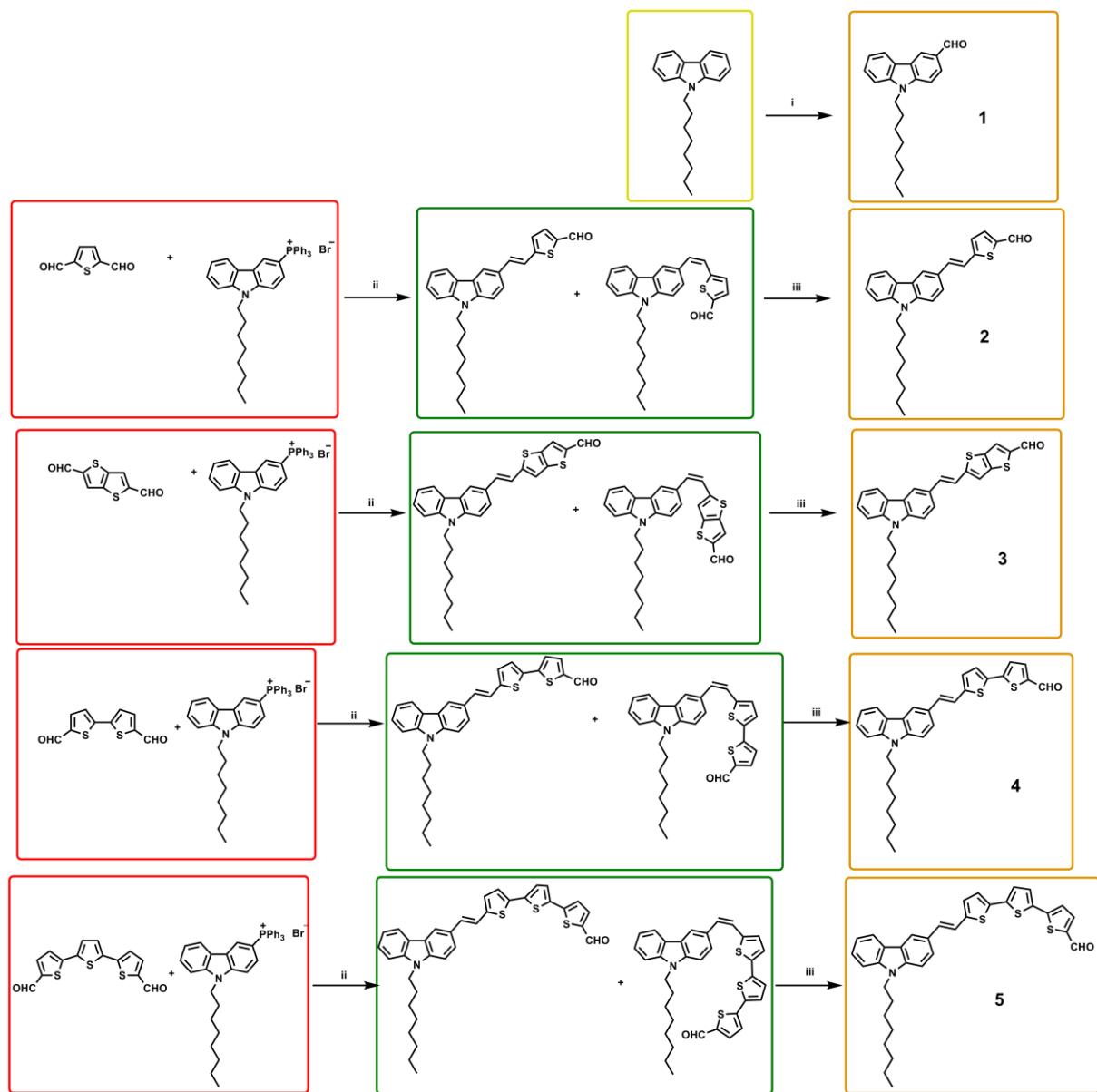


Systematic elongation of thiényl linkers and their effect on optical and electrochemical properties in Carbazole-BODIPY donor-acceptor systems

Alina Brzeczek, Katarzyna Piwowar, Wojciech Domagala, Mikołaj M. Mikołajczyk,
Krzysztof Walczak, Paweł Wagner

