

## Supporting Information

**Introduction of a novel nano sized *N*-Sulfonated Brönsted acidic catalyst for the  
promotion of the synthesis of polyhydroquinoline derivatives *via* Hantzsch  
condensation under solvent-free conditions**

Omid Goli-Jolodar, Farhad Shirini\* and Mohadeseh Seddighi

Department of Chemistry, Faculty of Sciences, University of Guilan, Rasht, zip code: 41335.

Post Box: 1914, I. R. Iran. Tel./Fax: +98 131 3233262, E-mail address: shirini@guilan.ac.ir

## Experimental

### Material

Chemicals were purchased from Fluka, Merck, Aldrich and Southern Clay Products Chemical Companies. Yields refer to isolated products. Products were characterized by their physical constants, comparison with authentic samples and FT-IR and NMR spectroscopy. The purity determination of the substrate and reaction monitoring were accomplished by TLC on silics-gel polygram SILG/UV 254 plates.

### Instrumentation

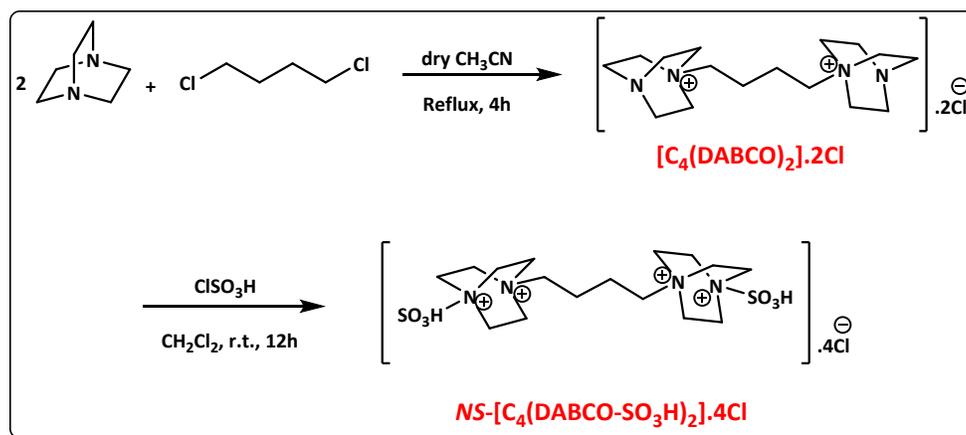
The FT-IR spectra were run on a VERTEX 70 Bruker company (Germany). Thermogravimetric analyses (TGA) were performed on TG/DTA6300 SII-Nonotechnology Company (Japan). Samples were heated from 25 to 700 °C at ramp 10 °C/min under N<sub>2</sub> atmosphere. Scanning electron microphotographs (SEM) were obtained on a SEM-Philips XL30. Wide-angle X-ray diffraction (XRD) measurements were performed at room temperature on a Siemens D-500 X-ray diffractometer (Germany), using Ni-filtered Co-K $\alpha$  radiation ( $\lambda=0.15418$  nm). The <sup>1</sup>H NMR (400 MHz) and <sup>13</sup>C NMR (100 MHz) were run on a Bruker AVANCE<sup>III</sup>-400 spectrometer in DMSO using TMS as an internal reference ( $\delta$  in ppm). The surface morphologies were characterized by atomic force microscope (AFM, Ara nanoscope, Iran).

### General procedure for the Synthesis of 1,1'-(butane-1,4-diyl)bis(1,4-diazabicyclo[2.2.2]octan-1-ium) chloride ([C<sub>4</sub>(DABCO)<sub>2</sub>].2Cl):

1,4-Dichlorobutane (0.547 mL, 5.0 mmol) was added to a solution DABCO (1.121 g, 10.0 mmol) in dry CH<sub>3</sub>CN (50 mL) and stirred for 4 hr under reflux conditions. After completion of the reaction the solvent was removed under vacuum. The white obtained solid was washed with diethyl ether. After drying at 50 °C 1,1'-(butane-1,4-diyl)bis(1,4-diazabicyclo[2.2.2]octan-1-ium)chloride ([C<sub>4</sub>(DABCO)<sub>2</sub>].2Cl) was obtained as a white solid in 98.7% yield (1.39 g).

**General procedure for the Synthesis of nano sized 4,4'-(butane-1,4-diyl)bis(1-sulfo-1,4-diazabicyclo[2.2.2]octane-1,4-dium) chloride ([NS-C<sub>4</sub>(DABCO-SO<sub>3</sub>H)<sub>2</sub>]<sub>2</sub>).4Cl):**

Chlorosulfonic acid (0.66 mL, 10 mmol) was added drop wise to a solution of C<sub>4</sub>(DABCO)<sub>2</sub>.2Cl (1.4 g, 5 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (25 mL) over a period of 30 min in an ice bath. After completion of the addition, the reaction mixture was stirred for 12h at room temperature. The residue was washed with dry diethyl ether (3 × 5 mL) and dried under vacuum to give nano sized 4,4'-(butane-1,4-diyl)bis(1-sulfo-1,4-diazabicyclo[2.2.2]octane-1,4-dium)chloride ([NS-C<sub>4</sub>(DABCO-SO<sub>3</sub>H)<sub>2</sub>]<sub>2</sub>).4Cl as a white puffy solid in 98.7% yield (4.42 g).



**Scheme 1.** Synthesis of the NS-[C<sub>4</sub>(DABCO-SO<sub>3</sub>H)<sub>2</sub>]<sub>2</sub>.4Cl.

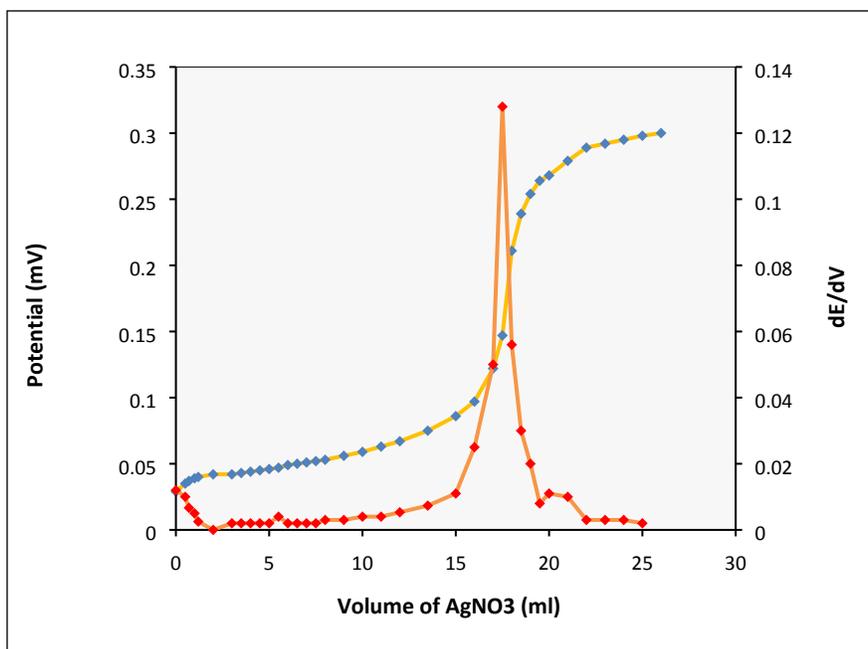


Fig. 1. Potentiometric titration and its first derivative curves of  $[C_4(DABCO)_2].2Cl$  with  $AgNO_3$ .

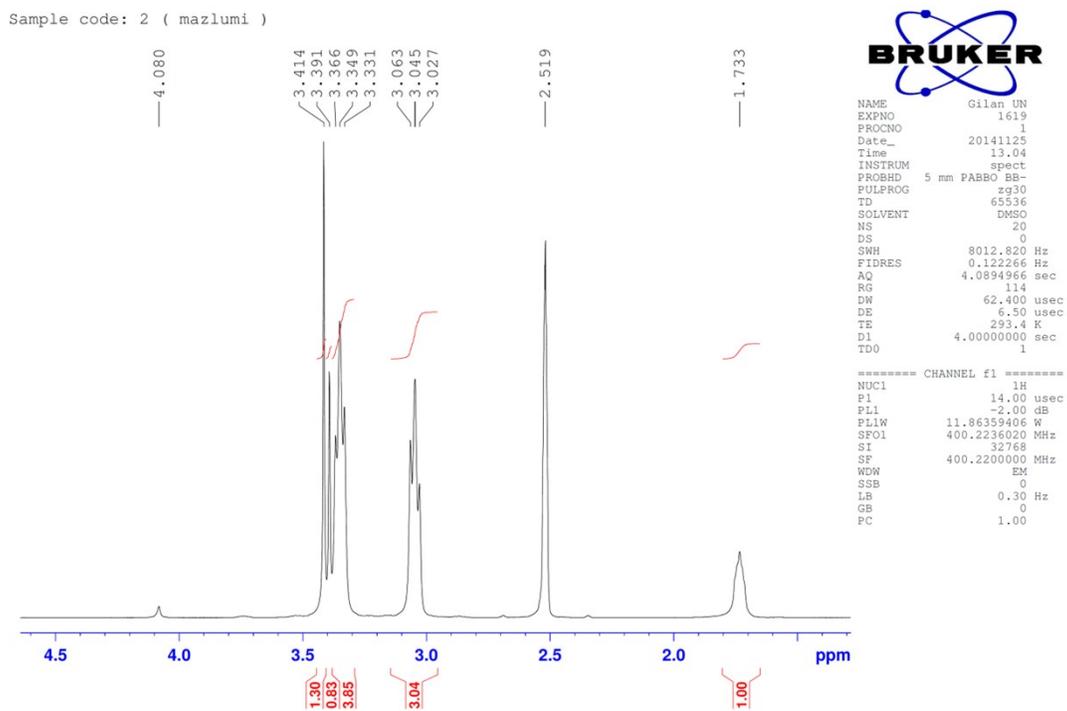


Fig. 2.  $^1H$  NMR spectra of  $[C_4(DABCO)_2].2Cl$ .

Sample code: 1 (gholi)

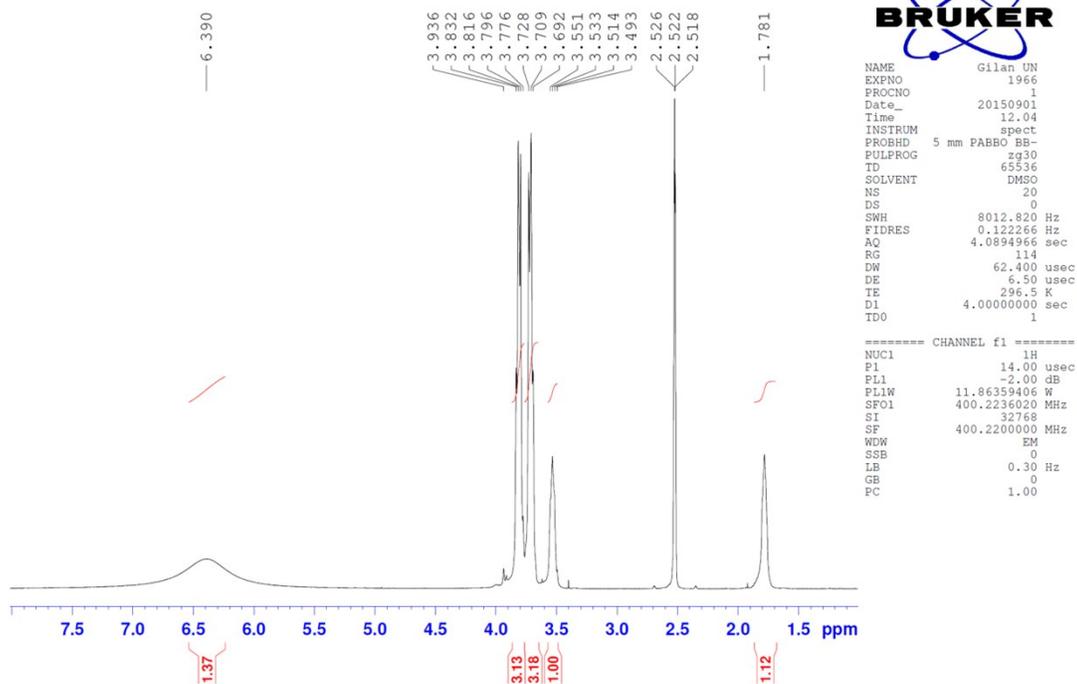
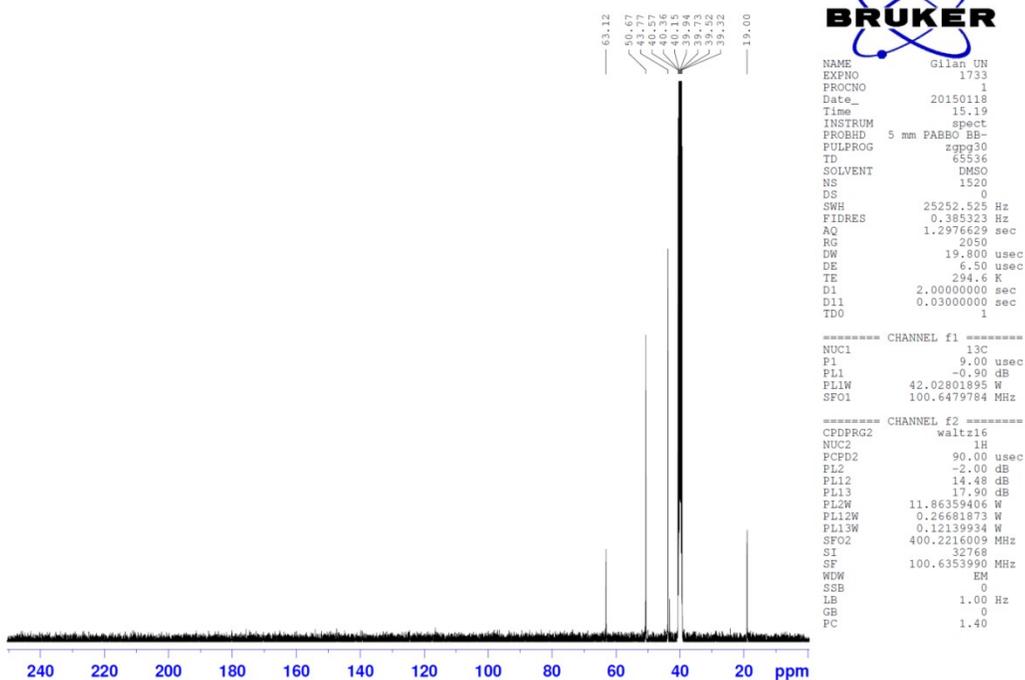
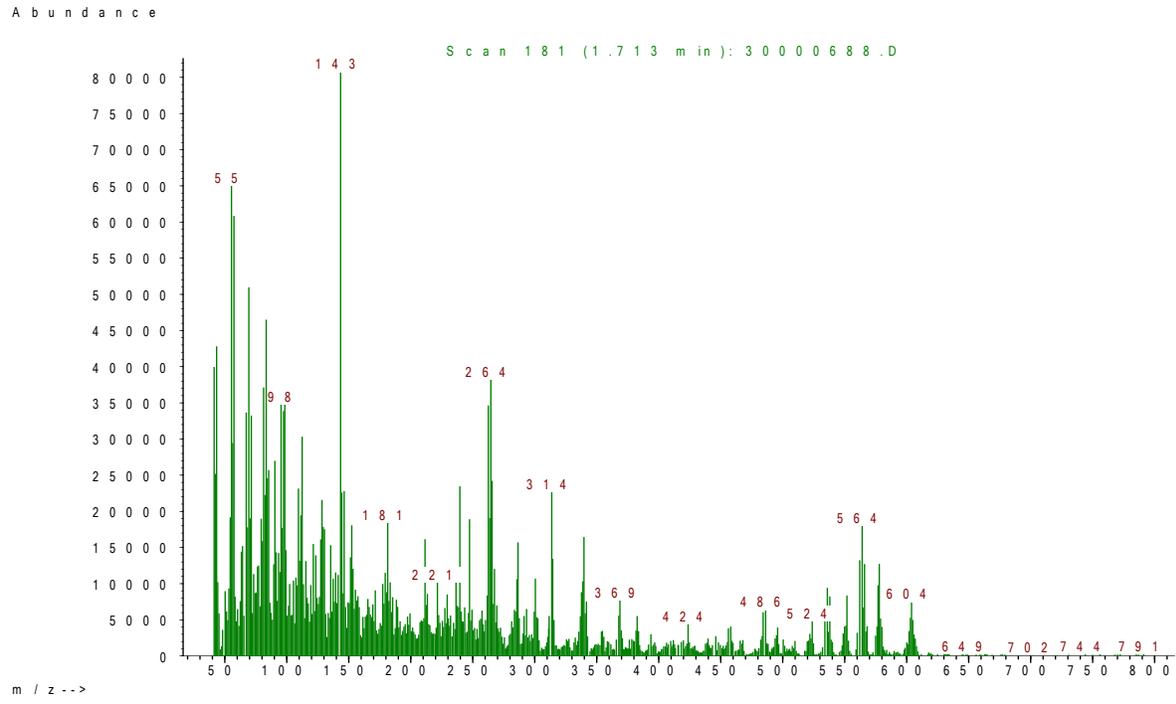


Fig. 3.  $^1\text{H}$  NMR spectra of  $\text{NS}[\text{C}_4(\text{DABCO-SO}_3\text{H})_2]_2 \cdot 4\text{Cl}$ .

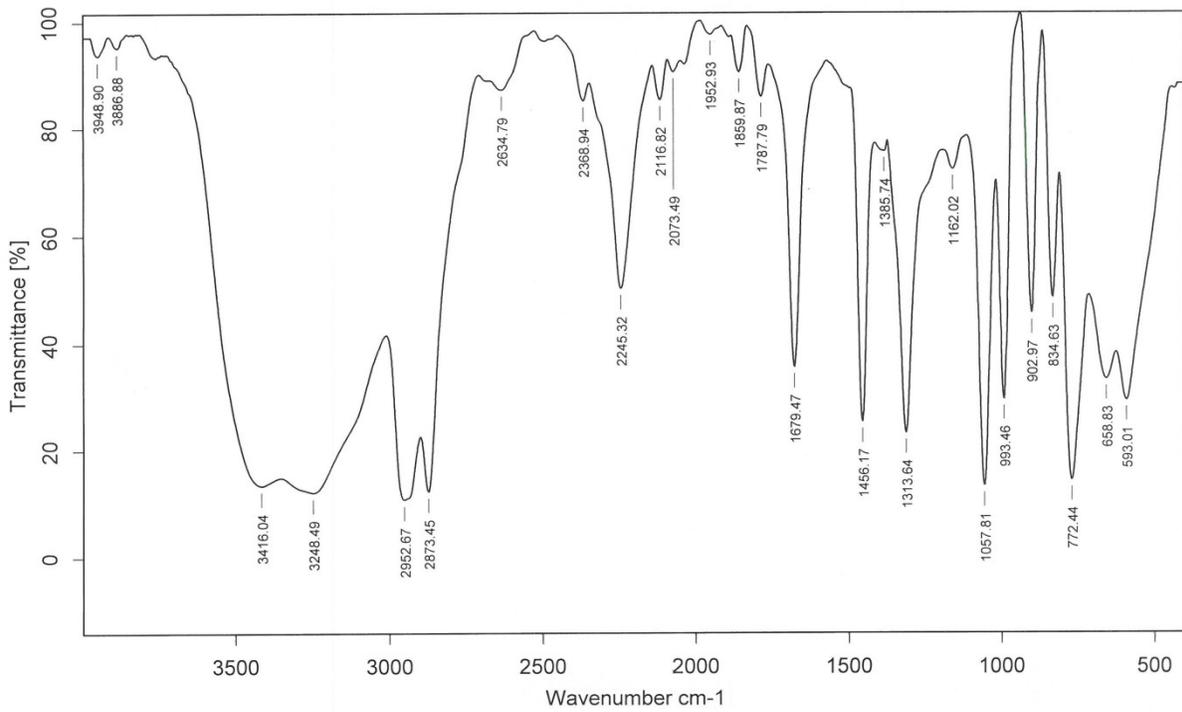
Sample code: DC (goli)



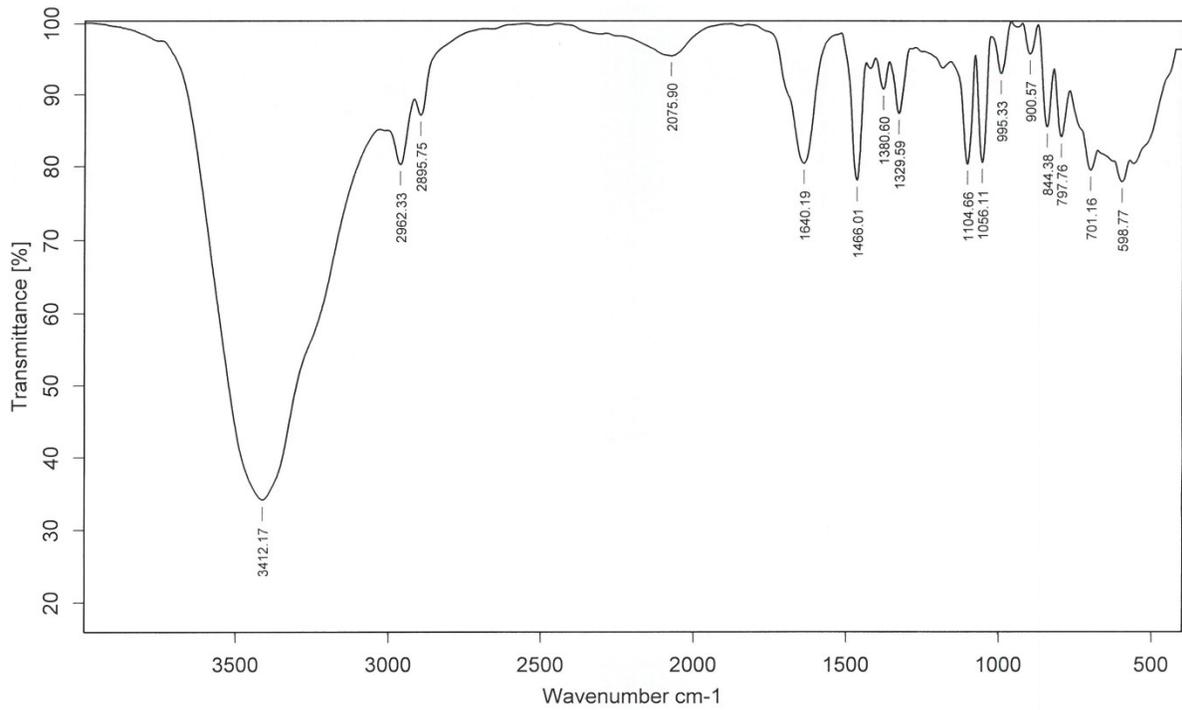
**Fig. 4.**  $^{13}\text{C}$  NMR spectra of NS-[C<sub>4</sub>(DABCO-SO<sub>3</sub>H)<sub>2</sub>].4Cl.



