

Supporting Information

Highly selective and sensitive colorimetric chemosensor for detection of Co²⁺ in a near-perfect aqueous solution

Seong Youl Lee, Jae Jun Lee, Kwon Hee Bok, So Young Kim, Cheal Kim*

Department of Fine Chemistry and Department of Interdisciplinary Bio IT Materials, Seoul National University of Science and Technology, Seoul 139-743, Korea. Fax: +82-2-973-9149; Tel: +82-2-970-6693; E-mail: chealkim@seoultech.ac.kr

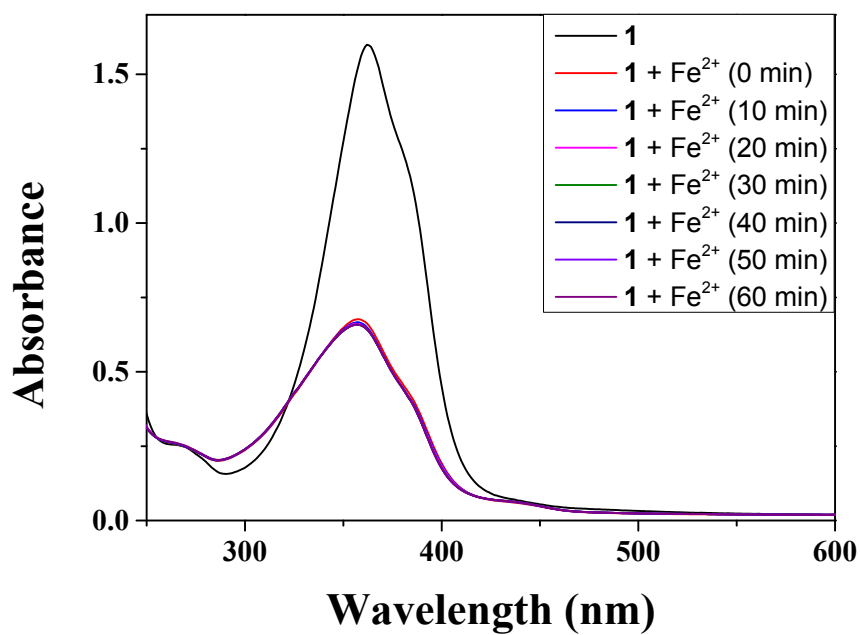


Fig. S1 Absorption spectral changes of **1** (20 μ M) for 1 h in the presence of 1.1 equiv of Fe²⁺ ion in bis-tris buffer solution (10 mM, pH 7).

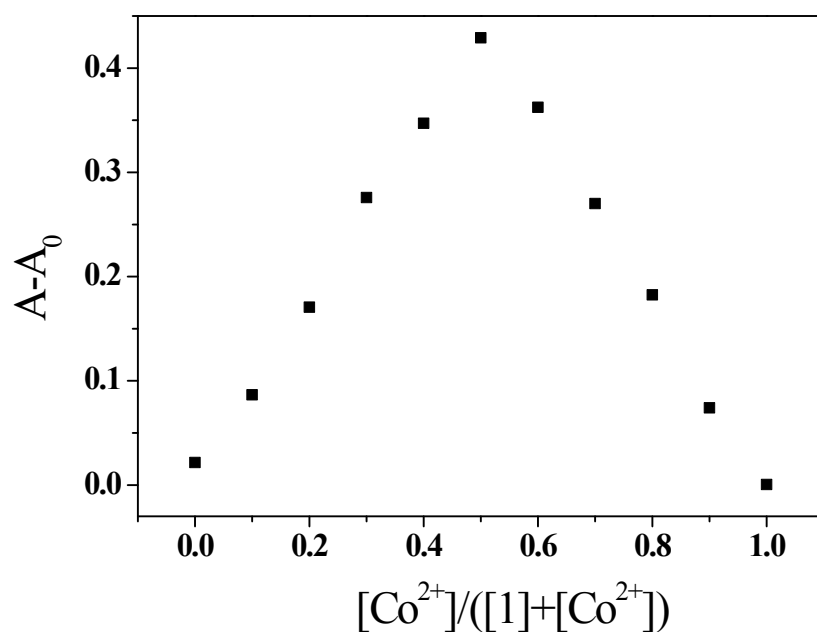


Fig. S2 Job plot of **1** and Co^{2+} in bis-tris buffer solution (10 mM, pH 7). The total concentrations of **1** and Co^{2+} were 150 μM .

