

# **Structural and Electronic Properties of the PbCrO<sub>4</sub> Chrome Yellow Pigment and of Its Light Sensitive Sulfate-Substituted Compounds**

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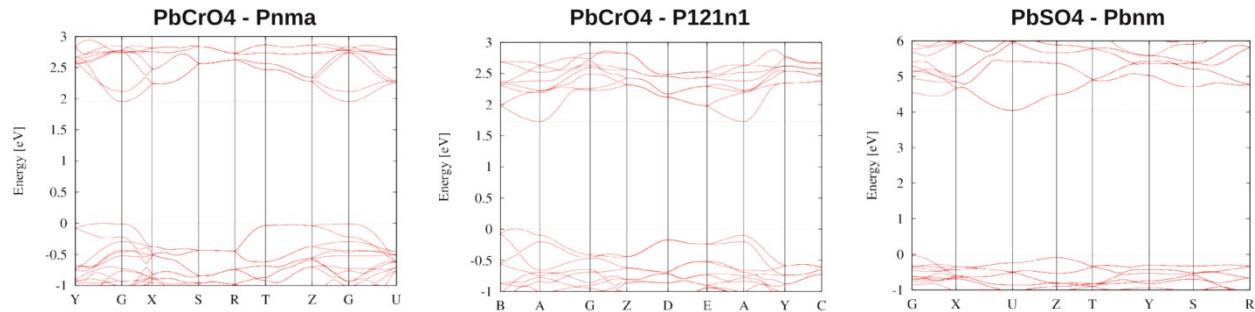
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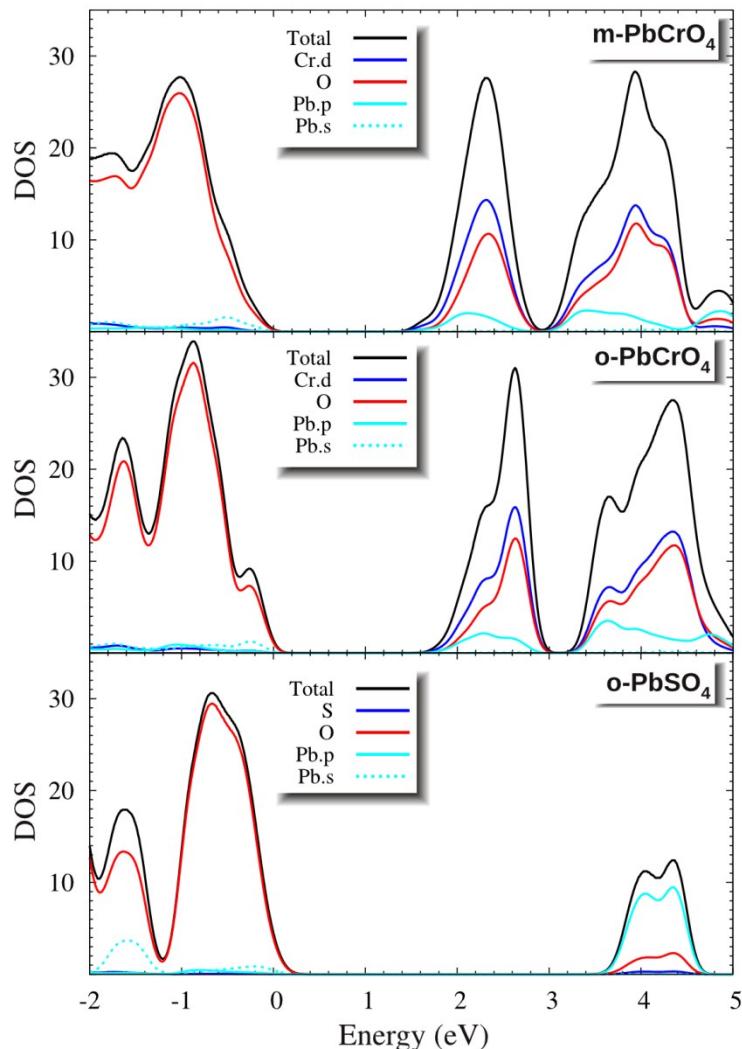
## **SUPPORTING INFORMATION**

**Table S1.** Cell parameters of the experimental and optimized cell structures of m-PbCrO<sub>4</sub>, o-PbCrO<sub>4</sub> and the o-PbSO<sub>4</sub>.