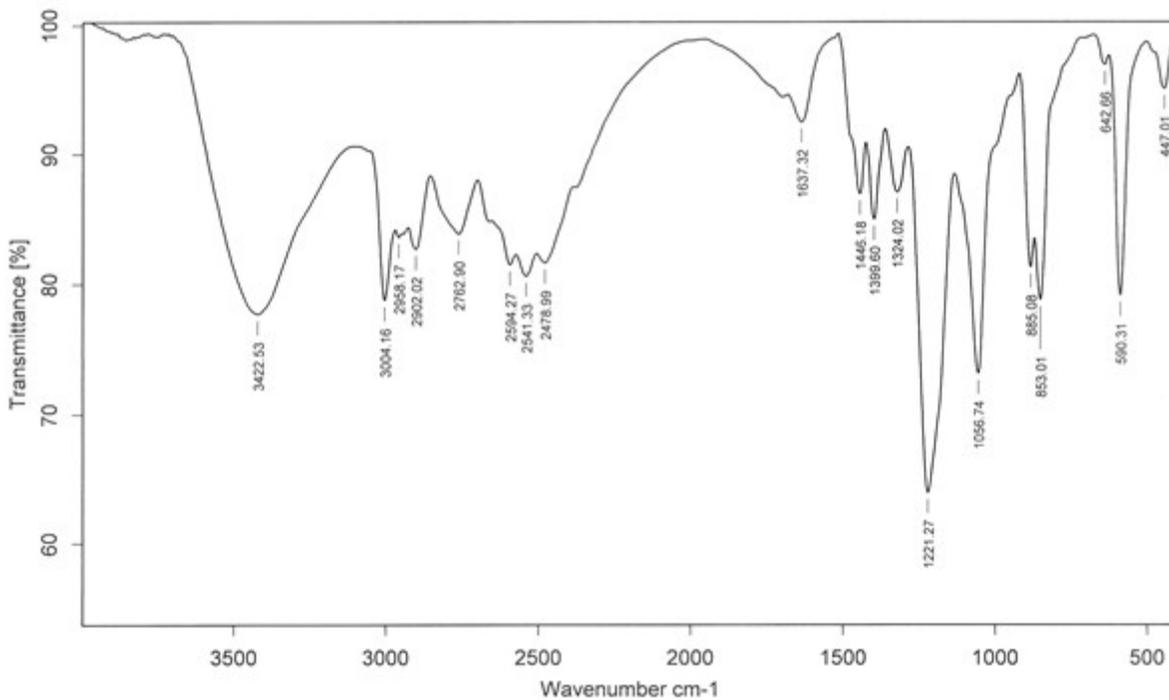
**Fig. 5.** Mass data of nano sized NS-[C<sub>4</sub>(DABCO-SO<sub>3</sub>H)<sub>2</sub>].4Cl.



**Fig. 6.** FT-IR spectra of DABCO.



**Fig. 7.** FT-IR spectra of  $[C_4(DABCO)_2] \cdot 2Cl$ .



**Fig. 8.** FT-IR spectra of  $NS-[C_4(DABCO-SO_3H)_2] \cdot 4Cl$  (c).

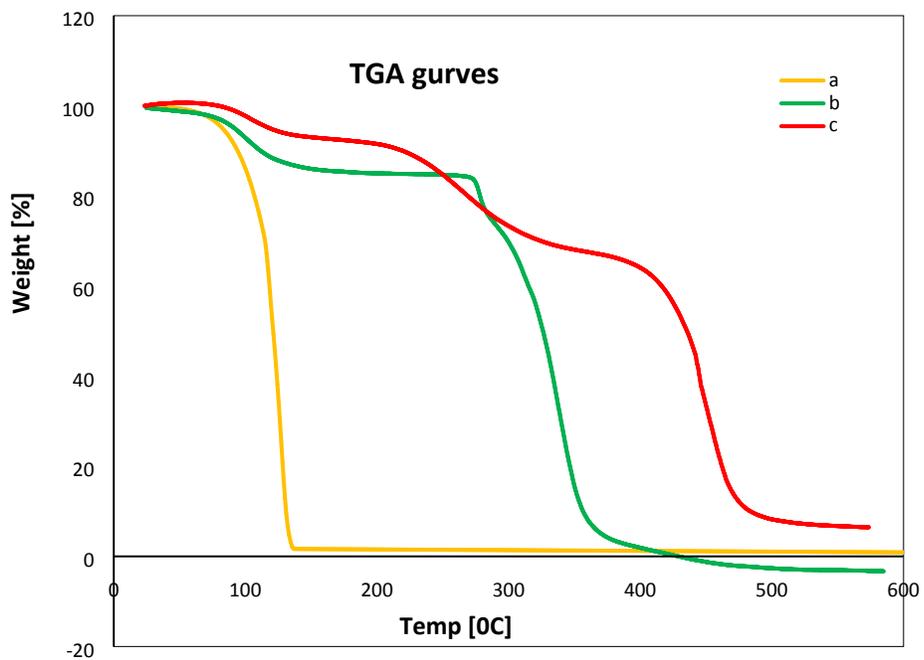


Fig. 9. TGA curves for DABCO,  $[C_4(DABCO)_2].2Cl$  and  $NS-[C_4(DABCO-SO_3H)_2].4Cl$ .

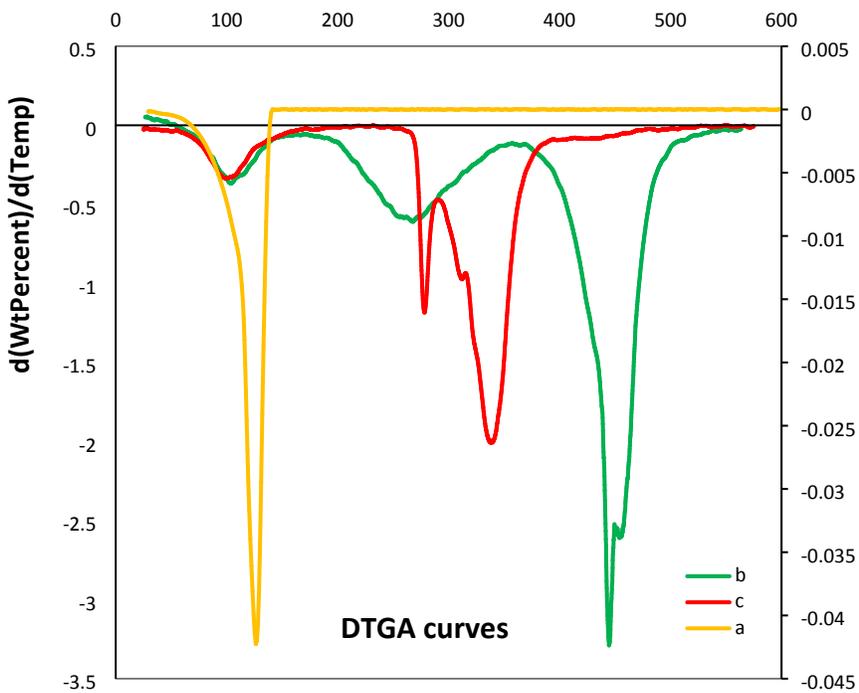
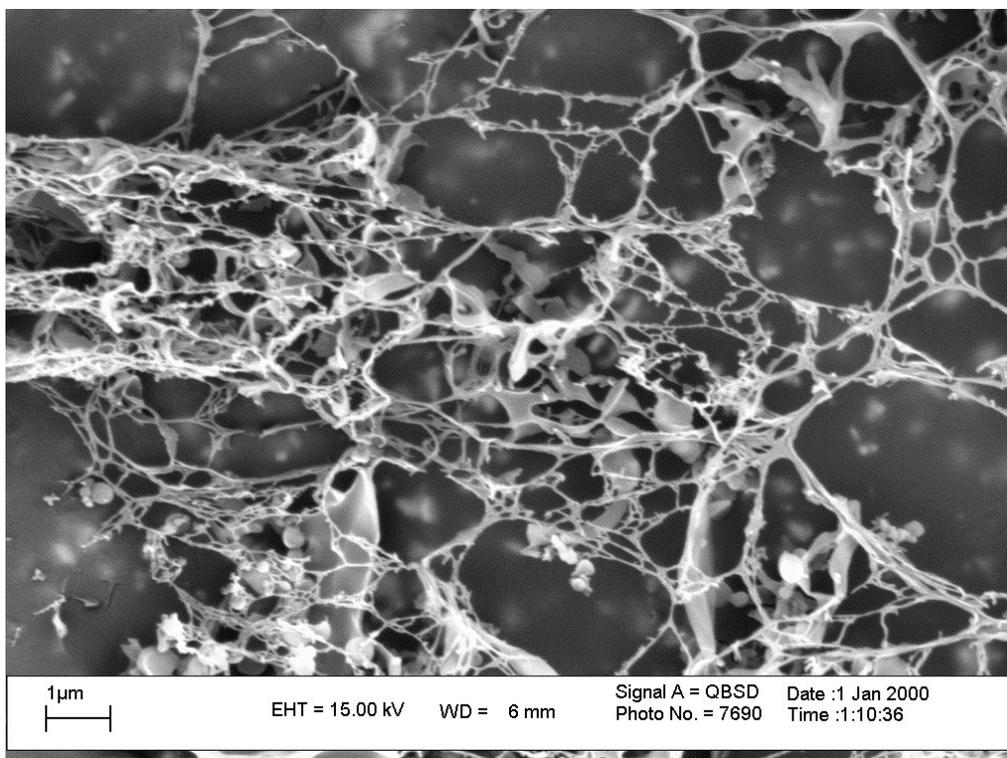
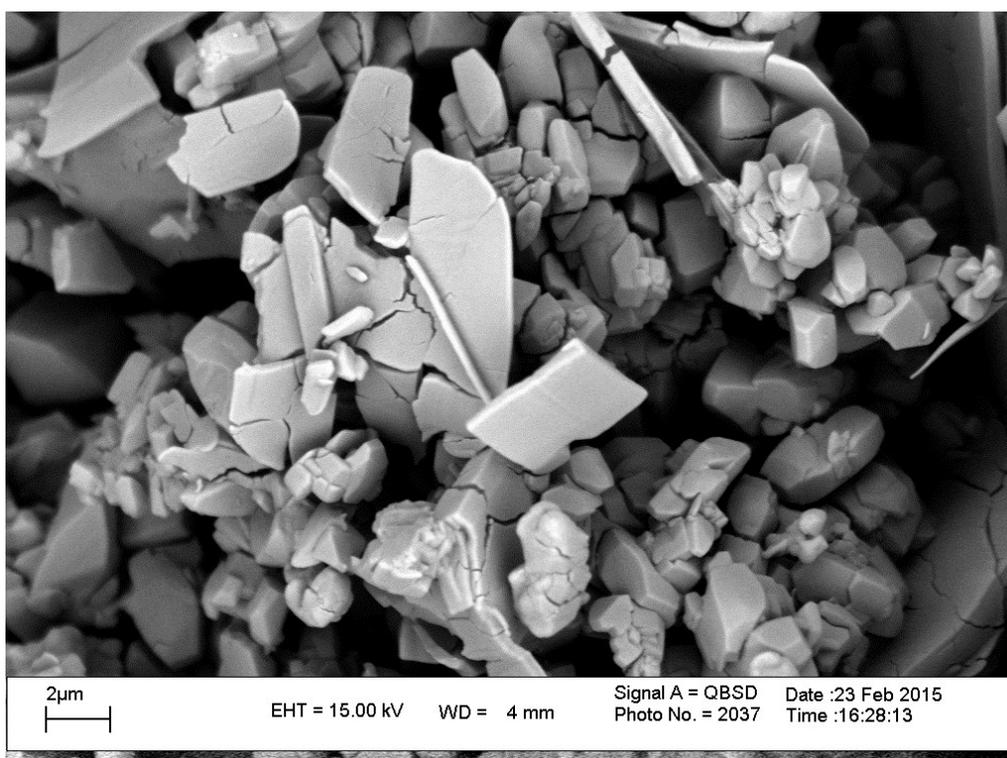


Fig. 10. DTGA curves for DABCO,  $[C_4(DABCO)_2].2Cl$  and  $NS-[C_4(DABCO-SO_3H)_2].4Cl$ .

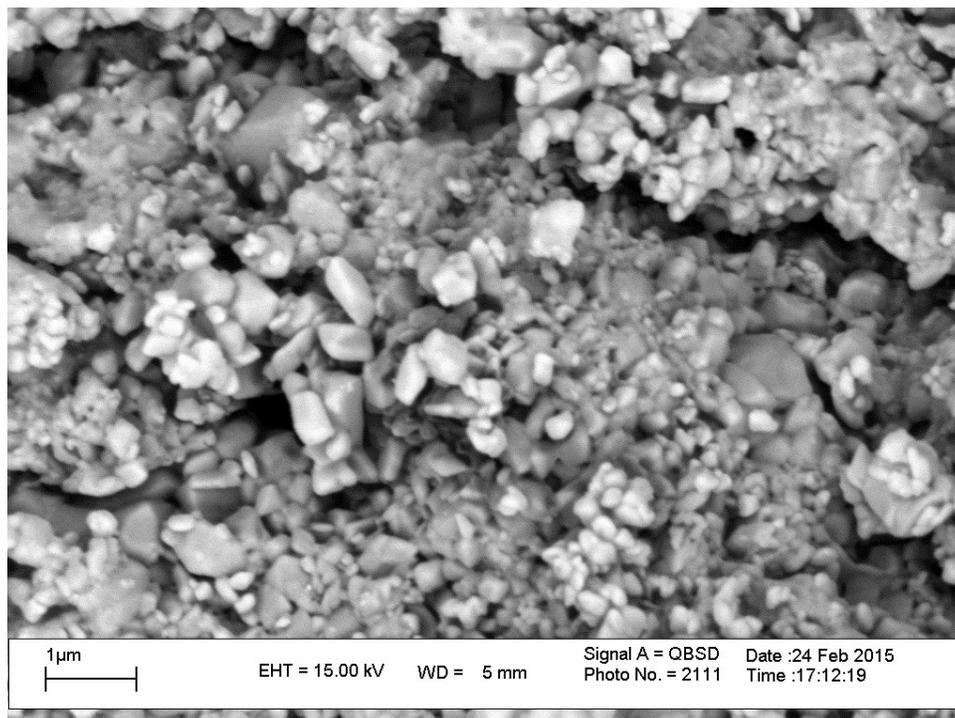


**Fig. 11.** SEM micrographs of DABCO.

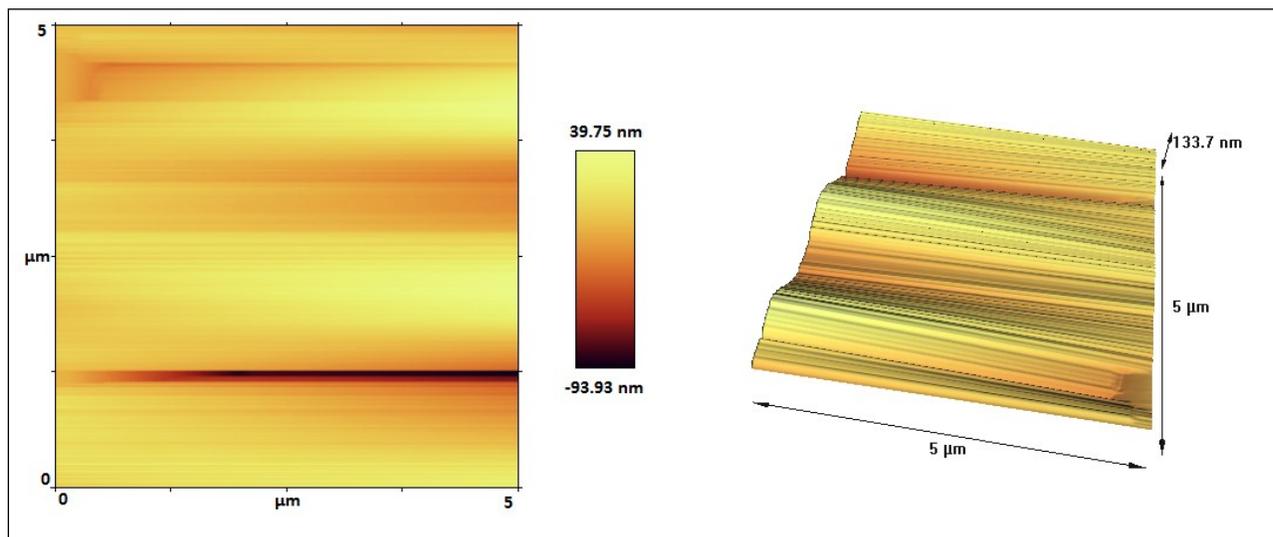
---



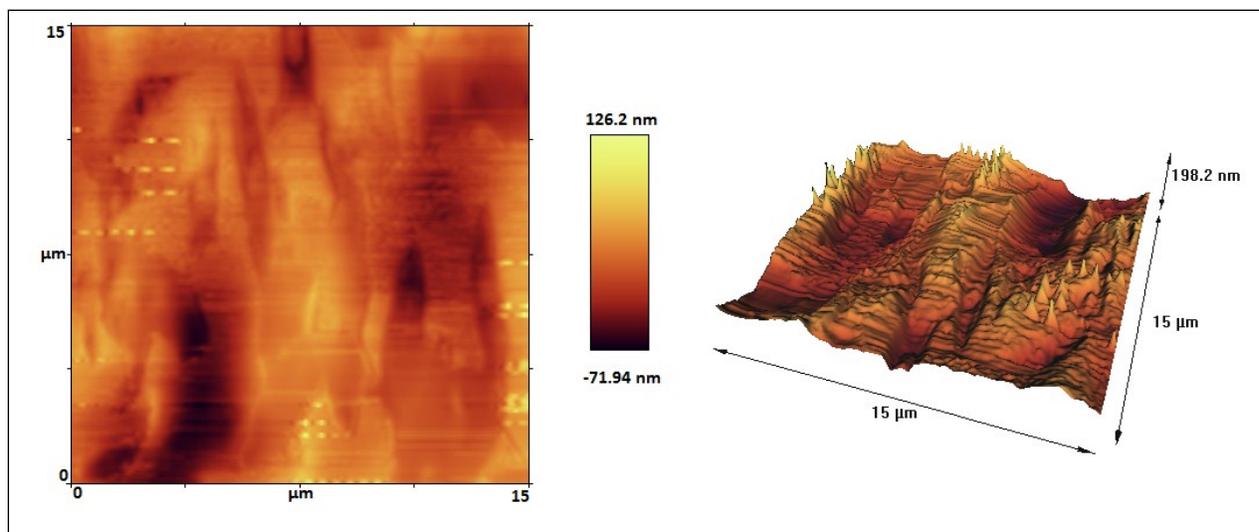
**Fig. 12.** SEM micrographs of  $[C_4(DABCO)_2] \cdot 2Cl$ .



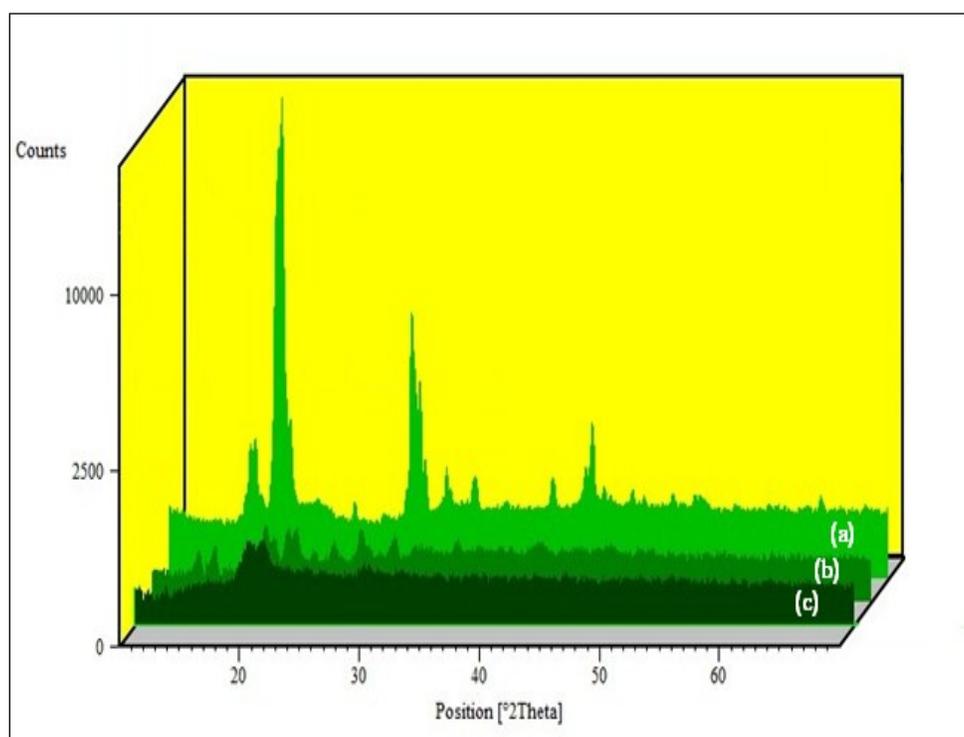
**Fig. 13.** SEM micrographs of NS- $[C_4(DABCO-SO_3H)_2] \cdot 4Cl$ .



**Fig. 14.** Atomic force microscopy (AFM) images in two- and three dimensions for  $[C_4(DABCO)_2] \cdot 2Cl$ .



**Fig. 15.** Atomic force microscopy (AFM) images in two- and three dimensions for  $[C_4(DABCO-SO_3H)_2].4Cl$ .



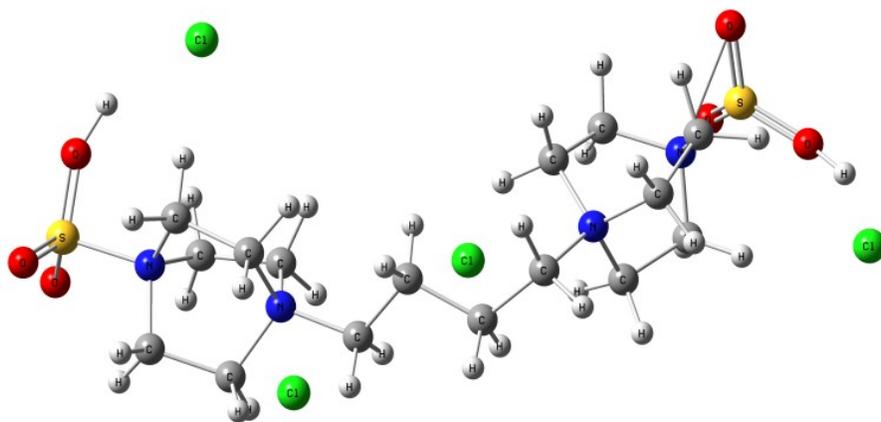
**Fig. 16.** XRD patterns of DABCO (a),  $[C_4(DABCO)_2].2Cl$  (b) and  $NS-[C_4(DABCO-SO_3H)_2].4Cl$  (c).

