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Supporting Information

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for

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Enhancing low pressure CO₂ adsorption of solvent-free derived mesoporous carbon by highly dispersed potassium species

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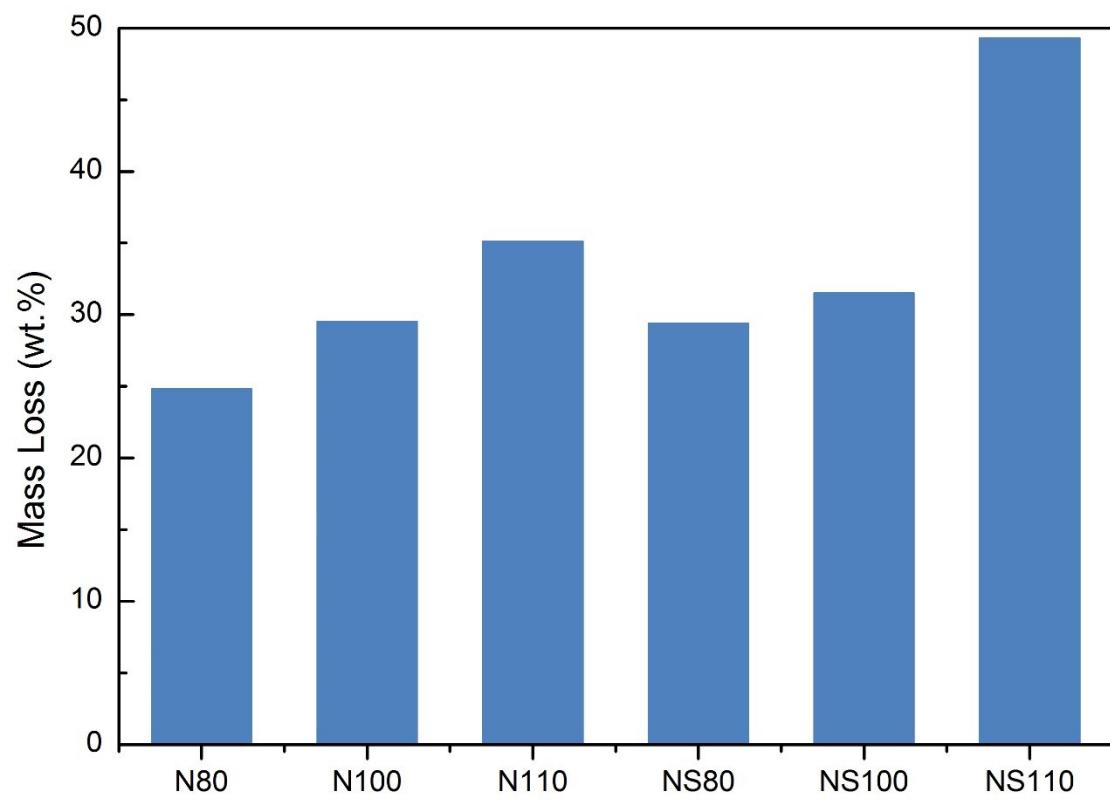


