

Supporting information:

Mesoporous Ni₃N/NiO composite with core-shell structure for room temperature, selective and sensitive NO₂ gas sensing

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Table 1. Summary of the different sample (precursor - Ni(HCO₃)₂ or Ni(OH)₂) with different Urea:Ni mol ratio, Solvothermal or Hydrothermal temperature, Reaction time and Solution.

Sample	Urea:Ni mol ratio	Solution	Temperature	Reaction time	Products
a	2:1	50 ml H ₂ O and 30 ml	100 °C	18 h	Ni(OH) ₂
b	5:1	50 ml H ₂ O and 30 ml	100 °C	18 h	Ni(OH) ₂
c	2:1	50 ml H ₂ O and 30 ml	160 °C	4 h	Ni(OH) ₂
d	5:1	50 ml H ₂ O and 30 ml	160 °C	4 h	Ni(OH) ₂
e	2:1	80 ml H ₂ O	160 °C	18 h	Ni(OH) ₂
f	5:1	80 ml H ₂ O	160 °C	18 h	Ni(HCO ₃) ₂
g	2:1	50 ml H ₂ O and 30 ml	160 °C	18 h	Ni(OH) ₂
h	5:1	50 ml H ₂ O and 30 ml	160 °C	18 h	Ni(HCO ₃) ₂

Figure S1. Schematic illustration of the homemade gas mixing line for sensor testing system.

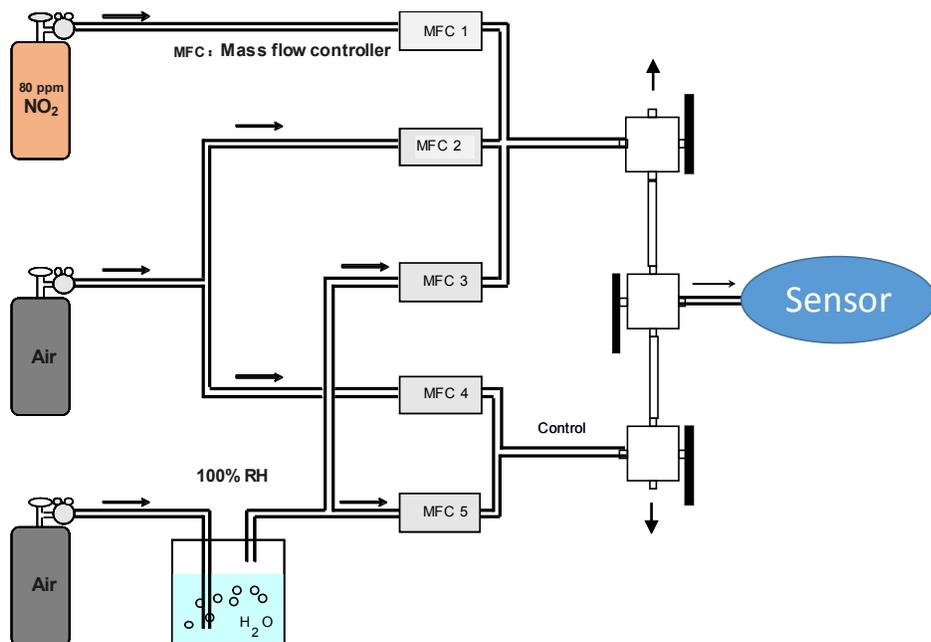


Figure S2. XRD patterns of Ni(OH)₂ (sample a-e and g listed in table1).

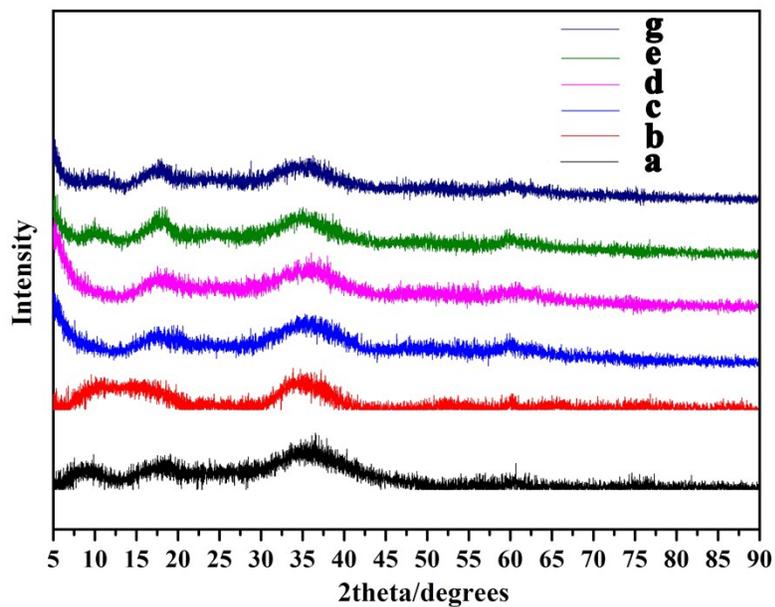


Figure S3. XRD patterns of $\text{Ni}(\text{HCO}_3)_2$ (sample f and h listed in table 1).

