## **Supporting Information**

# Reaction control in heterogeneous catalysis using montmorillonite: Switching between acid-catalysed and red-ox processes

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## **Table of Contents**

1. General methods	S1
2. Experimental procedures	S1
3. Characterization data	<b>S</b> 3
4. References	S6

## 1. General methods

All the chemicals were purchased from Aldrich Chemical Co and used without further purification unless stated otherwise. Yields refer to the chromatographically and spectroscopically (<sup>1</sup>H and <sup>13</sup>C) homogeneous materials, unless otherwise stated. All glassware utilized was flame-dried before use. Reactions were monitored by TLC carried out on 0.25 mm Macherey Nagel silica gel plates. Developed TLC plates were visualized under a short-wave UV lamp and by heating plates that were dipped in Ce(SO<sub>4</sub>)<sub>2</sub>. Flash column chromatography (FCC) was performed using flash silica gel (230–400) and employed a solvent polarity correlated with TLC mobility. NMR experiments were conducted on a Varian Unity and Bruker Avance 300 MHz instruments using CDCl<sub>3</sub> (99.9% D) as the solvent, with chemical shifts ( $\delta$ ) referenced to internal standards residual CHCl<sub>3</sub> and CDCl<sub>3</sub>,(7.26 ppm <sup>1</sup>H, 77.0 ppm <sup>13</sup>C) or Me<sub>4</sub>Si as an internal reference (0.00 ppm). Chemical shifts are in parts per million (ppm). Mass spectra were recorded on Jeol JS102 high-resolution mass spectrometer. The measurements of EPR were made with a Jeol JES-TE300 X band fashions spectrometer with a cylindrical cavity in the mode TE<sub>011</sub>. The external calibration of the magnetic field was made with a precision gaussmeter, Jeol ES-FC5.

#### 2. Experimental procedures

## Preparation of modified montmorillonite

The natural montmorillonite sample was the same as we described previously.<sup>1</sup> Modification and characterization of montmorillonite with  $CF_3SO_3H$  was realized according to the method previously reported.<sup>1</sup>

## General procedure for the synthesis of tetrasubstituted furans

A solution of hydroquinone (50% mmol) and modified montomorillonite (20%w) in anhydrous benzene (30 mL) was deoxygenated under an argon atmosphere. Then a solution of  $\alpha$ -hydroxyl ketones **1** (1 mmol) in deoxygenated benzene (5 mL) was added at room temperature. The reaction mixture was stirred for 12 h at reflux under argon atmosphere. After that time, the reaction mixture was filtered and concentrated in vacuum. The resulting residue was purified by flash column chromatography on silica gel.

**S1** 

## General procedure for the synthesis of 1,2-diketones

A solution of  $\alpha$ -hydroxyl ketones **1** (1 mmol) and modified montomorillonite (20%w) in anhydrous benzene (30 mL) was stirred for 6 h at reflux under argon atmosphere. After that time, the reaction mixture was filtered and concentrated in vacuum. The resulting residue was purified by flash column chromatography on silica gel.

#### Reuse of the catalyst

When the organic reactions were complete, the montmorillonite catalyst was removed by filtration, washed with ethyl acetate, dried overnight in an oven at 100 °C, and subjected to a second experimental run with the same substrate. After the completion of four runs, results indicated no significant variation in yield.

#### EPR analysis

A tube was prepared adding 40 mg of the modified montmorillonite and dried at 70 °C under vacuum; the tube was analyzed by EPR (Figure **S1A**). Then 2 mL of anhydrous benzene were added and the sample was analyzed (Figure **S1B**). The last experiment was carried out after adding benzoine to the tube, scanning every 5 minutes (Figure **S1C**).

