

## ELECTRONIC SUPPLEMENTARY INFORMATION

# Bis-arylsulfenyl- and bis-arylselanyl-benzo-2,1,3-thiadiazoles: synthesis and photophysical characterization

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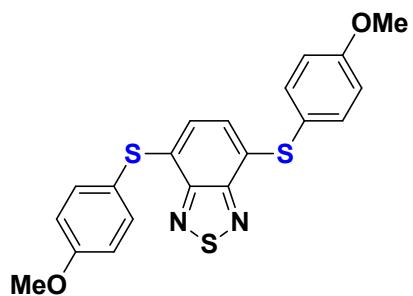
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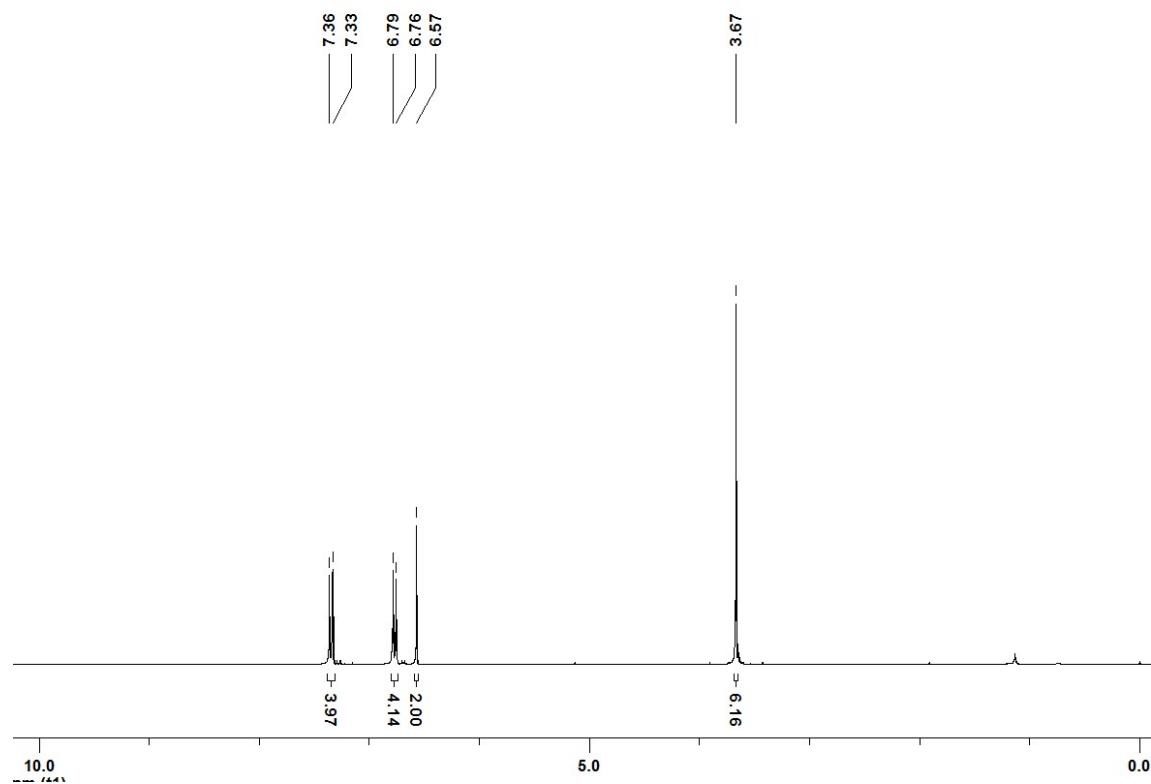
## Summary

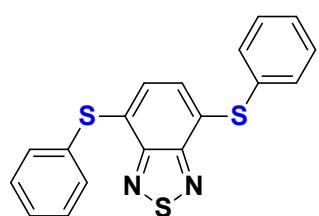
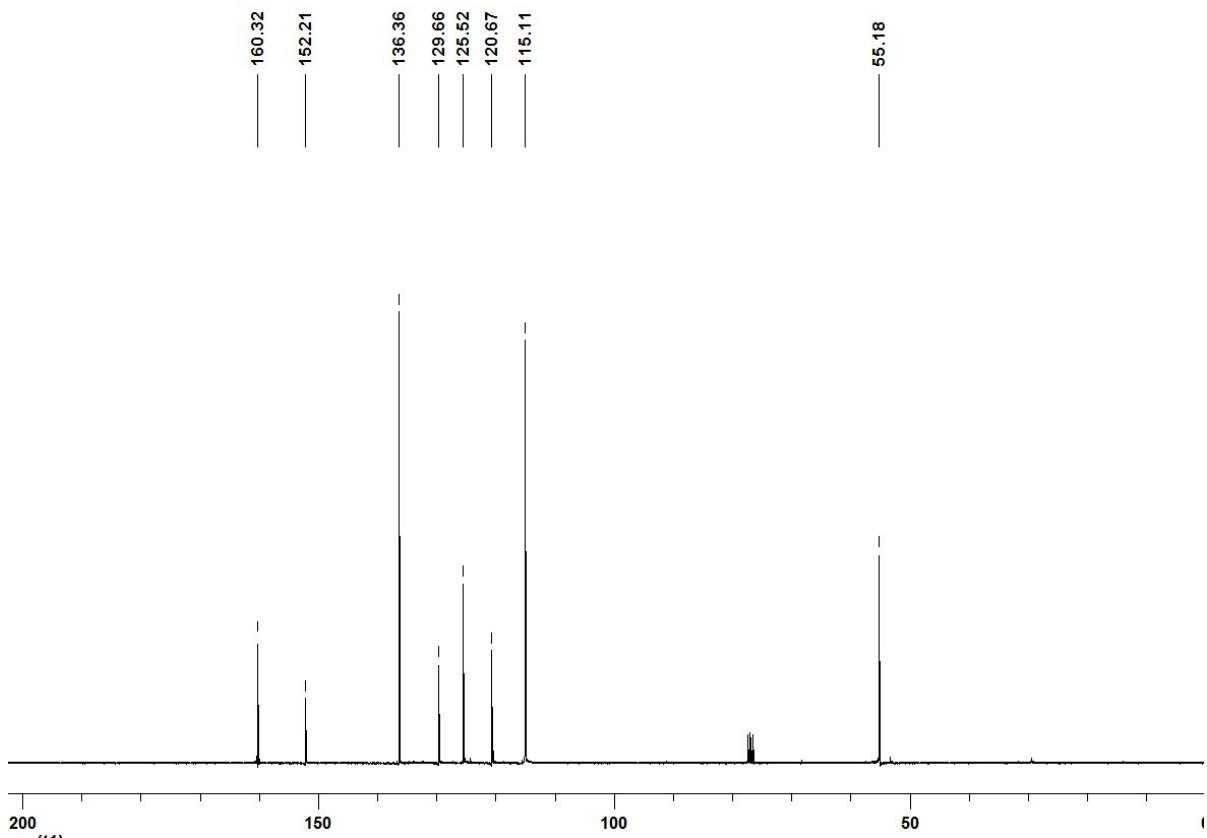
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## 1. Spectroscopic characterization