**Table 1.** XRD data of DABCO (1),  $[C_4(DABCO)_2].2Cl$  (2) and  $NS-[C_4(DABCO-SO_3H)_2].4Cl$  (3).

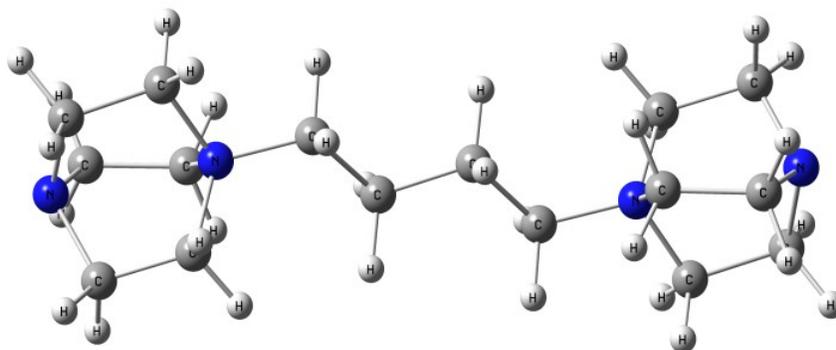
---

Entry	2 $\theta$	Peak width [FWHM] (degree)	Size [nm]
1	19.39	0.157	51.37
2	22.08	0.393	20.52
3	19.4	1.152	7

---

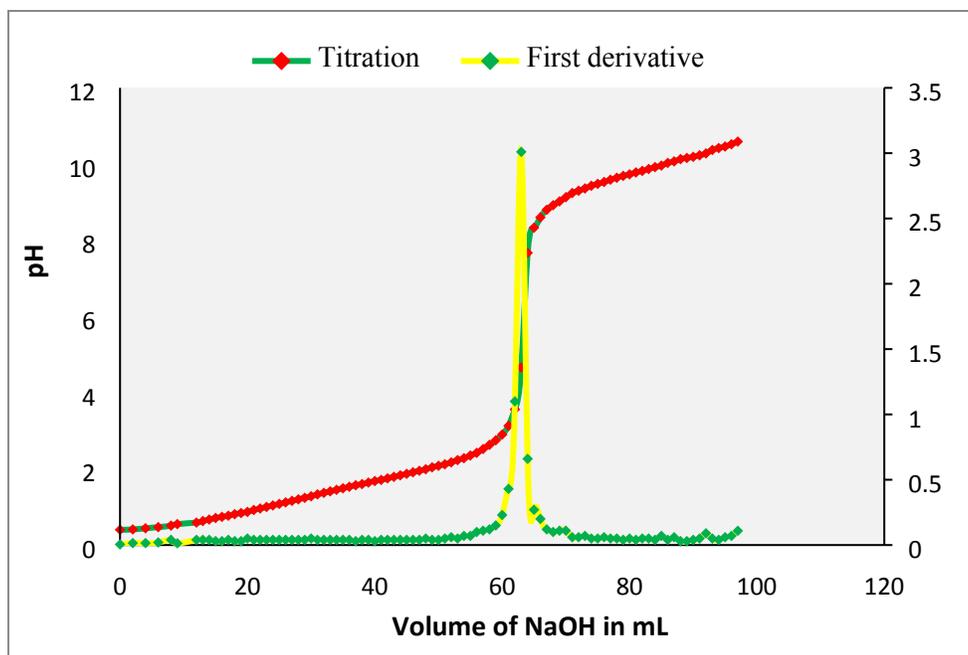


**Fig. 17.** The optimized geometry of NS-[C<sub>4</sub>(DABCO-SO<sub>3</sub>H)<sub>2</sub>].4Cl.



**Fig. 18.** The optimized geometry of [C<sub>4</sub>(DABCO)<sub>2</sub>].2Cl.

---



**Fig. 19.** Titration and its first derivative curves of NS-[C<sub>4</sub>(DABCO-SO<sub>3</sub>H)<sub>2</sub>].4Cl by NaOH.

#### **General procedure for the Synthesis of polyhydroquinolines:**

In a round-bottomed flask a mixture of aldehyde (1 mmol), 1,3-cyclohexanedione and/or dimedone (1 mmol), ammonium acetate (1.5 mmol), a  $\beta$ -ketoester derivative (1 mmol) and [NS-C<sub>4</sub>(DABCO-SO<sub>3</sub>H)<sub>2</sub>].4Cl was stirred in the different conditions that the best result was obtained in the presence of 5 mol% [NS-C<sub>4</sub>(DABCO-SO<sub>3</sub>H)<sub>2</sub>].4Cl and 100 °C. After completion of the reaction, as monitored by TLC (eluent: EtOAc:*n*-hexane), water was added to separate the catalyst and the crude product was separated and recrystallized by EtOH.

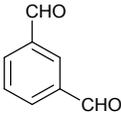
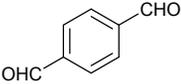
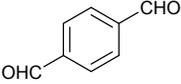
**Table 2.** Effect of the catalyst amount and temperature on the reaction between dimedone, 4-chlorobenzaldehyde, ethyl acetoacetate and ammonium acetate.

Entry	Catalyst amount (mol%)	Temp. (°C)	Time (min)	Conversion (%) <sup>a</sup>
1	2.5	25	60	60
2	2.5	80	45	75
3	2.5	100	40	80
4	5	25	40	80
5	5	80	30	100
<b>6</b>	<b>5</b>	<b>100</b>	<b>18</b>	<b>100</b>
7	10	100	20	100
8	-	100	6 h	20
9	DABCO (5)	100	6 h	23
10	[C <sub>4</sub> (DABCO) <sub>2</sub> ].2Cl (5)	100	6 h	60

<sup>a</sup> TLC or GC yields.

**Table 3.** Synthesis of polyhydroquinoline derivatives from dimedone, aryl aldehydes,  $\beta$ -ketoesters and ammonium acetate catalyzed by NS-[C<sub>4</sub>(DABCO-SO<sub>3</sub>H)<sub>2</sub>].4Cl at 100 °C under solvent-free condition.

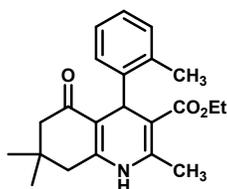
Entry	Aldehyde	R'	R''	Product	Time (min)	Yield (%) <sup>a</sup>	M.p. (°C)	
							Found	Reported <sup>ref.</sup>
1	C <sub>6</sub> H <sub>5</sub> –	CH <sub>3</sub>	OEt	a	8	98	220-222	224-226 <sup>42</sup>
2	2-Cl-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	b	10	95	202-204	207-209 <sup>42</sup>
3	2-OCH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	c	7	98	248-250	257-259 <sup>42</sup>
4	2-NO <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	d	25	87	200-202	204-206 <sup>42</sup>
5	2-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	e	4	98	210-212	-
6	3-Br-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	f	7	97	229-231	230-232 <sup>43</sup>
7	3-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	g	8	97	198-200	201-203 <sup>42</sup>
8	3-NO <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	h	22	90	170-172	173-175 <sup>42</sup>
9	4-Cl-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	i	18	92	241-243	245-246 <sup>42</sup>
10	4-Br-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	j	10	97	248-252	255-256 <sup>43</sup>
11	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	k	15	95	254-256	258-260 <sup>42</sup>
12	4-NO <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	l	27	87	238-240	242-244 <sup>42</sup>
13	4-OH-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	m	4	98	227-229	228-230 <sup>42</sup>
14	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	n	2	98	259-261	256-258 <sup>42</sup>
15	4-CN-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	o	15	90	140-142	140-142 <sup>42</sup>
16	4-CH <sub>3</sub> S-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	p	15	90	239-241	240-242 <sup>44</sup>
17	4-(CH <sub>3</sub> ) <sub>2</sub> N-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	q	50	88	231-233	233-235 <sup>42</sup>
18	4-Isopro-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OEt	r	40	85	179-181	182-184 <sup>42</sup>
19	C <sub>6</sub> H <sub>5</sub> –	H	OEt	s	10	95	250-252	240-242 <sup>45</sup>
20	4-Cl-C <sub>6</sub> H <sub>4</sub> –	H	OEt	t	5	98	243-246	234-236 <sup>45</sup>
21	4-Br-C <sub>6</sub> H <sub>4</sub> –	H	OEt	u	5	98	254-255	253-254 <sup>45</sup>
22	4-OH-C <sub>6</sub> H <sub>4</sub> –	H	OEt	v	3	98	244-246	222-224 <sup>45</sup>
23	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> –	H	OEt	w	5	98	194-195	192-193 <sup>17</sup>
24	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> –	H	OEt	x	12	95	251-254	241-243 <sup>45</sup>
25	4-(CH <sub>3</sub> ) <sub>2</sub> N-C <sub>6</sub> H <sub>4</sub> –	H	OEt	y	50	88	204-206	—
26	4-Cl-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OMe	z	20	89	220-221	220-221 <sup>46</sup>
27	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OMe	aa	5	98	248-252	249-251 <sup>46</sup>
28	4-CN-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OMe	bb	5	98	219-221	220-222 <sup>46</sup>
29	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OMe	cc	3	98	267-269	274-276 <sup>46</sup>
30	4-(CH <sub>3</sub> ) <sub>2</sub> N-C <sub>6</sub> H <sub>4</sub> –	CH <sub>3</sub>	OMe	dd	40	93	258-259	257-258 <sup>12</sup>
31	Butyraldehyde	CH <sub>3</sub>	OEt	ff	45	80	147-149	147-149 <sup>42</sup>
32	Isobutyraldehyde	CH <sub>3</sub>	OEt	gg	40	80	160-162	161-163 <sup>42</sup>
33	Furan-2-carbaldehyde	H	OEt	hh	5	87	207-209	210-212 <sup>19</sup>

34	Furan-2-carbaldehyde	CH <sub>3</sub>	OEt	ii	4	87	246-248	245-247 <sup>42</sup>
35		CH <sub>3</sub>	OEt	ll	15	100	281-282	—
36		H	OEt	mm	5	85	293-294	—
37		CH <sub>3</sub>	OEt	nn	7	98	305-307	294-296 <sup>43</sup>

<sup>a</sup> Isolated yields.

**The spectral data of the new compounds are as follow:**

**Ethyl 2,7,7-trimethyl-5-oxo-4-(o-tolyl)-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (e):** White solid., m.p. 210-212 °C., FT-IR (KBr, cm<sup>-1</sup>) : 3302, 3070, 2962, 1698, 1606, 1481, 1377, 1275, 1212, 1065 cm<sup>-1</sup>., <sup>1</sup>H NMR (DMSO-d<sub>6</sub>, 400 MHz): 0.8 (s, 3H, CH<sub>3</sub>), 1(S, 3H, CH<sub>3</sub>), 1.1 (t, *J*=7.2 Hz, 3H, CH<sub>3</sub>), 1.9 (d, *J*=16 Hz, 1H, CH<sub>2</sub>), 2.1 (d, *J*=16 Hz, 1H, CH<sub>2</sub>), 2.27 (d, *J*=17.2 Hz, 1H, CH<sub>2</sub>), 2.29 (s, 3H), 2.43 (d, *J*=17.2 Hz, 1H), 2.62 (s, 3H), 3.9-4.00 (m, 2H, OCH<sub>2</sub>), 4.93(s, 1H, CH), 6.91-6.96 (m, 2H, ArH), 7 (t, *J*=7.6 Hz, 1H, ArH), 7.12(d, *J*=7.6 Hz, 1H, ArH), 9.04 (NH, 1H).



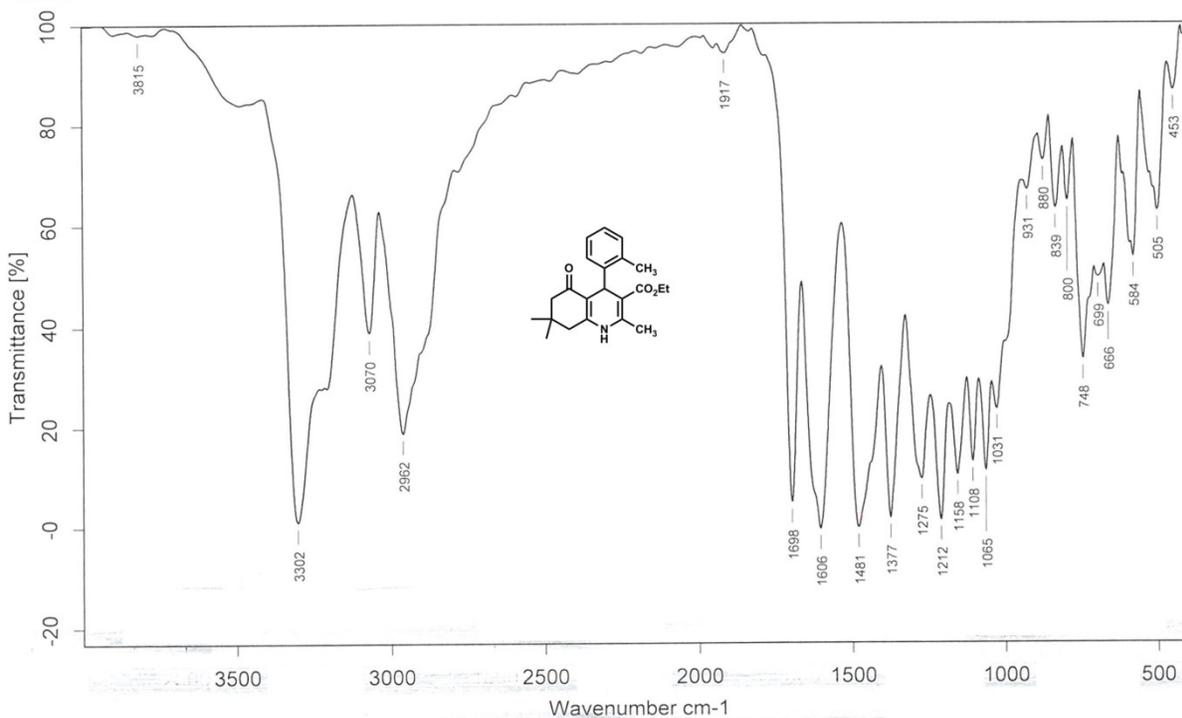
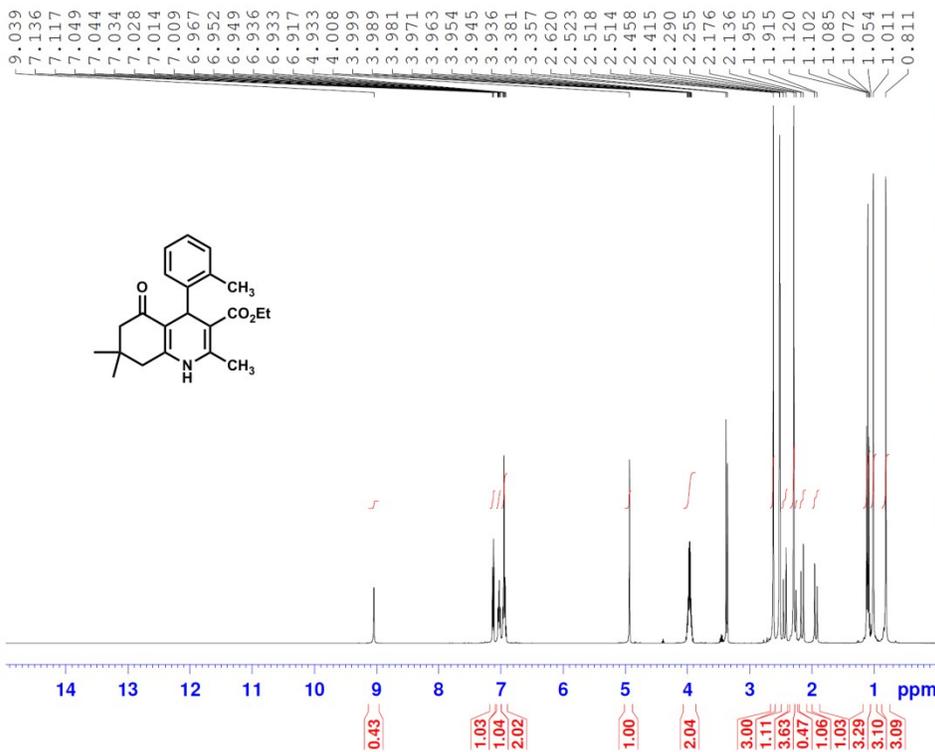


Fig. 20. FT-IR of ethyl 2,7,7-trimethyl-5-oxo-4-(o-tolyl)-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

Sample code: 1



**BRUKER**

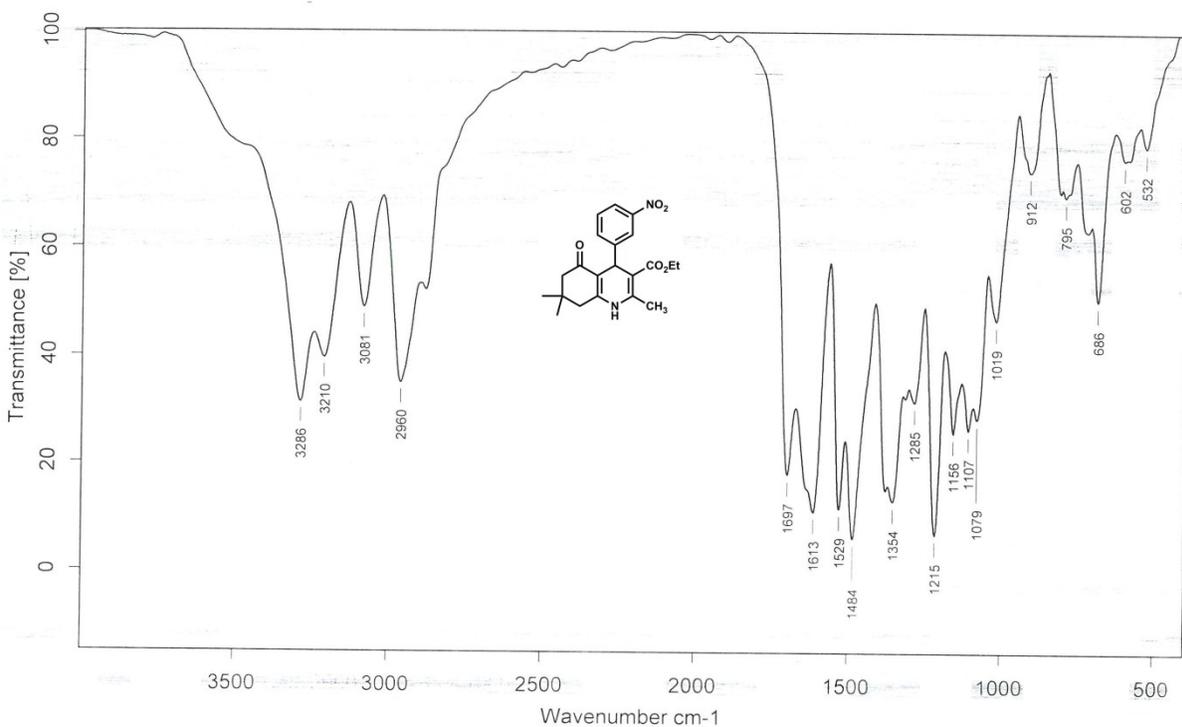
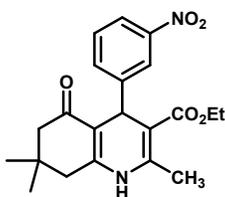
NAME	Gilan UN
EXPNO	1638
PROCNO	1
Date_	20141201
Time_	7.45
INSTRUM	spect
PROBHD	5 mm PABBO BB-
PULPROG	zg30
TD	65536
SOLVENT	DMSO
NS	20
DS	0
SWH	8012.820 Hz
FIDRES	0.122266 Hz
AQ	4.0894966 sec
RG	90.5
DM	62.400 usec
DE	6.50 usec
TE	292.9 K
D1	4.00000000 sec
TDO	1

----- CHANNEL f1 -----

NUC1	1H
P1	14.00 usec
PL1	-2.00 dB
PL1W	11.86359406 W
SFO1	400.2236020 MHz
SI	32768
SF	400.2200000 MHz
WDW	EM
SSB	0
LB	0.30 Hz
GB	0
PC	1.00

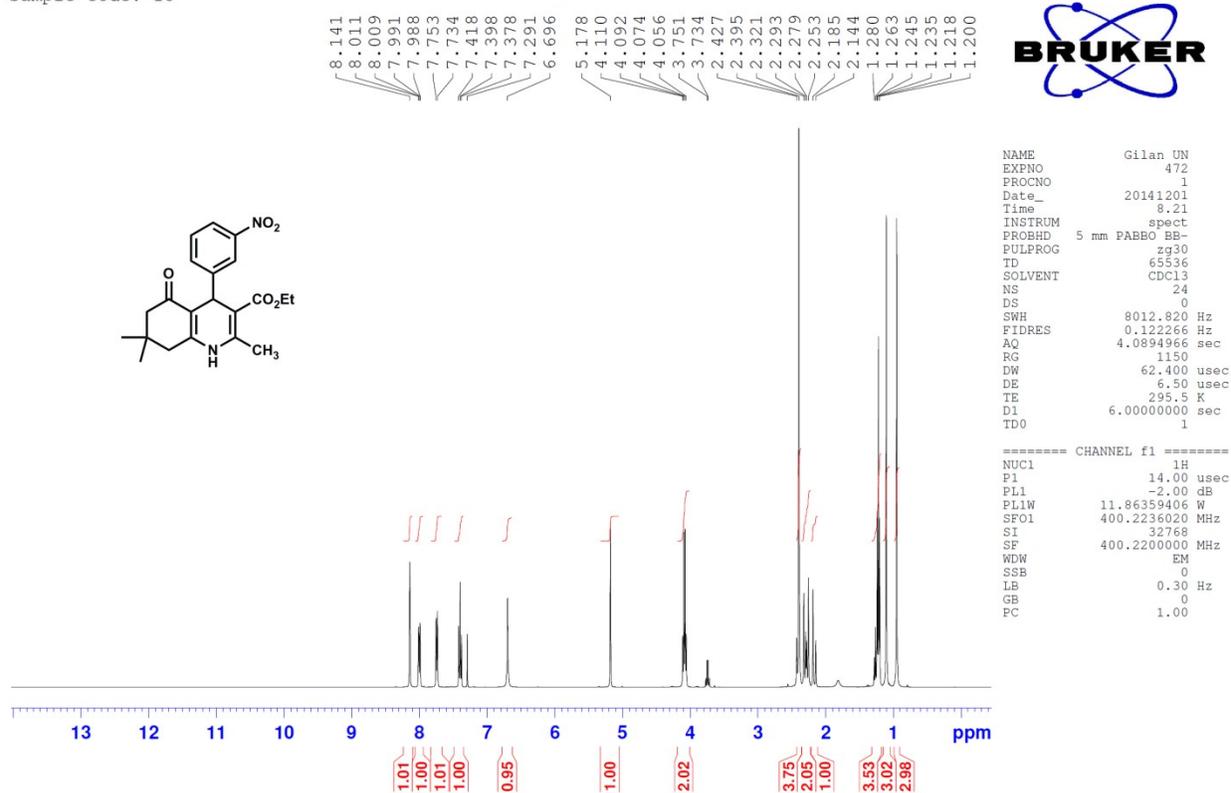
**Fig. 21.** H-NMR of ethyl 2,7,7-trimethyl-5-oxo-4-(*o*-tolyl)-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

**Ethyl-4-(3-nitrophenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (h):** Pale yellow., m.p.= 170-172 °C., FT-IR (KBr,  $\text{cm}^{-1}$ ): 3286, 3210, 3081, 2960, 1697, 1613, 1529, 1484, 1354, 1215, 1079, 686.,  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  0.949 (s, 3H,  $\text{CH}_3$ ), 1.03 (s, 3H,  $\text{CH}_3$ ), 1.22 (t,  $J=7.2$  Hz, 3H,  $\text{CH}_3$ ), 2.144-2.427 (m, 7H), 4.08 (q,  $J=7.2$  Hz, 2H,  $\text{OCH}_2$ ), 5.178 (s, 1H, CH), 6.696 (s, 1H, ArH), 7.398 (t,  $J=8$  Hz, 1H, ArH), 7.74 (d,  $J=8$  Hz, 1H, ArH), 8 (d,  $J=8$  Hz, 1H, ArH), 8.141 (s, 1H, NH).



