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Supporting Information

TiO₂ Microspheres Supported Polyoxometalate-based Ionic Liquids Induced

Catalytic Oxidative Deep-desulfurization

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Fig. S1 DRS spectra of (a): 0.05-C₁₆SiW-TiO₂, (b): 0.1-C₁₆SiW-TiO₂, (c): 0.2-C₁₆SiW-TiO₂, (d): 0.1-C₄SiW-TiO₂, (e): 0.1-C₈SiW-TiO₂.



Fig. S2 Wide-angle XRD patterns of (a): 0.05-C₁₆SiW-TiO₂, (b): 0.1-C₁₆SiW-TiO₂, (c): 0.2-C₁₆SiW-TiO₂, (d): 0.1-C₄SiW-TiO₂, (e): 0.1-C₈SiW-TiO₂.



Fig. S3. Effect of different catalysts on the removal of DBT. Experimental conditions: m(catalyst) = 0.01 g, $V([\text{Bmim}]\text{BF}_4) = 1\text{mL}$, $T = 50^{\circ}\text{C}$, t = 1h, n(O/S) = 2, V(model oil) = 5 mL.

Entry	Different ILs	Sulfur removal / %	
		EDS ^a	ECODS ^b
1	[Bmim]PF ₆	16.8	18.7
2	[Omim]PF ₆	31.3	50.5
3	[Bmim]BF ₄	18.9	95.3
4	[Omim]BF ₄	32.5	93.2

Table S1. The effect of different extractants on the removal of DBT

Experiment conditions: m(catalyst) = 0.01 g, $V_{\text{IL}} = 1 \text{ mL}$, n(O/S) = 2, $T = 50^{\circ}\text{C}$,

V(model oil) = 5 mL. a: t = 15 min; b: t = 1 h.



Fig. S4. Spatial structure of the different substrate. (a). DBT, (b). 4-MDBT, (c). BT and (d). 3-

MBT.