	Experimental			Optimized		
	m-PbCrO <sub>4</sub>	o--PbCrO <sub>4</sub>	o-PbSO <sub>4</sub>	m-PbCrO <sub>4</sub>	o-PbCrO <sub>4</sub>	o-PbSO <sub>4</sub>
<b>a (Å)</b>	7.120	8.799	6.958	7.104	8.195	7.124
<b>b (Å)</b>	7.430	5.734	8.480	7.427	5.705	8.632
<b>c (Å)</b>	6.790	7.274	5.398	6.762	7.478	5.530
<b>alpha (deg.)</b>	90	90	90	89.9	90	90
<b>beta (deg.)</b>	102.4	90	90	102.5	90	90
<b>gamma (deg.)</b>	90	90	90	90.1	90	90
<b>Volume (Å<sup>3</sup>)</b>	350.796	366.997	318.486	348.423	349.589	317.337



**Figure S1.** Band structures computed by SR-DFT on the experimental cell structures of m-PbCrO<sub>4</sub>, o-PbCrO<sub>4</sub> and the o-PbSO<sub>4</sub>. The zero has been set to the top of the valence band.



**Figure S2.** Total density of states for m-PbCrO<sub>4</sub> (top), o-PbCrO<sub>4</sub> (middle) and the o-PbSO<sub>4</sub> (bottom) structures at the experimental cell parameters. Contribution of the main orbitals to the valence and conduction bands. The DOS have been aligned using the Pb.d peaks (spin down) and the zero has been set to the top of the m-PbCrO<sub>4</sub> valence band.

**Table S2.** Relative stabilities (kcal/mol of the mixed PbCr<sub>0.5</sub>S<sub>0.5</sub>O<sub>4</sub> species with respect to the stablest one within the monoclinic or orthorhombic phase

	PbCr <sub>0.5</sub> S <sub>0.5</sub> O <sub>4</sub> (a)	PbCr <sub>0.5</sub> S <sub>0.5</sub> O <sub>4</sub> (b)	PbCr <sub>0.5</sub> S <sub>0.5</sub> O <sub>4</sub> (c)
Monoclinic	0.68	0.02	0.00
Orthorhombic	0.81	1.13	0.00

**Table S3.** High conduction band composition for the monoclinic species.

x=0            x=0.25            x=0.5(c)            x=0.75            x=1

Int. Limits	<b>3.01</b>	<b>3.00</b>	<b>3.03</b>	<b>3.06</b>	<b>3.16</b>
(eV)	<b>5.97</b>	<b>5.75</b>	<b>5.58</b>	<b>5.35</b>	<b>5.06</b>
Cr.1s					
Cr.2s					
Cr.d	<b>45</b>	<b>36</b>	<b>31</b>	<b>22</b>	
Cr.p					
OCr.p	<b>35</b>	<b>31</b>	<b>26</b>	<b>18</b>	
OCr.s	<b>5</b>	<b>5</b>	<b>3</b>	<b>3</b>	
OS.p			<b>3</b>	<b>5</b>	<b>11</b>
OS.s			<b>2</b>	<b>3</b>	<b>8</b>
Pb.d					
Pb.p	<b>11</b>	<b>23</b>	<b>31</b>	<b>44</b>	<b>75</b>
Pb.s	<b>1</b>	<b>1</b>	<b>1</b>	<b>1</b>	
S.p			<b>1</b>	<b>1</b>	<b>3</b>
S.s					<b>1</b>

**Table S4.** Low conduction band composition for the monoclinic species.

	<b>x=0</b>	<b>x=0.25</b>	<b>x=0.5(c)</b>	<b>x=0.75</b>	<b>x=1</b>
Int. Limits	<b>1.27</b>	<b>1.27</b>	<b>1.38</b>	<b>1.41</b>	
(eV)	<b>3.01</b>	<b>3.01</b>	<b>3.03</b>	<b>3.06</b>	
Cr.1s					
Cr.2s					
Cr.d	<b>52</b>	<b>52</b>	<b>51</b>	<b>51</b>	
Cr.p					
OCr.p	<b>36</b>	<b>36</b>	<b>35</b>	<b>34</b>	
OCr.s	<b>1</b>	<b>1</b>	<b>1</b>		
OS.p			<b>1</b>	<b>2</b>	
OS.s					
Pb.d					
Pb.p	<b>8</b>	<b>8</b>	<b>8</b>	<b>9</b>	
Pb.s					
S.p					
S.s					