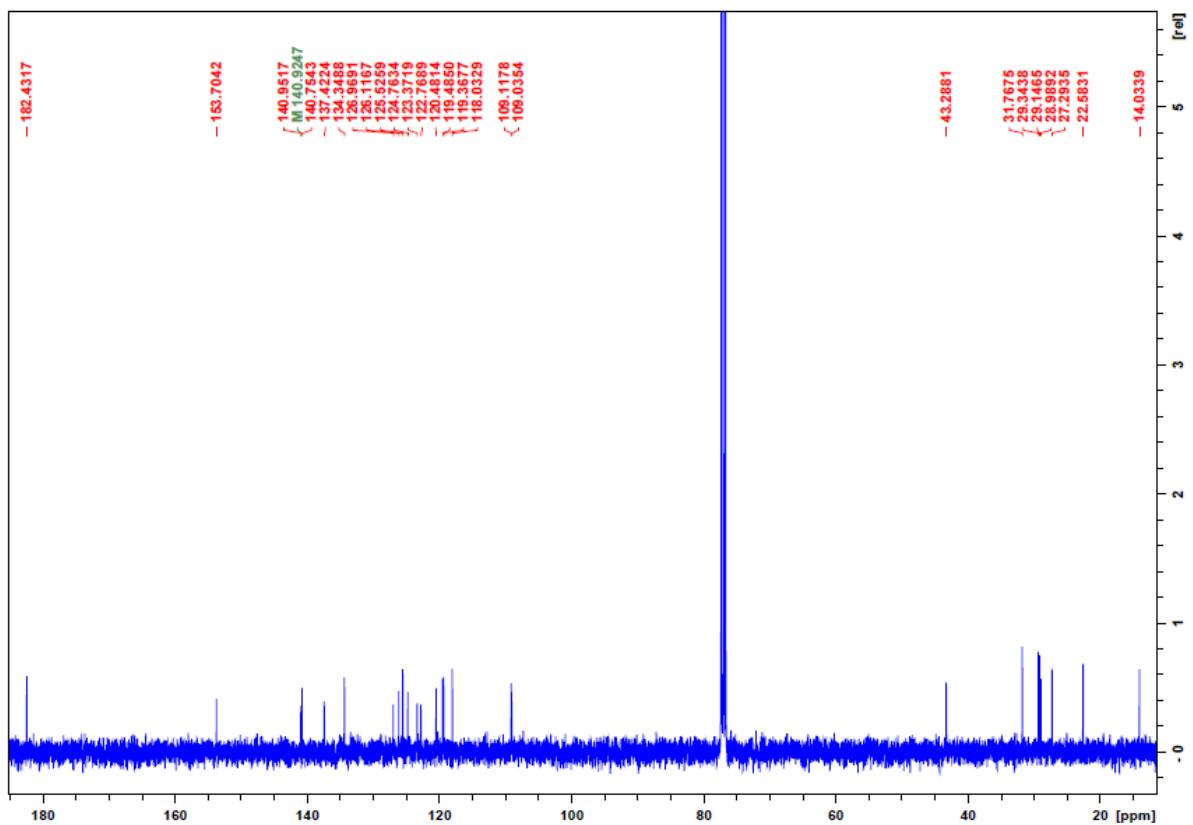
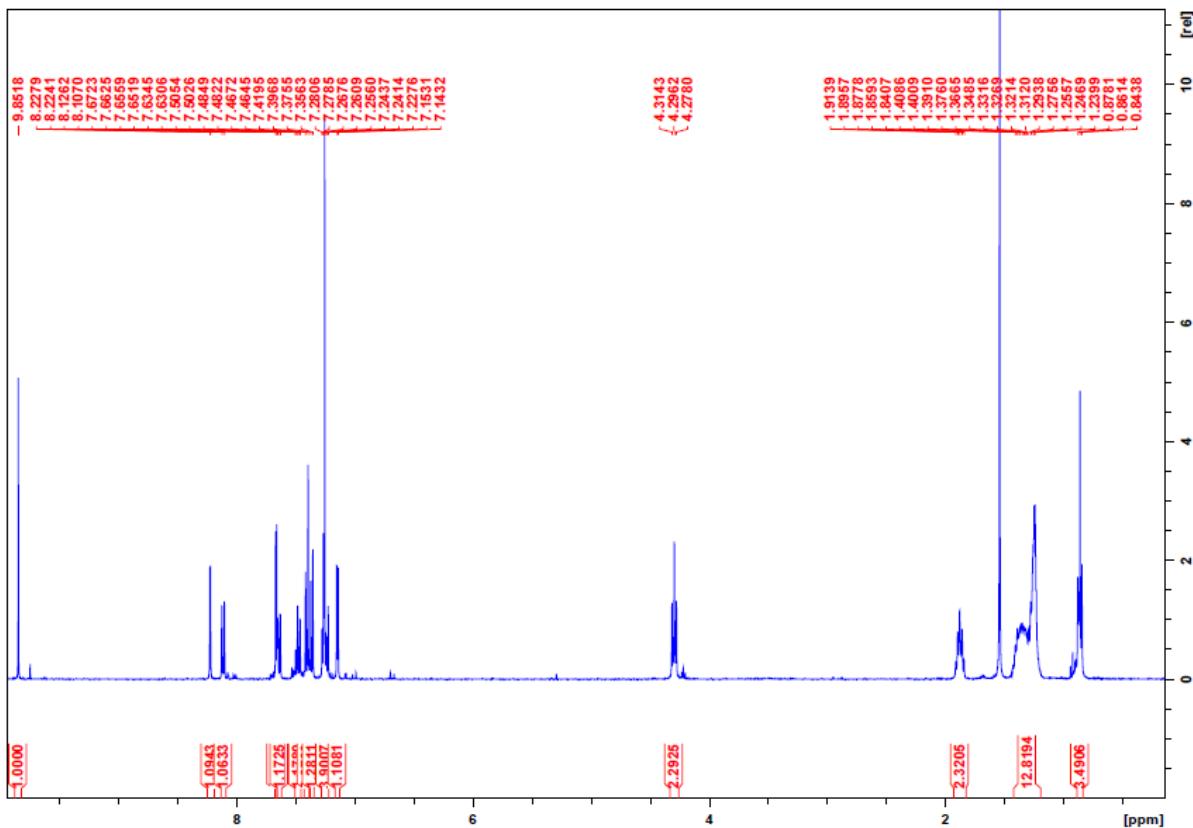
Electronic Supporting Information

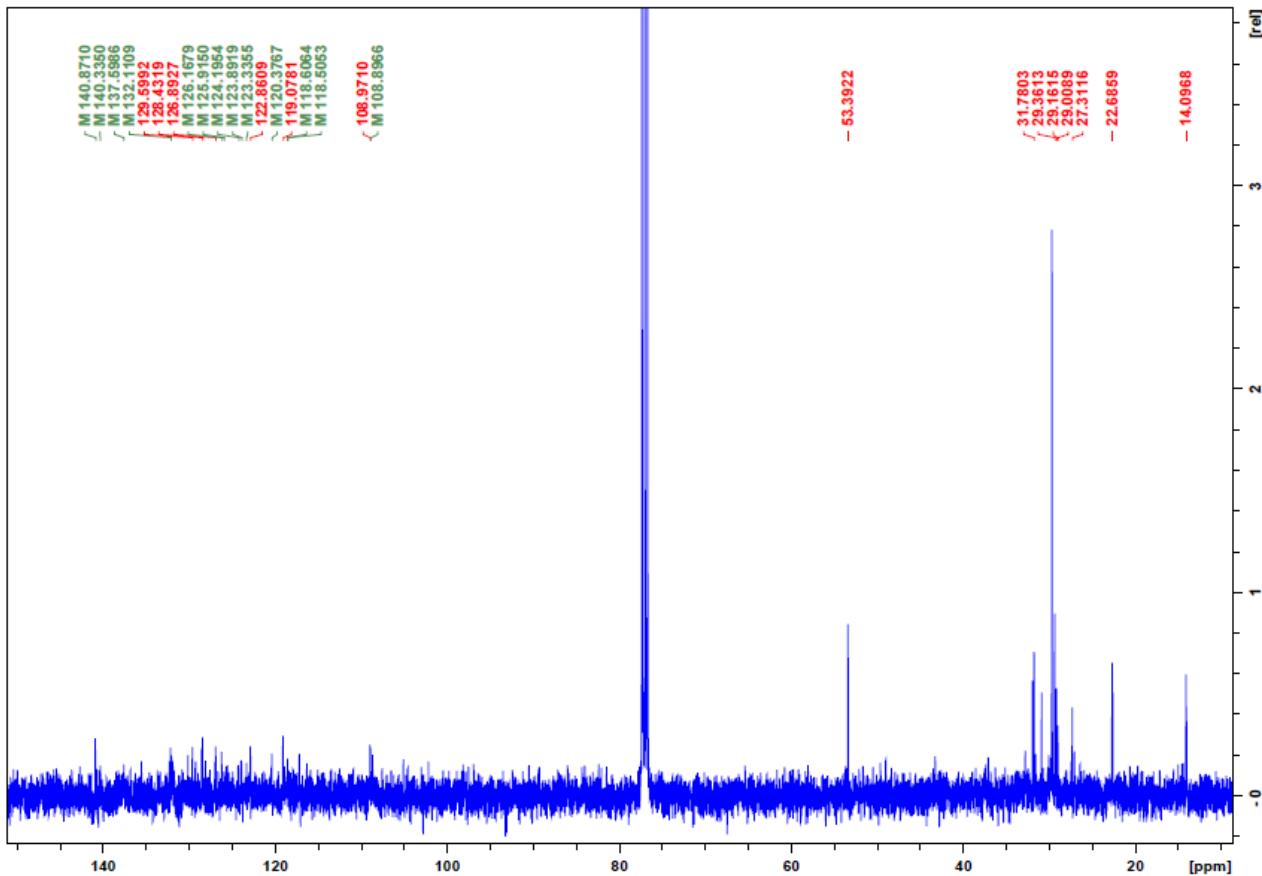
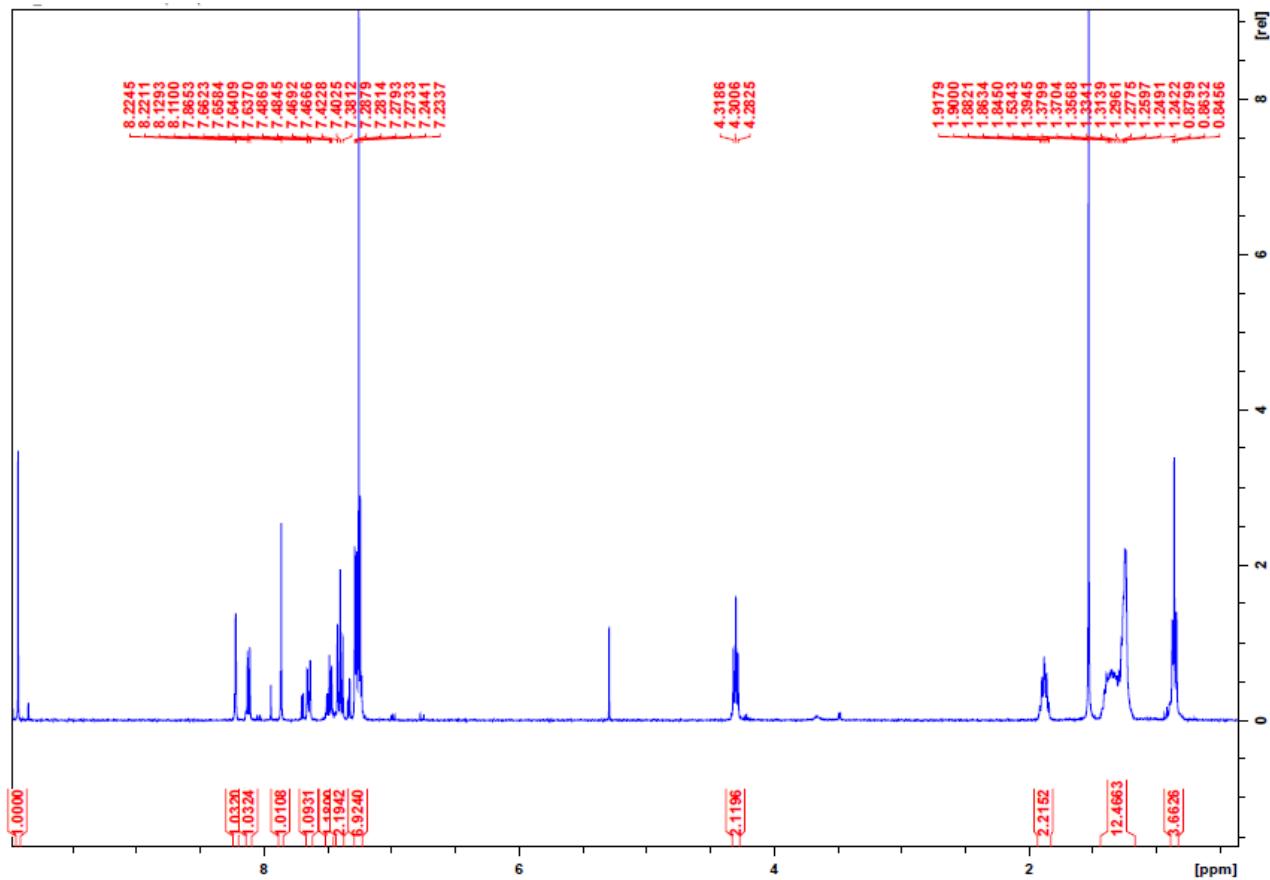