Fig. S1 Mass loss of oxidized MC at 1000°C

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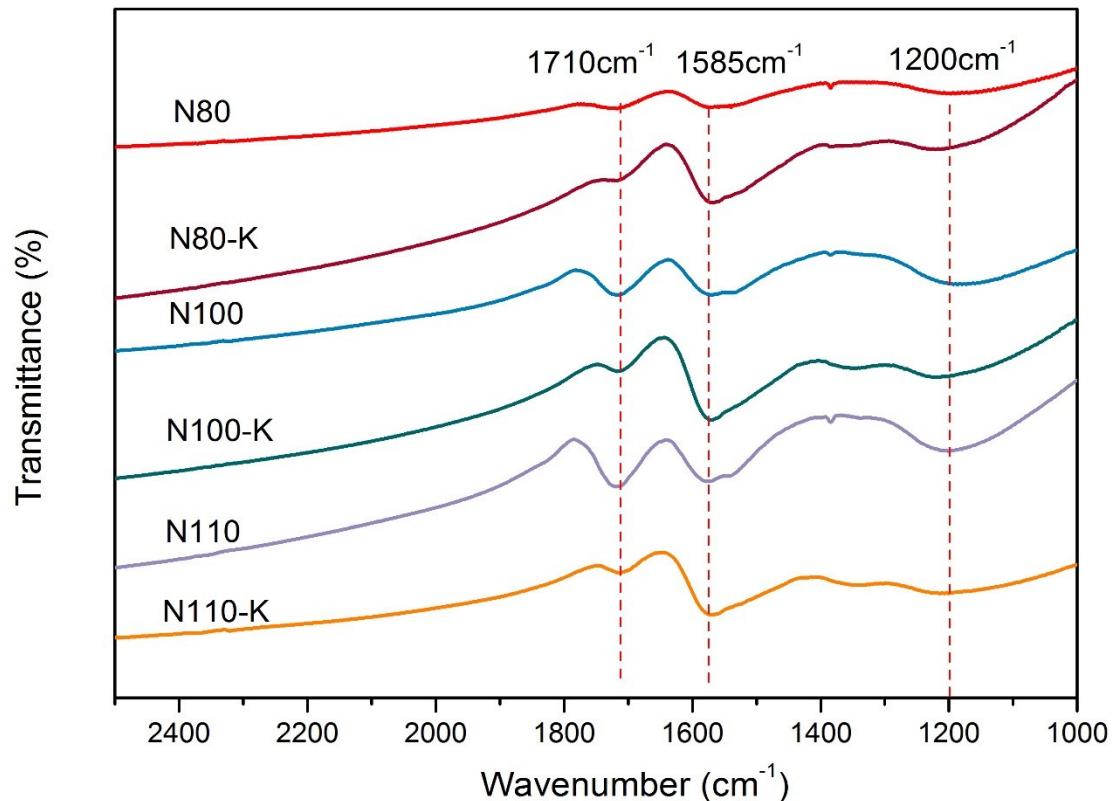


Fig. S2 FT-IR spectra of Nx samples

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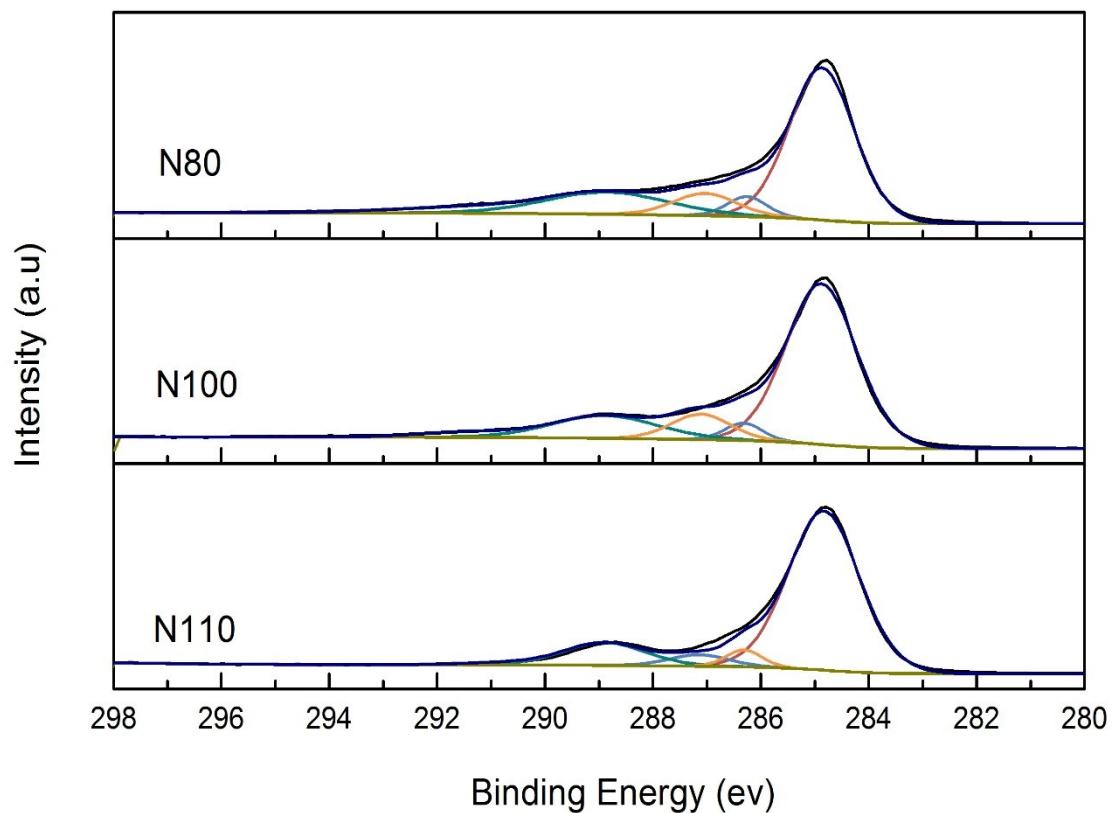


