

Supporting Information:

MoS₂ quantum dots decorated g-C₃N₄ composite photocatalyst with enhanced hydrogen evolution performance

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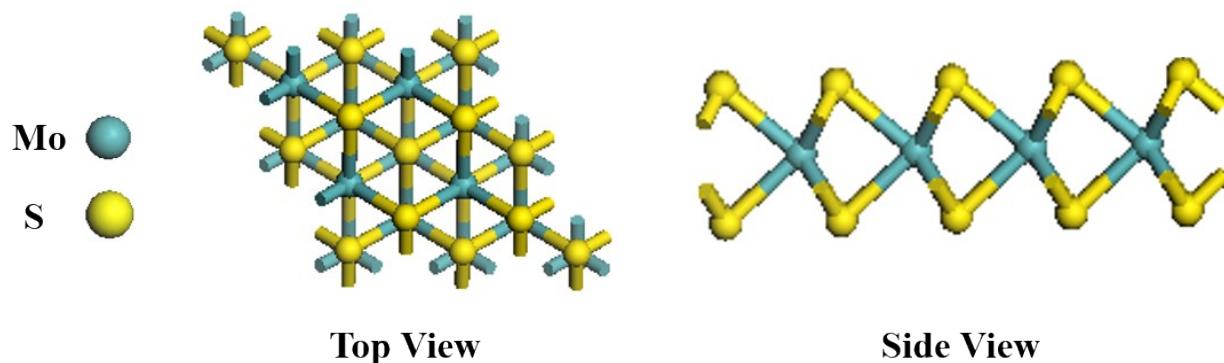


