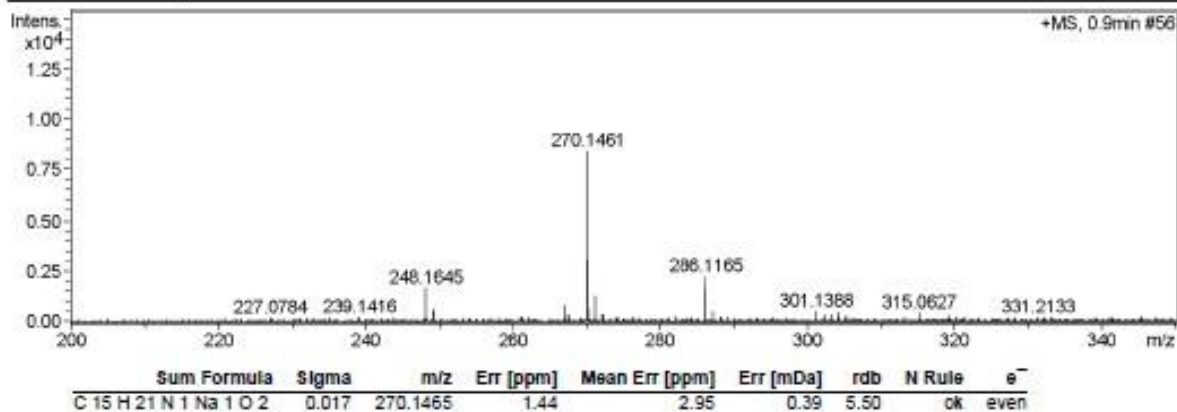
Acquisition Date 12/30/2015 2:16:51 PM  
 Operator Bruker Customer  
 Instrument / Ser# micrOTOF-Q 125

#### Acquisition Parameter

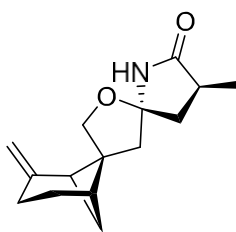
Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.3 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	3000 m/z	Set Collision Cell RF	400.0 Vpp	Set Divert Valve	Source

#### Generate Molecular Formula Parameter

Formula, min.	C <sub>8</sub> H <sub>21</sub> N <sub>1</sub> Na <sub>1</sub> O <sub>2</sub>		
Formula, max.			
Measured m/z	270.146	Tolerance	5 ppm
Charge		Charge	1
Check Valence	no	Minimum	0
Nitrogen Rule	yes	Electron Configuration	both
Filter H/C Ratio	yes	Minimum	0
Estimate Carbon	yes	Maximum	3



**Figure S23.  $^1\text{H}$  NMR (600 MHz,  $\text{DMSO-}d_6$ ) spectrum of compound 2**



sporulosine B (2)

