

## **Study of amorphous Cu-Te-Si thin films showing high thermal stability for application as a cation supply layer in conductive bridge random access memory devices**

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# 1. Kissinger analysis

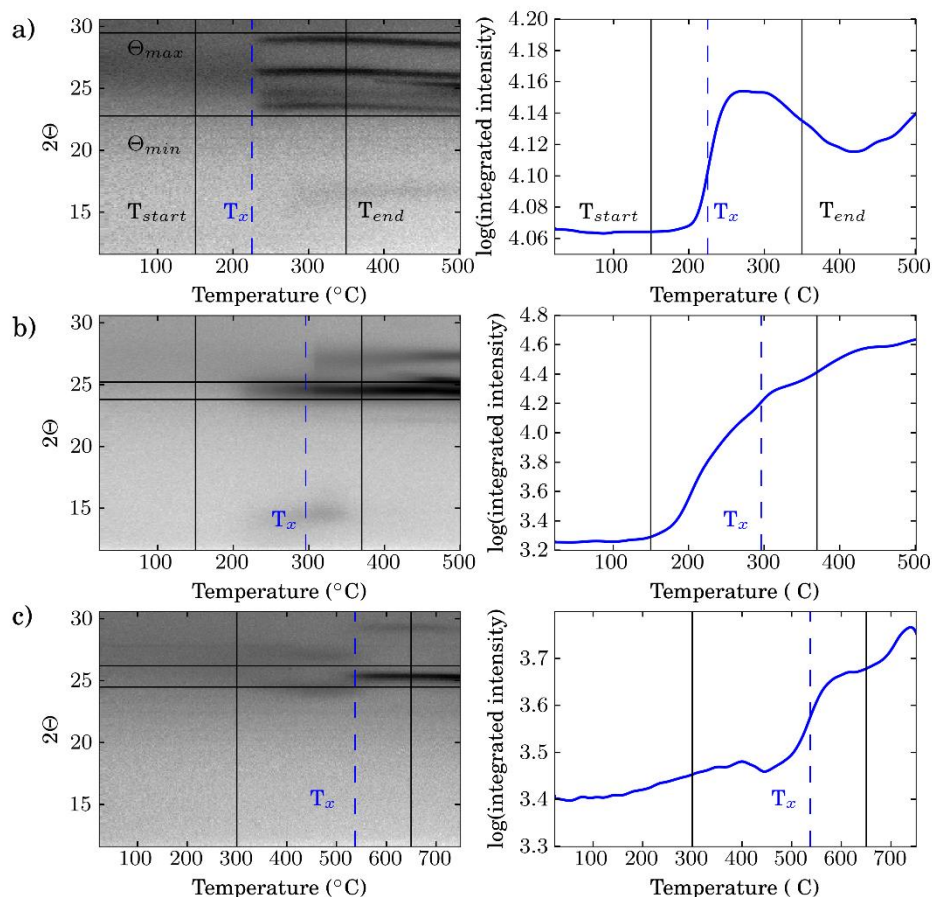
The activation energy for crystallization of a CuTeC20, CuTeGe20 and CuTeSi20 layer is determined by the Kissinger method [1]. The method requires the knowledge of the temperature at which an equivalent stage of reaction occurs for different heating rates. The method assumes a maximum reaction rate at the peak temperature  $T_p$  of crystallization. This also implies a constant degree of conversion at this stage [1]. In this work, in situ XRD is used to determine the crystallization temperature  $T_x$  of the alloy, which is used as  $T_p$ . The 50 nm CuTeC20 and CuTeGe20 layers were deposited by means of magnetron sputter deposition on 20 nm  $Al_2O_3/Si$  substrates, as described elsewhere [2, 3]. The composition of the CuTeC20 layer was verified by means of EDX (as described in [2]) and the composition of the CuTeGe20 layer by XRF (as described in [3]). The compositions are summarized in Table S1. The samples were then annealed at different heating rates ranging from 0.1°C/s to 5°C/s in He atmosphere and the XRD spectrum was recorded. The crystallization temperature  $T_x$  is then defined as the temperature in the temperature interval [ $T_{start}$ ,  $T_{end}$ ] where the maximum intensity change occurs in the selected  $2\Theta$  range [ $\Theta_{min}$ ,  $\Theta_{max}$ ], as is illustrated in Figure S1. This  $2\Theta$  range is selected based on the diffraction peaks of the phase that first crystallizes. This is illustrated in Figure S1 for the three different materials, annealed at a heating rate of 0.5°C/s. Using the relation [1]

$$\ln \left[ \left( \frac{dT}{dt} \right) \frac{1}{T_x^2} \right] = - \frac{E_A}{k_B T_x} + C,$$

the activation energy  $E_A$  can be extracted. In this equation  $(dT/dt)$  is the heating rate,  $T_x$  the crystallization temperature,  $k_B$  the Boltzmann constant and  $C$  is a constant. By plotting  $\ln((dT/dt)/T_x^2)$  as a function of  $1/k_B T_x$ ,  $E_A$  is extracted from a linear fit, as illustrated in Figure 4 in the manuscript.

**Table S1:** Compositions of the layers used for Kissinger analysis.

Material	Cu (at%)	Te (at%)	Alloying element (at%)
CuTeSi20	50.7	31.8	17.5 (Si)
CuTeC20	49.4	32.8	17.8 (C)
CuTeGe20	48	30.5	21.5 (Ge)



**Figure S1:** In situ XRD patterns of a (a) CuTeC<sub>20</sub>, (b) CuTeGe<sub>20</sub> and (c) CuTeSi<sub>20</sub> layer, collected at a heating rate of 0.5°C/s in He. The graphs at the right show the integrated intensity in the [ $\Theta_{min}$ ,  $\Theta_{max}$ ] region as a function of the temperature. The crystallization temperature  $T_x$  is determined as the temperature where the maximum increase in integrated intensity occurs in the temperature range [ $T_{start}$ ,  $T_{end}$ ].

## 2. References

- [1] Advances in Crystallization Processes, Edited by Y. Mastai, 2012
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- [3] Devulder, W.; Opsomer, K.; Rampelberg, G.; De Schutter, B.; Devloo-Casier, K.; Jurczak, M.; Goux, L. & Detavernier, C., *J. Mater. Chem. C*, **2015**, 3, 12469-12476