

Supplementary material

Coordination chemistry of pyrazine derivatives analogues of PZA: design, synthesis, characterization and biological activity.

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Keywords: Ru(III) complexes; Pyrazine derivatives; Spectrophotometric titration; Cyclic voltammetry; Biological assay

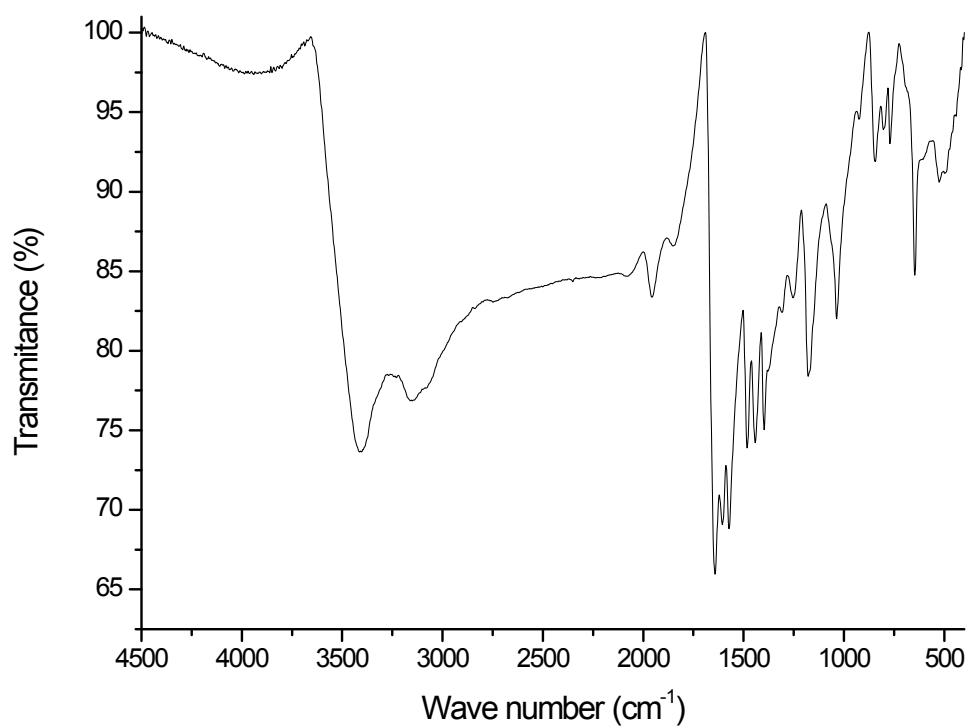


Fig.S1. FT-IR spectrum of $[\text{RuCl}(\text{PAOX})_2(\text{OH}_2)]\text{Cl}_2$.

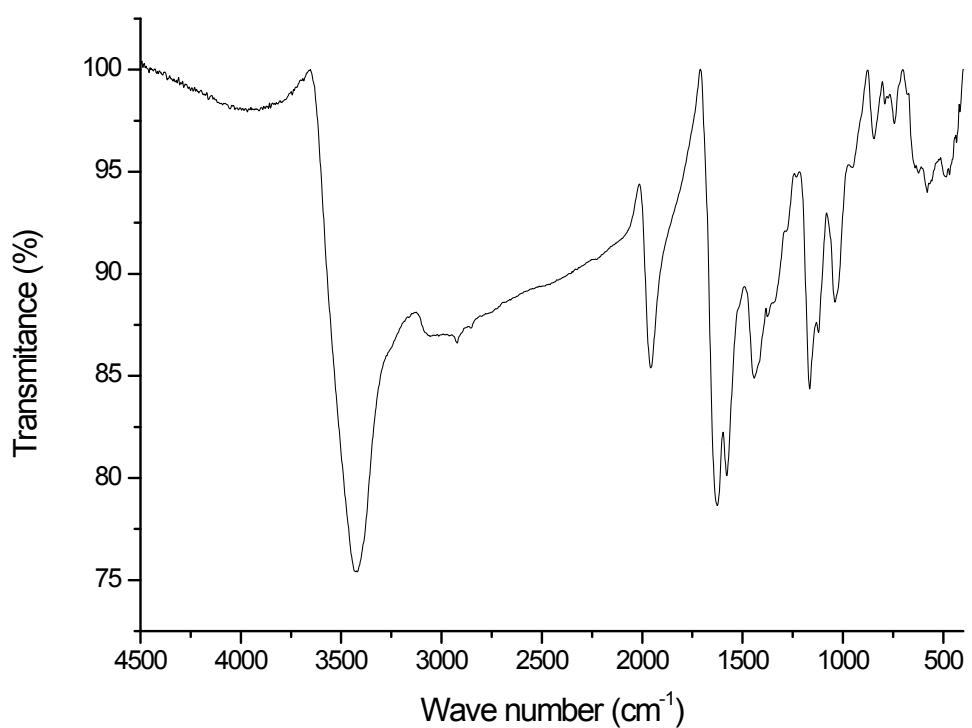


Fig. S2. FT-IR spectrum of $[\text{RuCl}(\text{PTCA})_2(\text{OH}_2)]\text{Cl}_2$.

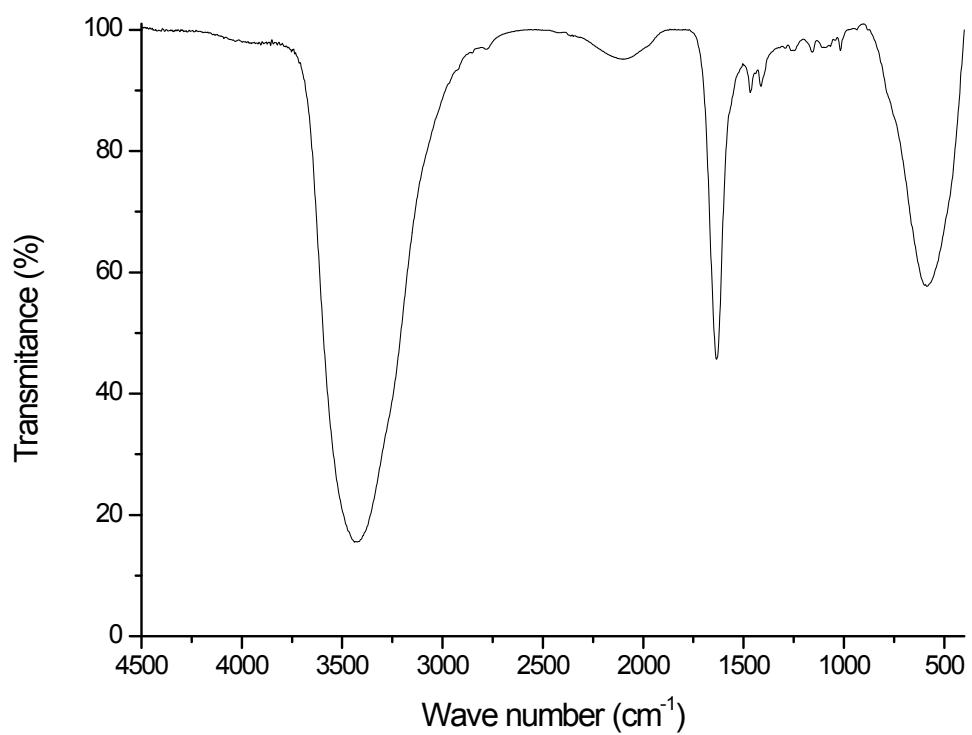


Fig. S3. FT-IR spectrum of $[\text{RuCl}(\text{DPP})(\text{OH}_2)_3]\text{Cl}_2$.