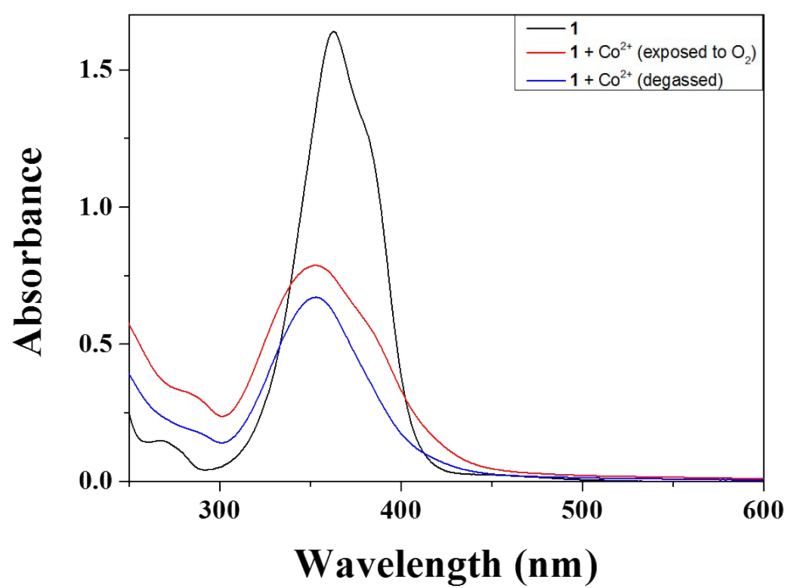


Fig. S3 Absorption spectra of **1** (20 μ M) and **1**- Co²⁺ under the anaerobic and aerobic conditions, respectively.

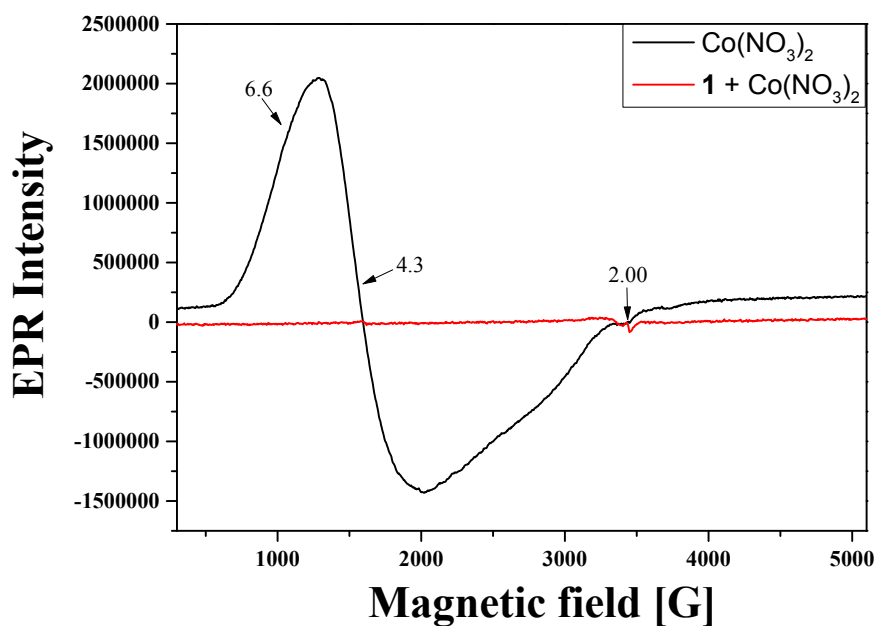


Fig. S4 X-band EPR spectra of $\text{Co}(\text{NO}_3)_2$ (black line) and $\mathbf{1} + \text{Co}(\text{NO}_3)_2$ (red line) recorded at 5 K. Dark line: the $\text{Co}(\text{NO}_3)_2$ (1 mM) solution in CH_3CN (or H_2O) was frozen in liquid nitrogen. Red line: the EPR sample was frozen in liquid nitrogen 5 min after $\mathbf{1}$ (1.2 mM) was mixed with $\text{Co}(\text{NO}_3)_2$ (1 mM) in CH_3CN (or H_2O) at room temperature.