**Fig. 22.** FT-IR of ethyl-4-(3-nitrophenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

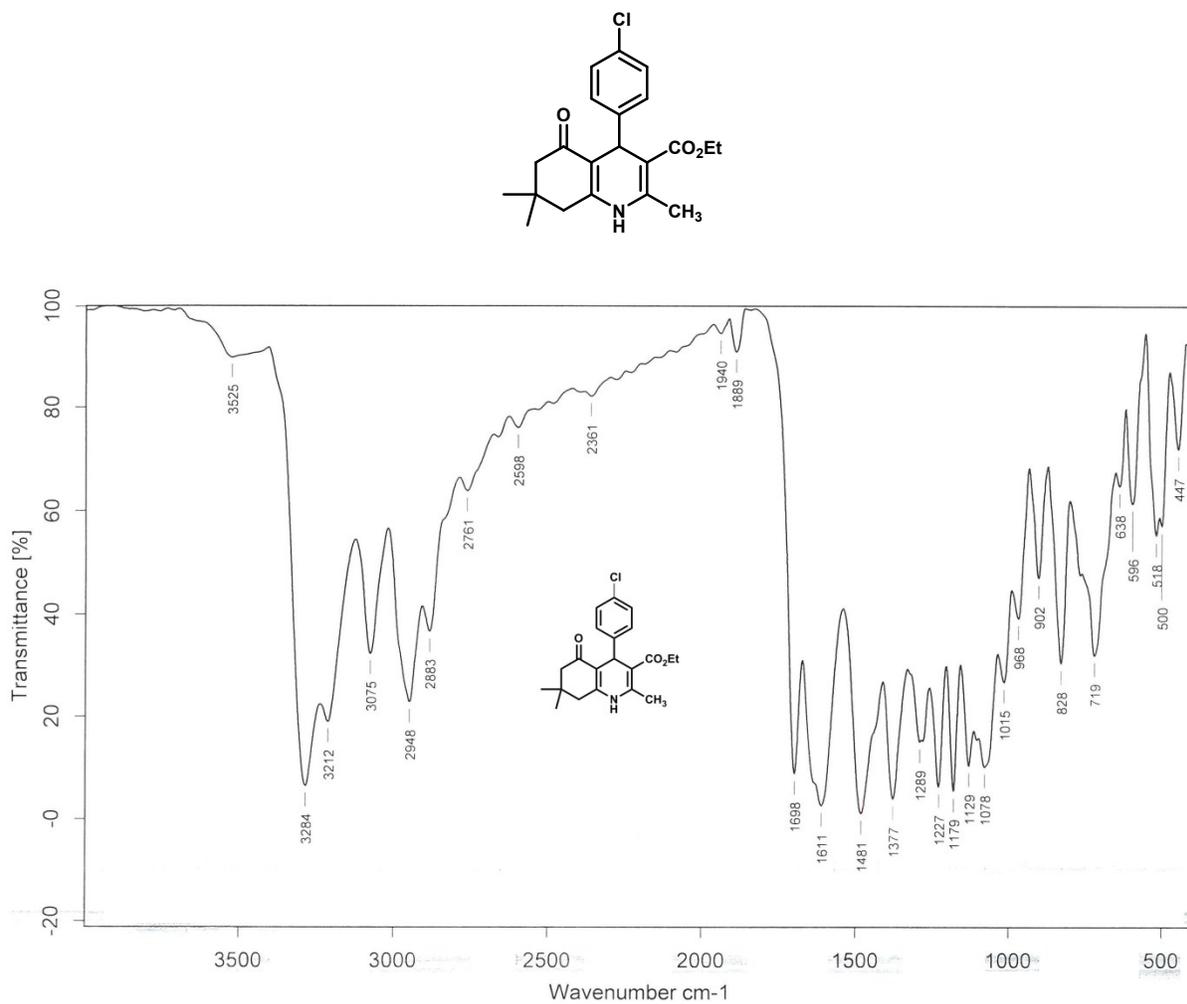
Sample code: 15



**Fig. 23.** <sup>1</sup>H NMR of ethyl-4-(3-nitrophenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (h).

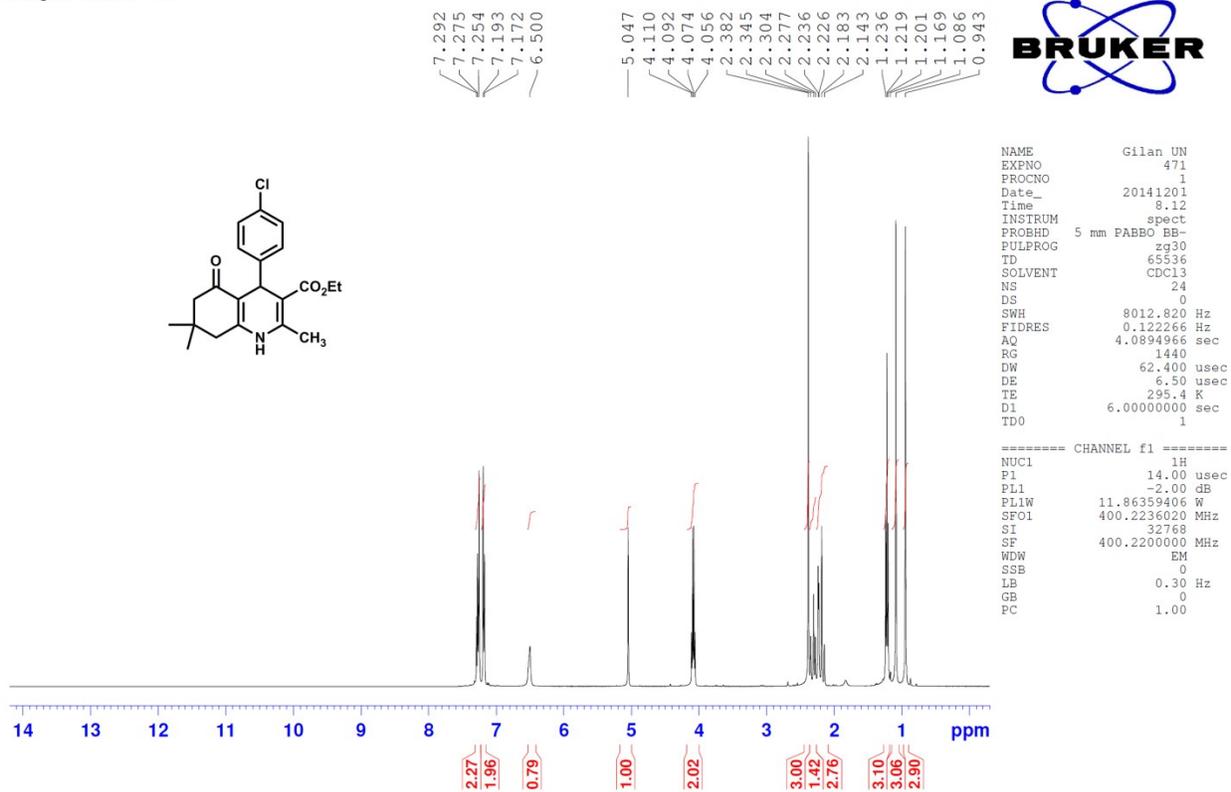
**Ethyl 4-(4-chlorophenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (i) :**

White solid., m.p.= 241-243 °C., FT-IR (KBr, cm<sup>-1</sup>): 3284, 3212, 3075, 2948, 1698, 1611, 1481, 1377, 1227, 1079, 828., <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 0.948 (s, 3H, CH<sub>3</sub>), 1.086 (s, 3H, CH<sub>3</sub>), 1.219 (t, *J*=7.2 Hz, 3H, CH<sub>3</sub>), 2.143-2.345 (m, 4H, CH<sub>2</sub>), 2.386 (s, 3H), 4.08 (q, *J* = 7.2 Hz, 2H), 5.047 (s, 1H, CH), 6.5 (br, 1H, NH), 7.18 (d, *J*= 8.3 Hz, 2H, ArH), 7.28 (d, *J* = 8.2 Hz, 2H, ArH).



**Fig. 24.** FT-IR of ethyl 4-(4-chlorophenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

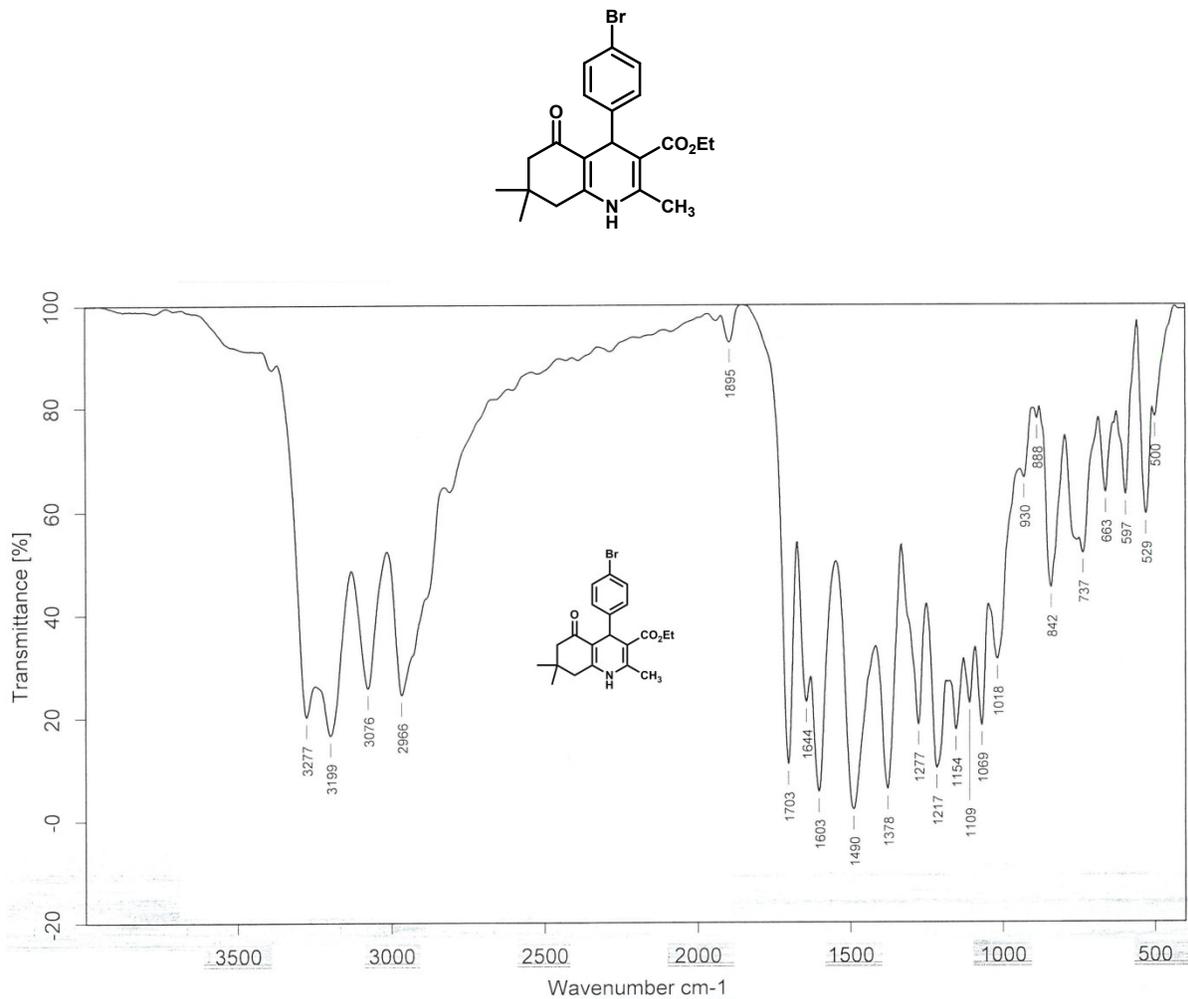
Sample code: 14



**Fig. 25.**  $^1\text{H}$  NMR of ethyl 4-(4-chlorophenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

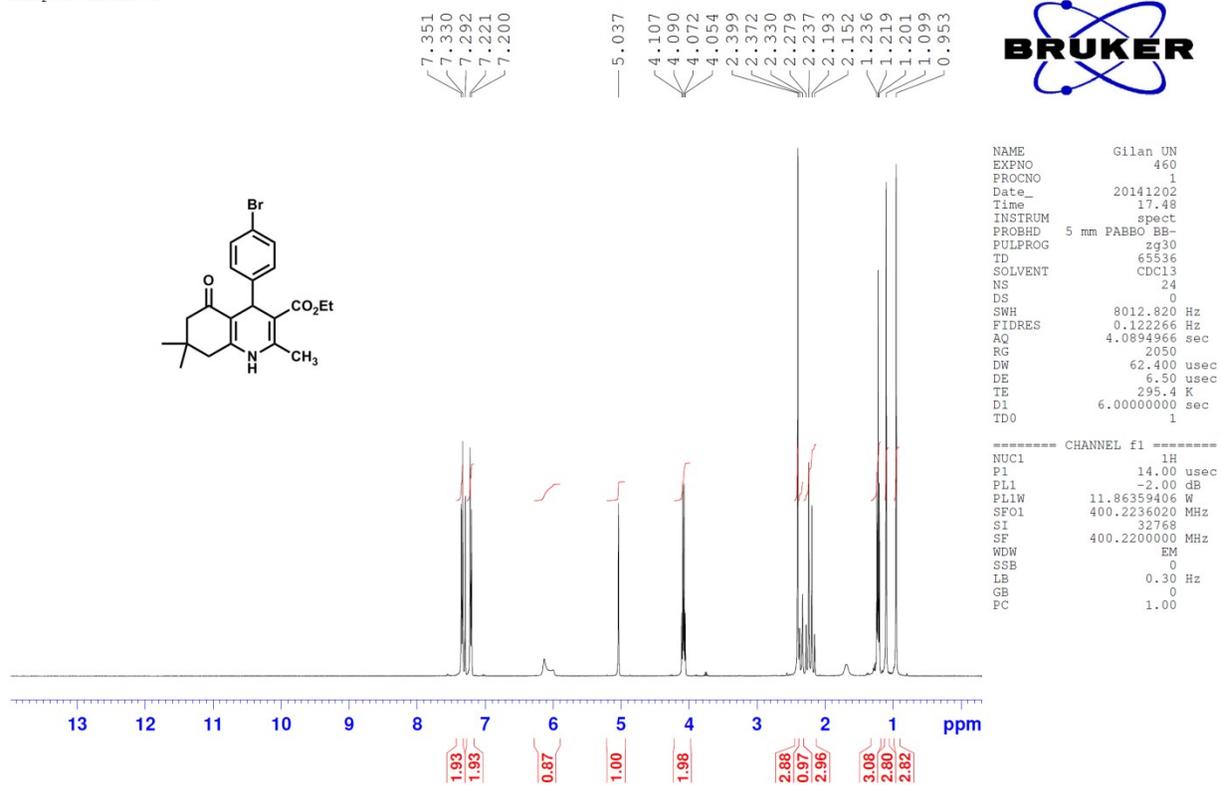
**Ethyl-4-(4-bromophenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (j):**

White solid, m.p.= 248-252 °C., FT-IR (KBr,  $\text{cm}^{-1}$ ): 3277, 3199, 3076, 2966, 1703, 1603, 1490, 1378, 1277, 1217, 1069, 842, 737, 529.,  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  0.953 (s, 3H,  $\text{CH}_3$ ), 1.099 (s, 3H,  $\text{CH}_3$ ), 1.219 (t,  $J$  = 7.2 Hz, 3H,  $\text{CH}_3$ ), 2.152–2.399 (m, 7H), 4.08 (q,  $J$  = 7.2 Hz, 2H,  $\text{OCH}_2$ ), 5.037 (s, 1H, CH), 6.01 (br, 1H, NH), 7.21 (d,  $J$  = 8.3 Hz, 2H, ArH), 7.34 (d,  $J$  = 8.5 Hz, 2H, ArH), 8.23 (s, 1H, NH).



**Fig. 26.** FT-IR of ethyl-4-(4-bromophenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

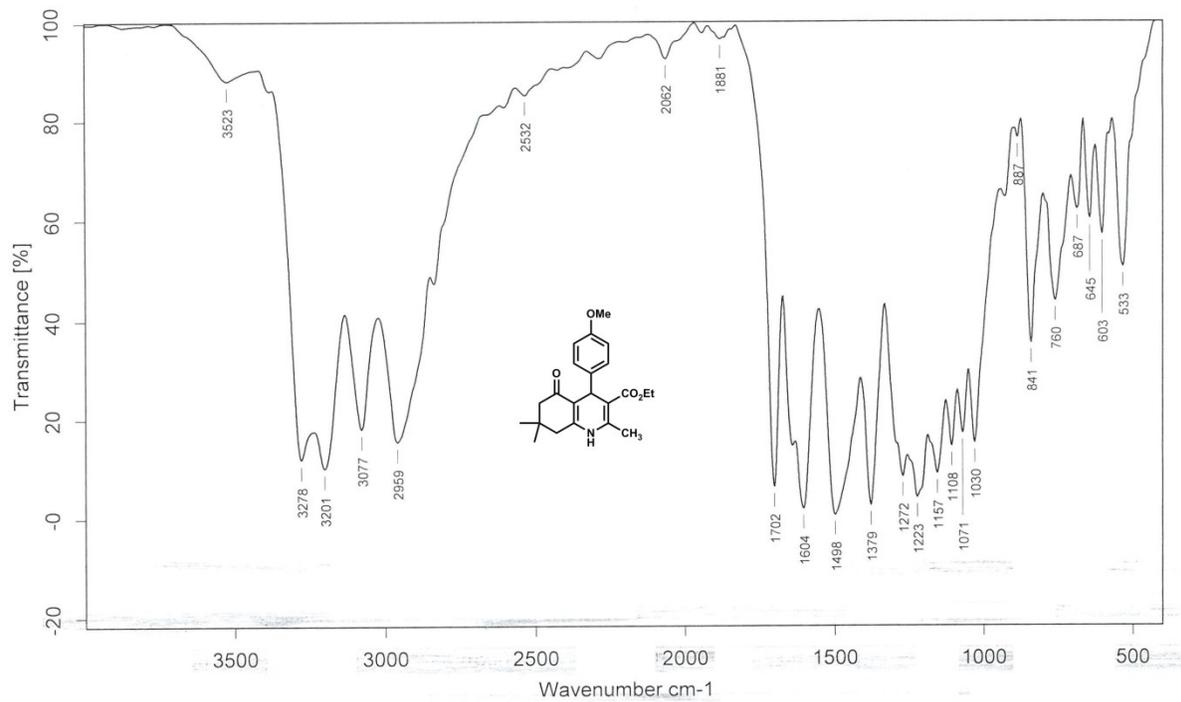
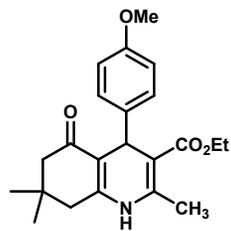
Sample code: 8



**Fig. 27.**  $^1\text{H}$ NMR of ethyl-4-(4-bromophenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

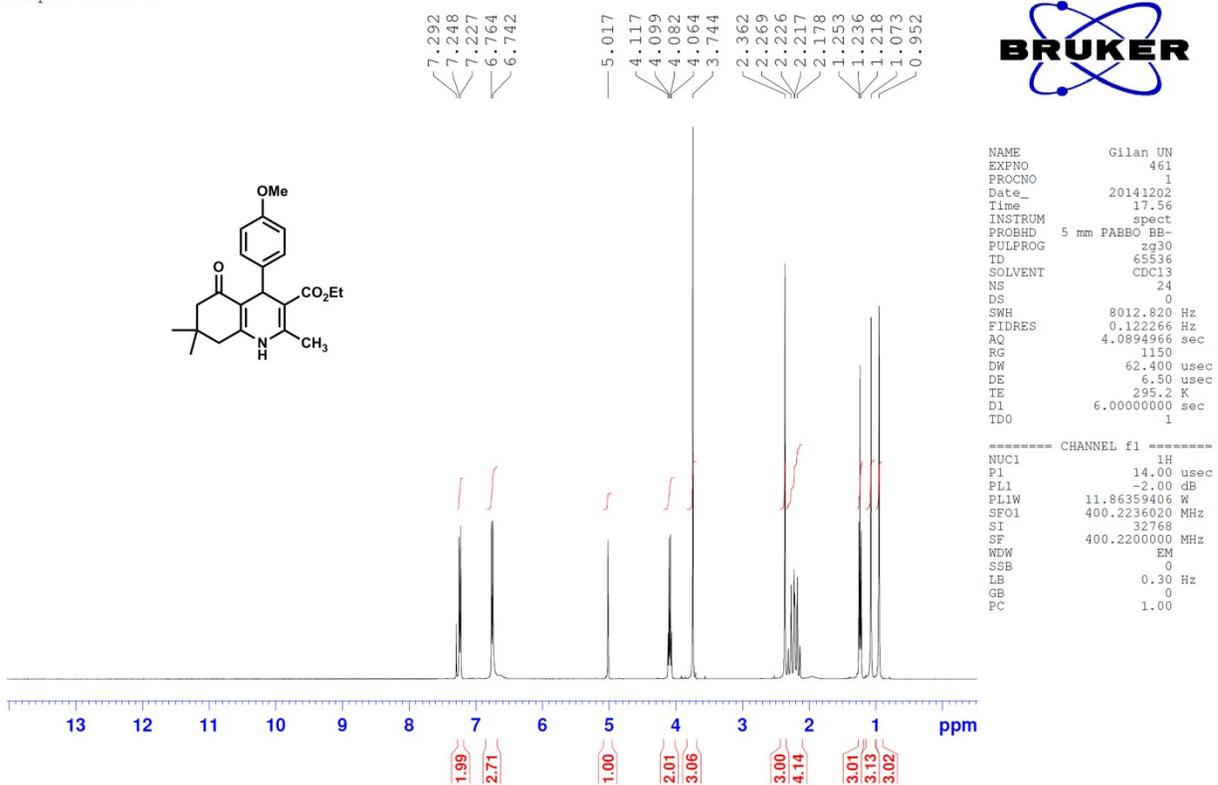
**Ethyl 4-(4-methoxyphenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (k):**

Pale yellow solid, m.p = 254-256 °C., FT-IR (KBr,  $\text{cm}^{-1}$ ) 3278, 3201, 3077, 2959, 1702, 1604, 1498, 1379, 1272, 1223, 1108, 1071, 1030, 841, 760.,  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  0.952 (s, 3H,  $\text{CH}_3$ ), 1.073 (s, 3H,  $\text{CH}_3$ ), 1.236 (t, 3H,  $J=7.4$  Hz,  $\text{CH}_3$ ), 2.178-2.269 (m, 4H), 2.362 (s, 3H), 3.744 (s, 3H), 4.085 (q, 2H,  $J=7.4$  Hz,  $\text{OCH}_2$ ), 5.17 (s, 1H, CH), 6.742-6.764 (d, 2H,  $J=8.8$  Hz, ArH), 7.23 (d, 2H,  $J=8.8$  Hz, ArH), 7.292 (s, 1H, NH).