Fig. S3 C1s XPS spectra of Nx samples

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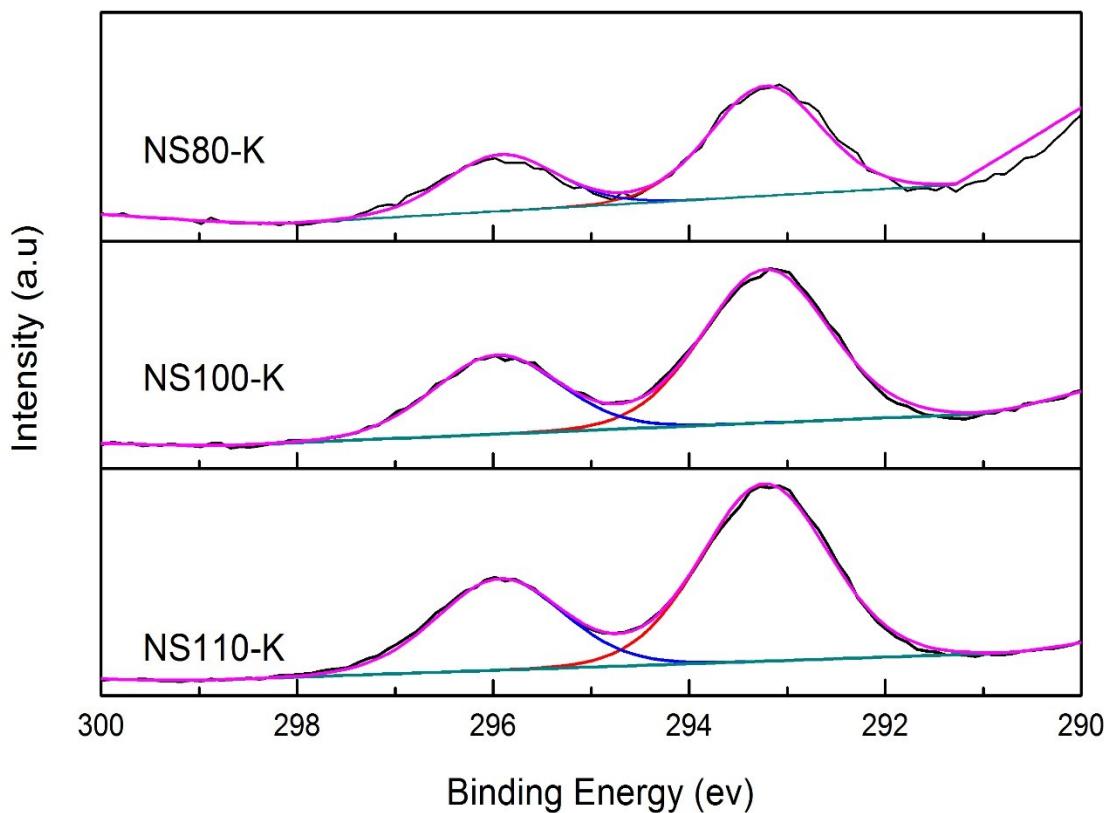
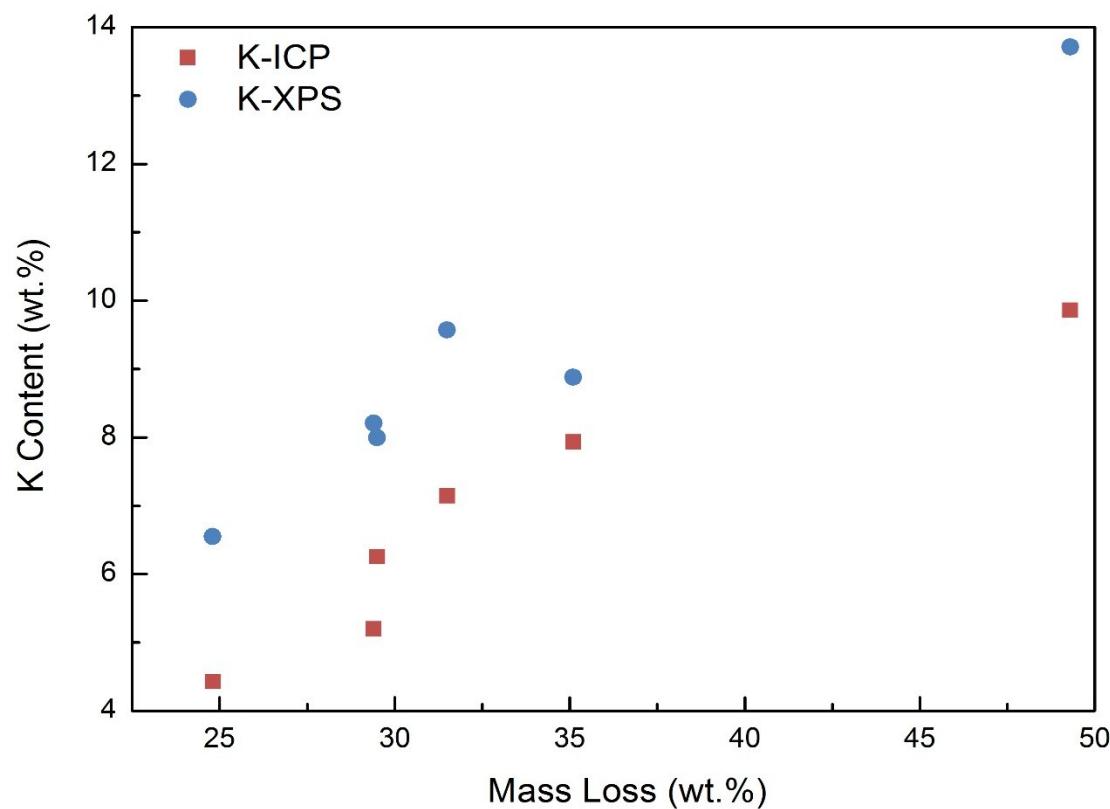


