

Supplementary Information (SI)

**Tuning the Composition of Gold-Silver Bimetallic Nanoparticles for the
Electrochemical Reduction of Hydrogen Peroxide and Nitrobenzene**

N.S.K. Gowthaman, Bharathi Sinduja and S. Abraham John*

Centre for Nanoscience and Nanotechnology
Department of Chemistry, Gandhigram Rural Institute
Gandhigram-624 302, Dindigul, Tamilnadu, India
E-mail : abrajohn@yahoo.co.in

*Corresponding author: Tel: +91 451 245 2371; Fax : + 91 451 245 3031

E-mail: abrajohn@yahoo.co.in , s.abrahamjohn@ruraluniv.ac.in

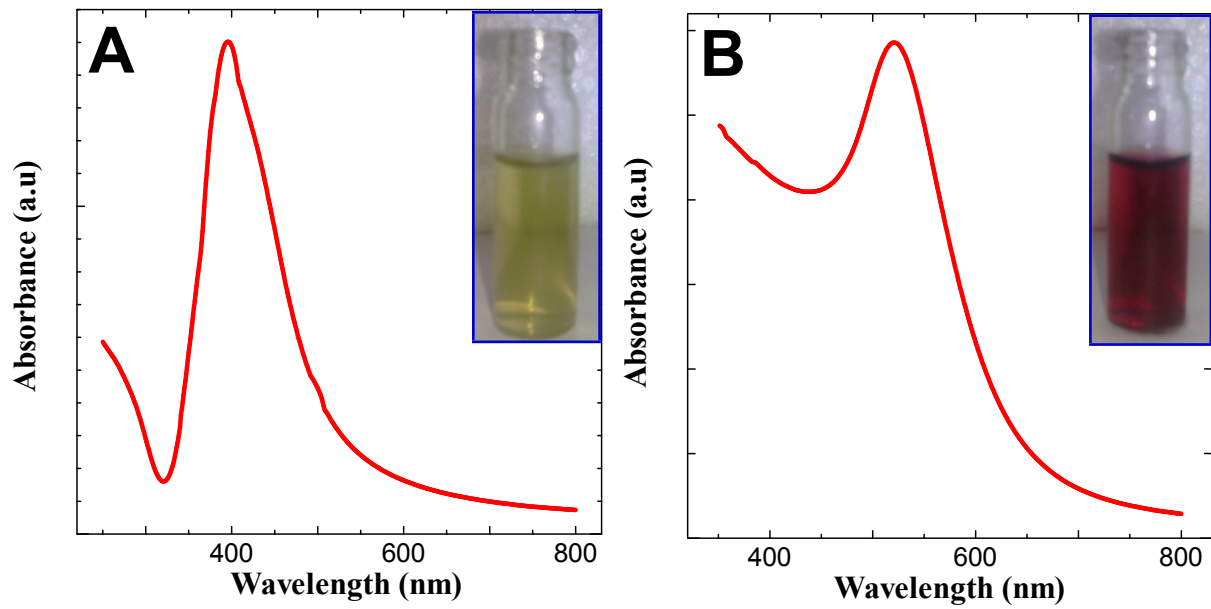


Figure S1. UV-vis spectra obtained for (A) AgNPs and (B) AuNPs. Insets: Photographs of (A) AgNPs and (B) AuNPs.