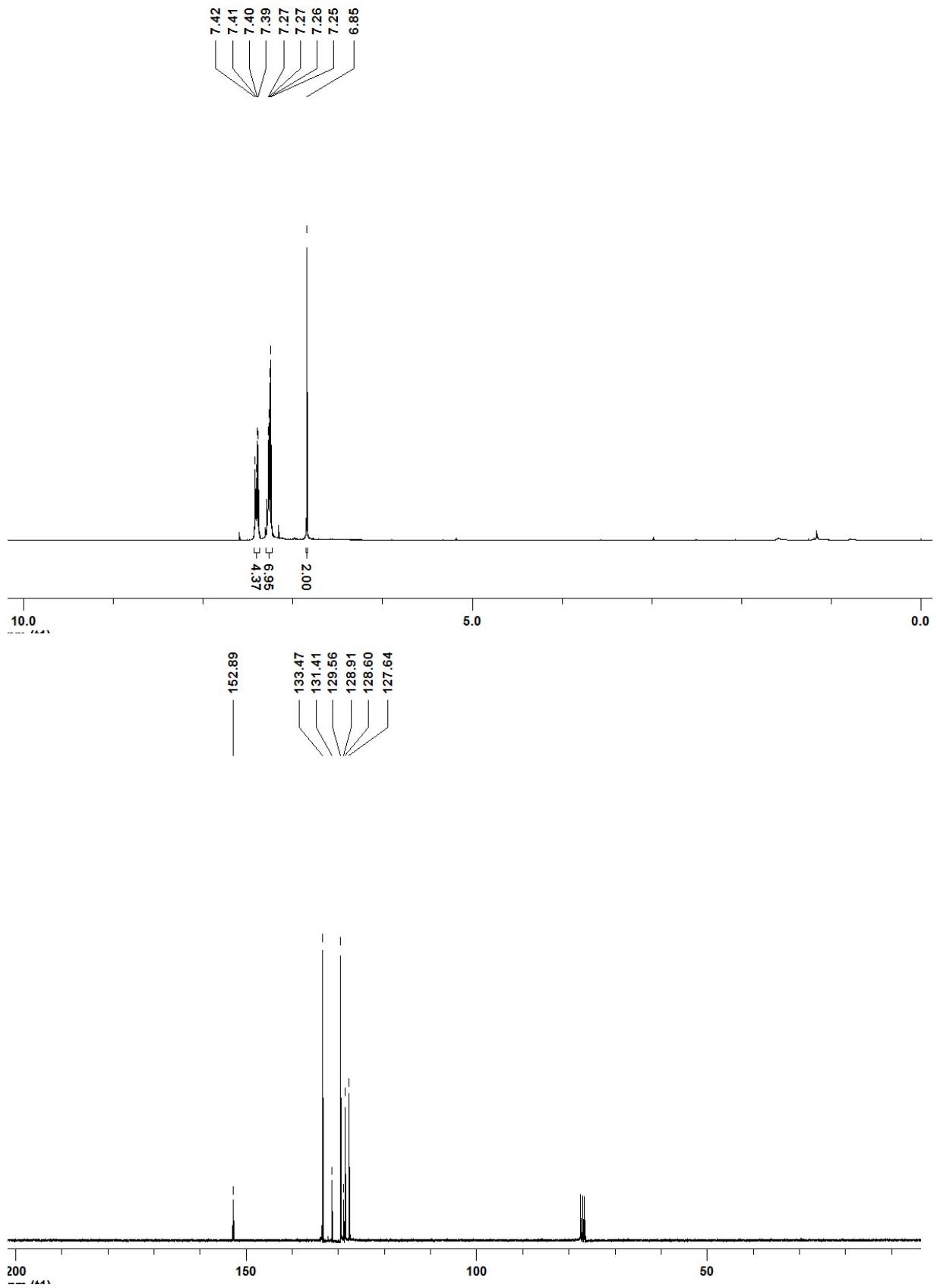


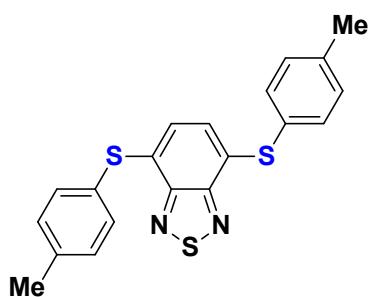
4,7-bis((4-methoxyphenyl)thio)benzo[c][1,2,5]thiadiazole: Yield: 0.179 g (87%); orange oil.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 300 MHz):  $\delta$  7.34 (d,  $J = 8.8$  Hz, 4H); 6.77 (d,  $J = 8.8$  Hz, 4H); 6.57 (s, 2H); 3.67 (s, 6H). RMN  $^{13}\text{C}$  ( $\text{CDCl}_3$  75 MHz):  $\delta$  160.32, 152.21, 136.36, 129.66, 125.52, 120.67, 115.11, 55.18. MS (relative intensity) m/z: 412 (59), 273 (100), 207 (23), 139 (28), 96 (16), 77 (11). HRMS calcd. for  $\text{C}_{20}\text{H}_{17}\text{N}_2\text{O}_2\text{S}_3$  [ $\text{M} + \text{H}]^+$  413.0446. Found: 413.0422.



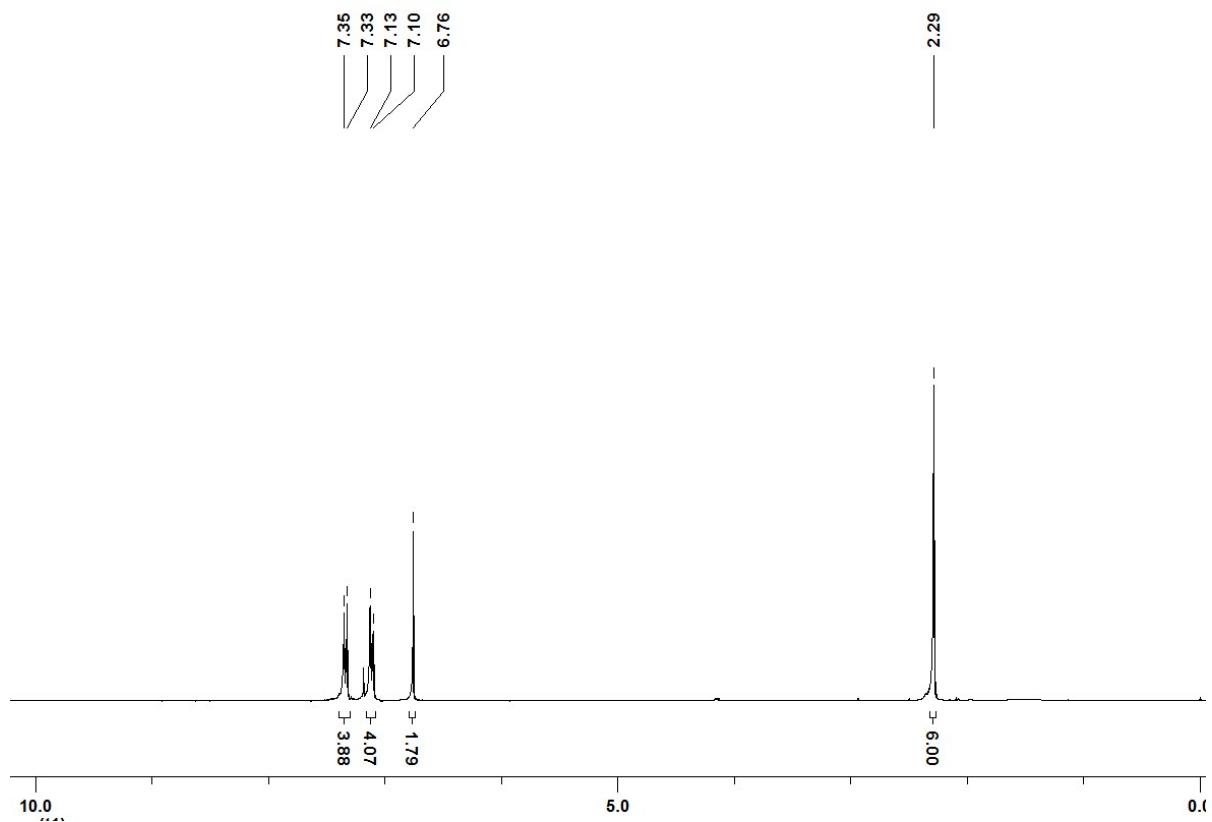


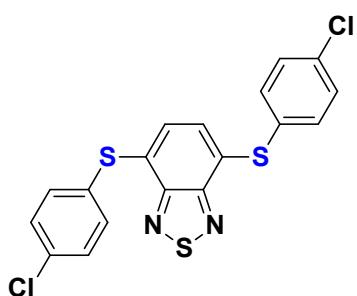
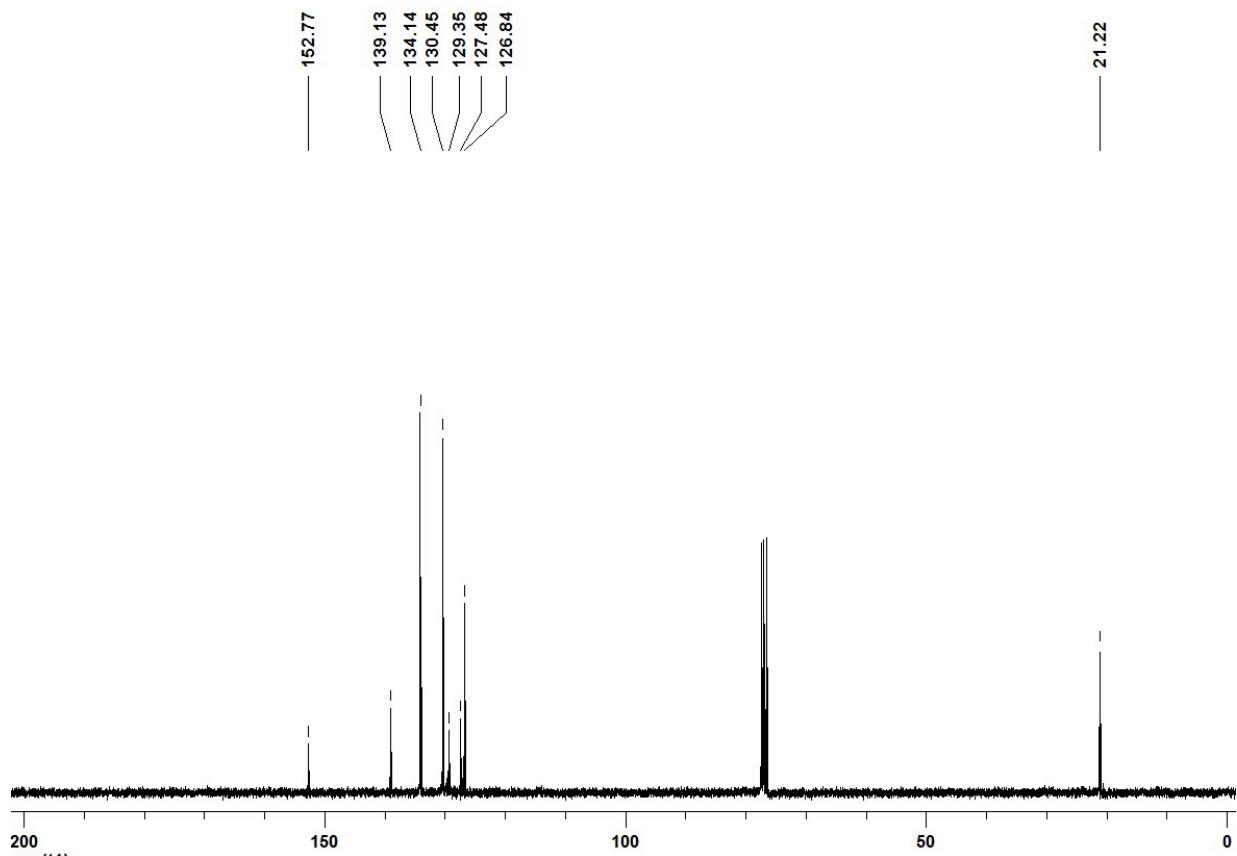
4,7-bis(phenylthio)benzo[1,2,5]thiadiazole: Yield: 0.151 g (86%); yellow solid; mp 125-126°C.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 300 MHz):  $\delta$  7.42-7.39 (m, 4H); 7.27-7.25 (m, 6H); 6.84 (s, 2H). RMN  $^{13}\text{C}$  ( $\text{CDCl}_3$  75 MHz):  $\delta$  152.89, 133.47, 131.41, 129.56, 128.91, 128.60, 127.64. MS (relative intensity) m/z: 352 (60), 243 (93), 207 (48), 109 (40), 77 (100), 51 (88). HRMS calcd. for  $\text{C}_{18}\text{H}_{13}\text{N}_2\text{S}_3$  [ $\text{M} + \text{H}]^+$  353.0235. Found: 353.0238.