Fig. S4 K2p XPS spectra of NS_x-K

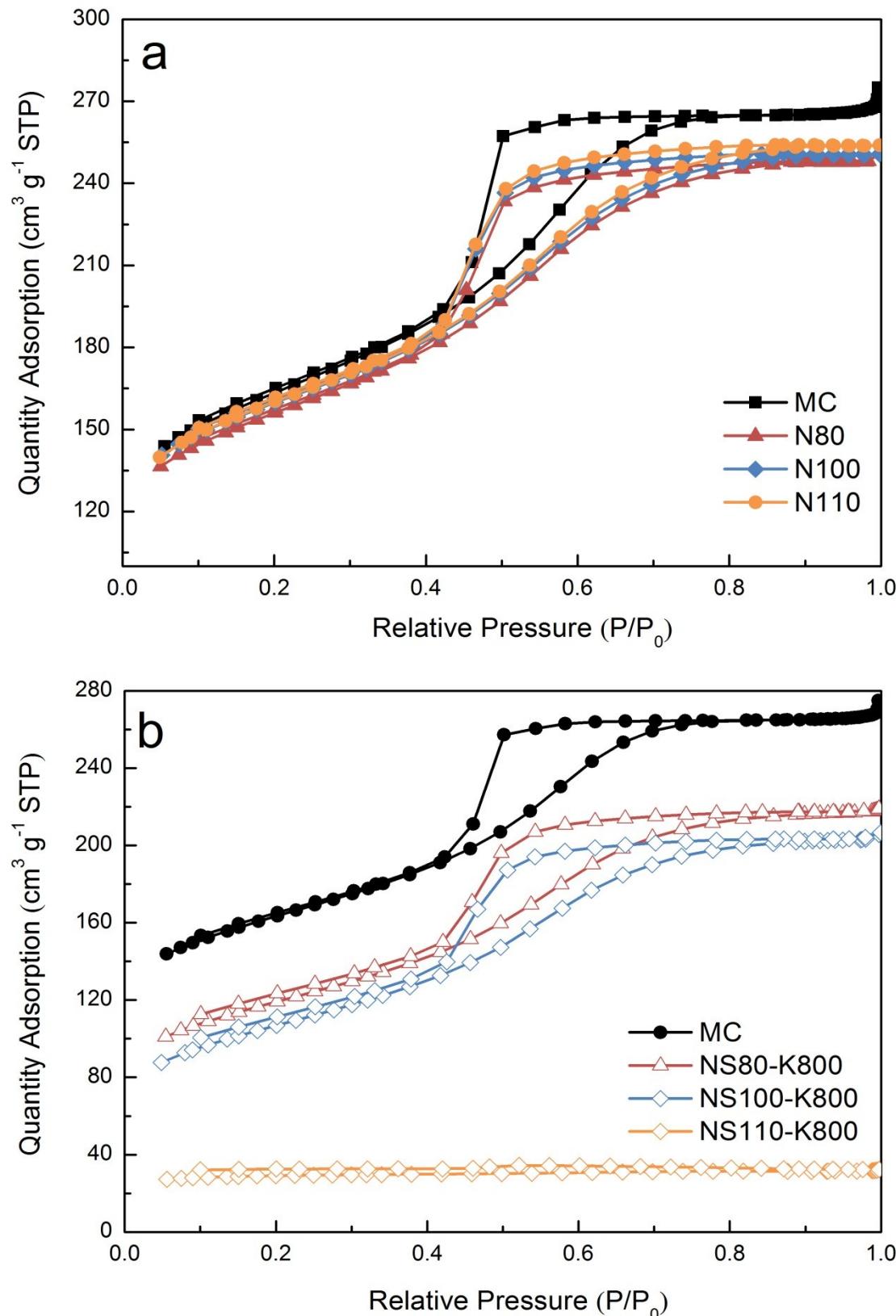
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