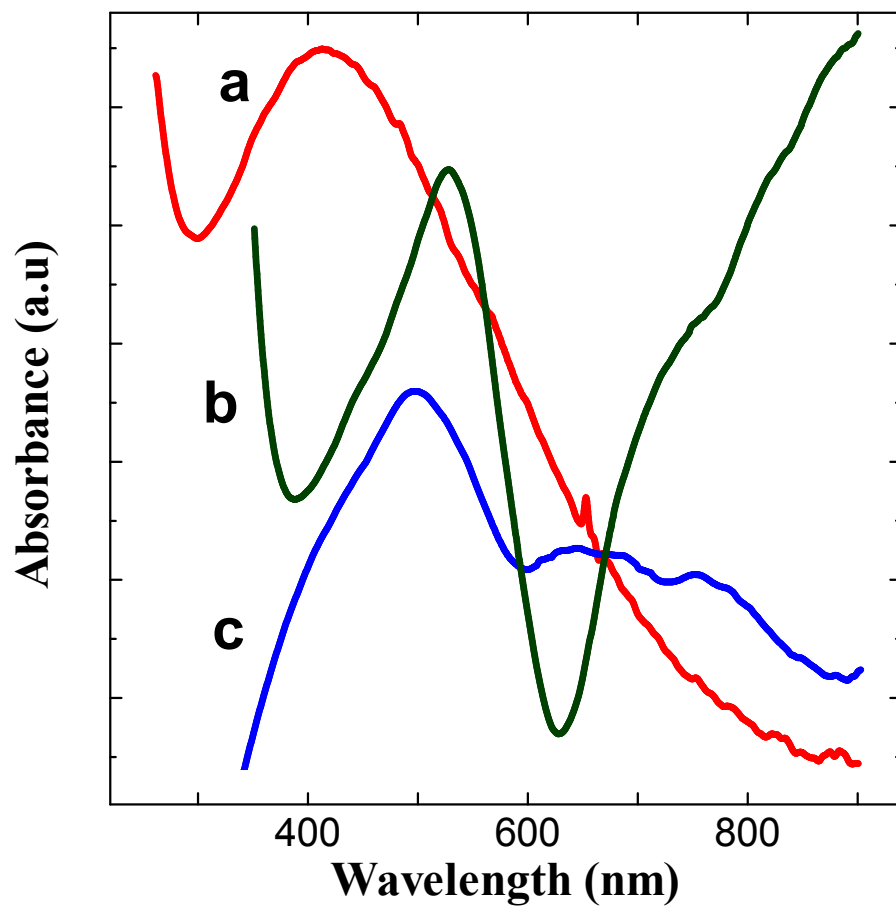


Figure S2. DRS obtained for (a) GC/HDA/AgNPs, (b) GC/HDA/AuNPs and (c) GC/HDA/Au-AgNPs electrodes.

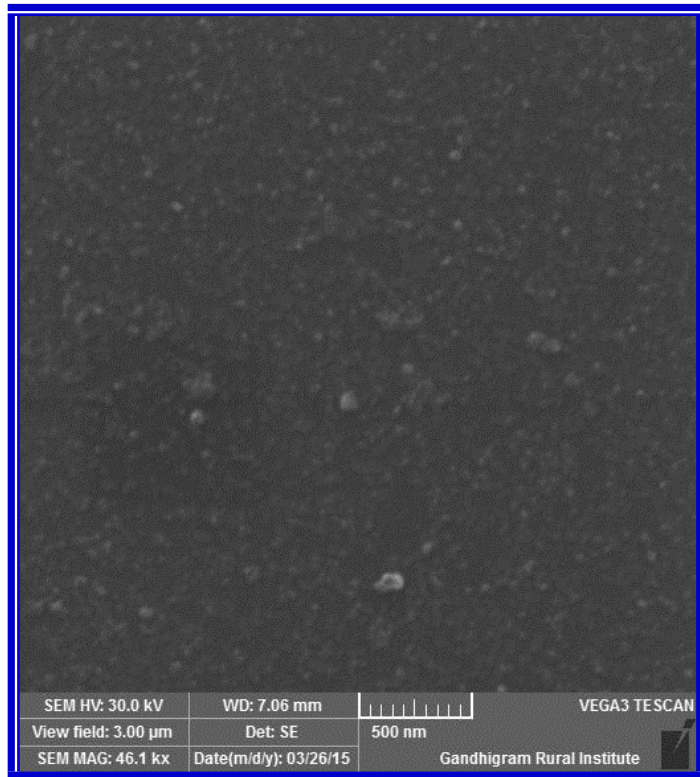


Figure S3. SEM image obtained for Au-AgNPs (Ag:Au=1:0.12) modified substrate.