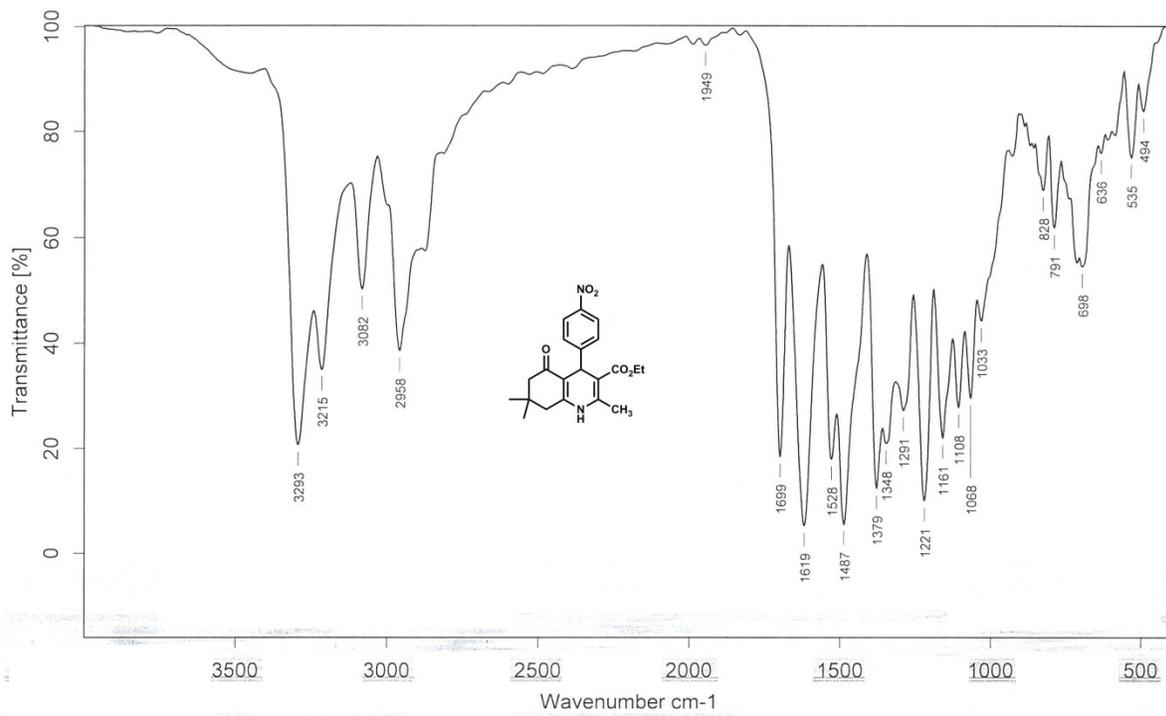
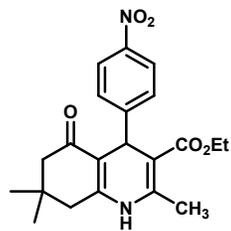
**Fig. 28.** FT-IR of ethyl 4-(4-methoxyphenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

Sample code: 9



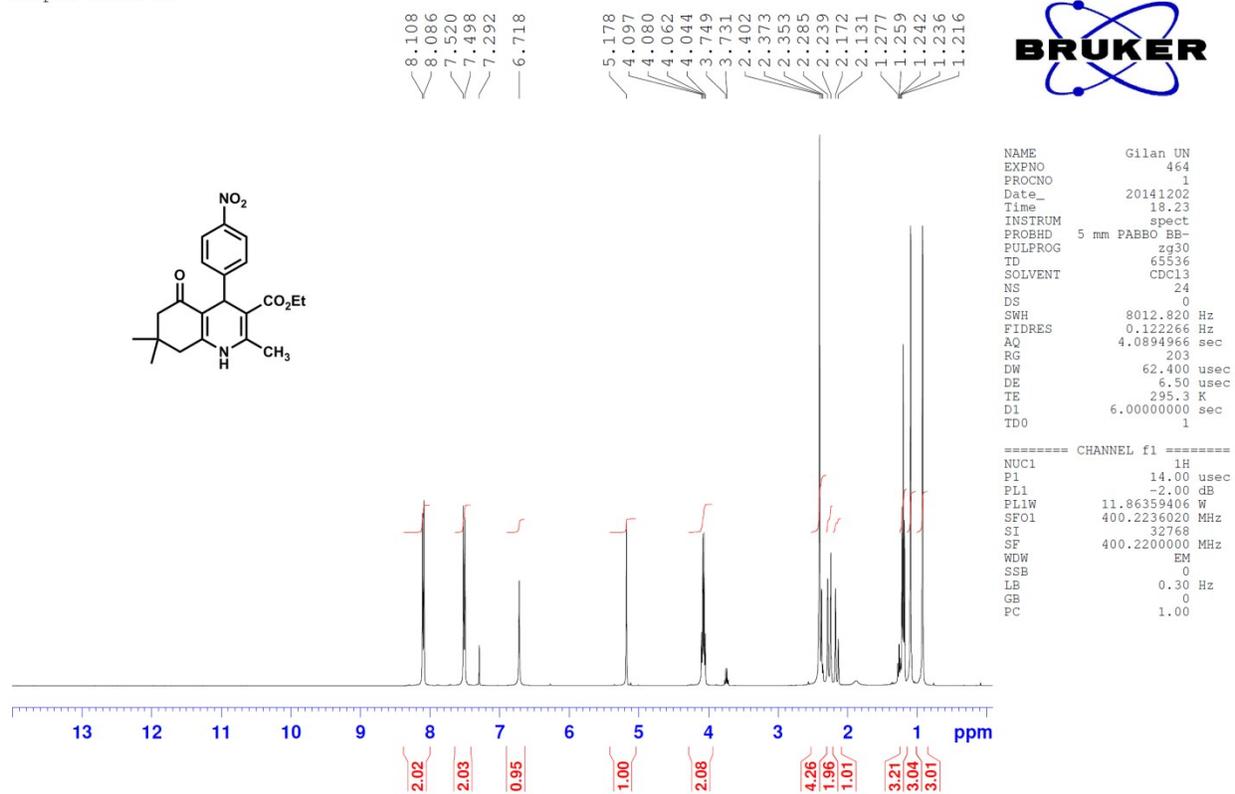
**Fig. 29.**  $^1\text{H}$  NMR of ethyl 4-(4-methoxyphenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

**Ethyl 2,7,7-trimethyl-4-(4-nitrophenyl)-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (I):** Pale yellow solid, m.p = 238-240 °C., FT-IR (KBr,  $\text{cm}^{-1}$ ) 3293, 3215, 3082, 2958, 1699, 1619, 1528, 1487, 1379, 1221, 1068, 698.,  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  0.920 (s, 3H,  $\text{CH}_3$ ), 1.092 (s, 3H,  $\text{CH}_3$ ), 1.199 (t, 3H,  $J=7.3$  Hz,  $\text{CH}_3$ ), 2.15 (dd, 2H,  $\text{CH}_2$ ), 2.26 (dd, 2H,  $\text{CH}_2$ ), 2.402 (s, 6H,  $\text{CH}_3$ ), 4.07 (q, 2H,  $J=7.4$  Hz,  $\text{OCH}_2$ ), 5.178 (s, 1H, CH), 6.718 (s, 1H, NH), 7.51 (d, 2H,  $J=9.4$  Hz, ArH), 8.09 (d, 2H,  $J=9.4$  Hz, ArH).



**Fig. 30.** FT-IR of ethyl 2,7,7-trimethyl-4-(4-nitrophenyl)-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

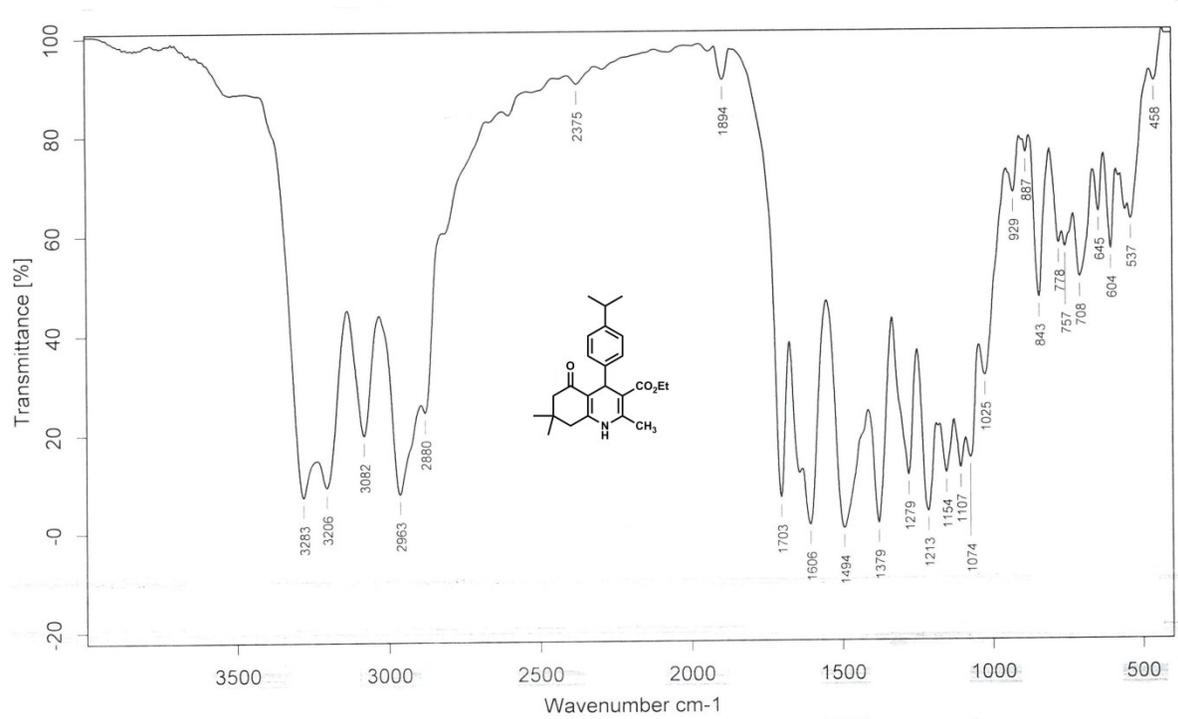
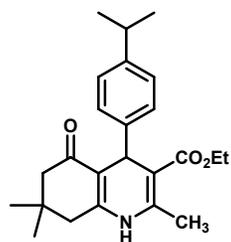
Sample code: 12



**Fig. 31.**  $^1\text{H}$  NMR of ethyl 2,7,7-trimethyl-4-(4-nitrophenyl)-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

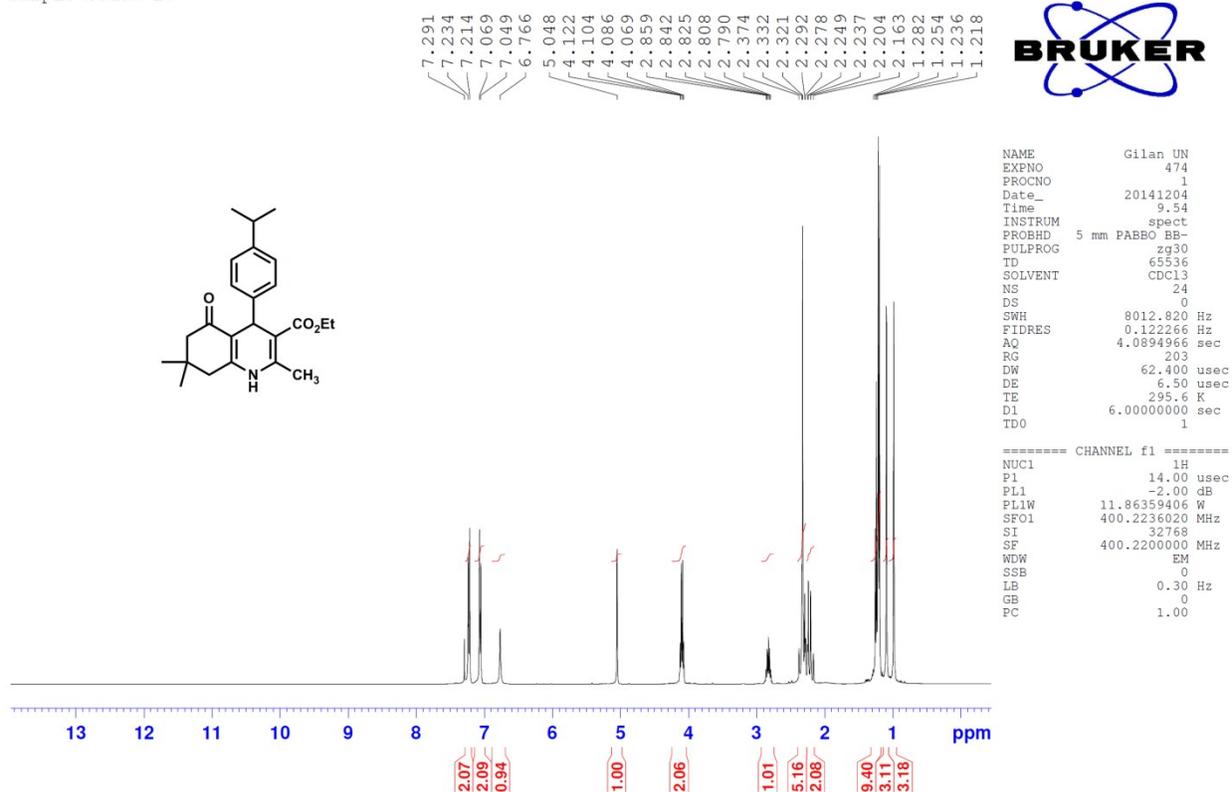
**Ethyl 4-(4-isopropylphenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (r):**

White solid, m.p = 179-181 °C.; FT-IR (KBr,  $\text{cm}^{-1}$ ) 3283, 3206, 3082, 2963, 1703, 1606, 1494, 1379, 1279, 1213, 843.,  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  0.983 (s, 3H,  $\text{CH}_3$ ), 1.091 (s, 3H,  $\text{CH}_3$ ), 1.192-1.282 (m, 9H), 2.163 (m, 6H), 2.79 (m, 1H, CH), 4.08 (q, 2H,  $J=7.4$  Hz,  $\text{OCH}_2$ ), 5.048 (s, 1H, CH), 6.766 (s, 1H, NH), 7.06 (d, 2H,  $J=8$  Hz, ArH), 7.22 (d, 2H,  $J=8$  Hz, ArH).



**Fig. 32.** FT-IR of ethyl 4-(4-isopropylphenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

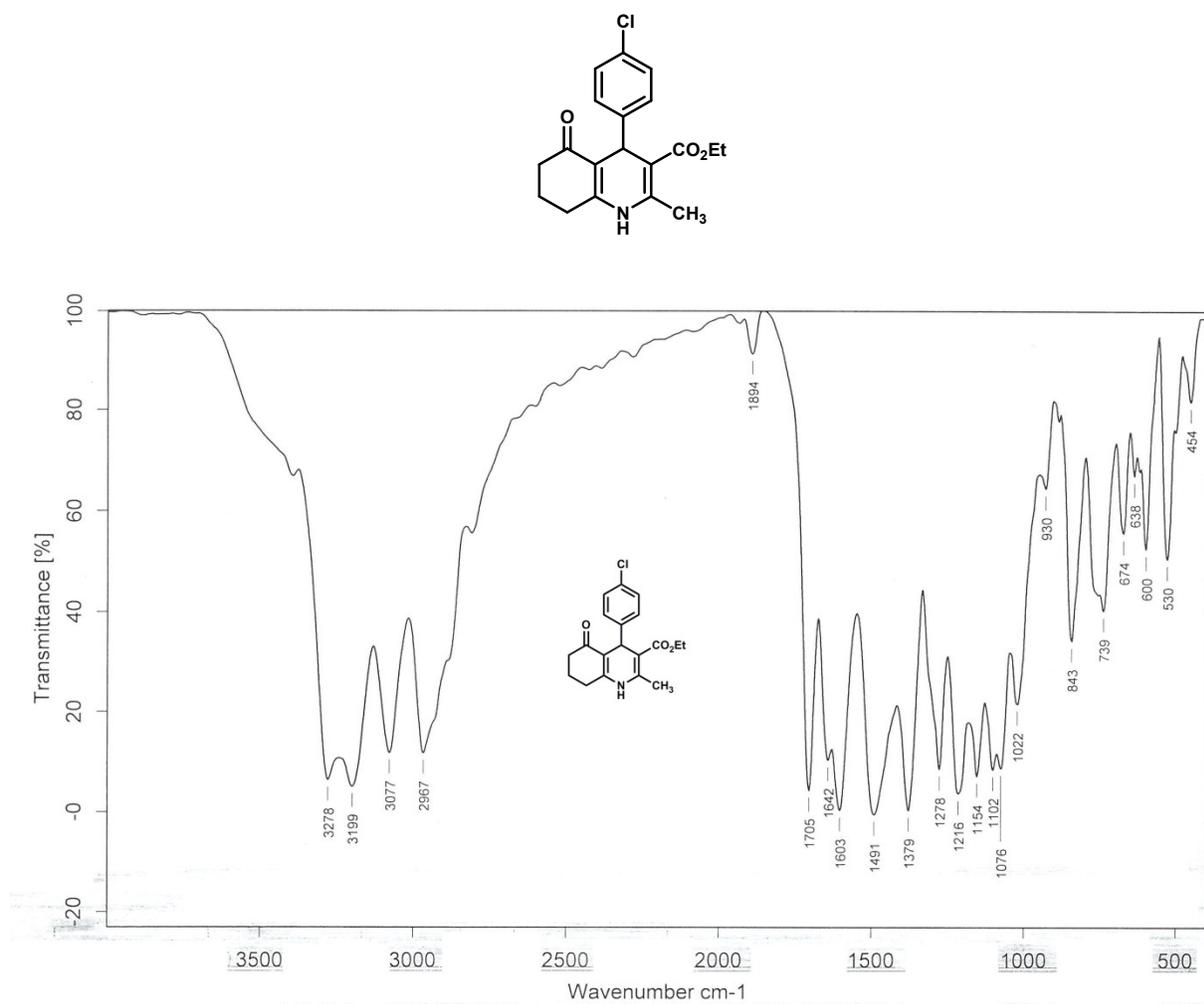
Sample code: 17



**Fig. 33.** <sup>1</sup>H NMR of ethyl 4-(4-isopropylphenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

**Ethyl 4-(4-chlorophenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (t):** White

solid, m.p.= 243-246 °C., FT-IR (KBr, cm<sup>-1</sup>): 3278, 3199, 3077, 2967, 1705, 1603, 1491, 1379, 1278, 1216, 1154, 1076, 843, 739., <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 1.20 (t, *J* = 7.2 Hz, 3H), 1.644 (s, 1H), 1.944-2.053 (m, 2H), 2.330-2.482 (m, 7H), 4.08 (q, *J* = 7.2 Hz, 2H), 5.087 (s, 1H, CH), 6.013 (br, 1H, NH), 7.19 (d, *J* = 8 Hz, 2H, ArH), 7.24 (d, *J* = 8 Hz, 2H, ArH).