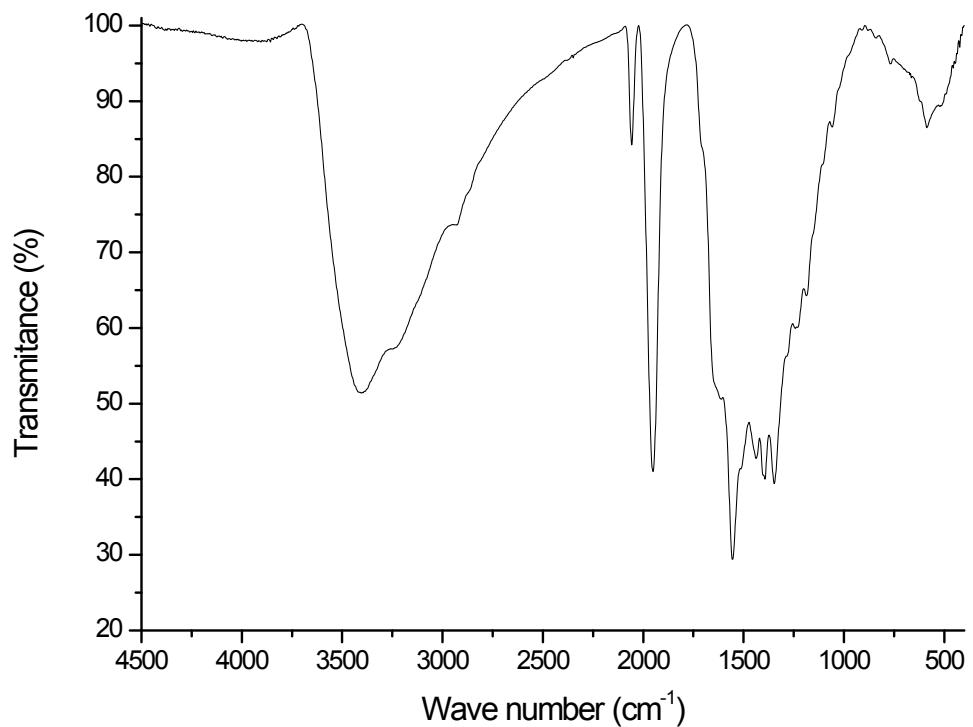


Fig. S4. FT-IR spectrum of $[\text{RuCl}_2(\text{ABMAP})_2]\text{Cl}$.

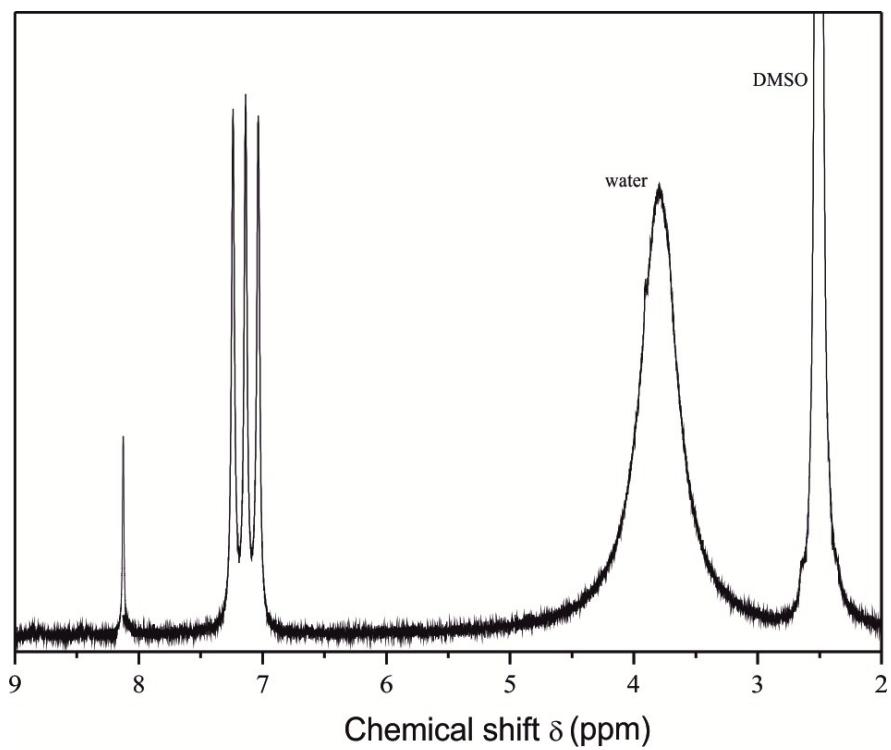


Fig. S5. ¹H NMR spectrum of $[\text{RuCl}(\text{PTCA})_2(\text{OH}_2)]\text{Cl}_2$ in DMSO-d_6 at 500 MHz.

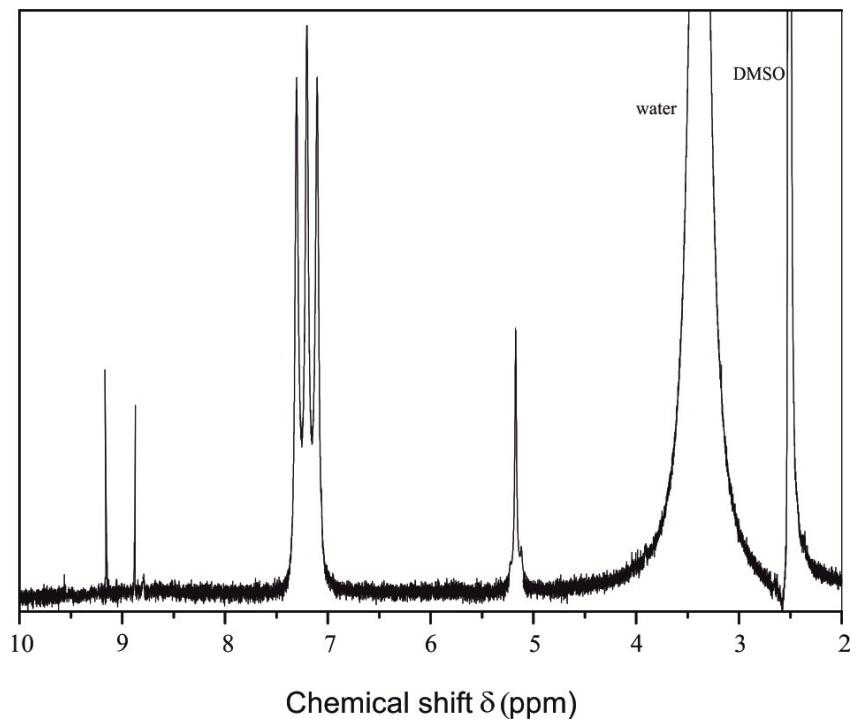


Fig. S6. ¹H NMR spectrum of $[\text{RuCl}(\text{PAOX})_2(\text{OH}_2)]\text{Cl}_2$ in DMSO-d_6 at 500 MHz.

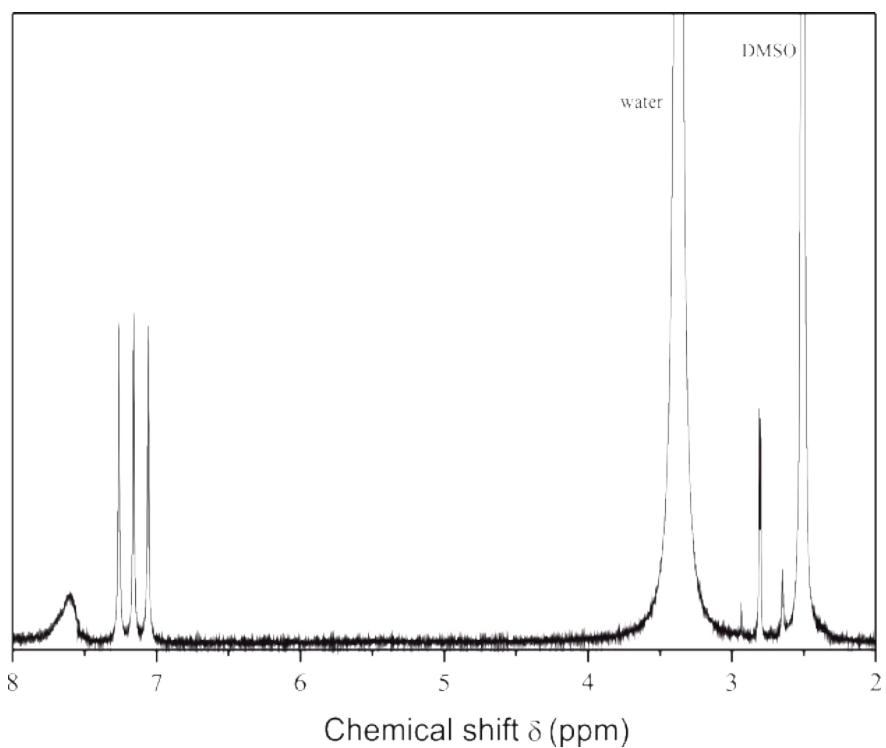


Fig. S7. ¹H NMR spectrum of $[\text{RuCl}_2(\text{ABMAP})_2]\text{Cl}$ in DMSO-d_6 at 500 MHz.

