Supplementary information

## Tertiary butyl hydroquinone as a novel additive for SEI film

## formation in lithium-ion batteries

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Fig. S1. Nyquist plots of the graphite electrodes at 3.0 V in the first discharge process

in the 1.0 M LiPF<sub>6</sub>-EC: DMC: DEC (1:1:1, v/v/v) electrolyte without and with TBHQ.



**Fig. S2.** Nyquist plots of the graphite electrodes at 0.4V in the first discharge process in the 1.0 M LiPF<sub>6</sub>-EC: DMC: DEC (1:1:1, v/v/v) electrolyte without and with TBHQ.



**Fig. S3.** Comparisons of EIS experimental data recorded at 0.7 V and 0.2V during discharge process with the simulation result obtained from the proposed equivalent circuit in the 1.0 M LiPF<sub>6</sub>-EC: DMC: DEC (1:1:1, v/v/v) electrolyte without and with

TBHQ.

## Table S1

Equivalent circuit parameters at 0.7 V in the discharge process in the 1.0 M  $LiPF_{6}$ -EC:

parameters	Without TBHQ		With TBHQ	
	value	Uncertainty(%)	value	Uncertainty (%)
<i>R</i> <sub>s</sub> (Ω)	15.83	0.34459	17.5	0.41333
<i>R</i> <sub>SEI</sub> (Ω)	8.552	9.9809	3.252	0.13218
Q <sub>SEI</sub> — n	2.4653×10 <sup>-4</sup>	14.145	4.3702×10 <sup>-4</sup>	13.655
$Q_{\rm SEI} - Y_0$	0.80492	6.3603	0.75998	4.9021
<i>R</i> <sub>ct</sub> (Ω)	22.4	4.3061	18.3	1.6547
$Q_{dl} - n$	0.013865	5.1693	0.011008	2.6914
$Q_{\rm dl} - Y_0$	0.54074	3.598	0.64036	1.6392
$Q_{\rm D}$ – n	0.18343	1.8816	0.57154	3.2999
$Q_{\rm D} - Y_0$	0.76753	1.4483	0.64975	1.7027

DMC: DEC (1:1:1, v/v/v) electrolyte without and with TBHQ.

## Table S2

Equivalent circuit parameters at 0.2 V in the discharge process in the 1.0 M  $LiPF_{6}$ -EC:

DMC: DEC (1:1:1, v/v/v) electrolyte without and with TBHQ

parameters	Without TBHQ		With TBHQ	
	value	Uncertainty (%)	value	Uncertainty (%)
<i>R</i> <sub>s</sub> (Ω)	17.11	0.27663	18.24	0.17332
$R_{\text{SEI}}(\Omega)$	4.445	12.759	2.216	10.917
Q <sub>SEI</sub> — n	5.8335×10 <sup>-2</sup>	14.842	1.8474×10 <sup>-2</sup>	14.39
$Q_{\rm SEI} - Y_0$	0.49221	13.137	0.5639	13.92
<i>R</i> <sub>ct</sub> (Ω)	4.29	13.017	3.472	10.724
$Q_{dl} - n$	3.1874×10 <sup>-4</sup>	14.833	8.4384×10 <sup>-5</sup>	14.194
$Q_{\rm dl} - Y_0$	0.75824	5.9823	0.92734	7.0895
<i>Q</i> <sub>D</sub> – n	1.576	12.438	1.474	11.13
$Q_{\rm D} - Y_0$	0.62624	6.98686	0.4452	6.427



**Fig. S4.** Variations of  $Q_{SEI}$  -n with the electrode potential in electrolytes without and with TBHQ.



**Fig. S5.** Variations of  $Q_{SEI}$  -Y with the electrode potential in electrolytes without and with TBHQ.



**Fig. S6.** Variations of  $Q_{dl}$  -n with the electrode potential in electrolytes without and with TBHQ.



**Fig. S7.** Variations of  $Q_{dl}$  -Y with the electrode potential in electrolytes without and with TBHQ.



**Fig. S8.** Variations of  $Q_D$  -n with the electrode potential in electrolytes without and with TBHQ.



**Fig. S9.** Variations of  $Q_D$  -Y with the electrode potential in electrolytes without and with TBHQ.