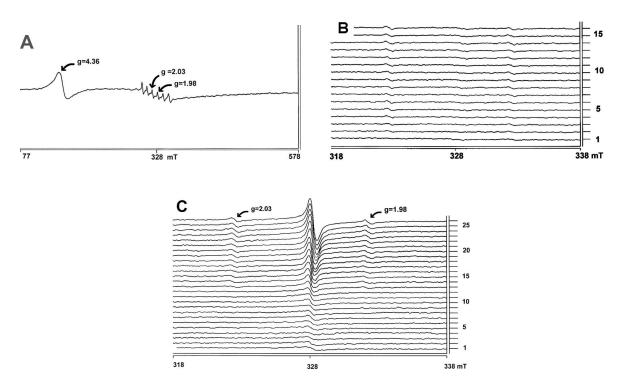


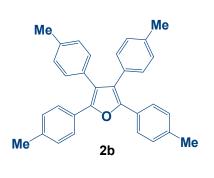
Figure S1

#### 3. Characterization data



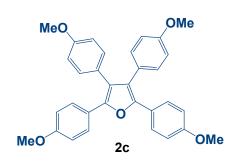
2,3,4,5-tetraphenylfuran **2a**:<sup>2</sup> Following the general procedure for the synthesis of tetrasubstituted furans, the reaction was carried out starting from benzoin 1a (212 mg, 1 mmol), hydroquinone (55 mg, 0.5 mmol) and modified montmorillonite (42 mg), to give 111 mg of a white solid (60%, mp 173-174 °C); <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 7.51-7.49 (m, 4H), 7.26-7.20 (m, 12H), 7.16-12.13 (m, 4H); <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 147.7, 133.1, 130.9, 130.4, 128.5,

128.3, 127.3, 127.1, 125.8, 125.1. HRMS (FAB) calcd for C<sub>28</sub>H<sub>20</sub>O 372.1514, found 372.1518.



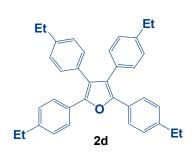
2,3,4,5-tetra-p-tolylfuran 2b:<sup>2</sup> Following the general procedure for the synthesis of tetrasubstituted furans, the reaction was carried out starting from benzoin 1b (240 mg, 1 mmol), hydroquinone (55 mg, 0.5 mmol) and modified montmorillonite (48 mg), to give 133 mg of a white solid (62%, mp 185-186 °C); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$ 7.37 (d, J = 8.4 Hz, 4H), 7.04 (d, J = 8.0 Hz, 4H), 7.02 (s,

8H), 2.36 (s, 12H); <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 147.5, 136.9, 136.5, 130.3, 130.2, 129.0, 128.7, 128.4, 125.8, 124.3, 29.7, 21.3. HRMS (FAB) calcd for C<sub>32</sub>H<sub>28</sub>O 428.2141, found 428.2139.



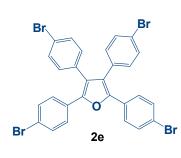
2,3,4,5-tetrakis(4-methoxyphenyl)furan 2c:<sup>3</sup> Following the general procedure for the synthesis of tetrasubstituted furans, the reaction was carried out starting from benzoin 1c (272 mg, 1 mmol), hydroquinone (55 mg, 0.5 mmol) and modified montmorillonite (54 mg), to give 160 mg of a white solid (65%, mp 207-209 °C); <sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 7.42 (d,

J = 8.8 Hz, 4 H), 7.04 (d, J = 8.8 Hz, 4 H), 6.79 (d, J = 8.2 Hz, 4 H), 6.77 (d, J = 8.4 Hz, 4 H), 3.78 (s, 12H); <sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 147.7, 133.1, 130.9, 130.4, 128.5, 128.3, 127.3, 127.1, 125.8, 125.1, 55.1, 55.0. HRMS (FAB) calcd for C<sub>32</sub>H<sub>28</sub>O<sub>5</sub> 492.1937, found 492.1941.



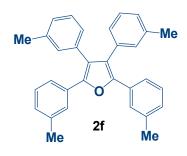
**2,3,4,5-tetrakis(4-ethylphenyl)furan 2c**:<sup>4</sup> Following the general procedure for the synthesis of tetrasubstituted furans, the reaction was carried out starting from benzoin **1d** (268 mg, 1 mmol), hydroquinone (55 mg, 0.5 mmol) and modified montmorillonite (52 mg), to give 157 mg of a white solid (65%, mp 194-196 °C); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  7.35 (d, *J* = 8.2

Hz, 4 H), 7.03-6.89 (m, 12H), 2.58-2.51 (m, 8H), 1.16 (t, J = 7.6 Hz, 6H), 1.15 (t, J = 7.6 Hz, 6H); <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  145.5, 143.2, 142.7, 130.6, 130.3, 128.7, 127.8, 127.7, 125.8, 124.4, 28.6, 28.5, 15.3, 15.1. HRMS (FAB) calcd for C<sub>36</sub>H<sub>36</sub>O 484.2766, found 484.2769.