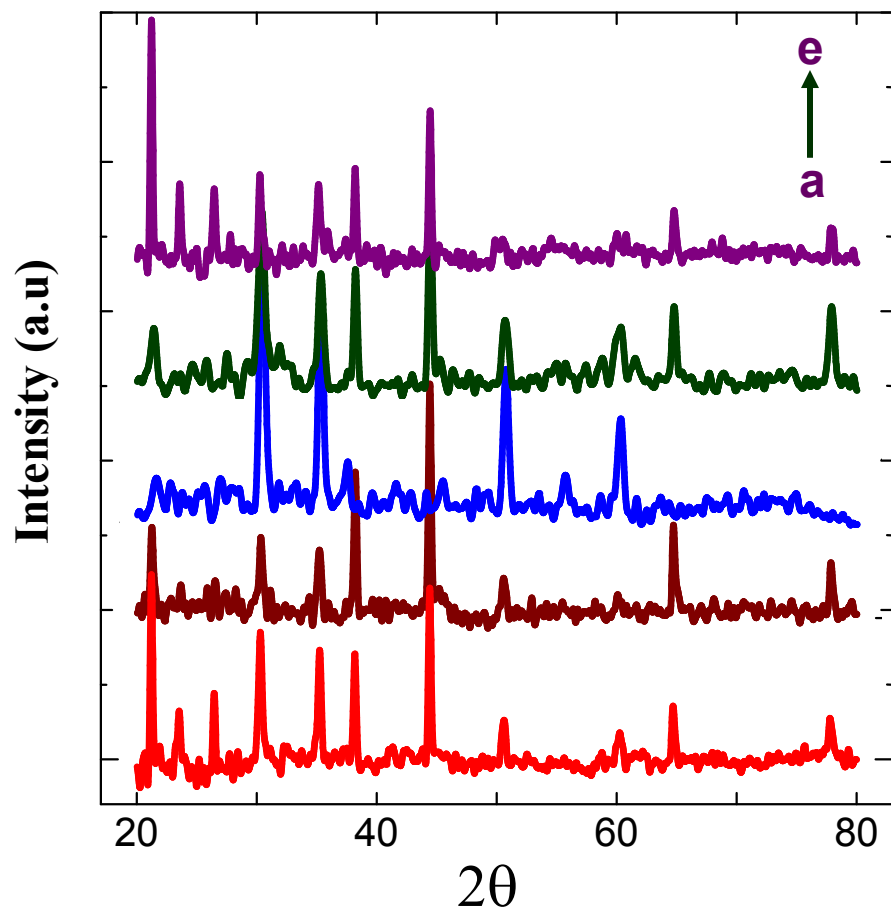


Figure S4. XRD patterns obtained for (a) AgNPs, (b) AuNPs and Au-AgNPs with the Ag:Au mole ratio of (c) 1:0.10, (d) 1:0.12 and (e) 1:0.14 modified ITO substrates.