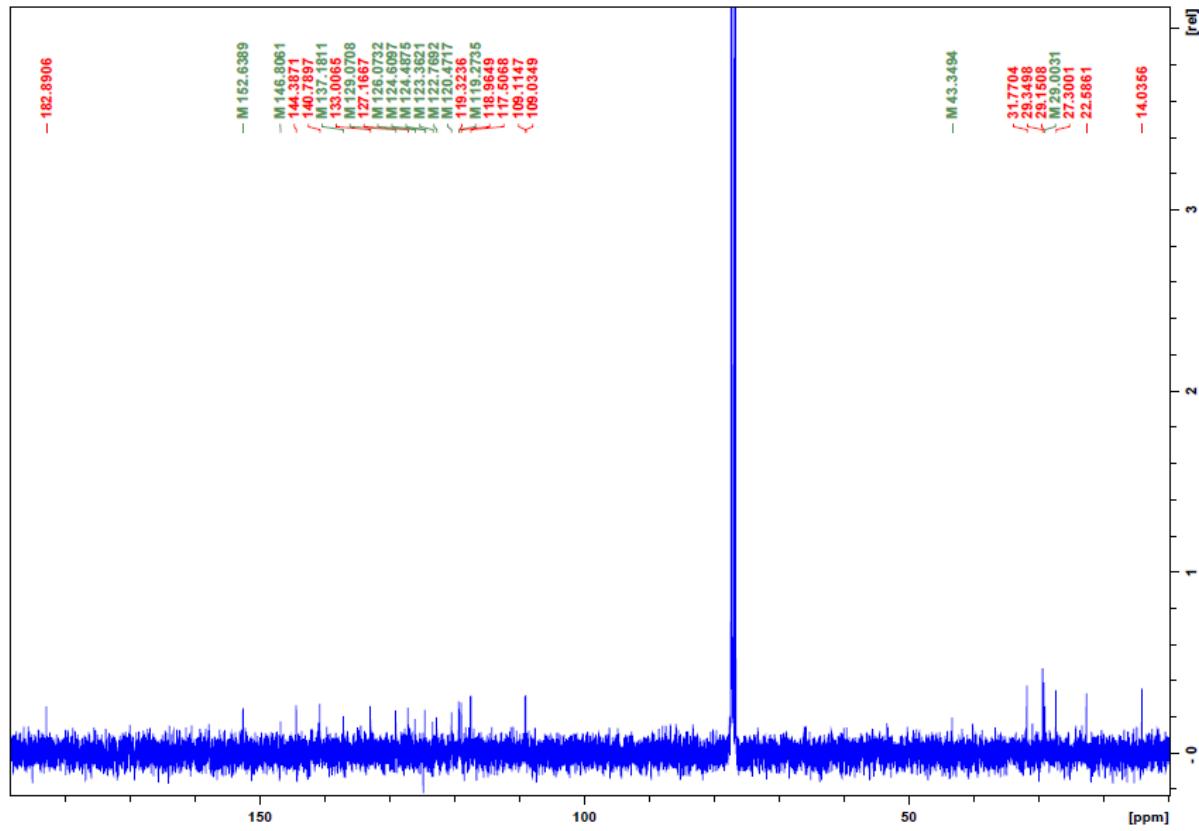
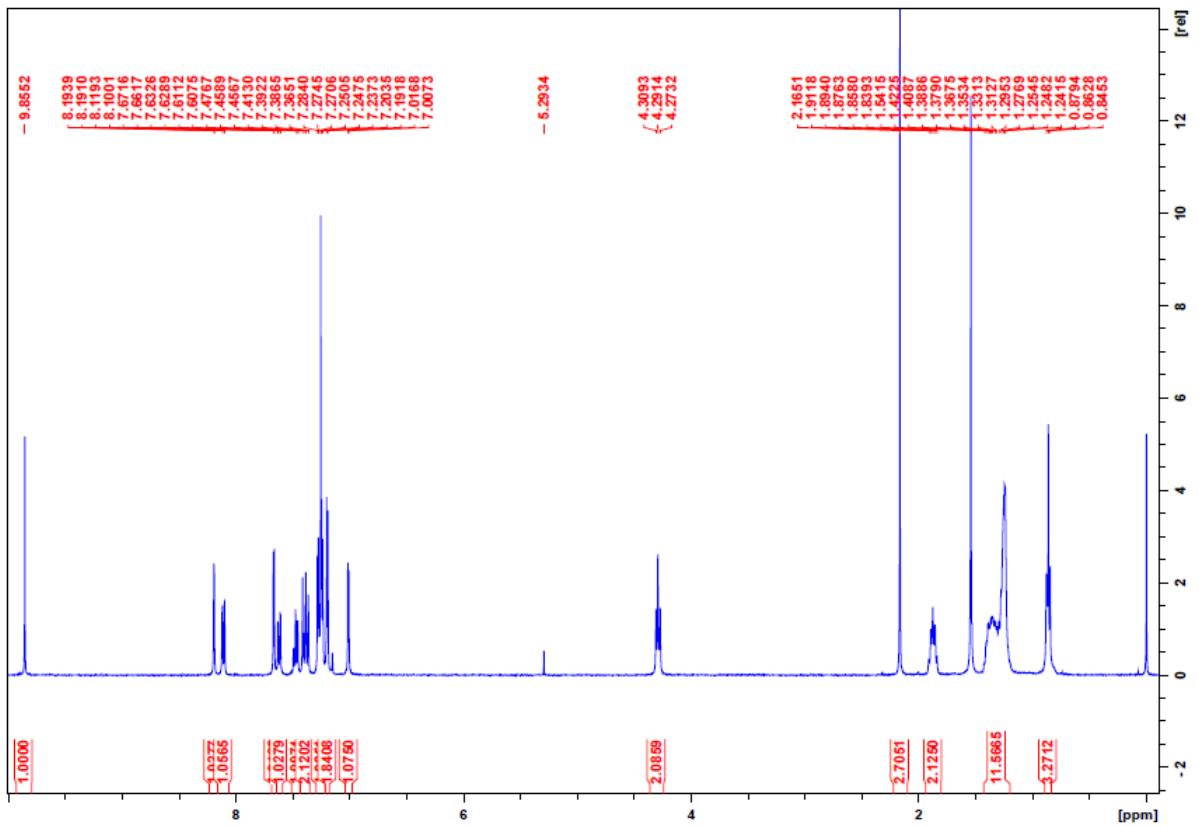
Scheme SI.1. Synthetic route for the preparation of aldehydes 2–5. Reagents and conditions: (i) POCl_3 , DMF, (ii) 18-Crown-6, K_2CO_3 , CH_2Cl_2 , reflux, 1,5h; (iii) TFA, CH_2Cl_2 , H_2O , r.t., 0,5h.