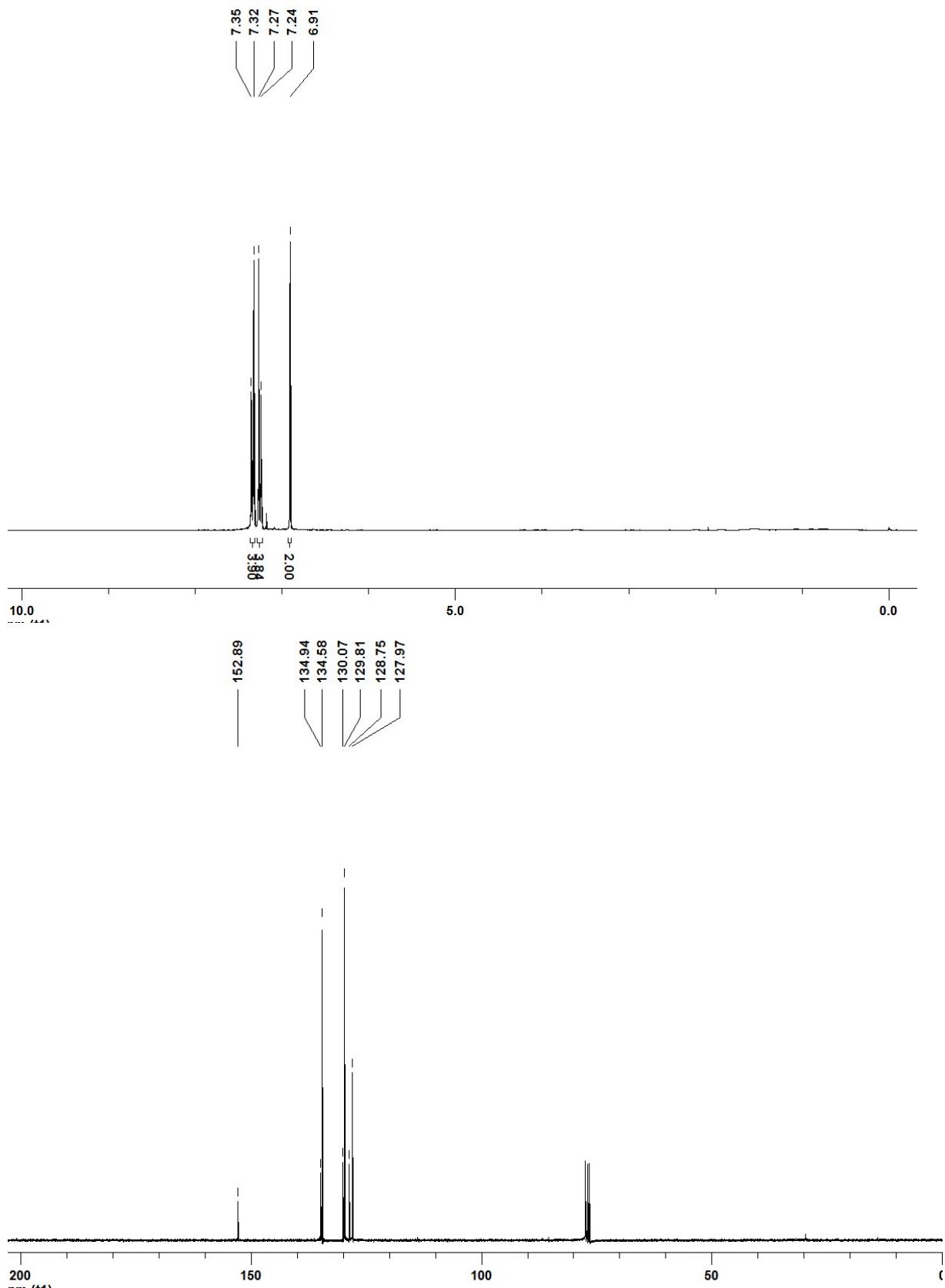


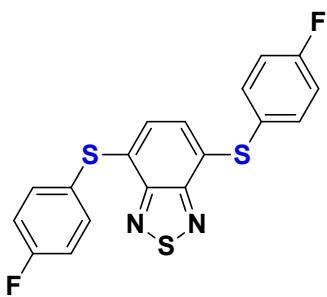
4,7-bis(p-tolylthio)benzo[c][1,2,5]thiadiazole: Yield: 0.158 g (83%); orange oil.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 300 MHz):  $\delta$  7.34 (d,  $J = 8.1$  Hz, 4H); 7.12 (d,  $J = 8.1$  Hz, 4H); 6.76 (s, 2H); 2.29 (s, 6H). RMN  $^{13}\text{C}$  ( $\text{CDCl}_3$  75 MHz):  $\delta$  152.77, 139.13, 134.14, 130.45, 129.35, 127.48, 126.84, 21.22. MS (relative intensity) m/z: 380 (14), 281 (25), 257 (25), 207 (100), 73 (20), 40 (41). HRMS calcd. for  $\text{C}_{20}\text{H}_{17}\text{N}_2\text{S}_3$  [ $\text{M} + \text{H}]^+$  381.0548. Found: 381.0548.