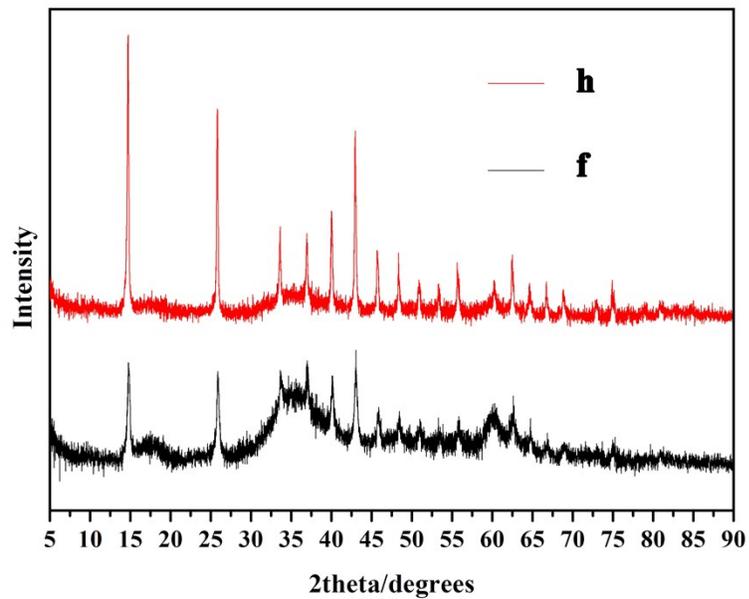


Figure S4. Refinement of PXRD of Ni_3N composite.

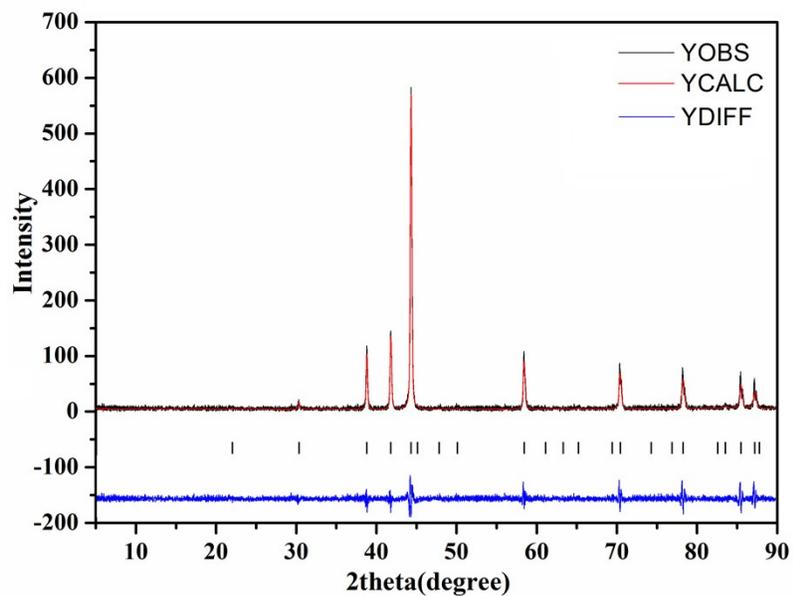


Figure S5. SEM images of $\text{Ni}(\text{OH})_2$ sample a-e and g (corresponded to images a-h) and $\text{Ni}(\text{HCO}_3)_2$ sample f and h (corresponded to images f and h) listed in table1.