**Table S5.** High valence band composition for the monoclinic species.

	<b>x=0</b>	<b>x=0.25</b>	<b>x=0.5(c)</b>	<b>x=0.75</b>	<b>x=1</b>
Int. Limits	<b>-1.45</b>	<b>-1.39</b>	<b>-1.34</b>	<b>-1.35</b>	<b>-1.26</b>
(eV)	<b>0.40</b>	<b>0.41</b>	<b>0.48</b>	<b>0.51</b>	<b>1.06</b>
Cr.1s					
Cr.2s					
Cr.d	<b>2</b>	<b>2</b>	<b>1</b>	<b>1</b>	
Cr.p					
OCr.p	<b>89</b>	<b>84</b>	<b>42</b>	<b>19</b>	
OCr.s	<b>1</b>	<b>1</b>			
OS.p		<b>6</b>	<b>49</b>	<b>72</b>	<b>93</b>
OS.s					
Pb.d					
Pb.p	<b>2</b>	<b>2</b>	<b>2</b>	<b>2</b>	<b>2</b>
Pb.s	<b>5</b>	<b>5</b>	<b>5</b>	<b>5</b>	<b>4</b>
S.p					
S.s					

**Table S6.** Low valence band composition for the monoclinic species.

	x=0	x=0.25	x=0.5(c)	x=0.75	x=1
Int. Limits (eV)	<b>-5.27</b> <b>-1.45</b>	<b>-5.14</b> <b>-1.39</b>	<b>-4.92</b> <b>-1.34</b>	<b>-4.75</b> <b>-1.35</b>	<b>-4.30</b> <b>-1.26</b>
Cr.1s					
Cr.2s					
Cr.d	<b>16</b>	<b>17</b>	<b>13</b>	<b>7</b>	
Cr.p					
OCr.p	<b>75</b>	<b>69</b>	<b>45</b>	<b>26</b>	
OCr.s	<b>2</b>	<b>2</b>	<b>1</b>	<b>1</b>	
OS.p		<b>3</b>	<b>31</b>	<b>54</b>	<b>86</b>
OS.s				<b>1</b>	<b>1</b>
Pb.d					
Pb.p	<b>3</b>	<b>4</b>	<b>4</b>	<b>4</b>	<b>4</b>
Pb.s	<b>3</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>4</b>
S.p			<b>1</b>	<b>1</b>	<b>1</b>
S.s					

**Table S7.** High conduction band composition for the orthorombic species.

	x=0	x=0.25	x=0.5(c)	x=0.75	x=1
Int. Limits (eV)	<b>3.44</b> <b>6.04</b>	<b>3.28</b> <b>5.38</b>	<b>3.14</b> <b>5.24</b>	<b>3.17</b> <b>5.23</b>	
Cr.1s					
Cr.2s					
Cr.d	<b>44</b>	<b>21</b>	<b>21</b>	<b>22</b>	
Cr.p					
OCr.p	<b>34</b>	<b>18</b>	<b>18</b>	<b>18</b>	
OCr.s	<b>5</b>	<b>2</b>	<b>2</b>	<b>2</b>	
OS.p		<b>4</b>	<b>4</b>	<b>5</b>	
OS.s		<b>2</b>	<b>2</b>	<b>3</b>	
Pb.d					
Pb.p	<b>13</b>	<b>44</b>	<b>44</b>	<b>45</b>	
Pb.s	<b>1</b>				
S.p				<b>1</b>	
S.s					

**Table S8.** Low conduction band composition for the orthorombic species.

	x=0	x=0.25	x=0.5(c)	x=0.75	x=1
Int. Limits (eV)	<b>1.67</b> <b>3.09</b>	<b>2.08</b> <b>3.28</b>	<b>1.94</b> <b>3.14</b>	<b>1.53</b> <b>3.17</b>	<b>2.69</b> <b>4.59</b>
Cr.1s					
Cr.2s					
Cr.d	<b>51</b>	<b>51</b>	<b>51</b>	<b>51</b>	
Cr.p					
OCr.p	<b>36</b>	<b>33</b>	<b>33</b>	<b>33</b>	
OCr.s	<b>1</b>				
OS.p		<b>2</b>	<b>2</b>	<b>2</b>	<b>11</b>
OS.s					<b>8</b>
Pb.d					
Pb.p	<b>9</b>	<b>10</b>	<b>10</b>	<b>10</b>	<b>75</b>
Pb.s					