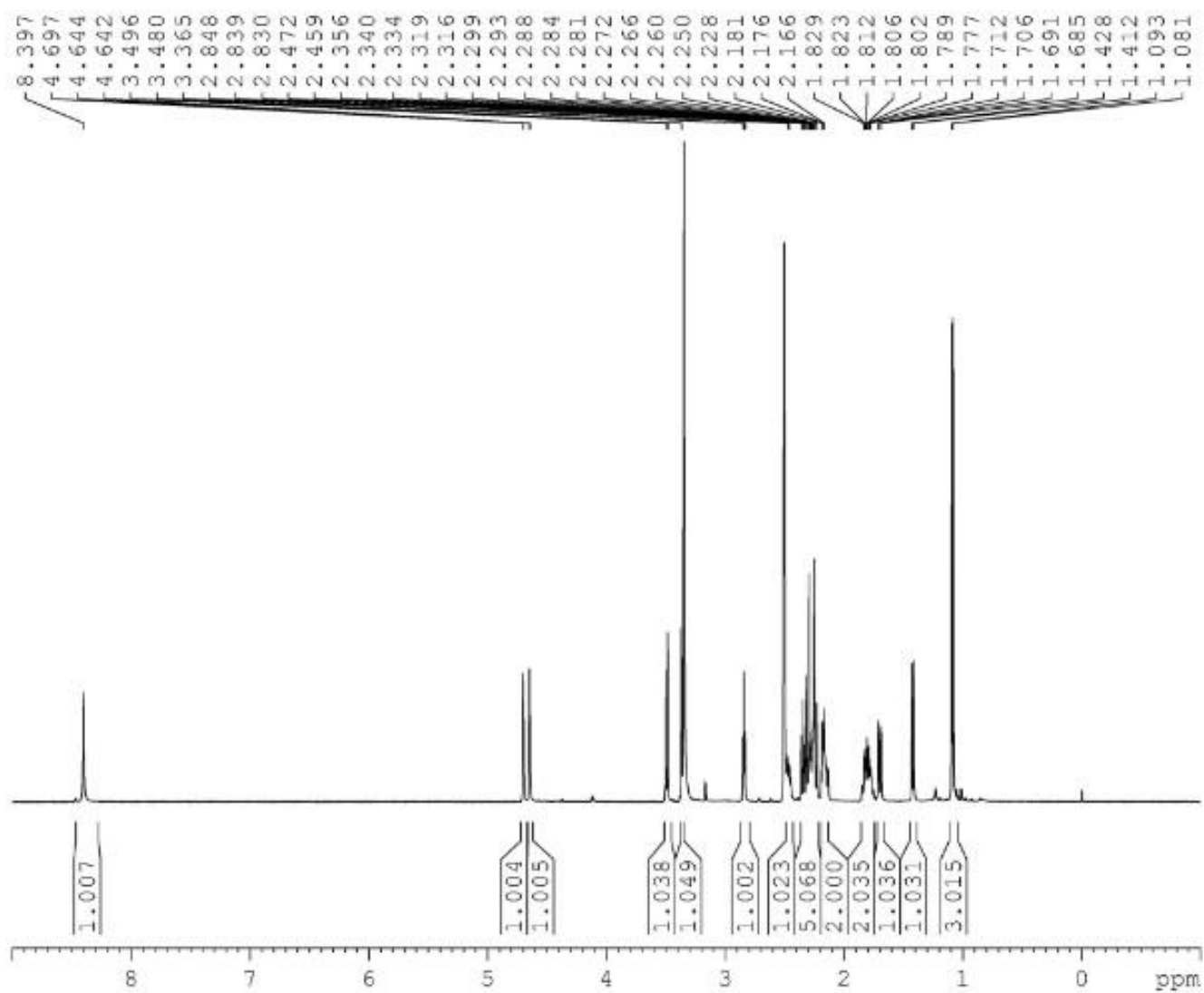
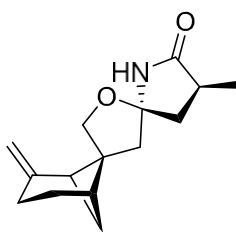


Figure S24.  $^{13}\text{C}$  NMR (150 MHz,  $\text{DMSO-}d_6$ ) spectrum of compound 2



sporulosine B (2)