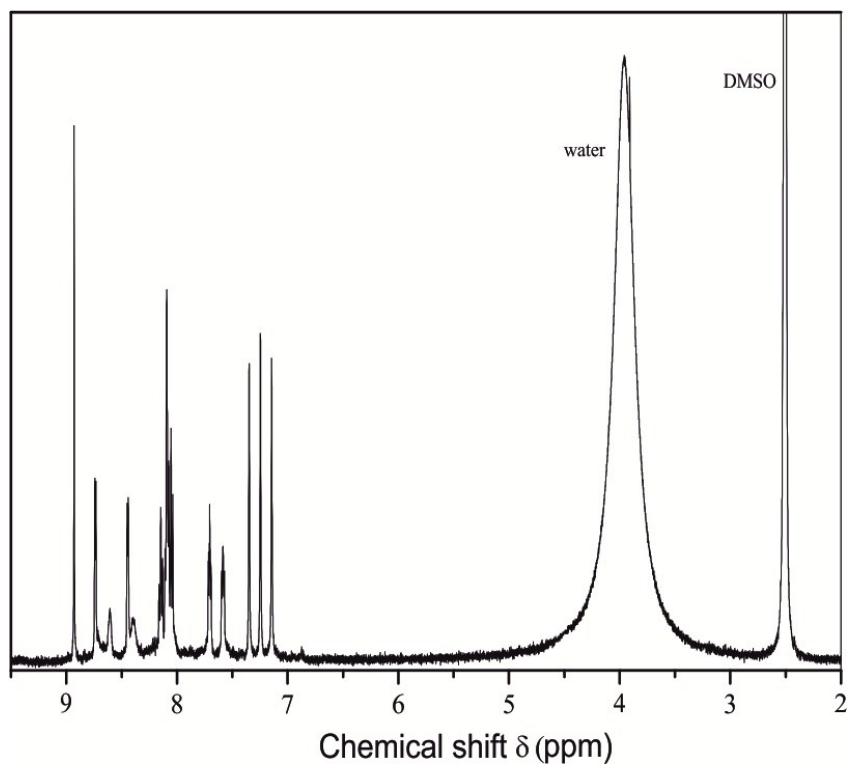


Fig. S8. ¹H NMR spectrum of [RuCl(DPP)(OH₂)₃]Cl₂ in DMSO-d₆ at 500 MHz.

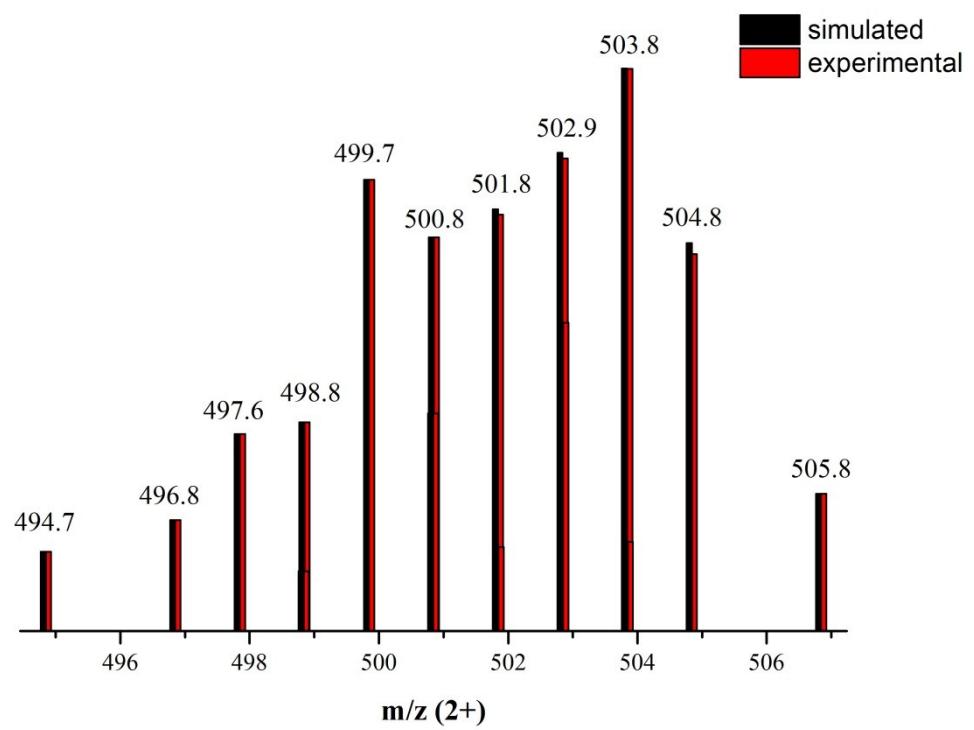


Fig. S9. ESI/MS of $[\text{RuCl}(\text{PTCA})_2(\text{OH}_2)]\text{Cl}_2$ showing the m/z^{2+} for the experimental (red columns) and simulated (black columns) spectra.

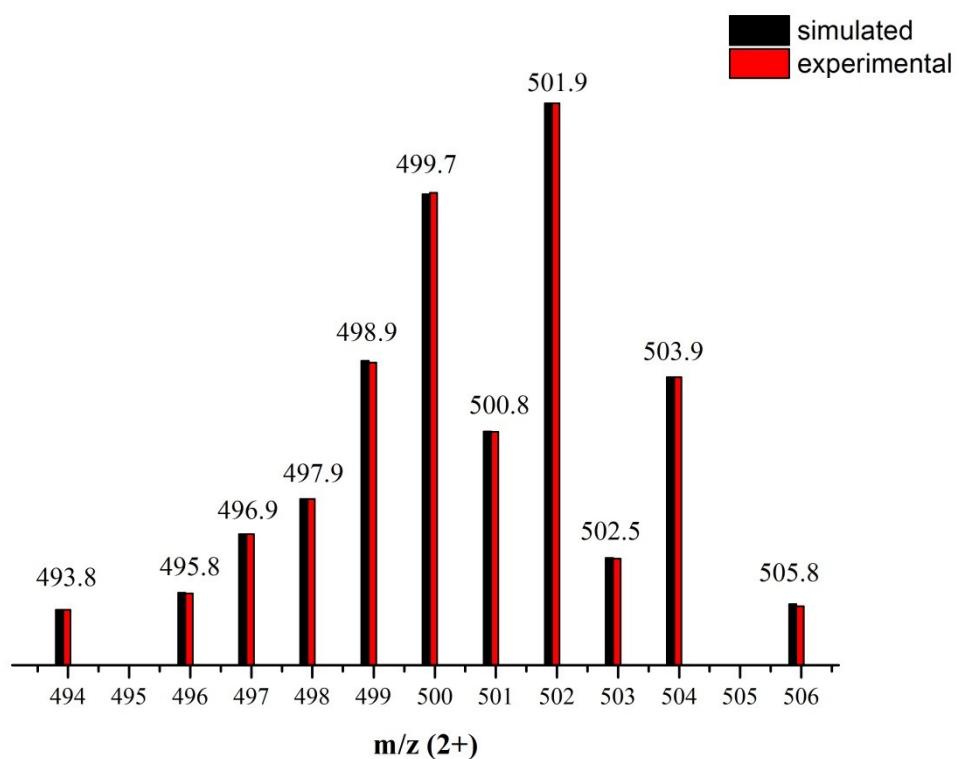


Fig. S10. ESI/MS of $[\text{RuCl}(\text{PAOX})_2(\text{OH}_2)]\text{Cl}_2$ showing the m/z^{2+} for the experimental (red columns) and simulated (black columns) spectra.