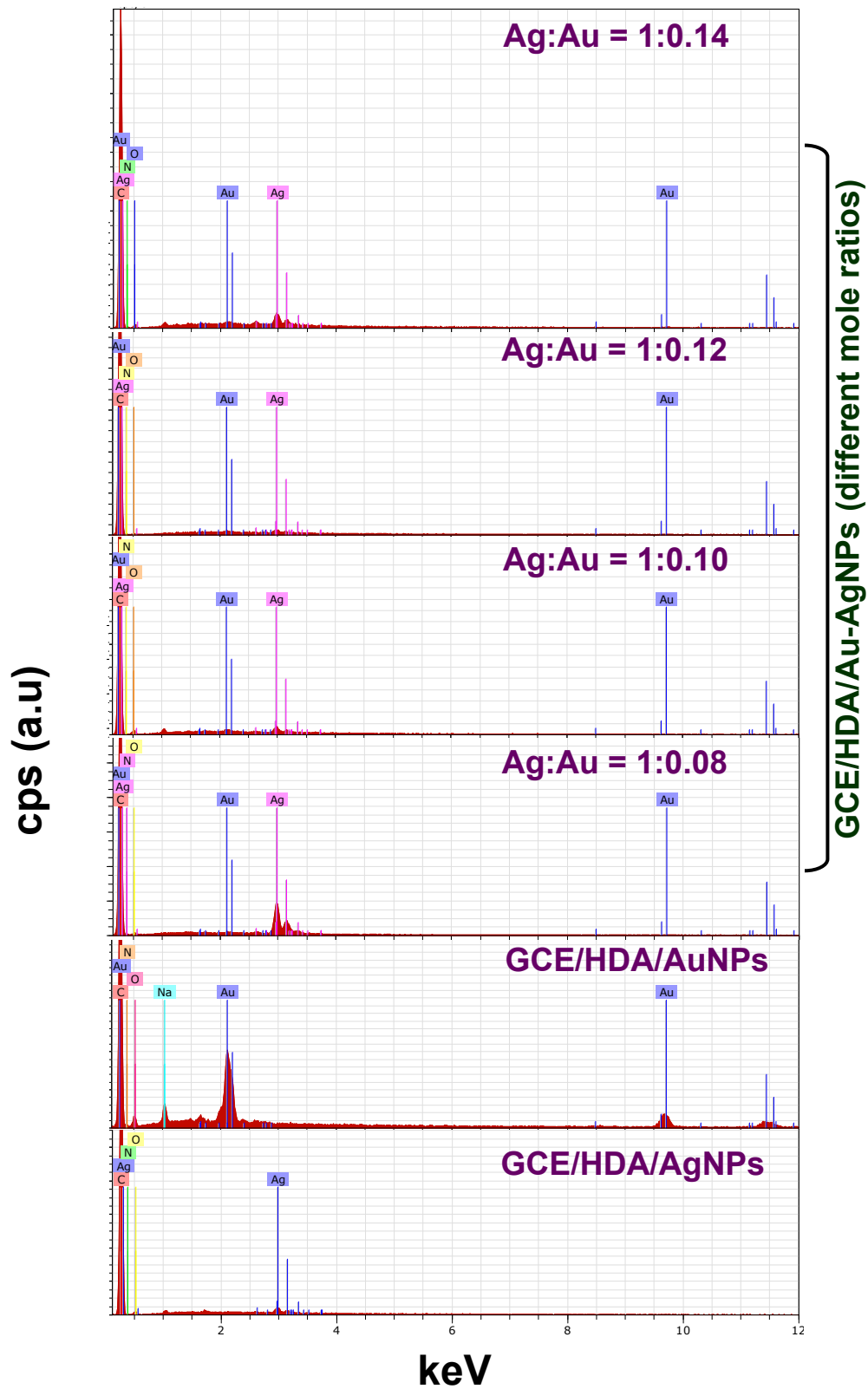


Figure S5. EDAX spectrum obtained for Au-AgNPs modified GC substrates.

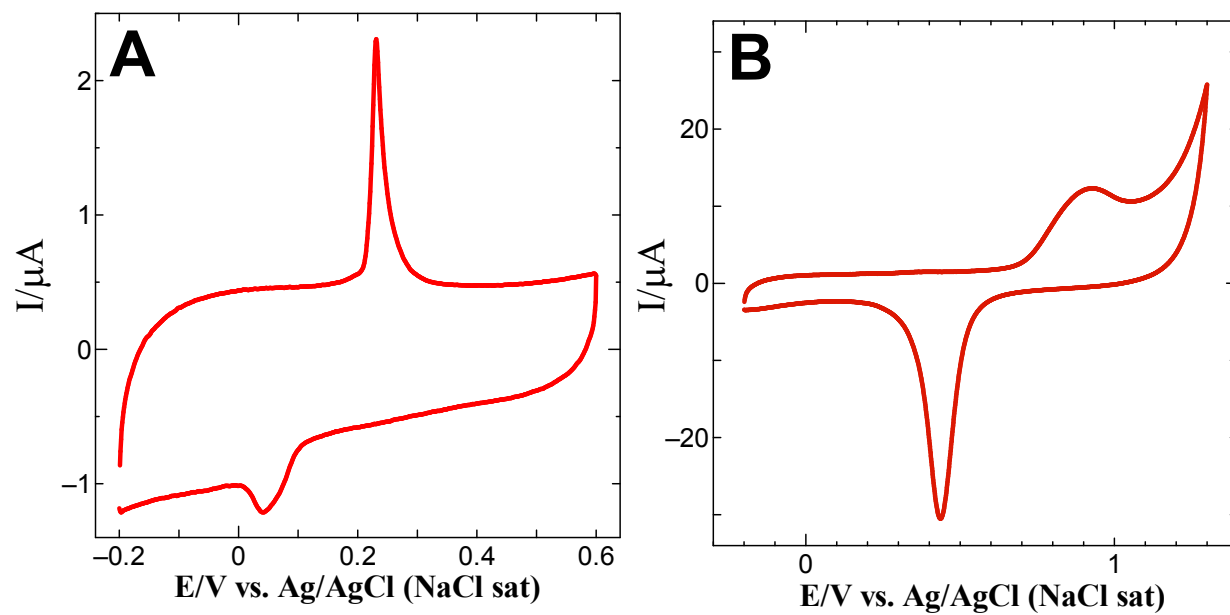


Figure S6. CVs obtained for (A) GC/HDA/AgNPs and (b) GC/HDA/AuNPs electrodes in 0.2 M PB solution (pH 7.2) at a scan rate of 50 mV s^{-1} .