Figure S1. The crystal structure of MoS₂

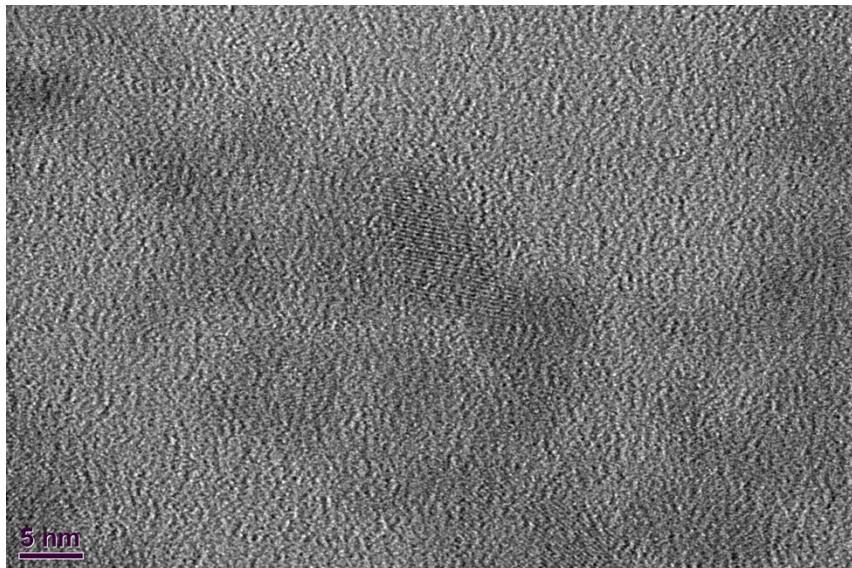


Figure S2. HRTEM images of 7wt% MoS₂-QDs/CN

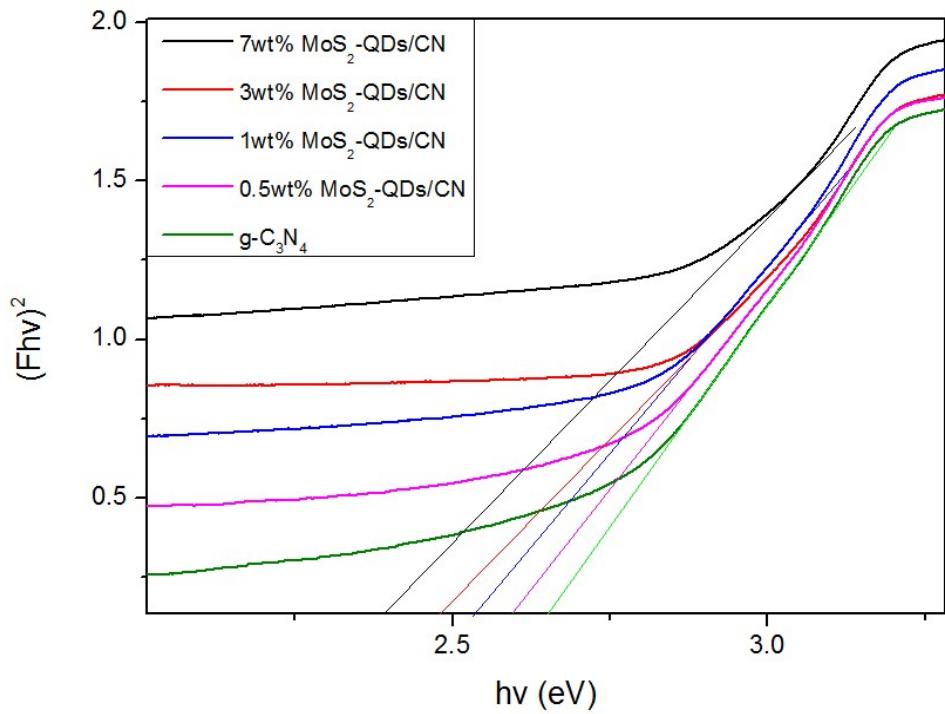


Figure S3. Band-gap calculation spectra of $\text{g-C}_3\text{N}_4$ and MoS_2 -QDs/CN composite

Table S1. Band-gap of g-C₃N₄ and MoS₂-QDs/CN composite

Sample	Band-gap (eV)
7wt% MoS ₂ -QDs/CN	2.38
3wt% MoS ₂ -QDs/CN	2.48
1wt% MoS ₂ -QDs/CN	2.53
0.5wt% MoS ₂ -QDs/CN	2.59
g-C ₃ N ₄	2.65

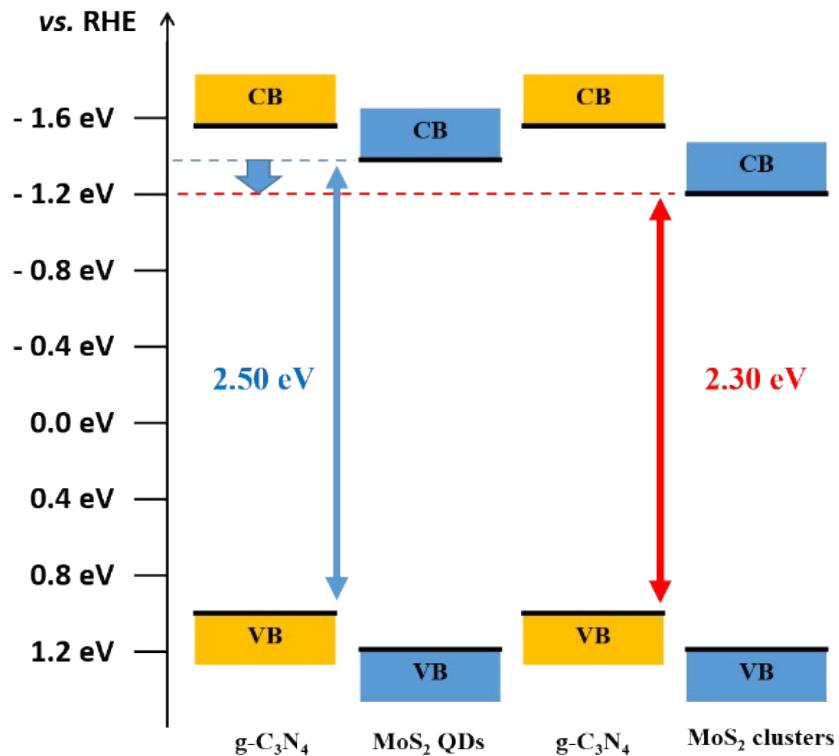


Figure S4. Explanation of the shrinkage of the band gap of the MoS_2 -QDs/CN composite: As increased amount of MoS_2 -QDs is introduced into $\text{g-C}_3\text{N}_4$, MoS_2 -QDs are more likely to form nanoclusters, thus narrow the band gap of MoS_2 -QDs and even the composite¹⁻⁴.