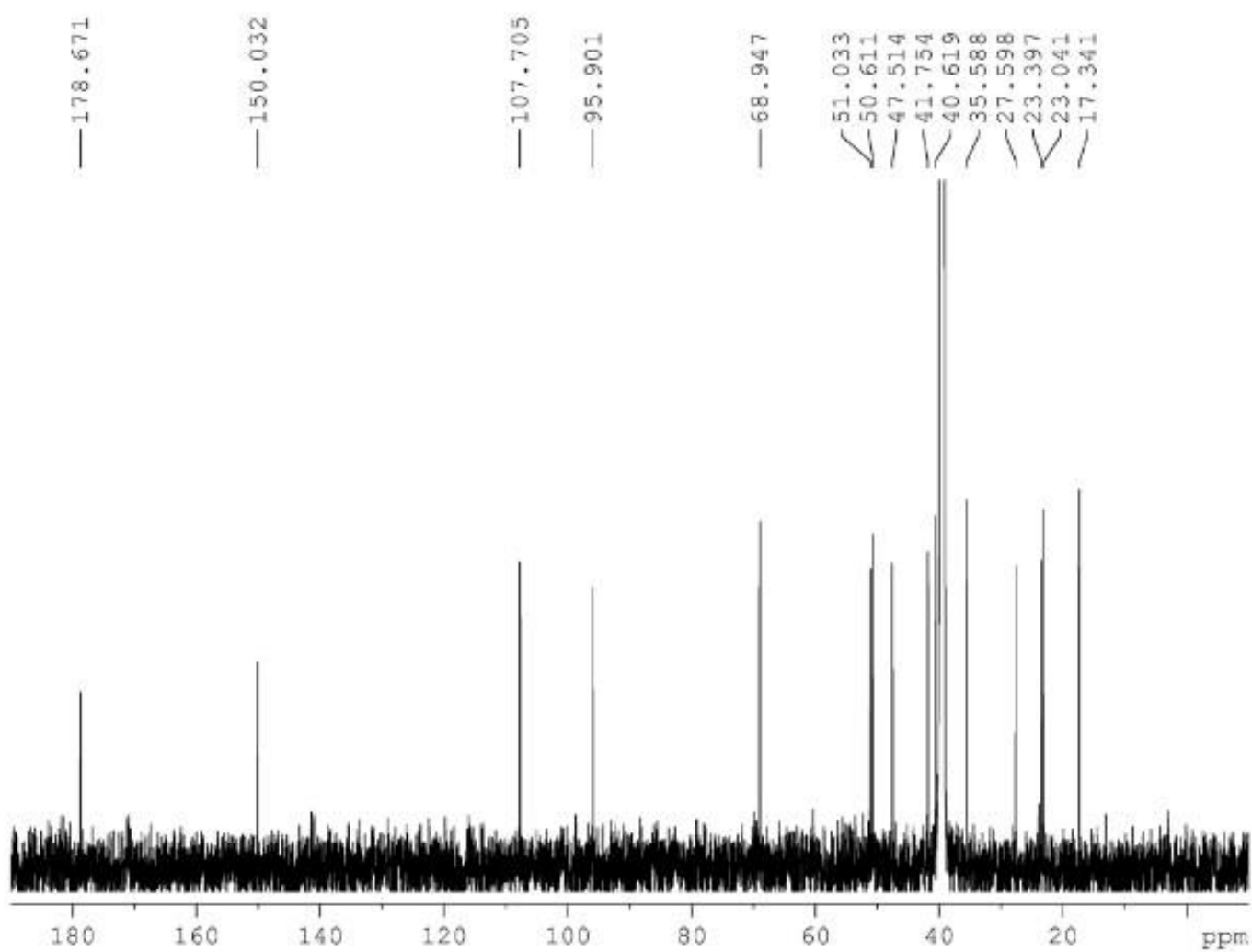
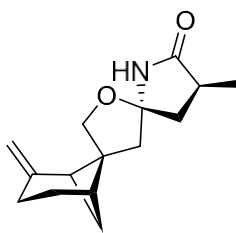


Figure S25. The HSQC correlations of compound 2 in HSQC (600 MHz, DMSO-*d*<sub>6</sub>) spectrum of the 1/2 mixture



sporulosine B (2)

AV-600-HSQC  
Sample:

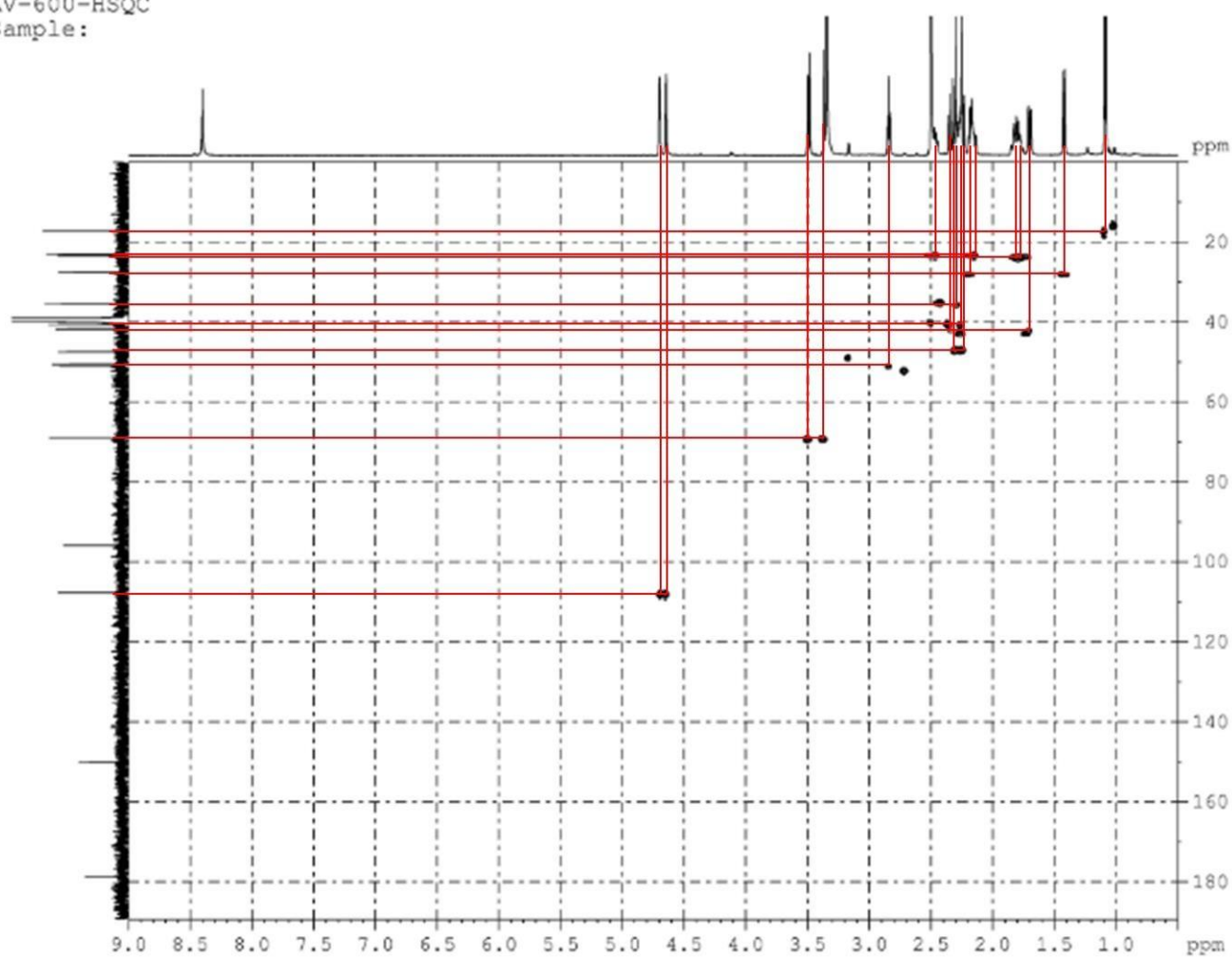
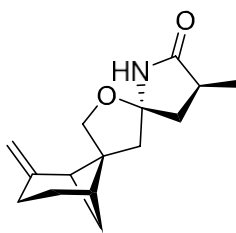