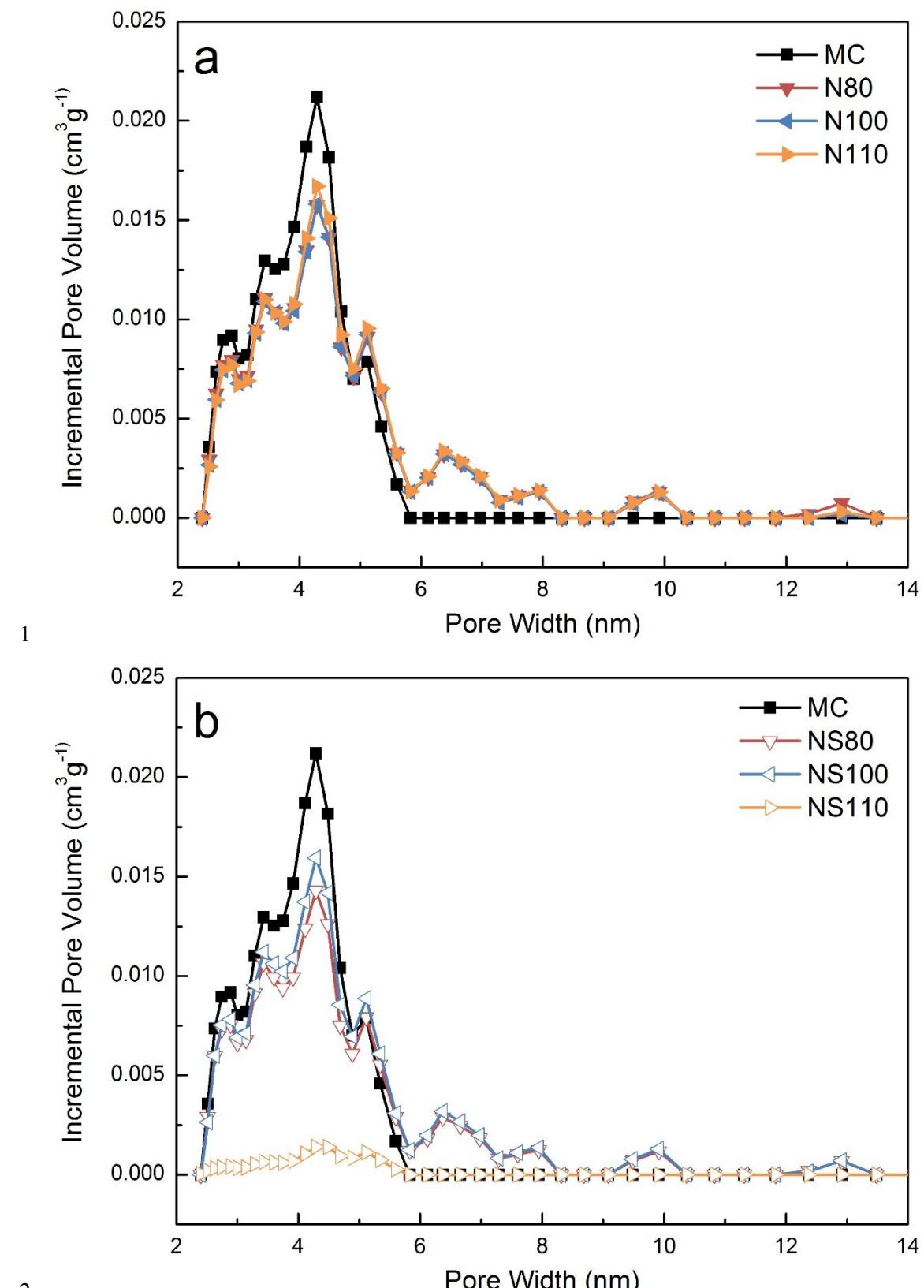
2 Fig. S5 The relationship between potassium loading and mass loss of the oxidized counterparts