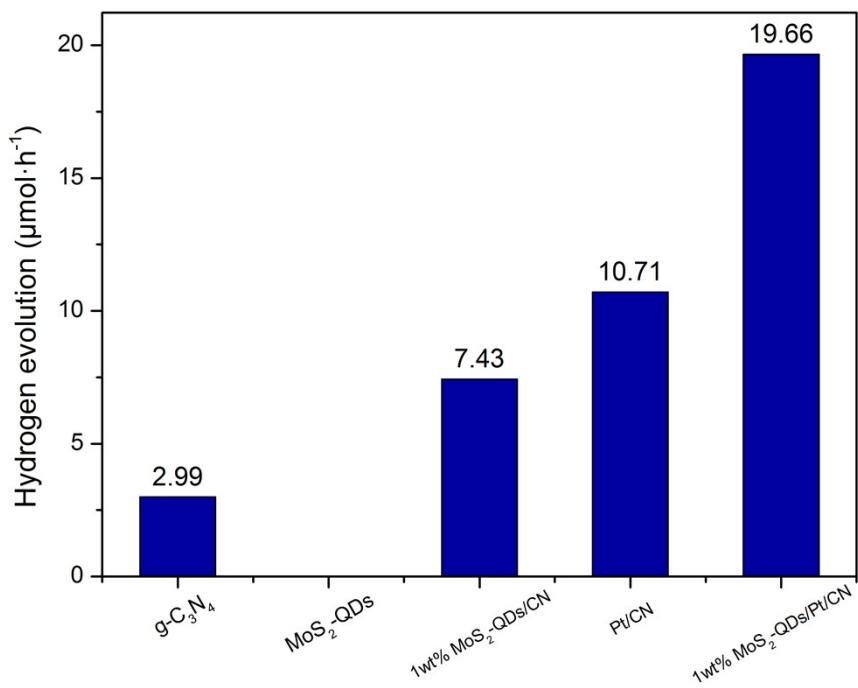


Figure S5. Photocatalytic H_2 evolution on $\text{g-C}_3\text{N}_4$ without Pt, $\text{MoS}_2\text{-QDs}$ without Pt, 1wt% $\text{MoS}_2\text{-QDs/CN}$ without Pt, $\text{g-C}_3\text{N}_4$ loaded with 1 wt% Pt (Pt/CN) and 1wt% $\text{MoS}_2\text{-QDs/CN}$ loaded with 1 wt% Pt (1wt% $\text{MoS}_2\text{-QDs/Pt/CN}$).

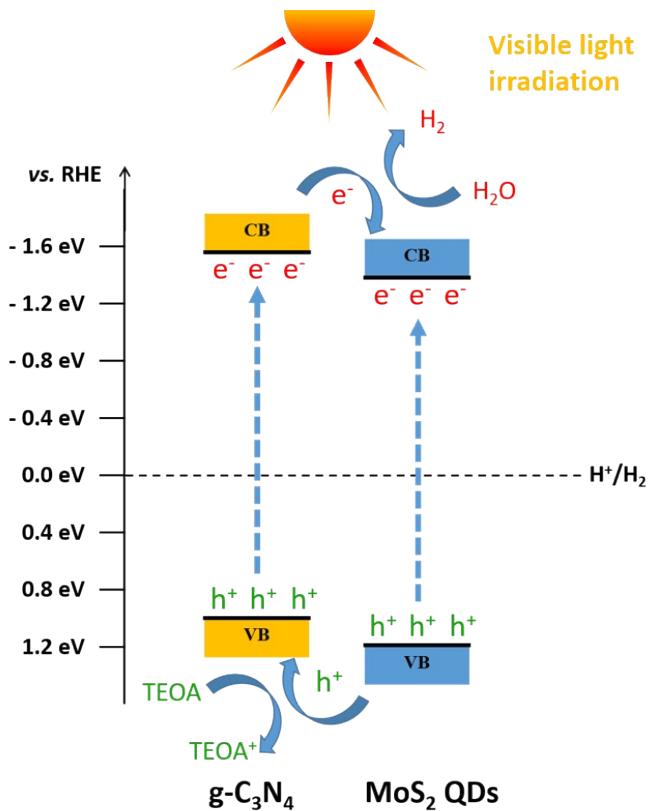


Figure S6. Schematic illustration of the electron-hole pair separation and transfer in the MoS₂-QDs/CN composite

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- 2 T. Z. Markus, M. Wu, L. Wang, D. H. Waldeck, D. Oron and R. Naaman, *J. Phys. Chem. C*, 2009, **113**, 14200-14206.
- 3 F. A. Frame, E. C. Carroll, D. S. Larsen, M. Sarahan, N. D. Browning and F. E. Osterloh, *Chem. Commun.*, 2008, **19**, 2206-2208.
- 4 L. Liao, Q. Zhang, Z. Su, Z. Zhao, Y. Wang, Y. Li, X. Lu, D. Wei, G. Feng, Q. Yu, X. Cai, J. Zhao, Z. Ren, H. Fang, F. Robles-Hernandez, S. Baldelli and J. Bao, *Nat. Nanotechnol.*, 2014, **9**, 69–73.