**Fig. 34.** FT-IR of ethyl 4-(4-chlorophenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

Sample code:4

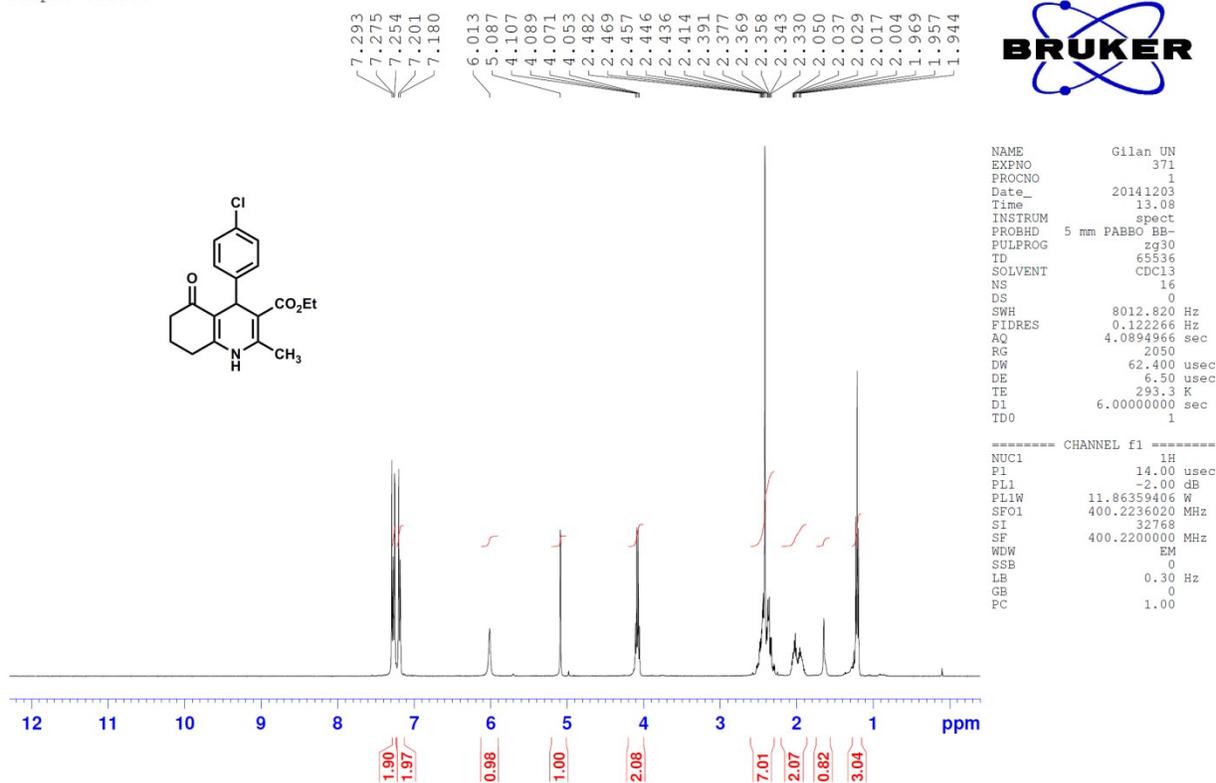
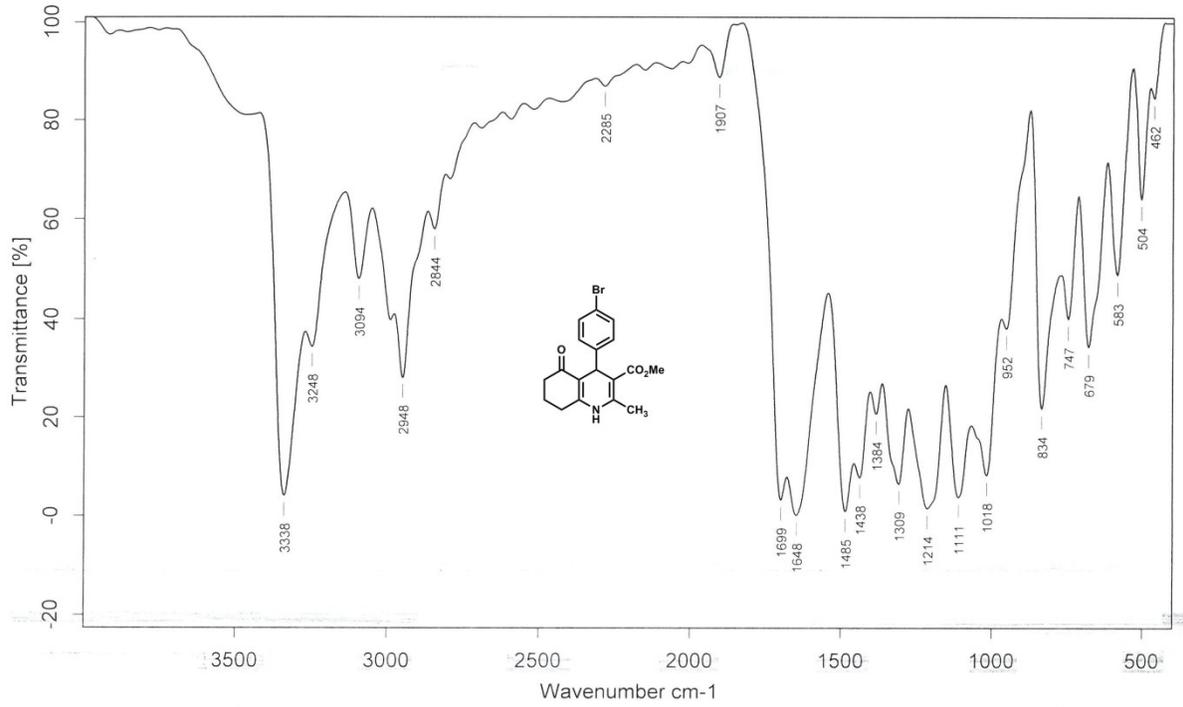
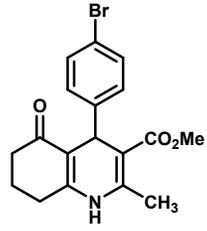


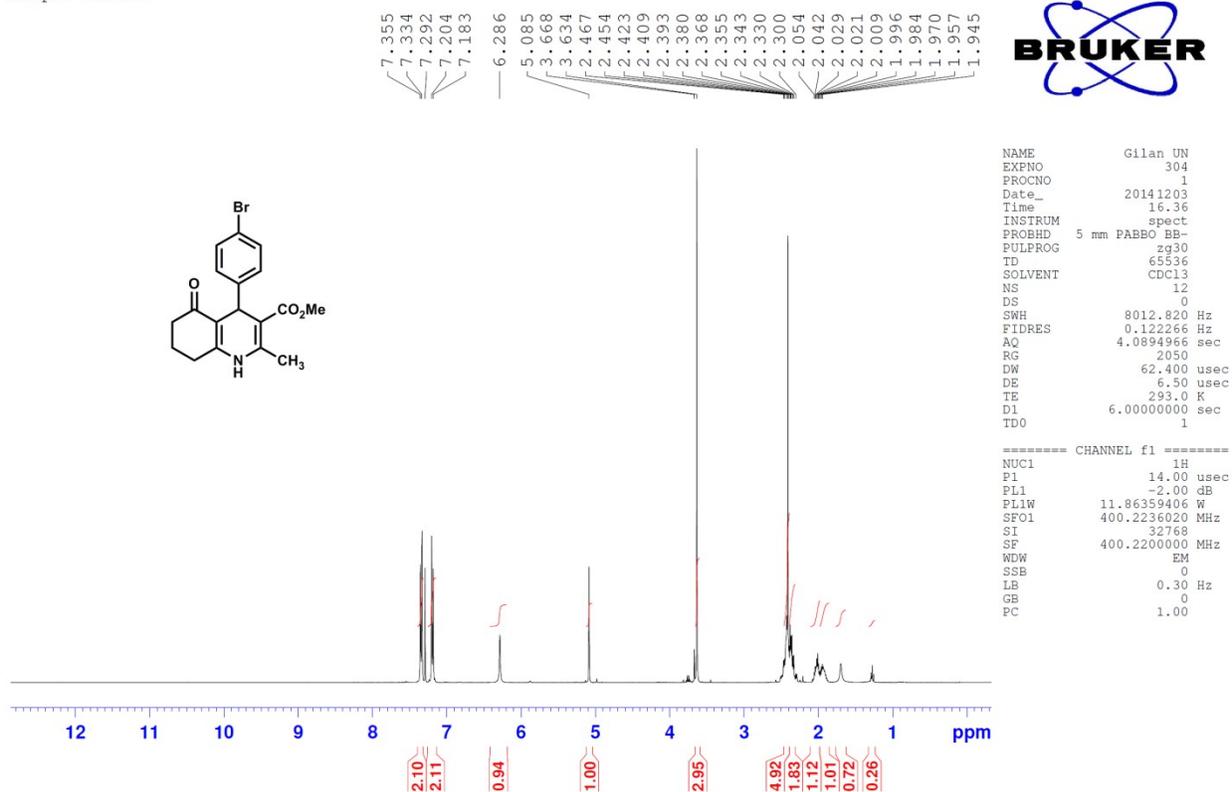
Fig. 35.  $^1\text{H}$  NMR of ethyl 4-(4-chlorophenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

**Methyl 4-(4-bromophenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (u):** White solid, m.p.= 254-255 °C., FT-IR (KBr,  $\text{cm}^{-1}$ ): 3338, 3248, 3094, 2948, 1699, 1648, 1485, 1309, 1214, 834.,  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  1.924-2.042 (m, 2H), 2.3-2.38 (m, 2H), 2.393-2.454 (m, 5H), 3.634 (s, 3H,  $\text{OCH}_3$ ), 5.085 (s, 1H, CH), 6.286 (s, 1H, NH), 7.19 (d, 2H,  $J=9.4$  Hz, ArH), 7.34 (d, 2H,  $J=9.4$  Hz, ArH).,  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  195.86, 167.70, 149.95, 146.03, 144.03, 131.11, 129.69, 119.90, 113.03, 105.23, 51.17, 36.99, 35.91, 27.41, 21.0, 19.46.



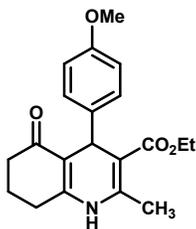
**Fig. 36.** FT-IR of methyl 4-(4-bromophenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

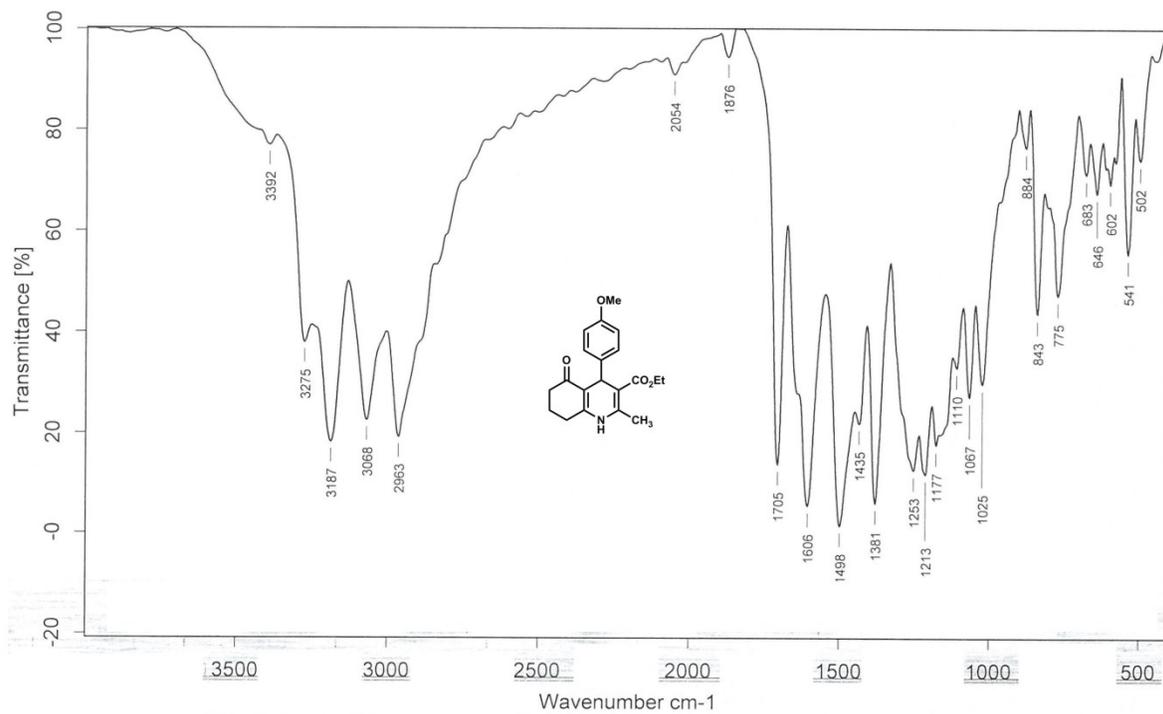
Sample code:1



**Fig. 37.** <sup>1</sup>H NMR of methyl 4-(4-bromophenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

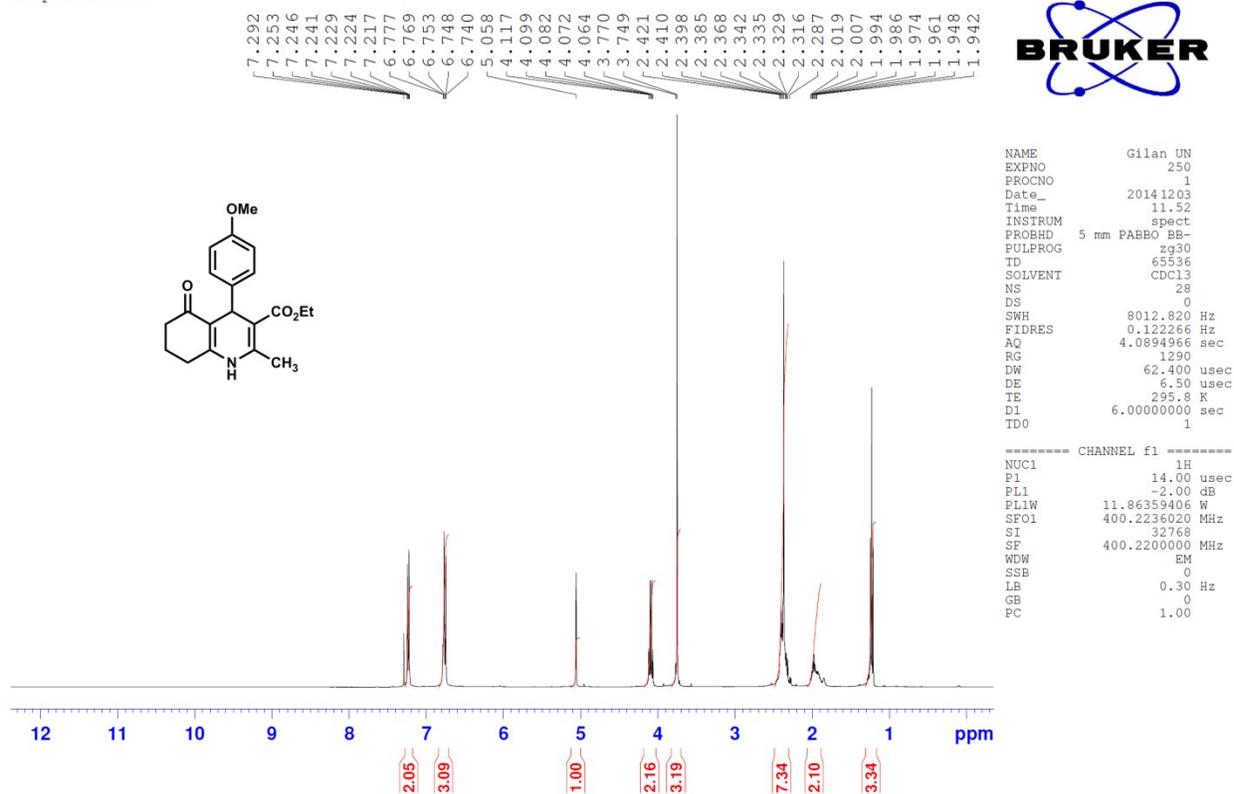
**Ethyl 4-(4-methoxyphenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (w):** Pale yellow solid, m.p.= 194-195 °C., FT-IR (KBr, cm<sup>-1</sup>): 3275, 3187, 3068, 2963, 1705, 1606, 1498, 1381, 1213, 1025, 843., <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 1.24 (t, *J* = 7.2 Hz, 3H), 1.91-2.02 (m, 2H), 2.29-2.42 (m, 7H), 3.75 (s, 3H), 4.08 (q, *J* = 7.2 Hz, 2H), 5.06 (s, 1H), 6.74-6.78 (m, 3H), 7.24 (d, *J* = 8.8 Hz, 2H, Ar-H).





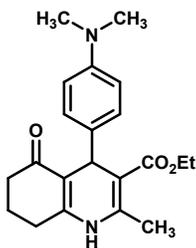
**Fig. 38.** FT-IR of ethyl 4-(4-methoxyphenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

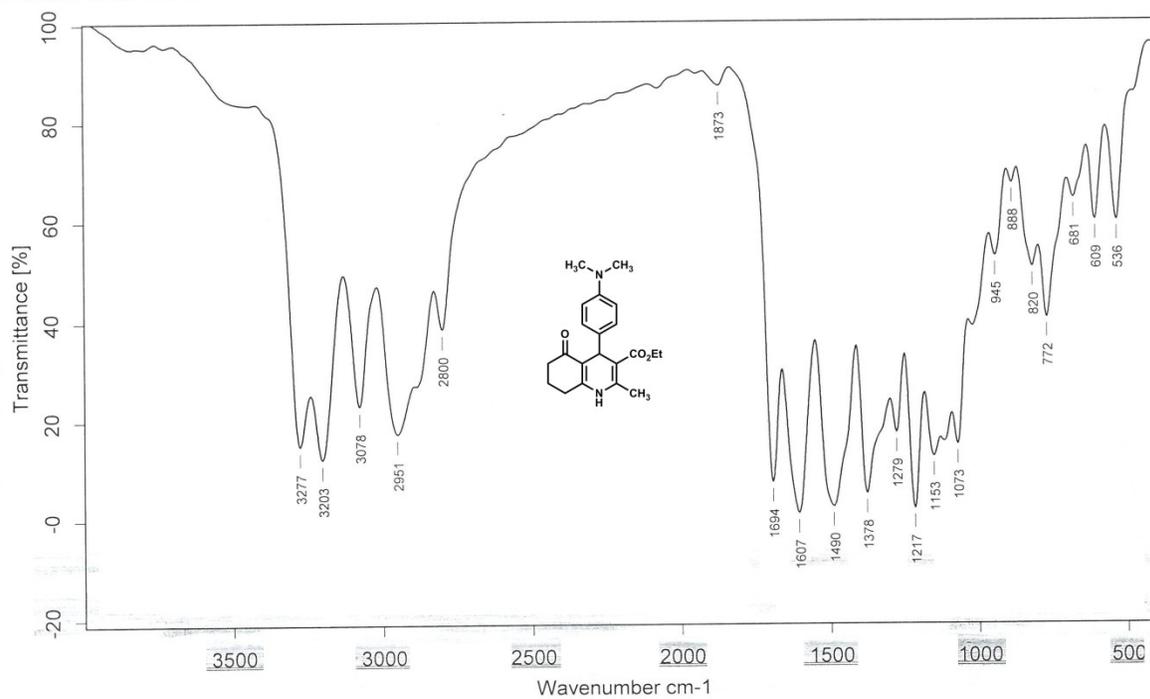
Sample code:1



**Fig. 39.** <sup>1</sup>H NMR of ethyl 4-(4-methoxyphenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

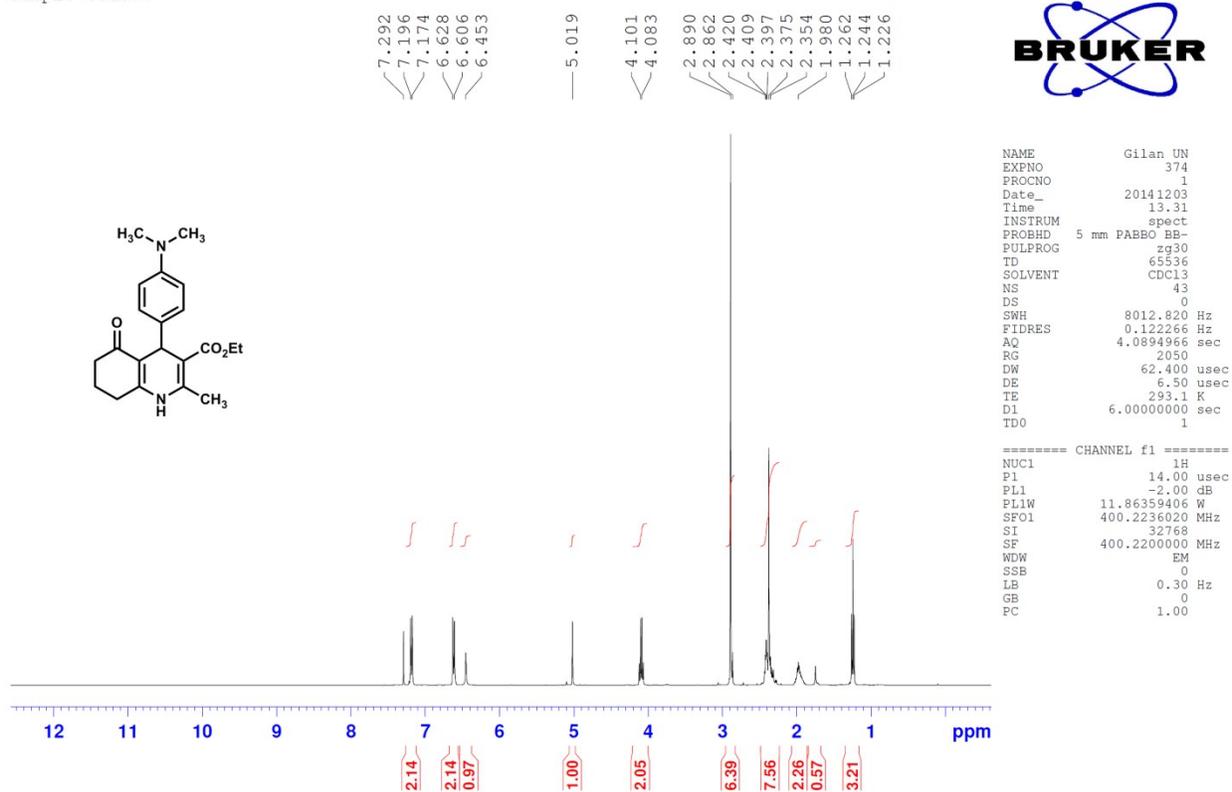
**Ethyl 4-(4-(dimethylamino)phenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (y):** Brown solid., m.p. = 204-206 °C., FT-IR (KBr, cm<sup>-1</sup>): 3277, 3203, 3078, 2951, 1694, 1607, 1490, 1378, 1217, 772., <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 0.978 (s, 3H), 1.091 (s, 3H), 2.167-2.344 (m, 4H), 2.393 (s, 3H, OCH<sub>3</sub>), 2.9 (s, 6H, NCH<sub>3</sub>), 4.966 (s, 1H, CH), 6.093 (br, 1H, NH), 6.64 (d, *J* = 8 Hz, 2H, ArH), 7.18 (d, *J* = 8 Hz, 2H, ArH).