Figure S26. The HMBC correlations of compound 2 in HMBC (600 MHz, DMSO- $d_6$ ) spectrum of the 1/2 mixture



sporulosine B (2)

AV-600-HMBC  
Sample:

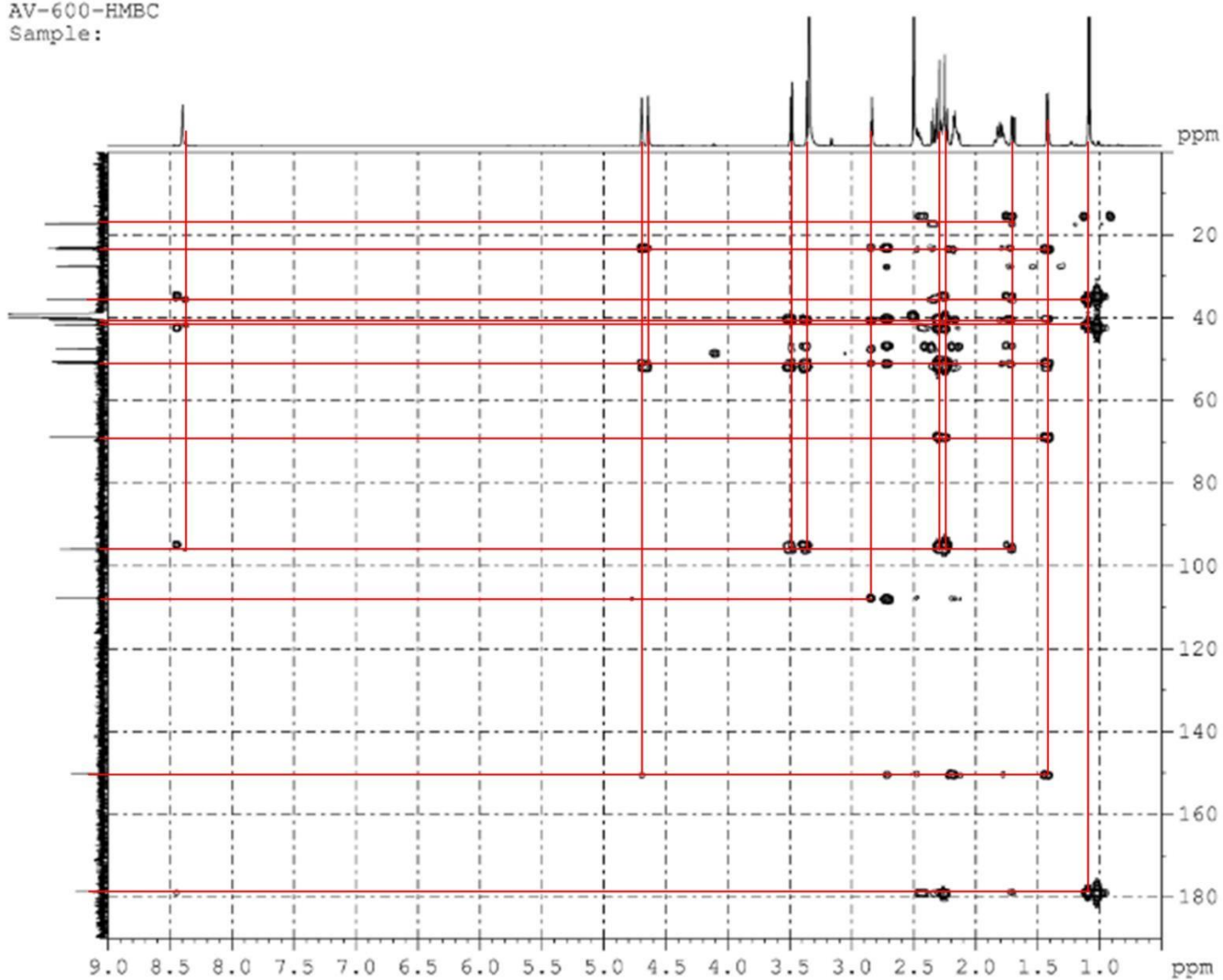
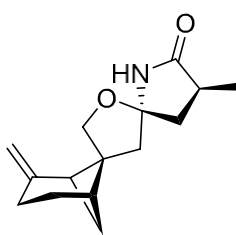