Figure SI.1 ^1H and ^{13}C NMR spectra of aldehydes **2**, **3**, **4**, **5**.

2







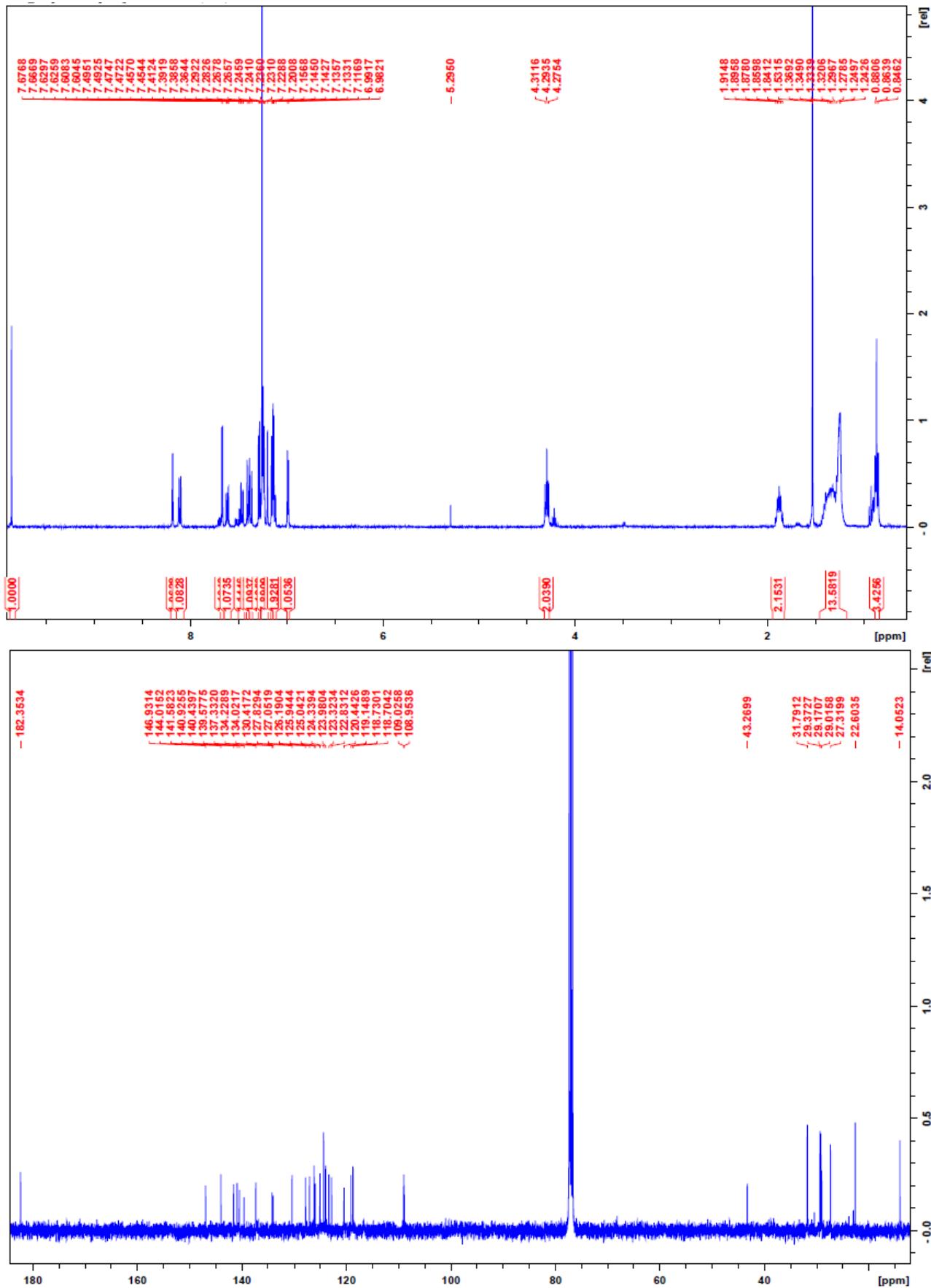
