

Modulating the Mixed Micellization of CTAB and an Ionic Liquid 1-Hexadecyl-3-methylimidazolium Bromide via Varying Physical States of Ionic Liquid

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Supporting Information

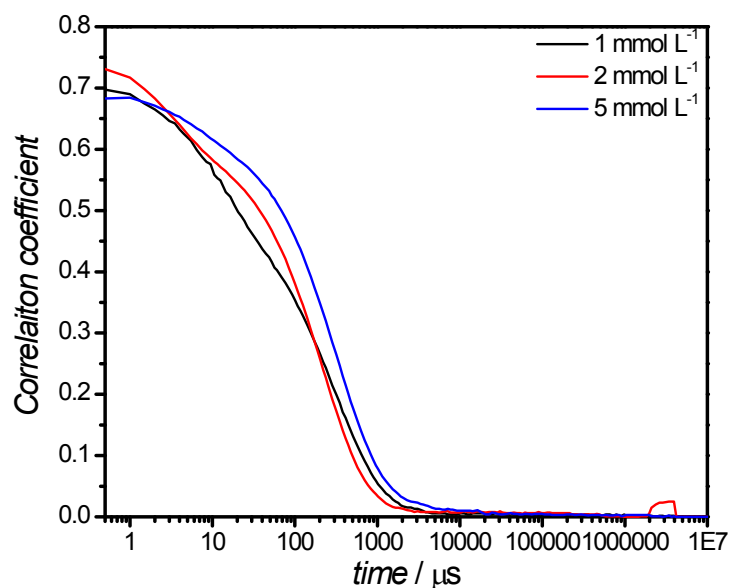


Figure S1. Correlogram obtained from DLS measurements for aqueous solutions of [C16mim][Br] in the absence of CTAB at different concentrations.

Table S1. A comparison of ^1H Spin-lattice relaxation Time (T_1) for different protons of IL (H_k and H_i) and CTAB (H_e) along with alkyl chain protons at different concentrations of IL, CTAB and IL-CTAB mixed systems below and above cmc at 298.15 K as per Figure 6.

Protons	^1H Spin-lattice relaxation Time (T_1)							
	CTAB		IL (1 mmol L $^{-1}$)	IL-CTAB (1 mmol L $^{-1}$)	IL / (2 mmol L $^{-1}$)	IL-CTAB 2 mmol L $^{-1}$	IL (5 mmol L $^{-1}$)	IL-CTAB (5 mmol L $^{-1}$)
	Below <i>cmc</i>	Above <i>cmc</i>						
H_k			2.23 \pm 0.35	2.09 \pm 0.29	1.90 \pm 0.24	1.70 \pm 0.25	1.810 \pm 0.29	1.63 \pm 0.13
H_i			2.12 \pm 0.30	2.04 \pm 0.32	2.09 \pm 0.29	1.90 \pm 0.28	1.94 \pm 0.22	1.95 \pm 0.22
H_e	0.89	0.61		0.18 \pm 0.03		0.34 \pm 0.01		0.50 \pm 0.07
H_b	0.88	0.63	0.78 \pm 0.10	0.514 \pm 0.11	0.71 \pm 0.01	0.51 \pm 0.14	0.67 \pm 0.04	0.57 \pm 0.05
H_a			1.25 \pm 0.14	1.20 \pm 0.23	1.26 \pm 0.07	1.11 \pm 0.08	1.16 \pm 0.08	0.68 \pm 0.18