Figure S27. NOESY spectrum (600 MHz, DMSO-*d*<sub>6</sub>) of compound 2



sporulosine B (2)

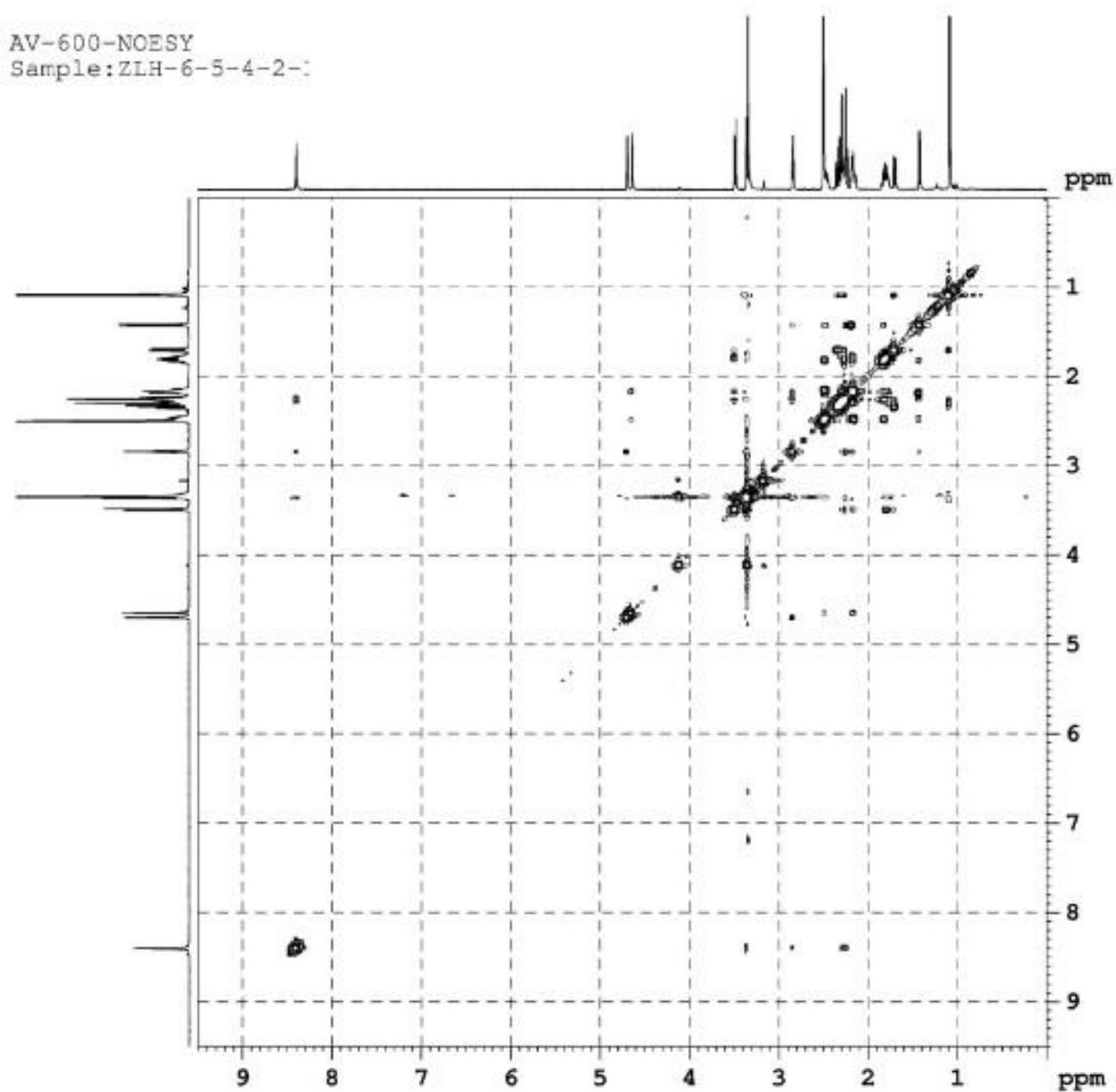
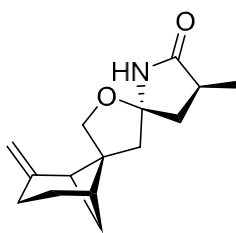