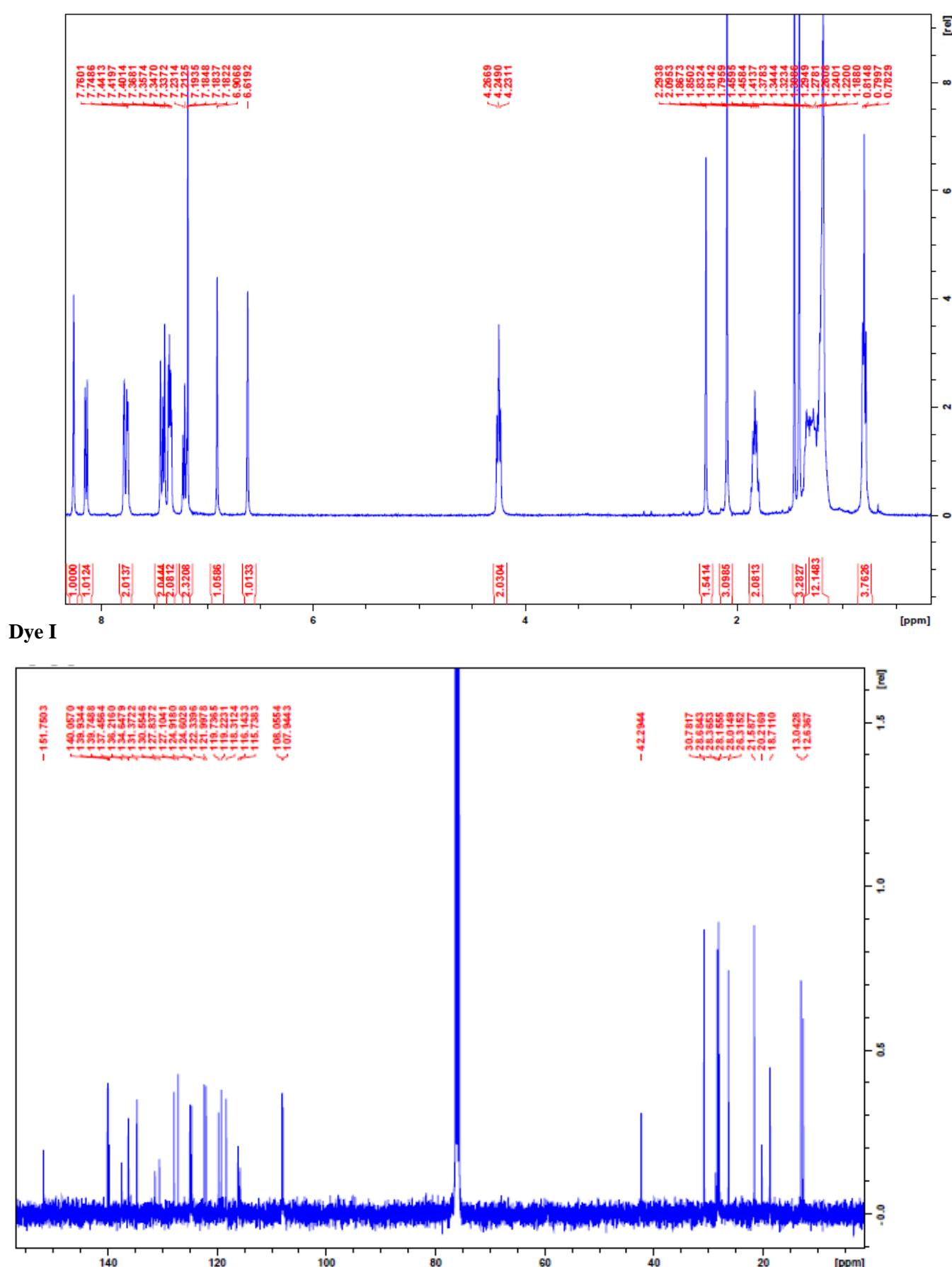
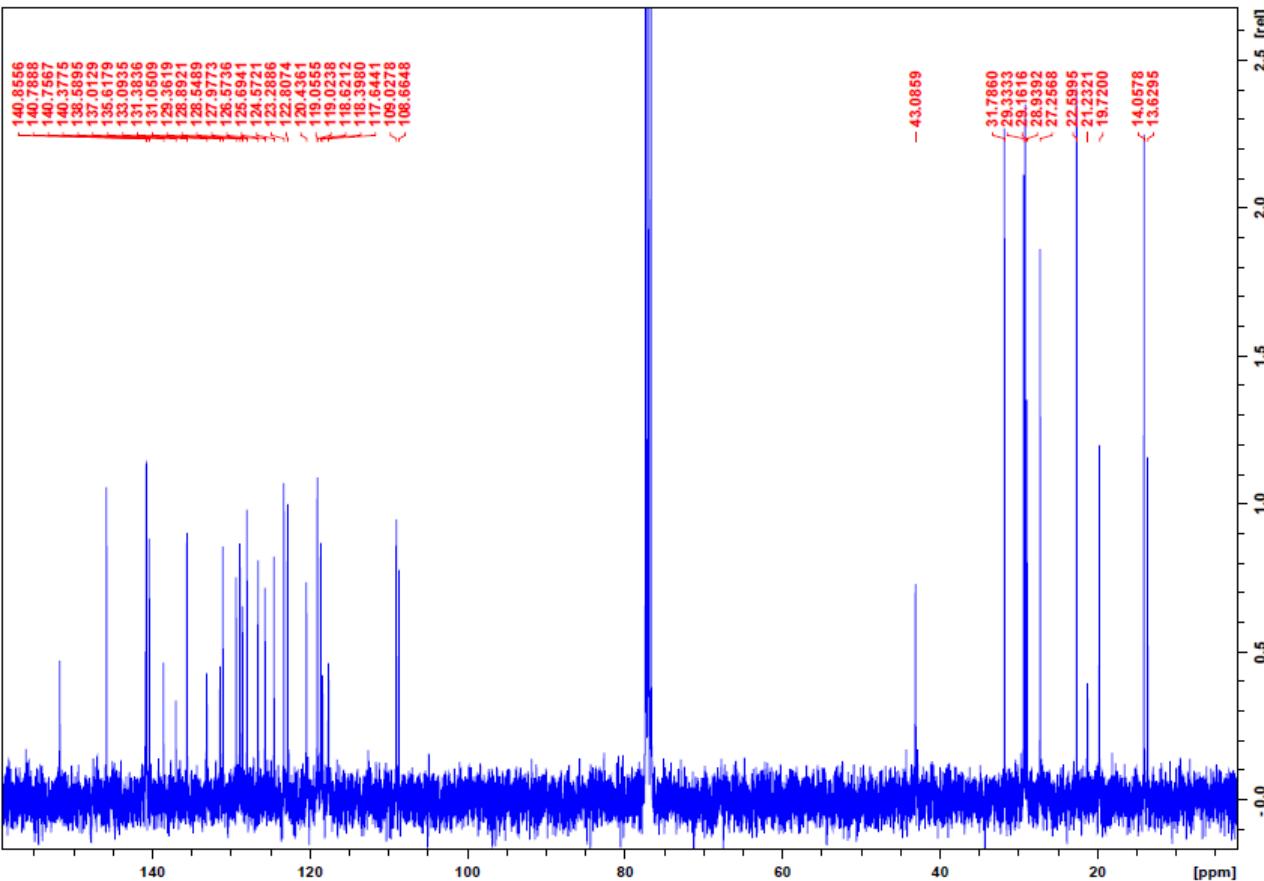
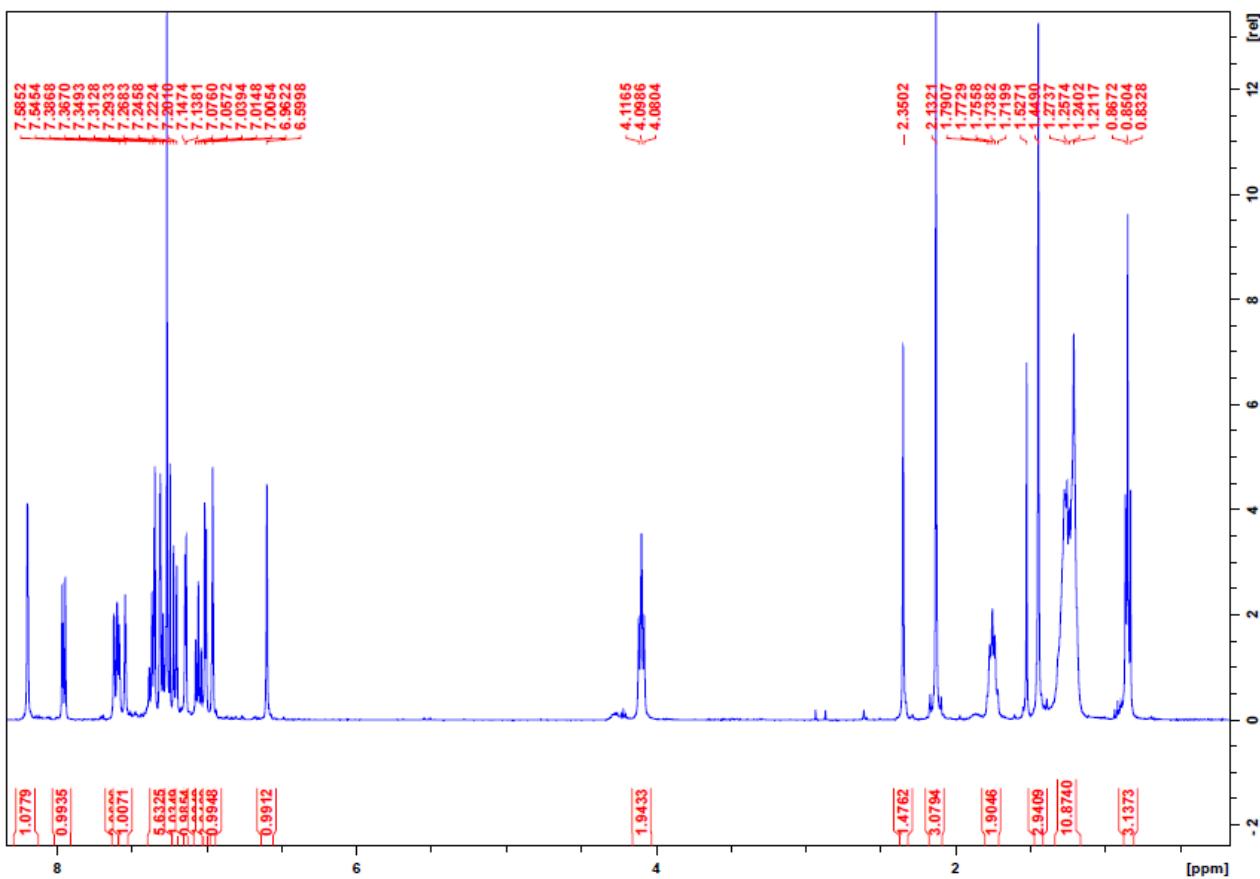