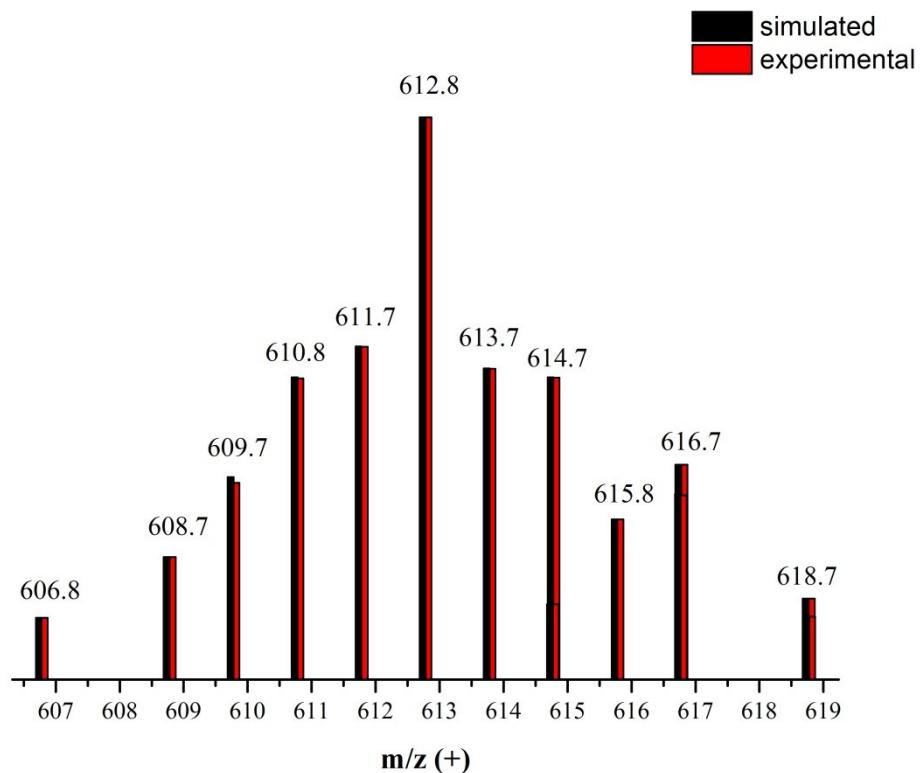


Fig. S11. ESI/MS of $[\text{RuCl}_2(\text{ABMAP})_2]\text{Cl}$ showing the m/z^{2+} for the experimental (red columns) and simulated (black columns) spectra.

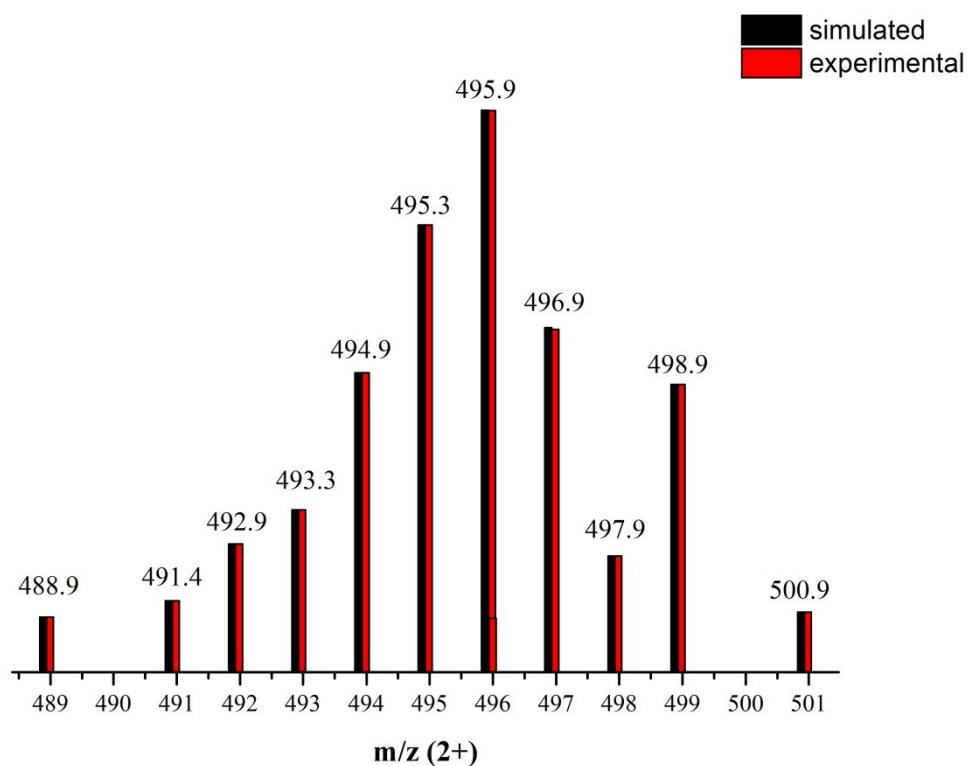


Fig. S12. ESI/MS of $[RuCl(DPP)(OH_2)_3]Cl_2$ showing the m/z^{2+} for the experimental (red columns) and simulated (black columns) spectra.

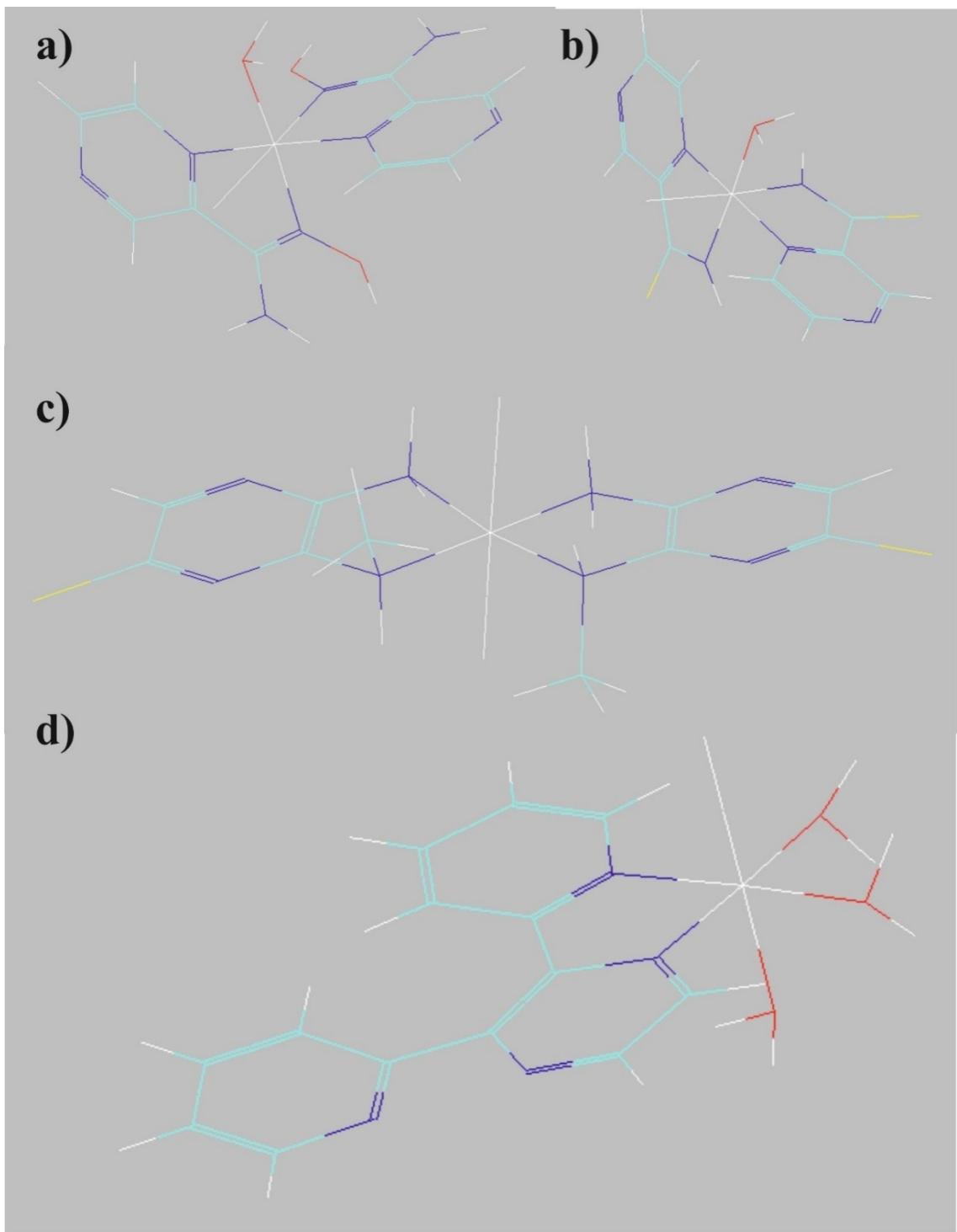


Fig. S13. Structure od Ru(III) complexes studied with pyrazine derivatives: **a)** pyrazine-2-amidoxime (*PAOX*); **b)** pyrazine-2-thiocarboxamide (*PTCA*); **c)** 2-amine-5-bromo-(3-methyl)-amine-pyrazine (*ABMAP*); **d)** 2,3-bis-(2-pyridyl)pyrazine (*DPP*).