Figure S28. IR spectrum of compound 2



sporulosine B (2)

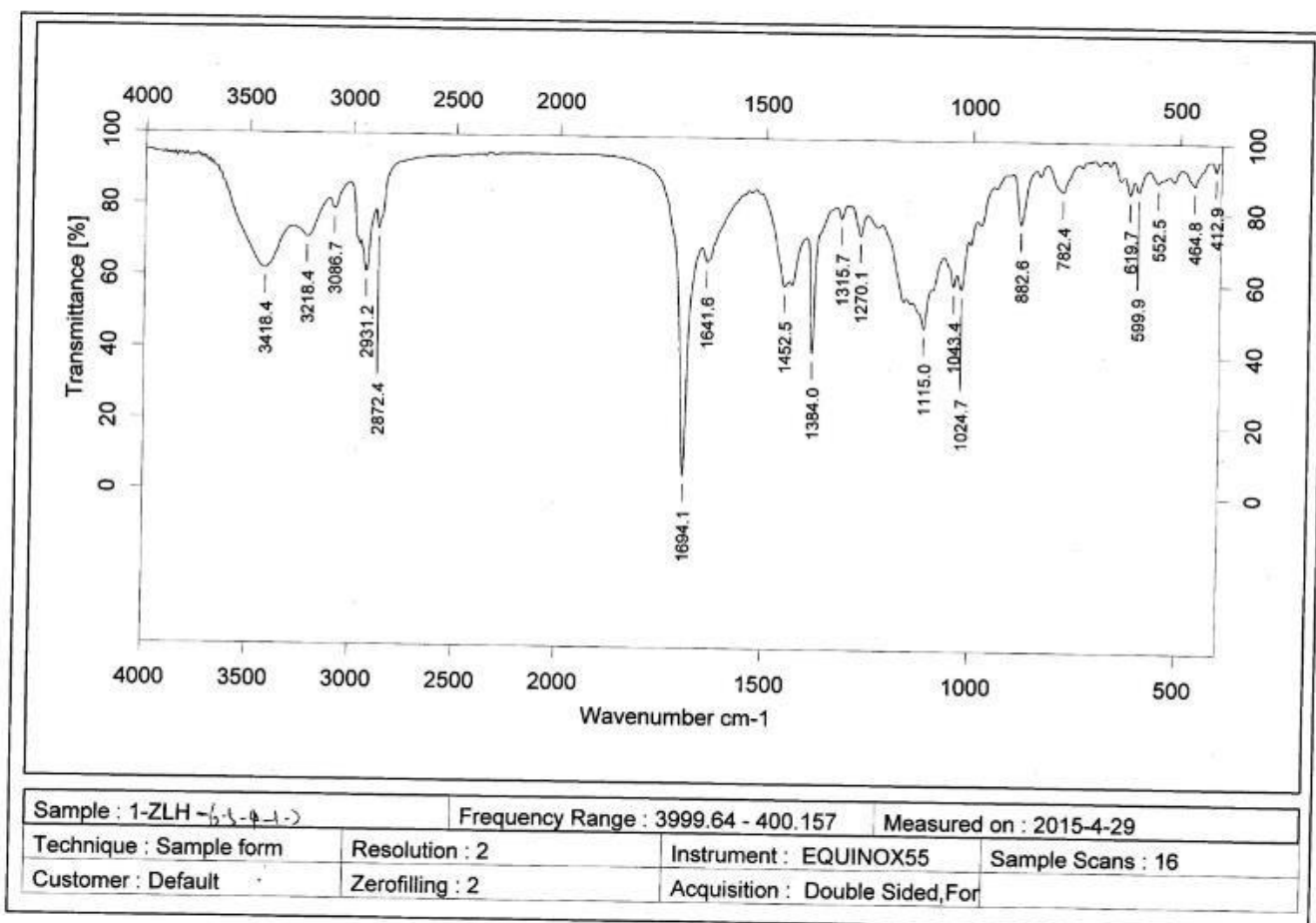
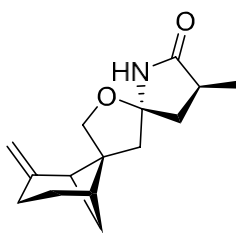
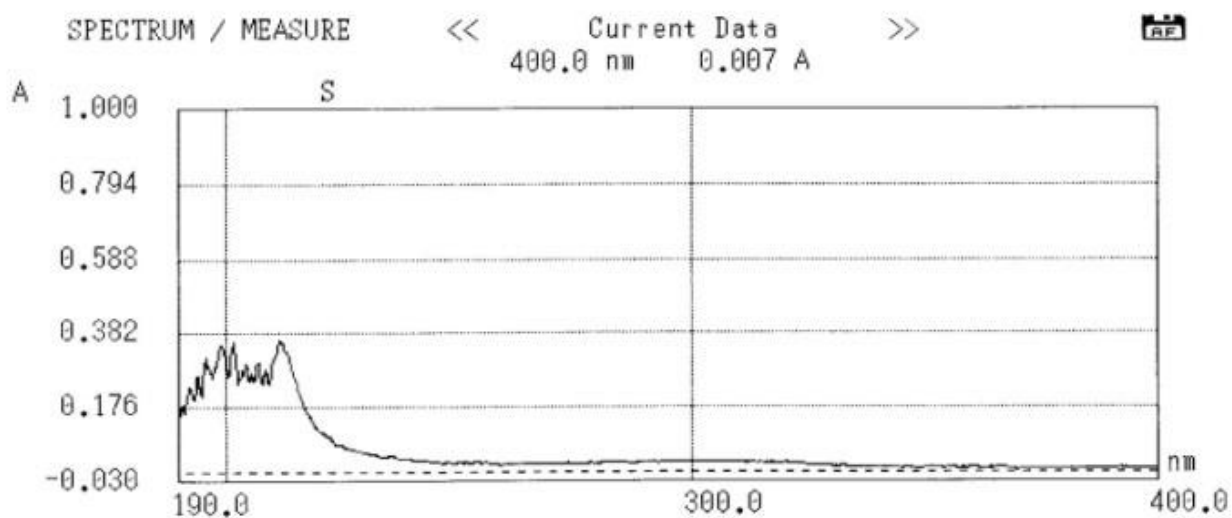


Figure S29. UV spectrum of compound 2



sporulosine B (2)

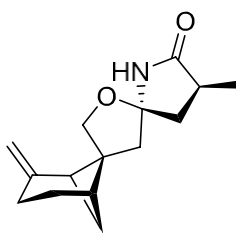


> Set sample then press [Start Scan].

⊕Enlarge/Reduce File (SP) Rec. Output Set Parameters Start Scan

NO.	ABSCISSA	PEAK	HEIGHT	ABSCISSA	VALLEY	HEIGHT
1	211.4	0.3604	0.1474	207.8	0.2423	-0.1171

**Figure S30. CD spectrum of compound 2**



sporulosine B (2)