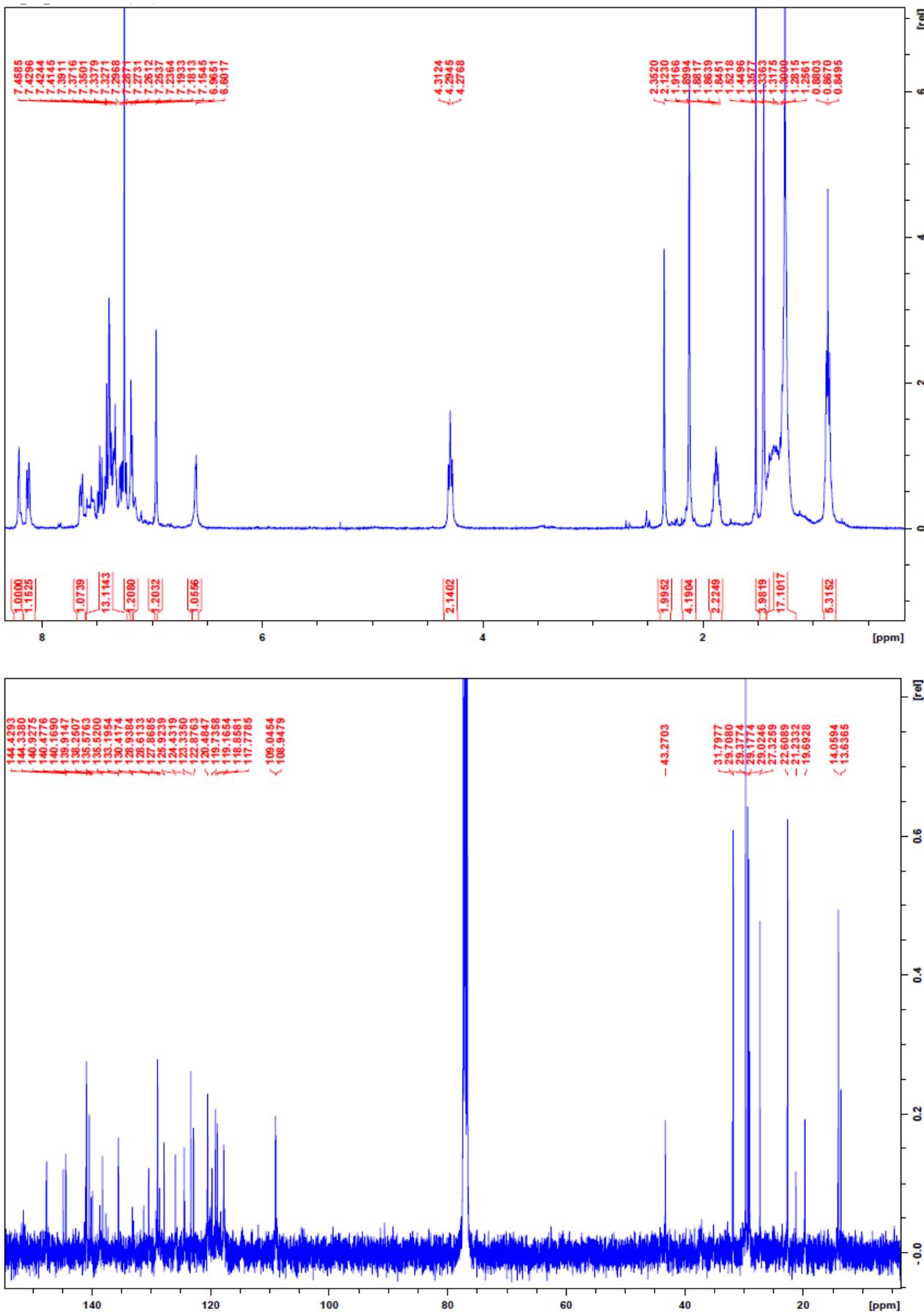
Figure SI.2 ^1H and ^{13}C NMR spectra of dyes **I**, **II**, **III**, **IV**, **V**.