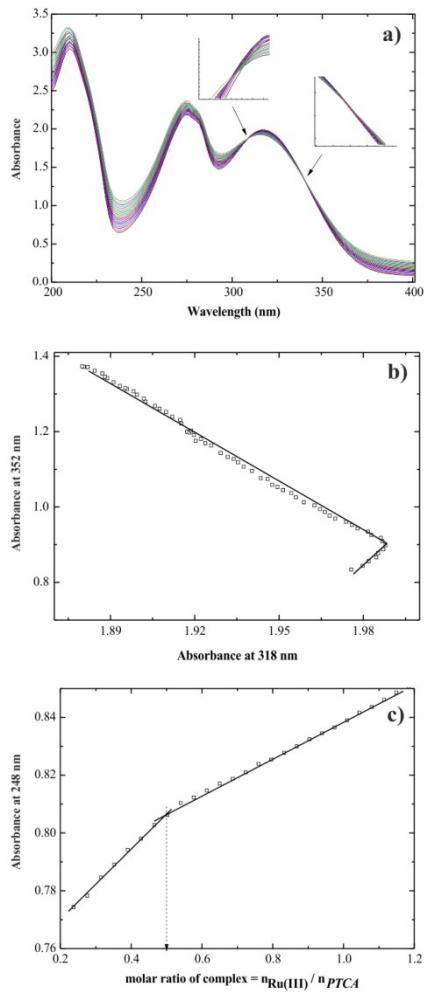


Fig. S14. **a)** Spectrophotometric titration curves of pyrazine-2-thiocarboxamide ($1.00 \cdot 10^{-4}$ mol L⁻¹) using a mixture of *PTCA* at the same concentration and RuCl_3 ($1.81 \cdot 10^{-3}$ mol L⁻¹). **b)** The A-diagram plot for Ru(III) ion with pyrazine-2-thiocarboxamide complexation process, major stoichiometry 1:2. **c)** Dependence of absorbance at 228 nm for *PTCA* as a function of molar ratio $n_{\text{Ru(III)}}/n_{\text{PTCA}}$.

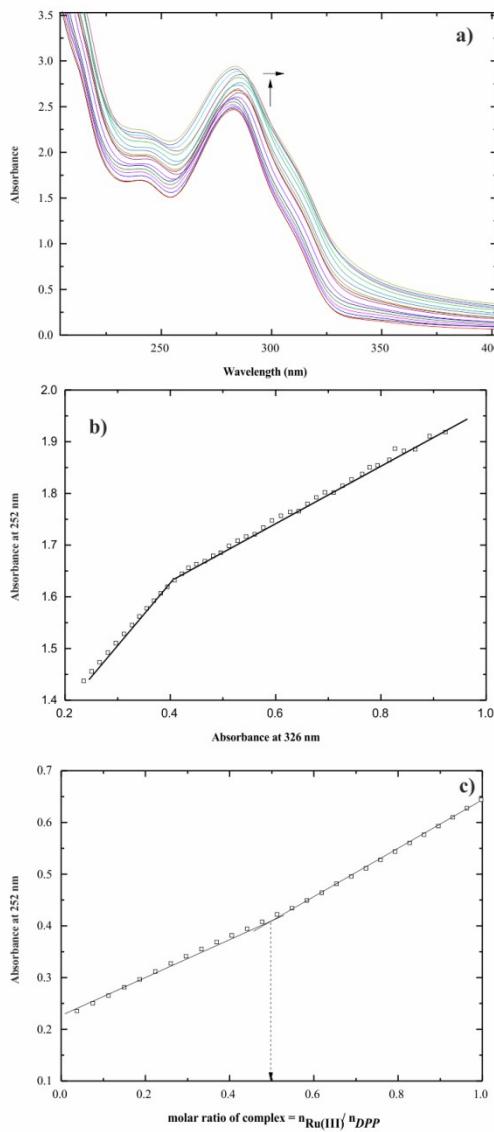


Fig. S15. **a)** Spectrophotometric titration curves of 2,3-bis(2-pyridyl)pyrazine ($5.15 \cdot 10^{-5}$ mol L $^{-1}$) using a mixture of DPP at the same concentration and RuCl $_3$ ($9.29 \cdot 10^{-4}$ mol L $^{-1}$). **b)** The A-diagram for Ru(III) ion with 2,3-bis(2-pyridyl)pyrazine for complexation process, major stoichiometry 1:2. **c)** Dependence of absorbance at 326 nm for DPP as a function of molar ratio $n_{\text{Ru(III)}}/n_{\text{DPP}}$.

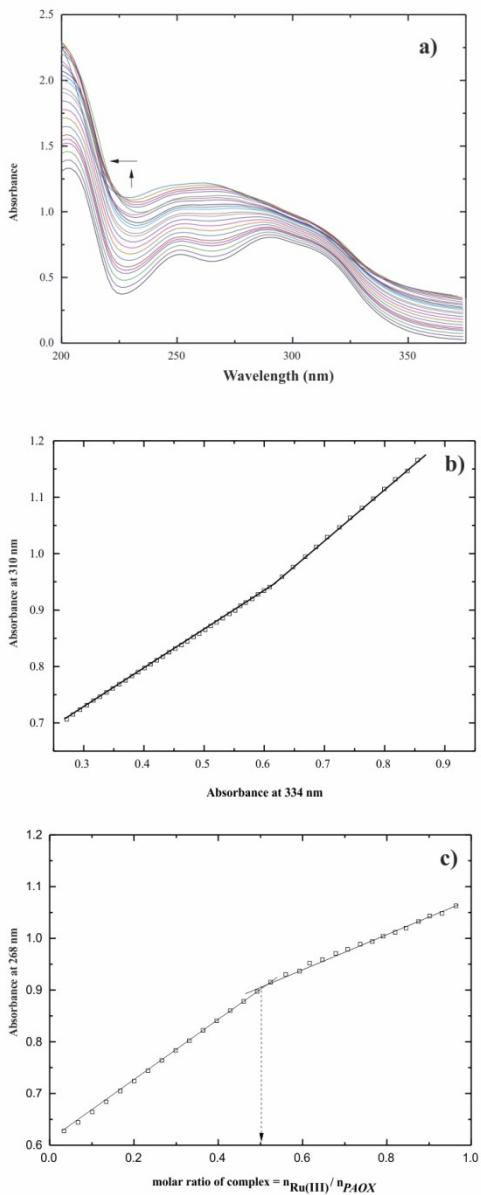


Fig. S16. a) Spectrophotometric titration curves of pyrazine-2-amidoxime ($1.70 \cdot 10^{-4} \text{ mol L}^{-1}$) using a mixture of *PAOX* at the same concentration and RuCl_3 ($2.76 \cdot 10^{-3} \text{ mol L}^{-1}$). b) The A-diagram for Ru(III) ion with pyrazine-2-amidoxime for complexation process, major stoichiometry 1:2. c) Dependence of absorbance at 268 nm for *PAOX* as a function of molar ratio $n_{\text{Ru(III)}}/n_{\text{PAOX}}$.