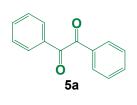
**2,3,4,5-tetrakis(4-bromophenyl)furan 2e**:<sup>2</sup> Following the general procedure for the synthesis of tetrasubstituted furans, the reaction was carried out starting from benzoin **1e** (368 mg, 1 mmol), hydroquinone (55 mg, 0.5 mmol) and modified montmorillonite (72 mg), to give 205 mg of a white solid (60%, mp 165-167 °C); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  7.42 (d, *J* = 8.0 Hz, 8 H),

7.32 (d, J = 8.0 Hz, 4 H), 6.98 (d, J = 8.0 Hz, 4 H); <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  147.3, 131.9, 131.7, 131.1, 128.9, 127.3, 121.9, 121.8 HRMS (FAB) calcd for C<sub>28</sub>H<sub>16</sub>BrO 683.7935, found 683.7939.



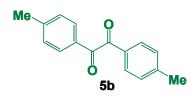
**2,3,4,5-tetra-***m***-tolylfuran**:<sup>2</sup> Following the general procedure for the synthesis of tetrasubstituted furans, the reaction was carried out starting from benzoin **1f** (240 mg, 1 mmol), hydroquinone (55 mg, 0.5 mmol) and modified montmorillonite (48 mg), to give 128 mg of a white solid (60%, mp 100-102 °C); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  7.41 (s, 2 H), 7.25 (d, *J* = 5.6 Hz, 2 H), 7.12 (t, *J* = 7.6 Hz, 4

H), 7.04-6.94 (m, 8 H), 2.30 (s, 6 H), 2.23 (s, 6 H);  ${}^{13}$ C NMR (CDCl<sub>3</sub>):  $\delta$  147.6, 137.9, 137.7, 133.2, 131.1, 128.1, 127.8, 127.5, 126.4, 125.1, 123.1, 21.5, 21.3. HRMS (FAB) calcd for C<sub>32</sub>H<sub>28</sub>rO 428.2140, found 428.2145.



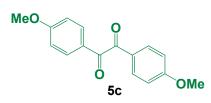
**Benzil 5a:**<sup>5</sup> Following the general procedure for the synthesis of 1,2-diketones, the reaction was carried out starting from benzoin **1a** (212 mg, 1 mmol) and modified montmorillonite (42 mg), to give 199 mg of a white solid (95%, mp 92-94 °C); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  8.01-7.94 (m, 4H), 7.70-7.64 (m, 2H), 7.56-7.50 (m, 4H); <sup>13</sup>C NMR

(CDCl<sub>3</sub>):  $\delta$  194.6, 134.9, 133.0, 129.9, 129.0. HRMS (FAB) calcd for C<sub>14</sub>H<sub>10</sub>O<sub>2</sub> 210.0681, found 210.0682.



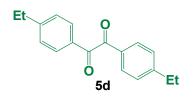
**Benzil 5b:**<sup>5</sup> Following the general procedure for the synthesis of 1,2-diketones, the reaction was carried out starting from benzoin **1b** (240 mg, 1 mmol) and modified montmorillonite (48 mg), to give 226 mg of a white solid (95%, mp 102-104 °C); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  7.68 (d, *J* = 8 Hz,

4H), 7.28 (d, *J* = 8 Hz, 4H), 2.40 (s, 6H); <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  194.6, 146.2, 130.9, 130.1, 129.8, 22.0 HRMS (FAB) calcd for C<sub>16</sub>H<sub>14</sub>O<sub>2</sub> 238.0994, found 238.0996.



**Benzil 5c:**<sup>5</sup> Following the general procedure for the synthesis of 1,2-diketones, the reaction was carried out starting from benzoin **1c** (272 mg, 1 mmol) and modified montmorillonite (54 mg), to give 256 mg of a white solid (95%, mp 131-133 °C); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$ 

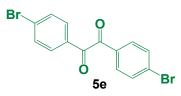
7.69 (d, J = 8 Hz, 4H), 6.92 (d, J = 8 Hz, 4H), 3.88 (s, 6H); <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  193.6, 165.0, 132.5, 126.5, 114.4, 55.8 HRMS (FAB) calcd for C<sub>16</sub>H<sub>14</sub>O<sub>4</sub> 270.0892, found 270.0893.