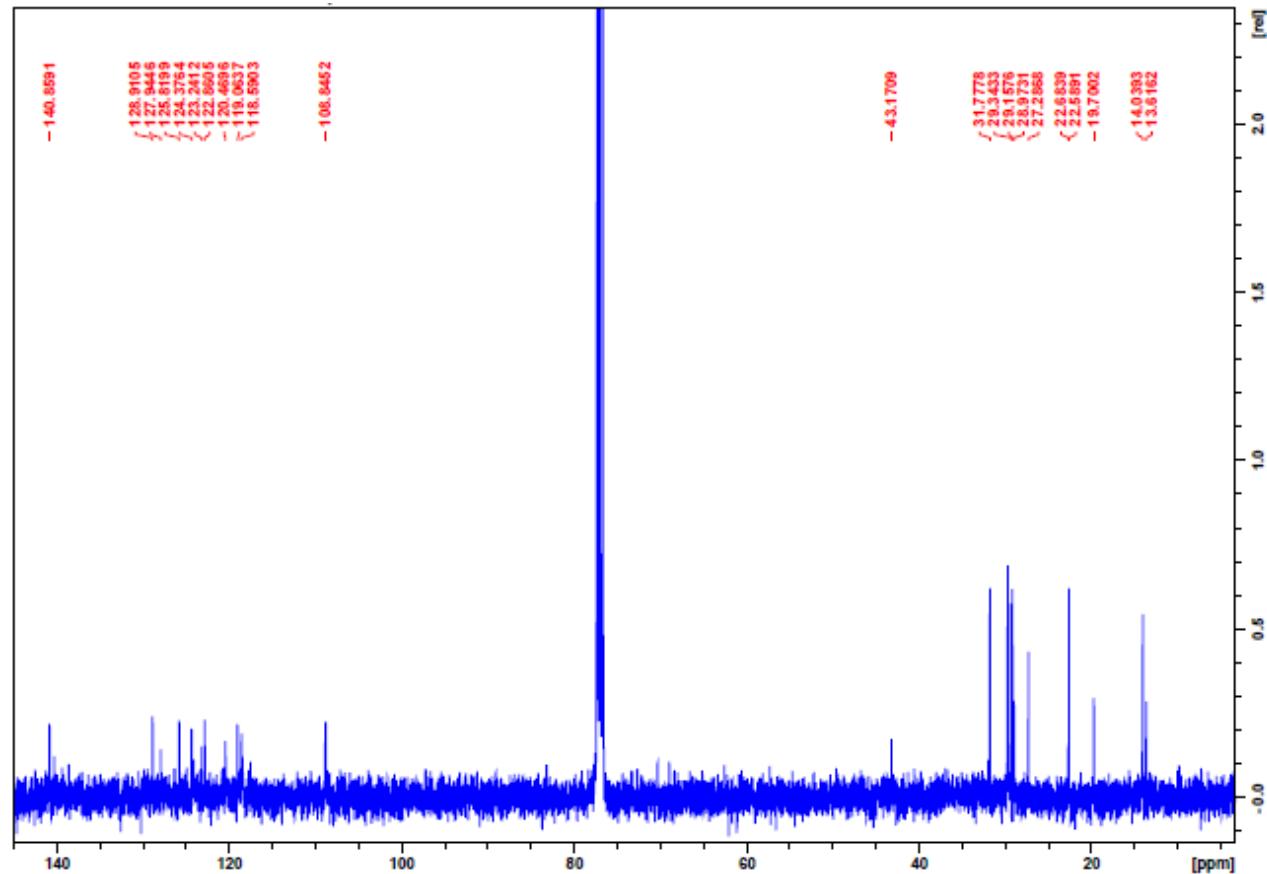
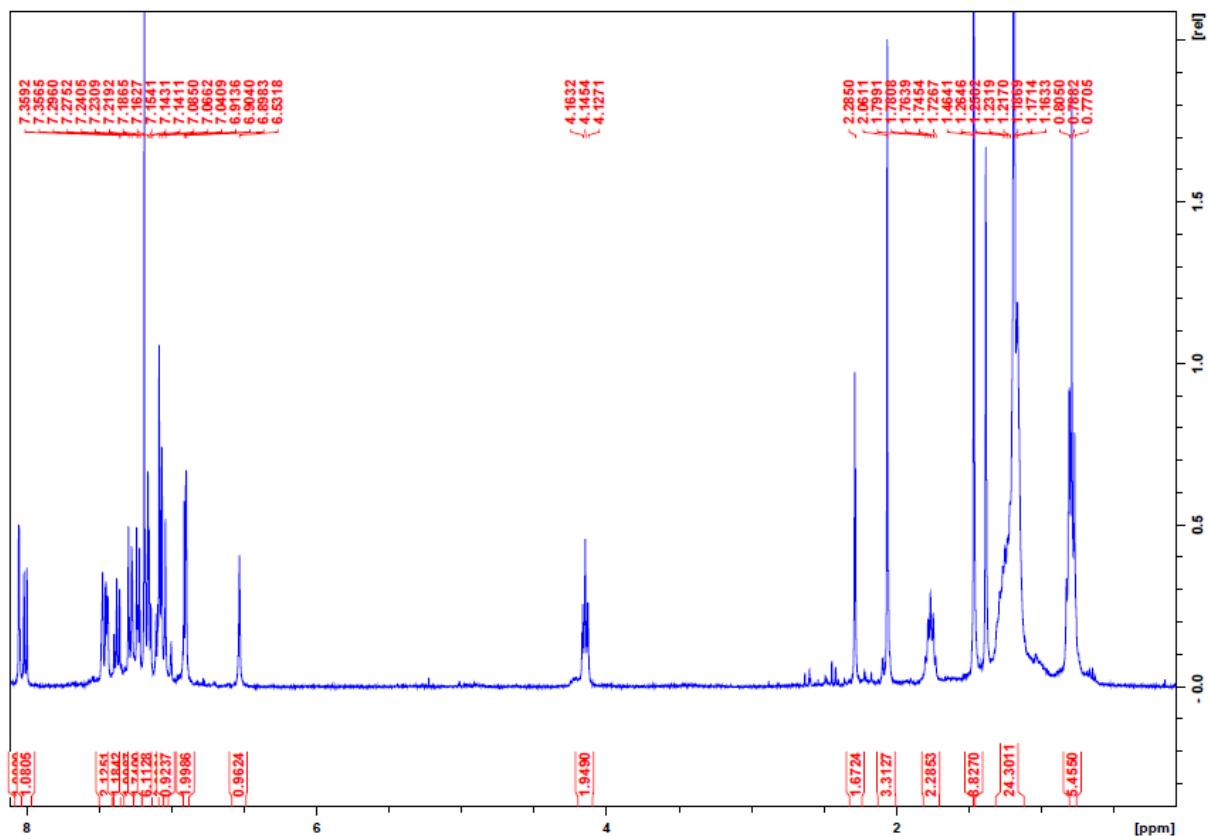
Dye II



Dye III



Dye IV



DYE V

Table SI.1. Formal potentials and peak-to-peak separation parameter of redox pairs of dyes **I** to **V** computed from cyclic voltammograms presented in figure 8. Due to close overlap of redox peaks of dyes **II** to **V** in the positive potentials' segment, meaningful determination of experimental formal potentials of redox pairs involved is unfeasible.