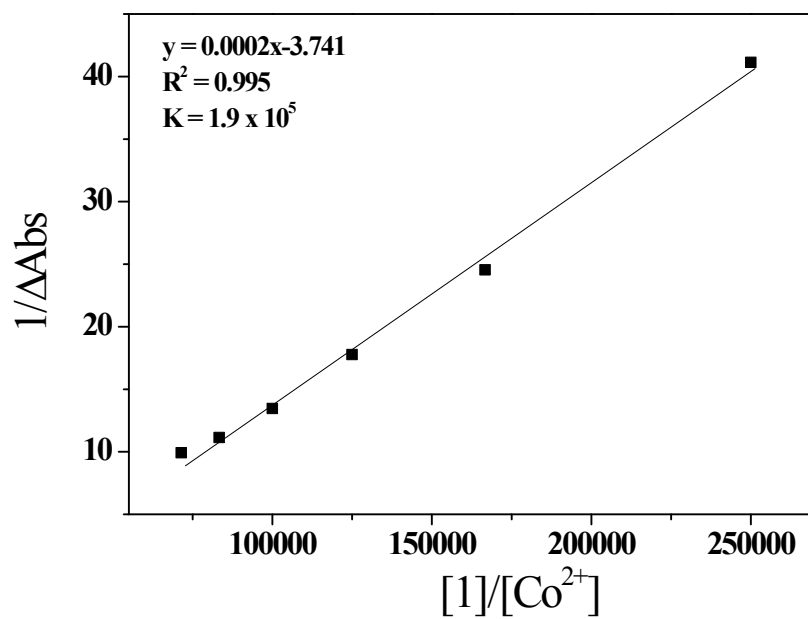


Fig. S5 Benesi-Hildebrand plot (absorbance at 420 nm) of **1** based on UV-vis titration, assuming 1:1 stoichiometry for association between **1** and Co^{2+} .

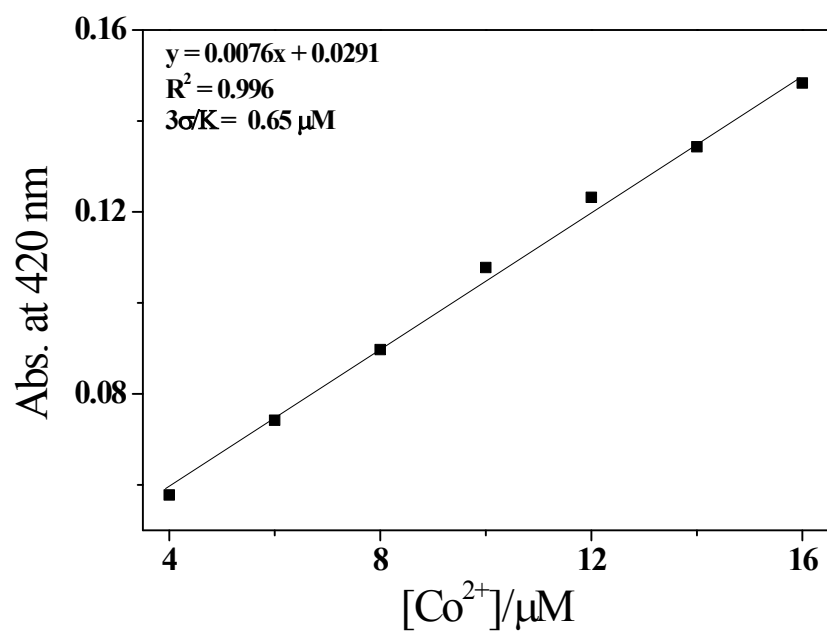
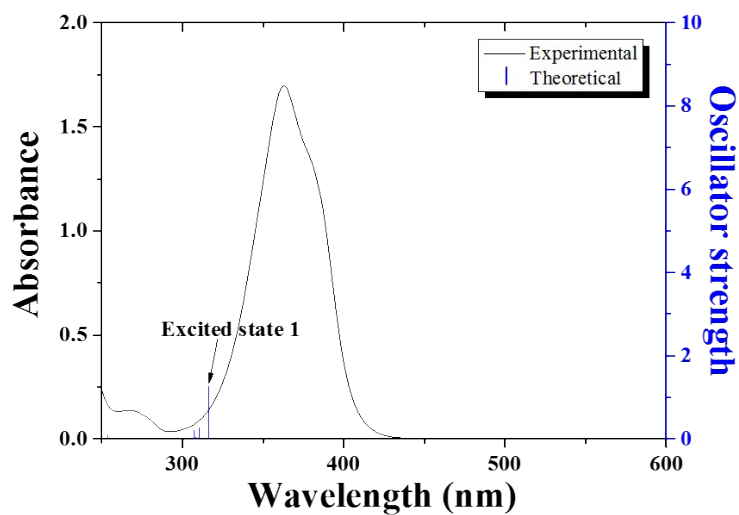