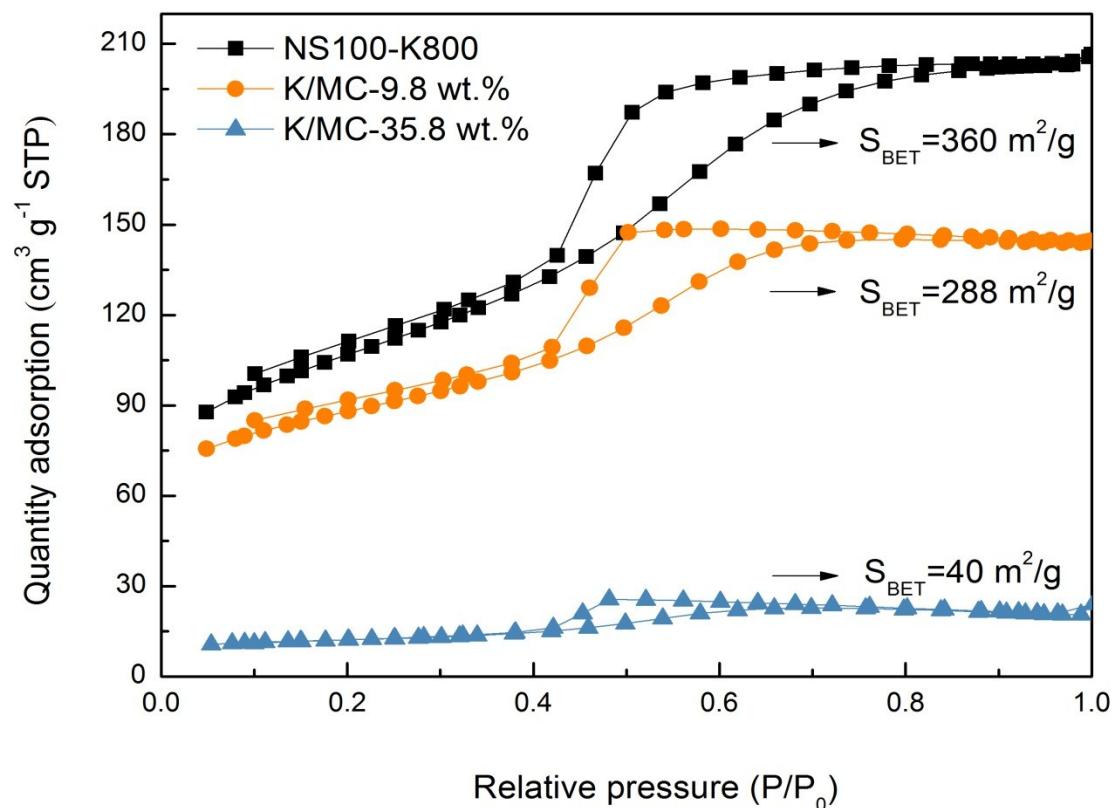
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3 Fig. S6 Low temperature (-196 °C) N₂ isotherms. (a) MC and Nx series, (b) MC and NSx series