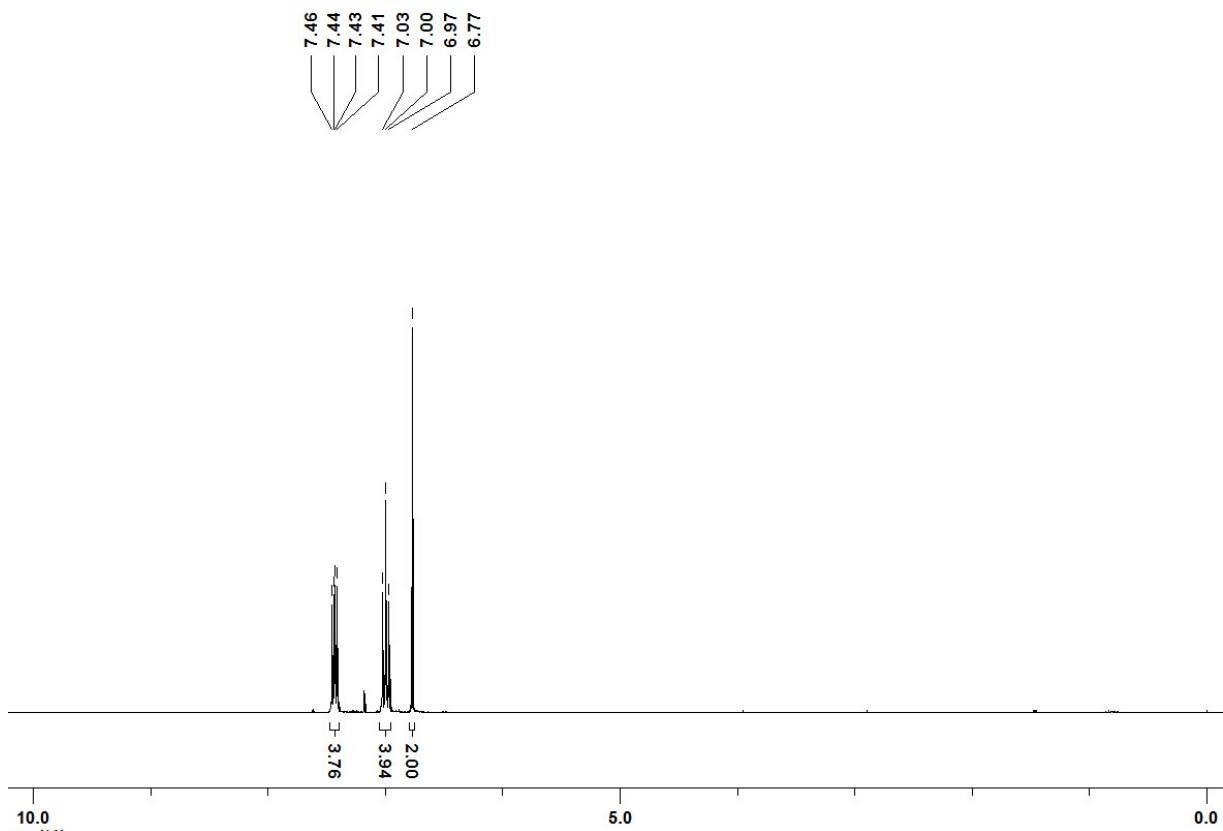


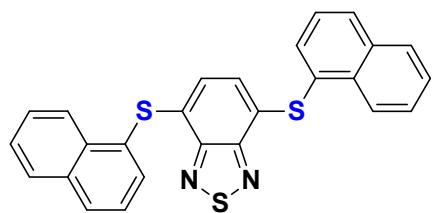
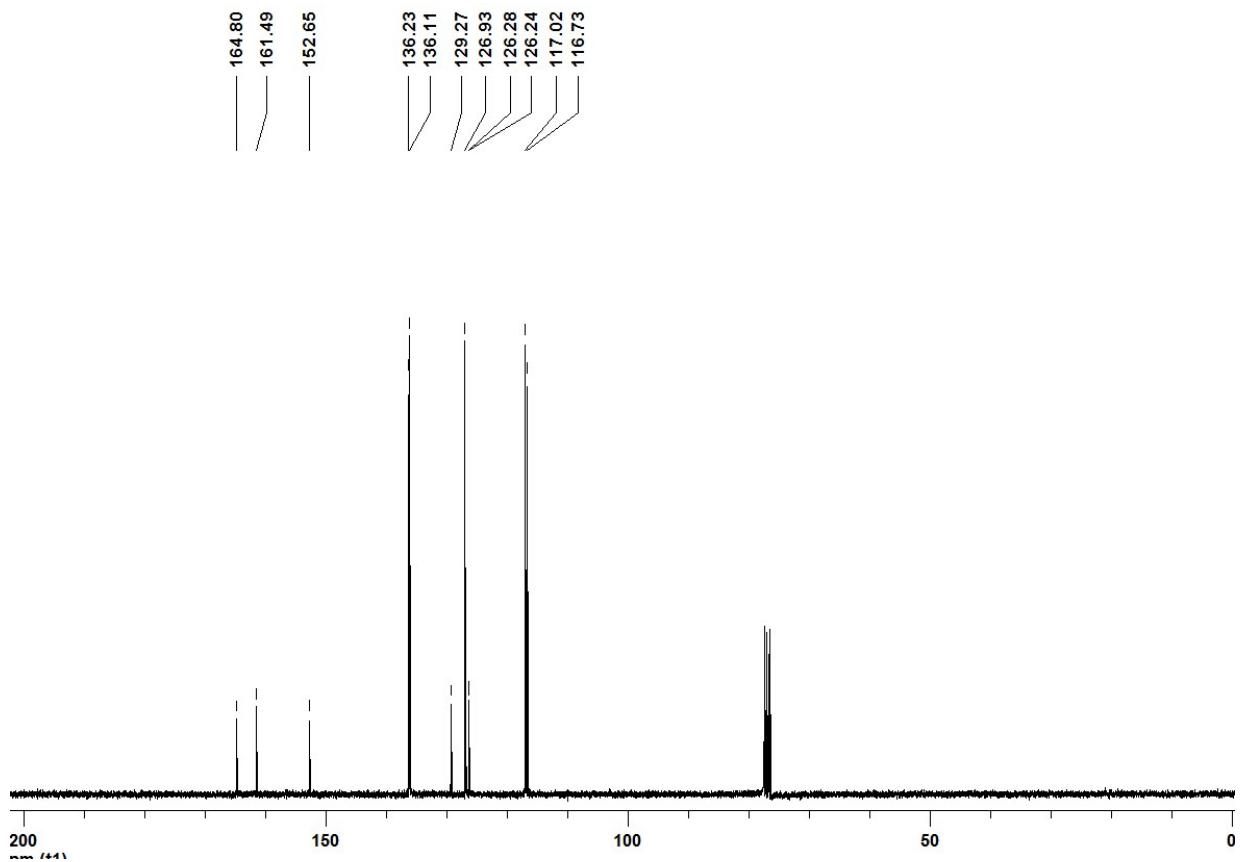
4,7-bis((4-chlorophenyl)thio)benzo[c][1,2,5]thiadiazole: Yield: 0.168 g (80%); orange solid; mp 110–112°C. <sup>1</sup>H NMR ( $\text{CDCl}_3$ , 300 MHz):  $\delta$  7.34 (d,  $J = 8.5$  Hz, 4H); 7.25 (d,  $J = 8.5$  Hz, 4H); 6.91 (s, 2H). RMN <sup>13</sup>C (CDCl<sub>3</sub> 75 MHz):  $\delta$  152.89, 134.94, 134.58, 130.07, 129.81, 128.75, 127.97. MS (relative intensity) m/z: 420 (15), 277 (100), 242 (39), 207 (17), 108 (32), 75 (26). HRMS calcd. for C<sub>18</sub>H<sub>11</sub>Cl<sub>2</sub>N<sub>2</sub>S<sub>3</sub> [M + H]<sup>+</sup> 420.9455. Found: 420.9451.



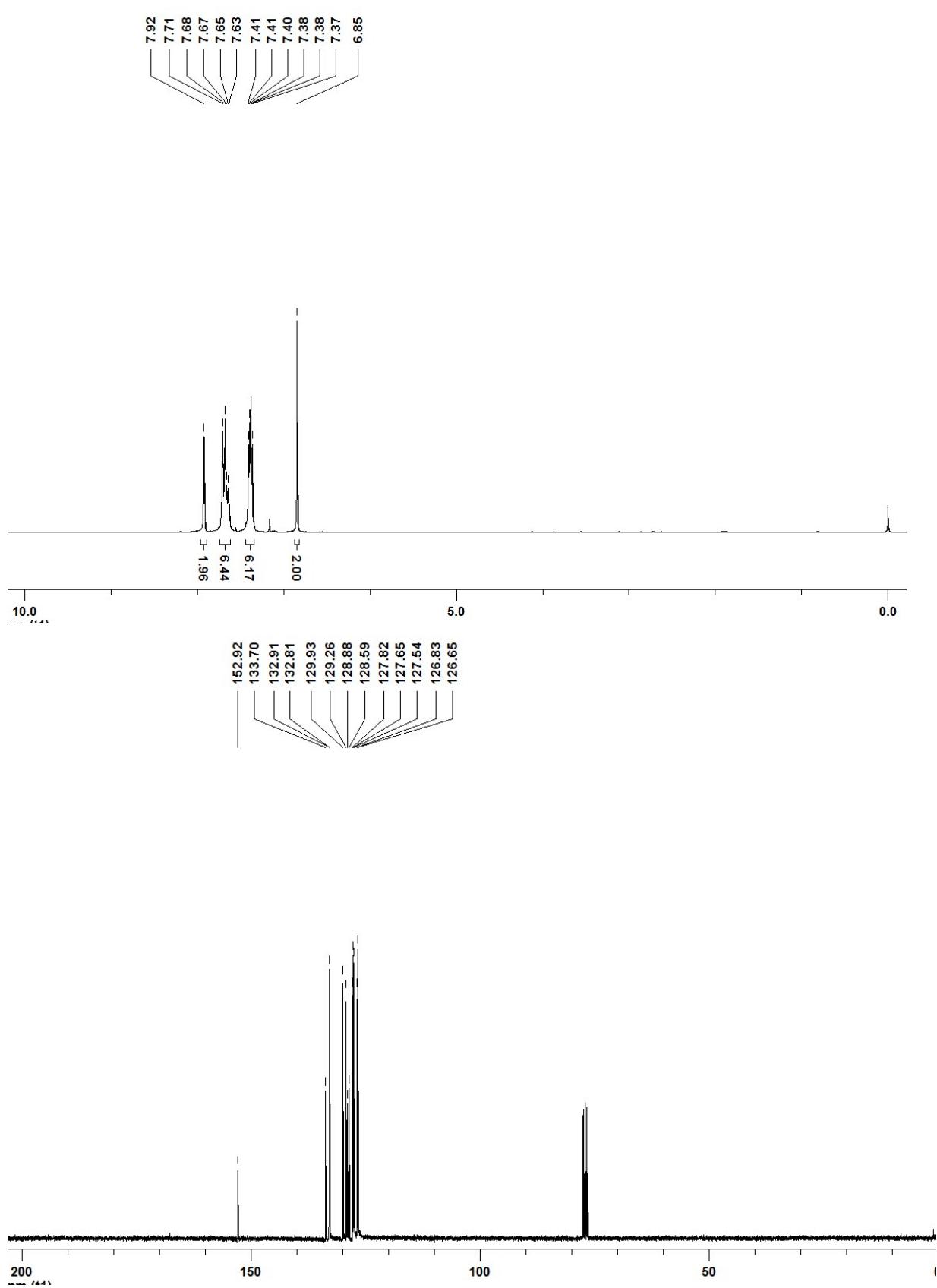


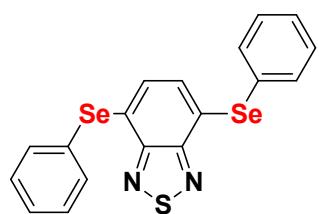
4,7-bis((4-fluorophenyl)thio)benzo[*c*][1,2,5]thiadiazole: Yield: 0.173 g (89%); yellow solid; mp 132-134°C.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 300 MHz):  $\delta$  7.44 (dd,  $J$  = 8.8 and 5.2 Hz, 4H); 7.00 (t,  $J$  = 8.8 Hz, 4H); 6.77 (s, 2H). RMN  $^{13}\text{C}$  ( $\text{CDCl}_3$  75 MHz):  $\delta$  163.14 (d,  $J$  = 249.7 Hz), 152.65, 136.17 (d,  $J$  = 8.4 Hz), 129.27, 126.93, 126.26 (d,  $J$  = 3.7 Hz), 116.87 (d,  $J$  = 22.0 Hz). MS (relative intensity) m/z: 388 (23), 281 (27), 261 (36), 207 (100), 73 (23). HRMS calcd. for  $\text{C}_{18}\text{H}_{11}\text{F}_2\text{N}_2\text{S}_3$  [M + H] $^+$  389.0046. Found: 389.0043.