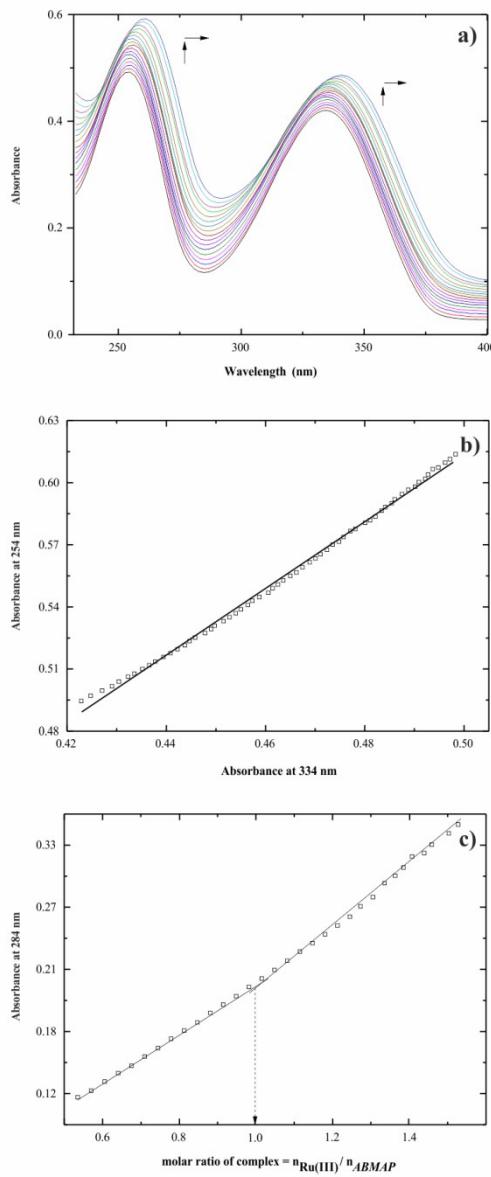


Fig. S17. a) Spectrophotometric titration curves of 2-amine-5-bromo(3-methylamine)-pyrazine ($5.01 \cdot 10^{-5}$ mol L $^{-1}$) using a mixture of ABMAP at the same concentration and RuCl $_3$ ($1.00 \cdot 10^{-3}$ mol L $^{-1}$). b) The A-diagram plot for Ru(III) ion with 2-amine-5-bromo(3-methylamine)-pyrazine for complexation process, major stoichiometry 1:1. c) Dependence of absorbance at 284 nm for ABMAP as a function of molar ratio $n_{\text{Ru(III)}}/n_{\text{ABMAP}}$.

Table S1. Selected bond lengths (\AA) of $[\text{RuCl}(\text{DPP})(\text{OH}_2)_3]\text{Cl}_2$ in gaseous phase from PM3 calculations. Atom labels as in Figure 4c.

Bond	Length (\AA)	Bond	Length (\AA)	Bond	Length (\AA)
C(2)-C(1)	1.3964	C(7)-C(5)	1.3905	H(27)-C(17)	1.1000
N(3)-C(2)	1.4014	C(8)-C(4)	1.3370	H(28)-C(18)	1.1000
H(20)-C(2)	1.1000	Ru(29)-N(6)	1.5061	C(9)-C(7)	1.4122
N(6)-C(1)	1.3899	N(14)-C(8)	1.3509	N(13)-C(7)	1.3872
H(19)-C(1)	1.1000	C(18)-C(8)	1.3858	C(12)-N(13)	1.3899
C(4)-N(3)	1.3968	C(15)-N(14)	1.3509	O(32)-Ru(29)	1.8900
C(5)-N(6)	1.3871	C(16)-C(15)	1.3858	O(33)-Ru(29)	1.8900
C(17)-C(18)	1.3910	H(25)-C(15)	1.1000	O(34)-Ru(29)	1.8900
C(10)-C(9)	1.3967	H(26)-C(16)	1.1000	Cl(31)-Ru(29)	2.2400
H(22)-C(10)	1.1130	H(21)-C(9)	1.1130	C(11)-C(12)	1.3965
H(37)-O(32)	0.9420	H(23)-C(11)	1.1130	H(24)-C(12)	1.1130
H(40)-O(33)	0.9420	H(38)-O(32)	0.9420	H(39)-O(33)	0.9420
H(41)-O(34)	0.9420	H(42)-O(34)	0.9420	H(30)-C(11)	1.1035

Table S2. Selected bond lengths (\AA) of $[\text{RuCl}(\text{PAOX})_2(\text{OH}_2)]\text{Cl}_2$ in gaseous phase from PM3 calculations. Atom labels as in Figure 4d.

Bond	Length (\AA)	Bond	Length (\AA)	Bond	Length (\AA)
C(2)-C(1)	1.3445	N(9)-C(7)	1.2744	N(13)-C(12)	1.2647
N(3)-C(2)	1.2652	N(8)-C(7)	1.2686	H(31)-C(12)	1.1026
H(25)-C(2)	1.1025	O(10)-N(9)	1.3230	C(14)-C(15)	1.3480
C(4)-N(3)	1.2665	N(16)-Ru(21)	1.9402	C(17)-C(15)	1.3463
C(5)-C(4)	1.3482	N(19)-Ru(21)	1.9448	N(18)-C(17)	1.2682
H(26)-C(4)	1.1013	O(22)-Ru(21)	1.9392	O(20)-N(19)	1.3221
N(6)-C(1)	1.2712	Cl(23)-Ru(21)	2.2477	H(32)-C(14)	1.1015
C(7)-C(5)	1.3482	C(11)-N(16)	1.2714	H(27)-N(8)	1.0481
Ru(21)-N(6)	1.9401	C(15)-N(16)	1.2748	H(28)-N(8)	1.0496
H(24)-C(1)	1.1065	C(12)-C(11)	1.3444	H(30)-C(11)	1.1035
H(29)-O(10)	0.9421	H(34)-N(18)	1.0499	H(36)-O(22)	1.0497
H(33)-N(18)	1.0481	H(35)-O(20)	0.9421	H(37)-O(22)	1.0595

Table S3. Selected bond lengths (\AA) of $[\text{RuCl}_2(ABMAP)_2]\text{Cl}$ in gaseous phase from PM3 calculations. Atom labels as in Figure 4a.

Bond	Length (\AA)	Bond	Length (\AA)	Bond	Length (\AA)
C(2)-C(1)	1.3432	H(12)-N(7)	1.0567	Br(29)-C(20)	1.8901
N(3)-C(2)	1.2648	H(32)-N(7)	1.0507	H(30)-C(14)	1.1022
Br(10)-C(2)	1.8928	H(33)-N(8)	1.0593	H(26)-N(13)	1.0569
C(4)-N(3)	1.2698	Cl(40)-Ru(37)	2.2510	H(38)-N(13)	1.0506
C(5)-C(4)	1.3540	Cl(41)-Ru(37)	2.2504	H(15)-C(9)	1.1147
N(8)-C(4)	1.2813	C(23)-N(21)	1.2827	H(16)-C(9)	1.1153
N(6)-C(1)	1.2644	C(28)-N(21)	1.5003	H(17)-C(9)	1.1148
N(7)-C(5)	1.2740	H(39)-N(21)	1.0595	H(34)-C(28)	1.1150
H(11)-C(1)	1.1024	N(22)-C(23)	1.2748	H(35)-C(28)	1.1153
C(9)-N(8)	1.4987	C(24)-N(13)	1.2737	H(36)-C(28)	1.1147
Ru(37)-N(8)	1.9492	N(25)-C(24)	1.2673	C(20)-N(22)	1.2688
N(13)-Ru(37)	1.9416	C(14)-N(25)	1.2647	N(21)-Ru(37)	1.9500

Table S4. Selected bond lengths (\AA) of $[\text{RuCl}(\text{PTCA})_2(\text{OH}_2)]\text{Cl}_2$ in gaseous phase from PM3 calculations. Atom labels as in Figure 4b.

Bond	Length (\AA)	Bond	Length (\AA)	Bond	Length (\AA)
C(2)-C(1)	1.3442	N(16)-Ru(10)	1.9422	H(26)-C(11)	1.1077
N(3)-C(2)	1.2642	N(9)-Ru(10)	1.9312	N(13)-C(12)	1.2646
H(23)-C(2)	1.1023	N(19)-Ru(10)	1.9295	H(27)-C(12)	1.1023
C(4)-N(3)	1.2657	O(21)-Ru(10)	1.9382	C(14)-C(15)	1.3506
C(5)-C(4)	1.3507	S(8)-C(7)	1.5771	C(17)-C(15)	1.3592
H(24)-C(4)	1.1029	Cl(20)-Ru(10)	2.2457	H(28)-C(14)	1.1029
N(6)-C(1)	1.2726	H(25)-N(9)	1.0078	S(18)-C(17)	1.5775
C(7)-C(5)	1.3577	C(11)-N(16)	1.2723	H(29)-N(19)	1.0080
Ru(10)-N(6)	1.9426	C(15)-N(16)	1.2787	H(30)-O(21)	1.0512
H(22)-C(1)	1.1077	C(12)-C(11)	1.3442	H(31)-O(21)	1.0560

Table S5. Selected bond angles (in degrees) of $[\text{RuCl}(\text{DPP})(\text{OH}_2)_3]\text{Cl}_2$ in gaseous phase from PM3 calculations. Atom labels as in Figure 4c.

