Supplementary Information

Chromogenic and fluorescence sensing of pH with a Schiff-base molecule[†]

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Fig. s1: Mass spectrum of HMQB



Fig. s2: FT-IR spectrum of HMQB



Fig. s3: ¹H NMR spectrum of HMQB in CDCl₃



Fig. s4: ¹³C NMR spectrum of HMQB in DMSO-d₆



Fig. s5: The plot of emission intensities at 464 nm vs pH of the medium



Fig. s6: Fluorescence intensity of (1) HMQB and HMQB in the presence of (2) Na⁺, (3) K⁺, (4) Zn²⁺, (5) Mg²⁺, (6) Cu²⁺, (7) Mn²⁺, (8) Fe³⁺, (9) Co²⁺, (10) Ca²⁺, (11) Ni²⁺, (12) NO₃⁻, (13) Cl⁻, (14) SO₄²⁻, (15) PO₄³⁻ and (16) BO₃³⁻ at pH 5.0



Fig. s7: Fluorescence intensity of (1) HMQB and HMQB in the presence of (2) Na⁺, (3) K⁺, (4) Zn²⁺, (5) Mg²⁺, (6) Cu²⁺, (7) Mn²⁺, (8) Fe³⁺, (9) Co²⁺, (10) Ca²⁺, (11) Ni²⁺, (12) NO₃⁻, (13) Cl⁻, (14) SO₄²⁻, (15) PO₄³⁻ and (16) BO₃³⁻ at pH 8.0



Fig. s8: UV-visible spectrum of HMQB (1×10^{-4} M) in water of river Ganga at room temperature



Fig. s9: Fluorescence spectrum of HMQB (1×10^{-4} M) in water of river Ganga at room temperature



Fig. s10: UV-visible spectrum of HMQB (1×10^{-4} M) in lemon juice at room temperature



Fig. s11: Fluorescence spectrum of HMQB $(1 \times 10^{-4} \text{ M})$ in lemon juice at room temperature



Fig. s12: % cell viability of HeLa cells treated with different concentrations (1 mM and 10 mM) of HMQB for 12 h determined by MTT assay. Results are given here as mean \pm S.D. of three independent experiments.