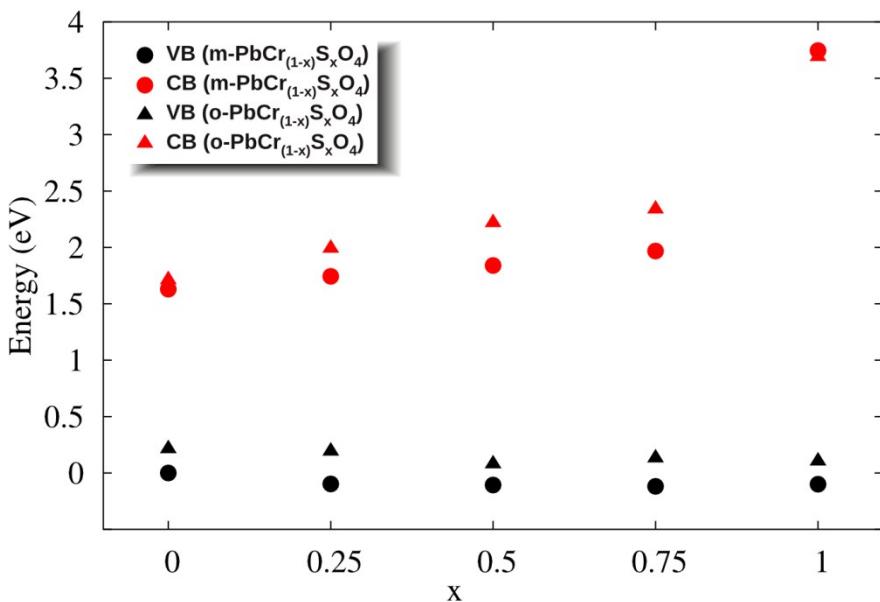
S.p					<b>3</b>
S.s					<b>1</b>

**Table S9.** Low valence band composition for the orthorombic species.

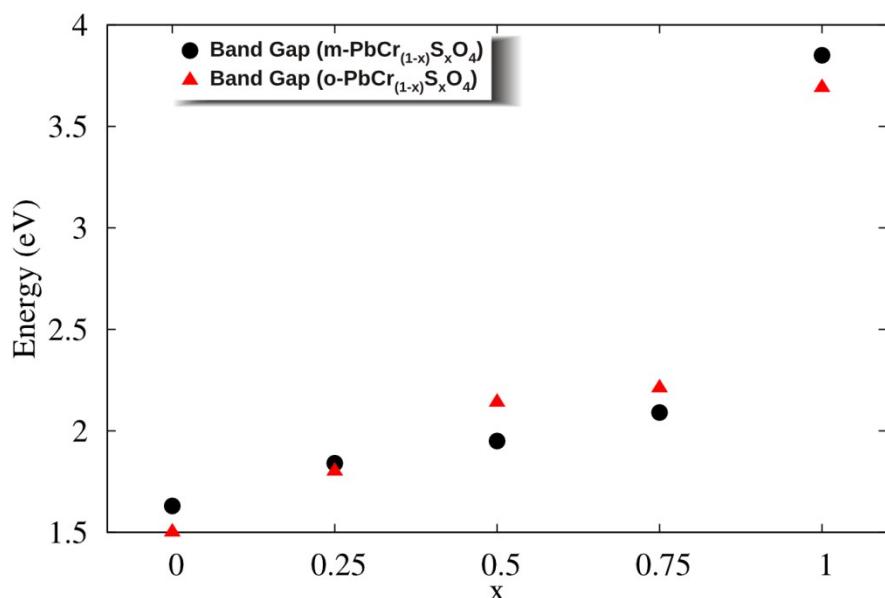
	<b>x=0</b>	<b>x=0.25</b>	<b>x=0.5(c)</b>	<b>x=0.75</b>	<b>x=1</b>
Int. Limits	<b>-0.26</b>	<b>-1.11</b>	<b>-1.25</b>	<b>-1.26</b>	<b>-1.73</b>
(eV)	<b>0.64</b>	<b>0.78</b>	<b>0.64</b>	<b>0.63</b>	<b>0.59</b>
Cr.1s					
Cr.2s					
Cr.d	<b>4</b>				
Cr.p					
OCr.p	<b>74</b>	<b>21</b>	<b>21</b>	<b>21</b>	
OCr.s	<b>1</b>				
OS.p		<b>72</b>	<b>72</b>	<b>72</b>	<b>93</b>
OS.s					
Pb.d					
Pb.p	<b>2</b>	<b>1</b>	<b>1</b>	<b>1</b>	<b>2</b>
Pb.s	<b>17</b>	<b>3</b>	<b>3</b>	<b>3</b>	<b>4</b>
S.p					
S.s					