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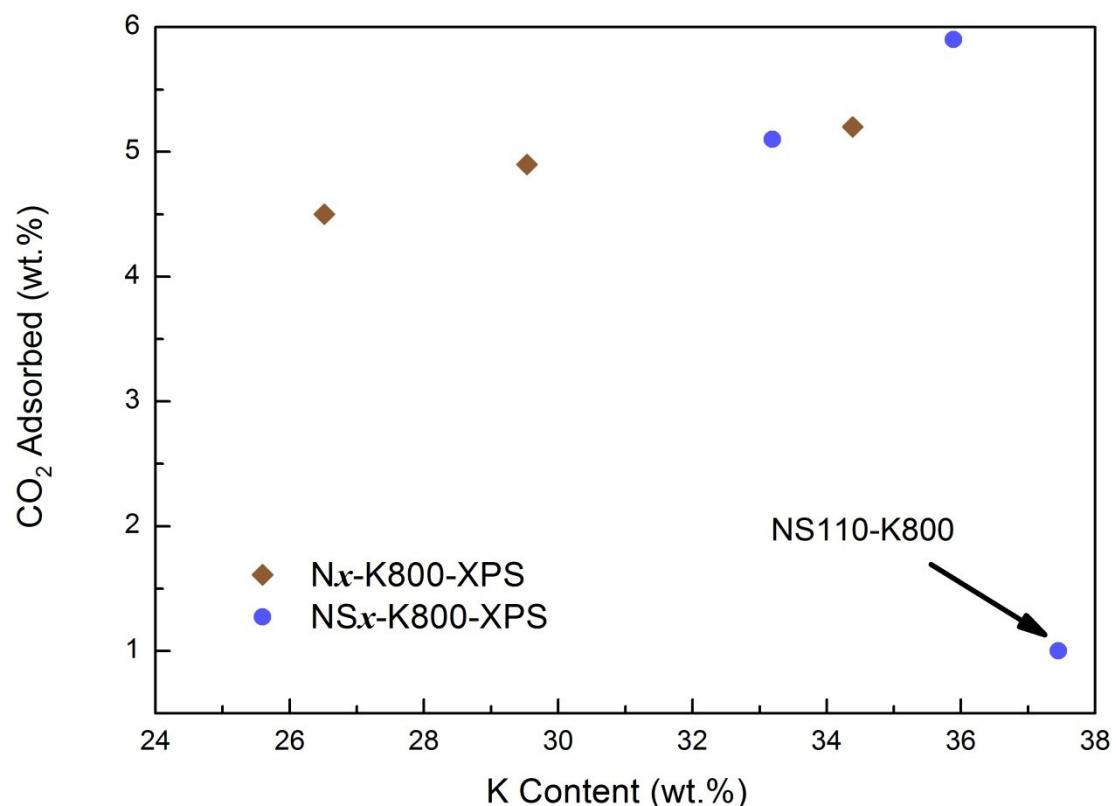