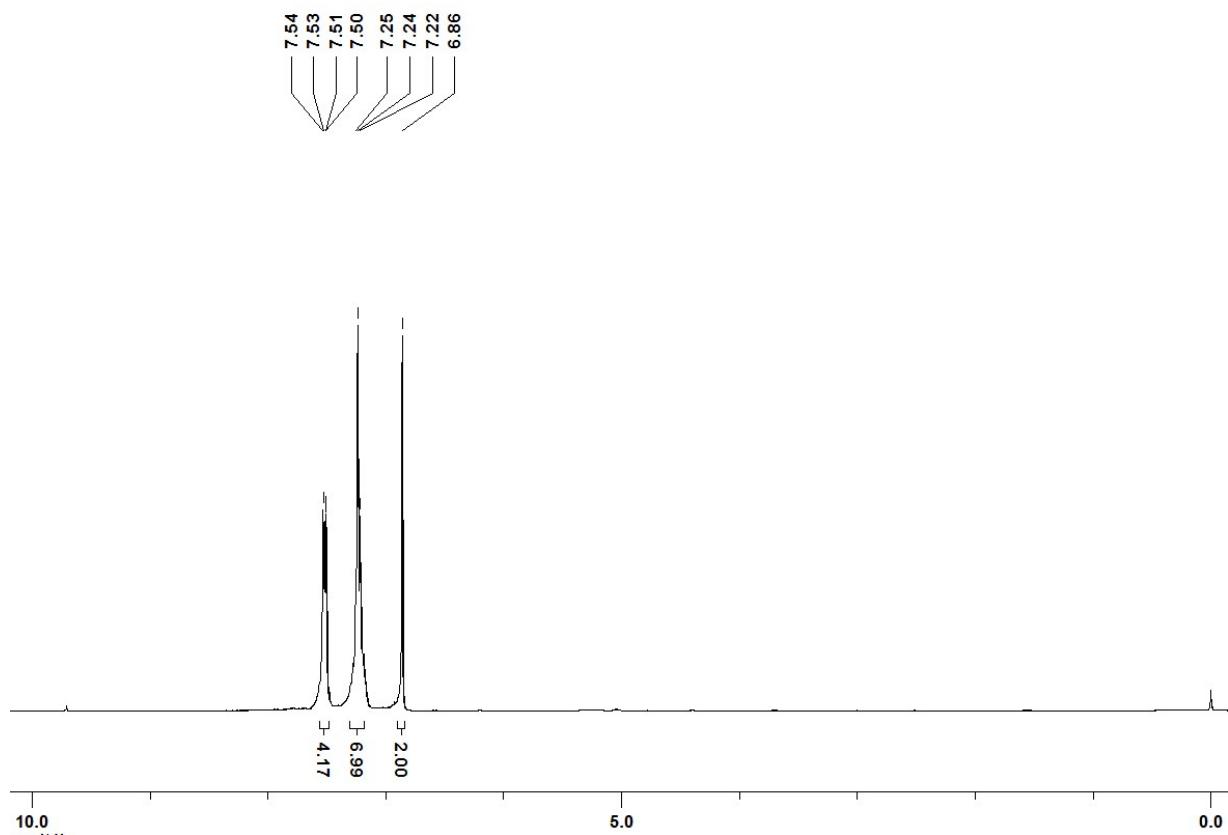


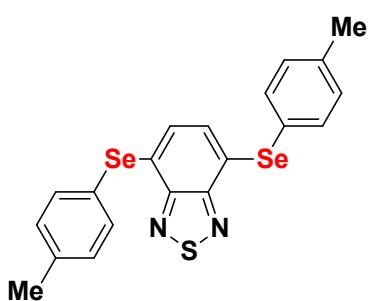
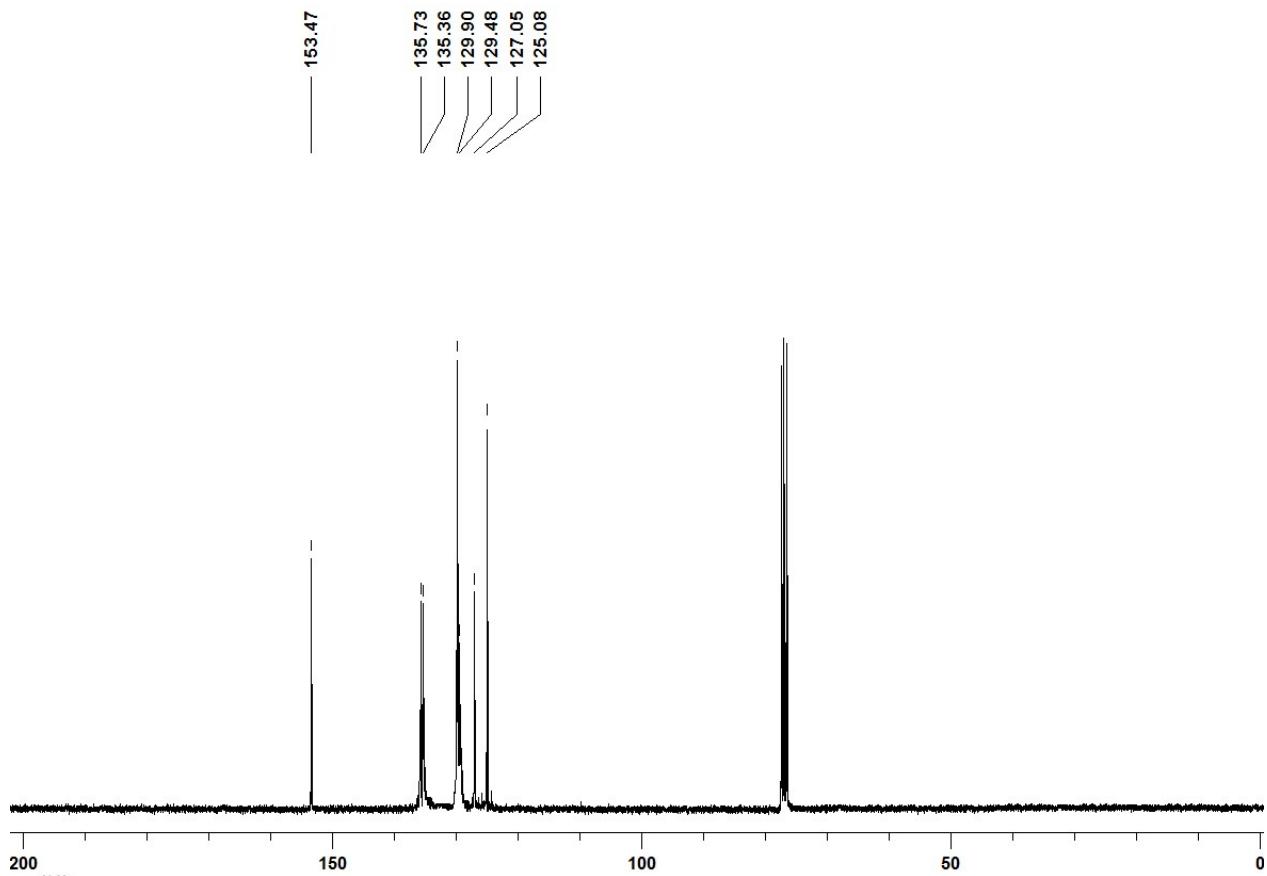
4,7-bis(naphthalen-1-ylthio)benzo[c][1,2,5]thiadiazole: Yield: 0.203 g (90%); yellow solid; mp 104-106°C.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 300 MHz):  $\delta$  7.92 (m, 2H); 7.71-7.63 (m, 6H); 7.41-7.37 (m, 6H); 6.85 (s, 2H). RMN  $^{13}\text{C}$  ( $\text{CDCl}_3$ , 75 MHz):  $\delta$  152.92, 133.70, 132.91, 132.81, 129.93, 129.26, 128.88, 128.59, 127.82, 127.65, 127.54, 126.83, 126.65. MS (relative intensity) m/z: 452 (2), 368 (11), 111 (25), 83 (23), 55 (100), 43 (93). HRMS calcd. for  $\text{C}_{26}\text{H}_{17}\text{N}_2\text{S}_3$  [ $\text{M} + \text{H}$ ]<sup>+</sup> 453.0548. Found: 453.0547.



