## **Electronic Supporting Information Materials**

Two hydrazone Copper (II) complexes: Synthesis, crystal structure, cytotoxicity, and action mechanism

Kun Hu,<sup>a</sup> Guimei Zhou,<sup>a</sup> Zhong Zhang,<sup>a</sup> Feiyan Li,<sup>a</sup> Jingui Li,<sup>a</sup> and Fupei Liang\*<sup>a,b</sup>

<sup>a</sup>State Key Laboratory Cultivation Base for the Chemistry and Molecular Engineering of Medicinal Resources, School of Chemistry and Pharmacy, Guangxi Normal University, 15 Yucai Road, Guilin 541004, P. R. China., Fax, 086-773-2120958 ; E-mail: fliangoffice@yahoo.com <sup>b</sup>Guangxi Key Laboratory of Electrochemical and Magnetochemical Functional Materials, College of Chemistry and Bioengineering, Guilin University of Technology, Guilin 541004, China Fig. S1. IR (KBr) spectra of H-L1.

Fig. S2. HRMS(ESI) spectra of H-L1. Calcd. for C<sub>18</sub>H<sub>16</sub>N<sub>3</sub>O<sub>3</sub> [M+H]<sup>+</sup> 322.11917, found 322.11743.

Fig. S3 NMR spectra of H-L1 in DMSO. A) <sup>1</sup>H; B) <sup>13</sup>C.

Fig. S4. IR (KBr) spectra of H-L2.

Fig. S5. HRMS(ESI) spectra of H-L2. Calcd. for C<sub>17</sub>H<sub>14</sub>N<sub>3</sub>O<sub>2</sub> [M+H]<sup>+</sup> 292.10860, found 292.10674.

Fig. S6 NMR spectra of H-L1 in DMSO. A) <sup>1</sup>H; B) <sup>13</sup>C.

Fig. S7. IR (KBr) spectra of complex 1.

Fig. S8. HRMS (ESI) spectra of complex **1**. Calcd. for  $C_{18}H_{14}N_3O_3Cu$  [M-NO<sub>3</sub>]<sup>+</sup> 383.03312, found 383.03246.

Fig. S9. IR (KBr) spectra of complex 2.

Fig. S10. HR-ESI-MS spectra of complex **2.** HRMS (ESI): Calcd. for  $C_{17}H_{12}N_3O_2Cu$  [M-NO<sub>3</sub>]<sup>+</sup> 353.02255, found 353.02136.

Fig. S11. LC-MS spectra for 1 and 2 ( $2.0 \times 10^{-6}$  M) monitored at 370 nm in TBS buffer (0.1 %) with 0 h and 48 h. Column: Phenyl-Hexyl (150mm×5.0 µm I.D.). Column temperature: 25 °C. Mobile phase: A) methol, B) H<sub>2</sub>O. Gradient: 70% to 100% in 20 min. Flow rate: 0.3 mL/min. Injection volume: 5. 0 µL. Complex 1: [M-NO<sub>3</sub>]<sup>+</sup>=383.03236; Complex 2:[M-NO<sub>3</sub>]<sup>+</sup>=353.02133, [M-NO<sub>3</sub>+CH<sub>3</sub>OH]<sup>+</sup>=385.04775.











Fig. S3



Fig. S4



Fig. S5



Fig. S6





Fig. S7



Fig. S8



Fig. S9







Fig. S11