**Table S10.** High valence band composition for the orthorombic species.

	<b>x=0</b>	<b>x=0.25</b>	<b>x=0.5(c)</b>	<b>x=0.75</b>	<b>x=1</b>
Int. Limits	<b>-5.46</b>	<b>-4.42</b>	<b>-4.56</b>	<b>-4.61</b>	<b>-4.77</b>
(eV)	<b>-0.26</b>	<b>-1.11</b>	<b>-1.25</b>	<b>-1.26</b>	<b>-1.73</b>
Cr.1s					
Cr.2s					
Cr.d	<b>16</b>	<b>7</b>	<b>7</b>	<b>7</b>	
Cr.p					
OCr.p	<b>75</b>	<b>25</b>	<b>25</b>	<b>25</b>	
OCr.s	<b>2</b>	<b>1</b>	<b>1</b>	<b>1</b>	
OS.p		<b>54</b>	<b>54</b>	<b>54</b>	<b>86</b>
OS.s		<b>1</b>	<b>1</b>	<b>1</b>	<b>1</b>
Pb.d					
Pb.p	<b>3</b>	<b>4</b>	<b>4</b>	<b>4</b>	<b>4</b>
Pb.s	<b>2</b>	<b>4</b>	<b>4</b>	<b>4</b>	<b>4</b>
S.p		<b>1</b>	<b>1</b>	<b>1</b>	<b>1</b>
S.s					



**Figure S3.** Valence and conduction band for the monoclinic and orthorhombic mixed species.



**Figure S4.** Band gap for the monoclinic and orthorhombic mixed species

**Table S11.** Average distances ( $\text{\AA}$ ), volumes ( $\text{\AA}^3$ ) and effective coordination numbers for the  $\text{CrO}_4^{2-}$  and  $\text{SO}_4^{2-}$  tetrahedra and the  $\text{PbO}_9$  polyhedra for the monoclinic structures.

		<b>x=0</b>	<b>x=0.25</b>	<b>x=0.5 (a)</b>	<b>x=0.75</b>
$\text{CrO}_4^{2-}$	<b>CrO dist. (<math>\text{\AA}</math>)</b>	1.654	1.654	1.654	1.654
	<b>Pol volume (<math>\text{\AA}^3</math>)</b>	2.315	2.314	2.309	2.311
	<b>Coord. Number</b>	3.997	3.997	3.998	3.998
$\text{SO}_4^{2-}$	<b>SO dist. (<math>\text{\AA}</math>)</b>	-	1.493	1.494	1.493
	<b>Pol volume (<math>\text{\AA}^3</math>)</b>	-	1.705	1.707	1.703

	<b>Coord. Number</b>	-	3.993	3.996	3.995
<b>PbO<sub>9</sub></b>	<b>PbO dist. (Å)</b>	2.675	2.677	2.679	2.685
	<b>Pol volume (Å<sup>3</sup>)</b>	37.404	37.505	37.632	37.909
	<b>Coord. Number</b>	7.562	7.709	7.949	8.104
	<b>Val. Band (eV)</b>	0.00	-0.10	-0.11	-0.12
	<b>Cond. Band (eV)</b>	1.63	1.74	1.84	1.97
	<b>B-gap (eV)</b>	1.63	1.84	1.98	2.09

**Table S12** Average distances (Å), volumes (Å<sup>3</sup>) and effective coordination numbers for the CrO<sub>4</sub><sup>2-</sup> and SO<sub>4</sub><sup>2-</sup> tetrahedra and the PbO<sub>x</sub> polyhedra for the orthorhombic structures.

		<b>x=0</b>	<b>x=0.25</b>	<b>x=0.5 (a)</b>	<b>x=0.75</b>
<b>CrO<sub>4</sub><sup>2-</sup></b>	<b>CrO dist. (Å)</b>	1.653	1.652	1.651	1.650
	<b>Pol volume (Å<sup>3</sup>)</b>	2.315	2.311	2.308	2.303
	<b>Coord. Number</b>	3.984	3.985	3.989	3.987
<b>SO<sub>4</sub><sup>2-</sup></b>	<b>SO dist. (Å)</b>	-	1.492	1.492	1.492
	<b>Pol volume (Å<sup>3</sup>)</b>	-	1.704	1.701	1.702
	<b>Coord. Number</b>	-	3.993	3.978	3.989
<b>PbO<sub>9</sub></b>	<b>PbO dist. (Å)</b>	2.541	2.707	2.802	2.788
	<b>Pol volume (Å<sup>3</sup>)</b>	19.102	29.564	42.860	42.456
	<b>Coord. Number</b>	5.688	5.896	6.179	7.183
	<b>Val. Band (eV)</b>	0.21	0.19	0.08	0.13
	<b>Cond. Band (eV)</b>	1.71	1.99	2.22	2.34
	<b>B-gap (eV)</b>	1.50	1.80	2.14	2.21