Dye	E_0' [V]	$\Delta E_{pp}'$ [mV]	E_0'' [V]	$\Delta E_{pp}''$ [mV]	E_0''' [V]	$\Delta E_{pp}'''$ [mV]
I	-1.52	88	0.22	77	0.47	77
II	-1.57	143	-	-	-	-
III	-1.39	113	-	-	-	-
IV	-1.31	80	-	-	-	-
V	-1.66	132	-	-	-	-

Table SI.2. Conjugation ratio expressing the number of atoms contributing their atomic orbitals to a given molecular orbital, to the total number of atoms in a molecule, set against ionization potentials and electron affinities of all investigated dyes.

Dye	Conjugation ratio for:		IP [eV]	EA [eV]
	HOMO	LUMO		
I	0.867	0.474	5.24	3.65
II	0.760	0.438	5.28	3.67
III	0.893	0.408	5.22	3.76
IV	0.793	0.379	5.27	3.84
V	0.698	0.515	5.32	3.53

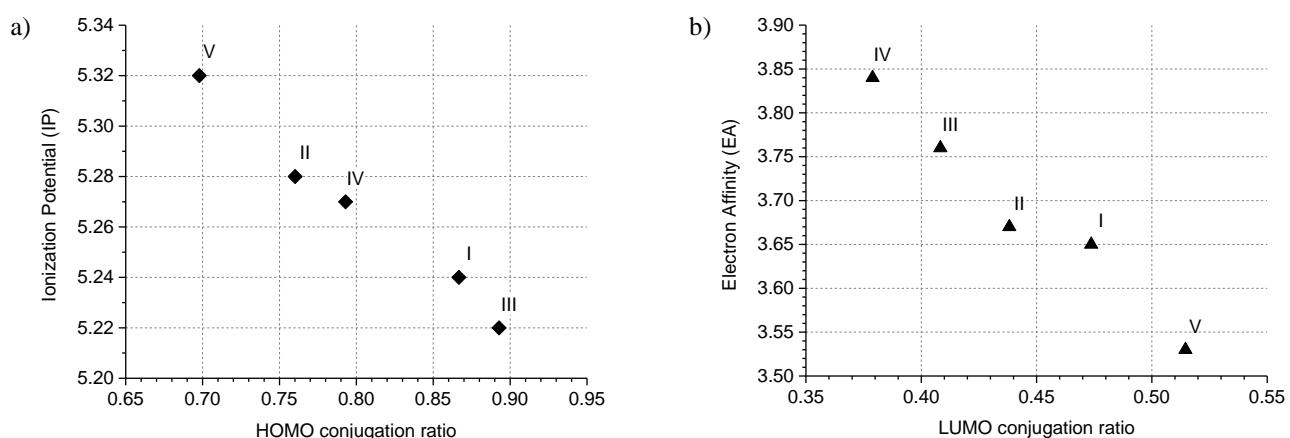


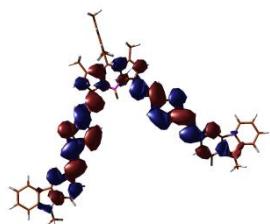
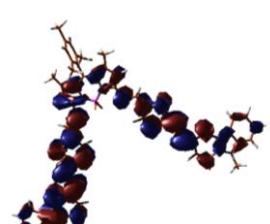
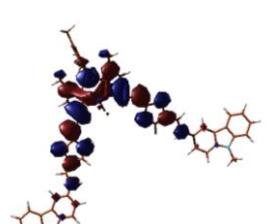
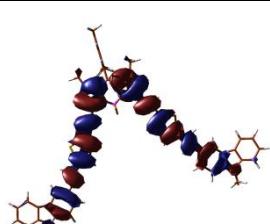
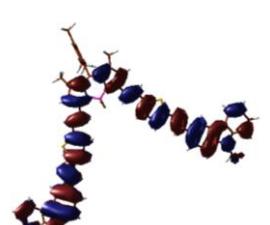
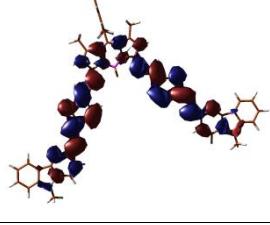
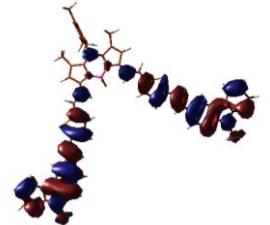
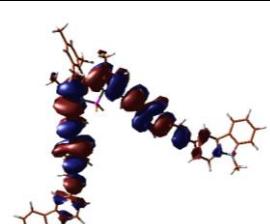
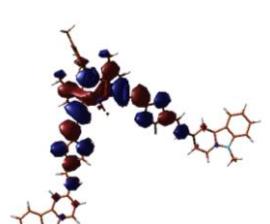
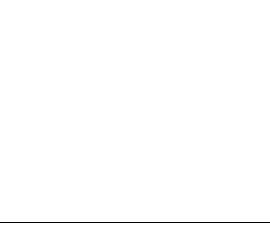
Figure SI.3. Correlation plots of characteristic electron transfer energies with conjugation ratio of given frontier molecular orbitals for dyes **I** through **V**: a) Ionization Potential vs HOMO, b) Electron Affinity vs LUMO.

Figure SI.4 Contours of selected molecular orbitals of neutral and singly charged states of dye: a) **I**, b) **II**, c) **III**, d) **IV**, e) **V**

a) Dye **I**

Radical cation			Neutral molecule			Radical anion		
No.	E [a.u.]	No.	E [a.u.]	No.	E [a.u.]	No.	E [a.u.]	No.
LUMO+1		199		200				
198	-0.15530	198	-0.05472	199	0.01043	200	0.09437	
SOMO	-0.30694					198		
196	-0.32188	197	-0.20403	197	-0.23464	198	-0.05713	
HOMO-1		196						-0.11075

b) Dye **II**

Radical cation			Neutral molecule			Radical anion		
No.		E [a.u.]	No.		E [a.u.]	No.		E [a.u.]
HOMO-1	LUMO+1		255			255		
251	252	-0.31187	252		-0.28963	253		-0.28329
253	254	-0.28329	254		-0.15140	254		-0.15140
252	253	-0.28963	253		-0.19975	253		-0.19975
251	254	-0.31187	254		-0.06417	255		-0.06417

c) Dye III

Radical cation			Neutral molecule			Radical anion		
No.		E [a.u.]	No.		E [a.u.]	No.		E [a.u.]
HOMO-1	LUMO LUMO+1		283			284		0.05144
279		-0.30743			-0.02717			0.04698
280	HOMO SOMO	-0.28525	281		-0.06706	283		-0.11993
281		-0.28077			-0.20144	282		-0.07360
282		-0.15271	282					

d) Dye **IV**

Radical cation			Neutral molecule			Radical anion		
No.	E [a.u.]	No.	No.	E [a.u.]	No.	E [a.u.]		
296	LUMO	296	297	-0.15138	298	-0.02965	0.04298	
295	SOMO	295	296	-0.27412	297	-0.06739	0.04025	
294	HOMO	294	295	-0.27685	295	-0.20055	-0.12023	
293	HOMO-1	293	294	-0.29971	295	-0.21556		

e) Dye V

Radical cation			Neutral molecule			Radical anion		
No.		E [a.u.]	No.		E [a.u.]	No.		E [a.u.]
	LUMO+1							
338		-0.15304	339			340		0.02703
337		-0.26489	338			339		0.02509
336		-0.26651	337			337		-0.07712
335		-0.29100	336			337		-0.12305