**Fig. 40.** FT-IR of ethyl 4-(4-(dimethylamino)phenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

Sample code:7



**Fig. 41.** <sup>1</sup>H NMR of ethyl 4-(4-(dimethylamino)phenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

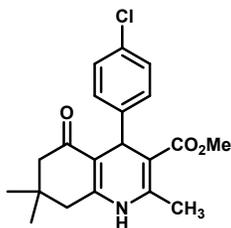
**Methyl-4-(4-chlorophenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (z):**

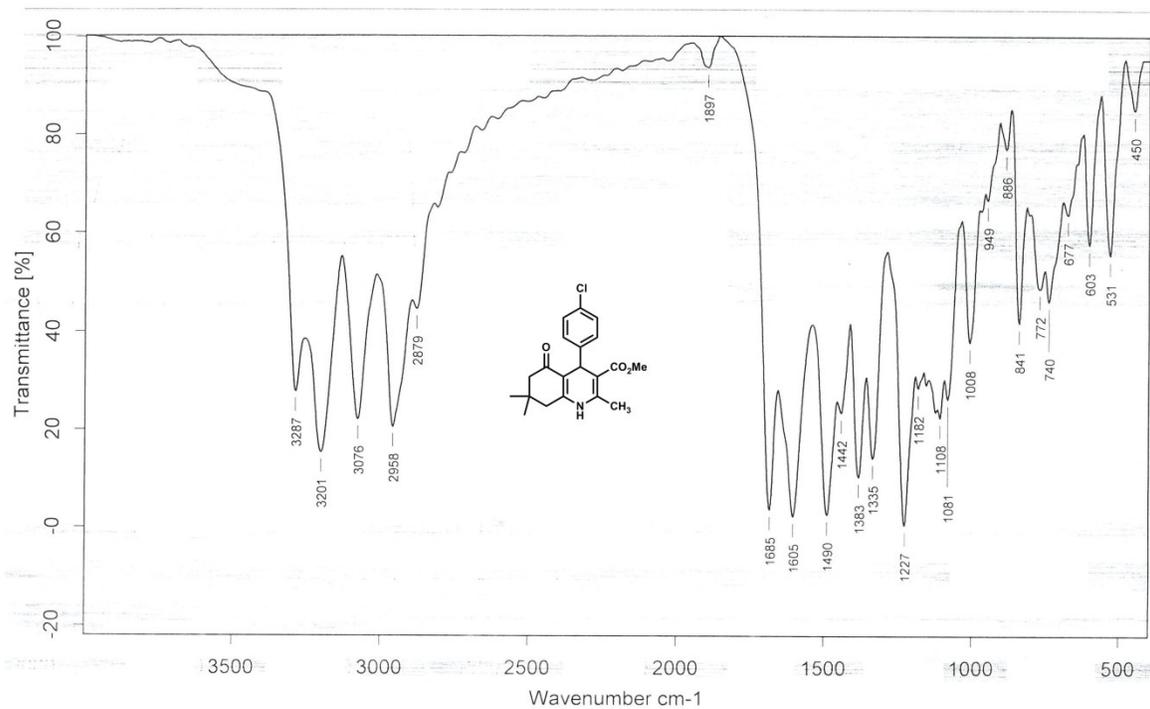
Yellow solid., m.p. = 220-221 °C., FT-IR (KBr, cm<sup>-1</sup>): 3287, 3201, 3076, 2958, 1685, 1605, 1409, 1383,

1335, 1227, 1008, 841., <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>) δ 0.948 (s, 3H, CH<sub>3</sub>), 1.102 (s, 3H, CH<sub>3</sub>), 2.243-

2.412 (m, 3H, CH<sub>3</sub>), 3.636 (s, 3H, OCH<sub>3</sub>), 5.062 (s, 1H, CH), 6.06 (br, 1H, NH), 7.18 (d, *J* = 8.6 Hz, 2H, ArH),

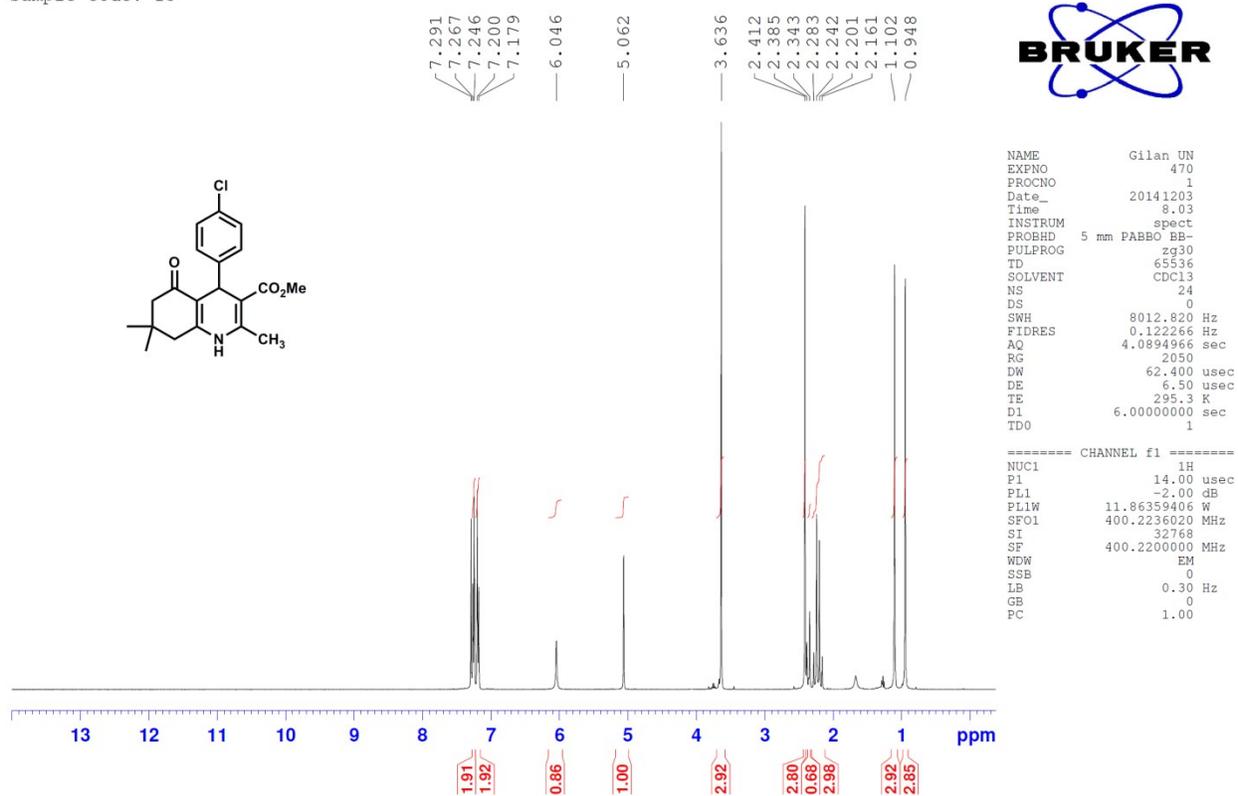
7.28 (d, *J* = 8.4 Hz, 2H, ArH).





**Fig. 42.** FT-IR of methyl-4-(4-chlorophenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

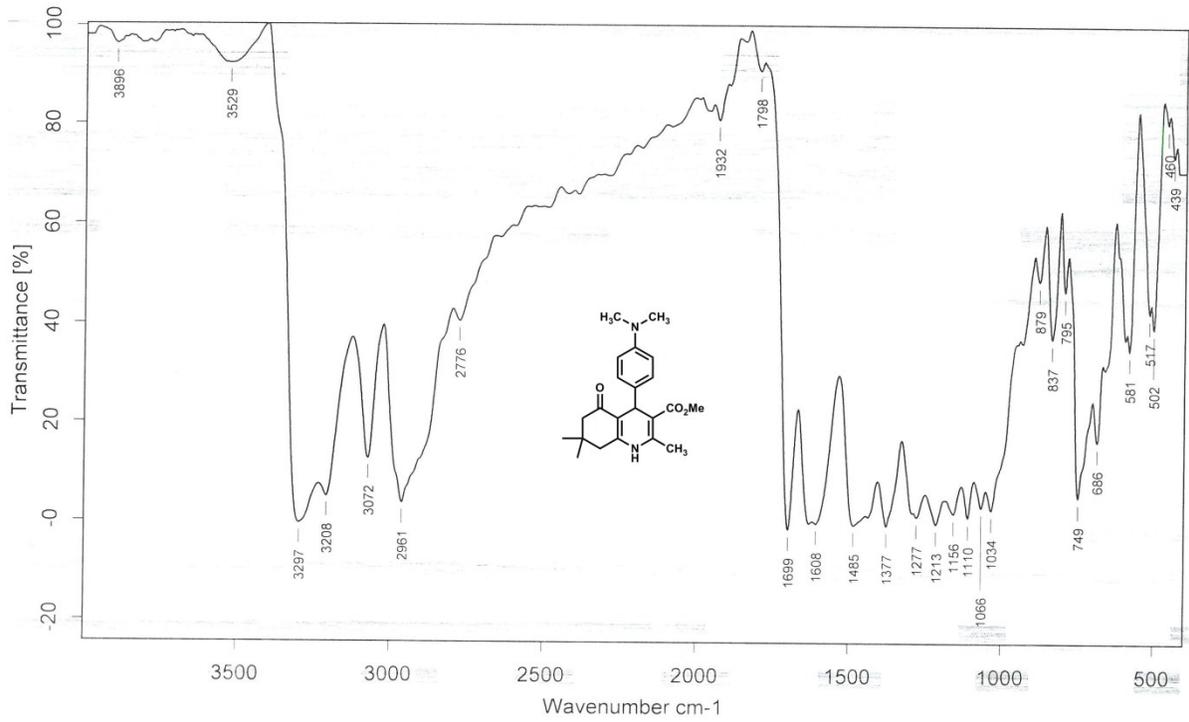
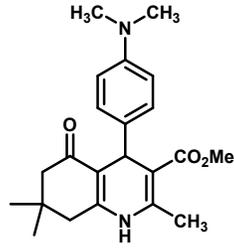
Sample code: 13



**Fig. 43.**  $^1\text{H}$  NMR of methyl-4-(4-chlorophenyl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

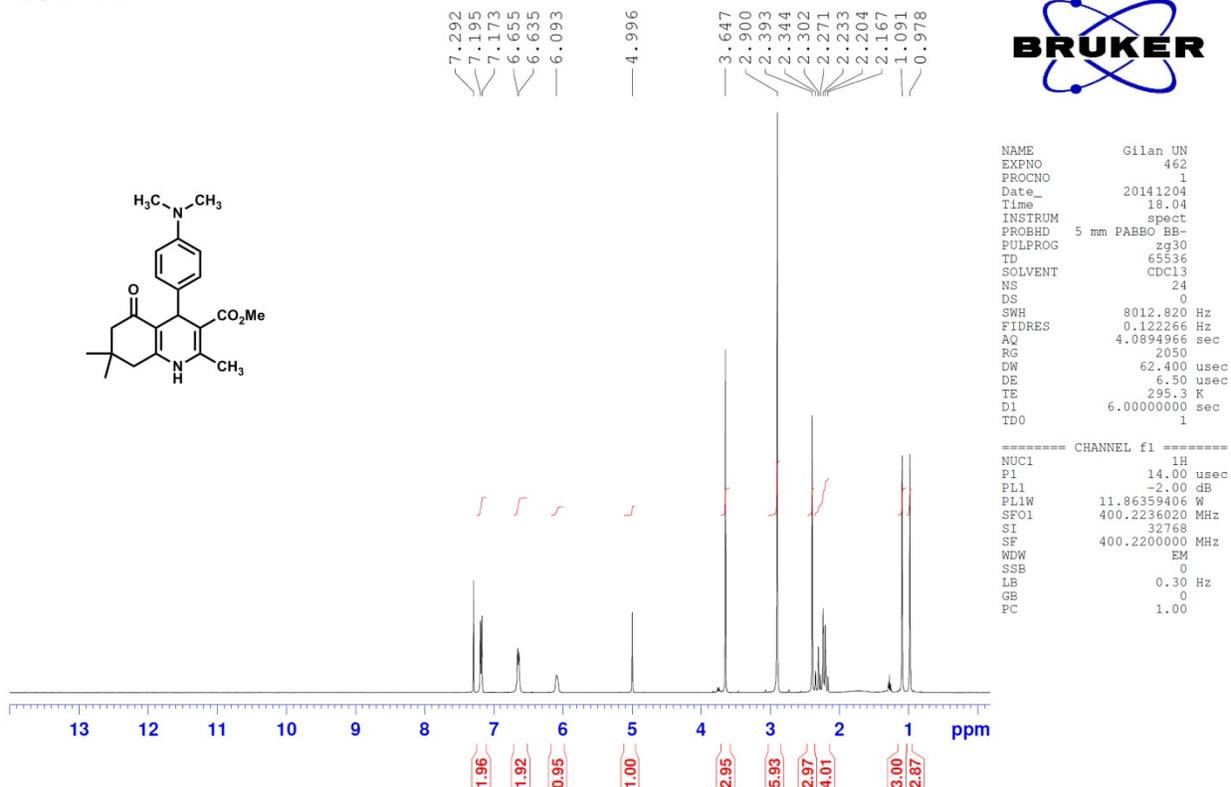
**Ethyl 4-(4-(dimethylamino)phenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (dd):**

Yellow solid, m.p. = 258-259 °C., FT-IR (KBr,  $\text{cm}^{-1}$ ): 3297, 3208, 3072, 2961, 2776, 1699, 1608, 1485, 1377, 1213, 749.,  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  1.24 (t,  $J = 7.2\text{Hz}$ , 3H), 1.98 (m, 2H), 2.35-2.42 (m, 7H), 2.89 (s, 6H), 4.09 (q,  $J = 7.2\text{ Hz}$ , 2H), 5.02 (s, 1H), 6.45 (s, 1H, NH), 6.62 (d,  $J = 8.8\text{ Hz}$ , 2H, Ar-H), 7.18 (d,  $J = 8.8\text{ Hz}$ , 2H, Ar-H),  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  196.08, 167.77, 149.83, 149.02, 143.05, 135.95, 128.6, 113.63, 112.36, 106.38, 59.77, 40.78, 37.14, 35.17, 27.37, 21.11, 19.34, 14.3.



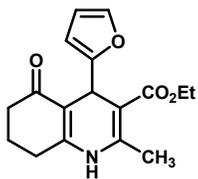
**Fig. 44.** FT-IR of ethyl 4-(4-(dimethylamino)phenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

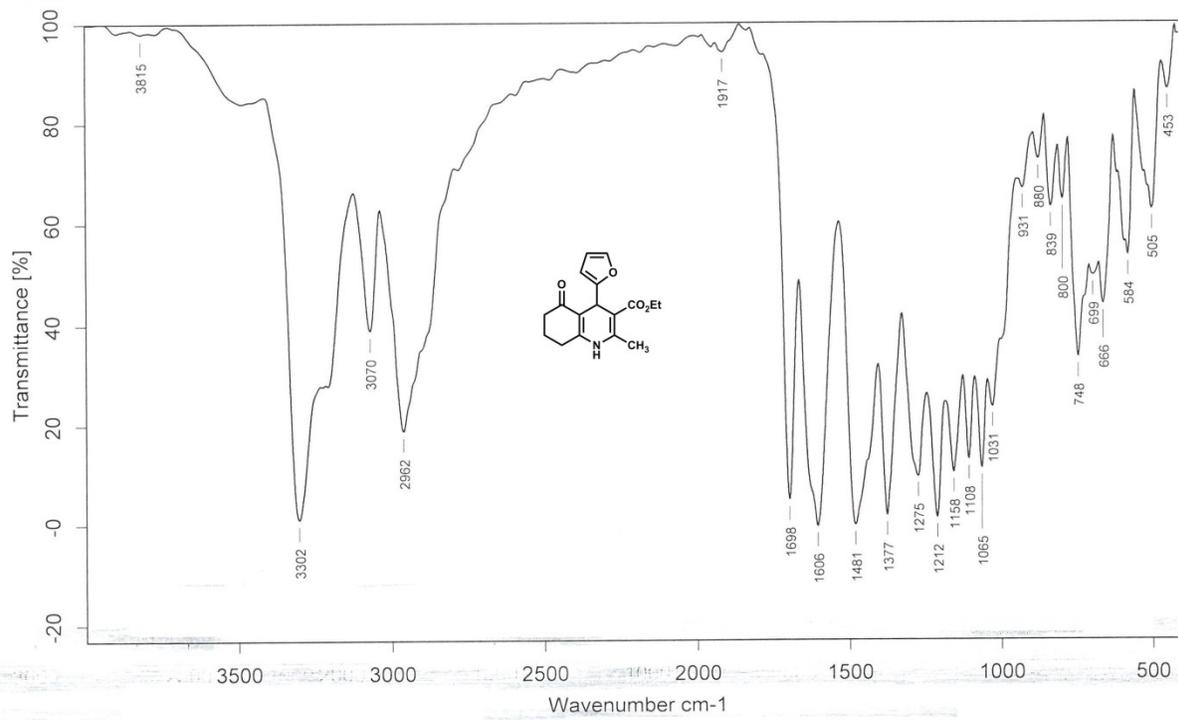
Sample code: 10



**Fig. 45.**  $^1\text{H}$  NMR of ethyl 4-(4-(dimethylamino)phenyl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

**Ethyl 4-(furan-2-yl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (hh):** brown solid, m.p = 207-209., FT-IR (KBr,  $\text{cm}^{-1}$ ): 3302, 3070, 2962, 1696, 1606, 1481, 1377, 1275, 1212, 1158, 1108, 1065, 748.,  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 400 MHz)  $\delta$  1.26 (t, 3H,  $J=7.2$  Hz), 2.01-2.059 (m, 2H), 2.363-2.397 (m, 6H), 2.465-2.487 (m, 3H), 4.11-4.21 (m, 2H,  $\text{OCH}_2$ ), 5.3 (s, 1H, CH), 6.01-6.018 (m, 1H, ArH), 6.14-6.23 (m, 2H, ArH), 7.22 (s, 1H, NH).





**Fig. 46.** FT-IR of ethyl 4-(furan-2-yl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

Sample code:6

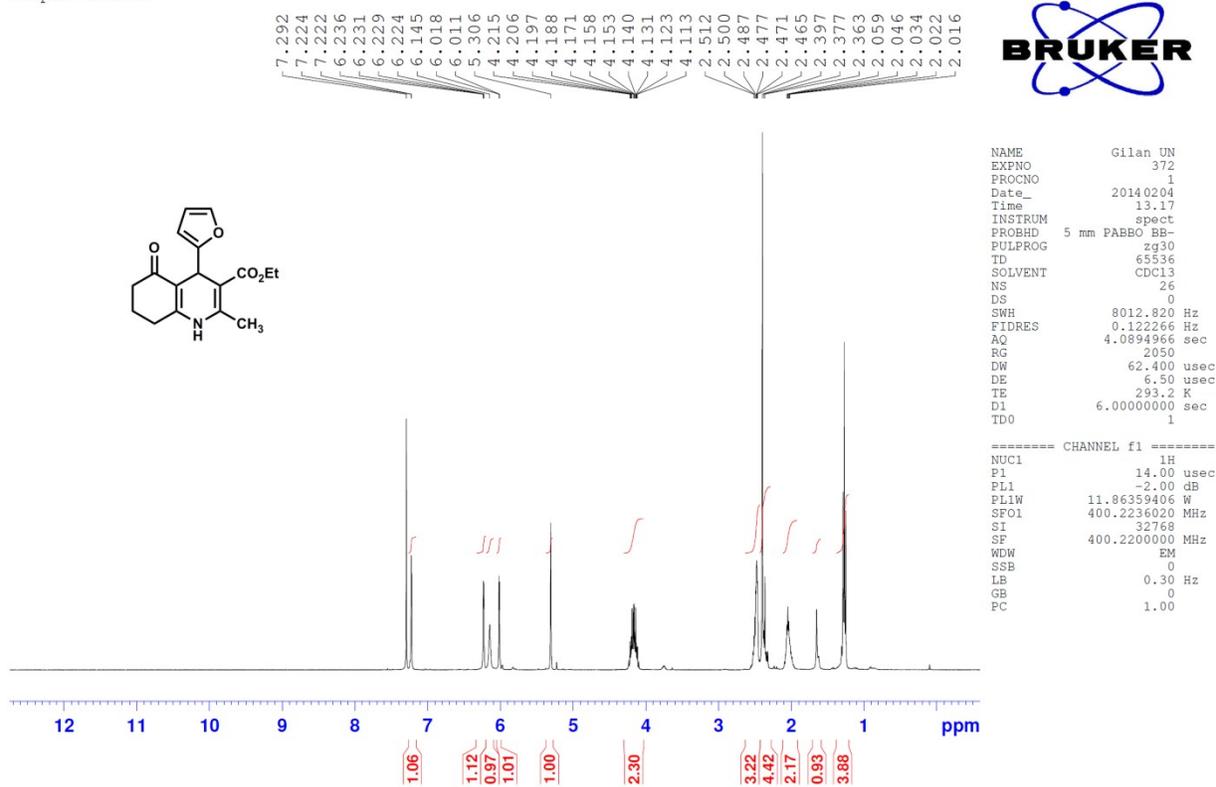
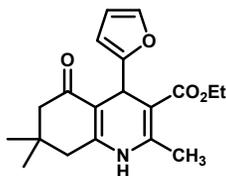
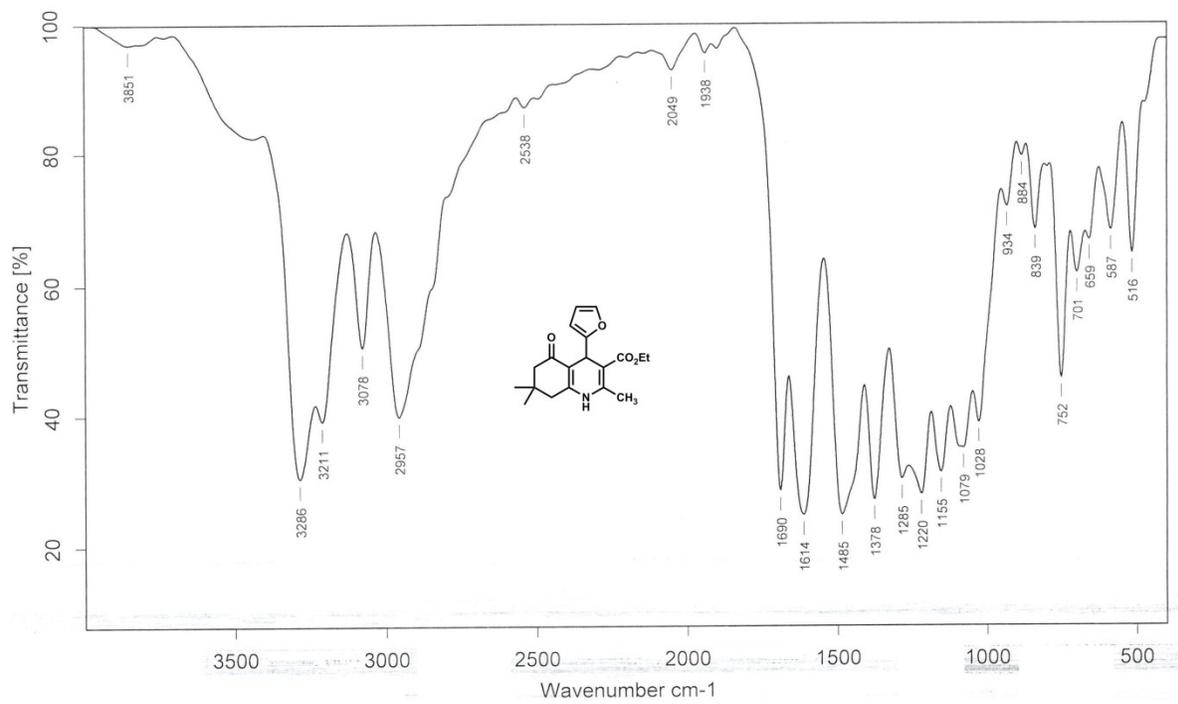