Angle	[Deg.]	Angle	[Deg.]
N(3)-C(2)-C(1)	120.5861	C(16)-C(15)-N(14)	123.7955
H(20)-C(2)-C(1)	119.7069	H(25)-C(15)-N(14)	118.1024
N(6)-C(1)-C(2)	118.6280	C(17)-C(18)-C(8)	118.2317
H(19)-C(1)-C(2)	120.6860	H(26)-C(16)-C(15)	120.8826
C(4)-N(3)-C(2)	120.2919	H(27)-C(17)-C(16)	120.3423
C(5)-N(6)-C(1)	121.8610	H(28)-C(18)-C(8)	120.8839
C(7)-C(5)-C(4)	131.2583	C(9)-C(7)-C(5)	131.2647
C(8)-C(4)-N(3)	120.4270	N(13)-C(7)-C(5)	109.2478
Ru(29)-N(6)-C(1)	127.3286	C(12)-N(13)-C(7)	121.8614
N(14)-C(8)-C(4)	118.0986	O(32)-Ru(29)-N(6)	108.5257
C(18)-C(8)-C(4)	118.0990	O(33)-Ru(29)-N(6)	89.9999
C(15)-N(14)-C(8)	116.6196	O(34)-Ru(29)-N(6)	103.4694
Cl(31)-Ru(29)-N(6)	89.9997	H(37)-O(32)-Ru(29)	119.9999
C(10)-C(9)-C(7)	119.1468	H(38)-O(32)-Ru(29)	119.9997
H(21)-C(9)-C(7)	106.9818	H(39)-O(33)-Ru(29)	120.0003
C(11)-C(12)-N(13)	118.6271	H(40)-O(33)-Ru(29)	119.9999
H(22)-C(10)-C(9)	106.6826	H(41)-O(34)-Ru(29)	119.9999
H(23)-C(11)-C(10)	106.6078	H(42)-O(34)-Ru(29)	120.0004
H(24)-C(12)-C(11)	107.1167		

Table S6. Selected bond angles (in degrees) of $[\text{RuCl}(\text{PAOX})_2(\text{OH}_2)]\text{Cl}_2$ in gaseous phase from PM3 calculations. Atom labels as in Figure 4d.

Angle	[Deg.]	Angle	[Deg.]
N(3)-C(2)-C(1)	122.1158	C(11)-N(16)-Ru(21)	120.2191
H(25)-C(2)-C(1)	120.8925	C(15)-N(16)-C(11)	117.0381
C(4)-N(3)-C(2)	116.4129	C(12)-C(11)-N(16)	121.3299
C(5)-C(4)-N(3)	122.6950	H(30)-C(11)-C(12)	117.3928
H(26)-C(4)-N(3)	114.9263	N(13)-C(12)-C(11)	122.2155
N(6)-C(1)-C(2)	120.7375	H(31)-C(12)-C(11)	120.8558
C(7)-C(5)-C(4)	126.7921	C(14)-C(15)-N(16)	120.6255
Ru(21)-N(6)-C(1)	120.9908	C(17)-C(15)-C(14)	125.5614
H(24)-C(1)-C(2)	117.9977	N(18)-C(17)-C(15)	122.2790
N(9)-C(7)-C(5)	113.2311	O(20)-N(19)-C(17)	116.1785
N(8)-C(7)-C(5)	121.0709	H(32)-C(14)-N(13)	114.9634
O(10)-N(9)-C(7)	115.4612	H(27)-N(8)-C(7)	121.6432
N(16)-Ru(21)-N(6)	110.7465	H(28)-N(8)-C(7)	119.4669
N(19)-Ru(21)-N(6)	161.7345	H(29)-O(10)-N(9)	111.1263
O(22)-Ru(21)-N(6)	92.0746	H(33)-N(18)-C(17)	121.8395
Cl(23)-Ru(21)-N(6)	89.0088	H(34)-N(18)-C(17)	118.8826
H(35)-O(20)-N(19)	110.8119	H(37)-O(22)-Ru(21)	106.1700
H(36)-O(22)-Ru(21)	105.3142		

Table S7. Selected bond angles (in degrees) of $[\text{RuCl}_2(\text{ABMAP})_2]\text{Cl}$ in gaseous phase from PM3 calculations. Atom labels as in Figure 4a.

Angle	[Deg.]	Angle	[Deg.]
N(3)-C(2)-C(1)	119.7568	C(28)-N(21)-C(23)	112.3373
Br(10)-C(2)-C(1)	120.2487	H(39)-N(21)-C(23)	107.7352
C(4)-N(3)-C(2)	119.5023	N(22)-C(23)-N(21)	126.2549
C(5)-C(4)-N(3)	119.6111	C(24)-N(13)-Ru(37)	103.2801
N(8)-C(4)-N(3)	128.8825	N(25)-C(24)-N(13)	125.8472
N(6)-C(1)-C(2)	122.0601	C(14)-N(25)-C(24)	115.7610
N(7)-C(5)-C(4)	112.2941	C(20)-N(22)-C(23)	111.8045
H(11)-C(1)-C(2)	121.7740	Br(29)-C(20)-C(14)	119.8779
C(9)-N(8)-C(4)	112.3896	H(30)-C(14)-C(20)	122.4016
Ru(37)-N(8)-C(4)	102.9789	H(26)-N(13)-C(24)	108.4755
N(13)-Ru(37)-N(7)	90.1454	H(38)-N(13)-C(24)	115.0964
N(21)-Ru(37)-N(7)	160.9081	H(15)-C(9)-N(8)	110.7970
H(12)-N(7)-C(5)	109.3388	H(16)-C(9)-N(8)	113.7069
H(32)-N(7)-C(5)	114.9945	H(17)-C(9)-N(8)	110.4408
H(33)-N(8)-C(4)	108.4065	H(34)-C(28)-N(21)	110.4075
Cl(40)-Ru(37)-N(7)	104.2654	H(35)-C(28)-N(21)	113.7423
Cl(41)-Ru(37)-N(7)	84.6328	H(36)-C(28)-N(21)	110.8934
C(23)-N(21)-Ru(37)	104.1034		

Table S8. Selected bond angles (in degrees) of $[\text{RuCl}(\text{PTCA})_2(\text{OH}_2)]\text{Cl}_2$ in gaseous phase from PM3 calculations. Atom labels as in Figure 4b.

Angle	Degree (°)	Angle	Degree (°)
N(3)-C(2)-C(1)	121.6435	H(25)-N(9)-C(7)	122.2726
H(23)-C(2)-C(1)	121.2212	C(11)-N(16)-Ru(10)	116.2602
C(4)-N(3)-C(2)	115.7469	C(15)-N(16)-Ru(10)	103.9610
C(5)-C(4)-N(3)	123.1702	C(12)-C(11)-N(16)	122.2288
H(24)-C(4)-N(3)	114.8885	H(26)-C(11)-C(12)	117.8395
N(6)-C(1)-C(2)	122.7640	N(13)-C(12)-C(11)	121.6544
C(7)-C(5)-C(4)	119.9335	H(27)-C(12)-C(11)	121.1570
Ru(10)-N(6)-C(1)	116.3274	C(14)-C(15)-N(16)	120.2453
H(22)-C(1)-C(2)	117.7442	C(17)-C(15)-C(14)	120.3962
N(16)-Ru(10)-N(6)	106.1515	H(28)-C(14)-N(13)	114.8301
N(9)-Ru(10)-N(6)	77.0872	S(18)-C(17)-C(15)	127.7193
N(19)-Ru(10)-N(6)	165.3079	H(29)-N(19)-Ru(10)	123.2818
O(21)-Ru(10)-N(6)	91.4324	H(30)-O(21)-Ru(10)	106.0504
S(8)-C(7)-C(5)	127.7973	H(31)-O(21)-Ru(10)	101.5025
Cl(20)-Ru(10)-N(6)	100.1856		

Table S9. Optimized structures of the favorable conformation of ruthenium(III) complexes with pyrazine derivatives: *PAOX*, *DPP*, *ABMAP* and *PTCA*, total energies (E in kcal/mol), and atom Cartesian (X,Y,Z) coordinates (Å). The data were obtained at the PM3 calculations.