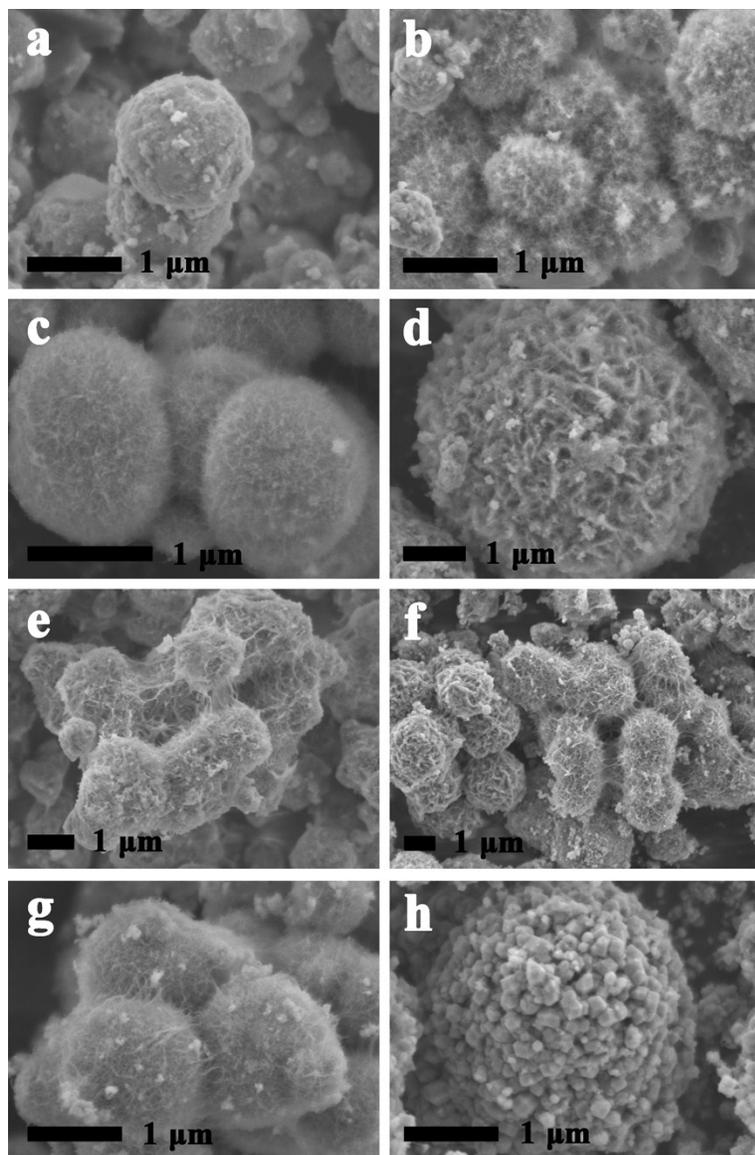


Figure S6. The self-life of Ni₃N/NiO-based sensor for 2 ppm NO₂ sensing at room temperature for two weeks.

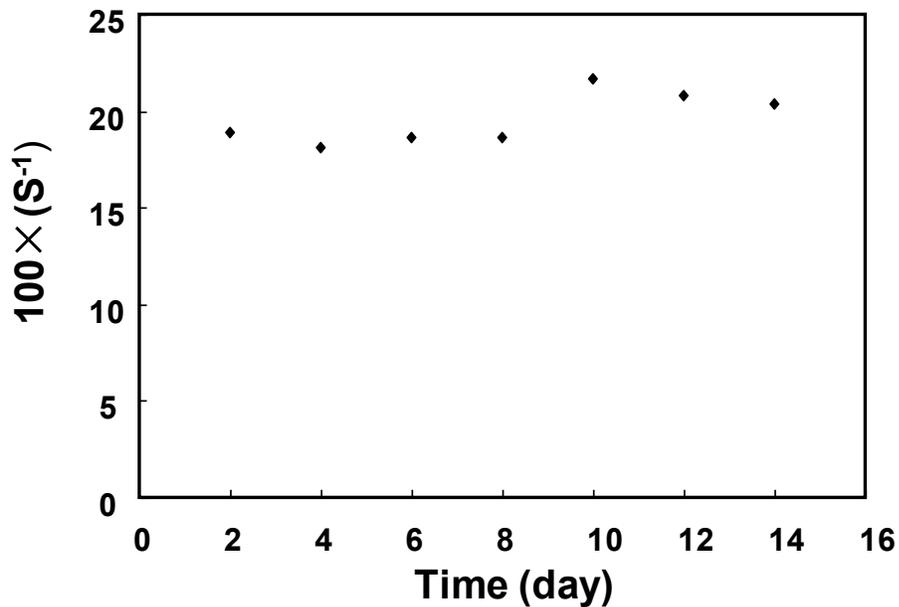


Figure S7. The room temperature sensing properties for Ni₃N, NiO, and Ni₃N/NiO mixture-based sensors to 2 ppm NO₂.