Fig. 47.  $^1\text{H}$  NMR of ethyl 4-(furan-2-yl)-2-methyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

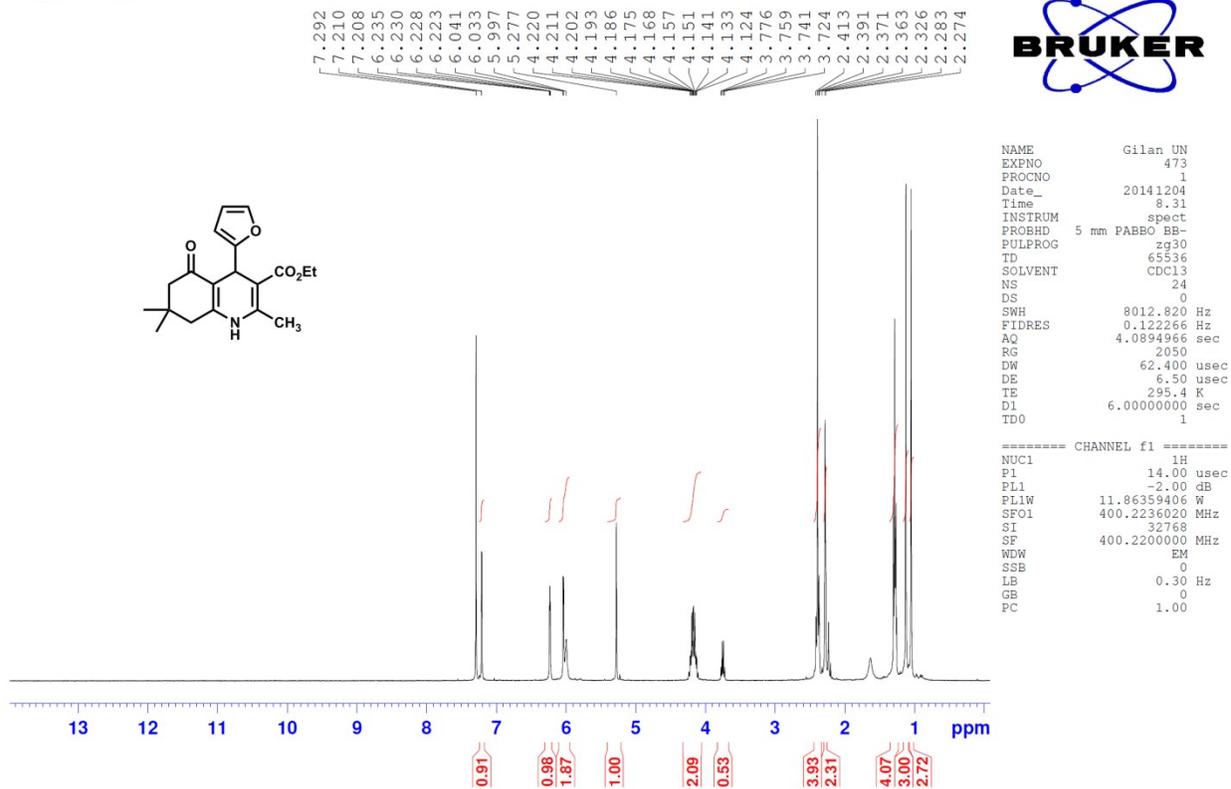
**Ethyl 4-(furan-2-yl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate (ii):** brown solid, m.p = 246-248 °C., FT-IR (KBr,  $\text{cm}^{-1}$ ): 3286, 3211, 3078, 1690, 1614, 1485, 1378, 1285, 1220, 1155, 1079, 752.,  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  1.047 (s, 3H,  $\text{CH}_3$ ), 1.125 (s, 3H,  $\text{CH}_3$ ), 1.28 (t, 3H,  $J=7.4$  Hz,  $\text{CH}_3$ ), 2.28 (m, 2H,  $\text{CH}_2$ ), 2.32-2.41 (m, 5H), 4.17 (q, 2H,  $J=7.3$  Hz,  $\text{OCH}_2$ ), 5.27 (s, 1H, CH), 5.99-6.04 (m, 2H, ArH), 6.22 (m, 1H, ArH), 7.21 (s, 1H, NH).





**Fig. 48.** FT-IR of ethyl 4-(furan-2-yl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

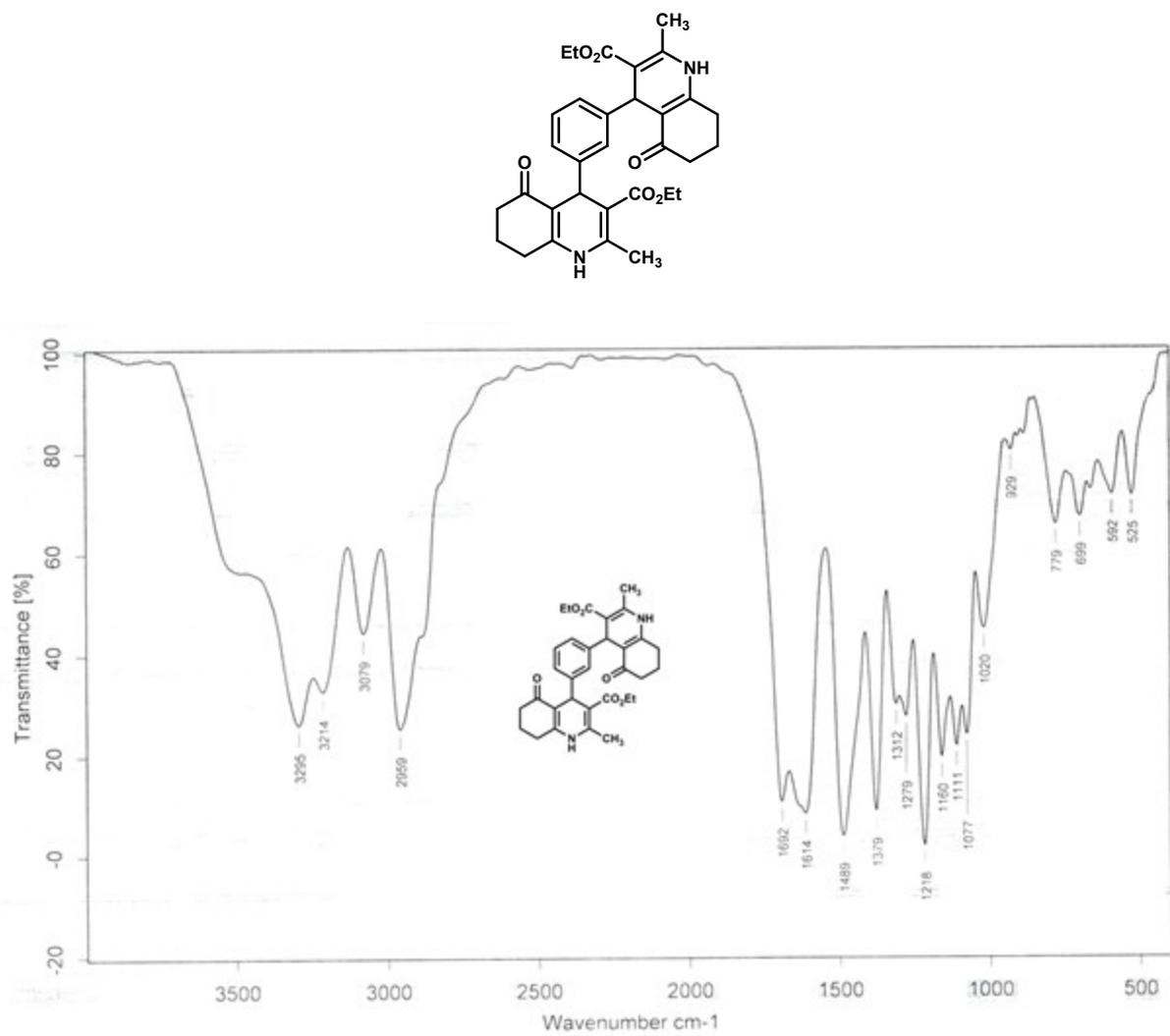
Sample code: 16



**Fig. 49.**  $^1\text{H}$  NMR of ethyl 4-(furan-2-yl)-2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate.

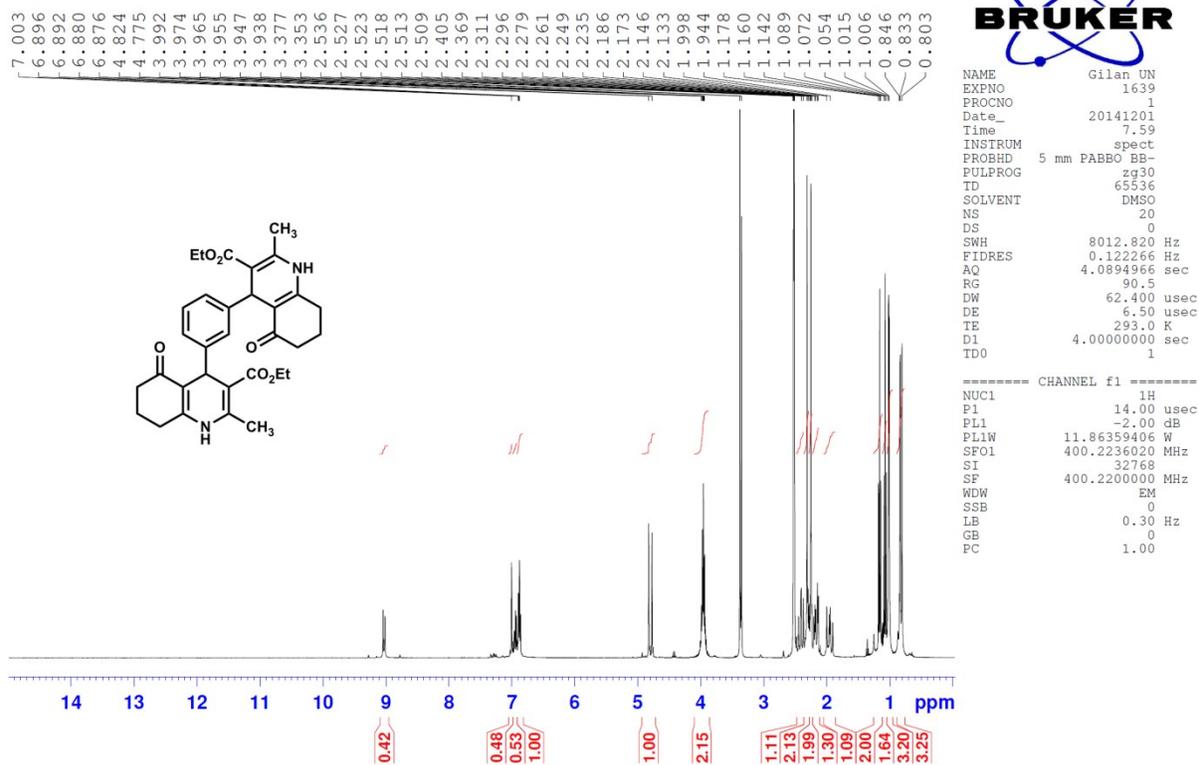
**Diethyl 4,4'-(1,3-phenylene)bis(2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate)**

**(II):** White solid, m.p. = 281-282 °C., FT-IR (KBr,  $\text{cm}^{-1}$ ): 3295, 3214, 3079, 2959, 1692, 1614, 1489, 1379, 1218  $\text{cm}^{-1}$ ,  $^1\text{H}$  NMR (DMSO, 400 MHz)  $\delta$  0.8 (s, 3H,  $\text{CH}_3$ ), 0.84 (s, 3H,  $\text{CH}_3$ ), 1.006 (s, 3H,  $\text{CH}_3$ ), 1.015 (s, 3H,  $\text{CH}_3$ ), 1.07 (t,  $J=7.2$  Hz, 3H,  $\text{CH}_3$ ), 1.17 (t,  $J=7.2$  Hz, 3H,  $\text{CH}_3$ ), 1.90-1.99 (m, 2H), 2.13-2.20 (m, 2H), 2.26-2.31 (8H, 2 $\text{CH}_3$  and  $\text{CH}_2$ ), 2.36-2.44 (m, 2H), 3.9-4.1 (m, 4H, 2 $\text{CH}_2$ ), 4.77 (s, 1H, CH), 4.83(s, 1H, CH), 6.81-7 (m, 3H, ArH), 7 (s, 1H, ArH), 9.00 (s, 1H, NH), 9.03(s, 1H, NH).



**Fig. 50.** FT-IR of diethyl 4,4'-(1,3-phenylene)bis(2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate).

Sample code: 2



**Fig. 51.**  $^1\text{H}$  NMR of diethyl 4,4'-(1,3-phenylene)bis(2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate).

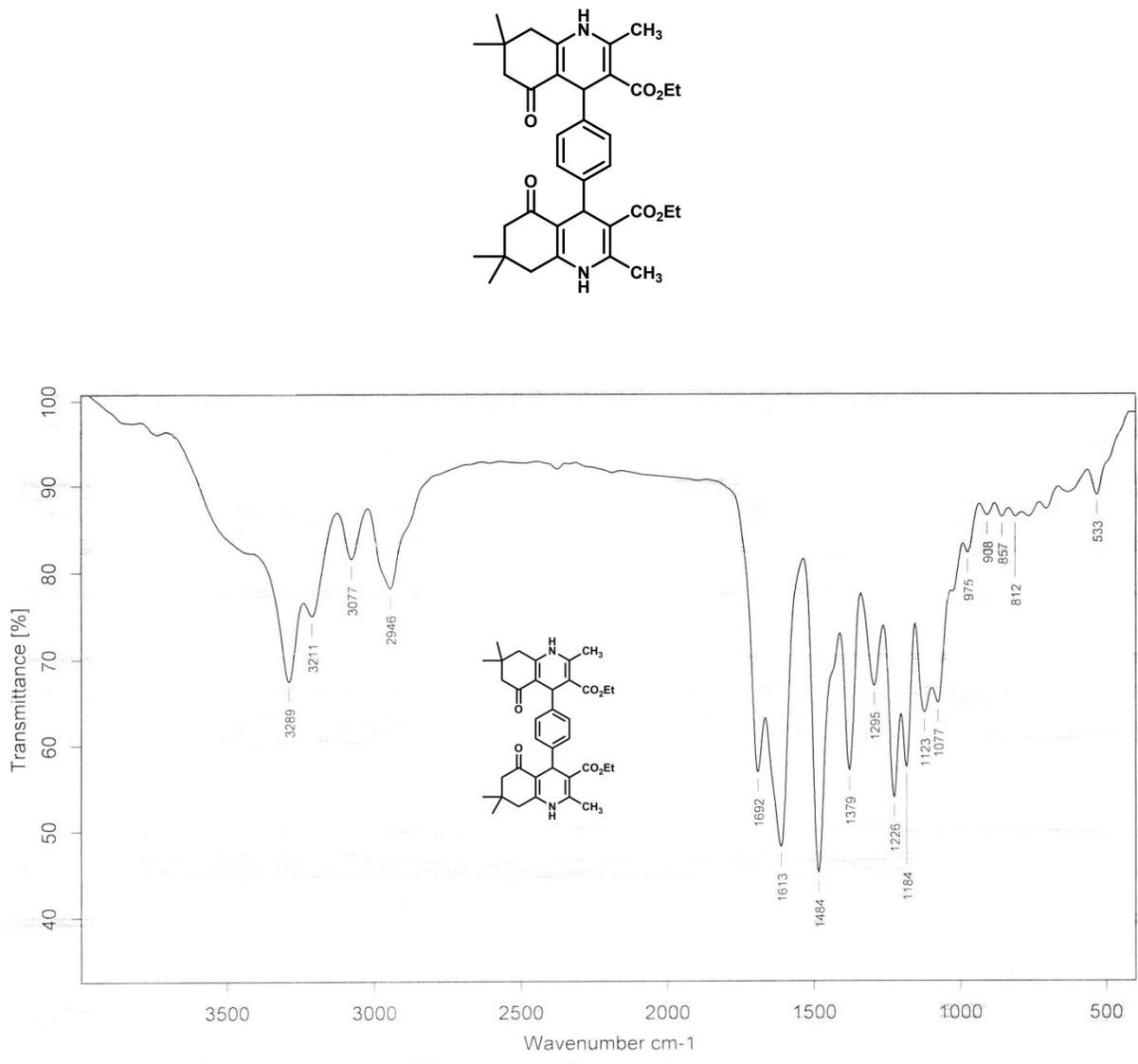
**Diethyl 4,4'-(1,4-phenylene)bis(2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate)**

**(mm):** White solid; m.p. 293-294 °C; FT-IR (KBr):  $\nu = 3289, 3211, 3077, 2946, 1692, 1613, 1484, 1379,$

$1226, 1184\text{ cm}^{-1}$ ,  $^1\text{H}$  NMR (DMSO, 400 MHz):  $\delta = 1.14$  (t, 6H, 2CH<sub>3</sub>), 1.79-1.81 (m, 4H, 2CH<sub>2</sub>), 1.88-1.91

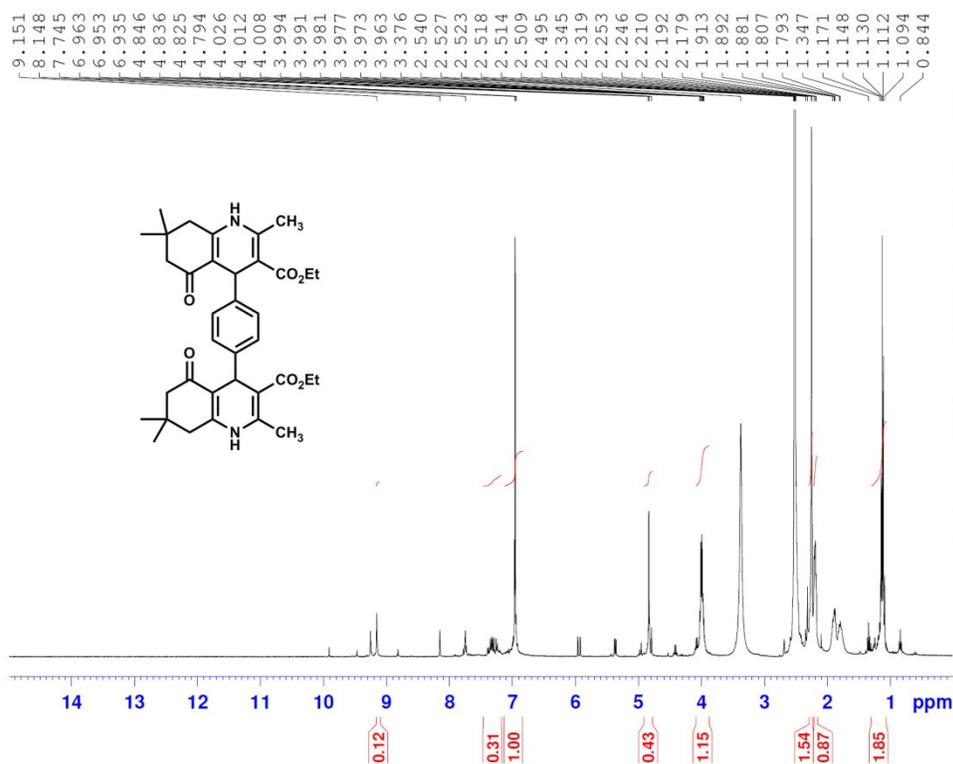
(m, 4H, 2CH<sub>2</sub>), 2.17-2.21 (m, 4H, 2CH<sub>2</sub>), 2.24 (s, 6H, 2CH<sub>3</sub>), 3.9-4.02 (m, 4H, 2CH<sub>2</sub>), 4.8 (s, 2H, CH), 6.95 (s,

4H, ArH), 9.15 (s, 1H, NH), 9.24 (s, 1H, NH).



**Fig. 52.** FT-IR of diethyl 4,4'-(1,4-phenylene)bis(2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate).

Sample code: 3



```
NAME      Gilan UN
EXPNO     1640
PROCNO    1
Date_     20141201
Time      8.03
INSTRUM   spect
PROBHD    5 mm PABBO BB-
PULPROG   zg30
TD         65536
SOLVENT   DMSO
NS         20
DS         0
SWH        8012.820 Hz
FIDRES     0.122266 Hz
AQ         4.0894966 sec
RG         181
DW         62.400 usec
DE         6.50 usec
TE         293.0 K
D1         4.00000000 sec
TD0        1

----- CHANNEL f1 -----
NUC1       1H
P1         14.00 usec
PL1        -2.00 dB
PL1W       11.86359406 W
SFO1       400.2236020 MHz
SI         32768
SF         400.2200000 MHz
WDW        EM
SSB        0
LB         0.30 Hz
GB         0
PC         1.00
```

Fig. 53.  $^1\text{H}$  NMR of diethyl 4,4'-(1,4-phenylene)bis(2,7,7-trimethyl-5-oxo-1,4,5,6,7,8-hexahydroquinoline-3-carboxylate).