Table S1. EDAX-Elemental analysis

S.No	Electrode	Weight % of the element present				
		C	N	O	Ag	Au
1	GCE/HDA/AgNPs	83.74	3.81	6.13	6.33	-
2	GCE/HDA/AuNPs	79.70	4.09	16.23	-	8.99
3	GCE/HDA/Au-AgNPs (Ag:Au = 1:0.08)	84.61	3.98	4.78	5.32	1.31
4	GCE/HDA/Au-AgNPs (Ag:Au = 1:0.10)	84.73	2.41	6.15	3.63	1.19
5	GCE/HDA/Au-AgNPs (Ag:Au = 1:0.12)	85.40	3.40	3.97	5.76	1.48
6	GCE/HDA/Au-AgNPs (Ag:Au = 1:0.14)	84.81	5.07	4.31	4.37	1.43

Table S2. Impedance spectral data

Parameter	GCE/HDA/ AgNPs	GCE/HDA/ AuNPs	GCE/HDA/Au-AgNPs (Ag: Au mole ratio)			
			1:0.10	1:0.12	1:0.14	1:0.16
R_s (k Ω)	0.138	0.129	0.136	0.151	0.131	0.150
CPE (μ F)	4.77×10^{-6}	1.49×10^{-5}	2.82×10^{-6}	6.11×10^{-6}	1.47×10^{-6}	5.96×10^{-6}
R_{CT} (k Ω)	49.94	27.70	18.23	17.16	19.73	23.64
k_{et} (cm s $^{-1}$)	5.39×10^{-8}	4.86×10^{-7}	3.29×10^{-7}	2.77×10^{-7}	3.40×10^{-7}	3.75×10^{-7}

Table S3. Comparison of onset reduction potential and limit of detection of HP obtained from the present modified electrode with the reported modified electrodes.

S.No	Electrodes	Onset potential (vs. AgCl)	pH	Linear range	Detection limit (μM)	Reference
1	Au@Ag nanorods/GCE	-0.30 V	6.5	0.02-7.02 mM	0.67	[20]
2	Au@Ag nanoparticles/GCE	-0.10 V*	7.2	0.005-15 mM	1.3	[34]
3	Au-Ag/BMT ^a -NF/GCE	-0.40 V	7.0	1-250 μM	-	[35]
4	Ag@Cu nanowires/GCE	-0.40 V	7.2	1-10 mM	3	[36]
5	Au@AgNPs/GCE	-0.40 V	7.2	10-110 μM	0.12	This work

^aBMT-NF: 1-Butyl-3-methylimidazolium tetrafluoroborate/Nafion, *(vs. SCE)

Table S4. Comparison of onset reduction potential and limit of detection of NB obtained from the present modified electrode with the reported modified electrodes.

S.No.	Electrodes	Onset potential (vs. AgCl)	pH	Linear range	Detection limit (μM)	Reference
1	Au/Ag-TPDT NRs/GCE	-0.50 V	7.0	-	-	[37]
2	TPDT-Ag NPs/GCE	-0.55 V	7.2	1-7 μM	1.0	[38]
3	AuNPs/GCE	-0.60 V	7.0	0.1-600 μM	0.016	[39]
4	PAA-AgNPs/GCE	-0.50 V	7.0	10-600 μM	1.68	[40]
5	RGO-AgNPs/GCE	-0.30 V	7.0	0.5-900 μM	0.26	[41]
6	Au@AgNPs/GCE	-0.30 V	7.2	10-120 μM	0.23	This work

TPDT: N^1 -[3-(trimethoxysilyl)propyl]diethylene triamine, *NRs*: Nanorods, *PAA-AgNPs*: AgNPs embedded in poly(amic) acid, *RGO*: reduced graphene oxide