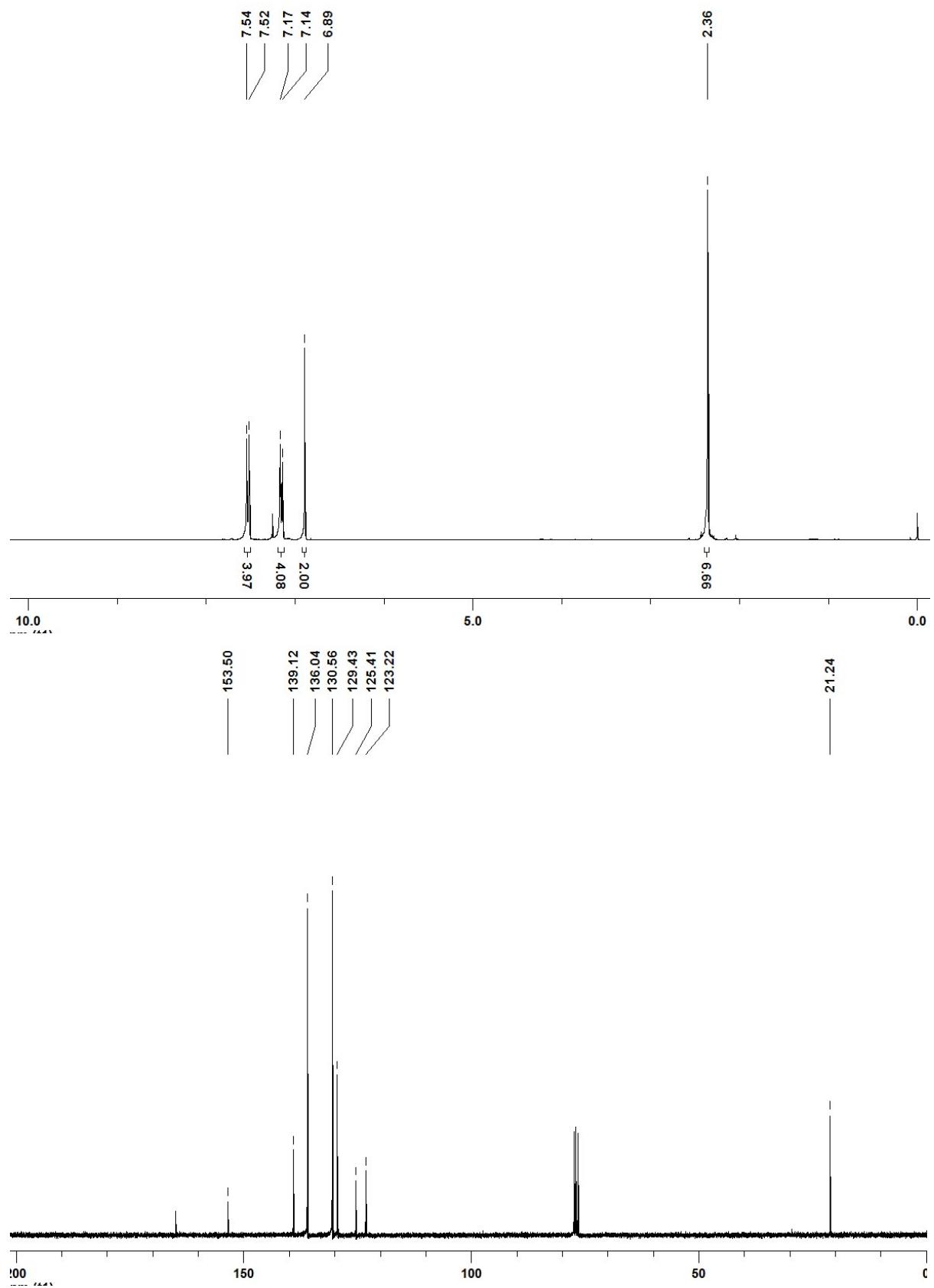


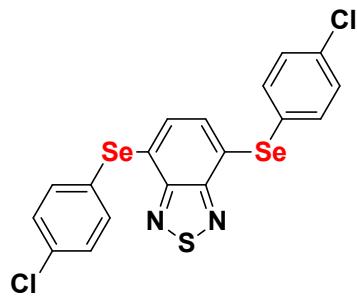
4,7-bis(phenylselanyl)benzo[c][1,2,5]thiadiazole: Yield: 0.175 g (78%); dark yellow solid; mp 129-131°C.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 300 MHz):  $\delta$  7.54-7.50 (m, 4H); 7.25-7.22 (m, 6H); 6.86 (s, 2H). RMN  $^{13}\text{C}$  ( $\text{CDCl}_3$  75 MHz):  $\delta$  153.47, 135.73, 135.36, 129.90, 129.48, 127.05, 125.08. MS (relative intensity) m/z: 448 (9), 291 (25), 207 (14), 77 (79), 51 (51), 40 (100). HRMS calcd. for  $\text{C}_{18}\text{H}_{13}\text{N}_2\text{SSe}_2$  [ $\text{M} + \text{H}]^+$  448.9126. Found: 448.9107.



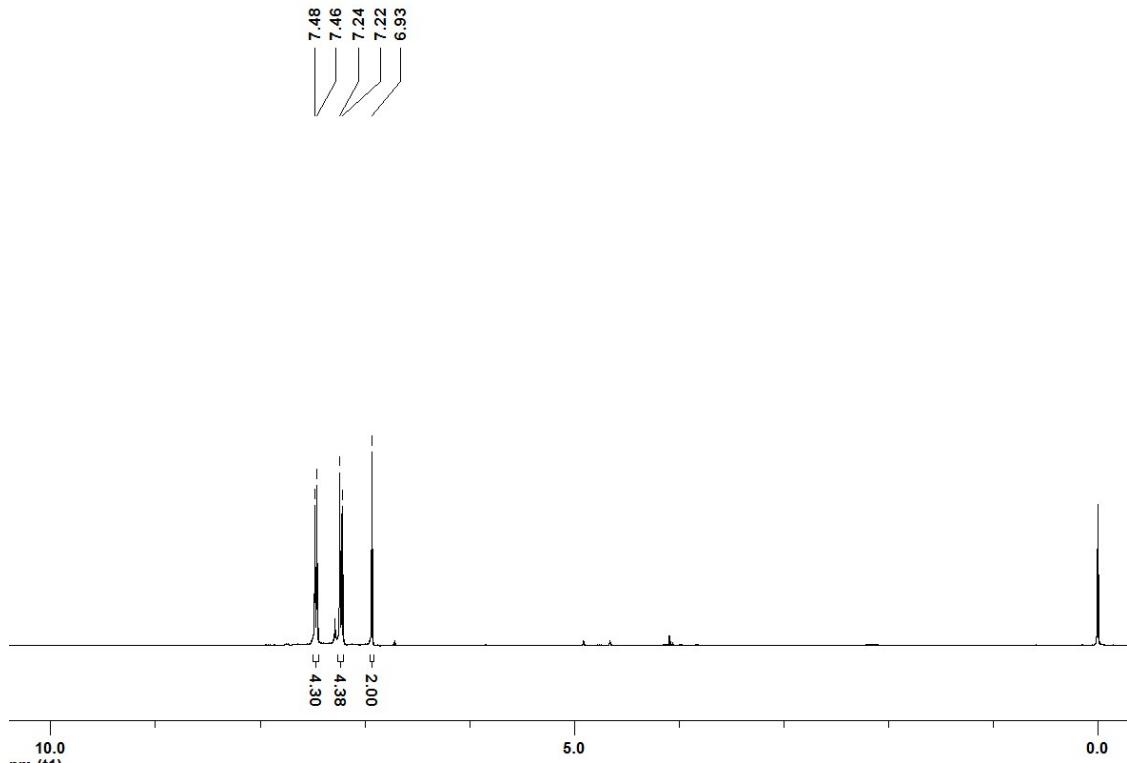


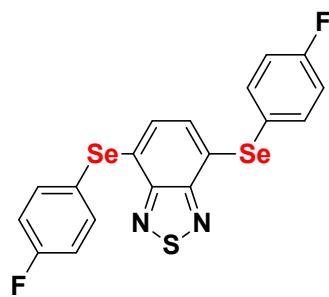
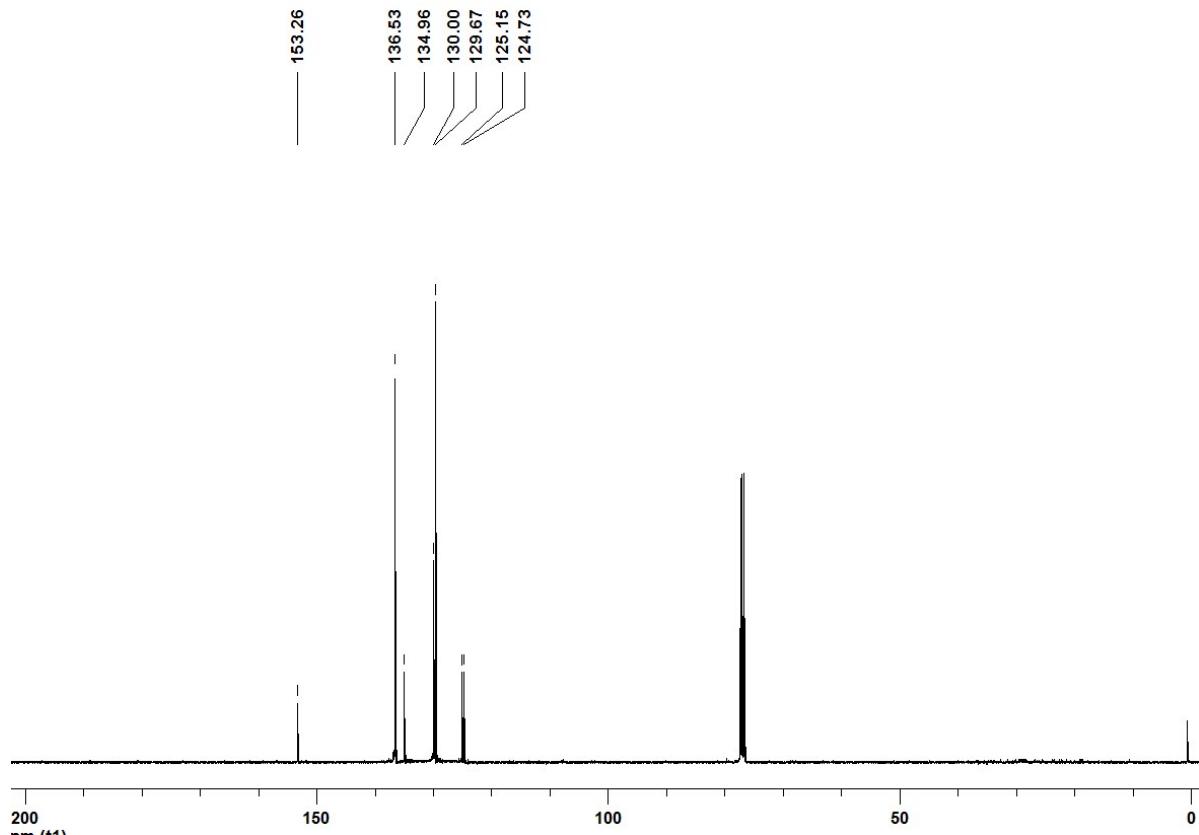
4,7-bis(p-tolylselanyl)benzo[c][1,2,5]thiadiazole: Yield: 0.164 g (69%); yellow solid; mp 101–103°C.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 300 MHz):  $\delta$  7.53 (d,  $J$  = 8.1 Hz, 4H); 7.15 (d,  $J$  = 8.1 Hz, 4H); 6.89 (s, 2H); 2.36 (s, 6H). RMN  $^{13}\text{C}$  ( $\text{CDCl}_3$  75 MHz):  $\delta$  153.50, 139.12, 136.04, 130.56, 129.43, 125.41, 123.22, 21.24. MS (relative intensity) m/z: 476 (5), 305 (12), 111 (17), 81 (57), 69 (100), 43 (88). HRMS calcd. for  $\text{C}_{20}\text{H}_{17}\text{N}_2\text{SSe}_2$  [M + H] $^+$  476.9439. Found: 476.9392.