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2 Fig. S8 Comparison of N₂ isotherm and surface area of impregnated sample with NS100-K800

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Fig. S9 Relationship between surface K content and CO₂ uptake (0.15bar)

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1 Table S1 Comparison on the adsorption capacities of NS100-K800 and other carbon adsorbents

Materials	Precursors	S_{BET} (m ² /g)	CO ₂ uptake (wt.%) (25 °C, 0.15 bar)	Ref.
NS100-K800	Phenolic resin	360	5.9	This work
KNC-A-K	p-diaminobenzene	614	7.9	¹
C-CA-6-KOH	Phenolic resin	1709	3.9	²
PR3_700	Phenolic resin	826	6.6	³
CTNC-C800	Polyacry- lonitrile-block-polymer	-	5.1	⁴
CEM-750	-	3360	4.3	⁵
PA-400-KOH-1-600	Biomass	837	6.1	⁶
C-KOH	-	2030	4.0	⁷
NMC250D8H	Melamine phenolic resin	385	5.3	⁸
MR-1-500	Melamine Phenolic resin	1286	5.4	⁹
VR-93	Petroleum pitch	2895	4.5	¹⁰

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1 **Reference**

- 2
- 3 1. Y. F. Zhao, X. Liu, K. X. Yao, L. Zhao and Y. Han, *Chem. Mat.*, 2012, 24, 4725-4734.
- 4 2. J. Choma, K. Jedynak, W. Fahrenholz, J. Ludwinowicz and M. Jaroniec, *Applied Surface*
5 *Science*, 2014, 289, 592-600.
- 6 3. J. J. Liu, N. N. Sun, C. G. Sun, H. Liu, C. Snape, K. X. Li, W. Wei and Y. H. Sun, *Carbon*,
7 2015, 94, 243-255.
- 8 4. M. J. Zhong, S. Natesakhawat, J. P. Baltrus, D. Luebke, H. Nulwala, K. Matyjaszewski and T.
9 Kowalewski, *Chem. Commun.*, 2012, 48, 11516-11518.
- 10 5. Y. D. Xia, R. Mokaya, G. S. Walker and Y. Q. Zhu, *Adv. Energy Mater.*, 2011, 1, 678-683.
- 11 6. J. Song, W. Z. Shen, J. G. Wang and W. B. Fan, *Carbon*, 2014, 69, 255-263.
- 12 7. J. Ludwinowicz and M. Jaroniec, *Carbon*, 2015, 94, 673-679.
- 13 8. Z. Z. Zhang, B. D. Wang, C. M. Zhu, P. Gao, Z. Y. Tang, N. N. Sun, W. Wei and Y. H. Sun,
14 *Journal of Materials Chemistry A*, 2015, 3, 23990-23999.
- 15 9. H. Cong, M. Zhang, Y. Chen, K. Chen, Y. Hao, Y. Zhao and L. Feng, *Carbon*, 2015, 92, 297-
16 304.
- 17 10. J. Silvestre-Albero, A. Wahby, A. Sepulveda-Escribano, M. Martinez-Escandell, K. Kaneko
18 and F. Rodriguez-Reinoso, *Chem. Commun.*, 2011, 47, 6840-6842.
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