Fig. S6 Determination of the detection limit based on change in the ratio (absorbance at 420 nm) of **1** (20 μM) with Co^{2+} .

(a)

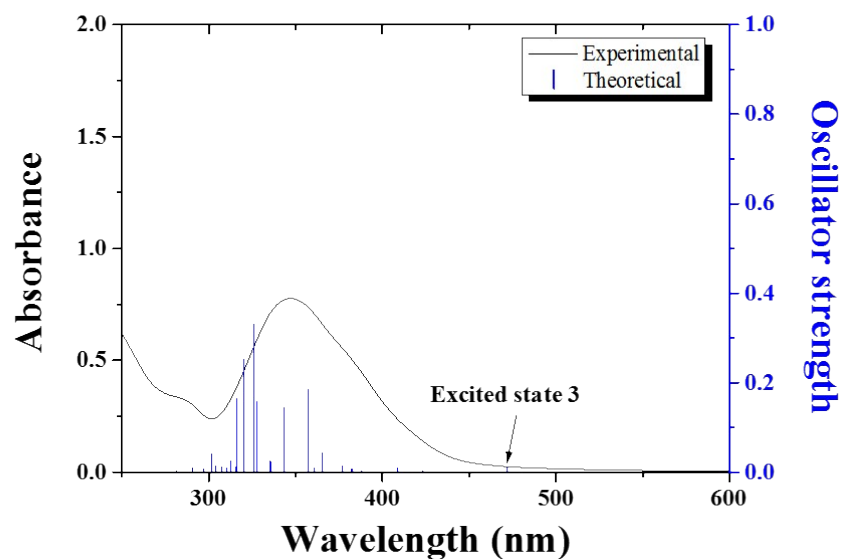


(b)

Excited state 1	Wavelength (nm)	Percent (%)	Main character	Oscillator strength
H → L	316.49	86	ICT	1.265
H - 1 → L + 1		12	ICT	

Fig S7. (a) The theoretical excitation energies and the experimental UV-vis spectrum of **1**. (b) The major electronic transition energies and molecular orbital contributions for **1** (H = HOMO and L = LUMO).

(a)



(b)

Excited state 3	Wavelength (nm)	Percent (%)	Main character	Oscillator strength
H - 1 → L + 3	472.03	45	LMCT	0.0102
H - 8 → L + 3		15	LMCT	
H - 6 → L + 3		9	LMCT	
H - 14 → L + 3		3	LMCT	
H - 2 → L + 3		2	LMCT	

Fig. S8 (a) The theoretical excitation energies and the experimental UV-vis spectrum of 1-Co³⁺. (b) The major electronic transition energies and molecular orbital contributions for 1-Co³⁺ (H = HOMO and L = LUMO).