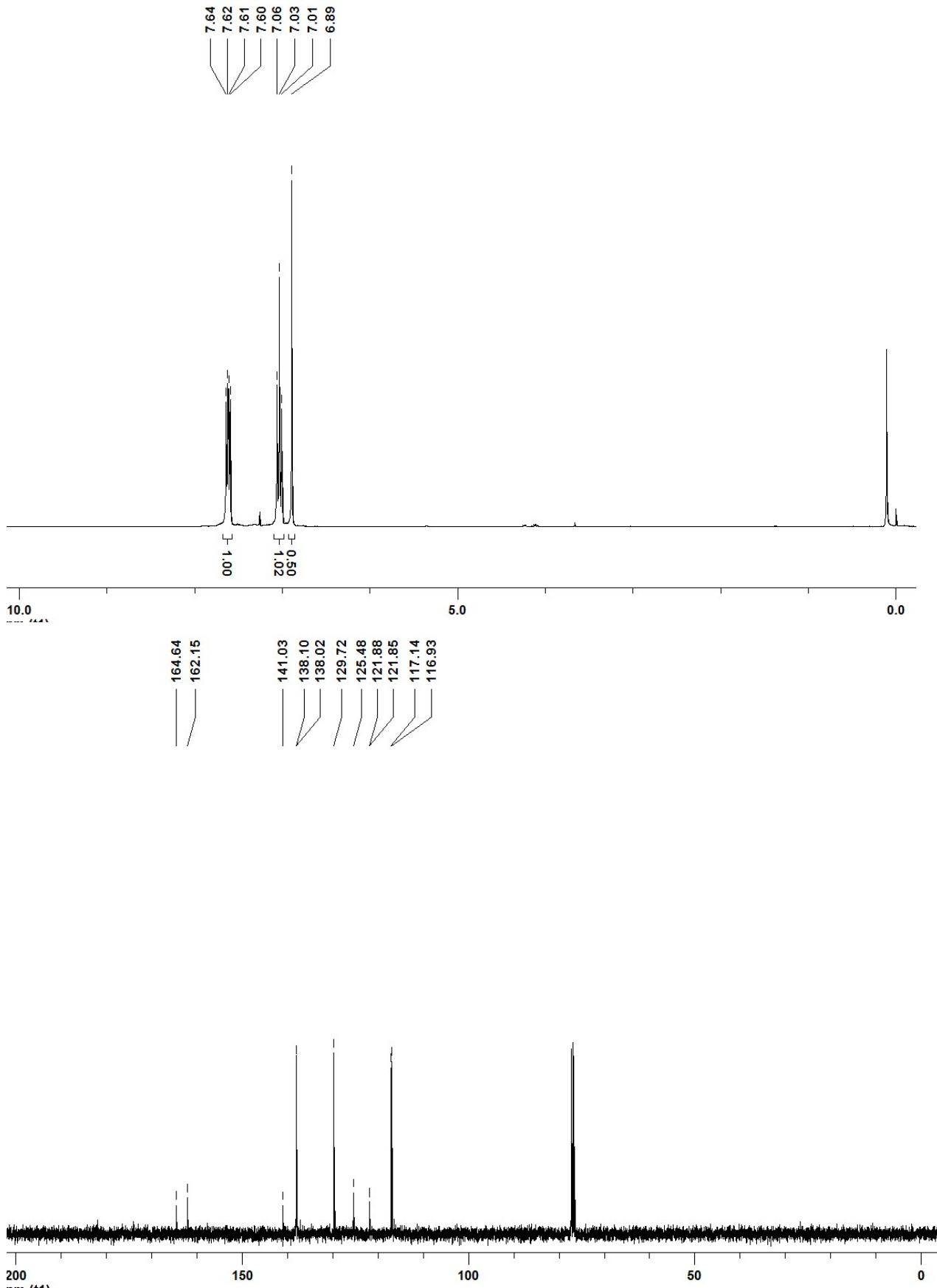


4,7-bis((4-chlorophenyl)selanyl)benzo[*c*][1,2,5]thiadiazole: Yield: 0.188 g (73%); orange solid; mp 87-89°C.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 300 MHz):  $\delta$  7.47 (d,  $J$  = 8.4 Hz, 4H); 7.23 (d,  $J$  = 8.4 Hz, 4H); 6.93 (s, 2H). RMN  $^{13}\text{C}$  ( $\text{CDCl}_3$  75 MHz):  $\delta$  153.26, 136.53, 134.96, 130.00, 129.67, 125.15, 124.73. MS (relative intensity) m/z: 516 (8), 325 (25), 290 (5), 81 (63), 69 (100), 41 (33). HRMS calcd. for  $\text{C}_{18}\text{H}_{11}\text{Cl}_2\text{N}_2\text{SSe}_2$  [ $\text{M} + \text{H}]^+$  516.8339. Found: 516.8318.



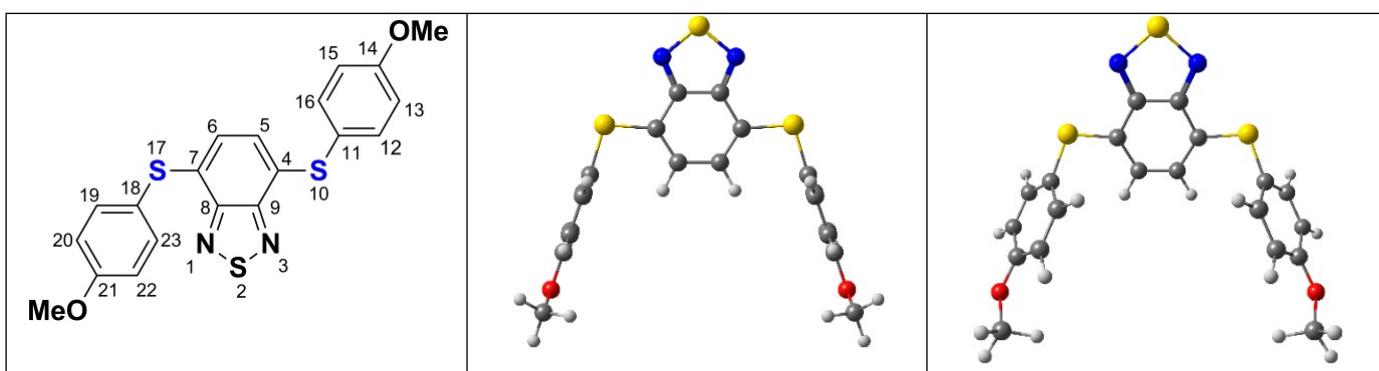


4,7-bis((4-fluorophenyl)selanyl)benzo[c][1,2,5]thiadiazole: Yield: 0.189 g (78%); orange solid; mp 151–153°C.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ , 300 MHz):  $\delta$  7.62 (dd,  $J$  = 8.8 and 5.4 Hz, 4H); 7.03 (t,  $J$  = 8.8 Hz, 4H); 6.89 (s, 2H). RMN  $^{13}\text{C}$  ( $\text{CDCl}_3$  75 MHz):  $\delta$  163.39 (d,  $J$  = 249.8 Hz), 141.03, 138.06 (d,  $J$  = 8.1 Hz), 129.72, 125.48, 121.87 (d,  $J$  = 3.6 Hz), 117.03 (d,  $J$  = 22.0 Hz). MS (relative intensity) m/z: 484 (39), 324 (40), 309 (100), 229 (30), 83 (35), 69 (24). HRMS calcd. for  $\text{C}_{18}\text{H}_{11}\text{F}_2\text{N}_2\text{SSe}_2$  [M + H] $^+$  484.8937. Found: 484.8906.



## 2. Theoretical calculations

**Table ESI1.** Theoretical structural data of **3a-f** and **5a-d** in different organic solvents, where the bond lengths (**r**) are presented in Å and angles (**a**) and dihedral angles (**d**) are given in degrees. Geometries calculated with PBE1PBE/jun-cc-pVTZ//PBE1PBE/cc-pVDZ. The representative equilibrium structures are plotted at the same level using hexane as solvent. The ground state structures are given on the left and first excited state structures are given on the right. In hexane, the structural data is also given with CAM-B3LYP/jun-cc-pVTZ//CAM-B3LYP/cc-pVDZ (marked with a \*). For all structures: d1 = C<sub>7</sub>-S<sub>17</sub>-C<sub>18</sub>-C<sub>19</sub> / r1= C<sub>7</sub>-S<sub>17</sub> / r2 = S<sub>17</sub>-C<sub>18</sub> / r3 = N<sub>1</sub>-S<sub>2</sub> / a1 = C<sub>7</sub>-S<sub>17</sub>-C<sub>18</sub> / a2 = N<sub>1</sub>-S<sub>2</sub>-N<sub>3</sub>, except for **3f**, where d1 = C<sub>7</sub>-S<sub>21</sub>-C<sub>22</sub>-C<sub>23</sub> / r1= C<sub>7</sub>-S<sub>21</sub> / r2 = S<sub>21</sub>-C<sub>22</sub> / r3 = N<sub>1</sub>-S<sub>2</sub> / a1 = C<sub>7</sub>-S<sub>21</sub>-C<sub>22</sub> / a3 = N<sub>1</sub>-S<sub>2</sub>-N<sub>3</sub>



Dye	Solvent	<b>S<sub>0</sub></b>						<b>S<sub>1</sub></b>					
		<b>d1</b>	<b>r1</b>	<b>r2</b>	<b>r3</b>	<b>a1</b>	<b>a2</b>	<b>d1</b>	<b>r1</b>	<b>r2</b>	<b>r3</b>	<b>a1</b>	<b>a2</b>
<b>3a</b>	Hexane	92.67	1.778	1.786	1.641	102.31	99.43	68.90	1.737	1.783	1.697	104.37	98.64
	Hexane	92.32	1.772	1.781	1.639	102.16	100.09	49.22	1.745	1.766	1.690	104.63	99.80
	Toluene	92.17	1.772	1.781	1.639	102.21	100.07	49.35	1.745	1.766	1.691	104.68	99.74
	DCM	91.94	1.772	1.781	1.639	102.44	99.97	49.39	1.743	1.766	1.693	104.92	99.52
	Ethanol	91.54	1.773	1.781	1.639	102.51	99.94	49.31	1.743	1.766	1.693	105.00	99.45
	DMF	91.51	1.773	1.781	1.639	102.53	99.93	49.30	1.742	1.766	1.693	105.02	99.43