**Benzil 5d:**<sup>5</sup> Following the general procedure for the synthesis of 1,2-diketones, the reaction was carried out starting from benzoin **1d** (268 mg, 1 mmol) and modified montmorillonite (52 mg), to give 253 mg of a white solid (95%, mp 122-124 °C); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$ 

7.80 (d, J = 8 Hz, 4H), 7.23 (d, J = 8 Hz, 4H), 2.62 (q, J = 7.6 Hz, 4H), 1.15 (t, J = 7.6 Hz, 6h); <sup>13</sup>C

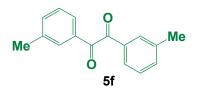
NMR (CDCl<sub>3</sub>):  $\delta$  194.7, 152.3, 130.9, 130.2, 128.6, 29.3, 15.2. HRMS (FAB) calcd for C<sub>18</sub>H<sub>18</sub>O<sub>2</sub> 266.1307, found 266.1309.



**Benzil 5e:**<sup>5</sup> Following the general procedure for the synthesis of 1,2-diketones, the reaction was carried out starting from benzoin **1e** (368 mg, 1 mmol) and modified montmorillonite (72 mg), to give 328 mg of a white solid (90%, mp 223-225 °C); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  7.83 (d, *J* = 8 Hz, 4H), 7.67 (d, *J* = 8 Hz, 4H); <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  192.7, 132.7,

131.7, 131.4, 130.9. HRMS (FAB) calcd for C<sub>14</sub>H<sub>8</sub>Br<sub>2</sub>O<sub>2</sub> 365.8891, found 365.8893.

Benzil 5f:<sup>5</sup> Following the general procedure for the synthesis of 1,2-diketones, the reaction



was carried out starting from benzoin **1f** (240 mg, 1 mmol) and modified montmorillonite (48 mg), to give 226 mg of a white solid (95%, mp 125-127 °C); <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  7.82-7.73 (m, 4H), 7.47 (d, *J* = 7.6 Hz, 2H), 7.39 (t, *J* = 7.6 Hz, 2H), 2.40 (s, 6H); <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  195.1,

139.1, 135.8, 133.2, 130.4, 129.0, 127.4, 21.4. HRMS (FAB) calcd for  $C_{16}H_{14}O_2$  238.0994, found 238.0998.

#### 4. References

 <sup>&</sup>lt;sup>1</sup> (a) G. Granados-Oliveros, V.; Gómez-Vidales, A.; Nieto-Camacho, J. A.; Morales-Serna, J. Cárdenas, and M. Salmón. RSC Adv., 2013, 3, 937; (b) R. Y. M. Vargas, H. I. Beltrán, V. E. Labastida, L. C. Lónaz and M. Salmón, L. Mater. Bas., 2007, 22, 788.

E. Labastida, L. C. López and M. Salmón, *J. Mater. Res.*, 2007, **22**, 788.

<sup>&</sup>lt;sup>2</sup>(*a*) L. Wang, Jun Li, Y. Lv, G. Zhao and S. Gao, *Synlett*, 2012, **23**, 1074, (*b*) Y. Wen, S. Zhu, H. Jiang, A. Wang and Z. Chen, *Synlett*, 2011, 1023; (*c*) A. Wang, H. Jiang and Q. Xu, *Synlett*, 2009, 929.

<sup>&</sup>lt;sup>3</sup> M. Nakano, H. Tsurugi, T. Satoh and M. Miura, Org. Lett., 2008, **10**, 1851.

<sup>&</sup>lt;sup>4</sup> M. Hussain, R. A. Khera, N. T. Hung and P. Langer, Org. Biomol. Chem., 2011, 9, 370.

 <sup>&</sup>lt;sup>5</sup> (a) Z., Zhishuo and Z. Xiaosong, Chin. J. Chem. 2012, **30**, 1683; (b) J.-D. Lou, Y.-C. Ma, N. Vatanian, Q. Wang and C. Zhang, Synth. React. Inorg., Met.-Org., Nano-Met. Chem., 2010, **40**, 157; (c) J.-D. Lou a, L. Li, N. Vatanian, X L. Lu and X. Yu, Synth. React. Inorg., Met.-Org., Nano-Met. Chem., 2008, **38**, 373.