Complex	Total energy and atom coordinates			
	Atoms	X	Y	Z
E = 262				
C(1)	2.641049	-1.681573	-0.840185	
C(2)	3.631160	-1.755560	-1.747350	
N(3)	3.968254	-0.612462	-2.511264	
C(4)	3.339951	0.503806	-2.356231	
C(5)	2.255031	0.585205	-1.365930	
N(6)	1.950426	-0.457307	-0.669318	
C(7)	1.447639	1.779223	-1.080414	
N(8)	0.555291	1.590150	-0.162870	
N(9)	1.647822	3.007996	-1.753569	
O(10)	-0.277791	2.488224	0.277099	
C(11)	-1.165968	0.662371	2.913894	
C(12)	-2.347357	0.901145	3.510834	
N(13)	-3.550001	0.658378	2.805902	
C(14)	-3.535212	0.211726	1.595851	
C(15)	-2.243342	-0.048380	0.942219	
N(16)	-1.146967	0.170881	1.585939	
C(17)	-2.069577	-0.553380	-0.425138	
N(18)	-0.830307	-0.702163	-0.763814	
N(19)	-3.166949	-0.850147	-1.268416	
[RuCl(<i>PAOX</i>) ₂ (OH ₂)]Cl ₂	O(20)	-0.413646	-1.146512	-1.914282
see Fig. S5a	Ru(21)	0.489097	-0.211516	0.597606
	Cl(22)	2.072353	0.362661	2.107750
	O(23)	0.540522	-1.992125	1.336475
	H(24)	2.385446	-2.524401	-0.270139
	H(25)	4.153992	-2.655191	-1.891296
	H(26)	3.611459	1.335057	-2.935928
	H(27)	1.051328	3.786622	-1.589521
	H(28)	2.380167	3.106192	-2.419420
	H(29)	-0.307513	3.368907	-0.044050
	H(30)	-0.272239	0.841513	3.432636
	H(31)	-2.380965	1.265833	4.495499
	H(32)	-4.442110	0.040136	1.097163
	H(33)	-3.027991	-1.203454	-2.187405
	H(34)	-4.099336	-0.698929	-0.957171
	H(35)	-0.988134	-1.397975	-2.611712
	H(36)	0.245249	-1.903905	2.266064
	H(47)	-0.179730	-2.487644	0.896394
E = 203				
	Atoms	X	Y	Z
	C(1)	-0.878892	-1.390374	-2.277264
	C(2)	-1.987850	-2.178895	-1.734610
	N(3)	-2.197560	-2.179982	-0.463625
	C(4)	-1.366290	-1.461940	0.435556

[RuCl(<i>DPP</i>)(OH ₂) ₃]Cl ₂ see Fig. S5d	C(5)	-0.302382	-0.792353	-0.061147
	N(6)	-0.116096	-0.739700	-1.465481
	C(7)	0.749478	-0.049507	0.656941
	C(8)	-1.760662	-1.493077	1.847679
	C(9)	0.975354	-0.074659	2.115404
	C(10)	1.990780	0.629914	2.641751
	C(11)	2.867858	1.393965	1.752823
	C(12)	2.653052	1.350934	0.426965
	N(13)	1.573986	0.596219	-0.108977
	N(14)	-2.142593	-0.403712	2.428838
	C(15)	-2.561256	-0.377779	3.781385
	C(16)	-2.589098	-1.512470	4.502680
	C(17)	-2.185503	-2.774267	3.869622
	C(18)	-1.786618	-2.767137	2.586153
	H(19)	-0.739496	-1.335102	-3.314801
	H(20)	-2.621000	-2.718344	-2.375581
	H(21)	0.386219	-0.660390	2.752256
	H(22)	2.171014	0.611606	3.676613
	H(23)	3.666566	1.951848	2.146197
	H(24)	3.291994	1.886054	-0.209841
	H(25)	-2.848325	0.533315	4.219175
	H(26)	-2.897490	-1.498425	5.506849
	H(27)	-2.210154	-3.672077	4.414868
	H(28)	-1.493756	-3.661134	2.117889
	Ru(29)	1.311513	0.454771	-2.047598
	Cl(31)	2.735921	-1.302222	-2.033068
	O(32)	2.722697	1.693577	-2.495282
	O(33)	0.104381	1.962417	-2.048797
	O(34)	0.947073	0.221666	-3.929769
	H(37)	2.309123	2.401313	-3.030661
	H(38)	3.311334	1.222873	-3.119955
	H(39)	-0.777228	1.600379	-2.273531
	H(40)	0.024087	2.248433	-1.116431
	H(41)	1.786432	-0.074436	-4.337370
	H(42)	0.798087	1.118598	-4.293157

E = 185

Atoms	X	Y	Z
C(1)	1.177651	-2.568560	-1.125722
C(2)	1.016449	-3.736468	-1.773587
N(3)	-0.186685	-3.990720	-2.475665
C(4)	-1.136900	-3.118769	-2.506557
C(5)	-0.959130	-1.842990	-1.798353
N(6)	0.137942	-1.607028	-1.156038
C(7)	-1.949194	-0.767198	-1.747683
S(8)	-3.504220	-0.857705	-2.404037
N(9)	-1.499697	0.343540	-1.015371
Ru(10)	0.269187	0.112825	-0.230858
C(11)	0.285915	3.003588	0.455171
C(12)	0.034398	4.046440	1.267118
[RuCl(<i>PTCA</i>) ₂ (OH ₂)]Cl ₂	N(13)	-0.565638	3.826006
			2.529901

see Fig. S5b	C(14)	-0.879007	2.636756	2.918402
	C(15)	-0.603903	1.498236	2.030097
	N(16)	-0.053244	1.693860	0.876804
	C(17)	-0.907028	0.102488	2.342481
	S(18)	-1.605451	-0.436596	3.784025
	N(19)	-0.535013	-0.770898	1.309558
	Cl(20)	1.192883	0.963983	-2.110563
	O(21)	2.060595	-0.073612	0.455783
	H(22)	2.067030	-2.375265	-0.603257
	H(23)	1.778097	-4.459980	-1.761079
	H(25)	-2.020875	-3.327195	-3.033255
	H(26)	0.729014	3.164930	-0.481741
	H(27)	0.280591	5.023007	0.968256
	H(28)	-1.323524	2.494845	3.858890
	H(29)	-2.090962	1.106941	-0.779499
	H(30)	2.389389	0.838908	0.592065
	H(31)	1.964300	-0.436812	1.359692
	H(34)	-0.534870	-1.759057	1.416922

E = 163

Atoms	X	Y	Z
C(1)	-4.624497	2.255132	-0.510529
C(2)	-4.380343	1.790804	-1.884141
N(3)	-3.306349	1.128015	-2.151691
C(4)	-2.359830	0.838003	-1.137901
C(5)	-2.584097	1.267744	0.114934
N(6)	-3.761037	1.995701	0.411555
N(7)	-1.600567	0.995299	1.114209
N(8)	-1.144381	0.121032	-1.411508
C(9)	-1.441780	-1.227223	-1.945848
Br(10)	-5.611803	2.156513	-3.243052
H(11)	-5.493245	2.797860	-0.278646
H(12)	-1.310921	1.868677	1.553192
N(13)	0.924399	0.346507	1.954452
C(14)	4.062837	-1.253871	2.935266
H(15)	-0.523792	-1.805980	-2.037711
H(16)	-2.135876	-1.757748	-1.293396
H(17)	-1.893799	-1.144194	-2.936826
C(20)	4.401950	-1.928574	1.673750
N(21)	1.549318	-0.873366	-0.382357
N(22)	3.623557	-1.801299	0.653018
C(23)	2.445982	-1.017300	0.731418
C(24)	2.135642	-0.407622	1.887423
N(25)	2.988131	-0.544462	3.008382
H(26)	0.357892	0.009996	2.732509
C(28)	2.240818	-0.242441	-1.529580
Br(29)	5.951235	-2.966918	1.539354
H(30)	4.687450	-1.348263	3.774221
H(32)	-2.003960	0.390859	1.828243
H(33)	-0.636961	0.635752	-2.133061
H(34)	2.990326	-0.926388	-1.934326

[RuCl₂(ABMAP)₂]Cl

see Fig. S5c

H(35)	2.741739	0.675998	-1.221400
H(36)	1.526380	-0.012741	-2.318939
Ru(37)	-0.051772	0.117746	0.246112
H(38)	1.147348	1.327619	2.114368
H(39)	1.259019	-1.809290	-0.670109
Cl(40)	-0.820361	-1.849996	1.082133
Cl(41)	0.651649	2.198719	-0.333521