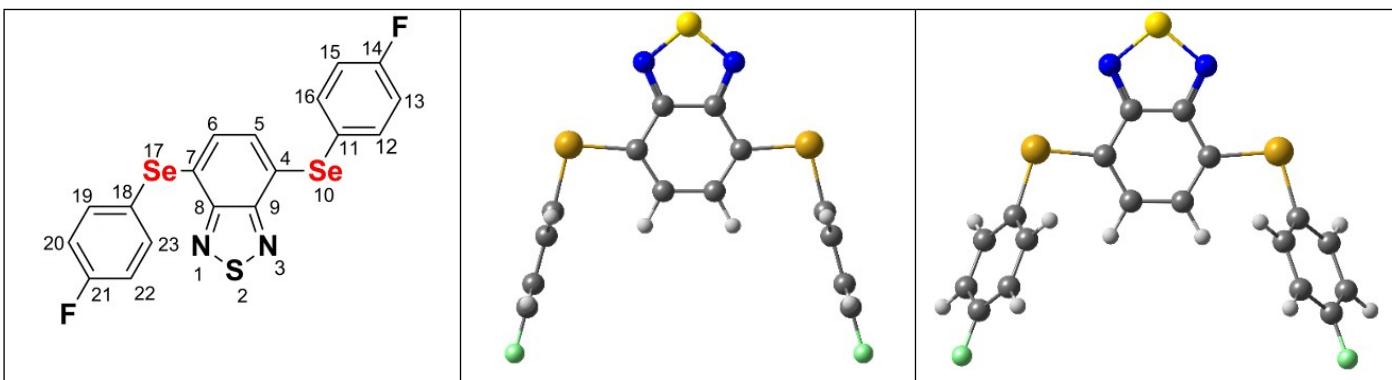








**Table ESI1.** continuation



Dye	Solvent	S <sub>0</sub>						S <sub>1</sub>					
		d1	r1	r2	r3	a1	a2	d1	r1	r2	r3	a1	a2
<b>5d</b>	Hexane	91.40	1.915	1.923	1.641	99.33	99.38	91.69	1.871	1.924	1.698	100.96	98.60
	Hexane	91.33	1.909	1.919	1.640	99.00	100.05	61.54	1.875	1.913	1.693	101.35	99.79
	Toluene	91.32	1.909	1.919	1.640	99.04	100.03	61.90	1.875	1.913	1.693	101.38	99.73
	DCM	91.29	1.910	1.919	1.640	99.20	99.94	61.32	1.873	1.912	1.695	101.56	99.51
	Ethanol	91.28	1.910	1.919	1.639	99.26	99.92	61.83	1.872	1.912	1.695	101.64	99.44
	DMF	91.28	1.910	1.919	1.639	99.27	99.91	61.71	1.872	1.912	1.695	101.66	99.42

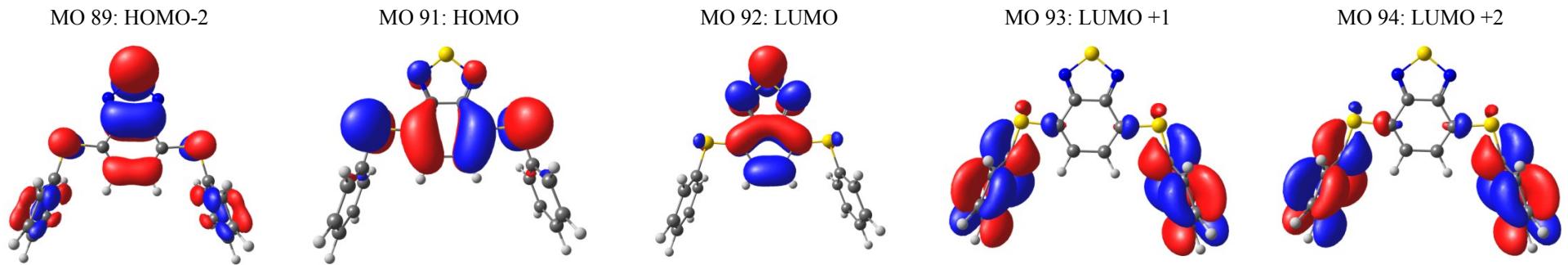


**Table ESI3.** Computed vertical absorption energies and vertical emission energies in hexane and in dichloromethane for **3b** structure of optimized S<sub>0</sub> and S<sub>1</sub> state at the CAM-B3LYP/jun-cc-pVTZ level.

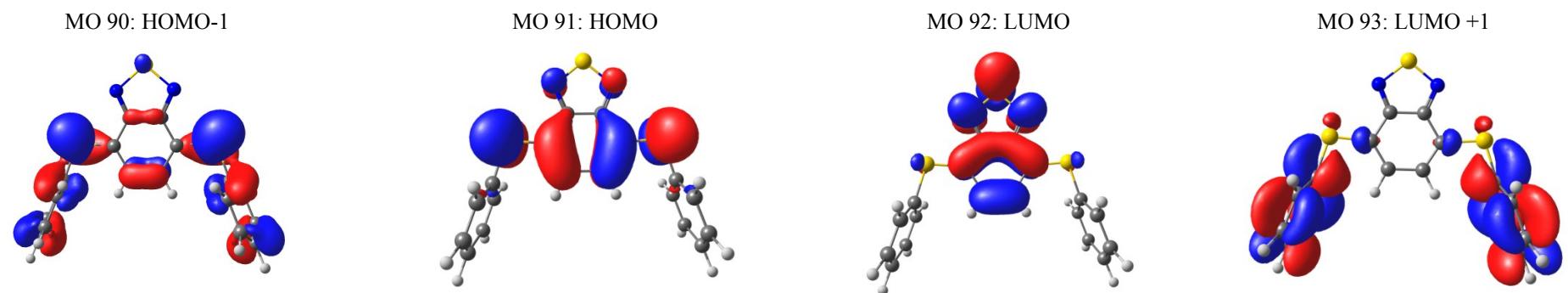
Structure*		Energy (eV)	Wavelength	Force	Transition	coefficient
<b>3b</b> Hexane (S <sub>0</sub> geometry)	S0 → S1	3.017	410.96	0.184	91 → 92	0.703
	S0 → S2	4.463	277.78	0.244	89 → 92	0.605
	S0 → S3	4.575	270.98	0.000	91 → 93	0.629
	S0 → S4	4.625	268.06	0.000	91 → 94	0.624
<b>3b</b> Hexane (S <sub>1</sub> geometry)	S0 → S1	2.292	541.05	0.166	91 → 92	0.704
	S0 → S2	4.086	303.41	0.148	90 → 92	0.493
	S0 → S3	4.327	286.55	0.079	91 → 93	0.650
<b>3b</b> DCM (S <sub>0</sub> geometry)	S0 → S1	3.011	411.83	0.184	91 → 92	0.70275
	S0 → S2	4.451	278.55	0.262	87 → 92	0.49671
	S0 → S3	4.669	265.53	0.000	91 → 93	0.62625
	S0 → S4	4.720	262.66	0.000	91 → 94	0.61892
<b>3b</b> DCM (S <sub>1</sub> geometry)	S0 → S1	2.219	558.72	0.2156	91 → 92	0.704
	S0 → S2	4.030	307.66	0.2715	90 → 92	0.425
	S0 → S3	4.306	287.92	0.0771	90 → 92	0.500



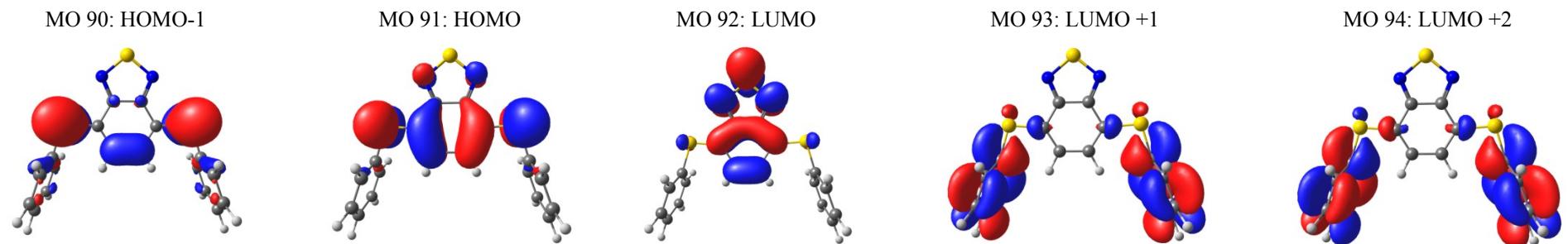
**Figure ESI1.** Molecular orbitals involved in the four vertical transitions of lower energy at the optimized ground state geometry calculated for 3b with CAM-B3LYP and PCM/hexane as solvent effect.



**Figure ESI2.** Molecular orbitals involved in the four vertical transitions of lower energy at the optimized first excited state geometry calculated for 3b with CAM-B3LYP and PCM/hexane as solvent effect.



**Figure ESI3.** Molecular orbitals involved in the four vertical transitions of lower energy at the optimized ground state geometry calculated for 3b with PBE1PBE and PCM/hexane as solvent effect.



**Figure ESI4.** Molecular orbitals involved in the four vertical transitions of lower energy at the optimized first excited state geometry calculated for 3b with PBE1PBE and PCM/hexane as solvent effect.

