Electronic Supplementary Information

for

Nanostructured charge transfer complex of $CuTCNQF_4$

for efficient photo-removal of hexavalent chromium

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Fig S1. FTIR spectra of pristine TCNQF₄ and CuTCNQF₄ obtained after the reaction of Cu foil with TCNQF₄ in acetonitrile.



Fig S2. XPS spectra of pristine TCNQF₄ powder showing C 1s, N 1s and F 1s core levels.



Fig S3. Diffuse reflectance spectra of pristine TCNQF₄ and CuTCNQF₄ grown on Cu foil.



Fig S4. Plot of the percentage reduction of Cr⁶⁺ under UV light irradiation for 30 min by CuTCNQF₄

over multiple cycles.



Fig S5. SEM image of $CuTCNQF_4$ obtained (a) before and (b) after the catalytic reaction.



Fig S6. Raman spectra of CuTCNQF₄ obtained (a) before and (b) after the catalytic reaction.

Reaction conditions	Reaction rate (min ⁻¹)
pH 2 / No donor / 25 °C / Dark CuTCNQF ₄	4.8 x 10 ⁻⁴
pH 2 / Methanol / 25 °C / Dark CuTCNQF4	2.5 x 10 ⁻³
pH 2 / Acetic acid / 25 °C / Dark CuTCNQF4	2.0×10^{-3}
pH 7 / Methanol / 25 °C / Dark CuTCNQF ₄	8.2 x 10 ⁻⁴
pH 3 / Methanol / 25 °C / Dark CuTCNQF4	3.5 x 10 ⁻⁴
pH 2 / Methanol / 25 °C / Dark CuTCNQF4	2.5 x 10 ⁻³
$pH~1$ / Methanol / 25 $^{\circ}C$ / Dark CuTCNQF4	2.4 x 10 ⁻³
pH 2 / Methanol / 5 °C / Dark CuTCNQF ₄	6.1 x 10 ⁻⁴
pH 2 / Methanol / 25 °C / Dark CuTCNQF4	2.5 x 10 ⁻³
pH 2 / Methanol / 50 °C / Dark CuTCNQF4	2.3×10^{-3}
pH 2 / Methanol / 25 °C / Control – No catalyst	7.6 x 10 ⁻⁴
pH 2 / Methanol / 25 °C / Dark CuTCNQF4	3.4 x 10 ⁻³
pH 2 / Methanol / 25 °C / UV CuTCNQF4	3.6 x 10 ⁻²

Table S1. Rates of reduction of Cr^{6+} ions under different reaction conditions.